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Supporting information

Molecular-iodine catalyzed selective construction of cyclopenta[b]indoles from indoles and acetone: a green gateway to indole-fused cycles

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1. Copies of ¹H and ¹³C NMR spectra



Fig. S1 400 MHz ¹H NMR spectrum of (2a)¹ in CDCl₃



Fig. S2 100 MHz 13 C NMR spectrum of $(2a)^1$ in CDCl₃



Fig. S3 400 MHz ¹H NMR spectrum of $(2b)^1$ in CDCl₃



Fig. S4 100 MHz 13 C NMR spectrum of $(2b)^1$ in CDCl₃



Fig. S5 400 MHz ¹H NMR spectrum of (2c)² in CDCl₃



Fig. S6 100 MHz 13 C NMR spectrum of $(2c)^2$ in CDCl₃



Fig. S7 400 MHz ¹H NMR spectrum of $(2d)^2$ in CDCl₃



Fig. S8 100 MHz ¹³C NMR spectrum of (2d)² in CDCl₃



Fig. S9 400 MHz ¹H NMR spectrum of 2e in CDCl₃



Fig. S10 100 MHz ¹³C NMR spectrum of 2e in CDCl₃



Fig. S11 400 MHz ¹H NMR spectrum of 2f in CDCl₃





Fig. S13 400 MHz ¹H NMR spectrum of (2g)¹ in CDCl₃



Fig. S14 100 MHz 13 C NMR spectrum of $(2g)^1$ in CDCl₃



Fig. S15 400 MHz ¹H NMR spectrum of (2h)² in CDCl₃



Fig. S16 100 MHz 13 C NMR spectrum of $(2h)^2$ in CDCl₃



Fig. S17 400 MHz ¹H NMR spectrum of 2i in CDCl₃



Fig. S18 100 MHz ¹³C NMR spectrum of 2i in CDCl₃



Fig. S19 400 MHz ¹H NMR spectrum of 2j in CDCl₃



Fig. S20 100 MHz ¹³C NMR spectrum of 2j in CDCl₃



Fig. S21 400 MHz ¹H NMR spectrum of 2k in CDCl₃



Fig. S22 100 MHz 13 C NMR spectrum of 2k in CDCl₃



Fig. S23 400 MHz ¹H NMR spectrum of 2l in CDCl₃



Fig. S24 100 MHz ¹³C NMR spectrum of 2l in CDCl₃

ψψ



Fig. S25 400 MHz ¹H NMR spectrum of 2m in CDCl₃



Fig. S26 100 MHz ¹³C NMR spectrum of 2m in CDCl₃



Fig. S27 400 MHz ¹H NMR spectrum of 2n in CDCl₃



Fig. S28 100 MHz 13 C NMR spectrum of 2n in CDCl₃



Fig. S29 400 MHz ¹H NMR spectrum of 20 in CDCl₃



Fig. S30 100 MHz 13 C NMR spectrum of 20 in CDCl₃



Fig. S31 400 MHz ¹H NMR spectrum of 2p in CDCl₃



Fig. S32 100 MHz 13 C NMR spectrum of 2p in CDCl₃



Fig. S33 400 MHz ¹H NMR spectrum of (2q)³ in CDCl₃



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Fig. S35 400 MHz ¹H NMR spectrum of $(3a)^2$ in CDCl₃



Fig. S36 100 MHz 13 C NMR spectrum of $(3a)^2$ in CDCl₃



Fig. S37 400 MHz ¹H NMR spectrum of (3b)² in CDCl₃



Fig. S38 100 MHz 13 C NMR spectrum of $(3b)^2$ in CDCl₃



Fig. S39 400 MHz ¹H NMR spectrum of 3c in DMSO- d_6



Fig. S40 100 MHz ¹³C NMR spectrum of 3c in DMSO- d_6



Fig. S41 400 MHz ¹H NMR spectrum of (3d)⁴ in CDCl₃



Fig. S42 100 MHz 13 C NMR spectrum of $(3d)^4$ in CDCl₃



Fig. S43 400 MHz ¹H NMR spectrum of (3e)⁴ in CDCl₃



Fig. S44 100 MHz ¹³C NMR spectrum of (3e)⁴ in CDCl₃



Fig. S45 400 MHz ¹H NMR spectrum of $(3a')^5$ in CDCl₃



Fig. S46 100 MHz 13 C NMR spectrum of (3a')⁵ in CDCl₃

2. Copies of HRMS spectra



Fig. S47 HRMS spectrum of 2c

Single Mass Analysis Tolerance = 100.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3 Monoisotopic Mass, Even Electron Ions 1033 formula(e) evaluated with 165 results within limits (all results (up to 1000) for each mass) Elements Used:

Mass	Calc. Mass	mDa	PPM	DBE	Formula	i-FIT	i-FIT Norm	Fit Conf %	С	н	N	0	
403.2385	403.2386	-0.1	-0.2	12.5	C26 H31 N2 O2	700.0	8.010	0.03	26	31	2	2	
	403.2391	-0.6	-1.5	5.5	C11 H27 N14 O3	695.2	3.185	4.14	11	27	14	3	
	403.2377	0.8	2.0	0.5	C10 H31 N10 O7	696.4	4.379	1.25	10	31	10	7	
	403.2404	-1.9	-4.7	-0.5	C14 H35 N4 O9	697.8	5.866	0.28	14	35	4	9	
	403.2359	2.6	6.4	13.5	C22 H27 N8	698.8	6.787	0.11	22	27	8		
	403.2417	-3.2	-7.9	4.5	C15 H31 N8 O5	697.4	5.381	0.46	15	31	8	5	•

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н

1: TOF MS ES+



PB-465_28122020_007 454 (3.949) AM2 (Ar,30000.0,556.32,0.00,LS 1); Cm (444:483)



Fig. S48 HRMS spectrum of 2d

3. Single crystal XRD data of 2d

(CCDC 2254620)



Fig. S49 ORTEP diagram of 2d with 50% probability ellipsoids

Crystal data	2d
Formula unit	$C_{26}H_{30}N_2O_2$
Formula weight (gmol ⁻¹)	402.52
Crystal system	orthorhombic
T [K]	100 K
a [Å]	8.0818(7)
<i>b</i> [Å]	16.4868(15)
c [Å]	32.475(3)
α [°]	90
β [°]	90
γ [°]	90
Volume [Å ³]	4327.1(7)
Space group	Pbca
Z	8
$D_{cal}[g/cm^3]$	1.236
$ R_1, wR2 $	0.0429, 0.0941
Instrument	Bruker CCD Apex II
CCDC No	CCDC 2254620

Table S1 Crystallographic parameters of structures 2d

Single crystal X-ray diffraction. Single crystal X-ray diffractions were collected on a Bruker SMART APEX-II CCD diffractometer using Mo K α ($\lambda = 0.71073$ Å) radiation. Bruker SAINT software has been employed for reducing the data and SADABS for correcting the intensities of absorption. Structure was solved and refined using SHELXL with anisotropic displacement parameters for non-H atoms. In the crystal structure H-atoms are located experimentally, whereas C–H atoms were fixed geometrically using the HFIX command in SHELX-TL. No missed symmetry observed in the final check of CIF file using PLATON. Information of crystallographic parameters for the structure is furnished in Table S1.

4. References

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