

Supporting Information

Ionic Plastic Crystals and Ionic Liquids Containing Quaternary Cations with Alkenyl Substituents: Chemical Phase Transformations by Bromine Vapor

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Table S1. Cell parameters

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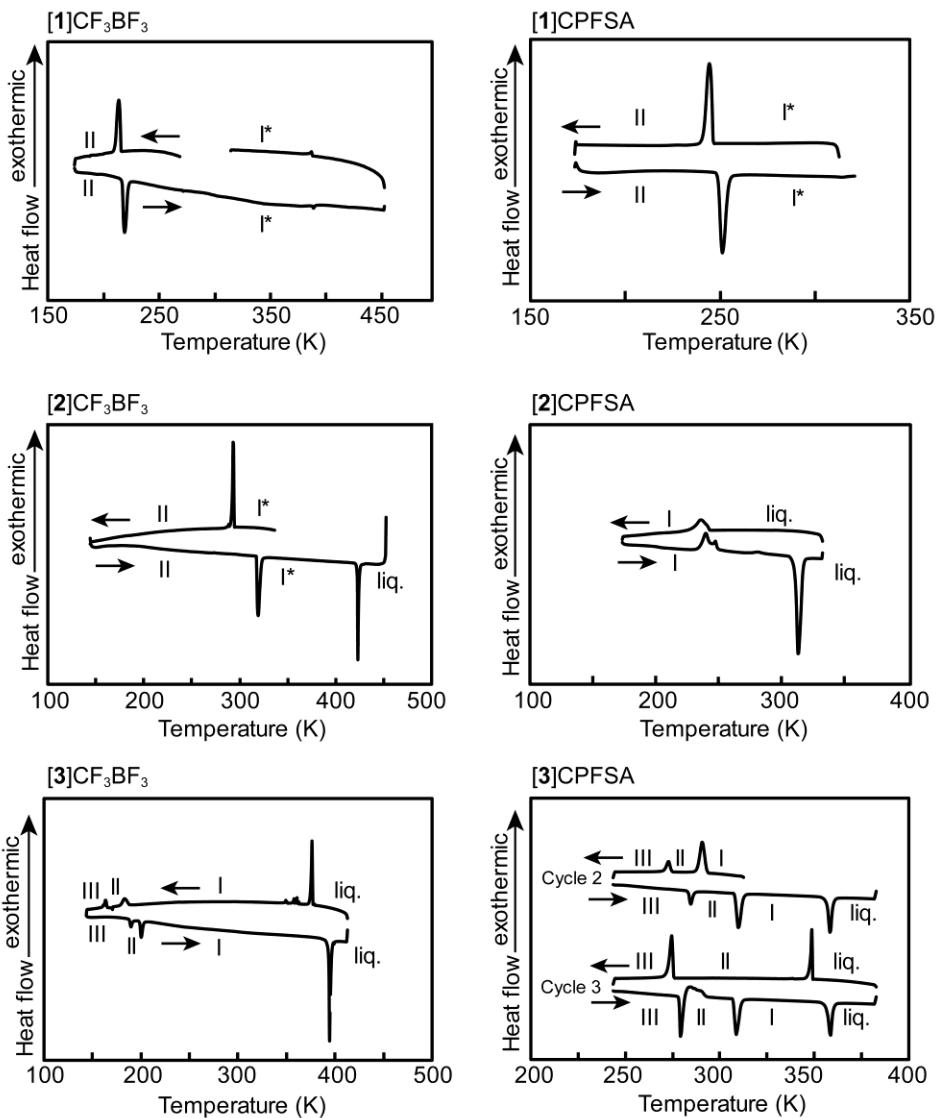


Fig. S1. DSC traces of $[1]X-[3]X$, where $X = \text{CPFSA}$ or CF_3BF_3 .

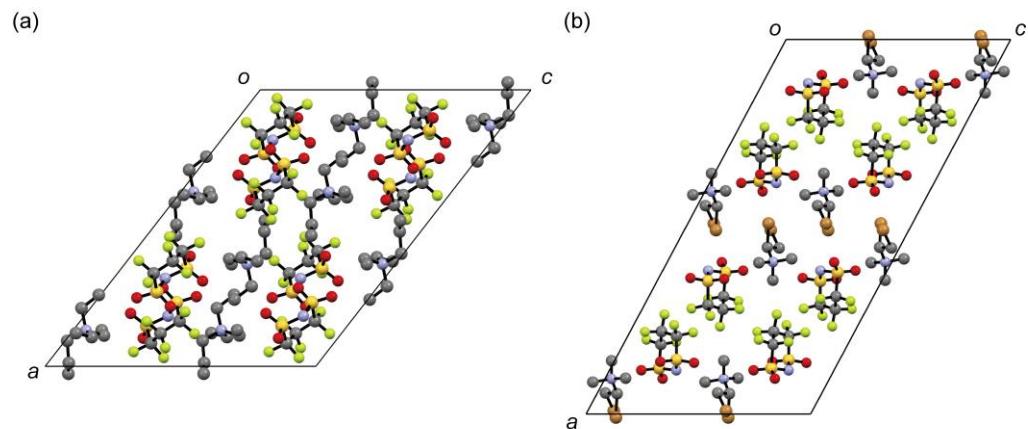


Fig. S2. Packing diagrams of (a) $[2]\text{CPFSA}$ and (b) $[1\text{-Br}]\text{CPFSA}$ at 90 K. Hydrogen atoms have been omitted.

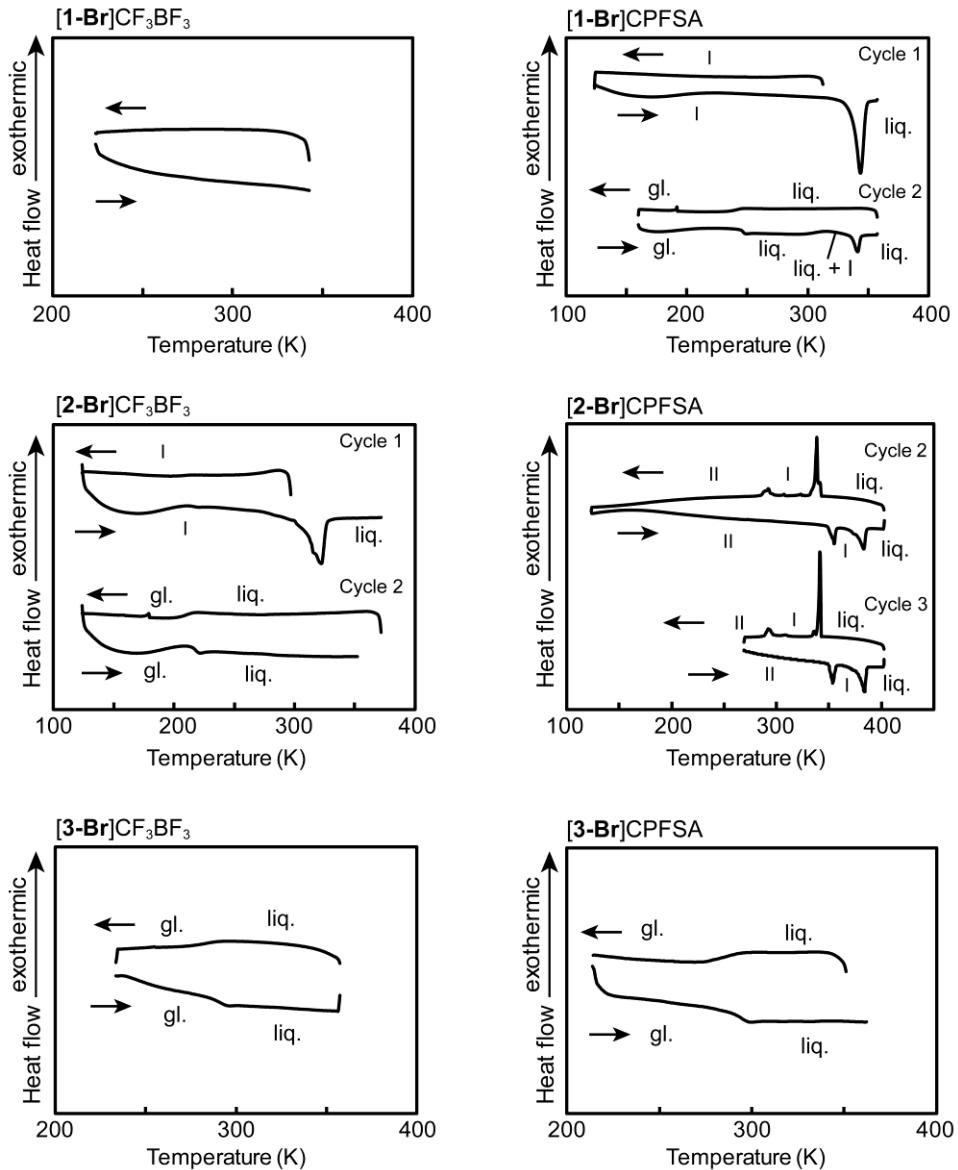


Fig. S3. DSC traces of $[1\text{-Br}]\text{X}$ – $[3\text{-Br}]\text{X}$, where $\text{X} = \text{CPFSA}$ or CF_3BF_3 , and liq. and gl. are the liquid and glassy states, respectively.

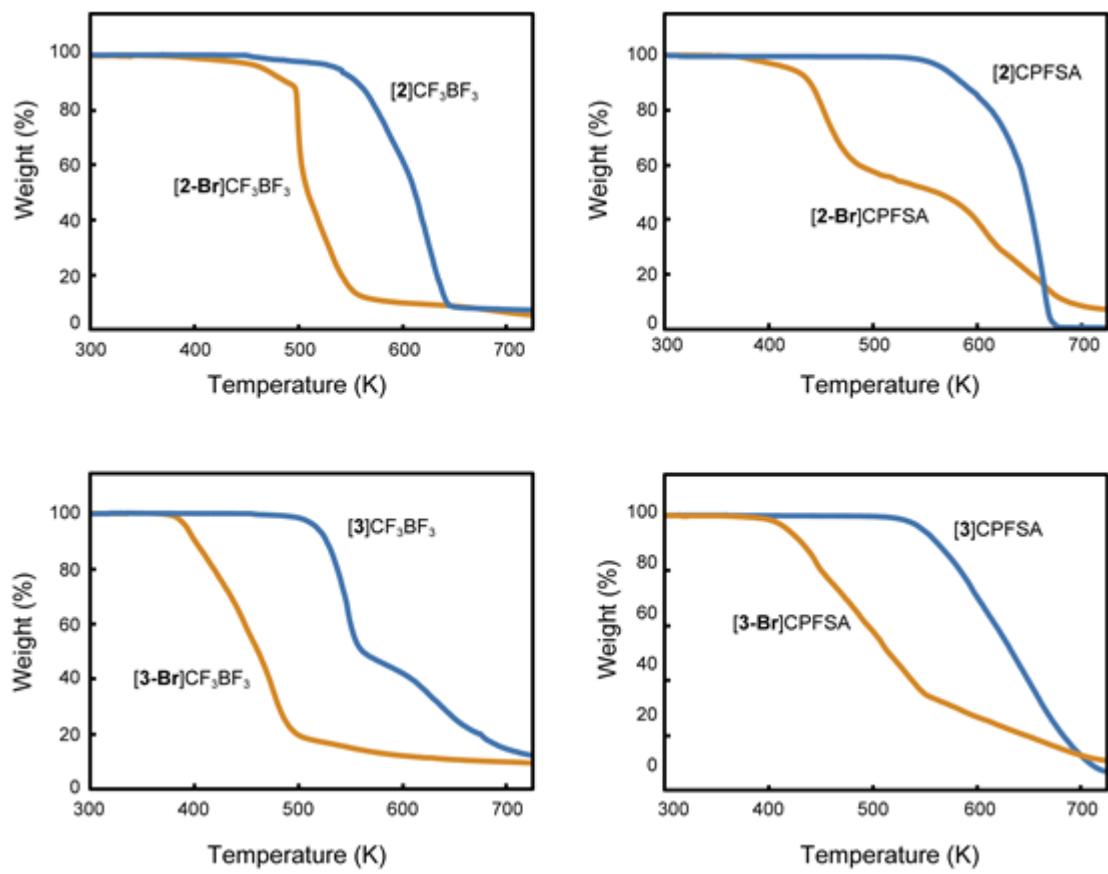
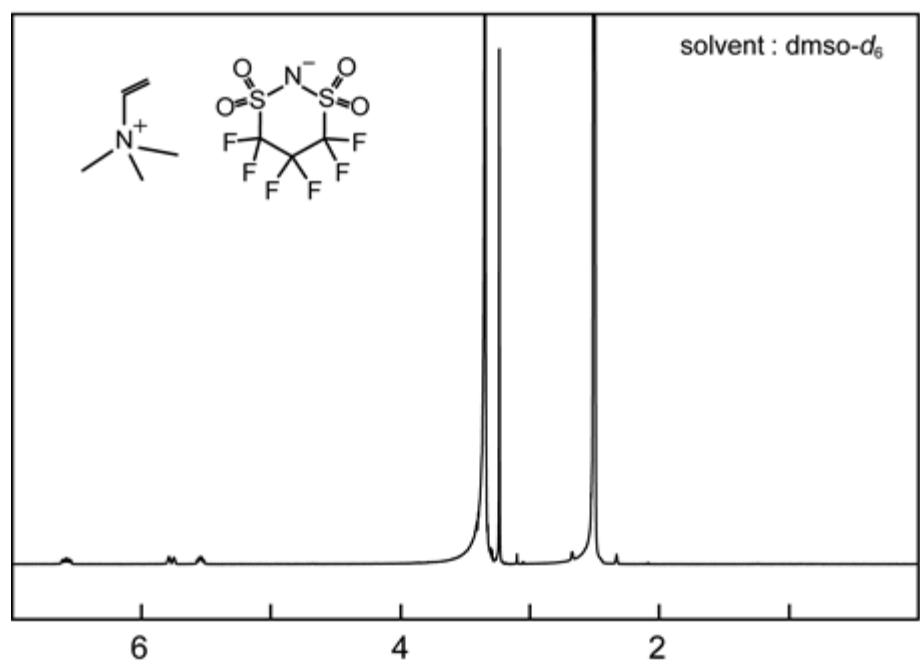
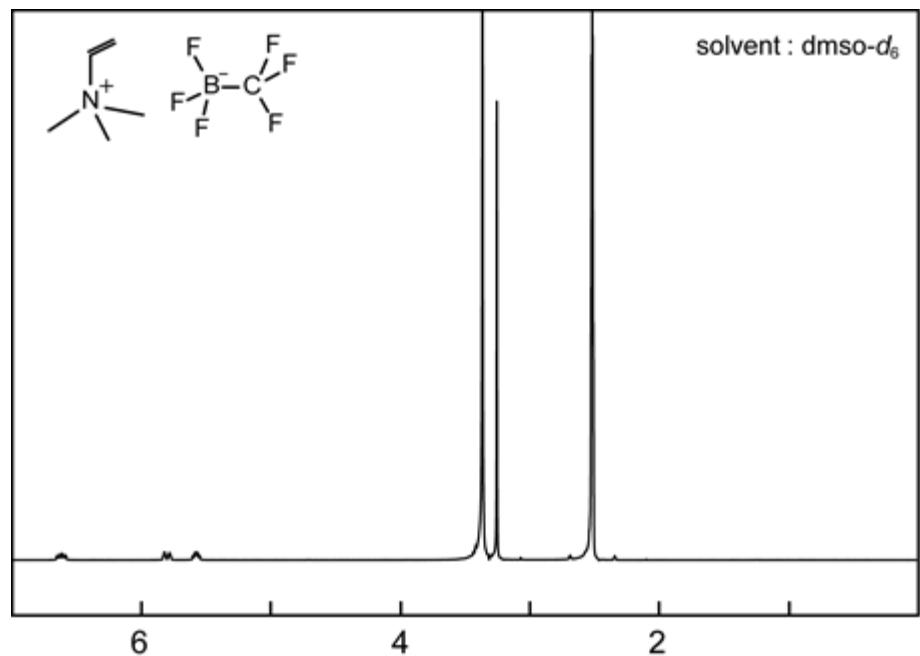
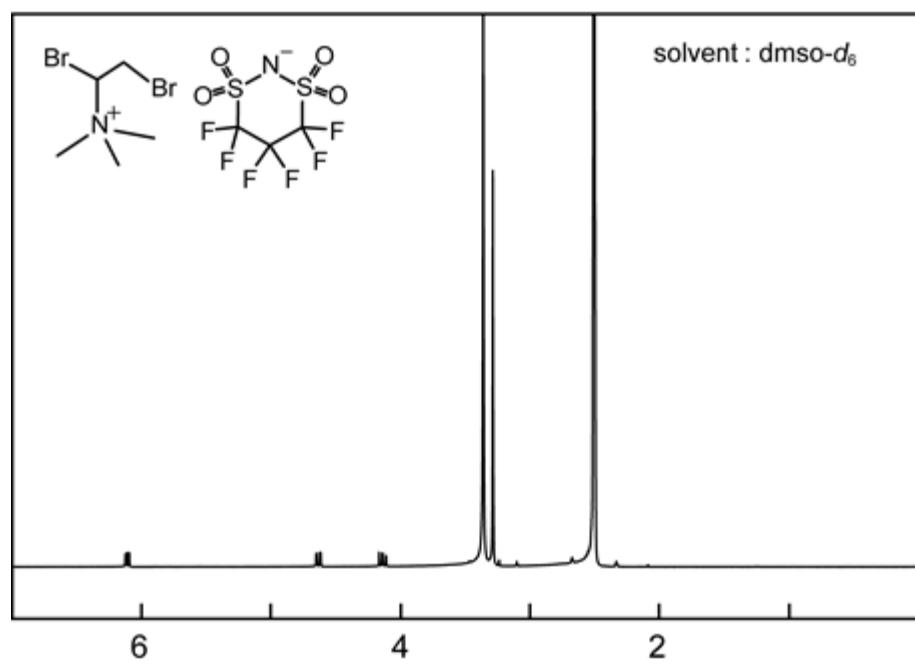
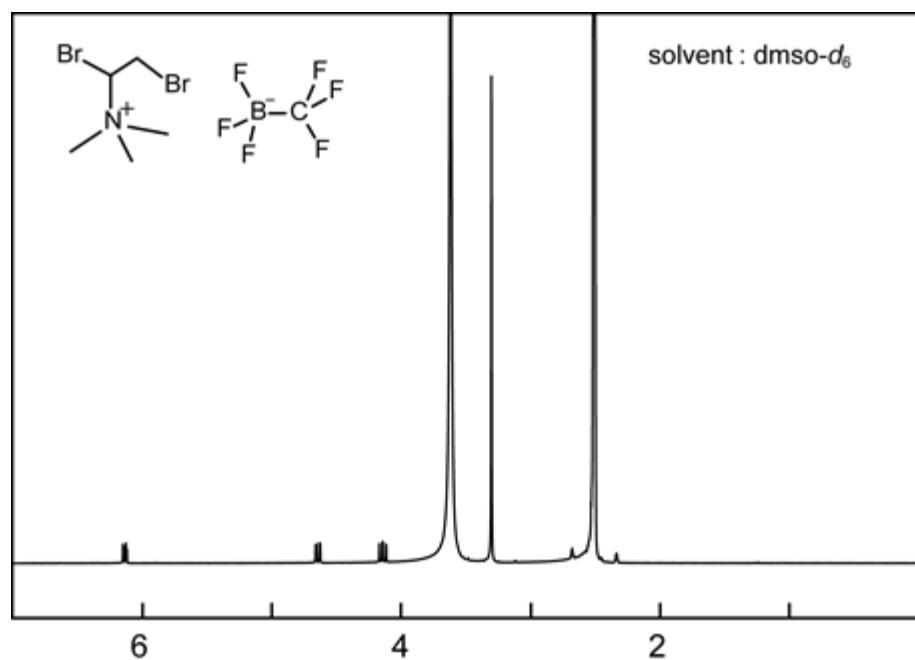
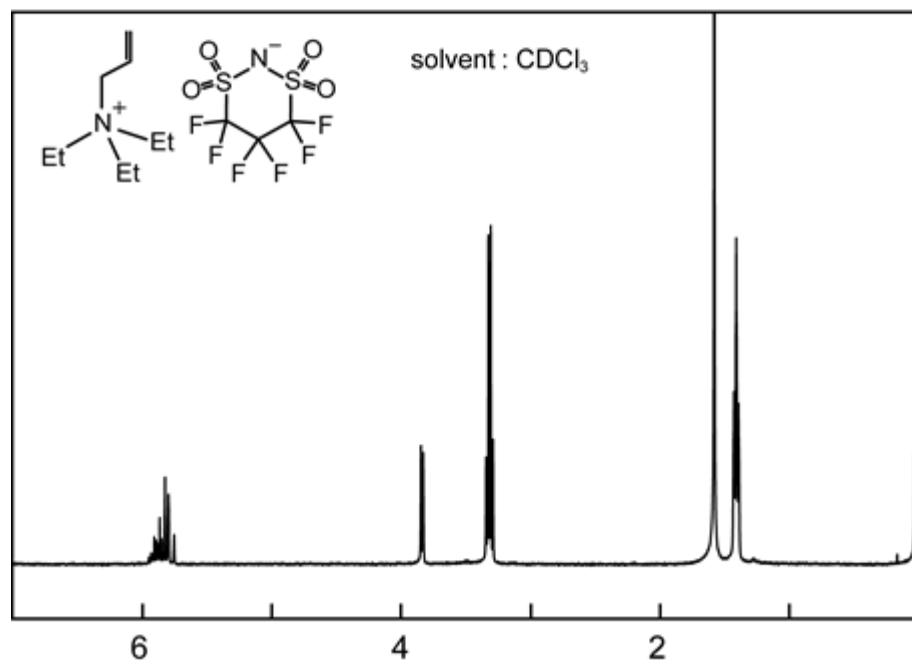
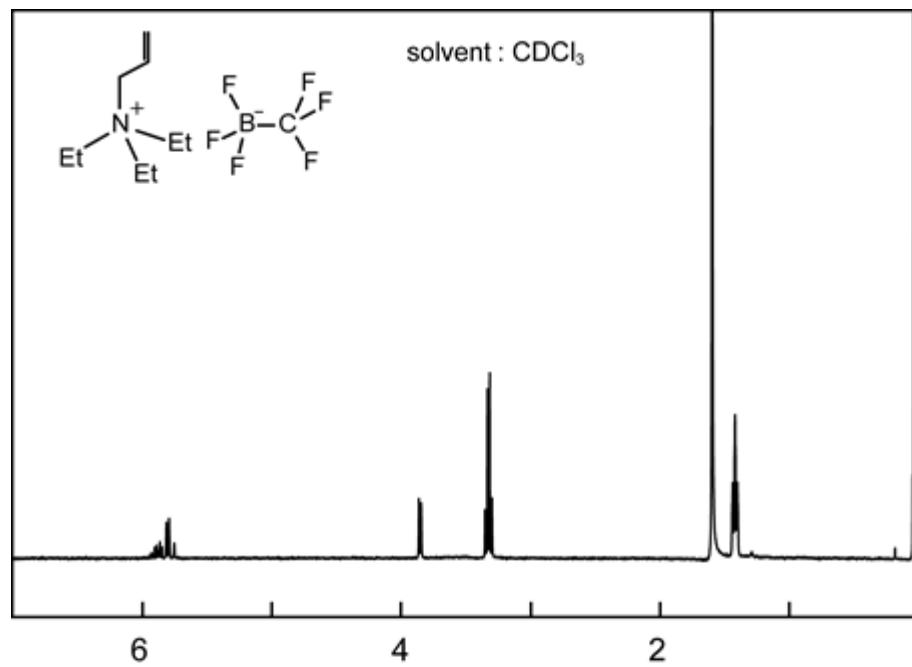
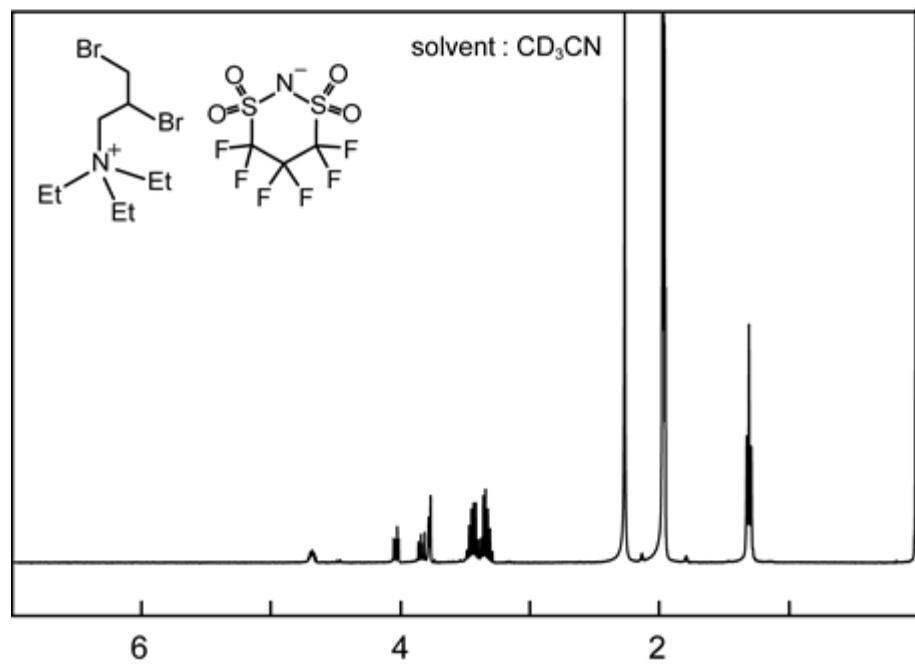
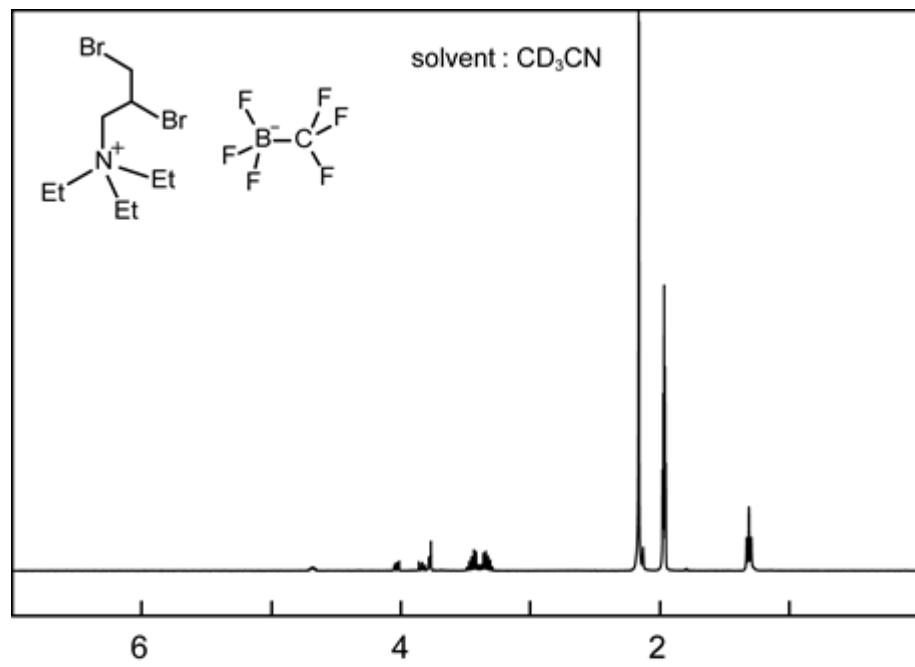


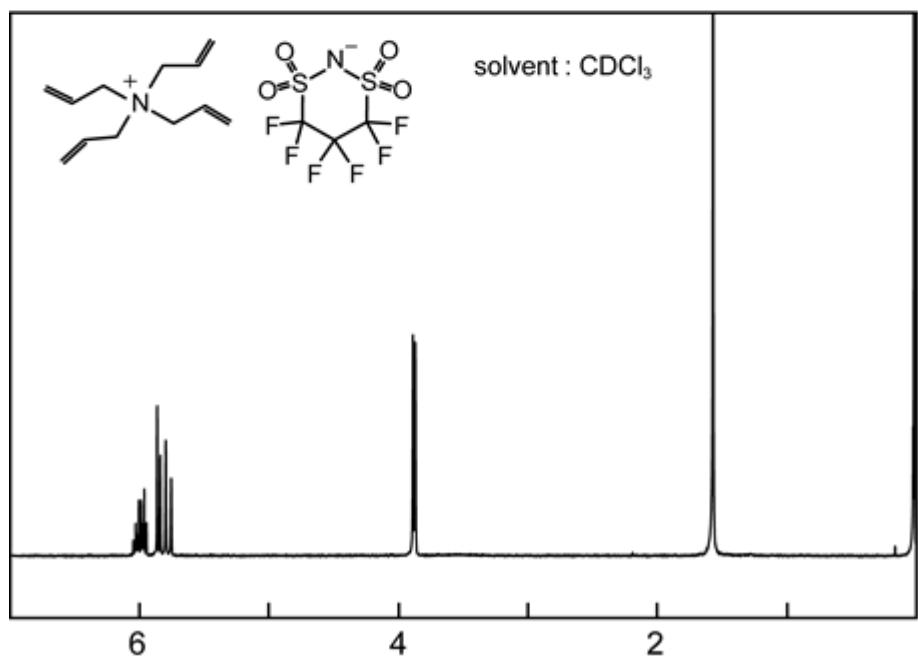
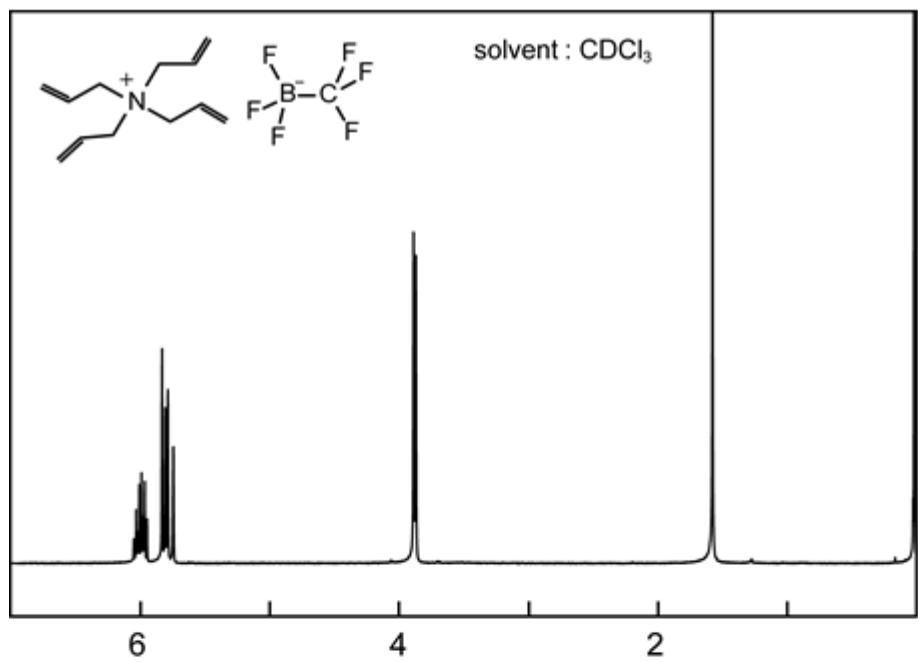
Fig. S4. TG traces of $[1]\text{X}-[3]\text{X}$ and $[1\text{-Br}]\text{X}-[3\text{-Br}]\text{X}$, where $\text{X} = \text{CPFSA}$ or CF_3BF_3 at 3 K min^{-1} under a nitrogen atmosphere.











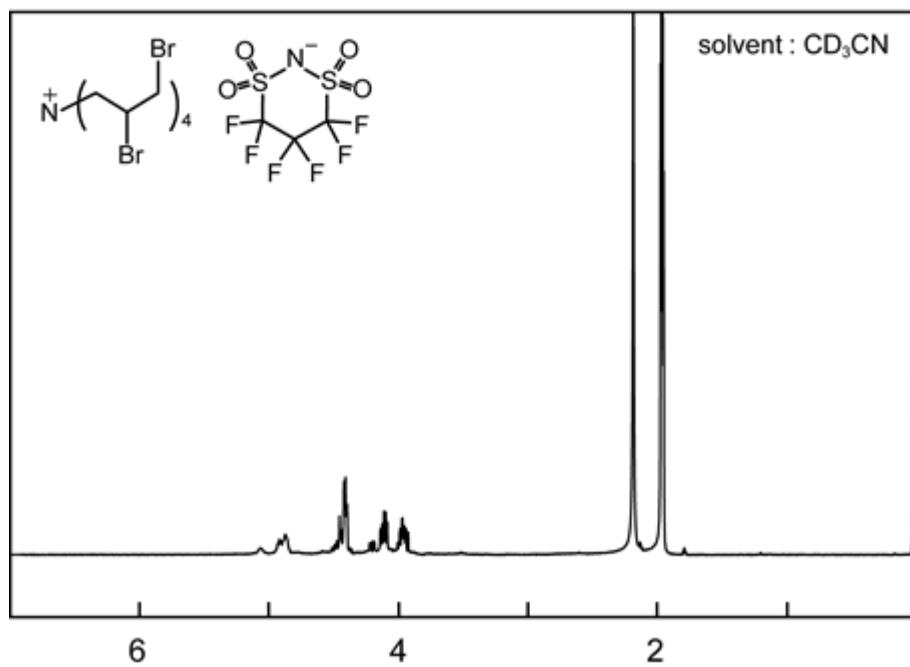
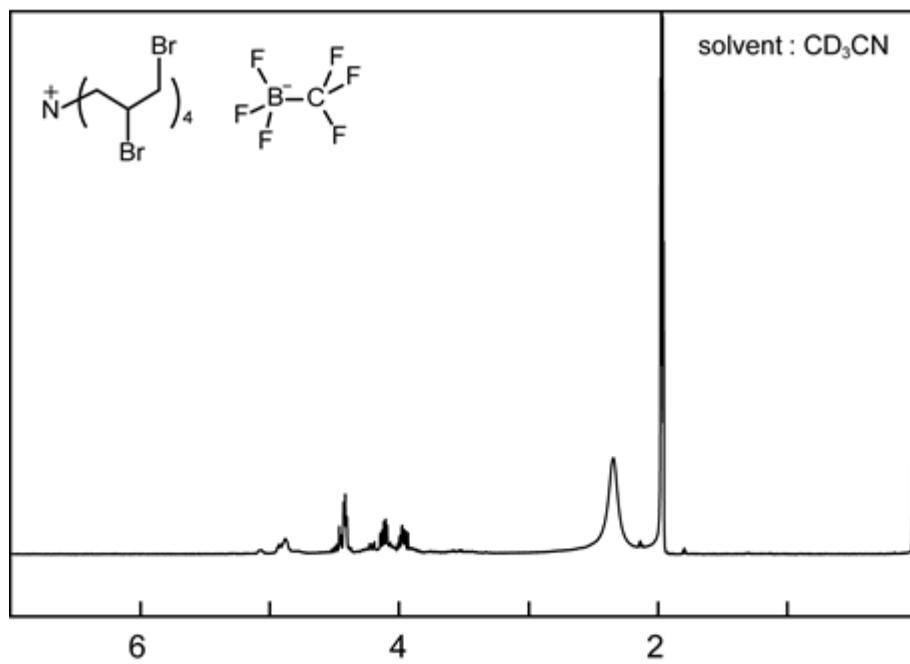


Fig. S5. ^1H NMR spectra of $[\mathbf{1}]X-[\mathbf{3}]X$ and $[\mathbf{1}\text{-Br}]X-[\mathbf{3}\text{-Br}]X$, where $X = \text{CPFSA}_\text{d}$ or CF_3BF_3 .

Table S1. Cell parameters

	[1]CF ₃ BF ₃	[1]CPFSA	[2]CF ₃ BF ₃
	193 K (Phase II)	203 K (Phase II)	273 K (Phase II)
Crystal system	triclinic	orthorhombic	monoclinic
<i>a</i> (Å)	12.31(2)	24.63(13)	13.785(11)
<i>b</i> (Å)	14.45(2)	9.69(9)	4.863(3)
<i>c</i> (Å)	7.840(14)	5.92(6)	6.984(5)
α (°)	97.01(5)		
β (°)	96.56(4)		90.66(4)
γ (°)	110.97(5)		
<i>V</i> (Å ³)	1273(4)	1414(20)	468.2(6)
<i>Z</i> ^{a)}	4	4	1

a) Estimated from molecular volume.

Table S2. Crystallographic parameters

	[2]CPFSA	[1-Br]CPFSA
Empirical formula	C ₁₂ H ₂₀ F ₆ N ₂ O ₄ S ₂	C ₈ H ₁₂ Br ₂ F ₆ N ₂ O ₄ S ₂
Formula weight	434.42	535.85
Crystal system	monoclinic	monoclinic
Space group	<i>C</i> 2/c	<i>C</i> 2/c
<i>a</i> / Å	23.447(6)	31.144
<i>b</i> / Å	10.992(3)	7.460
<i>c</i> / Å	17.992(7)	16.328(7)
β / °	127.576(3)	117.839(5)
<i>V</i> / Å ³	3675(2)	3355(2)
<i>Z</i>	8	8
ρ_{calkd} / g cm ⁻³	1.57	2.131
<i>F</i> (000)	1792	2096
Reflns collected	10214	9050
Independent reflns	4189	3824
Parameters	238	220
<i>R</i> (int)	0.0586	0.0461
R_1^a, R_w^b ($I > 2\sigma(I)$)	0.0498, 0.1017	0.0366, 0.0950
R_1^a, R_w^b (all data)	0.0978, 0.1171	0.0495, 0.1006
Goodness of fit	1.009	1.105
$\Delta\rho_{\text{max,min}}$ [e Å ⁻³]	0.425, -0.615	2.272, -1.030

^a $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$. ^b $R_w = [\sum w (F_o^2 - F_c^2)^2 / \sum w (F_o^2)^2]^{1/2}$.