Supporting material

Self-assembled hydrangea-like Fe₂P-CoP-NDC as efficient carrier material of Pt

nanoparticles for methanol oxidation reaction

Fei Xie^{1,*}, Qingchun Wang¹, Mengyu Gan^{2,**}, Li Ma²

¹ College of Automotive Engineering, Chengdu Aeronautic Polytechnic, Chengdu

610100 Sichuan, P. R. China

² College of Chemistry & Chemical Engineering, Chongqing University, Chongqing

400044, P. R. China

^{*}Corresponding author. E-mail address: xiefei0913@126.com.

^{**}Corresponding author. E-mail address: mlsys404@126.com.



Fig. S1 XRD pattern of hydrangea-like Pt/Fe2P-CoP-NDC catalyst



Fig. S2 TEM images of hydrangea-like Pt/Fe2P-CoP-NDC catalyst



Fig. S3 Nitrogen adsorption/desorption isotherms curve of Fe₂P-CoP-NDC carrier material



Fig. S4 XPS spectrum of hydrangea-like Pt/Fe₂P-CoP-NDC catalyst



Fig. S5 CV curves (a), mass activity (b) and chronoamperometric curves (c) of Pt/CoP-NDC, Pt/Fe₂P-NDC and Pt/Fe_2P -CoP-NDC catalysts.

Table S1 The exact amount of Pt in the Pt/Fe₂P-CoP-NDC and commercial Pt/C catalysts obtained by ICP-OES.

Samples	Pt amount / wt%		
Pt/Fe ₂ P-CoP-NDC	19.54		
Commercial Pt/C	20		

Table S2 Relative content of three types Pt for Pt/Fe₂P-CoP-NDC catalyst.

_

	Pt ⁰		Pt^{2+}		Pt ⁴⁺	
Samples	Binding	Ratio %	Binding	Ratio %	Binding	Ratio %
	energy / eV		energy / eV		energy / eV	
	71.48		72.58		73.77	
Pt/Fe ₂ P-CoP-NDC	74.88	60.15	76.08	25.06	77.78	14.79