

Supporting Information

Frogspawn inspired twin Mo₂C/Ni composite with conductive fibrous network as robust bifunctional catalyst for advanced anion exchange membrane electrolyzers

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Scherrer equation

The average particle size of the grains was calculated according to the **Equation S1**.

$$D = \frac{k\lambda}{\beta \cos \theta} \quad (1)$$

In equation S1, k is the scherrer constant (0.9), λ is the X-ray wavelength (0.15406 nm), β is the half width of the diffraction peak, θ is the diffraction angle.

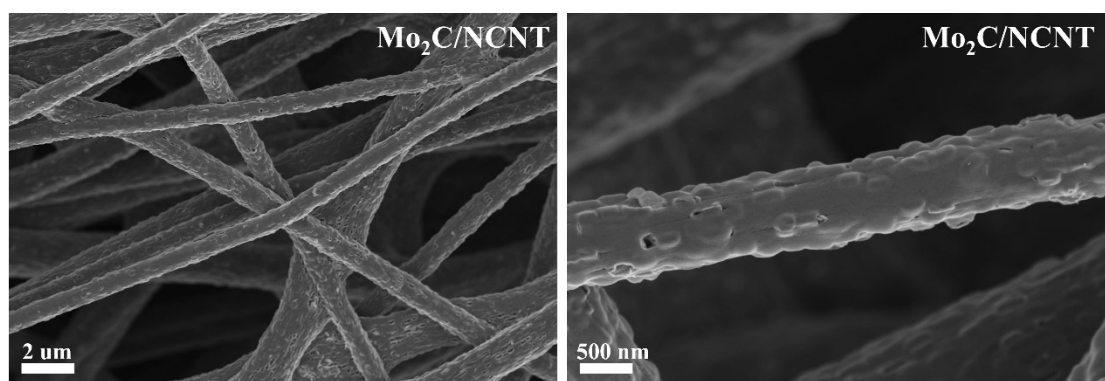


Figure S1. SEM image of the Mo₂C/NCNTs.

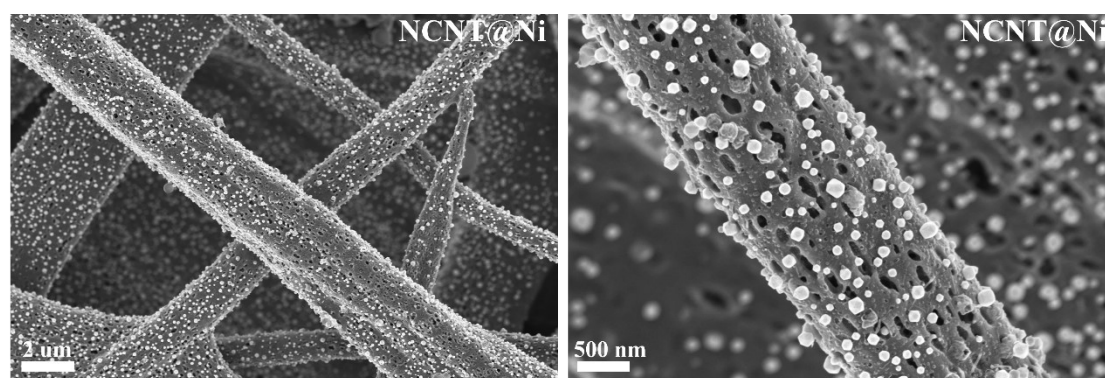


Figure S2. SEM image of the NCNTs@Ni.

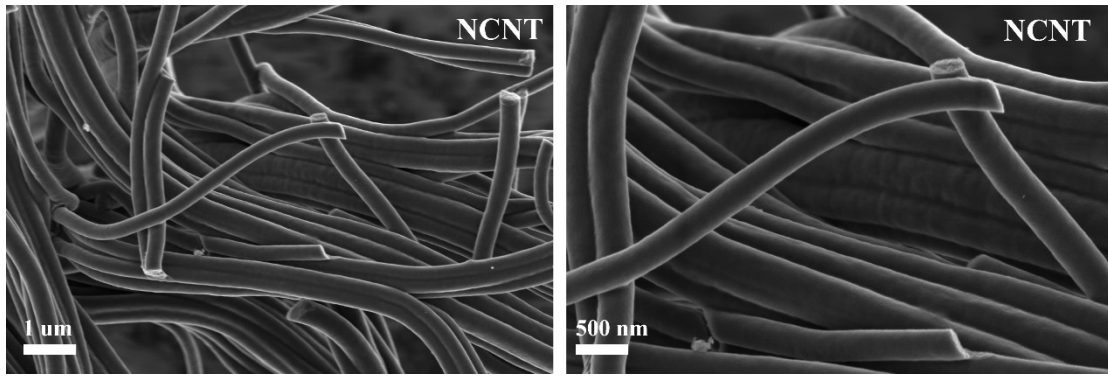


Figure S3. SEM image of the NCNTs.

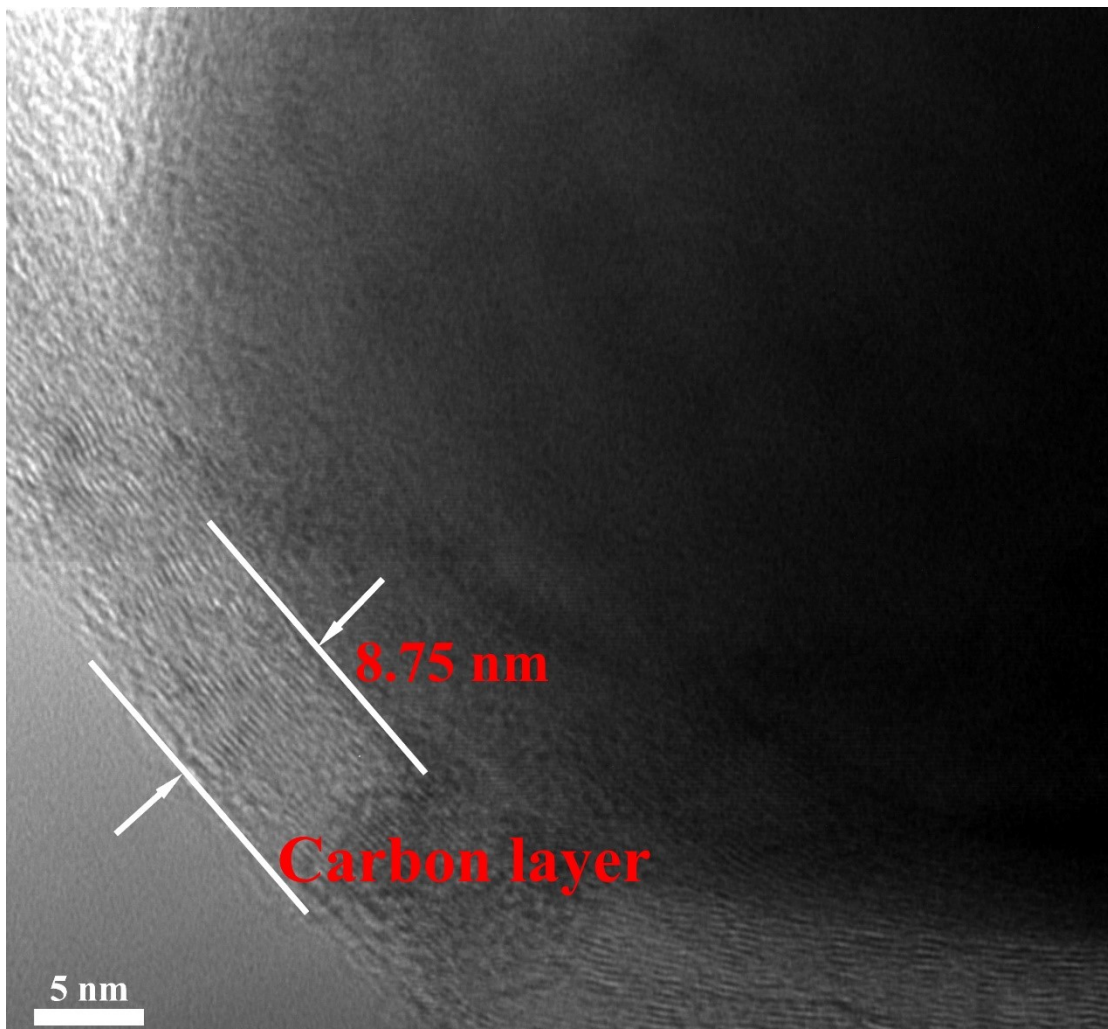


Figure S4. HAADF-STEM image of the Mo₂C/NCNTs@Ni.

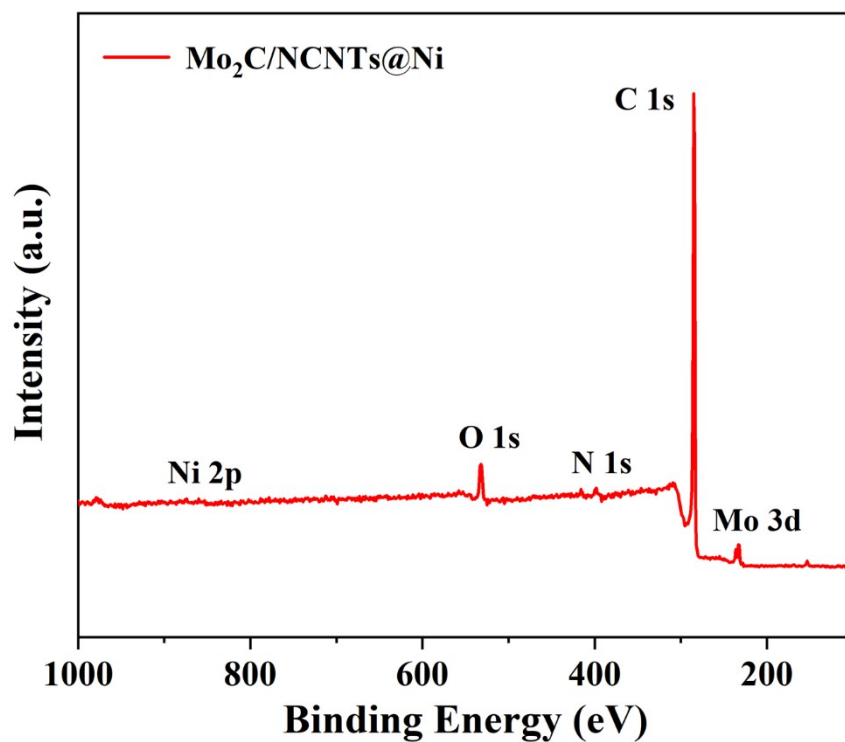


Figure S5. XPS full scan of the Mo₂C/NCNTs@Ni.

Table S1. Calculated R_{ct} values of the samples based on equivalent circuit models.

sample	R_{ct} (Ω)
Mo ₂ C/NCNTs@Ni	5.31
Mo ₂ C/NCNTs	8.20
NCNTs@Ni	6.98

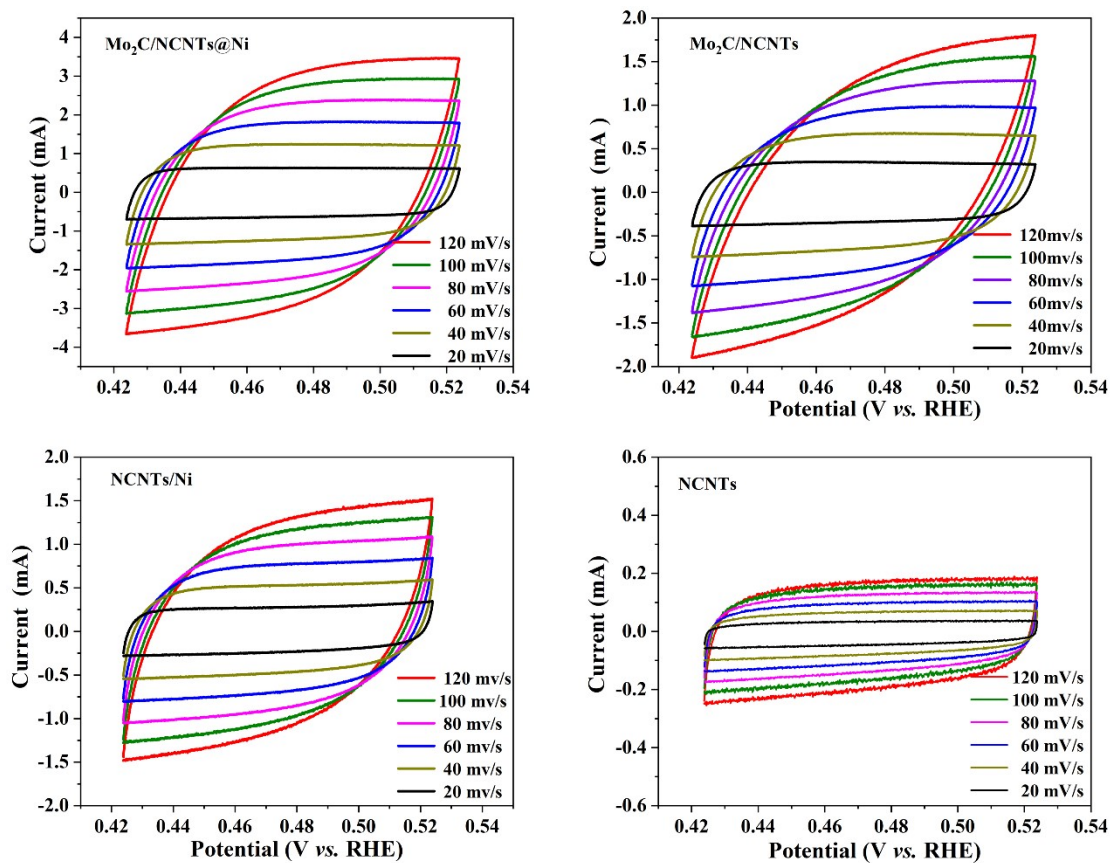


Figure S6. Cyclic voltammograms at different scan rates. (a) $\text{Mo}_2\text{C}/\text{NCNTs}@Ni$. (b) $\text{Mo}_2\text{C}/\text{NCNTs}$. (c) $\text{NCNTs}@Ni$. (d) NCNTs .

Table S2. Calculated R_{ct} values of the samples based on equivalent circuit models.

sample	R_{ct} (Ω)
$\text{Mo}_2\text{C}/\text{NCNTs}@Ni$	4.73
$\text{Mo}_2\text{C}/\text{NCNTs}$	6.49
$\text{NCNTs}@Ni$	6.43

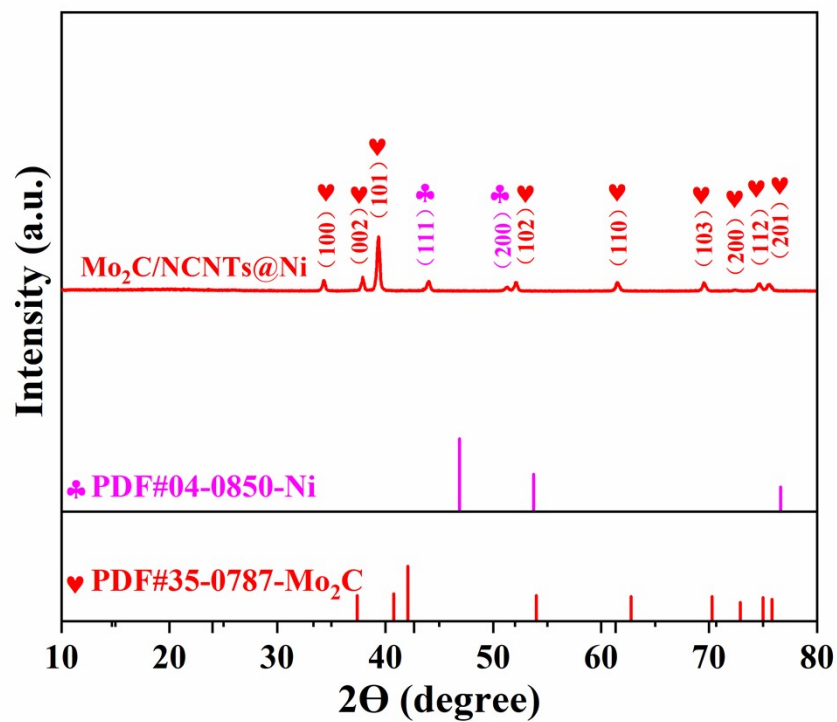


Figure S7. XRD patterns of the $\text{Mo}_2\text{C}/\text{NCNTs}@Ni$ after the HER durability test.

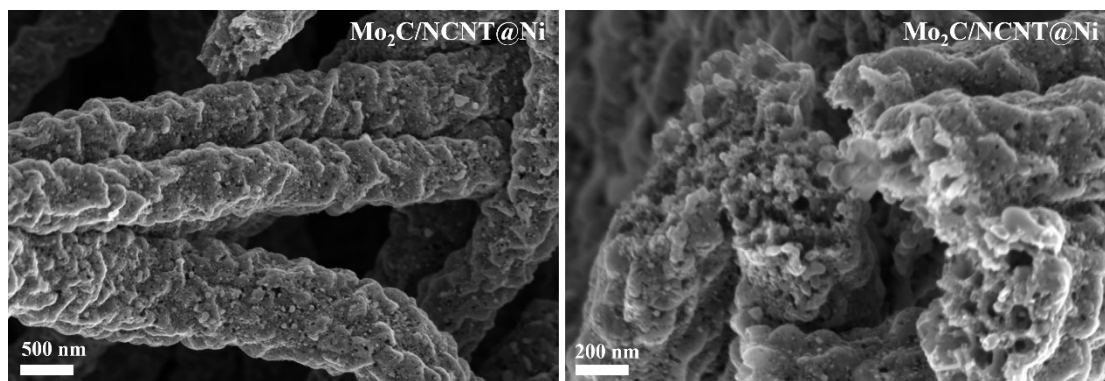


Figure S8. SEM image of the $\text{Mo}_2\text{C}/\text{NCNTs}@Ni$ after the HER durability test.

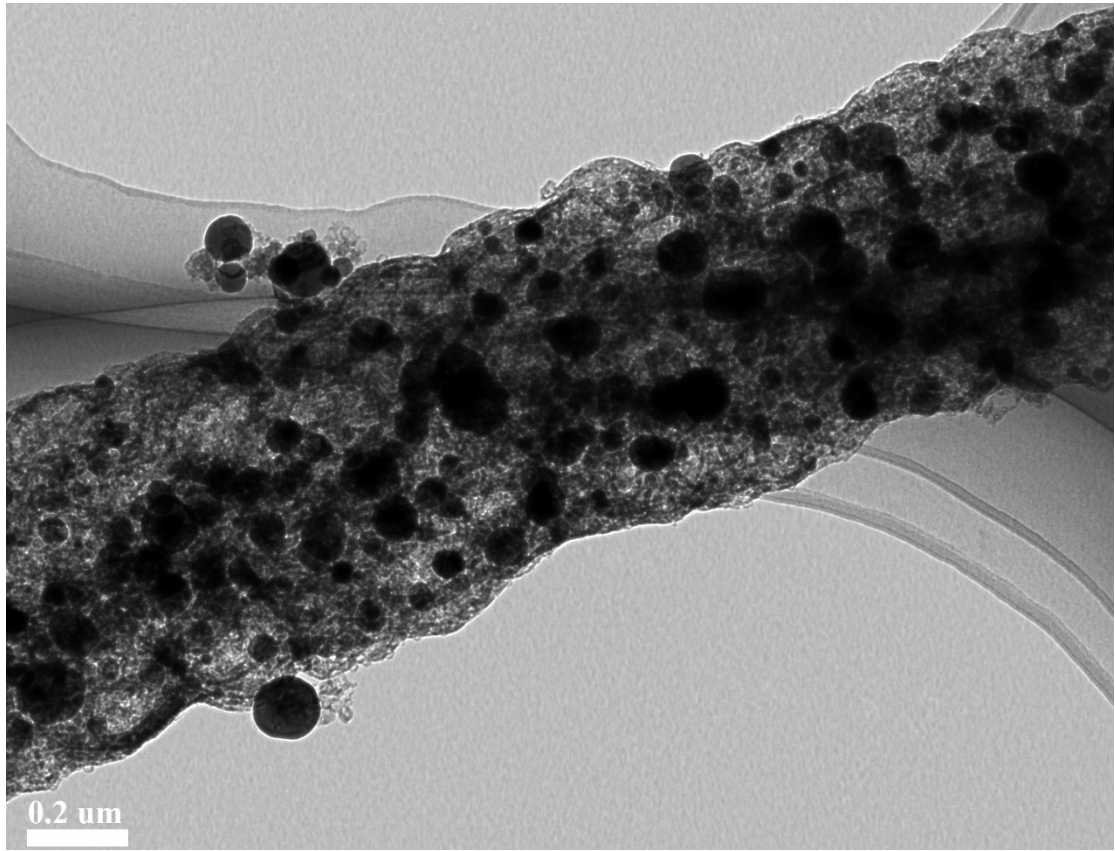


Figure S9. SEM image of the Mo₂C/NCNTs@Ni after the HER durability test.

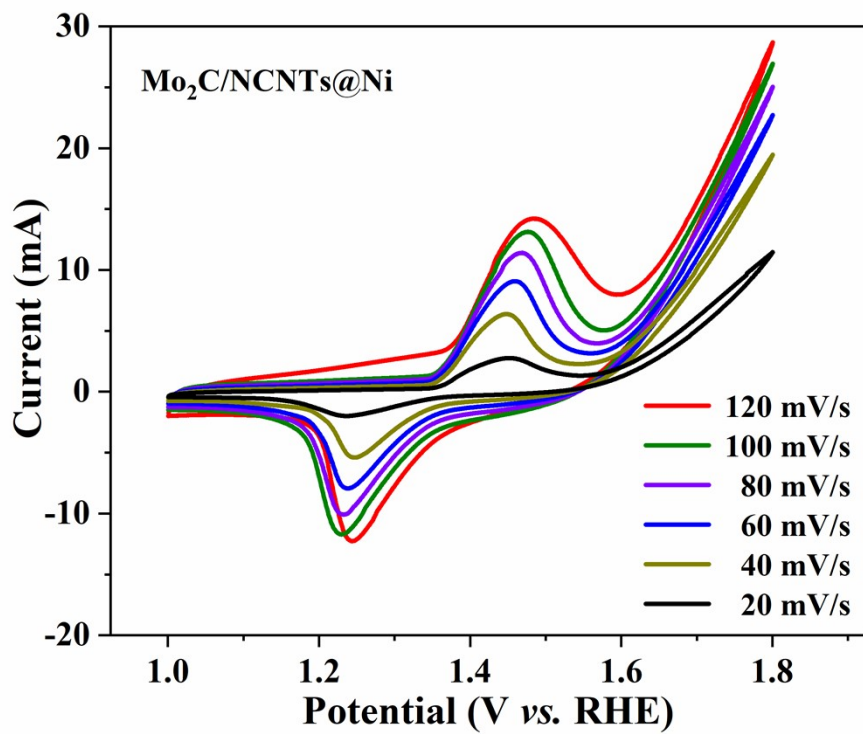


Figure S10. CVs for Mo₂C/NCNTs@Ni in the faradic capacitance current range at scan rates from 20 to 120 mV s⁻¹.

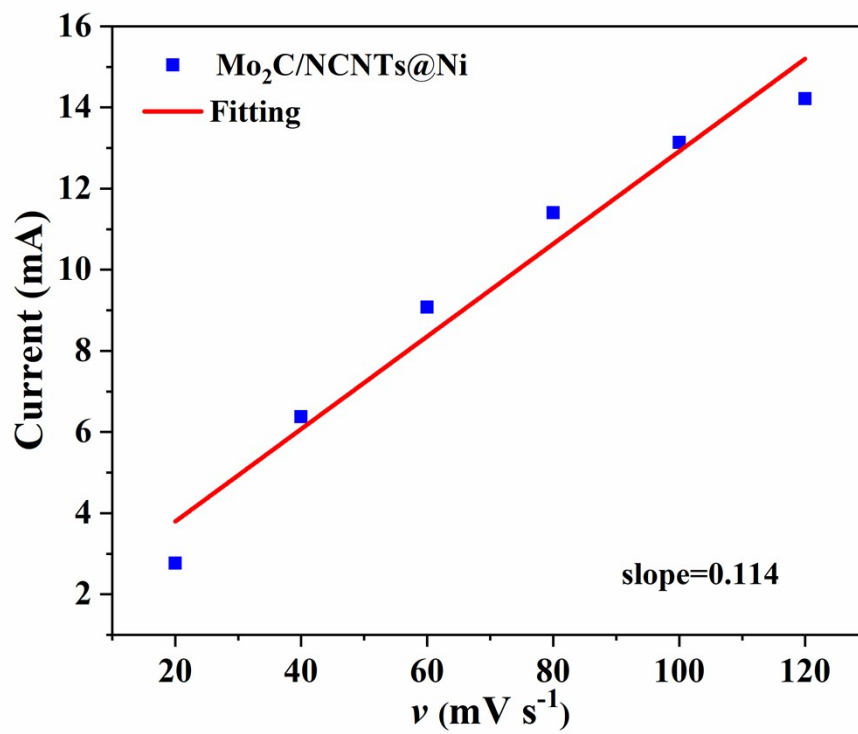


Figure S11. The corresponding plot of oxidation peak current versus the scan rate from CV test.

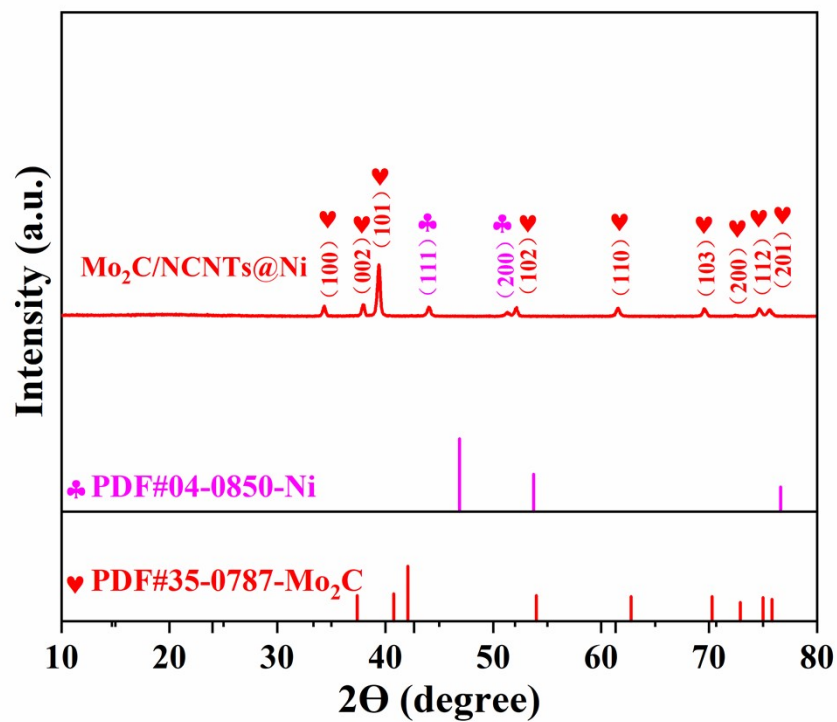


Figure S12. XRD patterns of the $\text{Mo}_2\text{C}/\text{NCNTs@Ni}$ after the OER durability test.

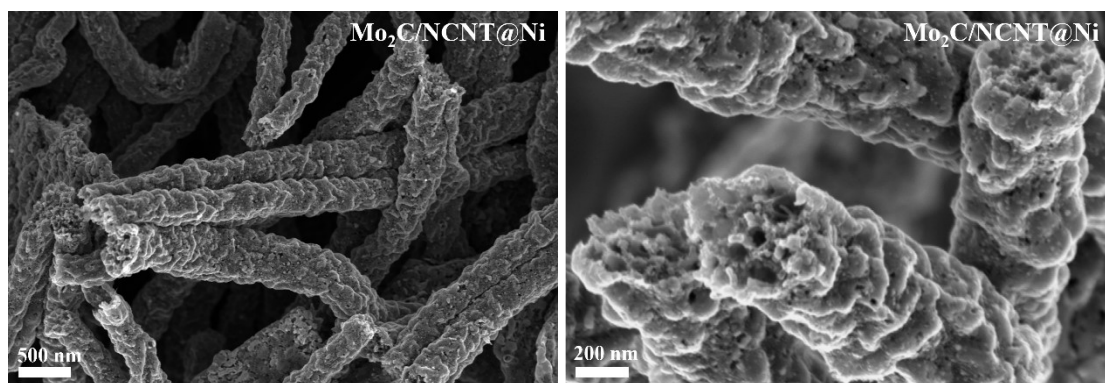


Figure S13. SEM image of the $\text{Mo}_2\text{C}/\text{NCNTs@Ni}$ after the OER durability test.

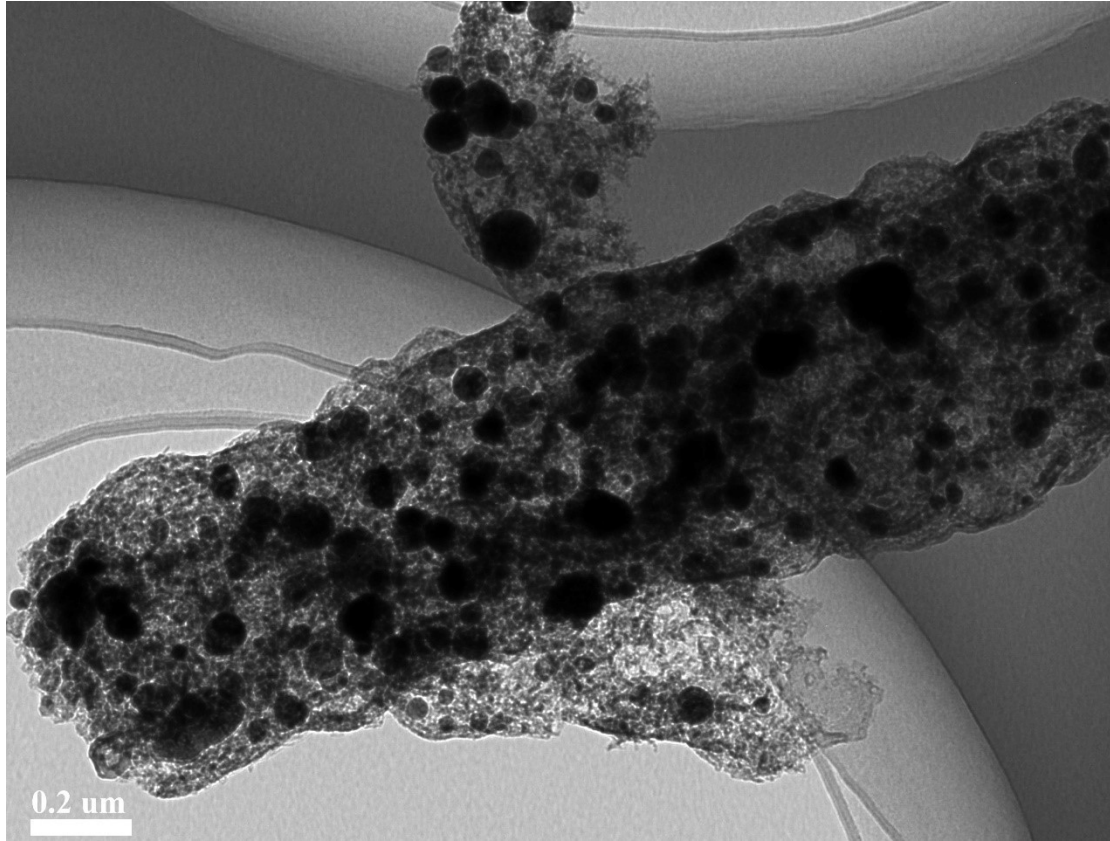


Figure S14. SEM image of the Mo₂C/NCNTs@Ni after the OER durability test.

Table S3. Activity comparison of AEM water electrolysis.

Catalyst	Temperature	Activity	Ref.
Mo ₂ C/NCNTs@Ni Mo ₂ C/NCNTs@Ni	room temperature	82.5 mA cm ⁻² at 1.99 V	This work
Co-Mo ₂ C@NC Co- Mo ₂ C@NC	room temperature	10 mA cm ⁻² at 1.83 V	1
Nickel nanopowders Li _{0.21} Co _{2.79} O ₄	45°C	300 mA cm ⁻² at 2.05 V	2
Ni/CP Ni/CP	50°C	150 mA cm ⁻² at 1.9 V	3
Co ₃ O ₄ eCuO Ni	40°C	92.11 mA cm ⁻² at 2 V	4
CuCoO _x Ni/(CeO ₂ - La ₂ O ₃)/C	60°C	74 mA cm ⁻² at 2 V	5
Cu _{0.7} Co _{2.3} O ₄ Pt	25 °C	73.33 mA cm ⁻² at 2 V	6
Cu _x Mg _{0.9-x} Co _{2.1} O ₄ Pt	40°C	66.67 mA cm ⁻² at 2 V	7
SG-LSFN-0.5 Pt/C	room temperature	100 mA cm ⁻² at 1.89 V	8

References:

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