

Dual-aptamer-based colorimetric assay for the accurate identification of circulating tumor cells via Fe₃O₄@Pt NP nanozymes and G-quadruplex/hemin for signal amplification

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The DNA sequence, buffer solutions and apparatus involved in this work were as follows:

All DNA was HPLC-purified and provided by Sangon Biotechnology Co., Ltd. (Shanghai, China). Their sequences were as follows: MUC1 aptamer (Apt_{MUC1}, 5'-biotin- TTT TTG CAG TTG ATC CTT TGG ATA CCC TGG -3'), AS1411 aptamer (Apt_{AS1411}, 5'-biotin- TTT TTG GTG GTG GTG GTT GTG GTG GTG GTG G-3'), signal probe (S_p, 5'-biotin- TTT TTT GGG TAG GGC GGG TTG GGA AA-3'). PBS buffer (8 mM Na₂HPO₄, 137 mM NaCl, 2 mM KH₂PO₄, 2.7 mM KCl, pH 7.4), PBST buffer (8 mM Na₂HPO₄, 137 mM NaCl, 2 mM KH₂PO₄, 2.7 mM KCl, 0.01% Tween 20, pH 7.4), acetate buffer (0.2 M HAc, 0.2 M NaAc, pH 4.0), TES buffer (10 mM Tris, 1 mM EDTA, 2 M NaCl, pH 7.5) and TEST buffer (5 mM Tris, 0.5 mM EDTA, 1 M NaCl, 0.01% Tween 20, pH 7.5).

Transmission electron microscopy (TEM) analysis was performed on an FEI Tecnai G2 F20 (FEI Co. Ltd., USA). The UV–vis spectra were collected on an Infinite 200 Pro spectrophotometer (Tecan Ltd., Austria).

Peroxidase-like activity of Fe₃O₄@Pt NPs-Apt_{AS1411}/S_p conjugates

50 µg/mL Fe₃O₄@Pt NPs-Apt_{AS1411}/S_p conjugates were resuspended to the 100 µL HAC-NaAC buffer (pH 4.0) in the presence of different concentrations of TMB or H₂O₂. The catalytic parameters were determined by fitting the absorbance data to Michaelis-Menten equation:

$$\frac{1}{V} = \frac{K_m}{V_{max}} \left(\frac{1}{[S]} + \frac{1}{K_m} \right)$$

The Michaelis–Menten equation describes the relationship between the rates of substrate conversion by an enzyme and the concentration of the substrate. In this equation, V is the initial velocity, V_{max} is the maximal reaction velocity, $[S]$ is the substrate concentration, and K_m is the Michaelis constant.

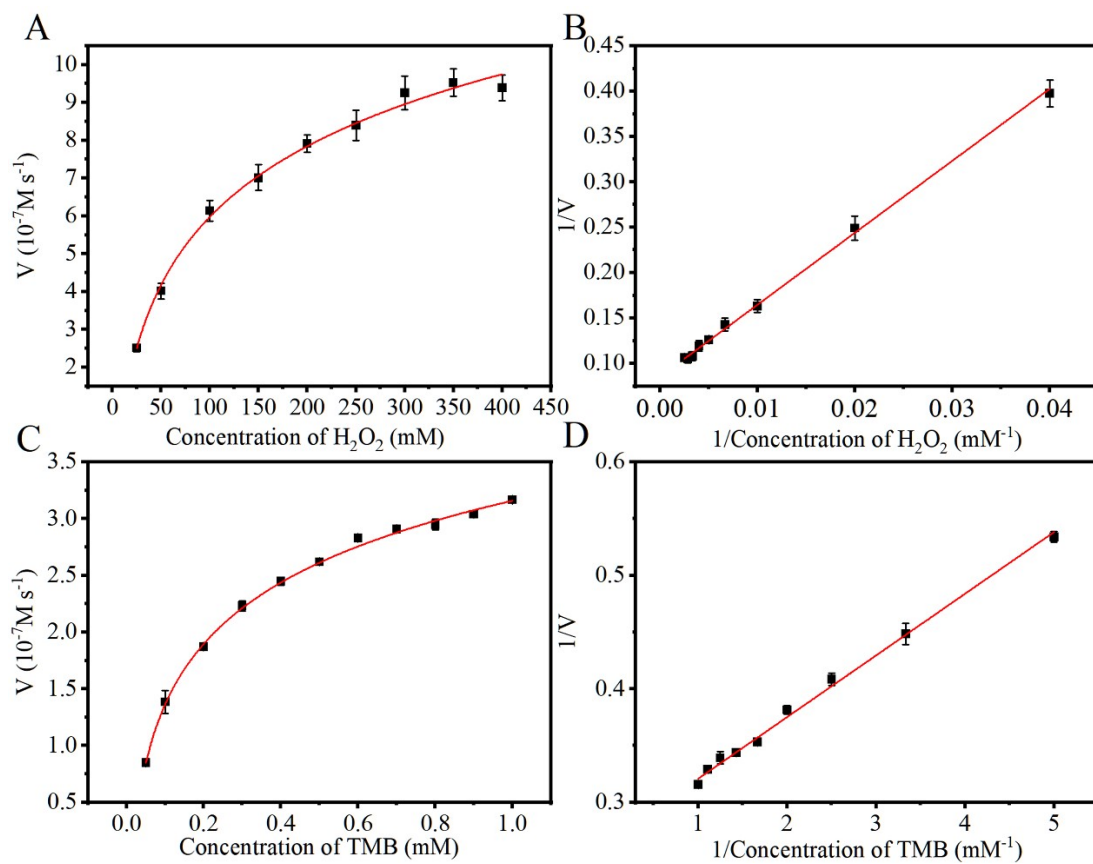


Figure S1 Steady-state kinetic assays and the corresponding double reciprocal (Lineweaver-Burk) plots of $\text{Fe}_3\text{O}_4\text{@Pt}$ NPs for H_2O_2 (A, B) and TMB (C, D).

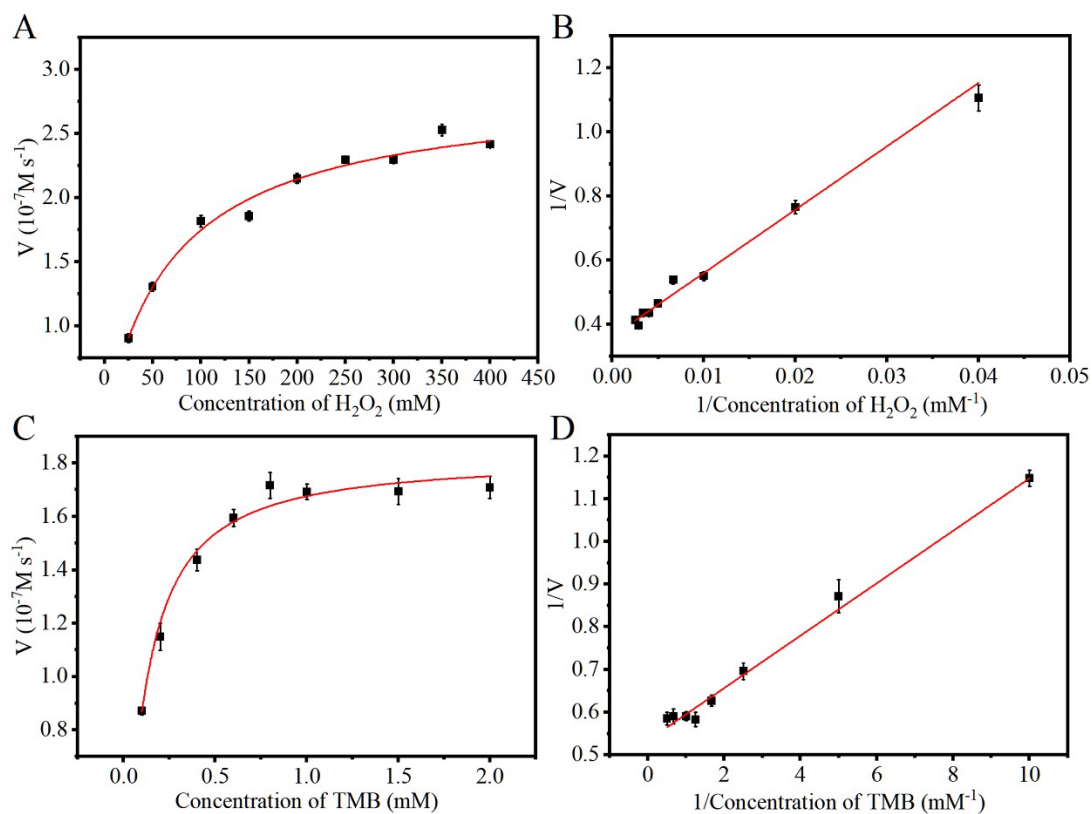


Figure S2 Steady-state kinetic assays and the corresponding double reciprocal (Lineweaver-Burk) plots of $\text{Fe}_3\text{O}_4@\text{Pt}$ NPs-Apt_{AS1411}/S_p conjugates for H_2O_2 (A, B) and TMB (C, D).

Table S1 Apparent kinetic parameters of Fe₃O₄@Pt NPs and Fe₃O₄@Pt NPs-Apt_{AS1411}/S_p conjugates as peroxidase mimetics.

Catalyst	Substrate	K_m (mM)	V_{max} (10 ⁻⁷ M s ⁻¹)
Fe ₃ O ₄ @Pt NPs-	H ₂ O ₂	54.7	2.8
Apt _{AS1411} /S _p conjugates	TMB	0.1	1.9
Fe ₃ O ₄ @Pt NPs	H ₂ O ₂	92.9	11.7
	TMB	0.2	3.8
HRP ¹	H ₂ O ₂	3.7	8.7
	TMB	0.4	10.0

K_m : Michaelis constant. V_{max} : maximal reaction velocity. HRP: horseradish peroxidase.

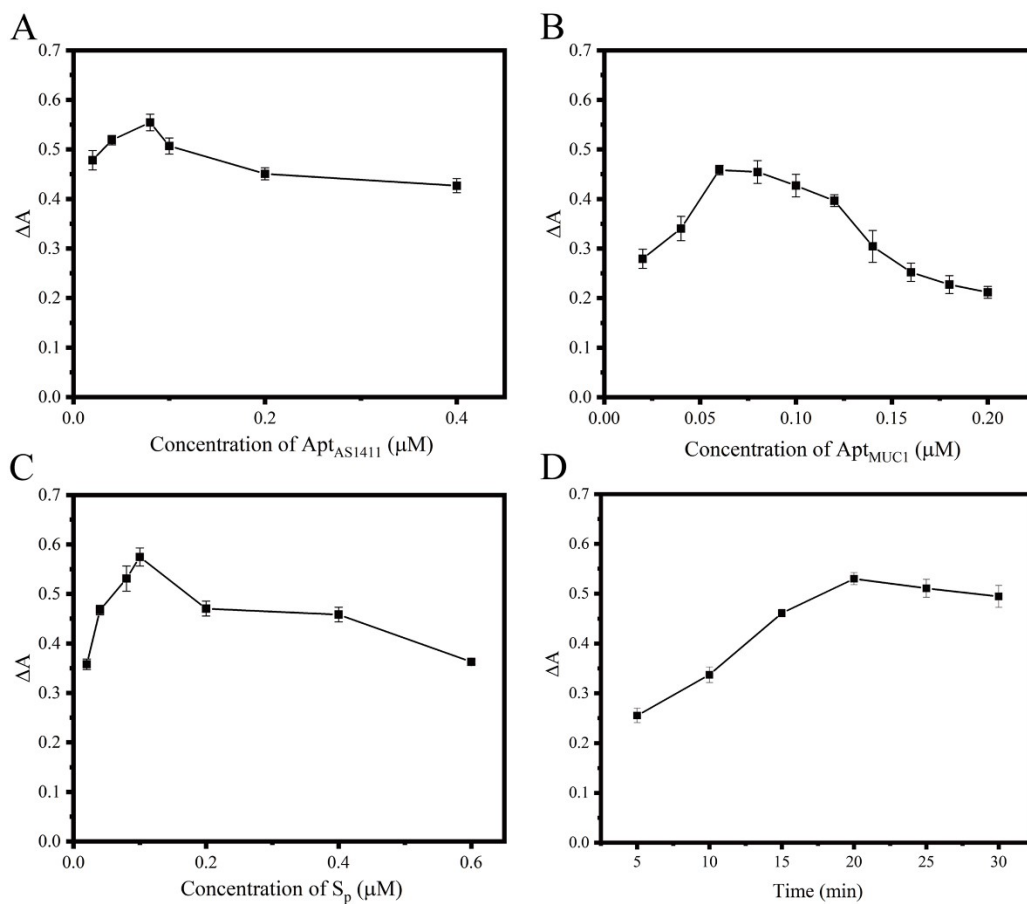


Figure S3 Effect of the concentration of $\text{Apt}_{\text{AS1411}}$ (A), the concentration of Apt_{MUC1} (B), the concentration of S_p (C) and the incubation time of $\text{Fe}_3\text{O}_4@\text{Pt}$ NPs- $\text{Apt}_{\text{AS1411}}/S_p$ conjugates and target MCF-7 cells (D) on the sensitivity of the method. The number of MCF-7 cells was 3000.

Table S2 Analytical performances of various methods for CTC detection.

Method	Cell type	Signal reporter	Linear range (cells)	Detection limit (cells)	Ref.
Electrochemistry	HepG2	methylene blue	10-50000	10	2
ICP-MS	MCF-7	TB	250-1×10000	87	3
Photoelectrochemistry	MCF-7	HCNT	100-100000	17	4
Electrochemistry	MCF-7	Ncomp	1×10-1000000	4	5
Colorimetry	HeLa	A ₃₀ AS1411-AuNFs	10-3000000	10	6
Fluorescence	MCF-7	rGO	100-20000	22	7
Electrochemistry	4T1	AuNPs	80-10000000	50	8
Colorimetry	MCF-7 and HT-29	GNCs	50-20,000	221	9
Colorimetry	MDA-MB-231	Fe ₃ O ₄ @MnO ₂ NPs	250-5×1000	186	10
Fluorescence and Colorimetry	HeLa	Pd NPs/CMC-COF-LZU1	0-1×1000000	100	11
Colorimetry	MCF-7	Fe ₃ O ₄ @Pt NPs-Apt _{AS1411} /S _p conjugates	50-4000	4	This work

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