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Supporting Information

Synthesis of functionalized disiloxanes with nonconventional fluorescence

by oxa-Michael addition reaction

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Experimental Section

Materials and Methods

All materials are used directly without purification unless otherwise specified. 1,3-Bis(3-hydroxypropyl)-1,1,3,3-tetramethyldisiloxane was provided by Shandong Dongyue Organosilicon Materials Co, LTD. Acrylonitrile (99%, containing MEHQ), methyl vinylsulfone (96%), phenylvinyl sulfone (99%), ethyl acrylate (99%, containing MEHQ), methyl acrylate (99.0%, containing ≤ 100 pppm MEHQ), N,N-dimethylpropanamide ($\geq 95.0\%$), phenyl acrylate (97%), and *t*-BuP₂ (100 µL, 2 M THF solution) were purchased from aladdin Co., LTD, China. Proton nuclear magnetic resonance (¹H NMR) spectra, carbon nuclear magnetic resonance (13 C NMR) and silicon nuclear magnetic resonance (29 Si NMR) spectra were recorded on a Bruker AVANCE 400 spectrometer at 25 °C using CDCl₃ as the solvent and without tetramethylsilane as an interior label. FT-IR were detected at room temperature with Bruker Tensor 27 Fourier Transform Infrared Spectrometer. Ultraviolet absorption (UV) spectra were detected at room temperature by using a Beijing TU-1901 double beam UV-vis

spectrophotometer. The fluorescent spectra of the samples were determined at room temperature with a Hitachi F-7000 fluorescence spectrophotometer using a monochromated Xe lamp as an excitation source.

Synthesis of 3,3'-((1,1,3,3-tetramethyldisiloxane-1,3-diyl)bis(propane-3,1-diyl))bis(oxy)di propionitrile (FDSi-1). Under argon atmosphere, 1,3-bis(3-hydroxypropyl)-1,1,3,3-tetramethyldisiloxane (BHTDS) (0.25 g, 1 mmol), acrylonitrile (0.106 g, 2 mmol), dichloromethane (5 mL), and t-BuP₂ (100 µL, 2 M THF solution) were added in a flask. The mixture was stirred at room temperature for 3 h. Then, acetic acid (2 mmol) was added in the mixture to neutralize the catalyst. The mixture was stood and the organic layer was separated and washed with water. After separating the organic layer, it was dried by anhydrous magnesium sulfate for 2 h. After filtration and removal of the solvent, the crude product was obtained and purified by column chromatography (using dichloromethane and methanol as eluents). The product was afforded as a brown liquid (0.333 g, yield: 93.5%). IR (KBr pellet cm⁻¹): 2933, 2874, 2250, 1415, 1368, 1255, 1187, 1116, 1055, 840. ¹H NMR (400 MHz, CDCl₃, ppm) δ 3.58 (t, *J* = 6.4 Hz, 4H), 3.38 (t, *J* = 6.9 Hz, 4H), 2.54 (t, *J* = 6.4 Hz, 4H), 1.61–1.48 (m, 4H), 0.52–0.40 (m, 4H), -0.01 (s, 12H). ¹³C NMR (100 MHz, CDCl₃, ppm) δ 118.0, 73.59, 64.74, 23.19, 18.85, 14.01, 0.30. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 7.73. HR-MS calcd for C₁₆H₃₂N₂O₃Si₂ [M+H]⁺: 357.2024, found 357.2031.

Synthesis of 1,1,3,3-tetramethyl-1,3-bis(3-(2-(methylsulfonyl)ethoxy)propyl)disiloxane (*FDSi-2*). The synthetic procedure and post-treatment of FDSi-2 were similar to those of FDSi-1 except acrylonitrile was replaced by methyl vinylsulfone (0.212 g, 2 mmol). The product was afforded as a yellow liquid (0.405 g, yield: 87.6%). IR (KBr pellet cm⁻¹): 2950, 1609, 1479, 1367, 1312, 1254, 1182, 1126, 1060, 782, 689, 487. ¹H NMR (400 MHz, CDCl₃, ppm) δ 3.80 (dd, *J* = 5.8, 4.9 Hz, 4H), 3.38 (t, *J* = 6.9 Hz, 4H), 3.20–3.14 (m, 4H), 2.97–2.88 (m, 6H), 1.59–1.46 (m, 4H), 0.49–0.40 (m, 4H), 0.00 (s, 12H). ¹³C NMR (100 MHz, CDCl₃, ppm) δ 73.88,

64.08, 55.03, 42.77, 23.01, 14.04, 0.29.²⁹Si NMR (75MHz, CDCl₃, ppm) δ 7.71. HR-MS calcd for C₁₆H₃₈O₇S₂Si₂ [M+NH₄]⁺: 480.1935, found 480.1926.

Synthesis of 1,1,3,3-tetramethyl-1,3-bis(3-(2-(phenylsulfonyl)ethoxy)propyl)disiloxane (*FDSi-3*). The synthetic procedure and post-treatment of FDSi-3 were similar to those of FDSi-1 except acrylonitrile was replaced by phenylvinyl sulfone (0.336 g, 2 mmol). The product was afforded as a yellow liquid (0.490 g, yield: 83.5%). IR (KBr pellet cm⁻¹): 2949, 1951, 1587, 1448, 1366, 1311, 1254, 1145, 1064, 840, 778, 733, 690, 531. ¹H NMR (400 MHz, CDCl₃, ppm) δ 7.94–7.88 (m, 4H), 7.68–7.60 (m, 2H), 7.57–7.52 (m, 4H), 3.77 (td, *J* = 6.3, 2.2 Hz, 4H), 3.40 (td, J = 6.3, 2.1 Hz, 4H), 3.25 (td, *J* = 6.9, 4.3 Hz, 4H), 1.43–1.33 (m, 4H), 0.40–0.30 (m, 4H), 0.01 (s, 12H). ¹³C NMR (100 MHz, CDCl₃, ppm) δ 139.72, 133.37, 128.84, 127.82, 73.71, 63.65, 56.04, 22.91, 13.87, 0.04. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 7.50. HR-MS calcd for C₂₆H₄₂O₇S₂Si₂ [M+NH₄]⁺: 604.2248, found 604.2235.

Synthesis of diethyl 3,3'-(((1,1,3,3-tetramethyldisiloxane-1,3-diyl))bis(propane-3,1-diyl)) bis(oxy))dipropionate (FDSi-4). The synthetic procedure and post-treatment of FDSi-4 were similar to those of FDSi-1 except acrylonitrile was replaced by ethyl acrylate (0.202 g, 2 mmol). The product was afforded as a pale yellow liquid (0.316 g, yield: 70.2%). IR (KBr pellet cm⁻¹): 2956, 1739, 1411, 1369, 1255, 1186, 1066, 841, 797. ¹H NMR (400 MHz, CDCl₃, ppm) δ 4.17– 4.02 (m, 4H), 3.65 (t, *J* = 6.2, 4.2 Hz, 4H), 3.52–3.28 (m, 4H), 2.59–2.44 (m, 4H), 1.67–1.49 (m, 4H), 1.26–1.14 (m, 6H), 0.52–0.41 (m, 4H), 0.01 (s, 12H). ¹³C NMR (101 MHz, CDCl₃, ppm) δ 171.46, 73.52, 65.77, 60.23, 34.94, 23.03, 22.32, 13.89, -0.54. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 7.72. HR-MS calcd for C₁₈H₃₈O₇Si₂ [M+NH₄]⁺: 468.2807, found 468.2736.

Synthesis of dimethyl 3,3'-(((1,1,3,3-tetramethyldisiloxane-1,3-diyl)bis(propane-3,1diyl))bis(oxy))dipropionate (FDSi-5). The synthetic procedure and post-treatment of FDSi-5 were similar to those of FDSi-1 except acrylonitrile was replaced by methyl acrylate (0.172 g, 2 mmol). The product was afforded as a pale yellow liquid (0.17 g, yield: 40.2%). IR (KBr pellet cm⁻¹): 2949, 1739, 1411, 1368, 1256, 1184, 1064, 840, 798, 704. ¹H NMR (400 MHz, CDCl₃, ppm) δ 4.12–3.96 (m, 4H), 3.74–3.58 (m, 6H), 3.43–3.26 (m, 4H), 2.54 (td, J = 6.4, 1.4 Hz, 4H), 1.69–1.44 (m, 4H), 0.53–0.40 (m, 4H), 0.04– -0.03 (m, 12H).¹³C NMR (100 MHz, CDCl₃, ppm) δ 171.90, 73.61, 66.75, 51.38, 34.70, 23.12, 22.34, 14.16, -0.28. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 8.30. HR-MS calcd for C₂₀H₄₂O₇Si₂ [M+NH₄]⁺: 440.2493, found 440.2407.

Synthesis of 3,3'-(((1,1,3,3-tetramethyldisiloxane-1,3-diyl)bis(propane-3,1-diyl))bis(oxy)) diphenyldipropionate (FDSi-6). The synthetic procedure and post-treatment of FDSi-6 were similar to those of FDSi-1 except acrylonitrile was replaced by phenyl acrylate (0.297 g, 2 mmol). The product was afforded as a pale yellow liquid (0.054 g, yield: 10.0%). IR (KBr pellet cm⁻¹): 2956, 1732, 1600, 1500, 1401, 1365, 1298, 1254, 1183, 1110, 1068, 988, 840, 797, 485. ¹H NMR (400 MHz, CDCl₃, ppm) δ 7.33–7.31 (m, 4H), 7.28–7.14 (m, 2H), 7.07–7.04 (m, 4H), 4.05 (t, *J* = 6.9 Hz, 2H), 3.51 (t, *J* = 6.9 Hz, 2H), 2.57 (t, *J* = 6.9 Hz, 4H), 1.65–1.48 (m, 4H), 0.50–0.42 (m, 4H), 0.00 (s, 12H). ¹³C NMR (101 MHz, CDCl₃, ppm) δ 170.76, 166.26, 130.43, 129.14, 128.32, 121.32, 119.74, 115.12, 73.64, 67.00, 35.09, 26.27, 22.35, 13.80, -0.15. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 7.77. HR-MS calcd for C₂₀H₄₄N₂O₅Si₂ [M+H]⁺: 449.2861, found 449.2865.

Synthesis of 3,3'-(((1,1,3,3-tetramethyldisiloxane-1,3-diyl)bis(propane-3,1diyl))bis(oxy))bis (N,N-dimethylpropanamide) (FDSi-7). The synthetic procedure and posttreatment of FDSi-7 were similar to those of FDSi-1 except acrylonitrile was replaced by N,Ndimethylpropanamide (0.1983 g, 2 mmol). The product was afforded as a pale yellow liquid (0.0381 g, yield: 8.5%). IR (KBr pellet cm⁻¹): 2929, 1648, 1401, 1366, 1298, 1255, 1183, 1110, 1060, 988, 840, 797, 485. ¹H NMR (400 MHz, CDCl₃, ppm) δ 3.69 (t, *J* = 6.9 Hz, 4H), 3.36 (t, *J* = 7.1 Hz, 4H), 2.99 (s, 6H), 2.90 (s, 6H), 2.57 (t, *J* = 6.9 Hz, 4H), 1.55–1.49 (m, 4H), 0.49– 0.41 (m, 4H), 0.00 (s, 12H). ¹³C NMR (101 MHz, CDCl₃, ppm) δ 170.76, 73.64, 66.63, 36.75, 34.97, 33.47, 23.17, 13.90, 0.14. ²⁹Si NMR (75 MHz, CDCl₃, ppm) δ 8.01. HR-MS calcd for C₂₀H₄₄N₂O₅Si₂ [M+H]⁺: 449.2861, found 449.2865.

Synthesis of 3,3'-(hexane-1,6-diylbis(oxy))dipropionitrile (FD-J). The synthetic procedure and post-treatment of FD-J were similar to those of FDSi-1 except BHTDS was replaced by 1,6-hexanediol (0.1181 g, 1 mmol). The product was afforded as a brown liquid (0.222 g, yield: 99.0%). IR (KBr pellet cm⁻¹): 2933, 2245, 1461, 1366, 1185, 1114, 1064, 989, 817, 747, 666, 487. ¹H NMR (400 MHz, CDCl₃, ppm) δ 3.65 (t, *J* = 6.4 Hz, 4H), 3.50 (t, *J* = 6.5 Hz, 4H), 2.61 (t, *J* = 6.4 Hz, 4H), 1.64–1.57 (m, 4H), 1.40 (ddd, *J* = 7.3, 4.5, 3.1 Hz, 4H).



Scheme S1 The possible mechanism of transesterification reaction of BHTDS and ethyl acrylate during the oxa-Michael addition reaction

HO OH + Catalyst N O O

Scheme S2 Synthetic route of FD-J



Fig. S1 ¹H NMR spectra of the mixture of 1,3-bis(3-hydroxypropyl)-1,1,3,3-



tetramethyldisiloxane and ethyl acrylate at different reaction times

Fig. S2 ¹H NMR spectra of the products from orthogonal experiments



Fig. S3 The photographs of functionalized disiloxanes (FDSi) in CH_2Cl_2 solution (M = 10⁻⁷ mol/L) under 365 nm UV light



Fig. S4 The photographs of functionalized disiloxane FDSi-1 (a) and FDSi-2 (b) at different concentrations in CH₂Cl₂ solution under 365 nm UV light



Fig. S5 The photographs of FDSi-1 (a) and FDSi-2 (b) in dichloromethane-petroleum ether solvent mixture (0.01 mol/L) under 365 nm UV light



Fig. S6 Fluorescence spectra of FDSi-2 dissolved in different solvents (0.01 mol/L, λ_{ex} =365



Fig. S8¹³C NMR spectra of FDSi-1



Fig. S9²⁹Si NMR spectra of FDSi-1



Fig. S10 FT-IR spectra of FDSi-1



Fig. S11 ¹H NMR spectra of FDSi-2



Fig. S12 ¹³C NMR spectra of FDSi-2



Fig. S14 FT-IR spectra of FDSi-2



Fig. S15 ¹H NMR spectra of FDSi-3



Fig. S16¹³C NMR spectra of FDSi-3



Fig. S17²⁹Si NMR spectra of FDSi-3



Fig. S18 FT-IR spectra of FDSi-3







Fig. S20¹³C NMR spectra of FDSi-4



Fig. S21 ²⁹Si NMR spectra of FDSi-4



Fig. S22 FT-IR spectra of FDSi-4











Fig. S25²⁹Si NMR spectra of FDSi-5



Fig. S26 FT-IR spectra of FDSi-5







Fig. S28 ¹³C NMR spectra of FDSi-6



Fig. S29²⁹Si NMR spectra of FDSi-6



Fig. S30 FT-IR spectra of FDSi-6







Fig. S32 ¹³C NMR spectra of FDSi-7



Fig. S33 ²⁹Si NMR spectra of FDSi-7



Fig. S34 FT-IR spectra of FDSi-7



Fig. S35 ¹H NMR spectra of FD-J



Fig. S36 FT-IR spectra of FD-J



Fig. S37 HR-MS spectra of FDSi-1



Fig. S38 HR-MS spectra of FDSi-2



Fig. S39 HR-MS spectra of FDSi-3



Fig. S40 HR-MS spectra of FDSi-4



Fig. S41 HR-MS spectra of FDSi-5



Fig. S42 HR-MS spectra of FDSi-6



Fig. S43 HR-MS spectra of FDSi-7