Electronic Supplementary Material (ESI) for New Journal of Chemistry. This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2023

Supporting information

Theoretical Study of the Electronic Structure of the Complexes of Gold, Silver, and Copper Mono- and Bimetallic Nanoclusters Decorated on Graphitic Carbon Nitride (g-C₃N₄): DFT and TD-DFT Study of Photocatalytic Activity

Hanieh Moradi^a, Hossein Farrokhpour*^a, Sayyed Mahdi Hosseini^a, Mehran Ghiaci^a

^aDepartment of Chemistry, Isfahan University of Technology, University Boulevard, Isfahan, 84156-83111, Iran

*Corresponding Author: Hossein Farrokhpour E-mail: h-farrokh@cc.iut.ac.ir

Summary of content: This Supporting Information file contains 21 Figures presented in 21 pages.



Fig. S1. The initial and optimized (two views) geometries of Ag₄/g-C₃N₄ complexes and their energies relative to the most stable structure.



Fig. S2. The initial and optimized (two views) geometries of Cu₄/g-C₃N₄ complexes and their energies relative to the most stable structure.



Fig. S3. The initial and optimized (two views) geometries of Au₂-Ag₂/g-C₃N₄ complexes and their energies relative to the most stable structure.



Fig. S4. The initial and optimized (two views) geometries of Au₂-Cu₂/g-C₃N₄ complexes and their energies relative to the most stable structure.



Fig. S5. The initial and optimized (two views) geometries of Ag₂-Cu₂/g-C₃N₄ complexes and their energies relative to the most stable structure.



Fig. S6. The colored representation of the partial charges of three optimized Au_4/g - C_3N_4 complexes.



Fig. S7. The colored representation of the partial charges of three optimized Ag_4/g -C₃N₄ complexes.



Fig. S8. The colored representation of the partial charges of three optimized $Cu_4/g-C_3N_4$ complexes.



Fig. S9. The colored representation of the partial charges of six optimized $Au_2-Ag_2/g-C_3N_4$ complexes.



Fig. S10. The colored representation of the partial charges of six optimized Au_2 - Cu_2/g - C_3N_4 complexes.



Fig. S11. The colored representation of the partial charges of six optimized $Ag_2-Cu_2/g-C_3N_4$ complexes.



Fig. S12. Two views of the HOMO and LUMO of (a) the bare $g-C_3N_4$ and the most stable (b) Au_2-Ag_2 , (c) Au_2-Cu_2 , and (d) $Ag_2-Cu_2/g-C_3N_4$ complexes.



Fig. S13. Two views of the HOMO and LUMO for the bare $g-C_3N_4$ and three optimized $Au_4/g-C_3N_4$ complexes.



Fig. S14. Two views of the HOMO and LUMO for three optimized Ag₄/g-C₃N₄ complexes.



Fig. S15. Two views of the HOMO and LUMO for three optimized $Cu_4/g-C_3N_4$ complexes.



Fig. S16. Two views of the HOMO and LUMO for six optimized $Au_2-Ag_2/g-C_3N_4$ complexes.



Fig. S17. Two views of the HOMO and LUMO for six optimized Au_2 - Cu_2/g - C_3N_4 complexes.



Fig. S18. Two views of the HOMO and LUMO for six optimized Ag_2 - Cu_2/g - C_3N_4 complexes.



Fig. S19. The comparison of the (a) TDOS diagram for the bare $g-C_3N_4$ and the PDOS diagrams for (b) Au_4 , (c) Ag_4 , (d) Cu_4 , (e) Au_2-Ag_2 , (f) Au_2-Cu_2 , and (g) $Ag_2-Cu_2/g-C_3N_4$ complexes (the $g-C_3N_4$ fragment).



Fig. S20. The UV spectra of: (A) Au₄/g-C₃N₄, (B) Cu₄/g-C₃N₄, (C) and (D) Au₂-Cu₂/g-C₃N₄ with different metal atom arrangements. The interior figures represent the UV absorption lines for each spectrum.



Fig. S21. The UV spectra of: (A) Ag₄/g-C₃N₄, (B) Cu₄/g-C₃N₄, (C) and (D) Ag₂-Cu₂/g-C₃N₄ with different metal atom arrangements. The interior figures represent the UV absorption lines for each spectrum.