

*Supporting Information*

**Engineering crystal planes and band structure of 2D tin sulfide nanosheets and its photocatalytic degradation performance**

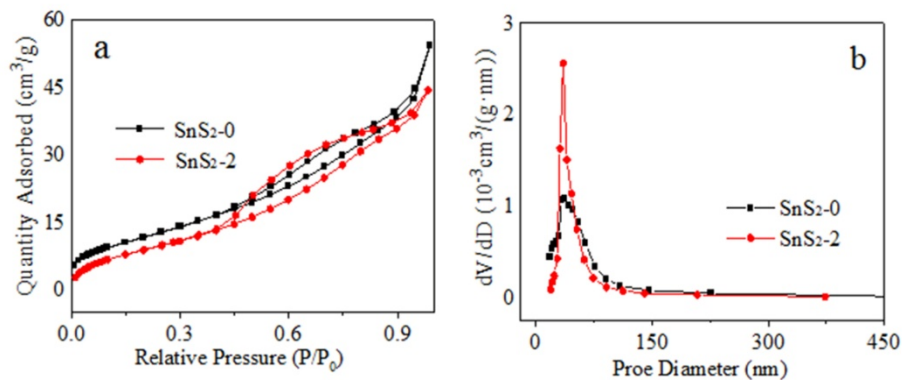
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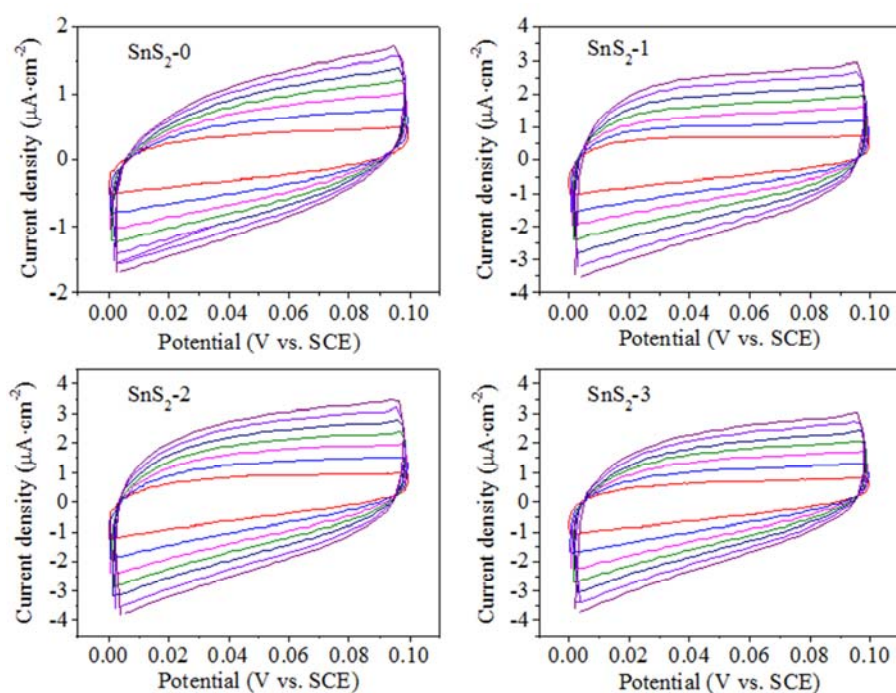
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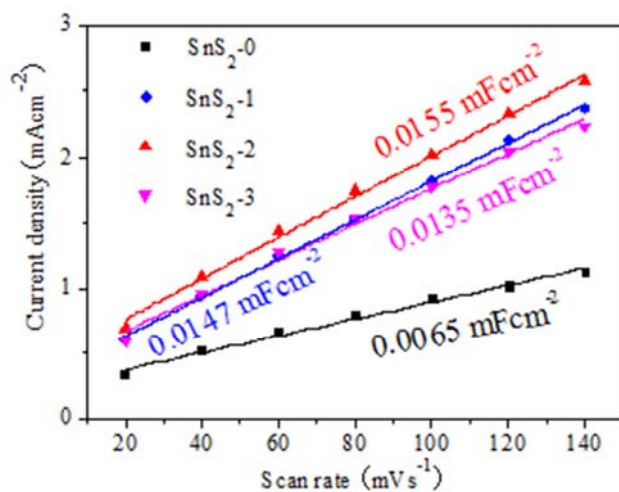
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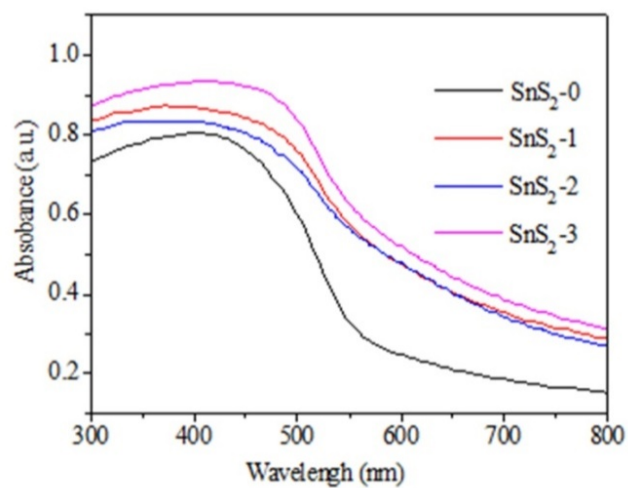
**Figure S1** (a)  $N_2$  adsorption-desorption isotherm and (b) pore size distribution of  $SnS_2-0$  and  $SnS_2-2$



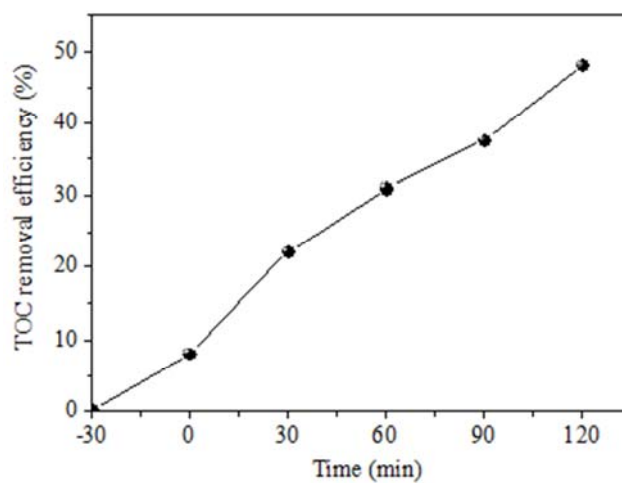
**Figure S2** Cyclic voltammety curves of  $SnS_2-x$  at different scan rates



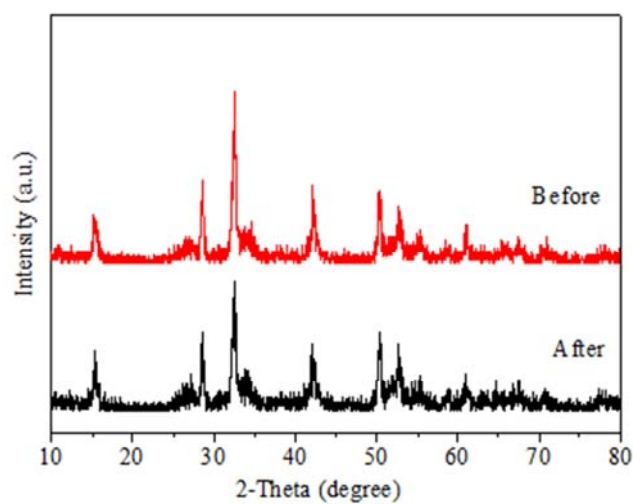
**Figure S3** The linear relationship between the current density and scan rate



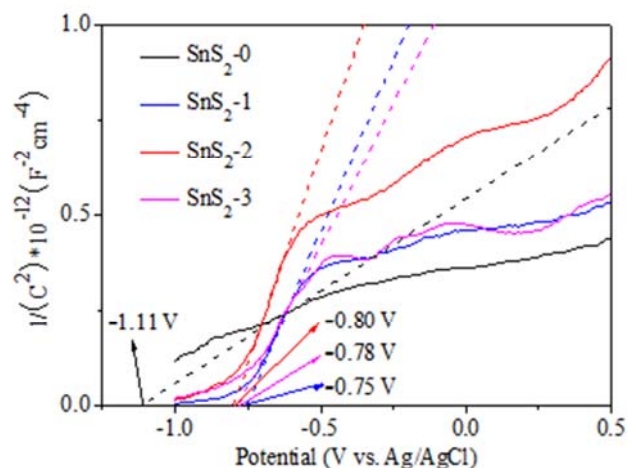
**Figure S4** UV-Vis diffuse reflectance spectra (DRS) of different SnS<sub>2</sub> materials



**Figure S5** TOC removal efficiency of SnS<sub>2</sub>-2 during the degradation of RhB



**Figure S6** XRD patterns of SnS<sub>2</sub>-2 before and after reactions



**Figure S7** Mott-Schottky plot of SnS<sub>2-x</sub> materials

**Table S1** the band gap, conduction band and valence band potential of SnS<sub>2-x</sub>

Catalyst	$E_g$	$E_{CB}$	$E_{VB}$
SnS <sub>2-0</sub>	1.91eV	-1.01eV	0.9eV
SnS <sub>2-1</sub>	1.87eV	-0.65eV	1.22eV
SnS <sub>2-2</sub>	1.42eV	-0.70eV	0.72eV
SnS <sub>2-3</sub>	1.55eV	-0.68eV	0.87eV

**Table S2** Photocatalytic activity of SnS<sub>2</sub> compared with other reported materials

Photocatalysts	Catal. dosage (g L <sup>-1</sup> )	Rh.B Concentration (mg L <sup>-1</sup> )	Light Source	Time (min)/ Degradation (%)	Ref.
SnO <sub>2</sub> /ZnS	1	10	500W Xe lamp	150/96	[1]
SnS <sub>2</sub> /NH <sub>2</sub> -MIL-125 (Ti)	1	40	300W Xe lamp 80 W	80/90.5	[2]
SnS nanoparticles	1	14.4	high-pressure mercury lamp	180/90.97	[3]
g-C <sub>3</sub> N <sub>4</sub> /BiOBr	1	10	300W Xe lamp 10 W	150/ 97.9	[4]
BiFeO <sub>3</sub> /SnS <sub>2</sub> NCs	0.1	15	tungsten halogen lamp	120/78	[5]
SnO <sub>2</sub> /TiO <sub>2</sub> /PVDF	N.A.	10	250W Xe lamp	240/92	[6]
SnS <sub>2</sub> -CdO	0.06	12	350W Xe lamp	210/86.11	[7]

ZnO/CA/SnS	0.6	4.8	Sunlight	175/99	[8]
SnS <sub>2</sub> -2	0.5	20	300W Xe lamp	120/94.5	This work

## References

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