

## Electromigration separation of lithium isotopes with B12C4, B15C5 and B18C6 systems

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### 3.1 The effect of crown ethers with different lithium salts

**Table S1.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B12C4 system.

lithium salts	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
LiCl	15.403	-4.798	-11.956	1.028	1.020	1.007
LiNTf <sub>2</sub>	2.408	-11.614	-17.558	1.020	1.014	1.006

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.

**Table S2.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B15C5 system.

lithium salts	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
LiCl	15.647	-0.640	-8.583	1.024	1.016	1.008
LiNTf <sub>2</sub>	3.040	-17.684	-24.982	1.029	1.021	1.007

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.

**Table S3.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B18C6 system.

lithium salts	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
LiCl	15.183	8.0129	4.258	1.011	1.007	1.004
LiNTf <sub>2</sub>	2.351	-6.245	-8.664	1.011	1.009	1.002

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.

### 3.2 The effect of crown ethers under different electric field

**Table S4.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B12C4 system.

voltage	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
2 V	7.294	-12.301	-10.513	1.018	1.020	0.998
6 V	7.437	-9.610	-1.221	1.009	1.017	0.992
12 V	7.927	-9.817	-13.100	1.021	1.018	1.003
16 V	15.043	-4.798	-11.956	1.028	1.020	1.007

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.

**Table S5.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B15C5 system.

voltage	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
2 V	13.614	-10.097	9.214	1.004	1.024	0.981
6 V	7.615	-8.311	-10.424	1.016	1.014	1.002
12 V	7.893	-5.409	-11.337	1.022	1.016	1.006
16 V	15.647	-0.640	-8.583	1.024	1.016	1.008

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.

**Table S6.** The  $\alpha$  and  $\delta^7\text{Li}$  value of B18C6 system.

voltage	$\delta^7\text{Li}_A$	$\delta^7\text{Li}_O$	$\delta^7\text{Li}_C$	$\alpha(\text{C}/\text{A})$	$\alpha(\text{O}/\text{A})$	$\alpha(\text{C}/\text{O})$
2 V	14.488	6.700	6.455	1.008	1.008	1.000
6 V	14.704	5.565	4.574	1.010	1.009	1.001
12 V	14.955	5.438	3.626	1.011	1.009	1.002
16 V	15.183	8.0129	4.258	1.011	1.007	1.004

where subscript “A”, “O” and “C” represent anolyte, organic solution and catholyte respectively.