## A phenylboronic acid-based smart photonic crystal hydrogel sensor for colorimetric detection of glucose

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## Preparation of superparamagnetic Fe<sub>3</sub>O<sub>4</sub> nanoparticles

The superparamagnetic Fe<sub>3</sub>O<sub>4</sub> nanoparticles with a hydrodynamic diameter ( $d_h$ ) of ~127 nm was synthesized according to our recent work.<sup>1,2</sup> Briefly, 2.5 g of PSSMA was dissolved in 40 mL of EG followed by addition of 0.497 g of ferric chloride hexahydrate (FeCl<sub>3</sub>·6H<sub>2</sub>O) and vigorous stirring for 20 min. Then, 3.0 g of sodium acetate trihydrate (NaAc·3H<sub>2</sub>O) was added, and the mixture was stirred for another 30 min to form a brownish-red solution. The solution was transferred into a Teflon-lined stainless-steel autoclave (100 mL) and reacted for 10 h at 200 °C. After natural cooling to room temperature, the black product was collected with a magnet and rinsed with water thrice by ultrasonication, and then dispersed in 20 mL of water for the subsequent use.



Fig. S1. Chemical structures of various sugars.



**Fig. S2**. SEM images of the PNAPB PCH with fixed Fe<sub>3</sub>O<sub>4</sub> particle chains before (a) and after (b) responding to glucose.

## Notes and references

- 1. L. Pan, Z. Peng, H. Yu, T. Liang and C. Cheng, New J. Chem., 2021, 45, 16511–16519.
- 2. J. Wen, H. Yu, T. Liang, X. Lv and C. Cheng, New J. Chem., 2023, 47, 10236–10244.