

## Supporting Information

### Preparation of metal phosphide derived from dual-ligand NiFe-MOF and its boosting activity toward electrolysis of water

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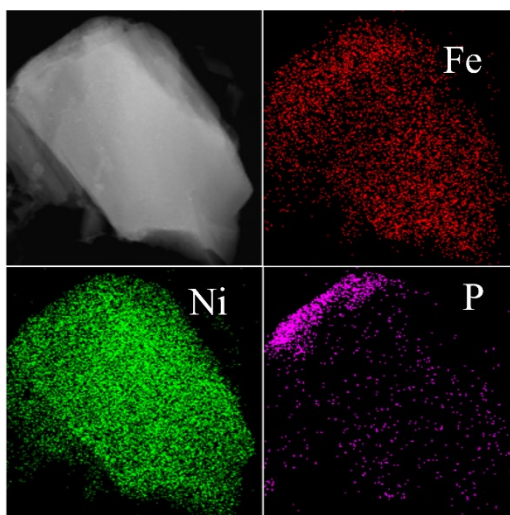


Fig. S1 elemental mapping image of P-FeNi

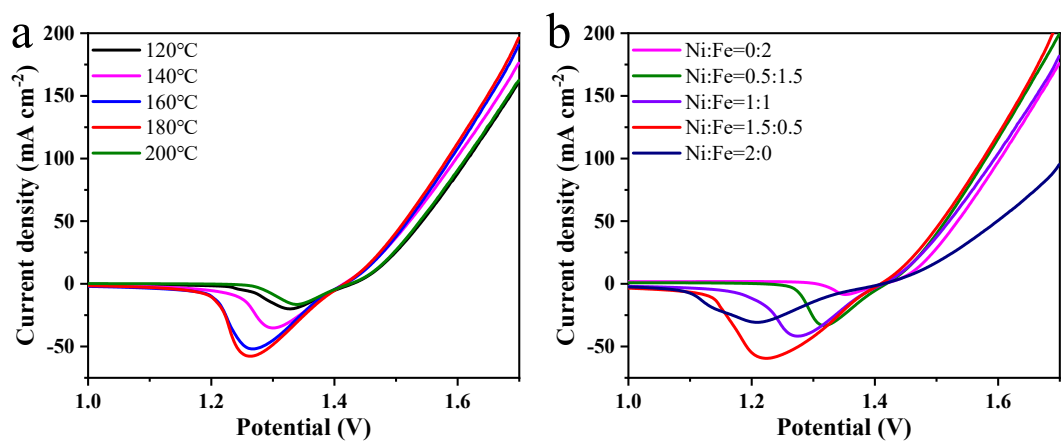


Fig. S2 LSV curves of different hydrothermal temperature (a) and different nickel-iron ratio (b) of P-FeNi@NFF.

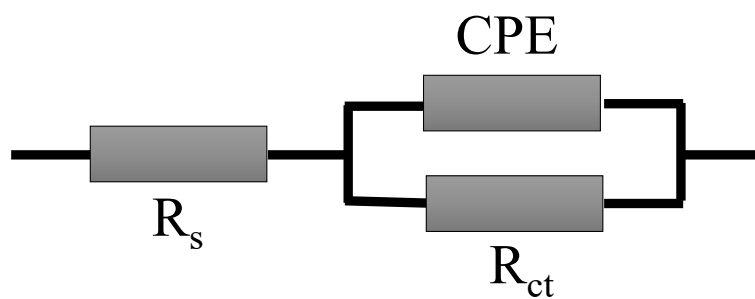


Fig. S3 The equivalent circuit model of EIS Nyquist diagram is fitted.  $R_s$ : solution resistance;  $R_{ct}$ : charge transfer resistance; CPE: Other resistors.

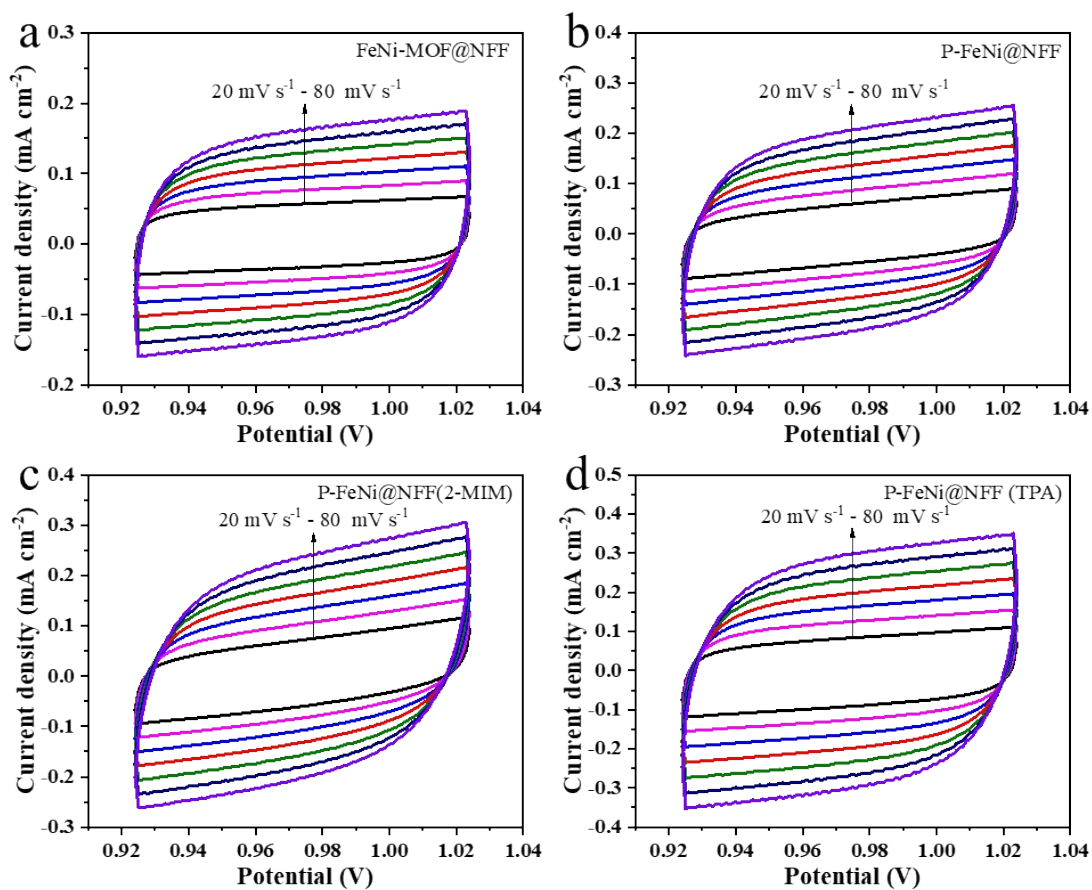


Fig. S4 CV diagrams of FeNi-MOF@NFF (a), P-FeNi@NFF (b), P-FeNi@NFF(2-MIM) (c), and P-FeNi@NFF(TPA) (d) at different sweep rates of 20-80 mV s<sup>-1</sup>.

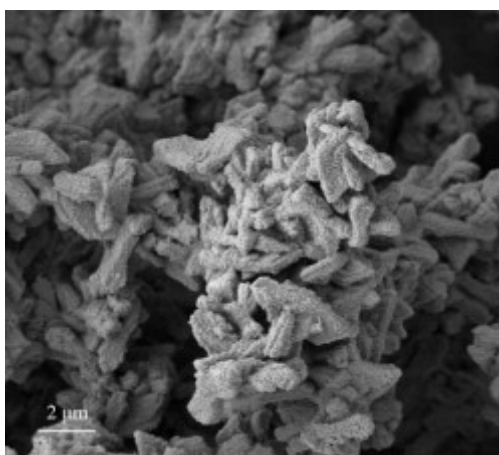


Fig. S5 SEM of P-FeNi@NFF after stability test

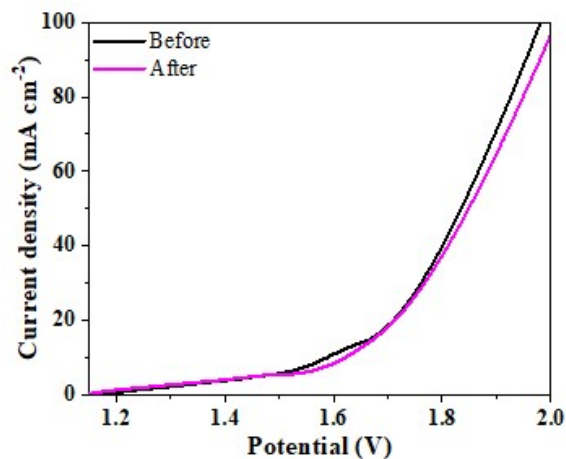


Fig. S6 LSV comparison of P-FeNi@NFF before and after stability test

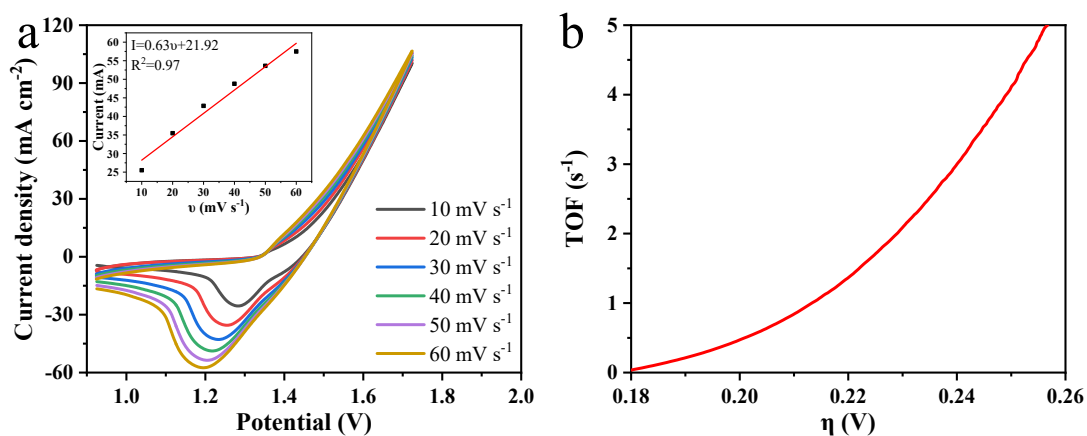


Fig. S7 CV curves of P-FeNi@NFF as the scanning rate increases from  $10 \text{ mV s}^{-1}$  to  $80 \text{ mV s}^{-1}$  (Inset: linear relationship between peak oxidation current and scanning rate). (d) The relationship between TOF of P-FeNi@NFF and overpotential.

## Tables

Table S1 Performance comparison of materials for OER

Catalysts	Overpotential/mV		$Z/\Omega$	
	at $10 \text{ mA cm}^{-2}$	at $100 \text{ mA cm}^{-2}$	$R_s$	$R_{ct}$
FeNi-MOF@NFF	202	282	3.75	6.91
P-FeNi@NFF	194	241	2.07	0.62
P-FeNi@NFF(2-MIM)	232	287	2.04	1.19
P-FeNi@NFF(TPA)	201	251	2.25	0.64

Table S2 Performance comparison of self-supported catalysts for OWS

Electrolytic cell	Cell voltage/at 10 mA cm <sup>-2</sup>	Ref.
P-FeNi@NFF    P-FeNi@NFF	1.59 V	This work
Fe-MOF/Au-8/FF    Fe-MOF/Au-8/FF	1.61 V	[1]
NiSP/NF    NiSP/NF	1.70 V	[2]
NiCoP/NF    NiCoP/NF	1.60 V	[3]
NiFe-MOF-74/NF    NiFe-MOF-74/NF	1.58 V	[4]
Fe,Rh-Ni2P/NF    Fe,Rh-Ni2P/NF	1.62 V	[5]
DLC FCP@NG    DLC FCP@NG	1.63 V	[6]
CoNi2S4/Ni3S2@NF    CoNi2S4/Ni3S2@NF	1.65 V	[7]

## References

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