Supporting Information

A moderate method for in-situ growing Fe-based LDHs on Ni foam for catalyzing the oxygen evolution reaction

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Fig. S1. Digital photograph of NiFe-LDH@NF.



Fig. S2. Ni 2p X-ray photoelectron spectroscopy (XPS) spectra of NF.



Fig. S3. The digital photographs of bare NF and NF treated by immersion in FeSO₄/Ni(NO₃)₂, FeSO₄, Fe(NO₃)₃, FeCl₃, O₂-free and O₂-rich FeSO₄/ Ni(NO₃)₂ solution.



Fig. S4. SEM images of nickel foam in (a) Ni(CH₃COO)₂, (b) NiCl₂.



Fig. S5. SEM-EDS spectra of CoFe-LDH@NF.



Fig. S6. SEM-EDS spectra of NiCoFe-LDH@NF.



Fig. S7. The LSV curves for OER without iR compensation.



Fig. S8. EIS equivalent circuit diagram.



Fig. S9. CV curves of (a) NF, (b) NiFe-LDH@NF with different scan

rates (20-100 mV s⁻¹) in 1 M KOH.



Fig. S10. SEM images of NiFe-LDH@NF after long-term testing.



Fig. S11. The OER activities of NiFe-LDH@NF prepared in solutions with different Ni/Fe molar ratio



Fig. S12. The OER activities of different periods of NiFe-LDH@NF.



Fig. S13. The OER activities of NiFe-LDH@NF prepared in oxygenenriched and oxygen-free Fe/Ni solutions.



Fig. S14. OER electrocatalytic properties of different temperatures of NiFe-LDH@NF.

Materials	$OER_{\eta 10} \left(mV \right)$	$OER_{\eta 100} (mV)$	Ref.
NiFe-LDH@NF	206	239	This
			Wor
			k
NiFeCe-LDH@CP	232	267	1
NiFeCo-LDH	249	*	2
NiFe-CuCoLDH	212	262	3
NiFeLDH/C on NF	210	*	4

Table S1. Comparison of OER activity data for various catalysts.

ex-Ir-Ni(OH) ₂	270	*	5
NiCo-LDH-10min	250	*	6
NiFe@TiO _{2-x}	*	300	7
Ni ₃ Fe _{0.9} Cr _{0.1} /CACC	239	302	8
S-FeOOH/IF	244	308	9
MoNiFe-27% (oxy)hydroxide	242	290	10
NiCoP@NiMnLDH/NF	*	293	11
Ni ₃ S ₂ @FeNi ₂ S ₄ @NF	235	379	12
FeIr/NF	220	*	13

Marked: CP, carbon paper; CACC, CO₂-Activated carbon cloth; IF, iron foam.

References

- Y. Y. Liao, R. C. He, W. H. Pan, Y. Li, Y. Y. Wang, J. Li and Y. X. Li, *Chem. Eng. J.*, 2023, 464, 142669.
- Y. S. Park, J. Y. Jeong, M. J. Jang, C. Y. Kwon, G. H. Kim, J. Jeong,
 J. H. Lee, J. Y. Lee and S. M. *Choi*, *J. Energy Chem.*, 2022, 75, 127-134.
- L. Yu, J. Y. Xiao, C. Q. Huang, J. Q. Zhou, M. Qiu, Y. Yu, Z. F. Ren,
 C. W. Chu and J. C. Yu, Proc. *Natl. Acad. Sci. U. S. A.*, 2022, 119, e2202382119.
- Z. H. Zhang, C. L. Wang, X. L. Ma, F. Liu, H. Xiao, J. Zhang, Z. Lin and Z. P. Hao, *Small.*, 2021, 17, 2103785.

- J. L. Liu, J. X. Xiao, Z. Y. Wang, H. M. Yuan, Z. G. Lu, B. C. Luo,
 E. K. Tian and G. I. N. Waterhouse, *ACS Catal.*, 2021, **11**, 5386-5395.
- H. M. Sun, Y. L. Miao, T. Wu and Q. Wang, *Chem. Commun.*, 2020, 56, 872-875.
- K. Zhang, T. T. Wan, H. Y. Wang, Y. H. Luo, Y. M. Shi, Z. S. Zhang,
 G. H. Liu and J. D. Li, J. *Colloid Interface Sci.*, 2023, 645, 66-75.
- J. F. Zheng, J. X. Zhang, L. H. Zhang, W. F. Zhang, X. J. Wang, Z. M. Cui, H. Y. Song, Z. X. Liang and L. Du, ACS Appl. Mater. Interfaces., 2022, 14, 19524-19533.
- X. Chen, Q. C. Wang, Y. W. Cheng, H. L. Xing, J. Z. Li, X. J. Zhu, L.
 B. Ma, Y. T. Li and D. M. Liu, *Adv. Funct. Mater.*, 2022, 32, 2112674.
- Z. Y. He, J. Zhang, Z. H. Gong, H. Lei, D. Zhou, N. A. Zhang, W. J.
 Mai, S. J. Zhao and Y. Chen, *Nat. Commun.*, 2022, 13, 2191.
- P. Wang, J. Qi, X. Chen, C. Li, W. P. Li, T. H. Wang and C. H. Liang, ACS Appl. *Mater. Interfaces.*, 2020, **12**, 4385-4395.
- Y. Y. Yang, H. X. Meng, C. Kong, S. H. Yan, W. X. Ma, H. Zhu, F.
 Q. Ma, C. J. Wang and Z. A. Hu, J. *Colloid Interface Sci.*, 2021, **599**, 300-312.
- J. L. Chen, Y. M. Wang, G. F. Qian, T. Q. Yu, Z. L. Wang, L. Luo,
 F. Shen and S. B. Yin, *Chem. Eng. J.*, 2021, **421**, 129892.