

Supporting information

Porous single crystal microcubes of niobium nitride for highly efficient electrocatalysis

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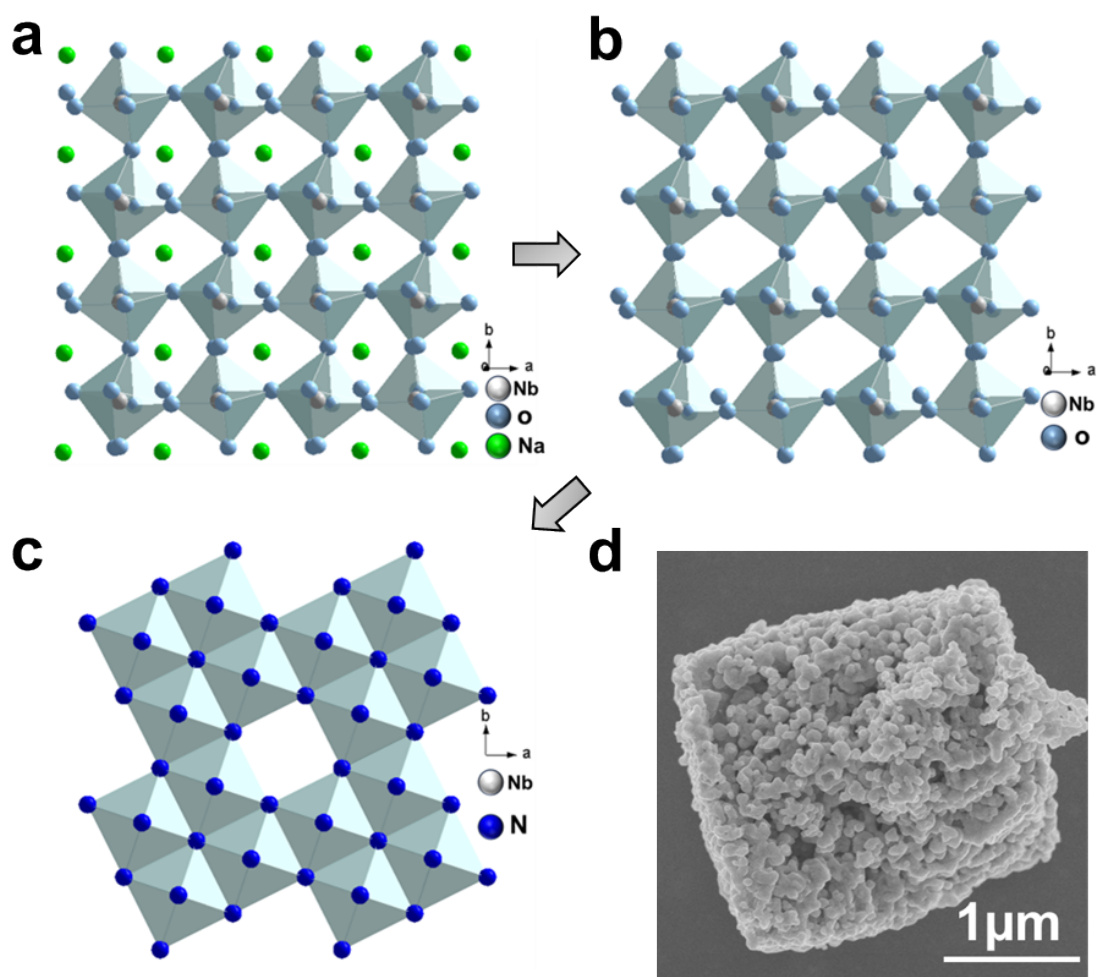


Fig. S1 (a) The lattice channel of atomic evaporation of the (100) facet of NaNbO₃ substrates. (b) (100) orientation of NaNbO₃ single crystal after evaporation of Na atoms. (c) (001) orientation of PSC Nb₄N₅ monoliths (Na atoms are green, O atoms are gray, Nb atoms are white, N atoms are blue). (d) SEM image of PPC Nb₄N₅

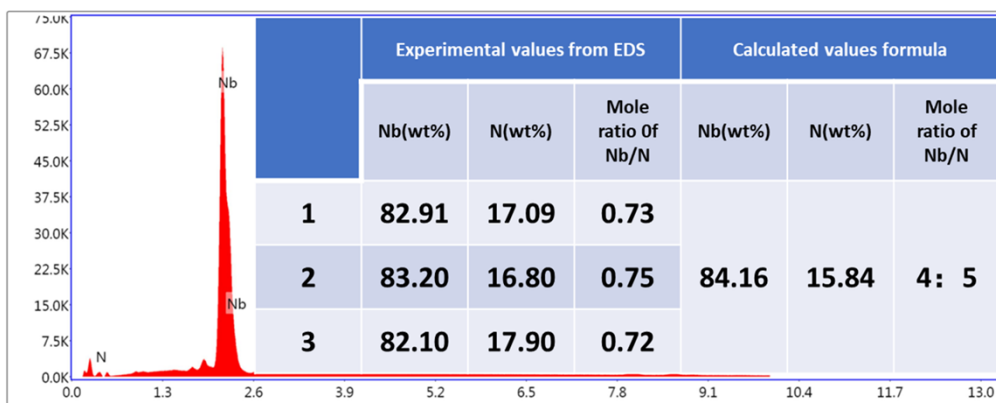


Fig. S2 The element analysis of PSC Nb₄N₅. No oxygen residual is observed from EDS elemental analysis. The mole ratio between Nb and N is approximately at 4:5.

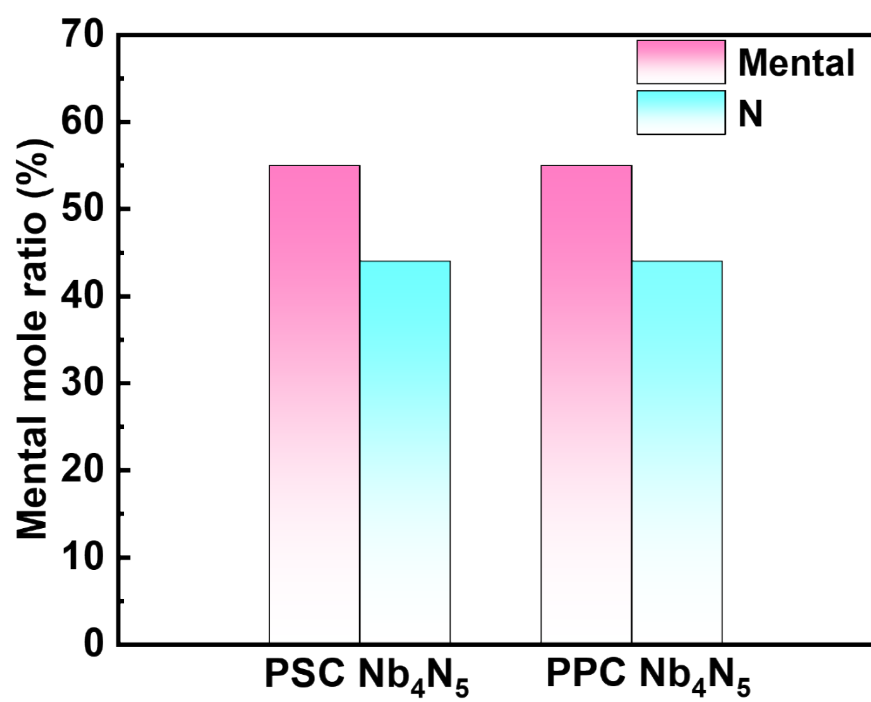


Fig. S3 ICP and CA results. Mole ratio between metal and nitrogen in PSC Nb₄N₅ and PPC Nb₄N₅.

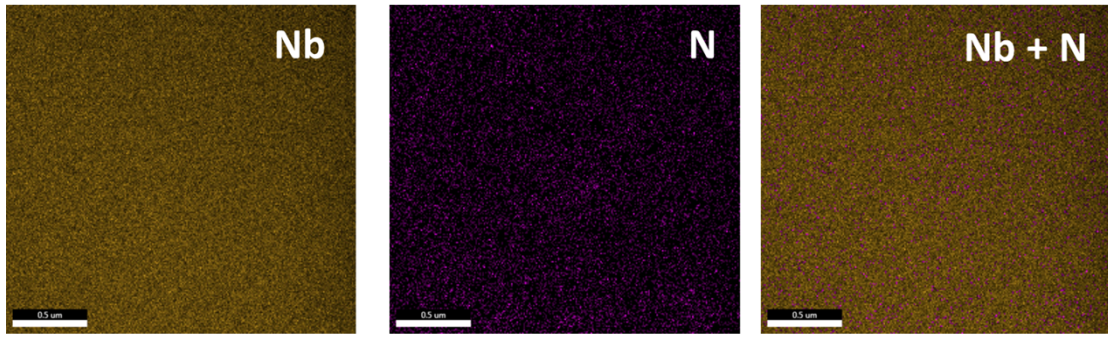


Fig. S4 EDS mapping images of PSC Nb₄N₅.

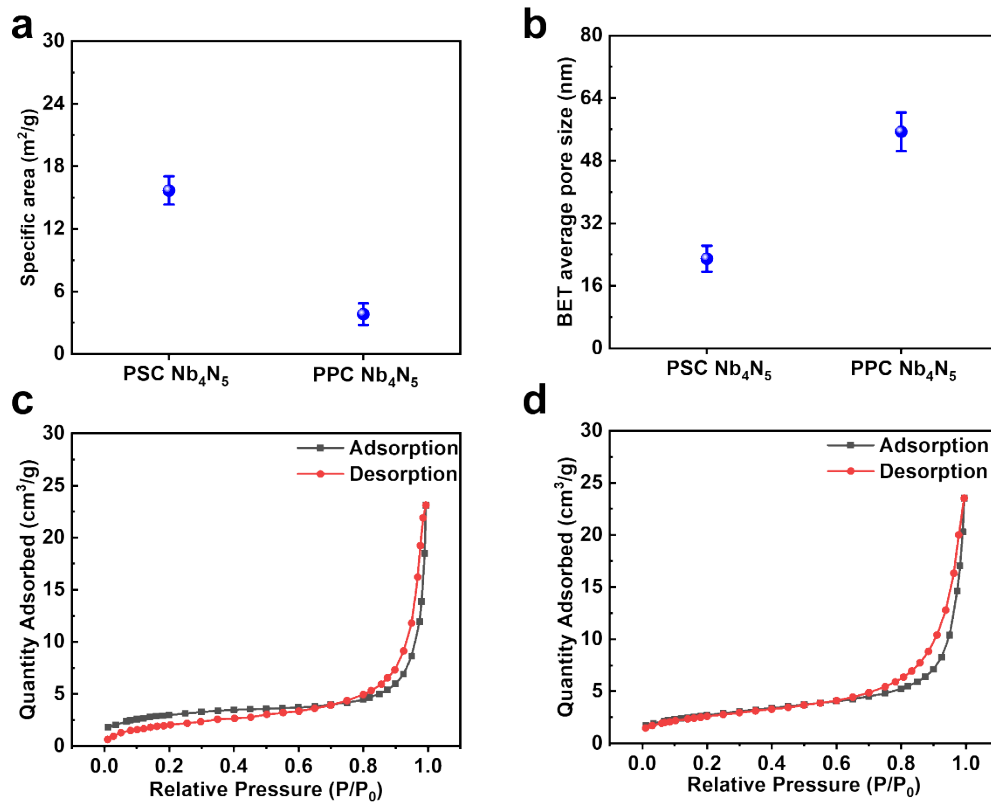


Fig. S5 (a, b) The surface specific area and BET average pore size of PSC Nb₄N₅ and PPC Nb₄N₅. (c) Nitrogen adsorption-desorption isotherm of PSC Nb₄N₅. (d) Nitrogen adsorption-desorption isotherm of PPC Nb₄N₅.

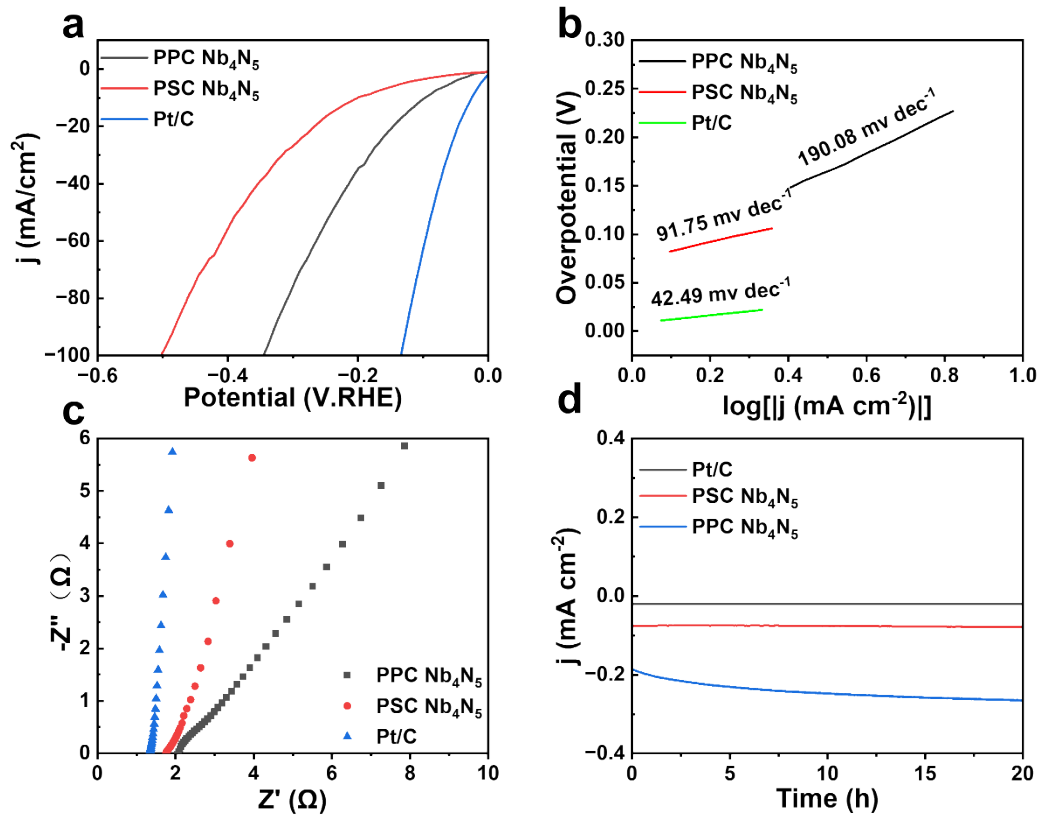


Fig. S6 (a) LSV curves of PSC Nb₄N₅, PPC Nb₄N₅ and Pt/C in 1 M KOH solution with a scan rate 5 mV s⁻¹. (b) Tafel slope of PSC Nb₄N₅, PPC Nb₄N₅ and Pt/C. (c) Nyquist plots of PSC Nb₄N₅, PPC Nb₄N₅ and Pt/C. (d) The durability test of PSC Nb₄N₅, PPC Nb₄N₅ and Pt/C for 20 h.

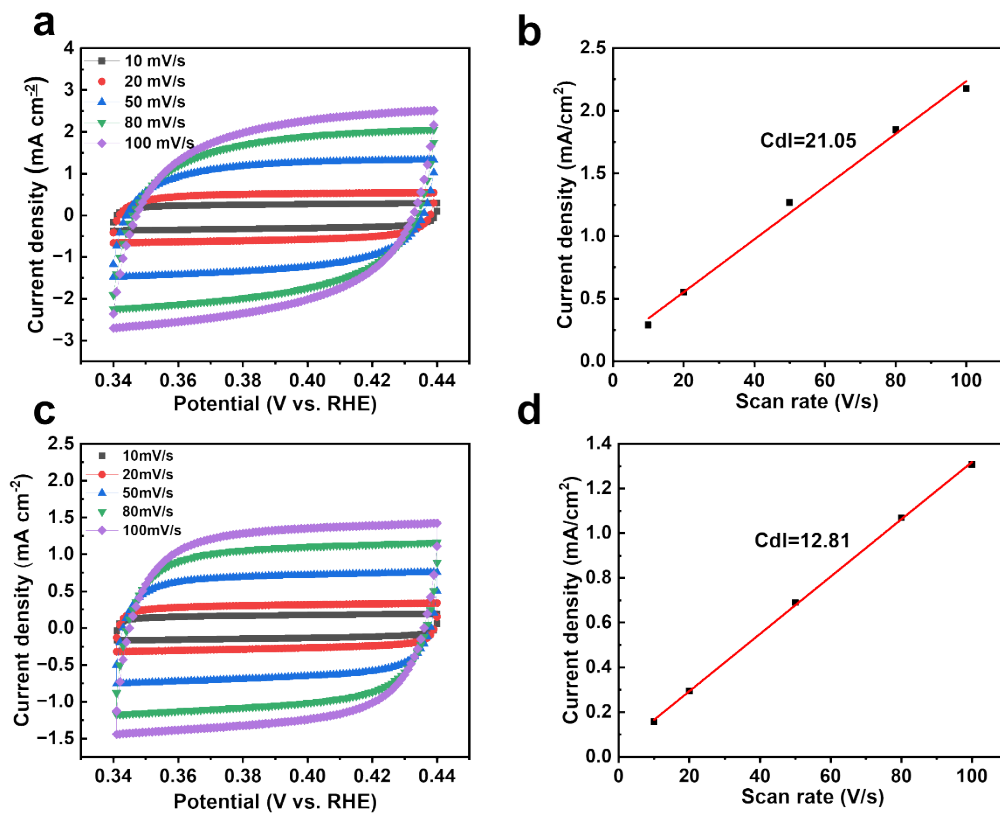


Fig. S7 CVs of (a) PSC Nb₄N₅ and (c) PPC Nb₄N₅ hexahedron catalysts at 10-100 mV s⁻¹ in 0.5 M H₂SO₄ solution; (b) Plots providing the C_{dl} value of PSC Nb₄N₅ catalyst. (d) Plots providing the C_{dl} value of PPC Nb₄N₅ catalyst.

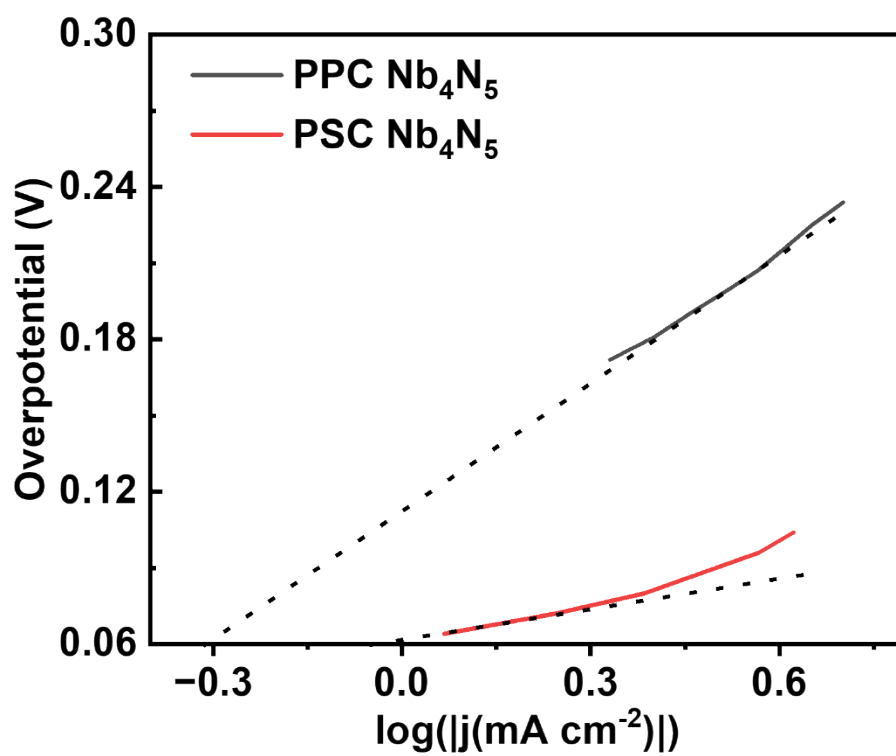


Fig. S8 Calculated exchange current densities of the PSC Nb₄N₅ and PPC Nb₄N₅ in 0.5 M H₂SO₄ electrolyte by applying extrapolation method to the Tafel plots.

Table S1. The comparison of HER electrocatalytic performance.

Catalyst	η (mV)at $j=10\text{mA cm}^{-2}$	Tafel slope	Electrolyte	references
PSC Nb ₄ N ₅	71.86 mV	70.26 mV dec ⁻¹	0.5 M H ₂ SO ₄	This work
PPC Nb ₄ N ₅	191.51 mV	169.02 mV dec ⁻¹	0.5 M H ₂ SO ₄	This work
Nb ₂ N	96.3 mV	92 mV dec ⁻¹	0.5 M H ₂ SO ₄	[1]
Sr ₂ RuO ₄	61 mV	51 mV dec ⁻¹	1 M KOH	[2]
Nb ₄ N ₅ -xOx/NG	39 mV	79 mV dec ⁻¹	0.5 M H ₂ SO ₄	[3]
PSC Mo ₂ N	73.13 mV	66.52 mV dec ⁻¹	1 M KOH	[4]
PSC VN	74.67 mV	68.30 mV dec ⁻¹	1 M KOH	[5]
Nb-Ti NNA	120 mV	52.90 mV dec ⁻¹	1 M KOH	[6]
Co-NG	30 mV	82 mV dec ⁻¹	0.5 M H ₂ SO ₄	[7]
C-MoS ₂	45 mV	46 mV dec ⁻¹	1 M KOH	[8]
NiN	210 mV	122 mV dec ⁻¹	1 M KOH	[9]

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