

Supporting information

**“On–off–on” fluorescent sensor based on Ti<sub>3</sub>C<sub>2</sub> quantum dots and  
CoOOH nanosheets for the detection of ascorbic acid**

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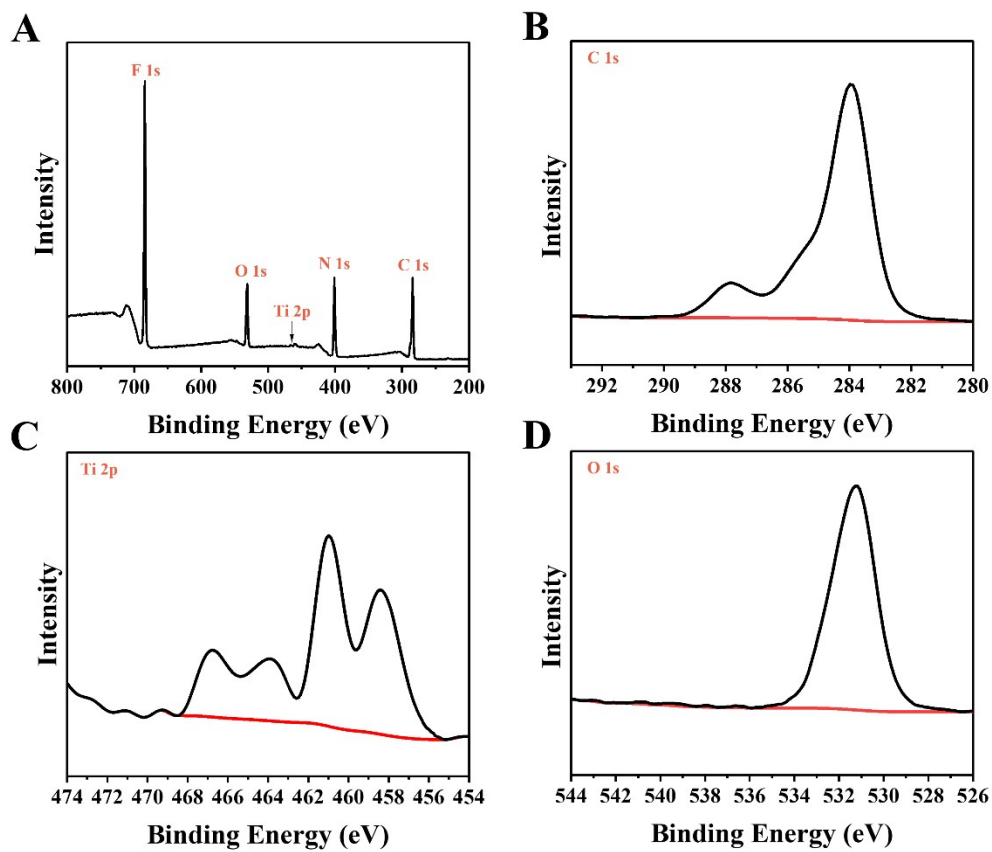
## Counts

Fig. S1. Graphs of XPS for  $\text{Ti}_3\text{C}_2$  QDs.

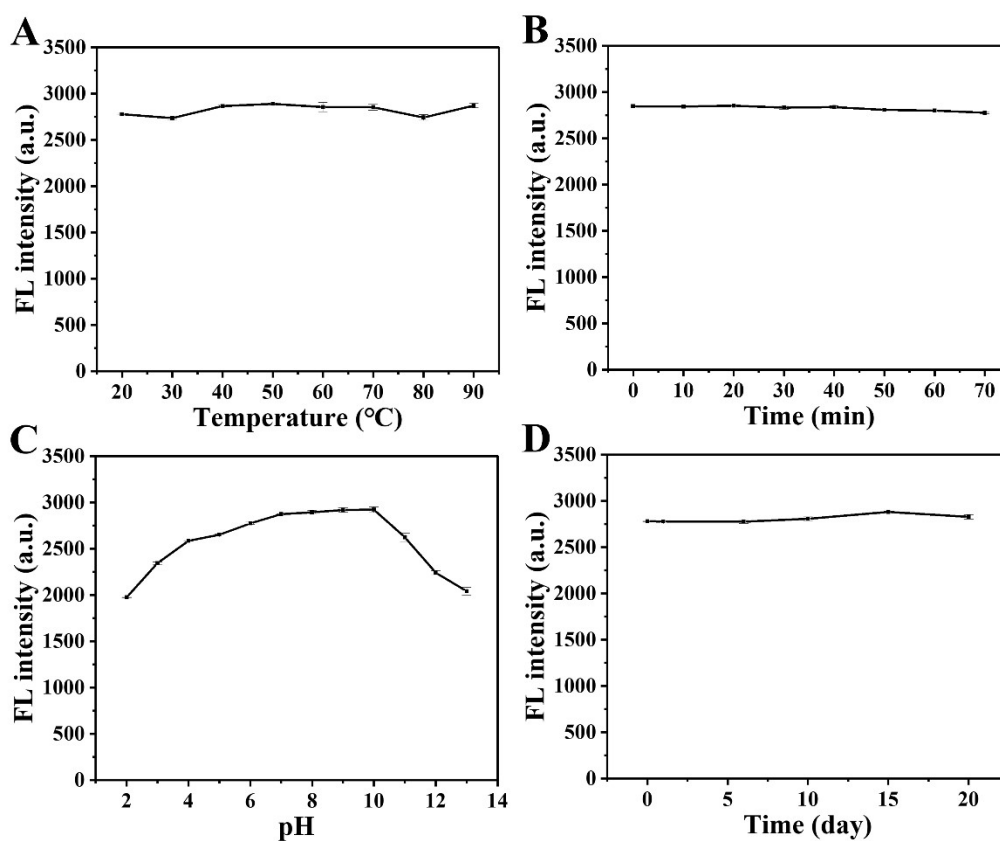
Fig. S2. Research on the stability of  $\text{Ti}_3\text{C}_2$  QDs, including temperature, time and pH.

Fig. S3. Research on the effect of  $\text{Co}^{2+}$  on the fluorescence of  $\text{Ti}_3\text{C}_2$  QDs.

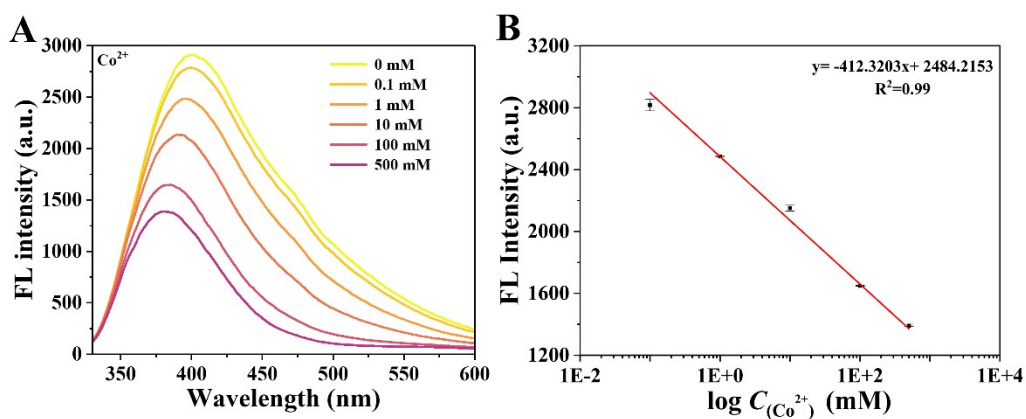
Table S1. Comparison of quantum yields in different reports.



**Fig. S1.** Graphs of XPS for  $\text{Ti}_3\text{C}_2$  QDs. (A) fully scanned spectra (B) C 1s spectrum of  $\text{Ti}_3\text{C}_2$  QDs. (C) Ti 2p (D) O 1s spectrum of  $\text{Ti}_3\text{C}_2$  QDs.



**Fig. S2.** The FL intensity of  $\text{Ti}_3\text{C}_2$  QDs changes with (A) temperature, (B) time (under 365 nm ultraviolet light), (C) pH and (D) time (stored in a refrigerator at 4 °C).



**Fig. S3.** (A) The effect of different concentrations of  $\text{Co}^{2+}$  on the fluorescence of  $\text{Ti}_3\text{C}_2$  QDs. (B) Linear relationships between the fluorescence of  $\text{Ti}_3\text{C}_2$  QDs and the concentration of  $\text{Co}^{2+}$ .

**Table S1** Comparison of quantum yields in different reports.

Materials	Quantum yields (%)	References
Ti <sub>3</sub> C <sub>2</sub> QDs	18.7	1
Ti <sub>3</sub> C <sub>2</sub> QDs	10	2
Ti <sub>3</sub> C <sub>2</sub> QDs	9.36	3
Ti <sub>3</sub> C <sub>2</sub> QDs	7.13	4
Ti <sub>3</sub> C <sub>2</sub> QDs	8.5	This work

**References:**

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