

## Supplementary Material

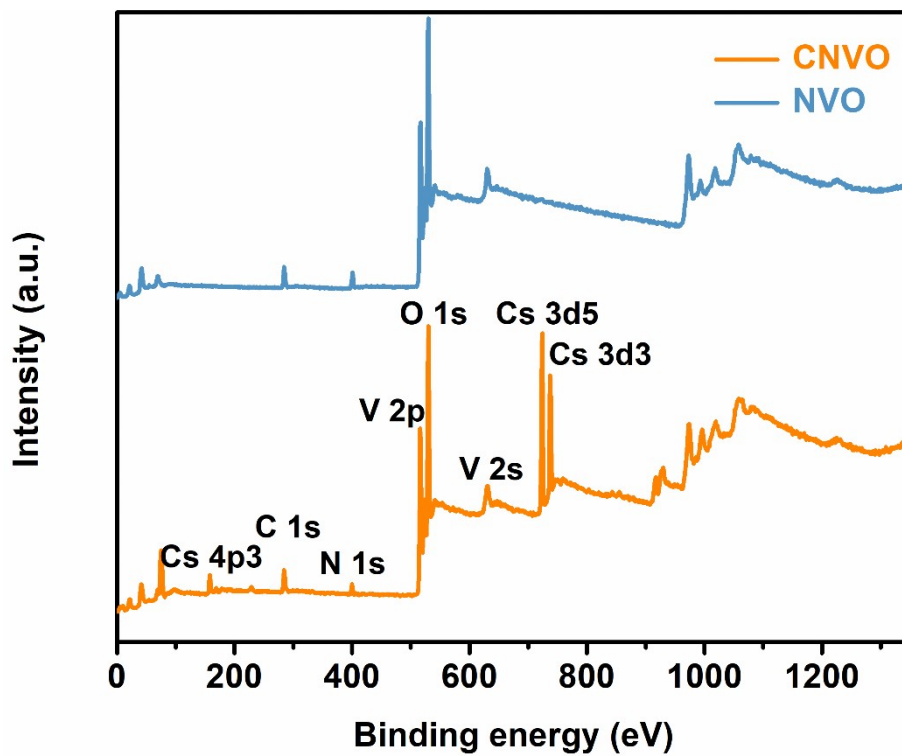


Fig. S1. XPS spectra of CNVO and NVO.

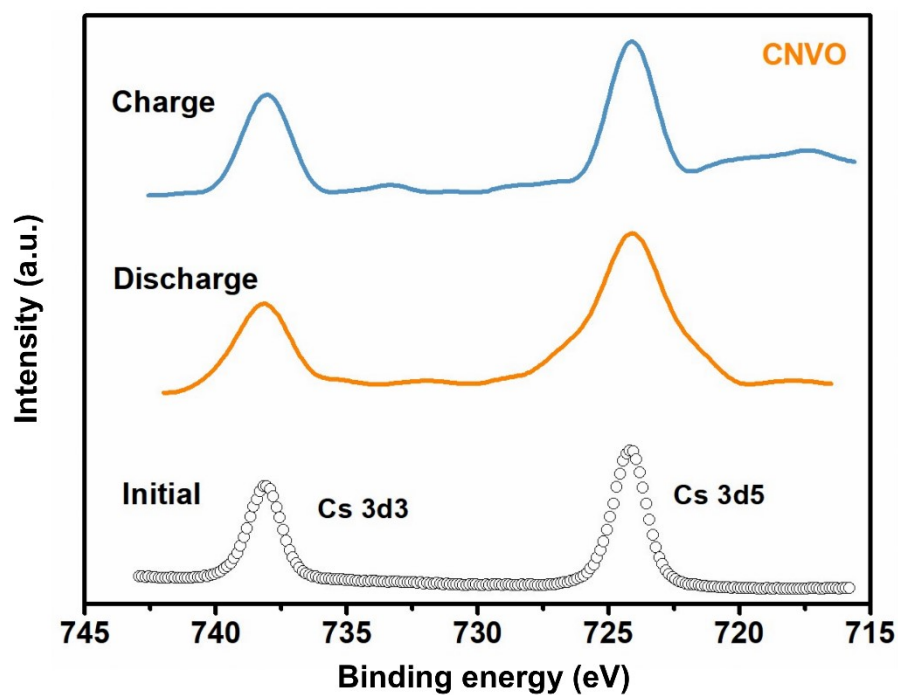


Fig. S2. XPS spectra of Cs 3d3 and Cs 3d5 of CNVO.

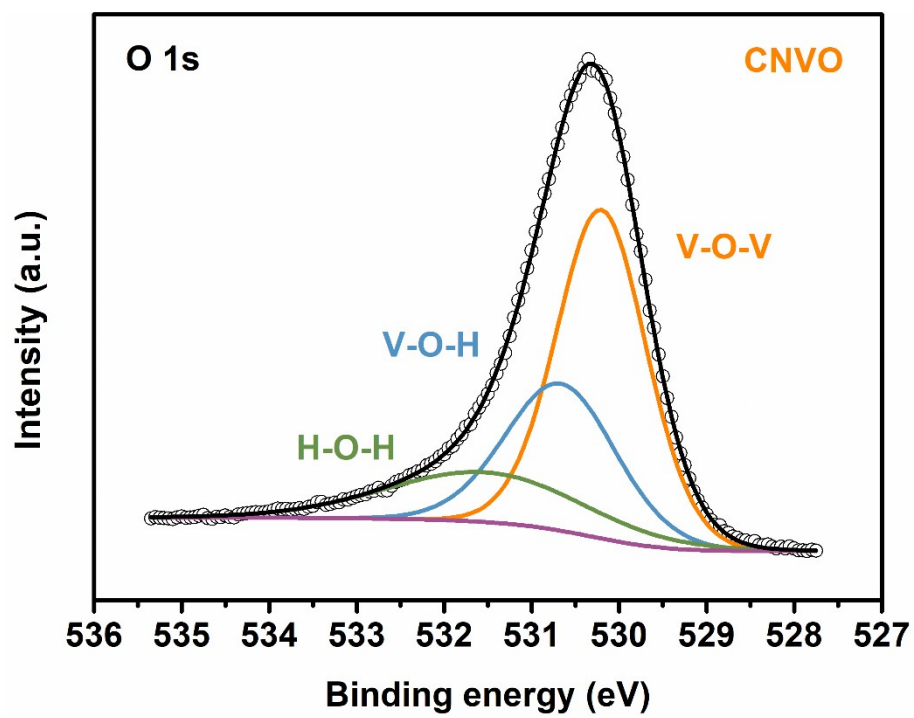


Fig. S3. XPS spectra of O 1s of CNVO.

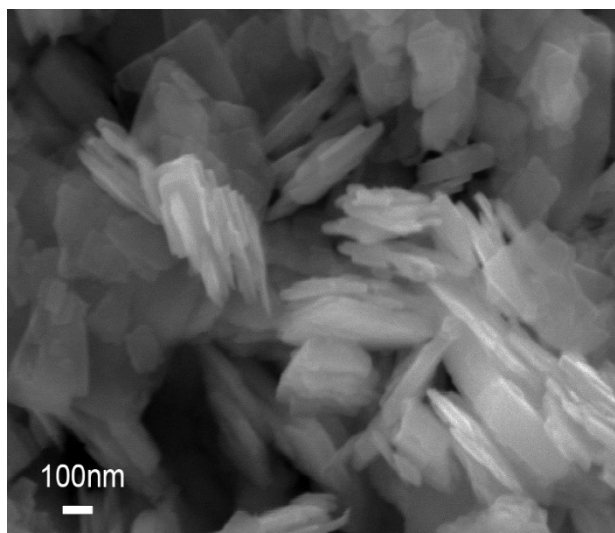
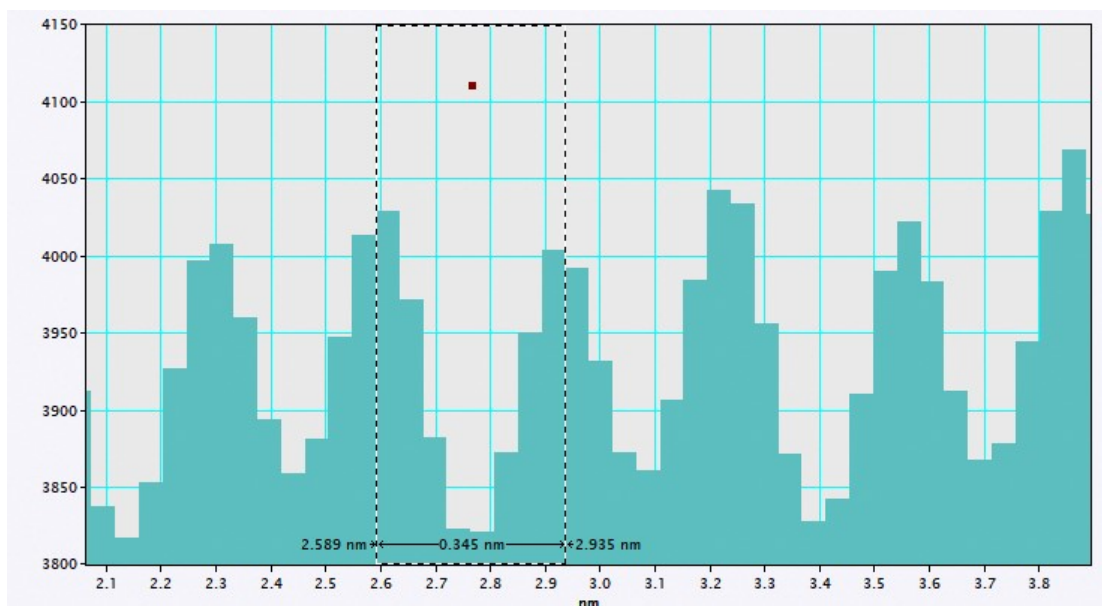
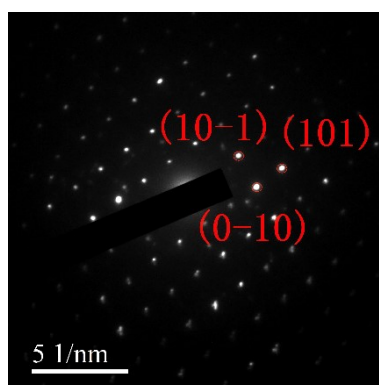


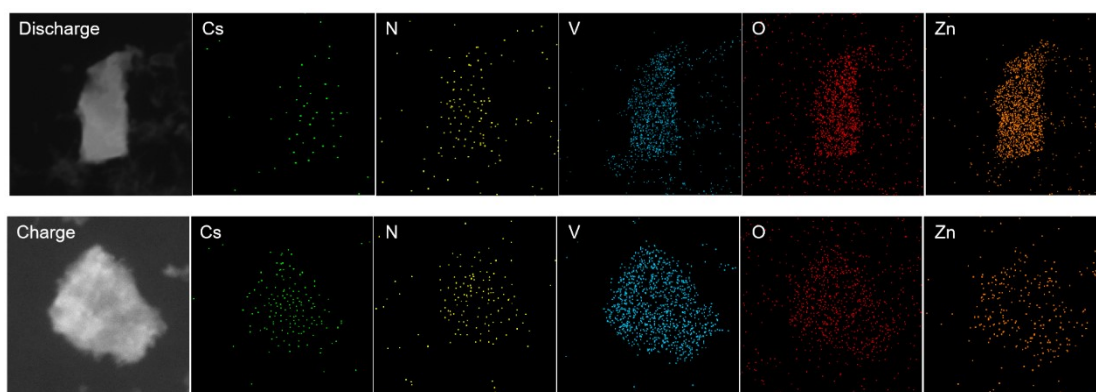
Fig. S4. Field Emission Scanning Electron Microscope (FE-SEM) image of CNVO.



**Fig.S5.** Lattice diffraction stripe size of CNVO.



**Fig. S6.** SAED image of CNVO.



**Fig.S7.** EDS mapping images of CNVO in different charging and discharging states.

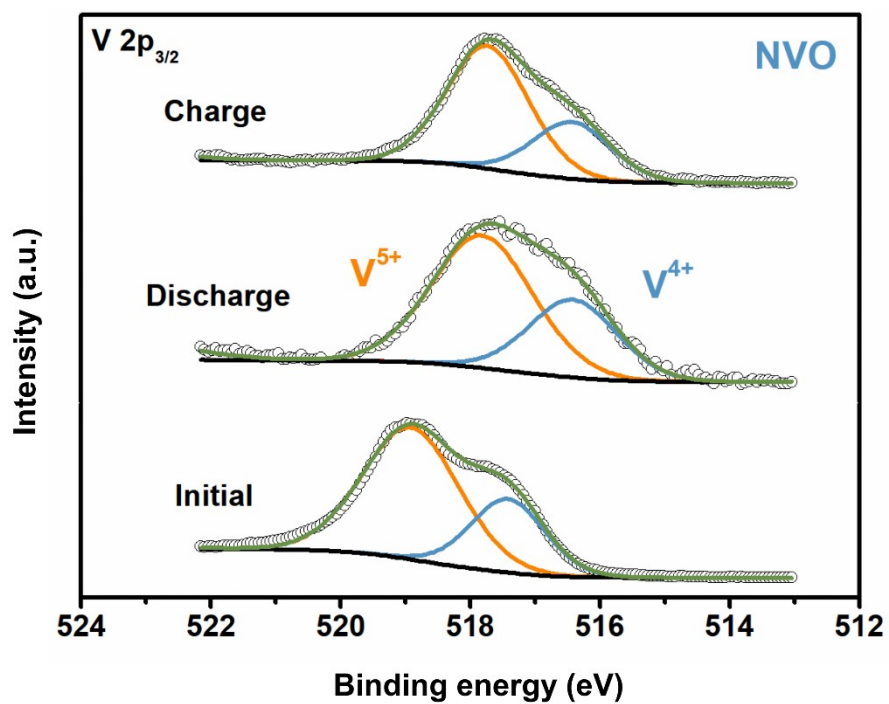


Fig. S8. XPS spectra of V 2p<sub>3/2</sub> of NVO.

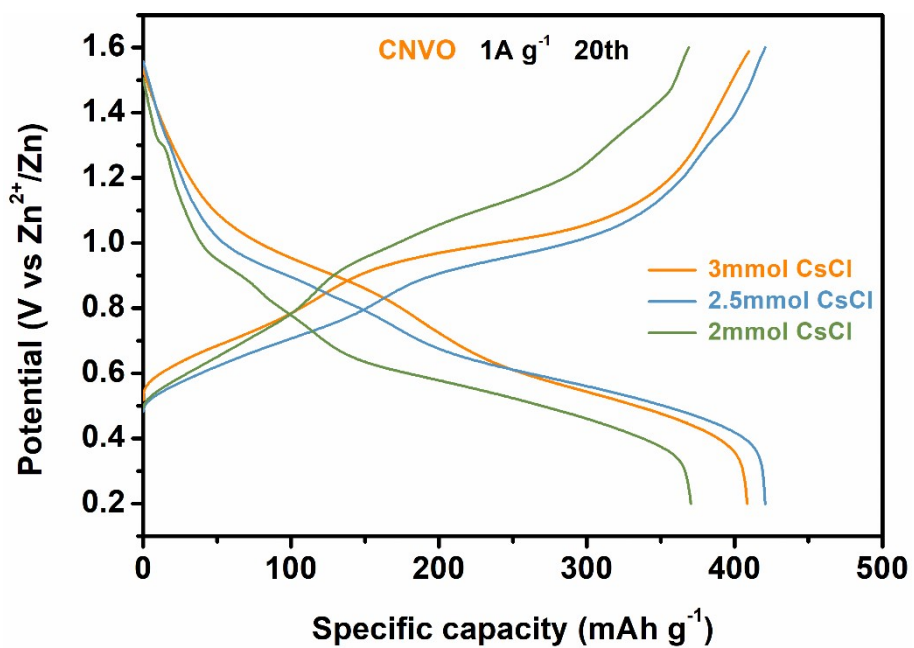


Fig. S9. 20th charge/discharge curves of CNVO with different Cs<sup>+</sup> molar ratios at 1 A g<sup>-1</sup> current density.

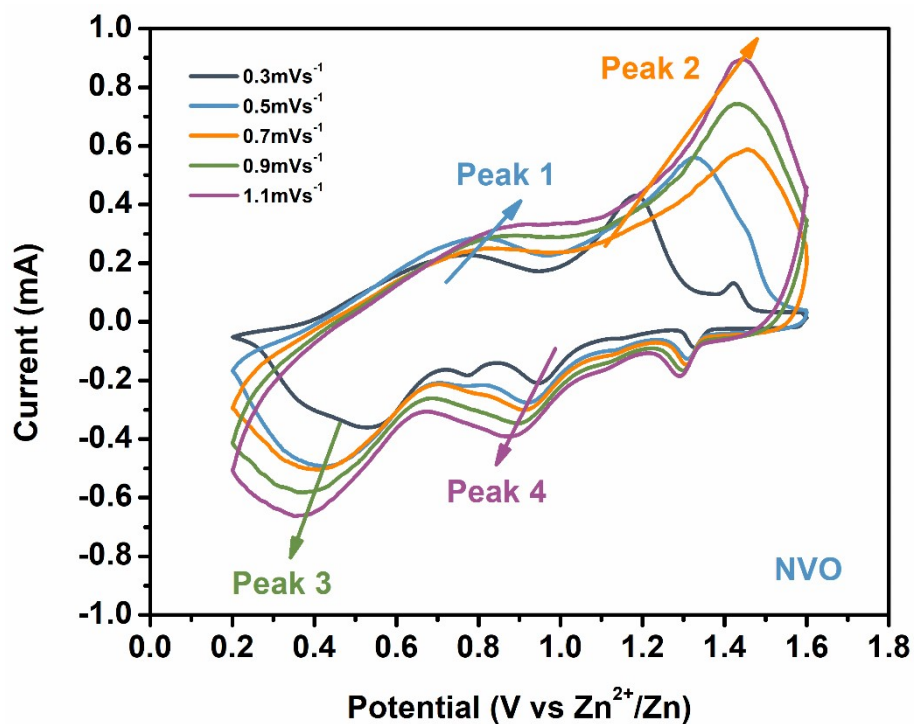


Fig. S10. CV curves of NVO at different scan rates.

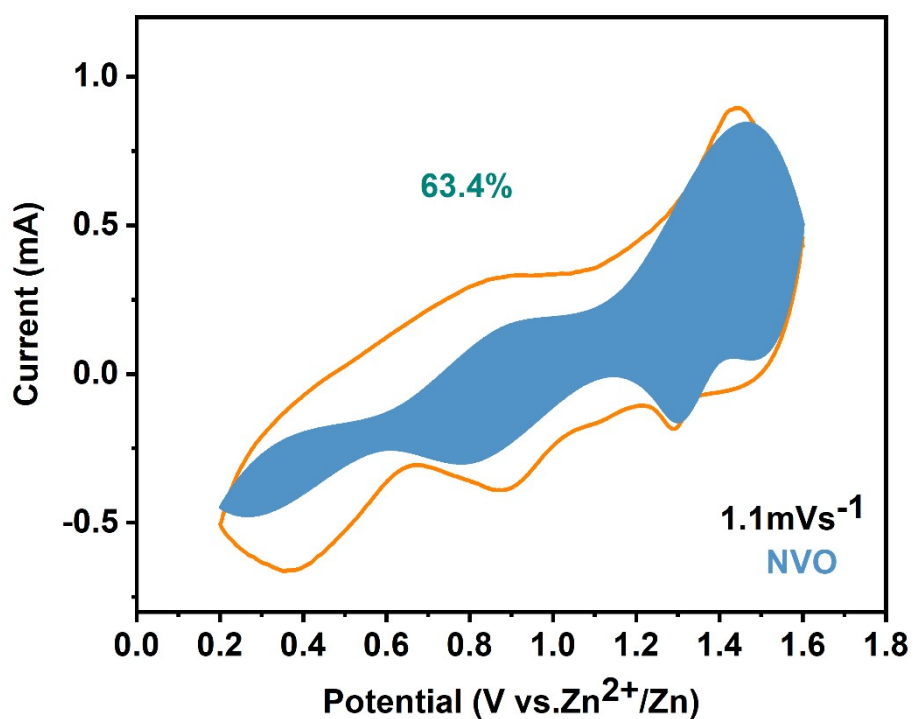


Fig. S11. Pseudocapacitance fitting results of CNVO at 1.1  $mVs^{-1}$  scan rate.

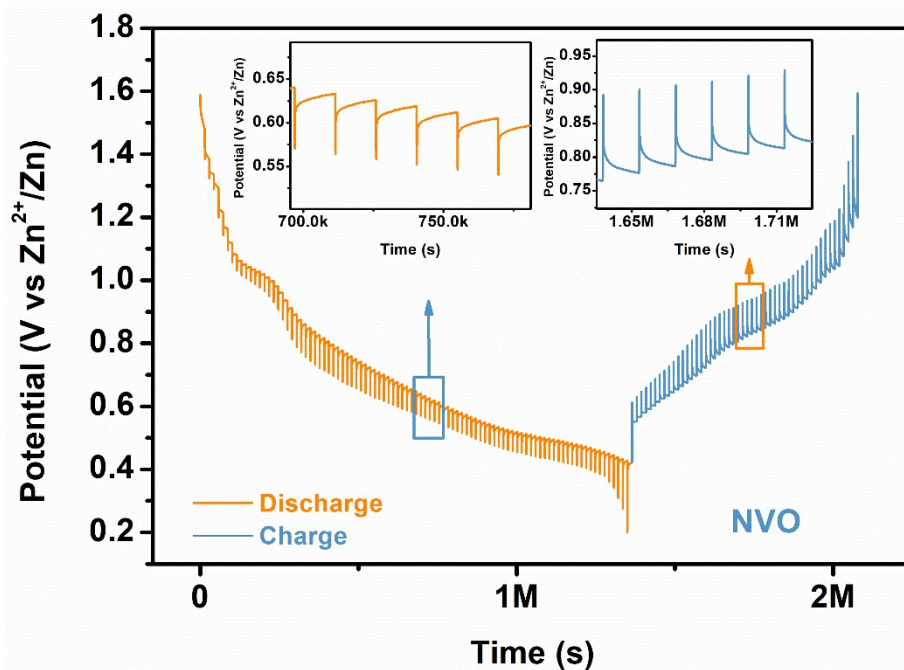


Fig. S12. GITT curves of NVO.

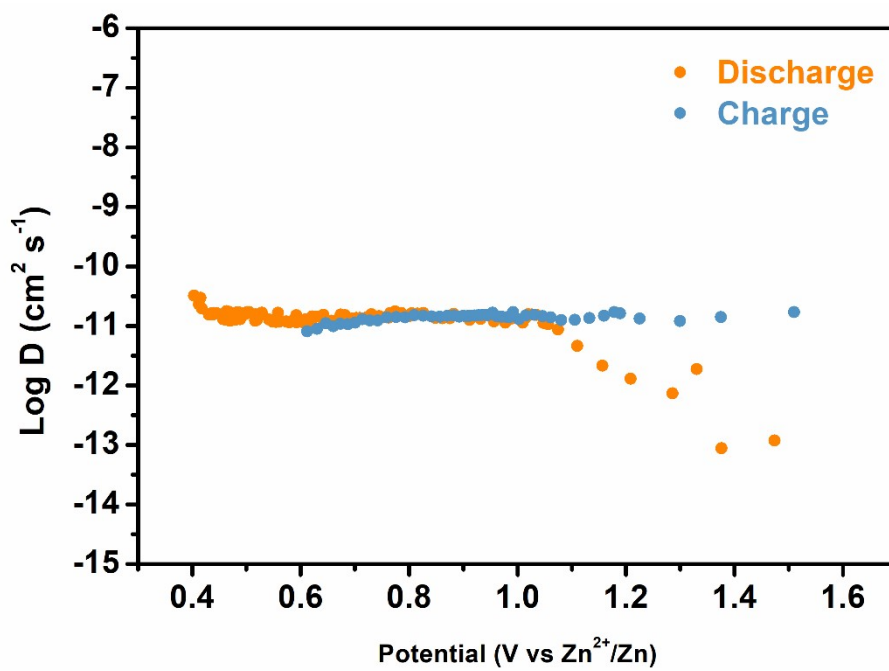
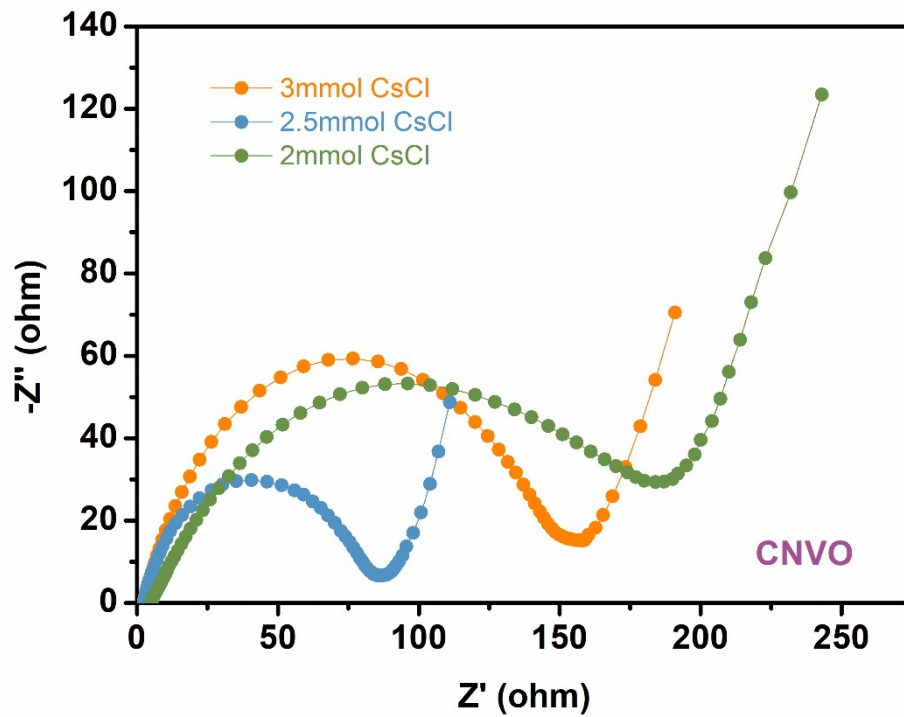
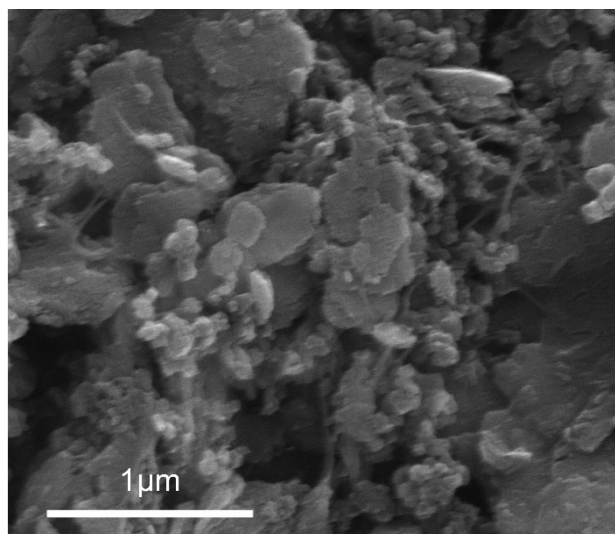


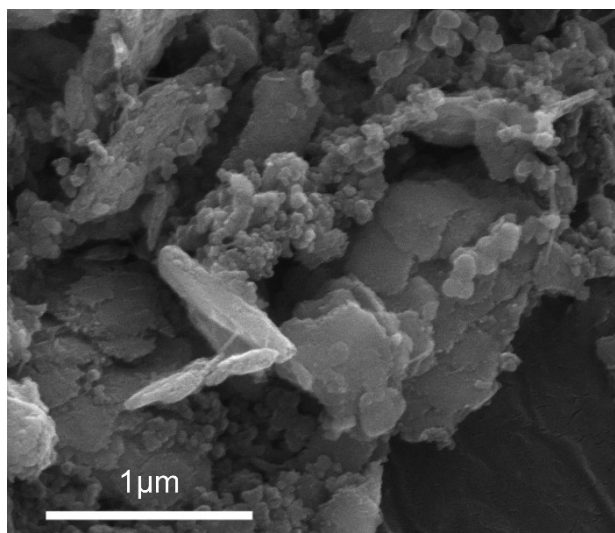
Fig. S13. Calculated diffusion coefficients of NVO.



**Fig. S14.** EIS curves of CNVO with different Cs<sup>+</sup> molar ratios.



**Fig. S15.** SEM image of CNVO cathode material charged to 1.6 V for the 20th time.



**Fig. S16.** SEM image of CNVO cathode material charged to 1.6 V for the 50th time.



**Table. S1.** Inductively coupled plasma-optical emission spectroscopy of CNVO samples.

Sample Quality (g)	Fixed volume $V_0$ (ml)	Test elements	Test element concentration $C_0$ (mg/L)	Dilution times $f$	The elemental concentration of digestion solution/original sample solution $C_1$ (mg/L)	Sample elemental content $C_x$ (mg/kg)	Sample elemental content $W$ (%)
0.1264	25.00	Cs	0.04	1000	35.80	7080.70	0.7081%
0.1264	25.00	V	2.08	1000	2080.83	411557.04	41.1557%
0.0125	25.00	N	21.96	1	21.96	43922.00	4.3922%

**Table. S2.** Comparison of the electrochemical performance of CNVO with other reported zinc-ion batteries.

Cathode	Electrolyte	Specific capacity	Rate performance	Cycling stability	Ref.
$(\text{NH}_4)_2\text{V}_4\text{O}_9 \cdot 0.5\text{H}_2\text{O}$	2 M $\text{ZnSO}_4$ aqueous solution	374.3 mAh $\text{g}^{-1}$ at 0.2 A $\text{g}^{-1}$	101 mAh $\text{g}^{-1}$ at 15 A $\text{g}^{-1}$	84% after 1000 cycles at 5 A $\text{g}^{-1}$	[1]
$\text{Mg}_{0.34}\text{V}_2\text{O}_5 \cdot 0.84\text{H}_2\text{O}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	352 mA h $\text{g}^{-1}$ at 100 mA $\text{g}^{-1}$	264 mA h $\text{g}^{-1}$ at 1000 mA $\text{g}^{-1}$	~97 % capacity retention for at least 2000 cycles at 5000 mA $\text{g}^{-1}$	[2]
$\delta\text{-Ni}_{0.25}\text{V}_2\text{O}_5 \cdot n\text{H}_2\text{O}$	---	402 mAh $\text{g}^{-1}$ at 0.2 A $\text{g}^{-1}$	225 mAh $\text{g}^{-1}$ at 5 A $\text{g}^{-1}$	98 % over 1200 cycles at 5 A $\text{g}^{-1}$	[3]
$\text{NaCa}_{0.6}\text{V}_6\text{O}_{16} \cdot 3\text{H}_2\text{O}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	347 mAh $\text{g}^{-1}$ at 0.1 A $\text{g}^{-1}$	154 mAh $\text{g}^{-1}$ at 5 A $\text{g}^{-1}$	94% after 2,000 cycles at 2 A $\text{g}^{-1}$	[4]
$\text{V}_6\text{O}_{13}@CC$	3 M $\text{ZnSO}_4$	431 mAh $\text{g}^{-1}$ at 0.2 A $\text{g}^{-1}$	227 mAh $\text{g}^{-1}$ at 9 A $\text{g}^{-1}$	nearly 99% after 1000 cycles at 9 A $\text{g}^{-1}$	[5]
Al-doped $\text{V}_{10}\text{O}_{24} \cdot 12\text{H}_2\text{O}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	290 mAh $\text{g}^{-1}$ at 0.375 A $\text{g}^{-1}$	294.5 mAh $\text{g}^{-1}$ at 5 A $\text{g}^{-1}$	98% capacity retention after 3000 cycles	[6]

## References

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