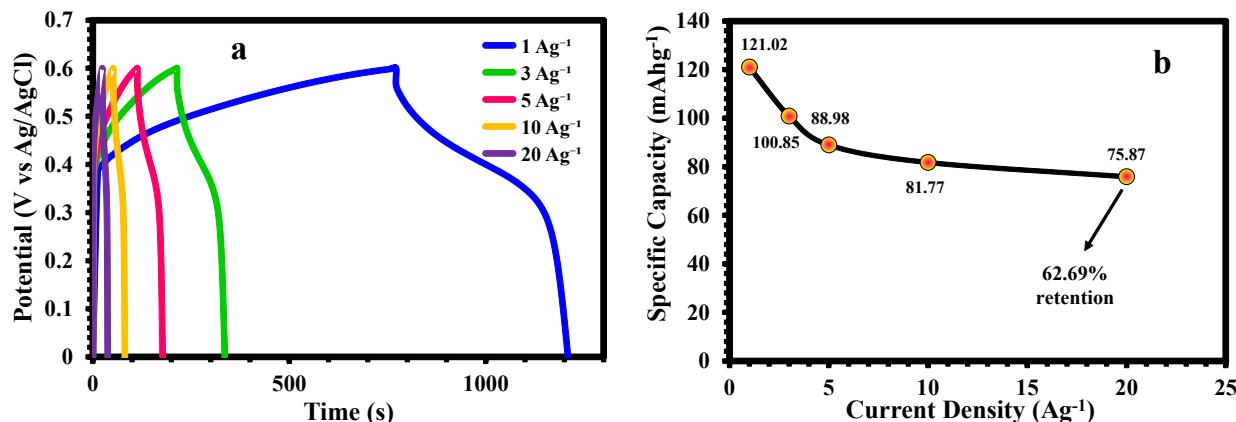


## Supporting Information

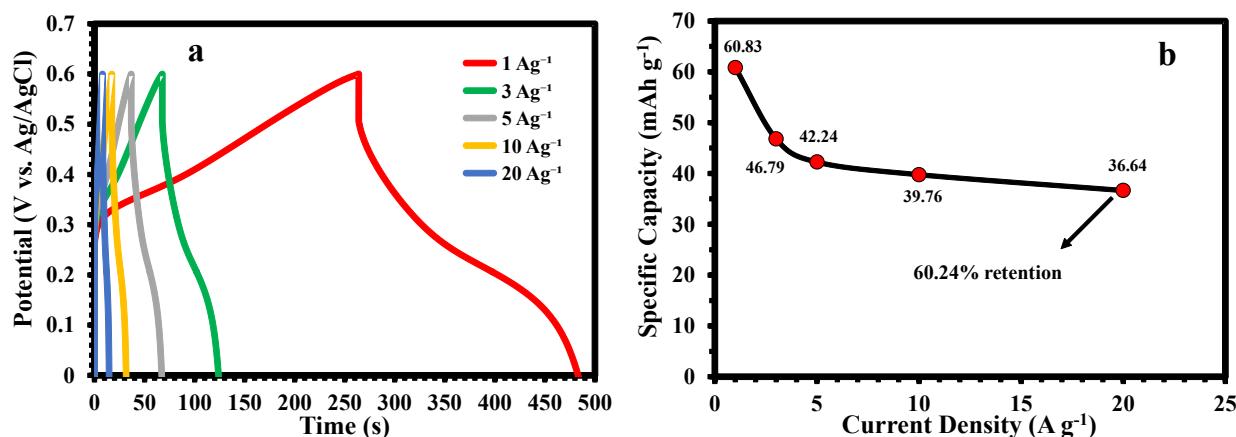
### Self-templated construction of hollow trimetallic MnNiCoP yolk-shell spheres assembled with nanosheets as a satisfactory electrode material for hybrid supercapacitors

Majid shirvani, Saied Saeed Hosseiny Davarani\*

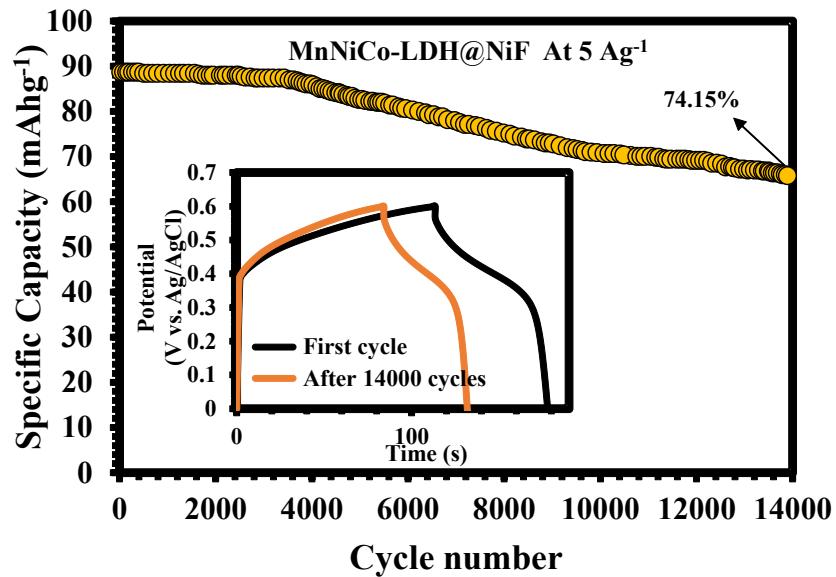
Department of Chemistry, Shahid Beheshti University, G. C., 1983963113, Evin,  
Tehran, Iran. E-mail: ss-hosseiny@sbu.ac.ir; Fax: +98 21 22431661;  
Tel: +98 21 22431661



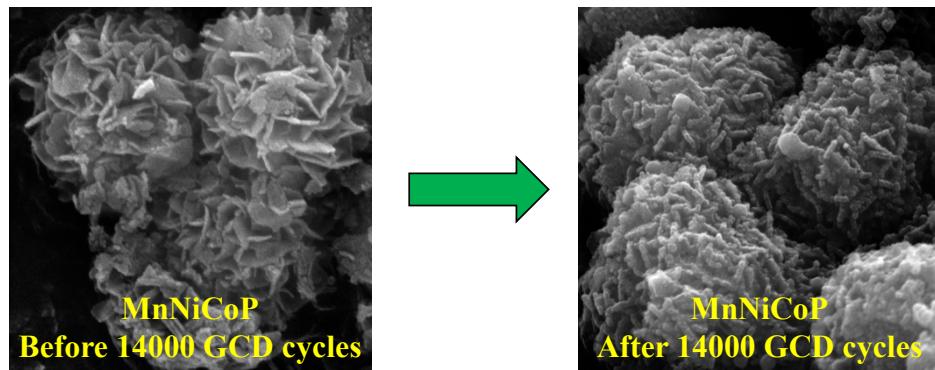
**Fig. S1** (a) GCD profiles of the MnNiCo-LDH@NiF electrode at different current densities (b) Rate performance of the MnNiCo-LDH@NiF electrode.



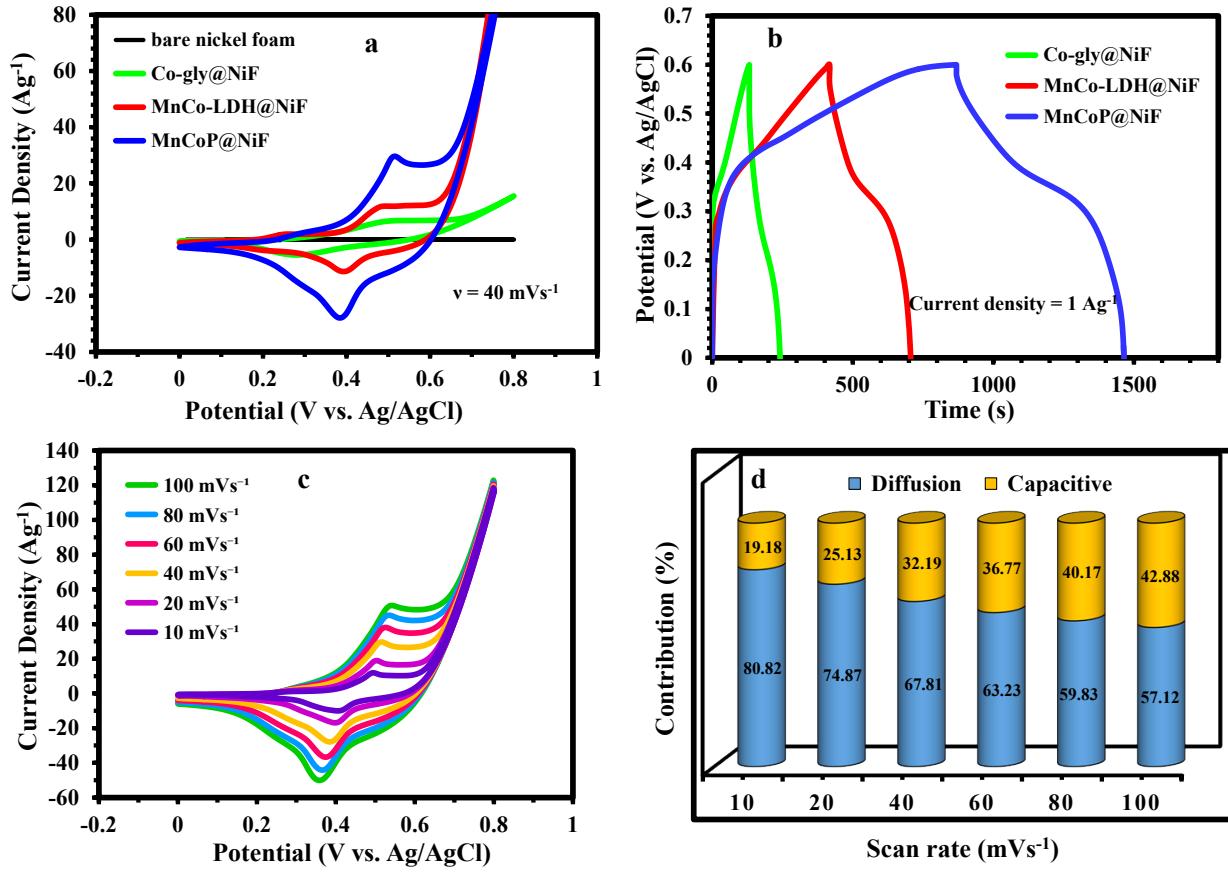
**Fig. S2** (a) GCD profiles of the NiCo-gly@NiF electrode at different current densities (b) Rate performance of the NiCo-gly@NiF electrode.



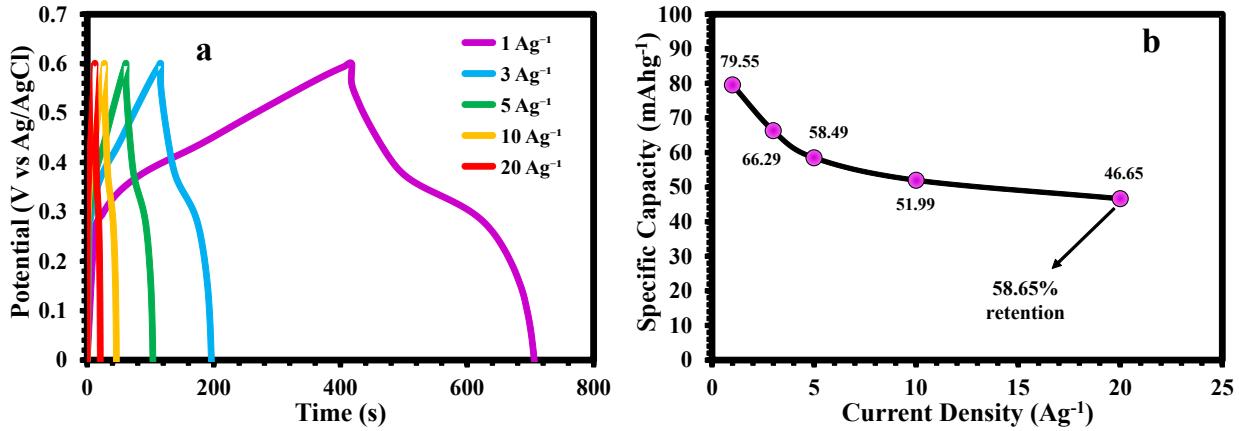
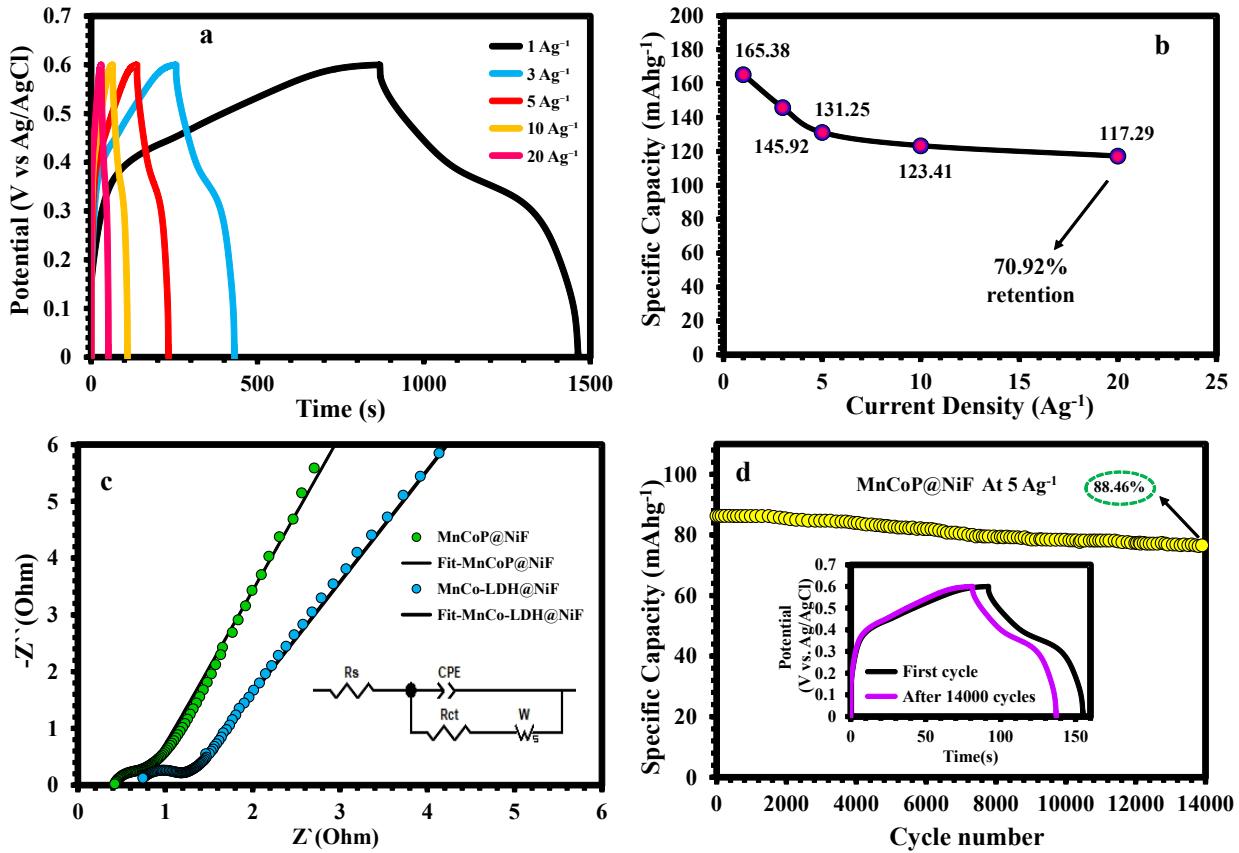
**Fig. S3** Durability of the MnNiCo-LDH@NiF electrode at  $5 \text{ Ag}^{-1}$ .



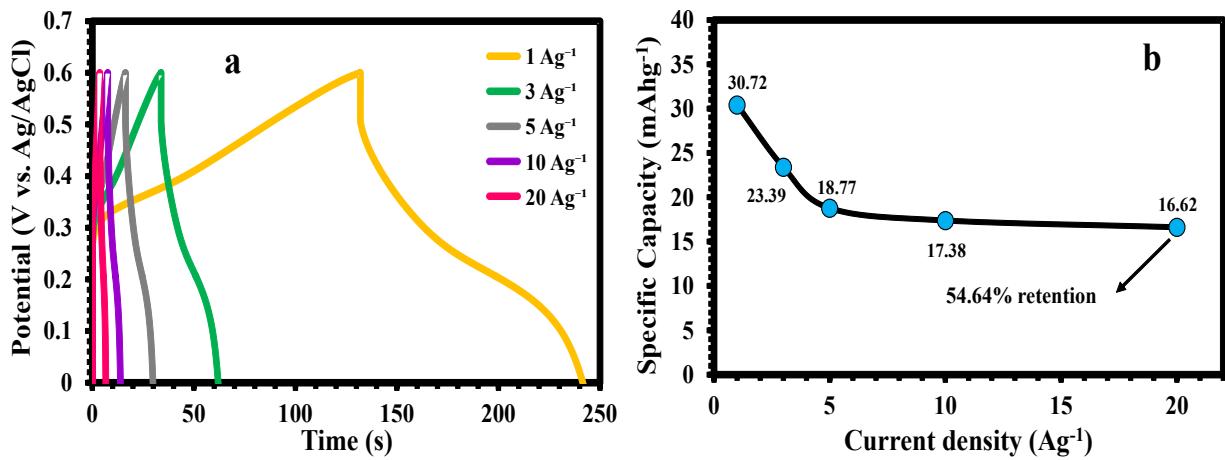
**Fig. S4** FE-SEM images of the MnNiCoP electrode material after 14,000 GCD cycles.



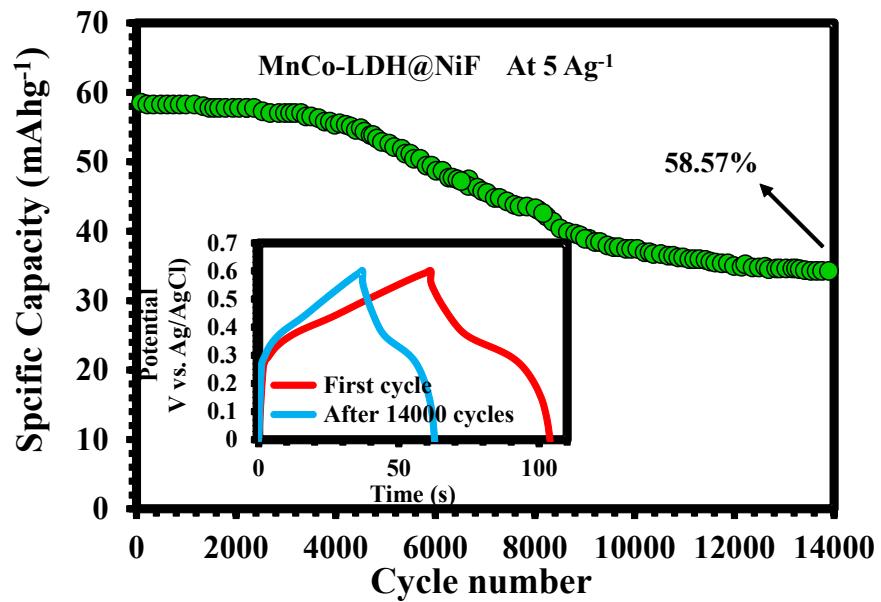
**Fig. S5** (a) CVs of the bare nickel foam, Co-gly@NiF, MnCo-LDH@NiF, and MnCoP@NiF electrodes at  $40 \text{ mVs}^{-1}$  (b) Charge-discharge curves of the Co-gly@NiF, MnCo-LDH@NiF, and MnCoP@NiF electrodes at  $1 \text{ Ag}^{-1}$  (c) CVs of the MnCoP@NiF electrode from 10 to  $100 \text{ mVs}^{-1}$  (d) The relative contribution of the capacitive and diffusion-controlled charge storage in the prepared MnCoP@NiF electrode at different scan rates



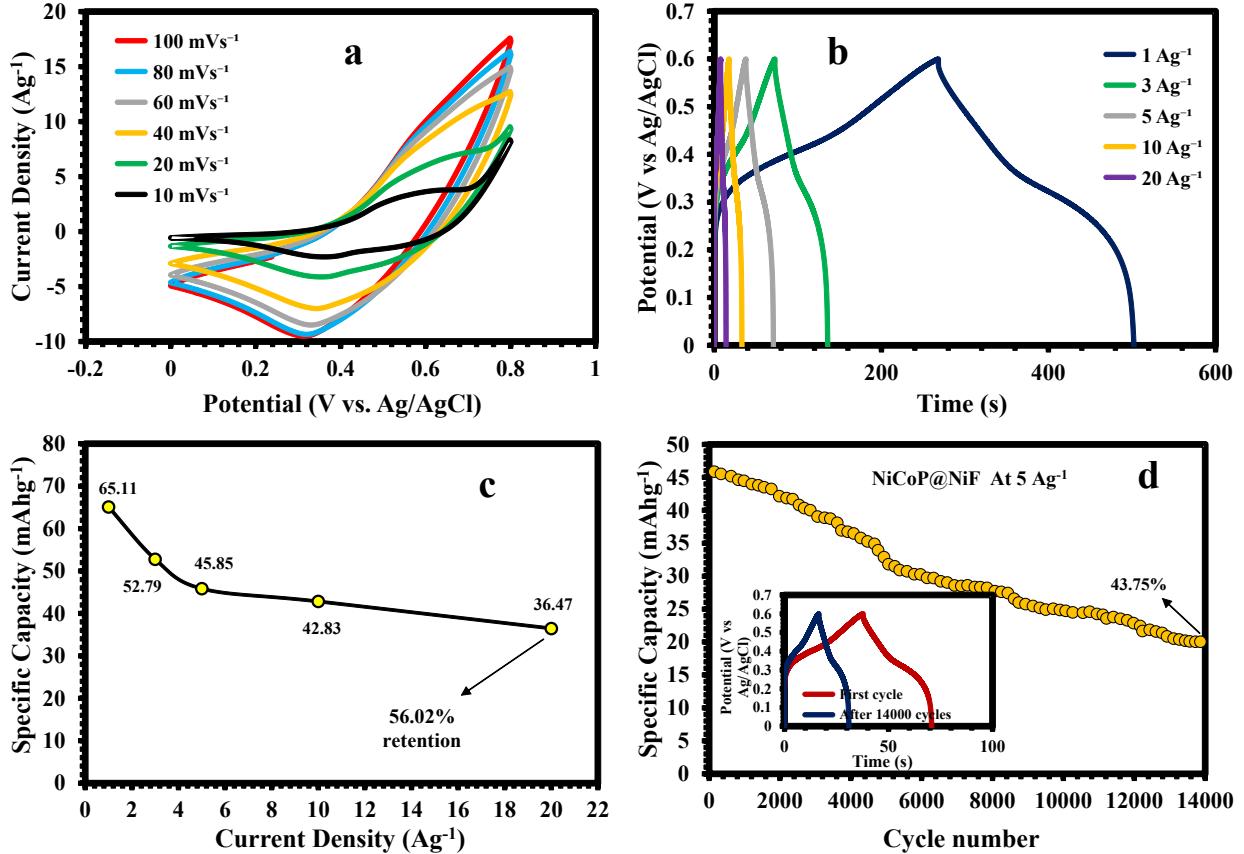
**Fig. S7** (a) GCD profiles of the MnCo-LDH@NiF electrode at different current densities (b) Rate performance of the MnCo-LDH@NiF electrode.



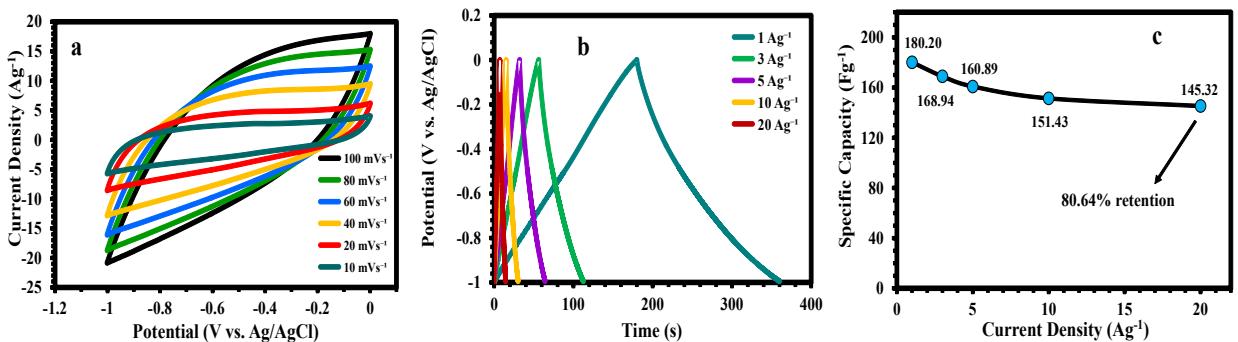
**Fig. S8** (a) GCD profiles of the Co-gly@NiF electrode at different current densities (b) Rate performance of the Co-gly@NiF electrode.



**Fig. S9** Durability of the MnCo-LDH@NiF electrode at 5  $\text{Ag}^{-1}$ .



**Fig. S10** (a) CVs of the NiCoP@NiF electrode at various scan rate of 10-100  $\text{mVs}^{-1}$  (b) GCD curves of the NiCoP@NiF electrode at various current densities of 1-20  $\text{Ag}^{-1}$  (c) Specific capacity vs. current density of the NiCoP@NiF electrode. (d) Durability of the NiCoP@NiF electrode at 5  $\text{Ag}^{-1}$ .



**Fig. S11** (a) CVs of the AC-based electrode at various scan rate of 10-100  $\text{mVs}^{-1}$  (b) GCD curves of the AC-based electrode at various current densities of 1-20  $\text{Ag}^{-1}$  (c) Specific capacitance vs. current density of AC-based electrode.

**Table S1** Comparison of the performance of the MnNiCoP electrode material with other previously reported materials

Composition	Capacity(mAhg <sup>-1</sup> )	Cycles, retention	Rate capability	ED(Wkg <sup>-1</sup> )	Reference
Ni <sub>2</sub> P/NiCoP	205.92	3000, 89.2%	75.5% at 20 Ag <sup>-1</sup>	44.5	1
O-CoxNiyP	199.19	5000, 95.1%	66.7% at 20 Ag <sup>-1</sup>	47.5	2
Cu-Co-P	110.6	10000, 89%	83.1% at 10 Ag <sup>-1</sup>	41.3	3
NiCoP/NC	172.18	8000, 75.5%	77.4% at 16 Ag <sup>-1</sup>	52.5	4
NiCoP	182.91	5000, 80.7%	66% at 30 Ag <sup>-1</sup>	41.3	5
Ni <sub>0.4</sub> Mn <sub>1.6</sub> P	176.66	2000, 75%	-	21.1	6
Ni-Co-P-3	213.1	5000, 85%	86% at 20 Ag <sup>-1</sup>	48.4	7
<b>MnNiCoP</b>	<b>291.24</b>	<b>14000, 91.30%</b>	<b>80% at 20 Ag<sup>-1</sup></b>	<b>57.03</b>	<b>This work</b>

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