An AIE-ligand-based coordination polymer with new wheel-shaped

Cd₆ cluster, unique bamboo-like architecture and highly efficient

dual-response sensing properties

Jinfang Zhang^{a,*}, Shunchang Zhao^a, Xingyu Tao^a, Qinghan Chen^a, Dejing Yin^b and Chi Zhang^{a,c,*}

^a International Joint Research Center for Photoresponsive Molecules and Materials, School of

Chemical and Material Engineering, Jiangnan University, Wuxi 214122, P. R. China

^b School of Biotechnology, Jiangnan University, Wuxi 214122, P. R. China

^c School of Chemical Science and Engineering, Tongji University, Shanghai 200092, P. R. China

Materials and methods

 H_2L ligand was obtained according to a previously reported procedure.¹ Other purchased chemicals were used without further purification. Powder X-ray diffraction (PXRD) patterns were obtained using Cu K α_1 radiation *via* a Bruker D8 X-ray diffractometer. Simulated PXRD patterns were achieved by Mercury 3.3. Thermal analysis was carried out using a TGA/1100SF thermal analyzer between 40 and 1000°C at a heating rate of 20°C min⁻¹ under a nitrogen atmosphere at a flow rate of 30 cm³ min⁻¹. IR spectra (4000-400 cm⁻¹) were obtained from KBr discs using a Nicolet Nexus 470 transform spectrometer. The HITACHI S-4800 field-emission scanning electron microscope was utilized to conduct Energy Dispersive Spectrometer (EDS). Particle size distribution is measured by zeta sizer nano ZS. Luminescent spectra were measured by PTI QM-TM fluorescent spectrometer. Uv-*vis* adsorption spectra were obtained using a TU-1950 spectrophotometer at 293 K.

References

 Q. Li, X. Wu, X. Huang, X. Xiao, S. Jia, Z. Lin and Y. Zhao, Crystal Growth & Design, 2018, 18, 912-920.

Bond	Lengths(Å)	Bond	Lengths(Å)
Cd(1)-O(1)#2	2.290(9)	Cd(2)-O(2)	2.309(10)
Cd(1)-O(2)#2	2.528(9)	Cd(2)-O(3)	2.277(9)
Cd(1)-O(4)#1	2.285(11)	Cd(2)-O(4)	2.506(12)
Cd(1)-O(5)	2.270(9)	Cd(2)-O(6)#3	2.251(10)
Cd(1)-O(6)	2.476(12)	Cd(2)-O(8)#3	2.360(7)
Cd(1)-O(8)	2.360(7)	Cd(2)-O(9)#4	2.499(9)
Cd(1)-O(9)	2.261(9)	Cd(2)-O(10)#4	2.272(9)

Table S1 Selected bond lengths(Å) for 1.

 Table S2 Selected bond angles for 1.

Bond	Angles(°)	Bond	Angles(°)
O(1)#2-Cd(1)-O(2)#2	53.8(3)	O(2)-Cd(2)-O(4)	97.9(4)
O(1)#2-Cd(1)-O(6)	146.0(4)	O(2)-Cd(2)-O(8)#3	89.9(3)
O(1)#2-Cd(1)-O(8)	137.3(3)	O(2)-Cd(2)-O(9)#4	81.3(3)
O(4)#1-Cd(1)-O(1)#2	88.9(4)	O(3)-Cd(2)-O(2)	94.0(3)
O(4)#1-Cd(1)-O(2)#2	91.1(4)	O(3)-Cd(2)-O(4)	55.2(4)
O(4)#1-Cd(1)-O(6)	81.4(4)	O(3)-Cd(2)-O(8)#3	125.4(3)
O(4)#1-Cd(1)-O(8)	74.3(3)	O(3)-Cd(2)-O(9)#4	146.9(4)
O(5)-Cd(1)-O(1)#2	95.5(3)	O(6)#3-Cd(2)-O(2)	164.2(3)
O(5)-Cd(1)-O(2)#2	147.8(4)	O(6)#3-Cd(2)-O(3)	98.5(4)
O(5)-Cd(1)-O(4)#1	98.4(4)	O(6)#3-Cd(2)-O(4)	81.4(4)
O(5)-Cd(1)-O(6)	54.4(4)	O(6)#3-Cd(2)-O(8)#3	75.0(4)
O(5)-Cd(1)-O(8)	125.3(3)	O(6)#3-Cd(2)-O(9)#4	93.4(4)
O(6)-Cd(1)-O(2)#2	157.7(3)	O(6)#3-Cd(2)-O(10)#4	89.6(4)
O(8)-Cd(1)-O(2)#2	86.9(3)	O(8)#3-Cd(2)-O(4)	70.3(3)
O(8)-Cd(1)-O(6)	70.9(3)	O(8)#3-Cd(2)-O(9)#4	87.5(3)
O(9)-Cd(1)-O(1)#2	98.3(4)	O(9)#4-Cd(2)-O(4)	157.8(3)

O(9)-Cd(1)-O(2)#2	81.6(3)	O(10)#4-Cd(2)-O(2)	98.9(4)
O(9)-Cd(1)-O(4)#1	163.7(4)	O(10)#4-Cd(2)-O(3)	95.5(3)
O(9)-Cd(1)-O(5)	95.5(4)	O(10)#4-Cd(2)-O(4)	147.1(4)
O(9)-Cd(1)-O(6)	100.0(4)	O(10)#4-Cd(2)-O(8)#3	137.5(3)
O(9)-Cd(1)-O(8)	90.7(3)	O(10)#4-Cd(2)-O(9)#4	53.6(3)

Symmetry transformations used to generate equivalent atoms: #1 -x, -x+y, z+1/2; #2 x-y+1, -y+2, z+1/2; #3 -x, -x+y, z-1/2; #4 x-y+1, -y+2, z-1/2

Analytes	HOMO (ev)	LUMO (eV)	Band Gap (eV)
L	-5.779	-2.336	3.443
TNP	-8.237	-3.897	4.340
1,3-DNB	-7.985	-3.431	4.554
2,4-DNT	-7.764	-3.217	4.547
2,6-DNT	-7.644	-3.287	4.357
NB	-7.591	-2.428	5.163
4-NT	-7.654	-2.790	4.864

Table S3 Calculated HOMO and LUMO energy levels of L from 1 and NACs.



Fig. S1 The asymmetric unit of **1** (C: grey, O: pink, S: yellow and Cd: green, and all H atoms are omitted for clarity).



Fig. S2 The angular L bridges linking two wheel-shaped $[Cd_6(COO)_{12}SO_4]^{2-}$ hexanuclear clusters (C: grey, O: pink, S: yellow and Cd: green, and all H atoms are omitted for clarity).



Fig. S3 The PXRD patterns of 1.



Fig. S4 The TGA curves of 1.



Fig. S5 The IR spectrum of 1.



Fig. S6 The Elemental mapping for 1.



Fig. S7 Excitation and emission spectra of 1.



Fig. S8 The particle size distributions of 1 dispersed in H_2O after 48h.