

Supporting Information

**Fluorescent Styryl Pyridine-N-Oxide Probes for Imaging Lipid Droplets**

Yogesh Dubey<sup>a</sup>, Paramasivam Mahalingavelar<sup>b</sup>, Deeksha Rajput<sup>a</sup>, Dipeshwari J. Shewale<sup>c</sup>,  
Virupakshi Soppina\* and Sriram Kanvah\*

<sup>a</sup>Department of Chemistry, Indian Institute of Technology Gandhinagar, Palaj, Gandhinagar, Gujarat – 382055: email: [sriram@iitgn.ac.in](mailto:sriram@iitgn.ac.in)

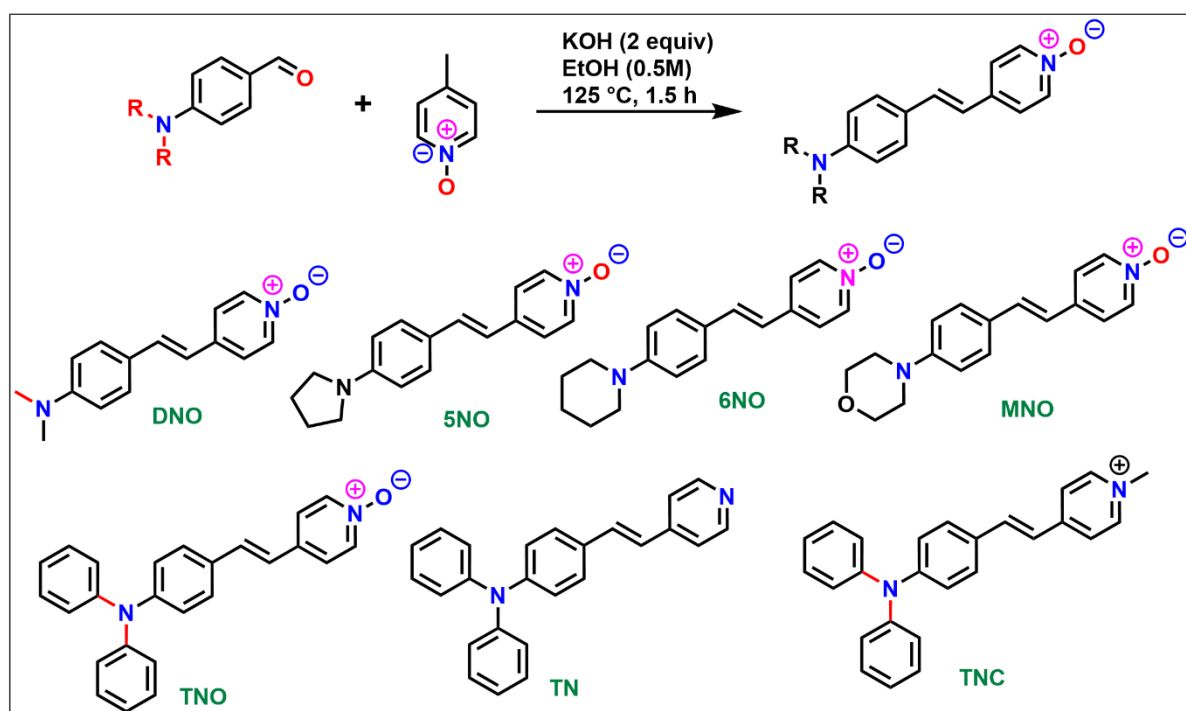
<sup>b</sup>School of Chemistry and Biochemistry, Georgia Institute of Technology Atlanta Georgia 30332

<sup>c</sup>Department of Biological Engineering, Indian Institute of Technology Gandhinagar, Palaj, Gandhinagar, Gujarat -382055 Email: [vsoppina@iitgn.ac.in](mailto:vsoppina@iitgn.ac.in);

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## S1. Synthetic procedure



### General procedure for the synthesis of DNO, 5NO and MNO:

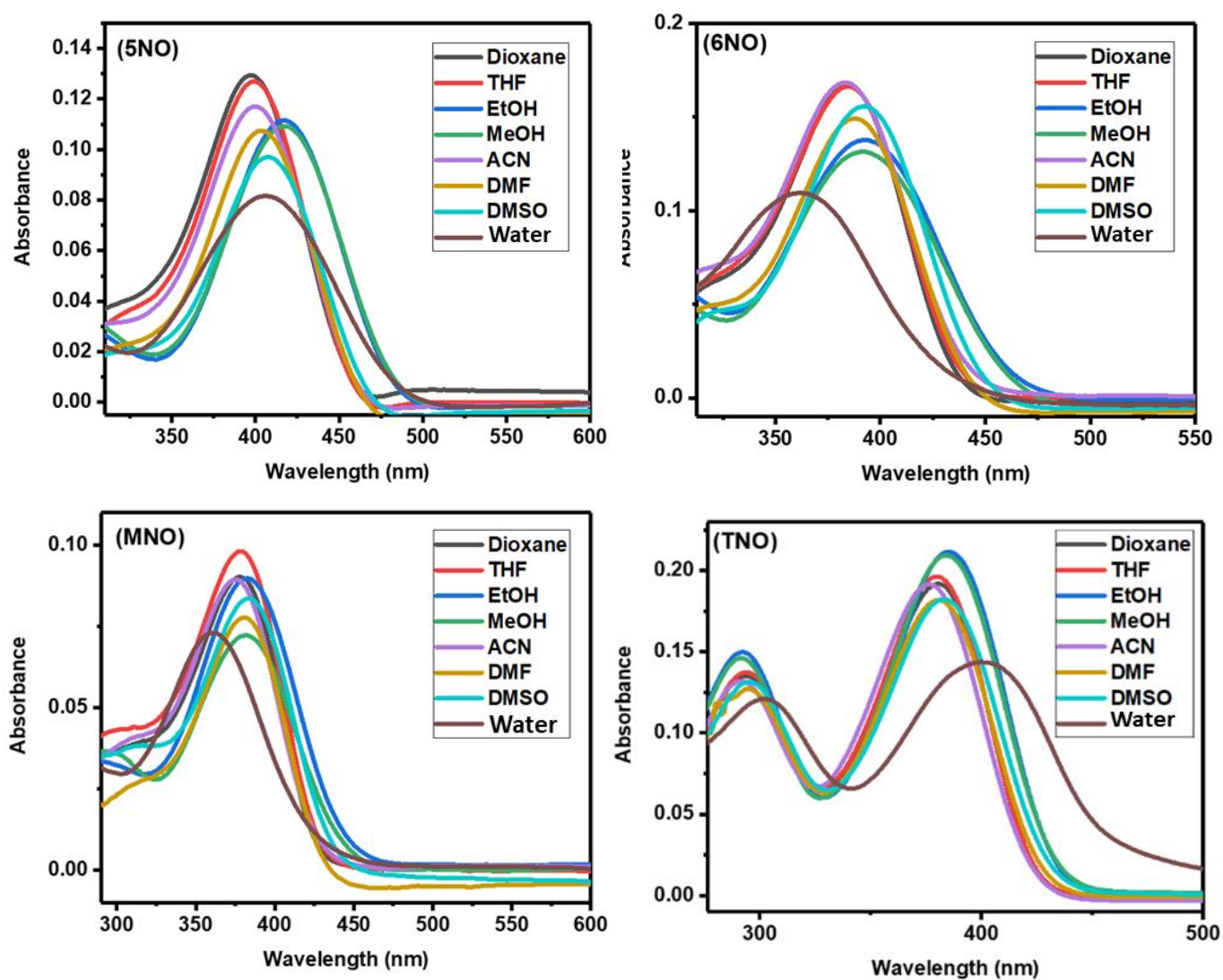
In a dry round bottom flask equipped with a magnetic bead, *p*-amino benzaldehyde (1 equiv), 4-methylpyridine-N-oxide (1 equiv.) and potassium hydroxide (2 equiv) were added. The reaction was purged with argon gas and added ethanol (0.5M). The reaction was heated at 125° C for 2 hours, and the progress of the reaction was monitored by TLC. The reaction mixture was filtered, and the insoluble solid was collected, washed multiple times with diethyl ether, dried and weighed.

### General procedure for the synthesis of TNO and 6NO:

In a dry round bottom flask equipped with a magnetic bead, *p*-amino benzaldehyde (1 equiv), 4-methylpyridine-N-oxide (1 equiv), and potassium hydroxide (2 equiv) were added. The reaction was purged with argon gas and added ethanol (0.5M). The reaction was heated at 125° C for 2 hours and the progress of the reaction was monitored by TLC. The reaction was concentrated on a rotary evaporator and purified by column chromatography on 100-200 mesh size silica gel with gradient elution from 1 to 5 % methanol in DCM.

Synthesis and optical properties of TN and TNC were referred from literature<sup>1,2</sup>.

**Figure S2.** Absorption spectral data for 5NO, 6NO, MNO and TNO



**Figure S3.** Emission spectral data of compounds 5NO, 6NO, and MNO in different solvents.

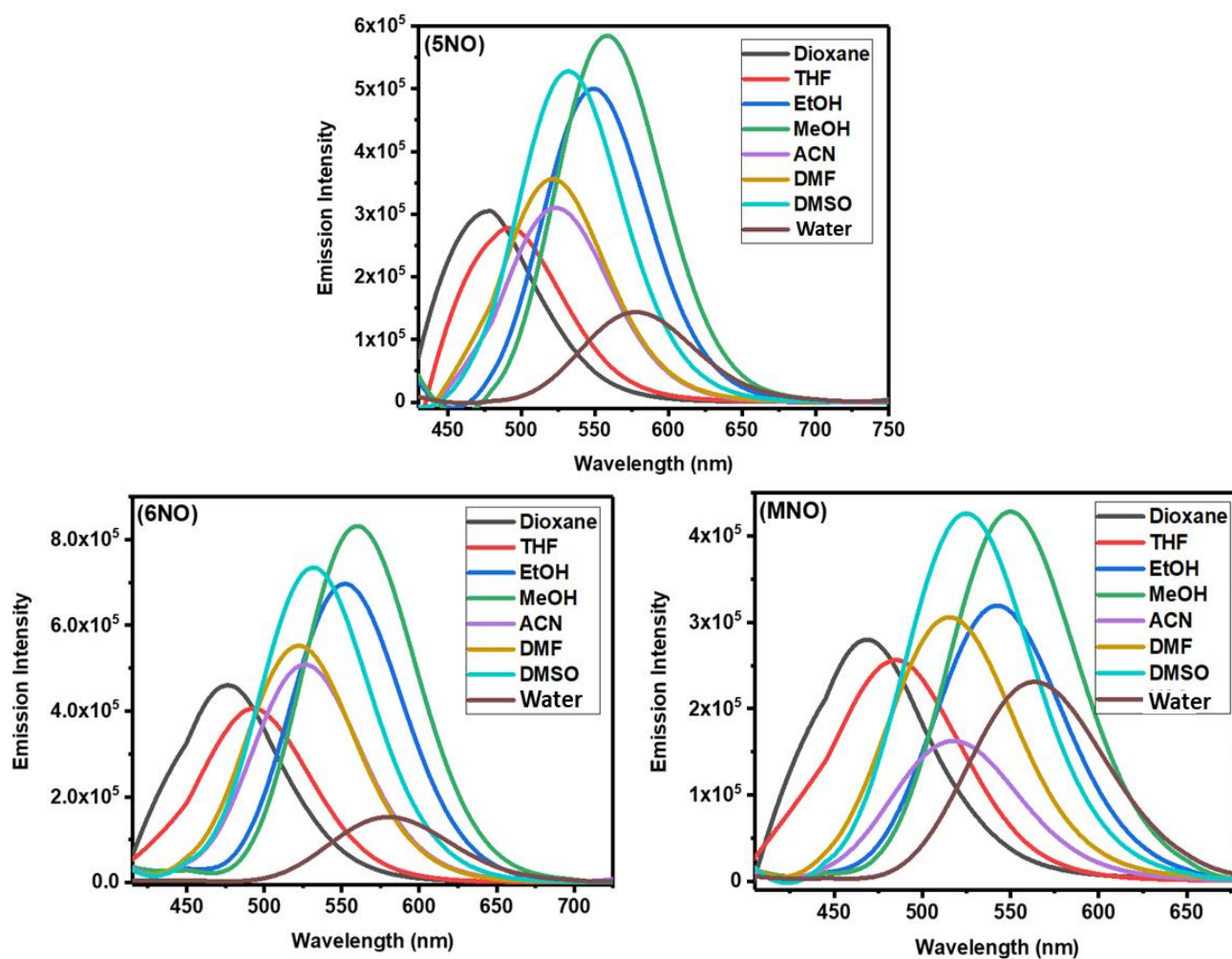
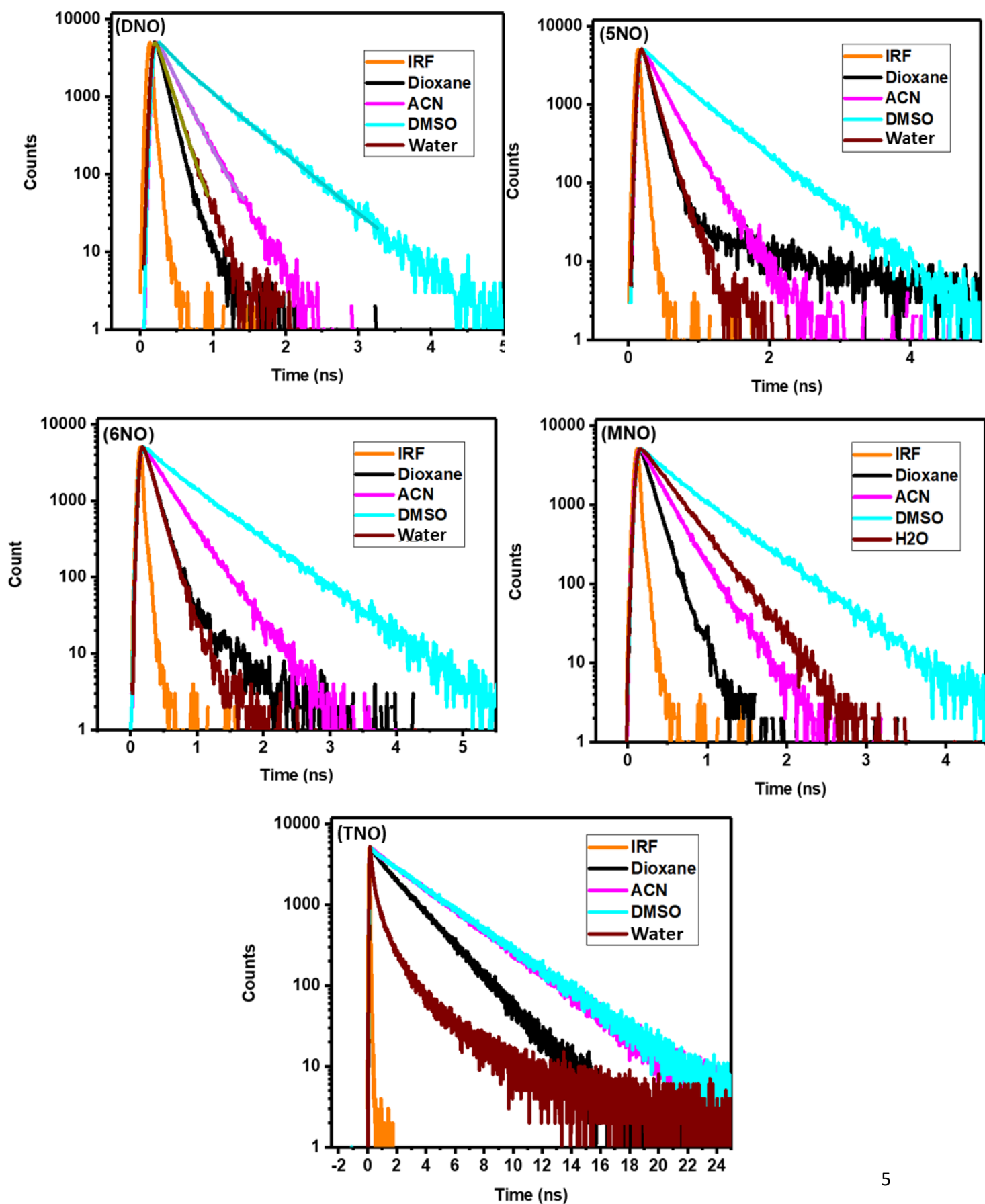
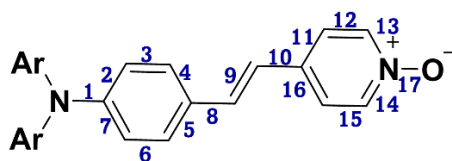


Figure S4. Fluorescence lifetime decay of the compounds



**Table S1:** Lifetime properties of compounds in various solvents

Compound	Solvents	$\tau_1$ (ns)	A1 (%)	$\tau_2$ (ns)	A2 (%)	$\tau_3$ (ns)	A3 (%)	Average lifetime ( $\tau$ ) (ns)	$\chi^2$
<b>DNO</b>	<b>Dioxane</b>	0.10	100	-	-	-	-	0.10	0.99
	<b>ACN</b>	0.23	100	-	-	-	-	0.23	1.18
	<b>DMSO</b>	0.13	9.36	0.55	90.64	-	-	0.43	1.10
	<b>water</b>	0.14	100	-	-	-	-	0.14	1.09
<b>5NO</b>	<b>Dioxane</b>	0.11	100	-	-	-	-	0.11	1.06
	<b>ACN</b>	0.23	100	-	-	-	-	0.23	1.19
	<b>DMSO</b>	0.27	13.76	0.62	86.24	-	-	0.52	1.05
	<b>water</b>	0.13	100	-	-	-	-	0.13	1.02
<b>6NO</b>	<b>Dioxane</b>	0.13	100	-	-	-	-	0.13	1.16
	<b>ACN</b>	0.30	100	-	-	-	-	0.30	1.23
	<b>DMSO</b>	0.15	6.52	0.67	93.48	-	-	0.55	0.99
	<b>water</b>	0.13	100	-	-	-	-	0.13	1.09
<b>MNO</b>	<b>Dioxane</b>	0.12	100	-	-	-	-	0.12	1.23
	<b>ACN</b>	0.23	100	-	-	-	-	0.23	1.31
	<b>DMSO</b>	0.17	9.08	0.58	90.92	-	-	0.48	1.17
	<b>water</b>	0.31	100	-	-	-	-	0.31	1.20
<b>TNO</b>	<b>Dioxane</b>	0.43	3.36	2.20	96.64	-	-	1.94	1.06
	<b>ACN</b>	3.30	100	-	-	-	-	3.30	1.13
	<b>DMSO</b>	3.40	100	-	-	-	-	3.40	1.13
	<b>water</b>	0.55	33.65	1.80	30.70	0.13	33.65	0.30	1.08

**Table S2** Geometrical coordinates of the N-oxide derivatives

	5NO	6NO	DNO	PNO	TNO
<b>N1-C2</b>	1.365	1.438	1.374	1.411	1.397
<b>C2-C3</b>	1.420	1.400	1.416	1.404	1.414
<b>C3-C4</b>	1.382	1.390	1.385	1.386	1.387
<b>C4-C5</b>	1.410	1.407	1.407	1.406	1.407
<b>C5-C6</b>	1.408	1.407	1.409	1.408	1.409
<b>C6-C7</b>	1.384	1.388	1.382	1.384	1.384
<b>C7-C2</b>	1.416	1.400	1.420	1.412	1.408
<b>C5-C8</b>	1.454	1.463	1.455	1.457	1.458
<b>C8-C9</b>	1.352	1.348	1.352	1.351	1.350
<b>C9-C10</b>	1.454	1.457	1.455	1.456	1.456
<b>C10-C11</b>	1.408	1.408	1.410	1.407	1.409
<b>C11-C12</b>	1.377	1.374	1.375	1.378	1.374
<b>C12-N13</b>	1.367	1.372	1.371	1.367	1.371
<b>N13-C14</b>	1.371	1.367	1.367	1.37	1.367
<b>C14-C15</b>	1.374	1.378	1.377	1.374	1.377
<b>C15-C16</b>	1.410	1.406	1.407	1.409	1.406
<b>N14-O17</b>	1.288	1.284	1.287	1.286	1.285

**Table S3** TDDFT results of the TN, TNC and TNO derivatives provided with computed excitation energy, oscillator strength, major transitions involved in terms of frontier molecular Orbitals, and dipole moments.

	$\lambda_{\omega X97BD}^a$ (nm)	$f_{\omega X97BD}^a$	Major transitions	$\mu_g^b$ (Debye)	$\mu_e^c$ (Debye)
<b>TN</b>	361.3	1.34	HOMO->LUMO (86%), H-1->LUMO (11%)	5.76	5.26
<b>TNC</b>	430.2	1.23	HOMO(A)->LUMO(A) (24%), HOMO(B)->LUMO(B) (55%) H-1(B)->LUMO(B) (9%),	8.48	8.53
<b>TNO</b>	375.2	1.89	HOMO->LUMO (91%), H-1->L+4 (2%), HOMO->L+4 (3%)	7.8	7.8

<sup>a</sup>TDDFT simulated absorption maximum and oscillator strength using  $\omega X97BD/6-31G^{**}/C-PCM(THF)$  level of theory; <sup>b</sup> $\mu_g$ : ground state dipole moment; <sup>c</sup> $\mu_e$ : transient dipole moment.

**Table S4** Comparison of absorption properties of the molecules with computed excitation energy, oscillator strength, major transitions involved in terms of frontier molecular Orbitals, and dipole moments.

	$\lambda_{exp}^a$ (nm)	$\lambda_{CAM-B3LYP}^b$ (nm)	$f_{CAM-B3LYP}^b$	Major transitions	$\mu_g^c$ (Debye)	$\mu_e^d$ (Debye)
<b>5NO</b>	403	361.8	1.91	HOMO->LUMO (93%)	12.0	11.6
<b>6NO</b>	383	322.5	1.78	HOMO->LUMO (91%), H- 2->LUMO (2%)	6.5	6.7
<b>DNO</b>	395	357.5	1.1.85	HOMO->LUMO (93%)	11.3	10.9
<b>MNO</b>	377	349.4	1.88	HOMO->LUMO (93%)	8.1	7.8
<b>TNO</b>	380	365.3	1.89	HOMO->LUMO (89%), H-1->L+4 (2%), HOMO->L+4 (4%)	7.8	7.7

<sup>a</sup>Absorption spectra measured in THF in the concentration of  $1 \times 10^{-5}$  M at ambient temperature; <sup>b</sup>TDDFT simulated absorption maximum and oscillator strength using **B3LYP/6-31G<sup>\*\*</sup>/C-PCM(THF)** level of theory; <sup>c</sup> $\mu_g$ : ground state dipole moment; <sup>d</sup> $\mu_e$ : transient dipole moment.

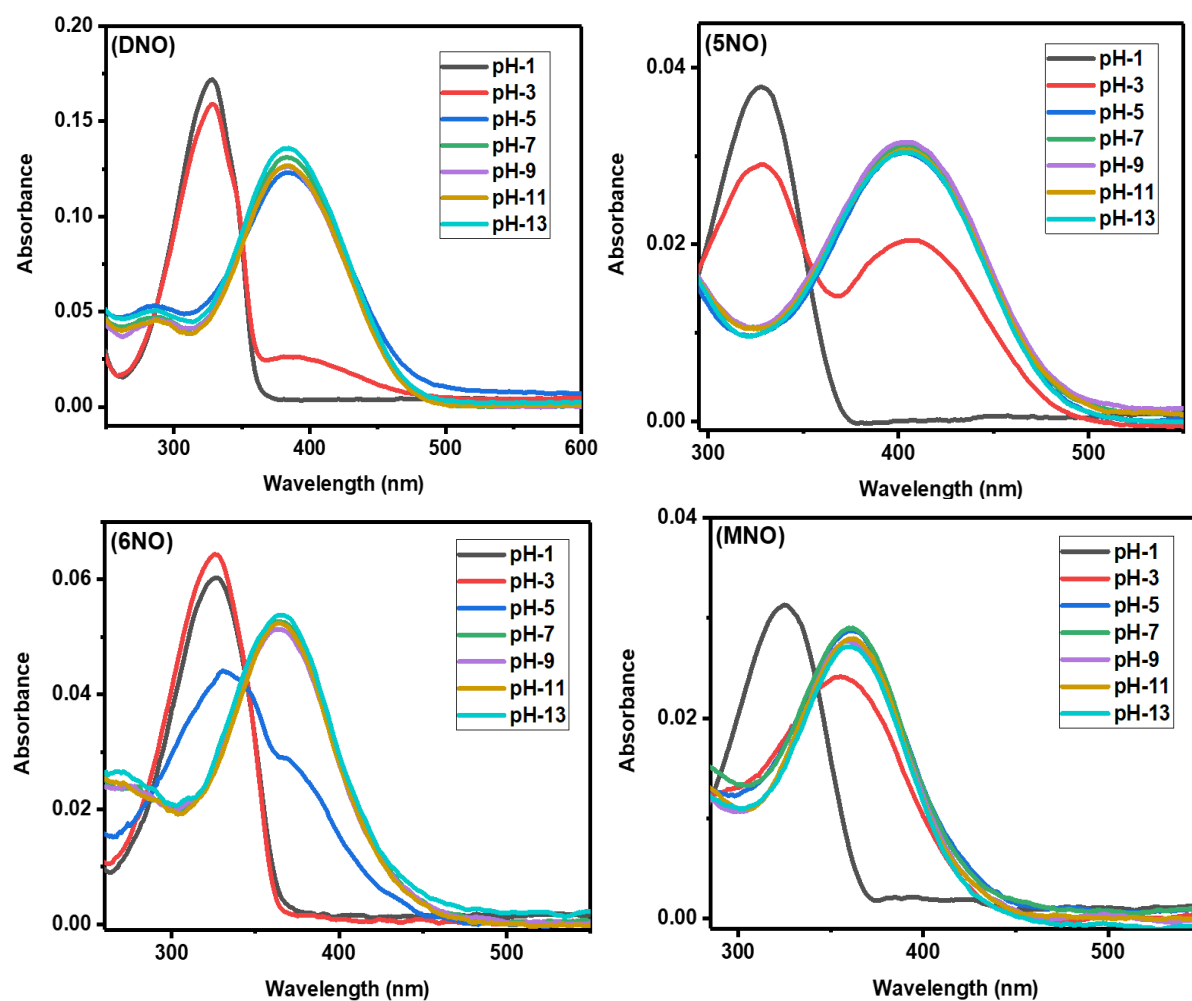


**Table S5** Comparison of absorption properties of the molecules with computed excitation energy, oscillator strength, major transitions involved in terms of frontier molecular Orbitals, and dipole moments.

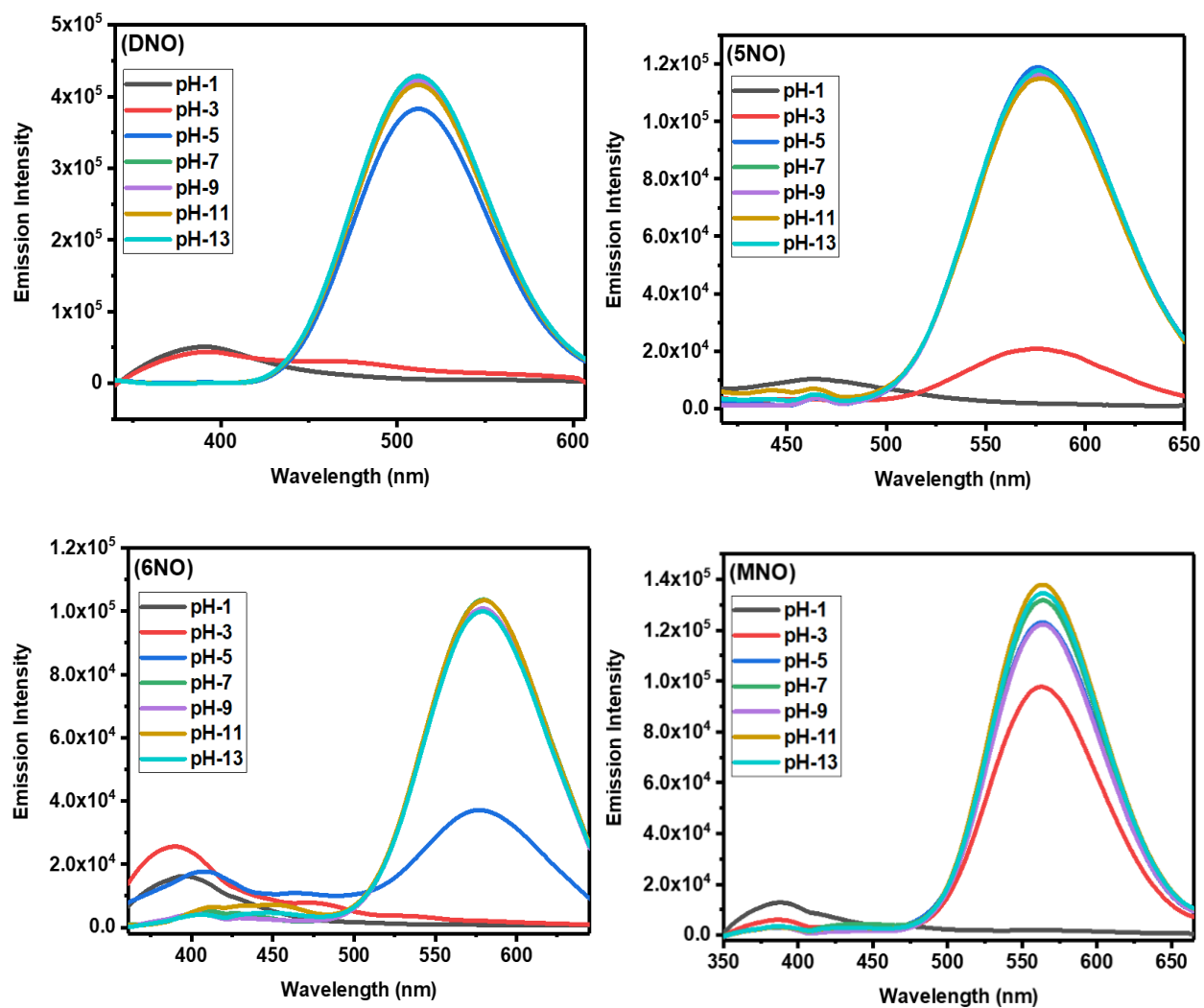
	$\lambda_{\text{exp}}^{\text{a}}$ (nm)	$\lambda_{\text{CAM-B3LYP}}^{\text{b}}$ (nm)	$f_{\text{CAM-B3LYP}}^{\text{b}}$	Major transitions	$\mu_{\text{g}}^{\text{c}}$ (Debye)	$\mu_{\text{e}}^{\text{d}}$ (Debye)
<b>5NO</b>	403	361.8	1.91	HOMO->LUMO (93%)	12.0	11.6
<b>6NO</b>	383	322.5	1.78	HOMO->LUMO (91%), H-2->LUMO (2%)	6.5	6.7
<b>DNO</b>	395	357.5	1.1.85	HOMO->LUMO (93%)	11.3	10.9
<b>MNO</b>	377	349.4	1.88	HOMO->LUMO (93%)	8.1	7.8
<b>TNO</b>	380	365.3	1.89	HOMO->LUMO (89%), H-1->L+4 (2%), HOMO->L+4 (4%)	7.8	7.7

<sup>a</sup>Absorption spectra measured in THF in the concentration of  $1 \times 10^{-5}$  M at ambient temperature; <sup>b</sup>TDDFT simulated absorption maximum and oscillator strength using **CAM-B3LYP/6-31G\*\*/C-PCM(THF)** level of theory; <sup>c</sup> $\mu_{\text{g}}$ : ground state dipole moment; <sup>d</sup> $\mu_{\text{e}}$ : transient dipole moment. Both CAM-B3LYP and B3LYP gave identical results.

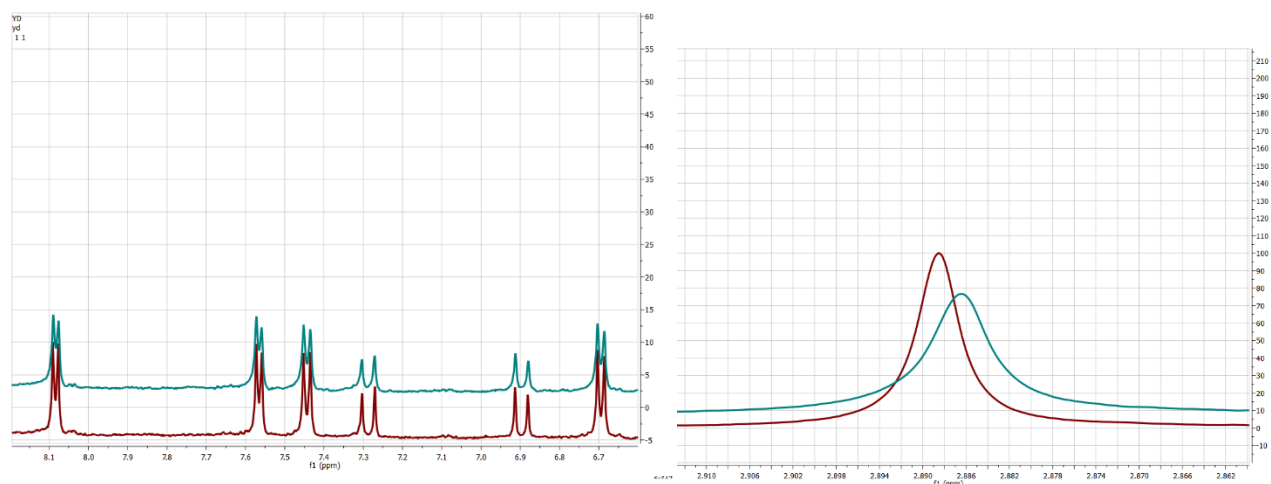
**Figure. S5.** pH-dependent absorbance titration for DNO, 5NO, 6NO and MNO



**Figure S6.** pH-dependent emission spectra for DNO, 5NO, 6NO and MNO

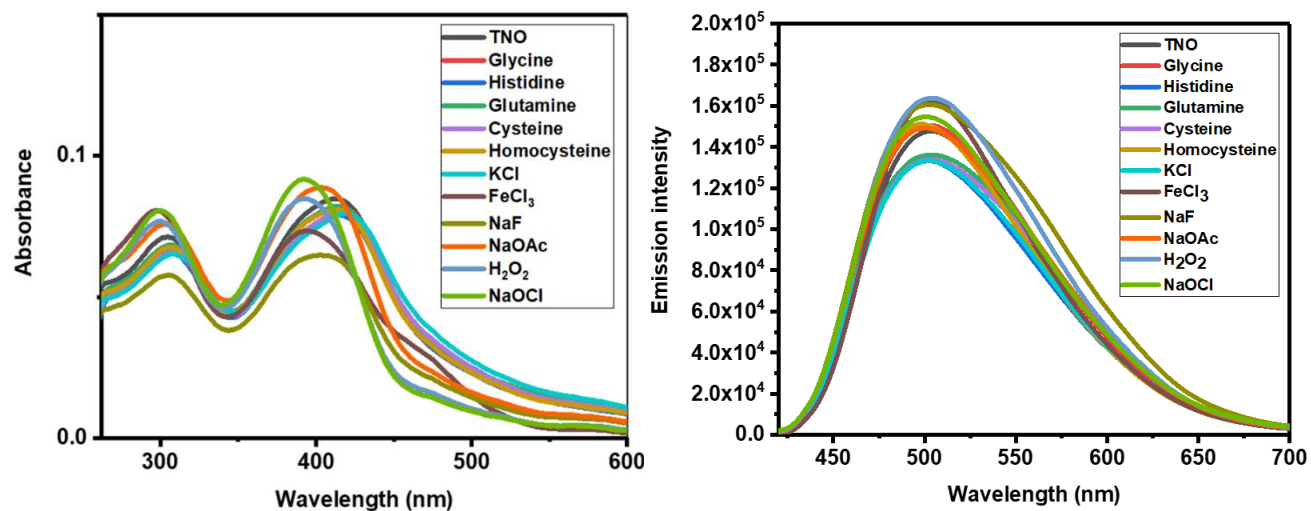


**Figure S7.** Overlay pH-dependent NMR for DNO



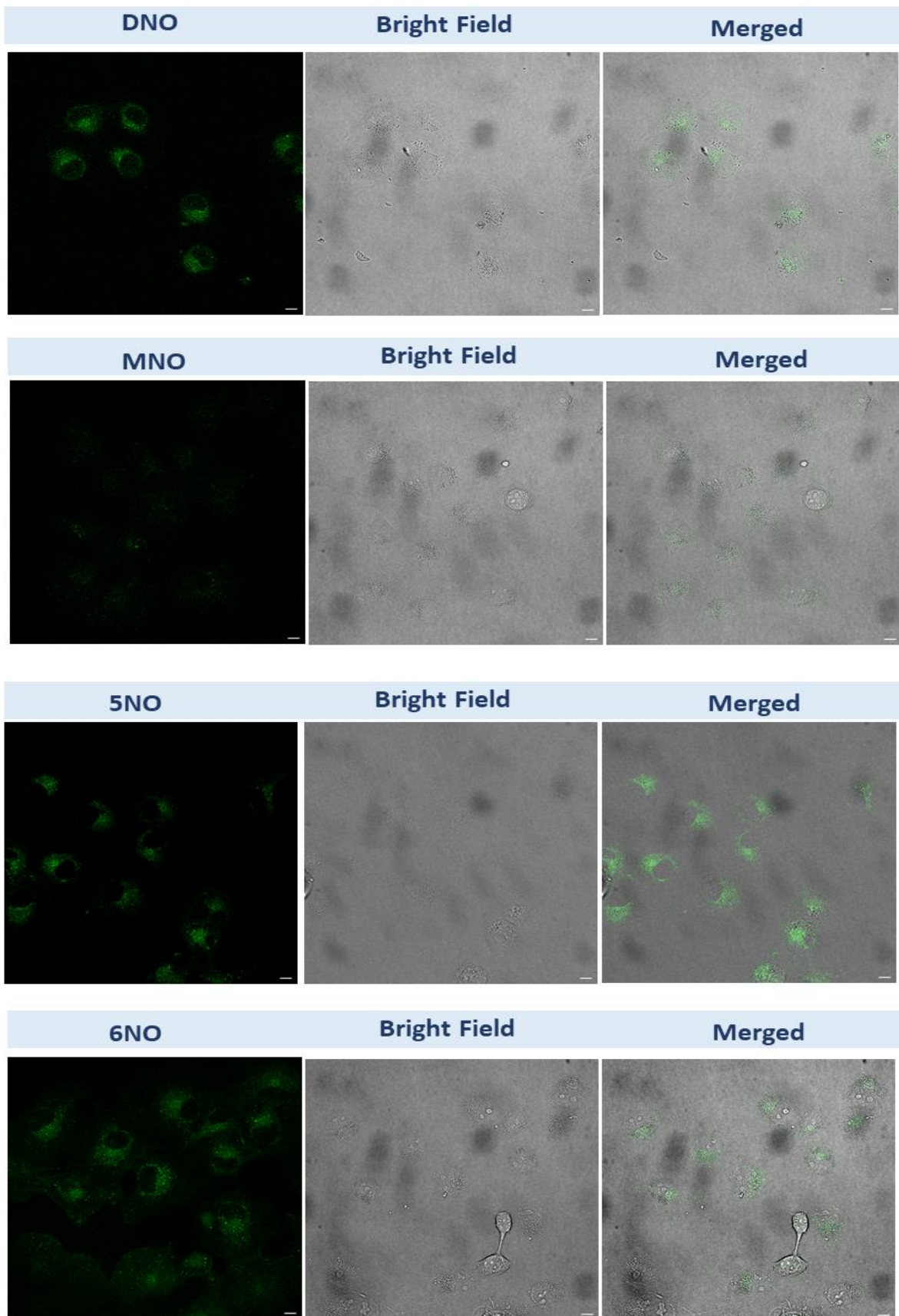
Overlay NMR spectra of DNO in pH-3 (red line) and pH-7 (green line). Calibrated axis - 2.5 ppm (DMSO-d<sub>6</sub>).

**Figure S8: Influence of added analytes on the absorption and emission of TNO**

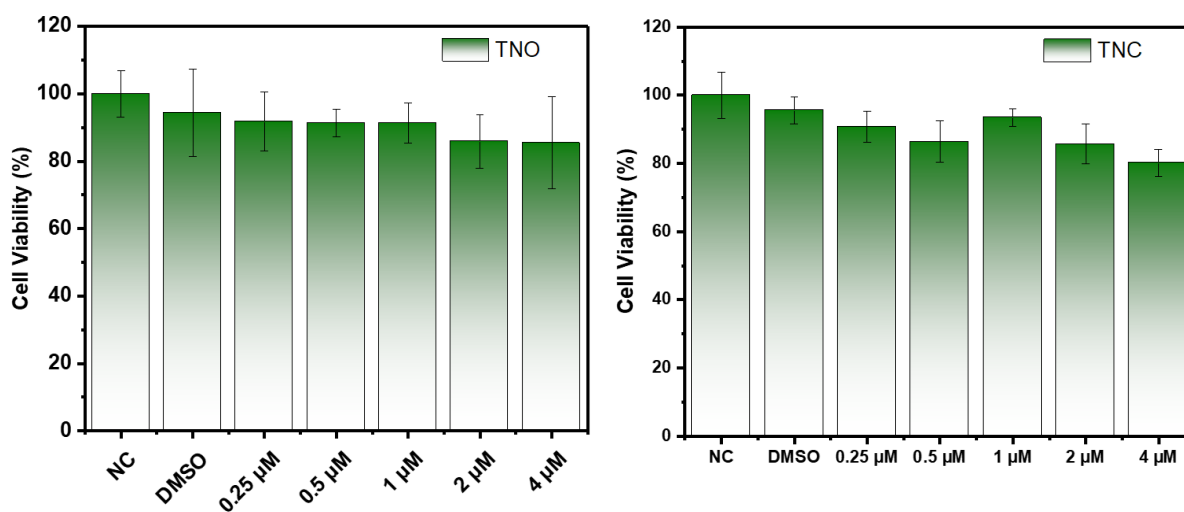


(Excitation Wavelength 380nm. Absorption and emission is recorded after 5 minutes of addition of the analyte)

**Figure S9:** Live cell imaging in COS-7 cells: DNO, 5NO, 6NO, and MNO. The data suggests weaker specificity of the molecules Scale bar 10 $\mu$ m



**Figure S10:** MTT Assay for TNO and TNC. The compounds are good for cellular imaging studies with over 80% cell viability at all concentrations. The imaging is performed at 1  $\mu\text{M}$ ,



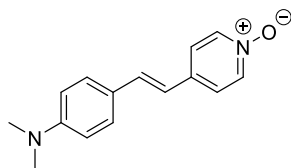
**Table S6:** Molar absorption coefficient ( $\epsilon$ ) of all compounds in different solvents

Solvent	DNO	5NO	6NO	MNO	TNO
	$\epsilon$	$\epsilon$	$\epsilon$	$\epsilon$	$\epsilon$
Dioxane	19.7	6.9	10.1	5.6	9.7
THF	19.8	6.8	10.5	7.1	9.9
EtOH	17.1	5.9	7.4	4.5	9.8
MeOH	16.4	5.7	8.2	5.0	10.0
ACN	19.2	6.2	10.5	6.3	9.9
DMF	18.9	5.7	10.2	5.9	9.5
DMSO	18.3	5.2	7.9	5.4	9.0
H <sub>2</sub> O	13.2	4.2	5.8	4.1	6.6

$\epsilon$  ( $\text{M}^{-1} \text{cm}^{-1}$ )  $\times 10^3$

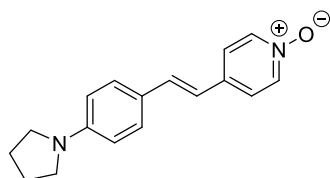
## S11. Characterisation data

### (E)-4-(4-(dimethylamino)styryl)pyridine 1-oxide (DNO)



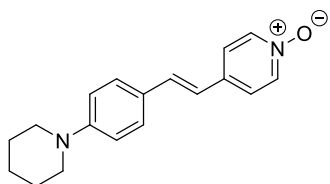
**<sup>1</sup>H NMR** (500 MHz, DMSO-d<sub>6</sub>): δ 8.11 (d, *J* = 6.5 Hz, 2H), 7.51 (d, *J* = 7 Hz, 2H), 7.44 (d, *J* = 8.5 Hz, 2H), 7.27 (d, *J* = 16.5 Hz, 1H), 6.94 (d, *J* = 16.5 Hz, 1H), 6.72 (d, *J* = 8.5 Hz, 2H), 2.95 (s, 6H). **<sup>13</sup>C NMR** (126 MHz; DMSO-d<sub>6</sub>) δ 150.92, 138.96, 135.64, 132.56, 128.57, 124.50, 123.09, 119.92, 112.58, 40.58. **HRMS** (ESI) *m/z* calcd for C<sub>15</sub>H<sub>17</sub>N<sub>2</sub>O<sup>+</sup> [M + H]<sup>+</sup> 241.1335, found 241.1320.

### (E)-4-(4-(pyrrolidin-1-yl)styryl)pyridine 1-oxide (5NO)



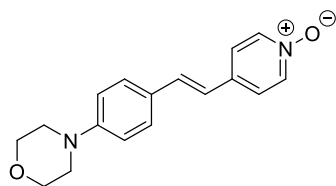
**<sup>1</sup>H NMR** (500 MHz, DMSO-d<sub>6</sub>): δ 8.10 (d, *J* = 7 Hz, 2H), 7.5 (d, *J* = 7 Hz, 2H), 7.43 (d, *J* = 8.5 Hz, 2H), 7.26 (d, *J* = 16.5 Hz, 1H), 6.91 (d, *J* = 16.5 Hz, 1H), 6.55 (d, *J* = 8.5 Hz, 2H), 3.26 (t, *J* = 6 Hz, 4H), 1.95 (t, *J* = 6.5 Hz, 4H). **<sup>13</sup>C NMR** (126 MHz; DMSO-d<sub>6</sub>) δ 148.31, 138.94, 135.82, 132.87, 128.76, 123.76, 122.98, 119.24, 112.22, 47.71, 25.43. **HRMS** (ESI) *m/z* calcd for C<sub>17</sub>H<sub>19</sub>N<sub>2</sub>O<sup>+</sup> [M + H]<sup>+</sup> 267.1492, found 267.1477.

### (E)-4-(4-(piperidin-1-yl)styryl)pyridine 1-oxide (6NO)



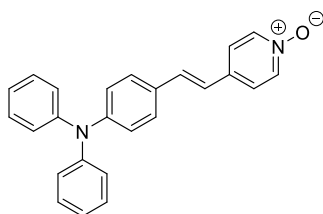
**<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>): δ 8.00 (d, *J* = 6.5 Hz, 2H), 7.27 (d, *J* = 9 Hz, 2H), 7.15 (d, *J* = 6.5 Hz, 2H), 6.95 (d, *J* = 16 Hz, 1H), 6.78 (d, *J* = 8.5 Hz, 2H), 6.63 (d, *J* = 16.5 Hz, 1H), 3.13 (t, *J* = 5 Hz, 4H), 1.60-1.49 (m, 7H). **<sup>13</sup>C NMR** (126 MHz; CDCl<sub>3</sub>) δ 151.14, 137.90, 135.75, 131.88, 127.17, 124.78, 121.34, 118.93, 114.40, 48.45, 24.50, 23.26. **HRMS** (ESI) *m/z* calcd for C<sub>18</sub>H<sub>21</sub>N<sub>2</sub>O<sup>+</sup> [M + H]<sup>+</sup> 281.1648, found 281.1634.

**(E)-4-(4-morpholinostyryl)pyridine 1-oxide (MNO)**



**<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>): δ 8.12 (d, J= 7 Hz, 2H), 7.43 (d, J= 8.5 Hz, 2H), 7.31 (d, J= 7 Hz, 2H), 7.09 (d, J= 16 Hz, 1H), 6.90 (d, J= 8.5 Hz, 2H), 6.81 (d, J= 16 Hz, 1H), 3.87 (t, J= 4.5 Hz, 4H), 3.23 (t, J= 5 Hz, 4H) **<sup>13</sup>C NMR** (126 MHz; CDCl<sub>3</sub>) δ 151.57, 139.03, 136.47, 132.61, 128.20, 127.17, 122.53, 120.83, 115.15, 66.72, 48.45 **HRMS** (ESI) m/z calcd for C<sub>17</sub>H<sub>19</sub>N<sub>2</sub>O<sub>2</sub><sup>+</sup> [M + H]<sup>+</sup> 283.1441, found 283.1427.

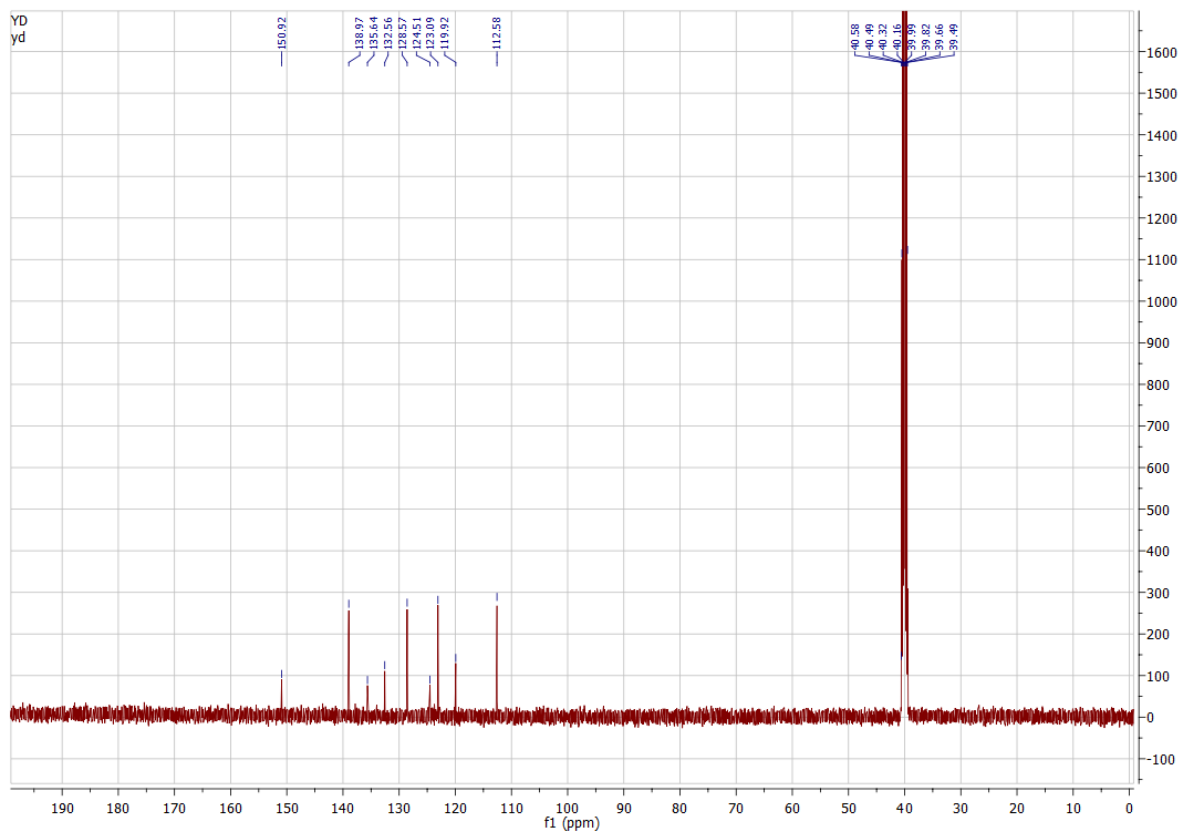
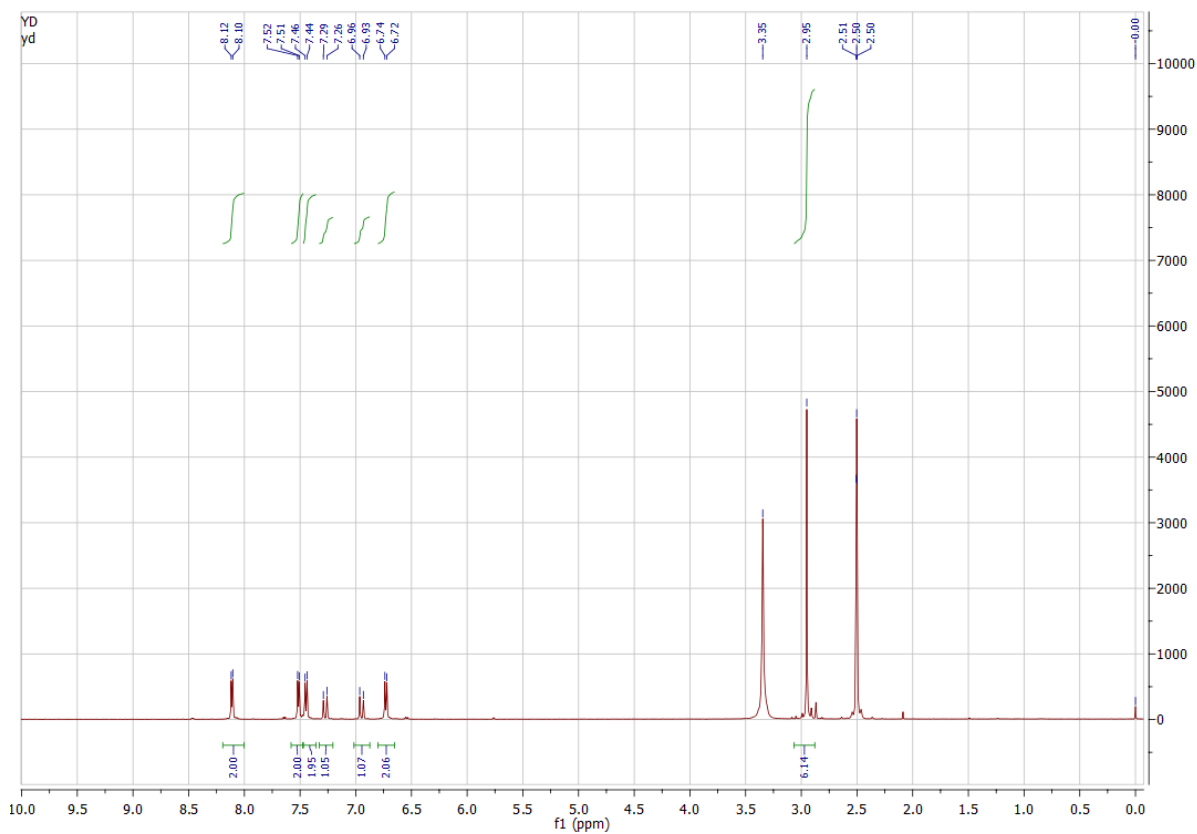
**(E)-4-(4-(diphenylamino)styryl)pyridine 1-oxide (TNO)**

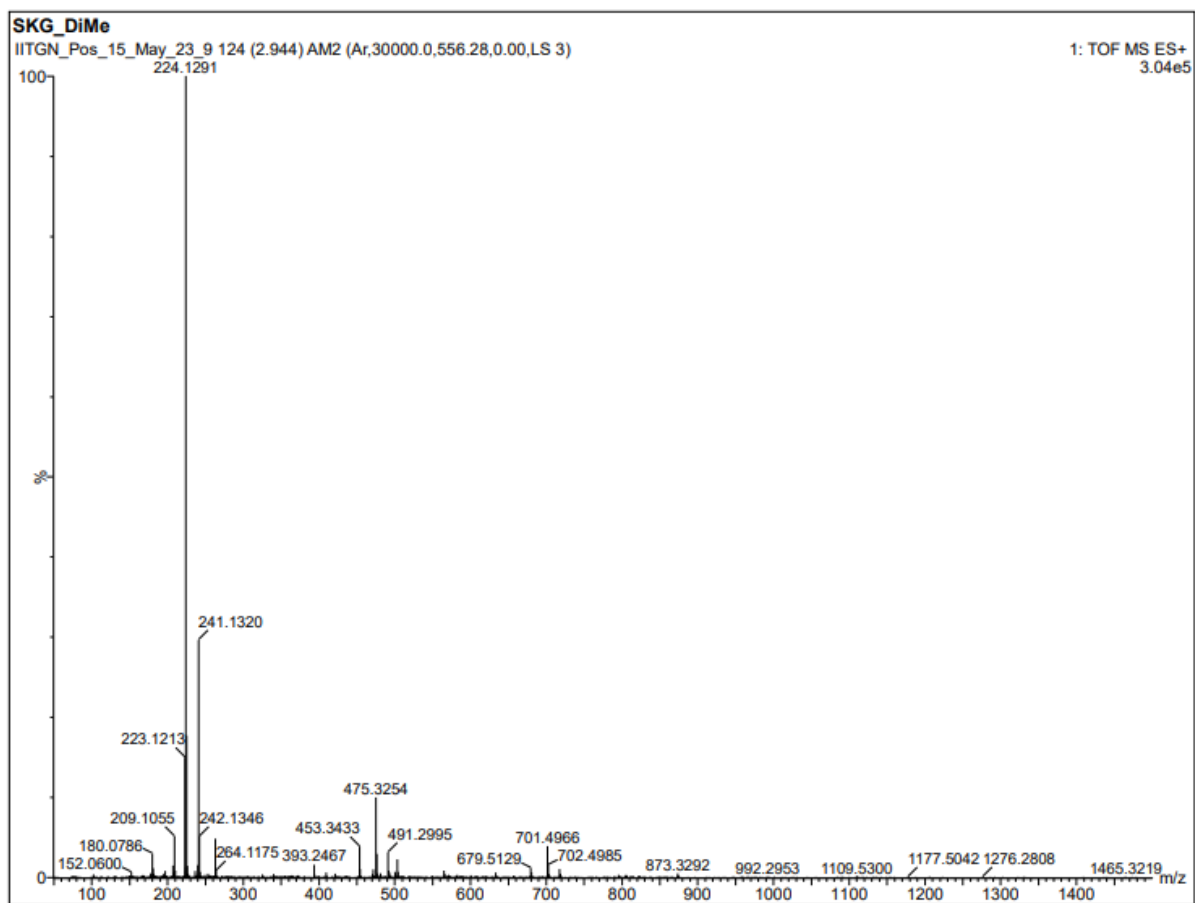


**<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>): δ 8.39 (d, J= 6 Hz, 2H), 7.24 (d, J= 8.5 Hz, 2H), 7.17-7.07 (m, 7H), 6.98 (d, J= 8 Hz, 4H), 6.93-6.90 (m, 4H), 6.72 (d, J= 16.5 Hz, 1H). **<sup>13</sup>C NMR** (126 MHz; CDCl<sub>3</sub>) δ 148.97, 147.35, 146.16, 143.89, 131.60, 128.69, 128.33, 126.91, 123.80, 122.77, 122.42, 121.71, 119.56 **HRMS** (ESI) m/z calcd for C<sub>25</sub>H<sub>21</sub>N<sub>2</sub>O<sup>+</sup> [M + H]<sup>+</sup> 365.1648, found 365.1635.

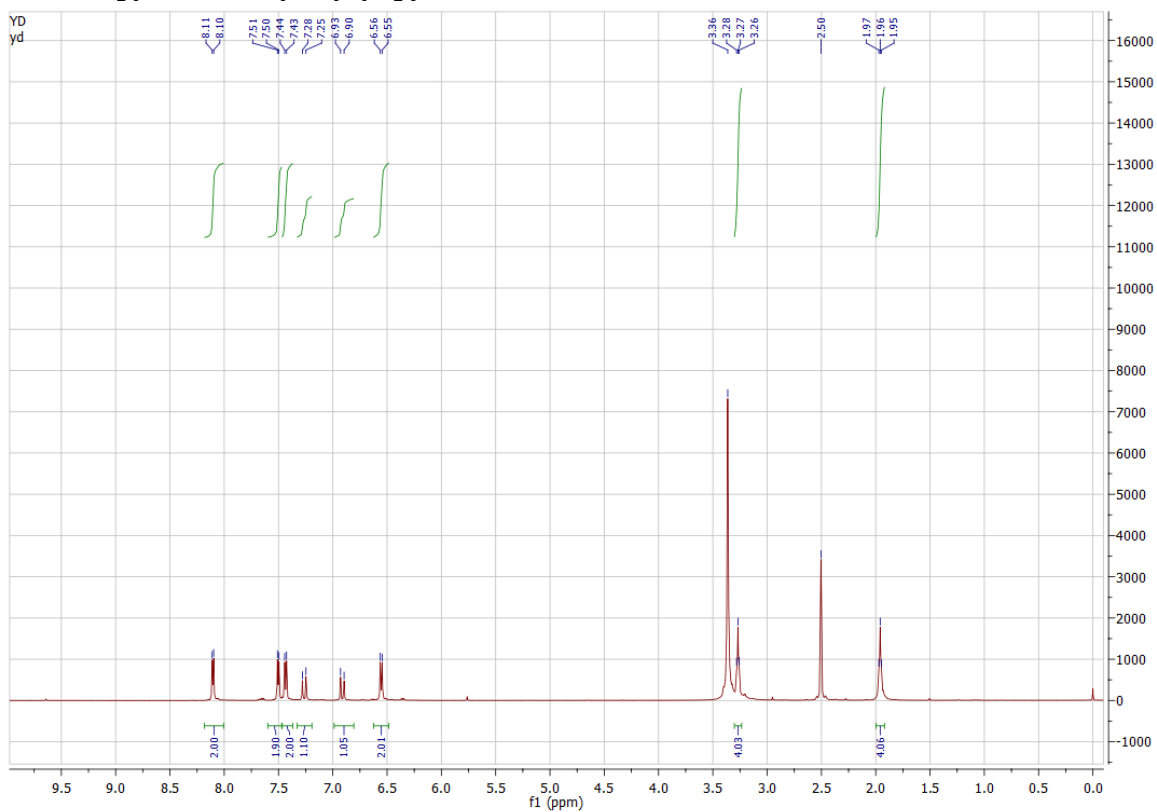


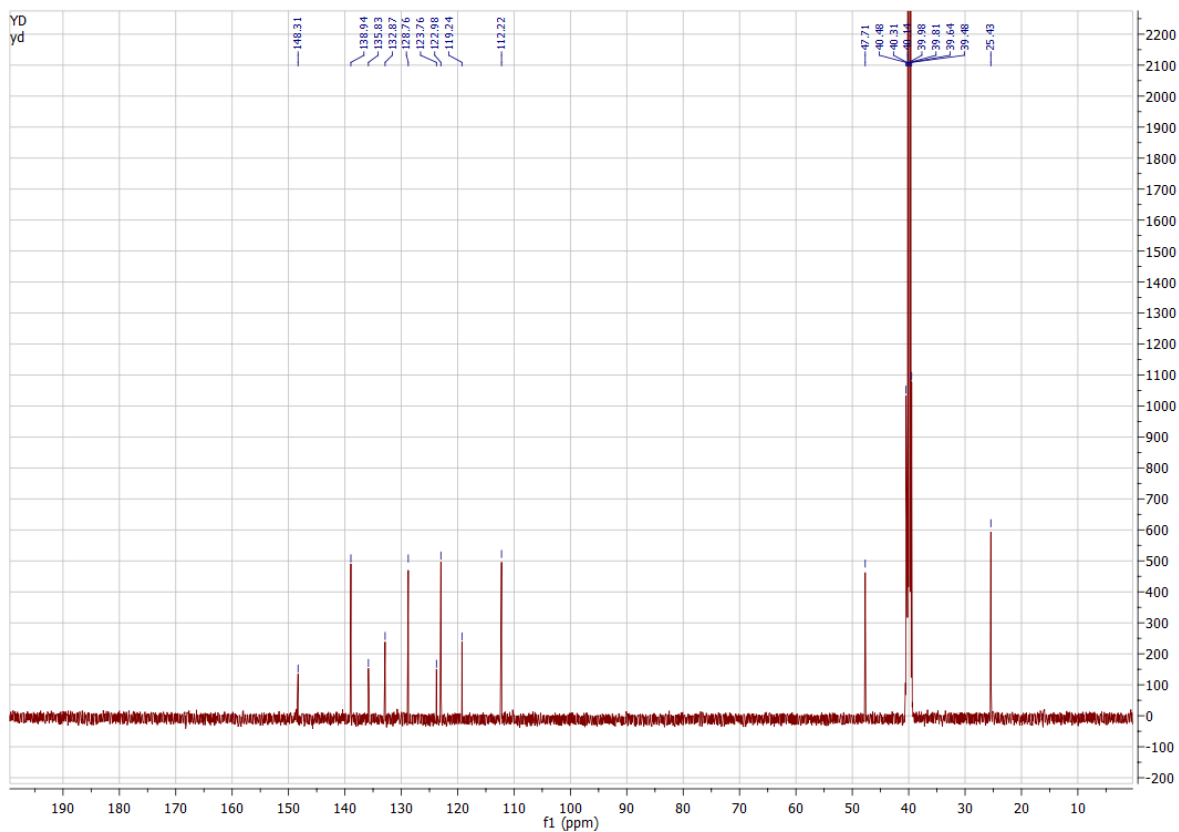
### (E)-4-(4-(dimethylamino)styryl)pyridine 1-oxide (DNO)





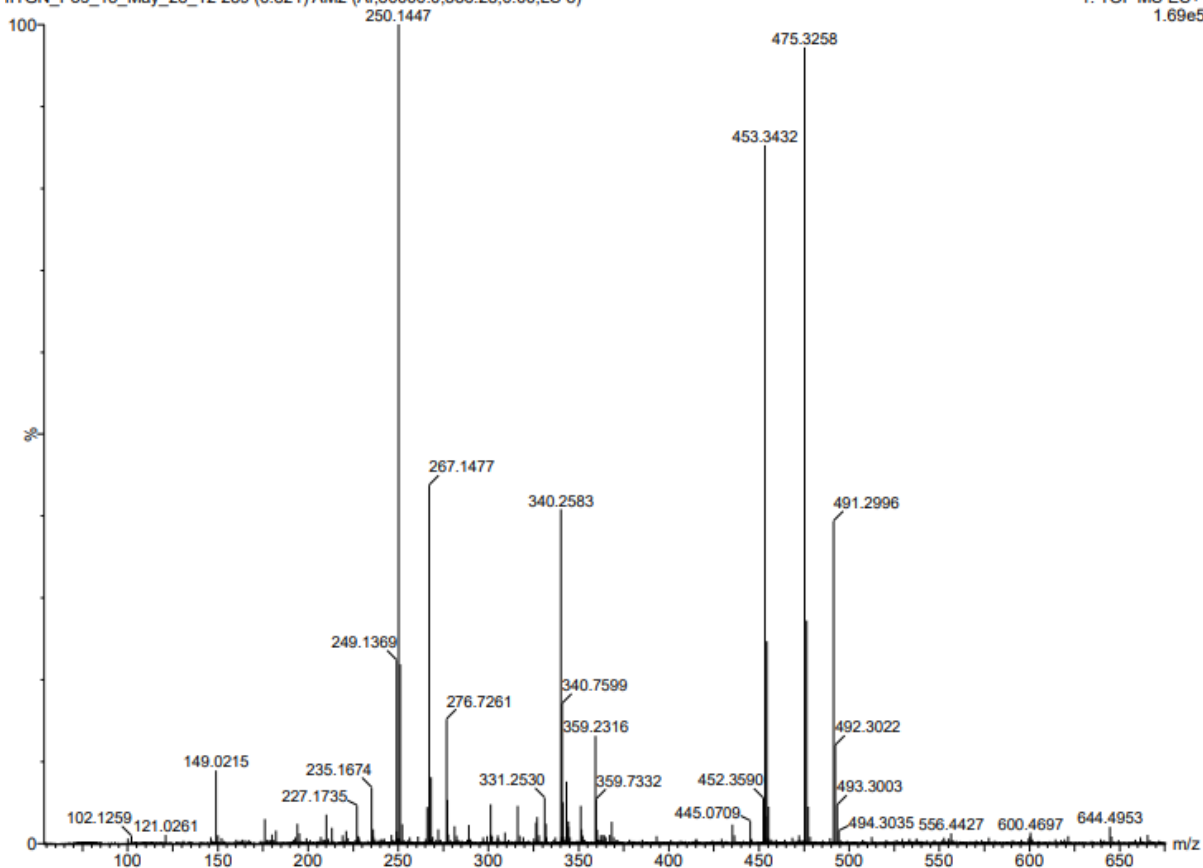
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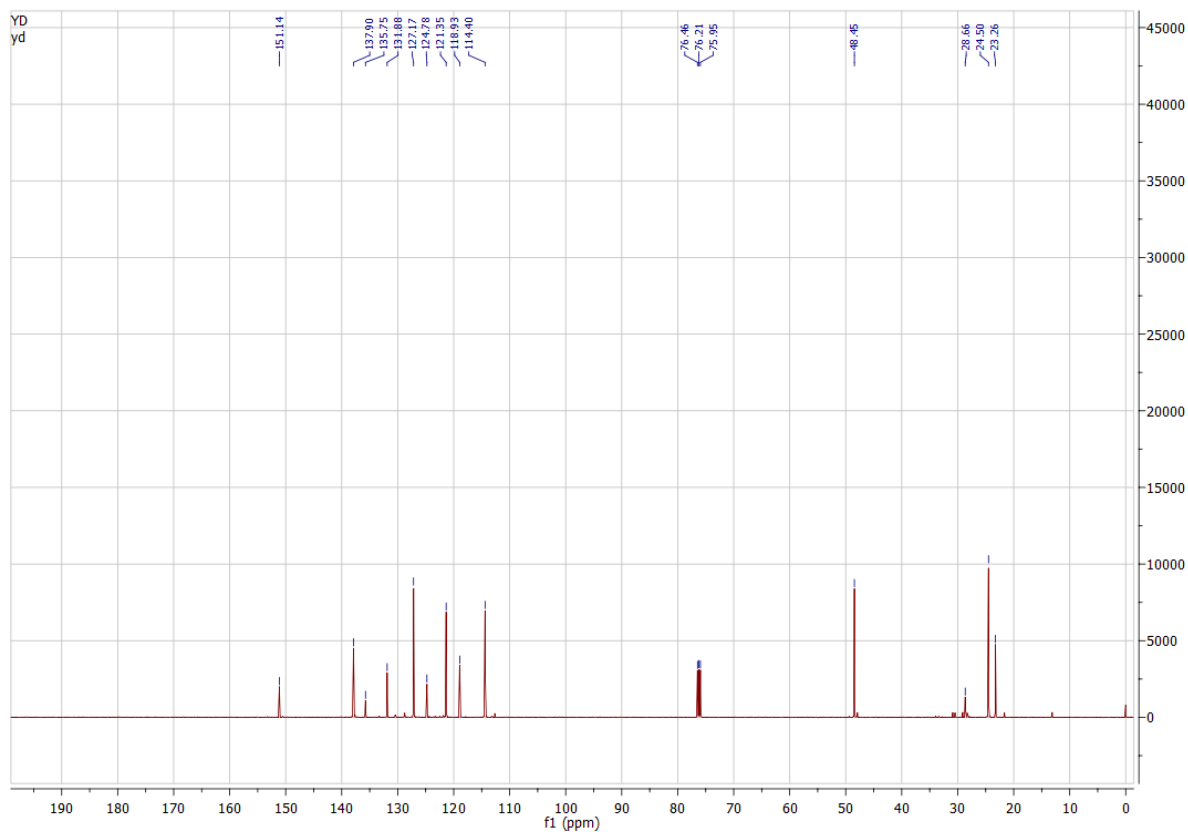
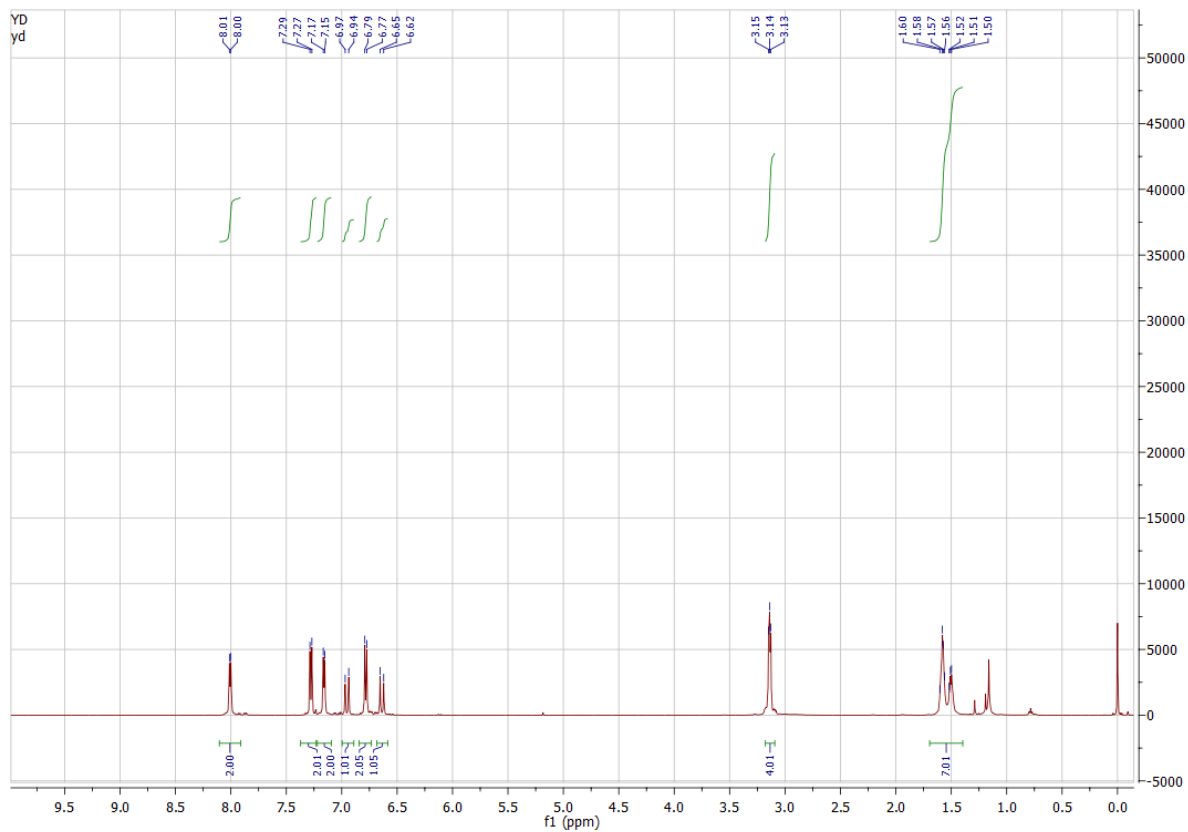


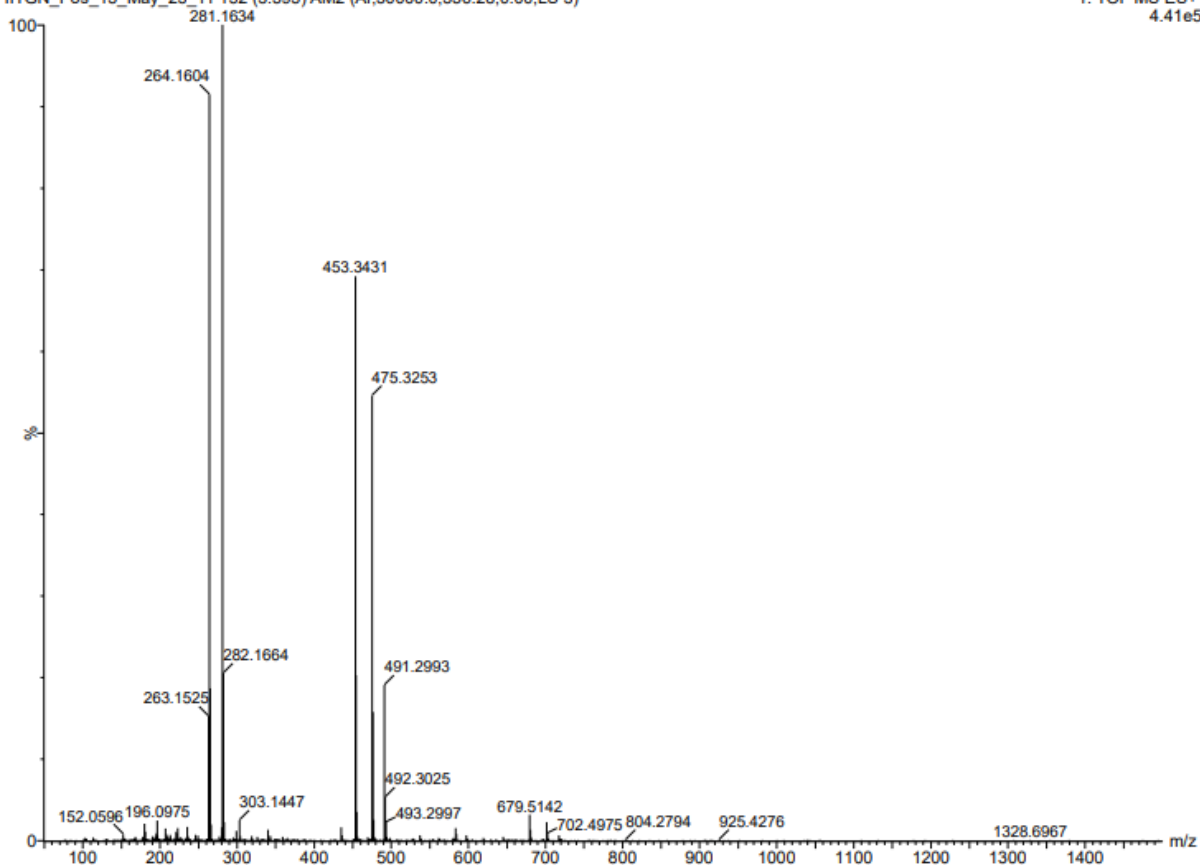
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1: TOF MS ES+  
1.69e5

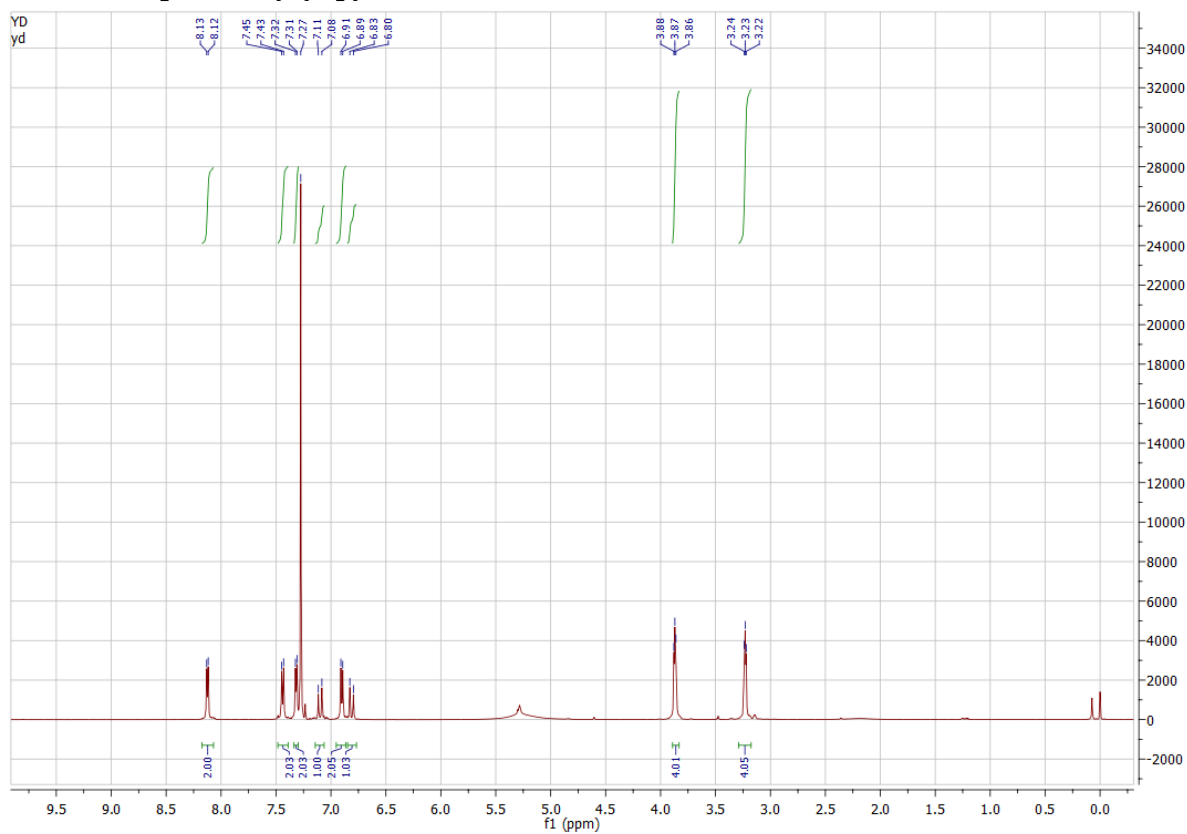


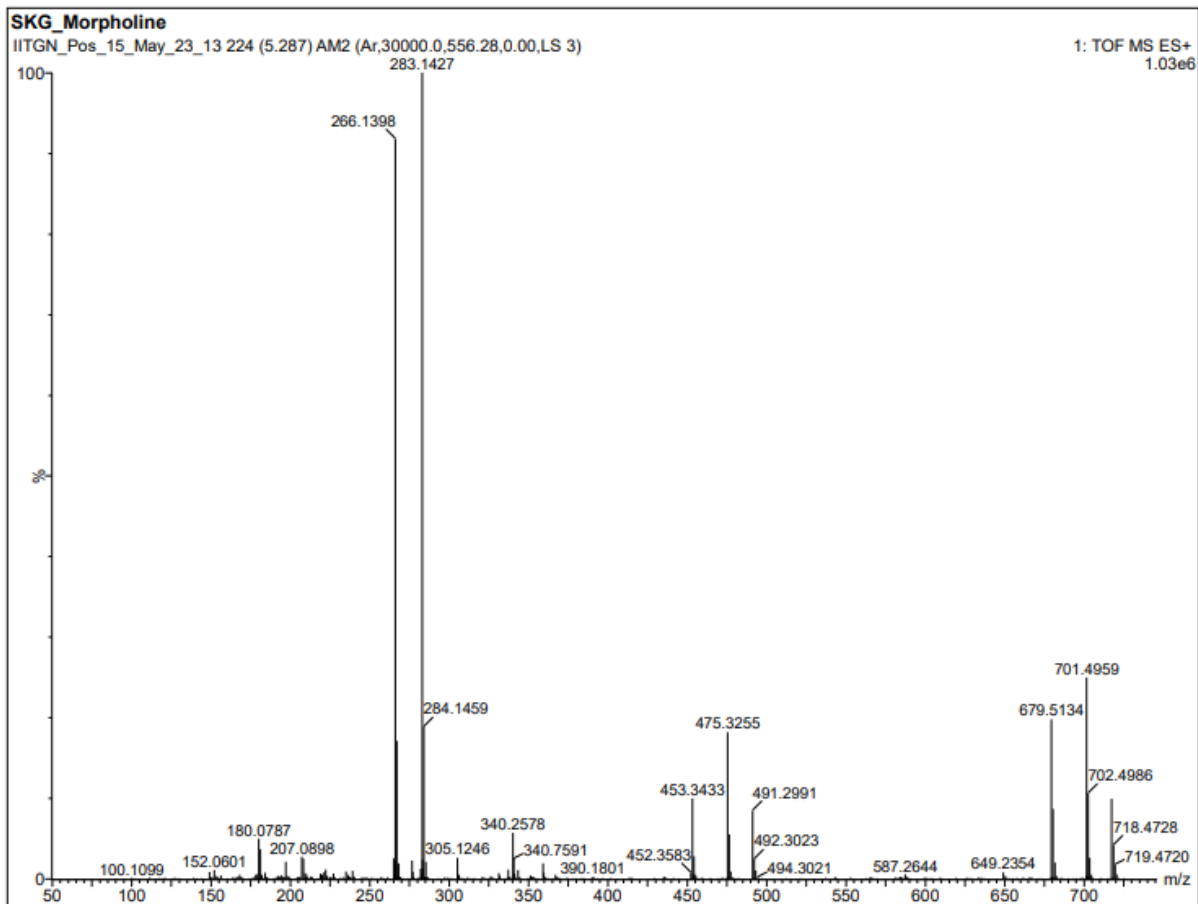
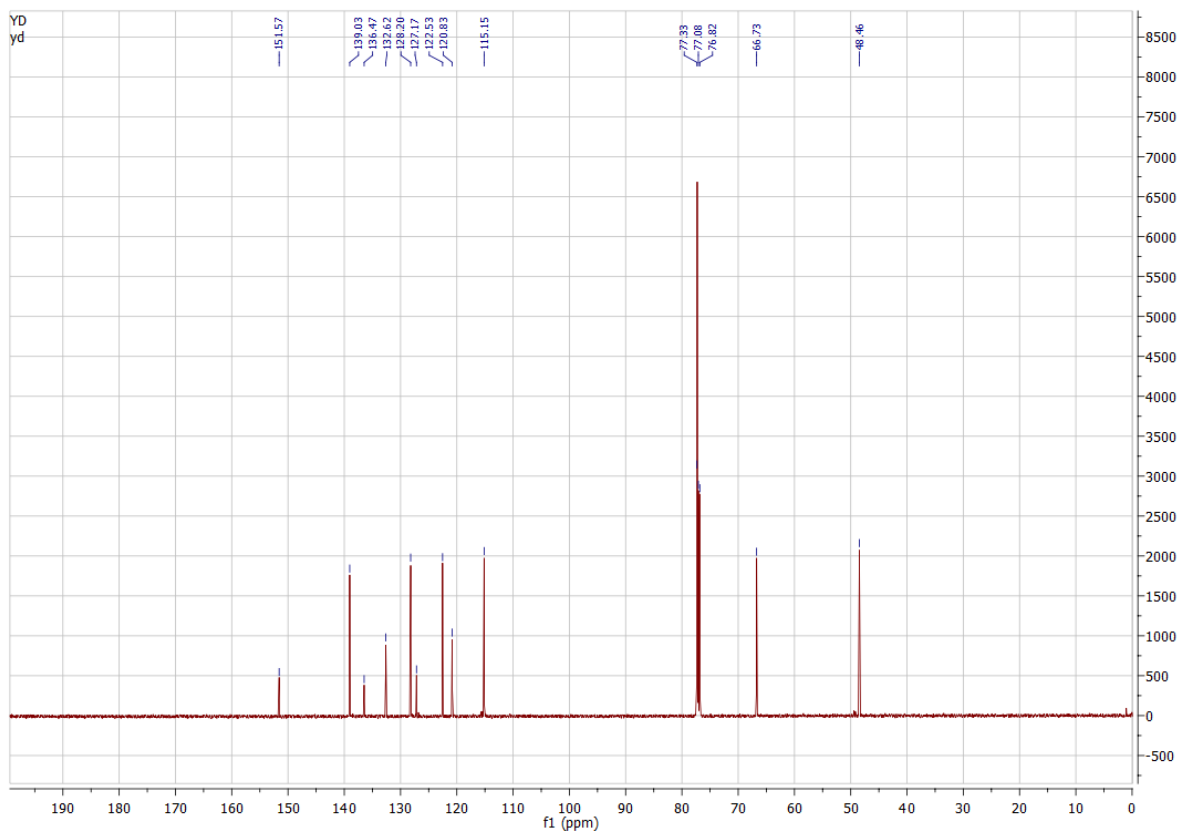
**(E)-4-(4-(piperidin-1-yl)styryl)pyridine 1-oxide (6NO)**



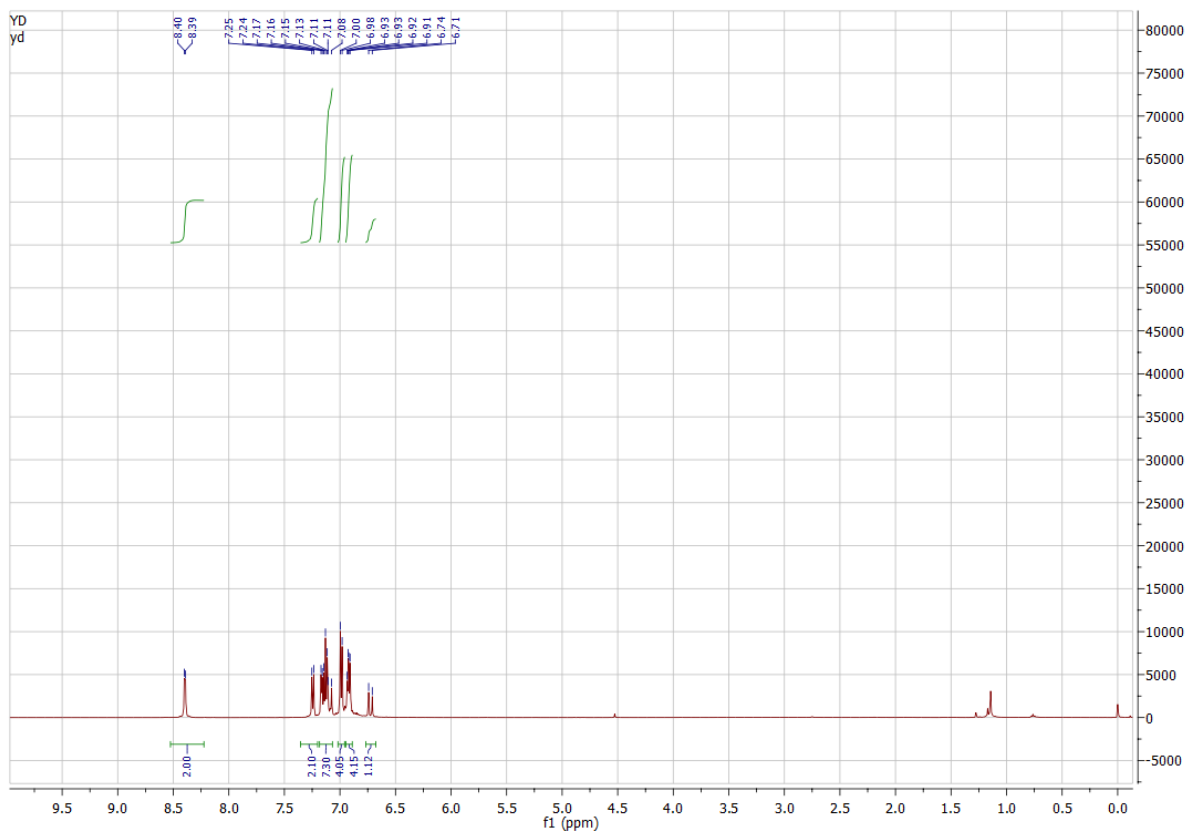


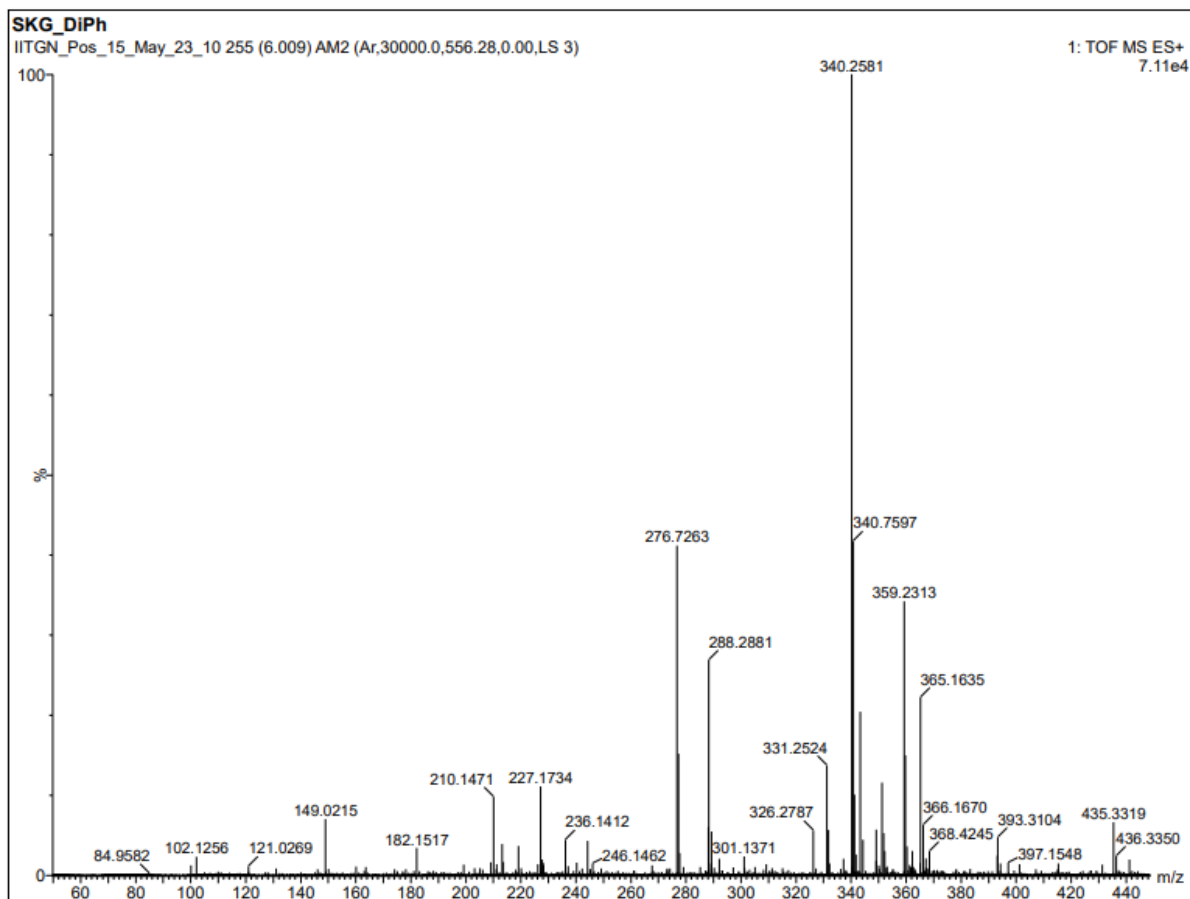
**(E)-4-(4-morpholinostyryl)pyridine 1-oxide (MNO)**





**(E)-4-(4-(diphenylamino)styryl)pyridine 1-oxide (TNO)**





## References

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2. Kumari, B., Singh, A., Jana, P., Radhakrishna, M., & Kanvah, S. *New Journal of Chemistry*, 2019, **43(29)**, 11701-11709.