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Electronic Supplementary Information (ESI)

Concise Synthesis of A Fused Energetic Pyrazolotriazine with Good

Performances

Haichuan Shi^a, Yuqing Wang^a, Xinyu Wang^a, Xiaotian Zhang^a, Ming Huang^b, Haijun Yang^{*a}, and Zhenghang Luo^{*a}

a.School of Materials and Chemistry, Southwest University of Science and Technology,

Mianyang, 621010, China.

b.Institute of Chemical Materials, China Academy of Engineering Physics, Mianyang, 621999,

China.

*E-mail address of Corresponding author: yanghaijun@swust.edu.cn

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Experimental section

Caution!

As energetic materials have explosive risks, all operations should be carried out with helmets and behind safety shields. Mechanical actions such as friction and impact, as well as static electricity should be avoided.

General Methods:

Materials: 95% fuming nitric acid with an analytical purity was obtained from Chengdu Kelong Chemical Co., Ltd.; Other chemicals were purchased from Aladdin Reagent Company (Shanghai, China) and were used as supplied.

Product characterization: NMR spectra were recorded on Bruker Avance 600 nuclear magnetic resonance spectrometer; IR spectra were determined on Spectrum 100 Fourier transform infrared spectrometer by KBr pellet technique in the range of 400-4000 cm⁻¹; High-resolution mass spectra were performed on a Waters SYNAPT G2-Si mass spectrometer using electrospray ionization (ESI); TG/DSC tests were carried out on Jupiter STA449C comprehensive thermal analyzer; Impact and friction sensitivity measurements were made using a standard BAM Fallhammer and a BAM friction tester. The mechanical sensitivities of **GD-1** were tested according to GJB772 A-97. In fallhammer experiments, the weight of the fallhammer was 10 ± 0.01 kg, and the mass of **GD-1** was 50 ± 1 mg for each shot. The intensity data sets of **GD-1** were collected on a Agilent Xcalibur, Eos, Gemini CCD diffractometer equipped with a graphite-monochromated Mo Ka radiation (I = 0.71073 Å) at 293 K. The data sets were reduced by the CrysAlisPro program. The exposure time is set as 150 s with a scan width of 1 degree.



Synthesis of 4-amino-3-nitropyrazolo[5,1-*c*][1,2,4]triazine (2)

To a 20% aqueous solution of H_2SO_4 (50-100 mL) cooled with an ice-water bath was added 5aminopyrazole (1.00 g, 12.05 mmol). Then, NaNO₂ (1.20 g, 24.10 mmol) dissolved in 5 mL water was added dropwise at 0-5 °C. The resulting mixture was stirred at room temperature for 2h, giving solution A. To 6 mL of ice water was added NaOH (1.00 g, 24.00 mmol) and Nitroacetonitrile (NAN) (2.10 g, 24.00 mmol) with stirring, giving solution B. Solution B was added dropwise to solution A cooled with an ice-water bath. The reaction solution was stirred in ice-water bath overnight. A yellow precipitate was filtered, washed with water and dried, giving 1.95 g of **2** with a yield of 90%. ¹H NMR (DMSO-*d*₆, 600MHz) δ = 7.31 (s, 1H), 8.56 (s, 1H), 9.53 (s, 1H), 9.99 (s, 1H) ppm. ¹³C NMR (DMSO-*d*₆, 150MHz) δ = 100.51, 135.93, 139.46, 147.84, 149.71 ppm. FT-IR (KBr): v = 3446, 1647, 1552, 1489, 1460, 1422, 1329, 1236, 1225, 1192, 839, 804, 669, 588 cm⁻¹. MS (ESI): m/z = 181.04651 [M+H]⁺.

Synthesis of 4-amino-3,8-dinitro-pyrazolo[5,1-*c*][1,2,4]triazine (GD-1)

To 95% fuming HNO₃ (10 mL) was added **2** (1.00 g, 5.60 mmol) with stirring at room temperature. The mixture was stirred at r.t. for 12 h, and then poured onto crushed ice. Filtration, washing with water and drying gave 1.01 g of bright yellow **GD-1** with a yield of 81%. ¹H NMR (DMSO-*d*₆, 600MHz) δ = 9.25 (s, 1H), 9.93 (s, 1H), 10.50 (s, 1H) ppm. ¹³C NMR (DMSO-*d*₆, 150MHz) δ = 123.90, 139.07, 139.84, 143.55, 144.27 ppm. FT-IR (KBr): *v* = 3299, 3067, 2361, 1674, 1632, 1571, 1520, 1442, 1399, 1342, 1325, 1289, 1226, 1192, 1014, 955, 844, 818, 767, 689 cm⁻¹. MS (ESI): *m/z* = 226.03124 [M+H]⁺.

NMR and MS Spectra





Figure S3 MS spectrum of 2





Figure S6 MS spectrum of GD-1

Crystallographic data

(1) $GD-1\cdot 2H_2O$

The crystal structure of $GD-1\cdot 2H_2O$ is based on extremely weak data (CCDC2239586).

$1 \cdot 2H_2O$	
CCDC	2239586
Empirical formula	$C_5H_7N_7O_6$
Formula weight	261.18
Temperature/K	293(2)
Crystal system	monoclinic
Space group	P2 ₁ /m
a/Å	9.2737(13)
b/Å	6.2094(8)
c/Å	9.4529(12)
α/°	90
β/°	114.864(16)
$\gamma/^{\circ}$	90
Volume/Å ³	493.88(13)
Z	2
pcalcg/cm ³	1.756
µ/mm ⁻¹	0.160
F(000)	270.0
Crystal size/mm ³	$0.22\times0.2\times0.18$
Radiation	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/°	8.086 to 58.292
Index ranges	-12 \leq h \leq 12, -8 \leq k \leq 8, -12 \leq l \leq 12
Reflections collected	6317
Independent reflections	1318 [$R_{int} = 0.0383$, $R_{sigma} = 0.0966$]
Data/restraints/parameters	1318/0/109
Goodness-of-fit on F ²	0.947
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0837, wR_2 = 0.1914$
Final R indexes [all data]	$R_1 = 0.1584, wR_2 = 0.2129$
Largest diff. peak/hole / e Å-3	0.71/-0.22

Table S1 Crystal data and structure refinement for GD-

(2) **GD-1**·CH₃OH

Table I Crystal data and stru	cture remiement for GD-1-CH ₃ OH
CCDC	2239587
Empirical formula	$C_6H_7N_7O_5$
Formula weight	257.19
Temperature/K	298.15
Crystal system	orthorhombic
Space group	Pna2 ₁
a/Å	8.5784(8)
b/Å	26.550(2)
c/Å	4.5184(5)
α/°	90
β/°	90
$\gamma/^{\circ}$	90
Volume/Å ³	1029.09(17)
Z	4
$\rho_{calc}g/cm^3$	1.660
µ/mm ⁻¹	0.814
F(000)	528.0
Crystal size/mm ³	0.22 imes 0.2 imes 0.18
Radiation	$GaK\alpha (\lambda = 1.34138)$
2Θ range for data collection/°	10.682 to 114.222
Index ranges	$\textbf{-10} \leq h \leq 10,\textbf{-32} \leq k \leq 33,\textbf{-5} \leq l \leq 3$
Reflections collected	8708
Independent reflections	1830 [$R_{int} = 0.0501, R_{sigma} = 0.0389$]
Data/restraints/parameters	1830/1/165
Goodness-of-fit on F ²	1.111
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0371, wR_2 = 0.0852$
Final R indexes [all data]	$R_1 = 0.0492, wR_2 = 0.0932$
Largest diff. peak/hole / e Å $^{\text{-}3}$	0.14/-0.20
Flack parameter	0.0(6)

Table 1 Crystal data and structure refinement for GD-1·CH₃OH



Figure S8 TG curve of previously dried GD-1 (10 °C/min)

GD-1 sample for TG analysis was obtained via drying in vacuum at 100 °C for 6 h and then stored in air at room temperature for more than 24 h.