

Supporting Information

1T-WS₂ “electron pump” regulate charge transfer over ZnCdS/NiV-LDH p-n heterostructures for enhanced photocatalytic hydrogen evolution

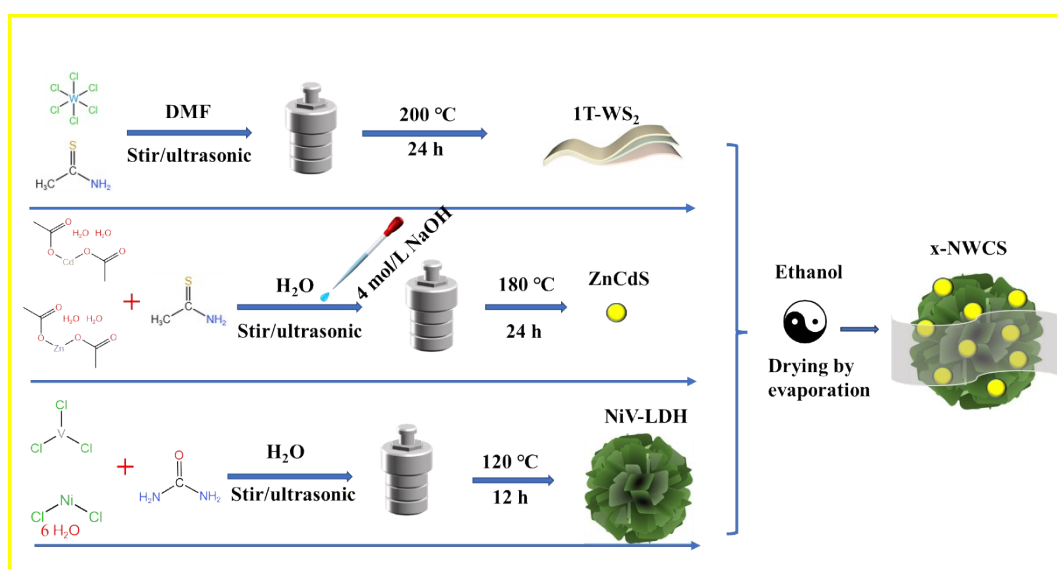


Fig. S1 Compound Catalyst preparation process.

Table S1. The S_{BET} , pore volume and pore diameter distribution for each catalyst.

Samples	S_{BET} ($\text{m}^2 \cdot \text{g}^{-1}$)	Pore volume ($\text{cm}^3 \cdot \text{g}^{-1}$)	Average pore size (nm)
ZnCdS	7.35	0.09	49.38
1T-WS ₂	5.02	0.02	20.13
NiV-LDH	65.69	0.53	32.20
10-WCS	8.79	0.06	27.69
20-NWCS	10.56	0.13	47.52

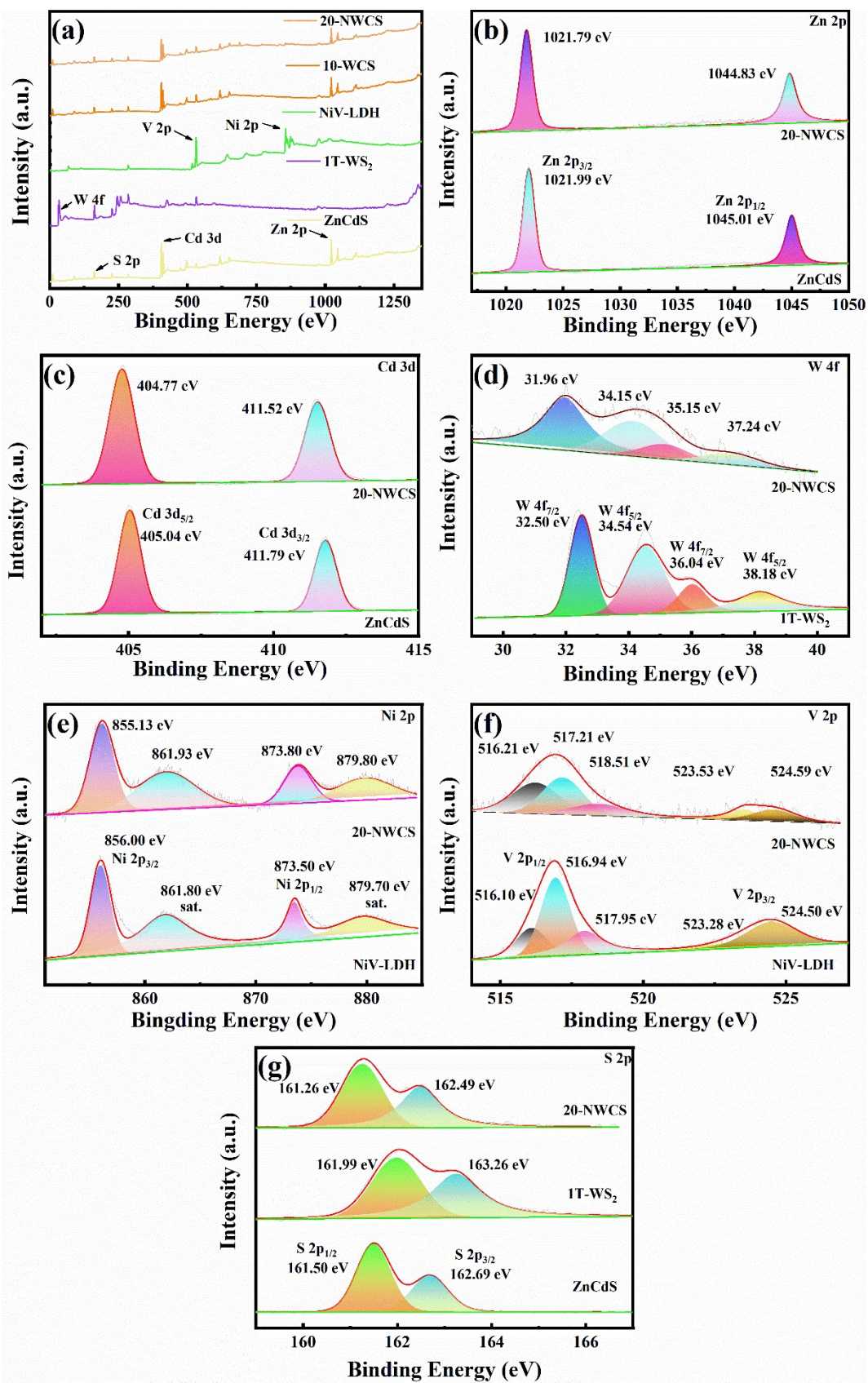


Fig. S2 XPS of (a) full spectrum; (b) Zn 2p; (c) Cd 3d; (d) W 4f; (e) Ni 2p; (f) V 2p and (g) S 2p.

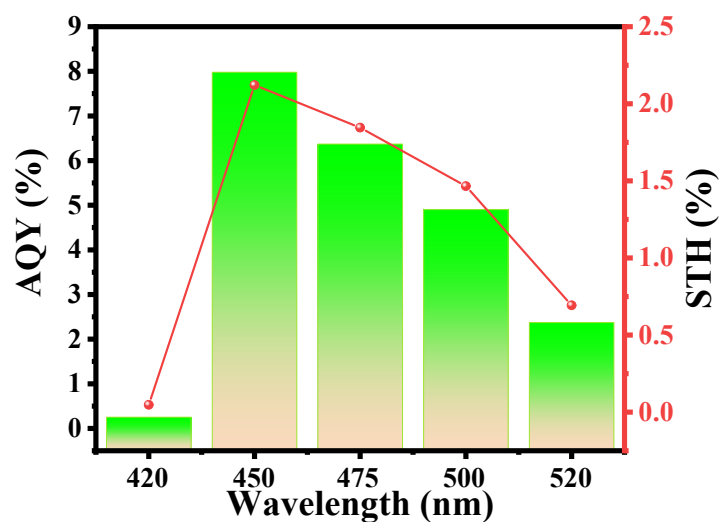


Fig. S3 AQE testing and STH conversion efficiency of 20-NWCs.

Table S2 Comparison of ZnCdS based composite materials for photocatalytic hydrogen evolution.

Photocatalyst	Light Source	Sacrificial agents	Production rate	AQE	Refs
				(%)/n m	
ZnCdS/1T-WS ₂ /NiV-LDH	5 W LED	Lactic acid	22.37 mmol·g ⁻¹ ·h ⁻¹	7.98 (450)	This
ZnCdS/NiCoP	5 W LED	Lactic acid	11.64 mmol·g ⁻¹ ·h ⁻¹	7.93 (450)	[1]
ZnCdS/P-graphdiyne	5 W LED	Lactic acid	10.39 mmol·g ⁻¹ ·h ⁻¹	3.35 (420)	[2]
ZnCdS/Ce-MOF	300 W Xe lamp	Na ₂ S/Na ₂ SO ₃	3.95 mmol·g ⁻¹ ·h ⁻¹	No	[3]
ZnCdS/Co ₃ P	5 W LED	Na ₂ S/Na ₂ SO ₃	17.76 mmol·g ⁻¹ ·h ⁻¹	No	[4]
ZnCdS/CuS/Cu ₉ S ₅	5 W LED	Na ₂ S/Na ₂ SO ₃	13.82 mmol·g ⁻¹ ·h ⁻¹	8.87 (475)	[5]
ZnCdS/Ni ₃ C	300 W Xe lamp	Na ₂ S/Na ₂ SO ₃	3.31 mmol·g ⁻¹ ·h ⁻¹	No	[6]
ZnCdS/NiB	LED	Lactic acid	7.70 mmol·g ⁻¹ ·h ⁻¹	13.3 (not found)	[7]
ZnCdS/In ₂ O ₃	300 W Xe lamp	Na ₂ S/Na ₂ SO ₃	1.1 mmol·g ⁻¹ ·h ⁻¹	0.32	[8]

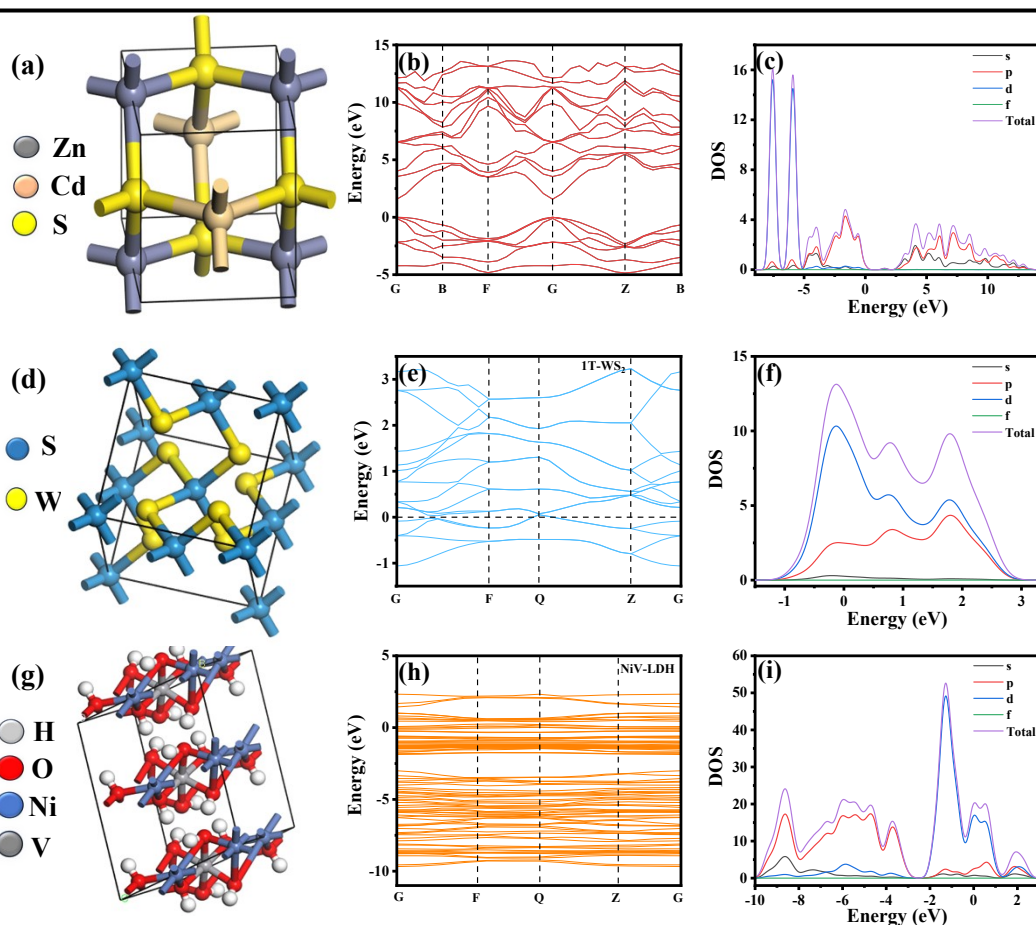


Fig. S4 Rhythm optimization model of (a) ZnCdS; (d) 1T-WS₂; (g) NiV-LDH; band structure of (b) ZnCdS; (e) 1T-WS₂; (h) NiV-LDH; DOS and PDOS of (c) ZnCdS; (f) 1T-WS₂; (i) NiV-LDH.

References

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