

BP23C7: high-yield synthesis and application in constructing [3]rotaxanes and responsive pseudo[2]rotaxanes

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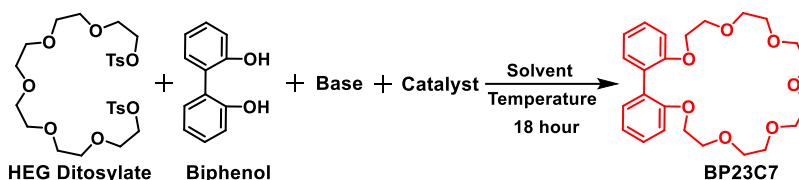
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Electronic Supplementary Information

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1. Optimization table for BP23C7 synthesis



Entry	Solvent ^b	Base (equiv.)	Catalyst (equiv.)	Temperature	Yield (%) ^c
1	THF	NaH (3)	KPF ₆ (0.25)	Room temp.	44
2	THF	NaH (6)	KPF ₆ (0.25)	Room temp.	25
3	DMF	NaH (3)	KPF ₆ (0.25)	Room temp.	23
4	DMF	NaH (6)	KPF ₆ (0.25)	Room temp.	15
5	THF	NaH (3)	KPF ₆ (0.25)	80°C	83
6	THF	NaH (3)	KI (0.25)	80°C	77
7	THF	K ₂ CO ₃ (3)	-	80°C	66
8	THF	KO ^t Bu (3)	-	80°C	71
9	THF	Cs ₂ CO ₃ (3)	-	80°C	71
10	DMF	NaH (3)	KPF ₆ (0.25)	80°C	51
11	DMF	Cs ₂ CO ₃ (3)	-	80°C	64
12	ACN	Cs ₂ CO ₃ (3)	-	80°C	86

^a100 mg scale of HEG ditosylate has been used throughout. ^bTHF = Tetrahydrofuran, DMF = N,N'-Dimethylformamide and ACN = Acetonitrile. ^cIsolated yield.

Table S1. Optimization reaction condition for **BP23C7** synthesis. Heating condition afforded better yields. Three equivalents of base worked better than six equivalents of base. Templates such as KPF₆ and KI are effective, and bases comprising of alkali metal ions performed additional role of templating. Cs₂CO₃ in acetonitrile medium at 80°C afforded the highest yield (86%, entry 12).

2. Nominal ESI-MS of a 1:1 mixture of DBA•PF₆ & BP23C7

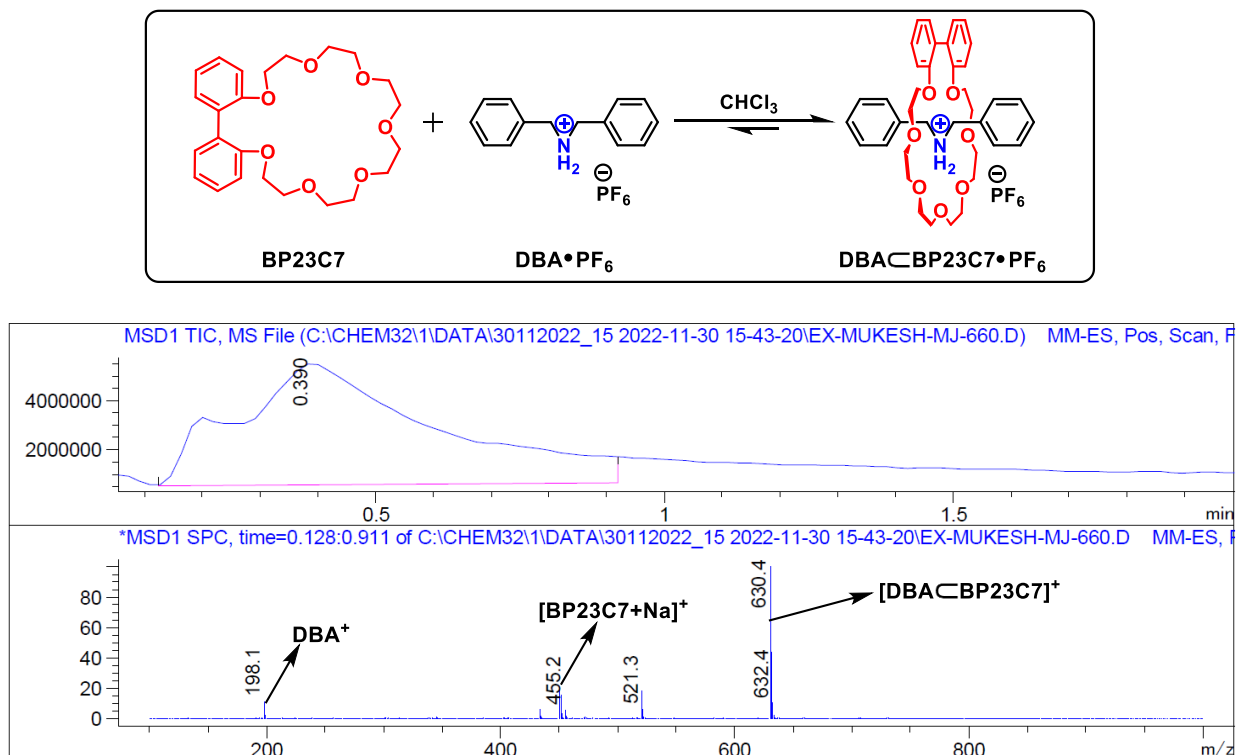


Figure S1. ESI mass spectrum of the equimolar mixture of **DBA•PF₆** & **BP23C7** displayed base peak at $m/z = 630.4$ $[M-PF_6]^+$, where “M” corresponds to the pseudo[2]rotaxane species **DBA⊂BP23C7•PF₆**.

3. ^1H NMR spectrum of a 1:1 mixture of $\text{DBA}\cdot\text{PF}_6$ & BP23C7

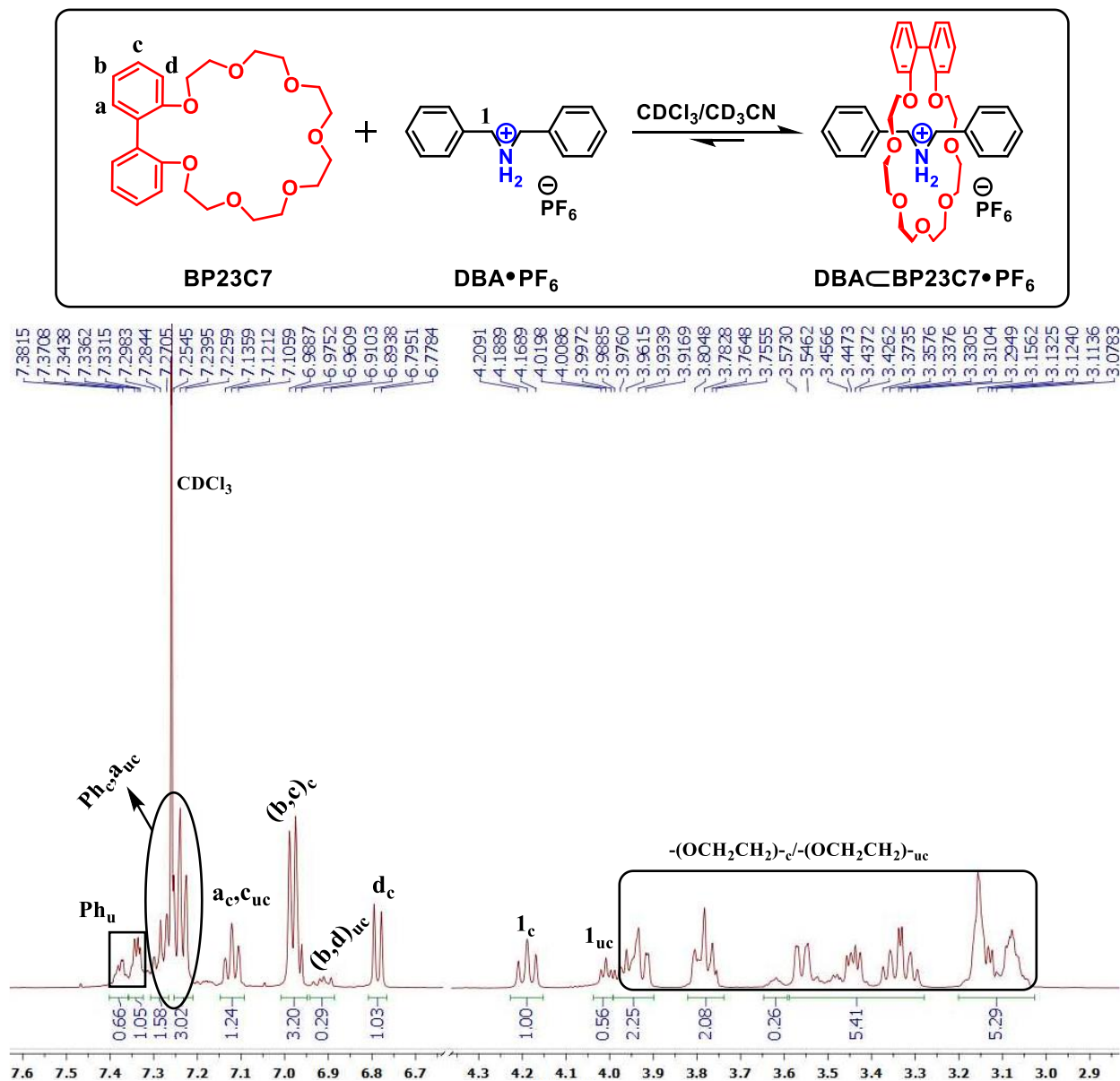


Figure S2. ^1H NMR spectrum (500 MHz, $\text{CDCl}_3/\text{CD}_3\text{CN} = 11:1$) of a 5mM equimolar mixture of $\text{DBA}\cdot\text{PF}_6$ & BP23C7 . K_a value calculated from integrations of the complexed and the uncomplexed peaks “1” of $\text{DBA}\cdot\text{PF}_6$ is $[(1.00/1.56) \times 5.00 \times 10^{-3}] / [(1 - 1.00/1.56) \times 5.00 \times 10^{-3}]^2 = 1.00 \times 10^3 \text{ M}^{-1}$.

4. Nominal ESI-MS of a 1:1 mixture of 1-H•PF₆ & BP23C7

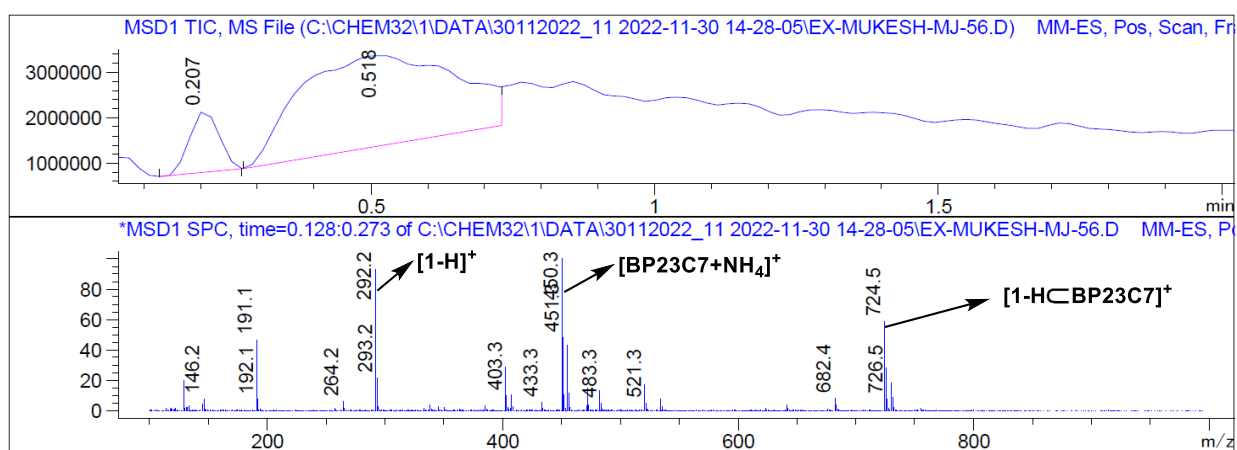
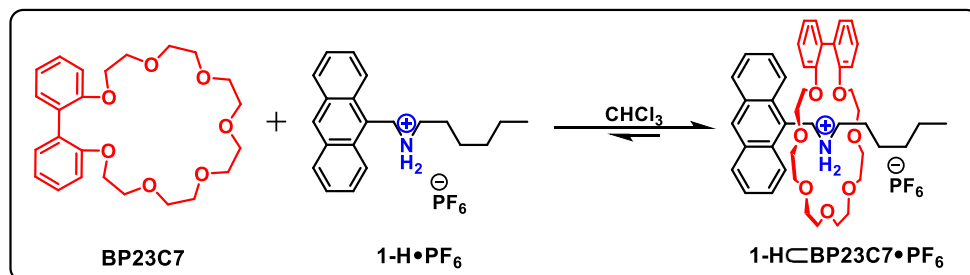


Figure S3. ESI mass spectrum of the equimolar mixture of 1-H•PF₆ & BP23C7 displayed base peak at $m/z = 724.5$ [M-PF₆]⁺, where “M” corresponds to the pseudo[2]rotaxane species 1-H⊂BP23C7•PF₆.

5. ^1H NMR spectrum of a 1:1 mixture of $1\text{-H}\cdot\text{PF}_6$ & BP23C7

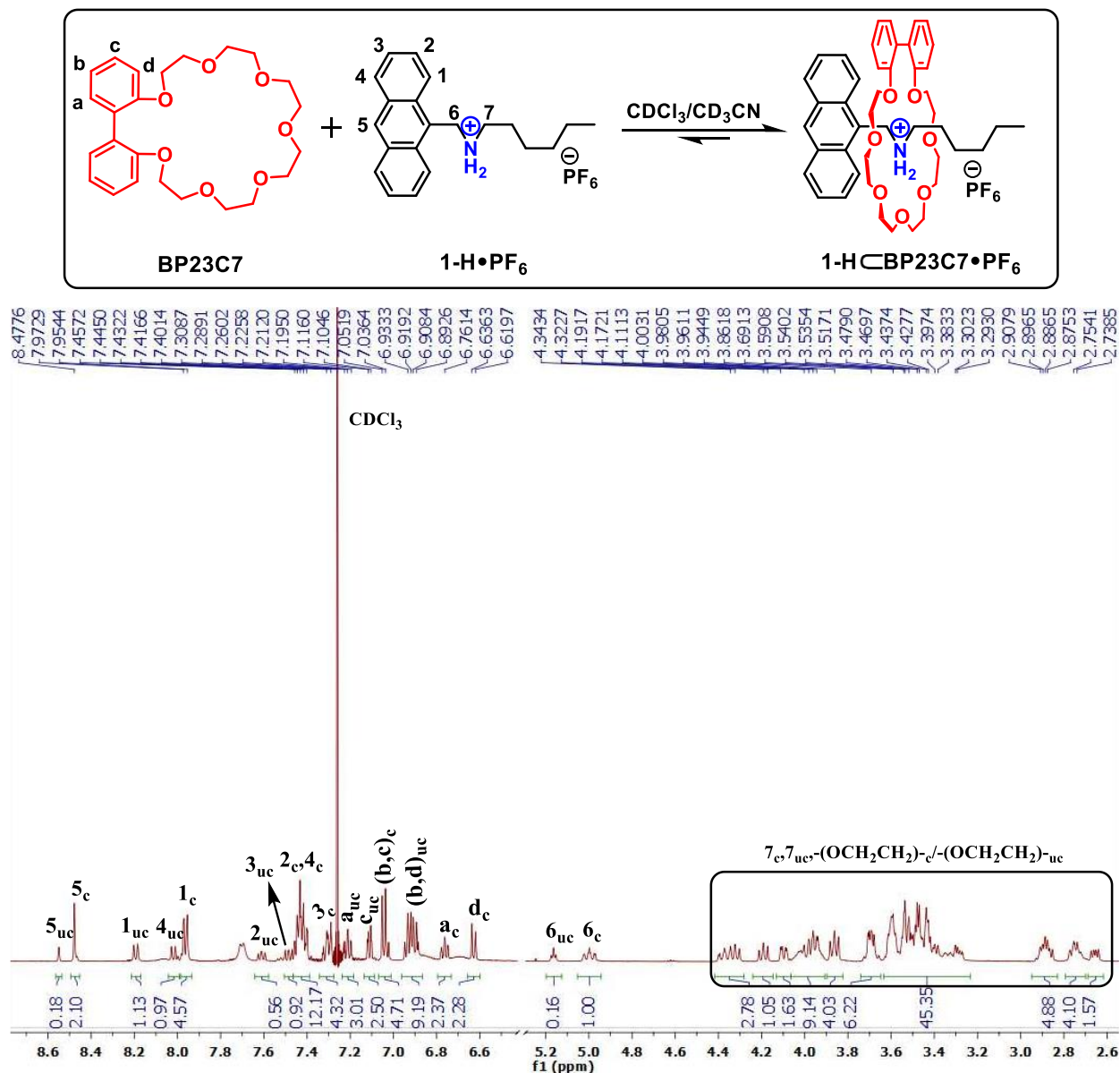


Figure S4. ^1H NMR spectrum (500 MHz, $\text{CDCl}_3/\text{CD}_3\text{CN} = 11:1$) of a 5mM equimolar mixture of $1\text{-H}\cdot\text{PF}_6$ & BP23C7. K_a value calculated from integrations of the complexed and the uncomplexed peaks “6” of $1\text{-H}\cdot\text{PF}_6$ is $[(1.00/1.16) \times 5.00 \times 10^{-3}] / [(1 - 1.00/1.16) \times 5.00 \times 10^{-3}]^2 = 9.0 \times 10^3 \text{ M}^{-1}$.

6. Optical data of the mixture of 1-H•PF₆ & BP23C7

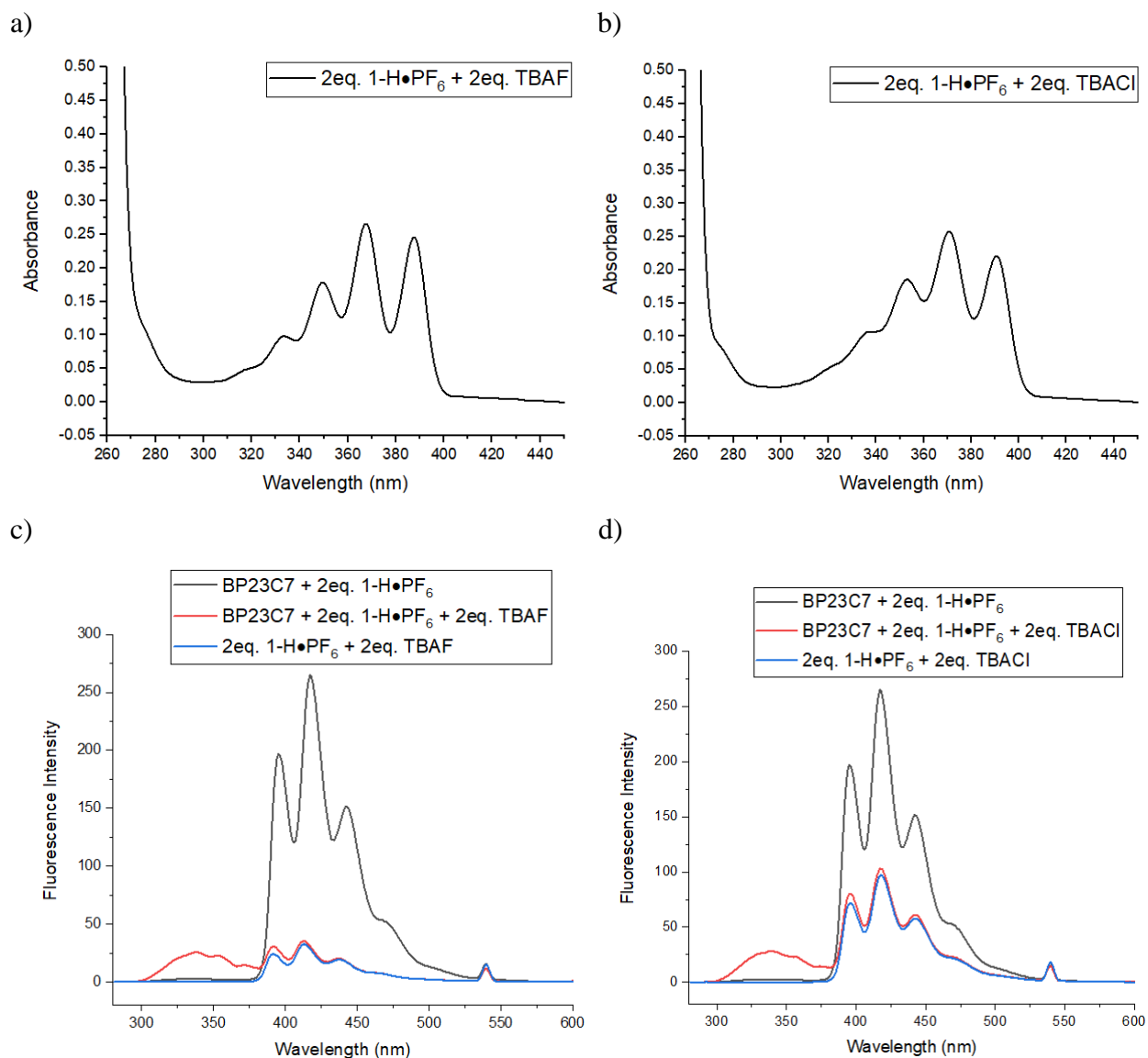


Figure S5. Absorption spectra (CH₂Cl₂, 298 K) of a) 2eq. 1-H•PF₆ ($4 \times 10^{-5} \text{ M}^{-1}$) + 2eq. TBAF; and b) 2eq. 1-H•PF₆ ($4 \times 10^{-5} \text{ M}^{-1}$) + 2eq. TBACl.

Fluorescence spectra (CH₂Cl₂, $\lambda_{\text{exc}} = 270 \text{ nm}$, 298 K) of c) BP23C7 ($2 \times 10^{-5} \text{ M}^{-1}$) + 2eq. 1-H•PF₆ (black), BP23C7 + 2eq. 1-H•PF₆ + 2eq. TBAF (red), 2eq. 1-H•PF₆ + 2eq. TBAF (blue); and d) BP23C7 ($2 \times 10^{-5} \text{ M}^{-1}$) + 2eq. 1-H•PF₆ (black), BP23C7 + 2eq. 1-H•PF₆ + 2eq. TBACl (red), 2eq. 1-H•PF₆ + 2eq. TBACl (blue).

7. Nominal ESI-MS of a 1:1 mixture of 2-H•PF₆ & BP23C7

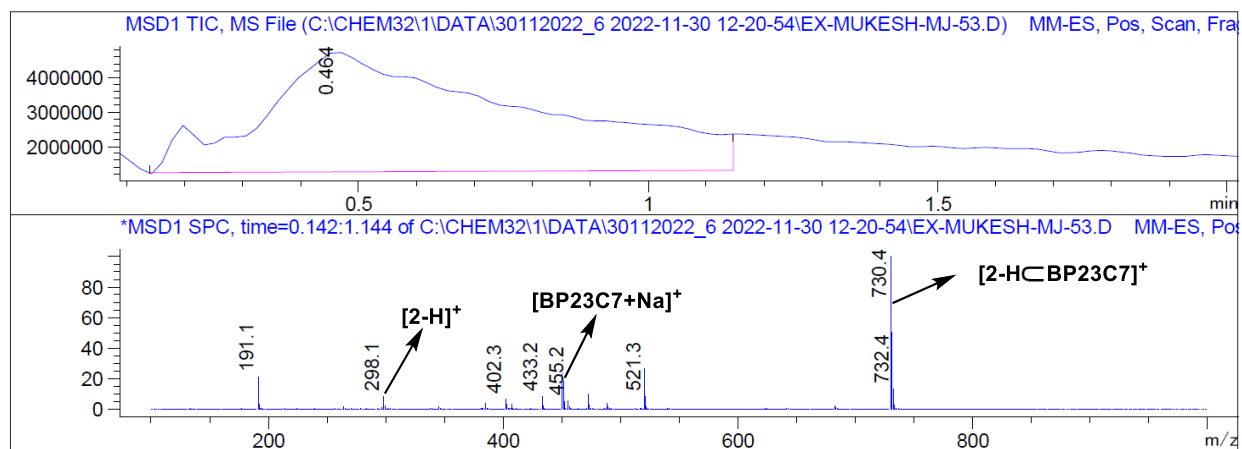
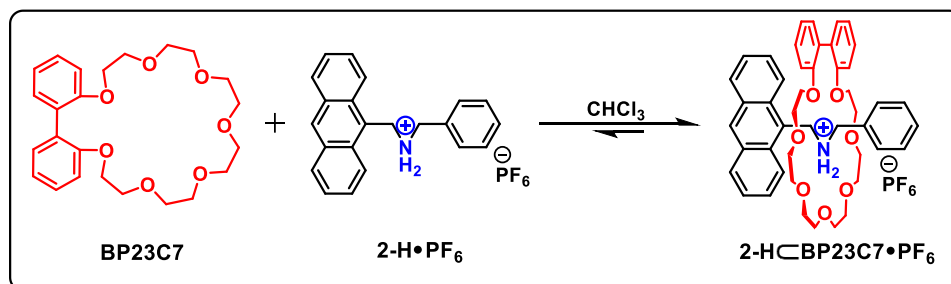


Figure S6. ESI mass spectrum of the equimolar mixture of 2-H•PF₆ & BP23C7 displayed base peak at $m/z = 730.4$ [M-PF₆]⁺, where “M” corresponds to the pseudo[2]rotaxane species 2-H⊂BP23C7•PF₆.

8. ^1H NMR spectrum of a 1:1 mixture of $2\text{-H}\cdot\text{PF}_6$ & BP23C7

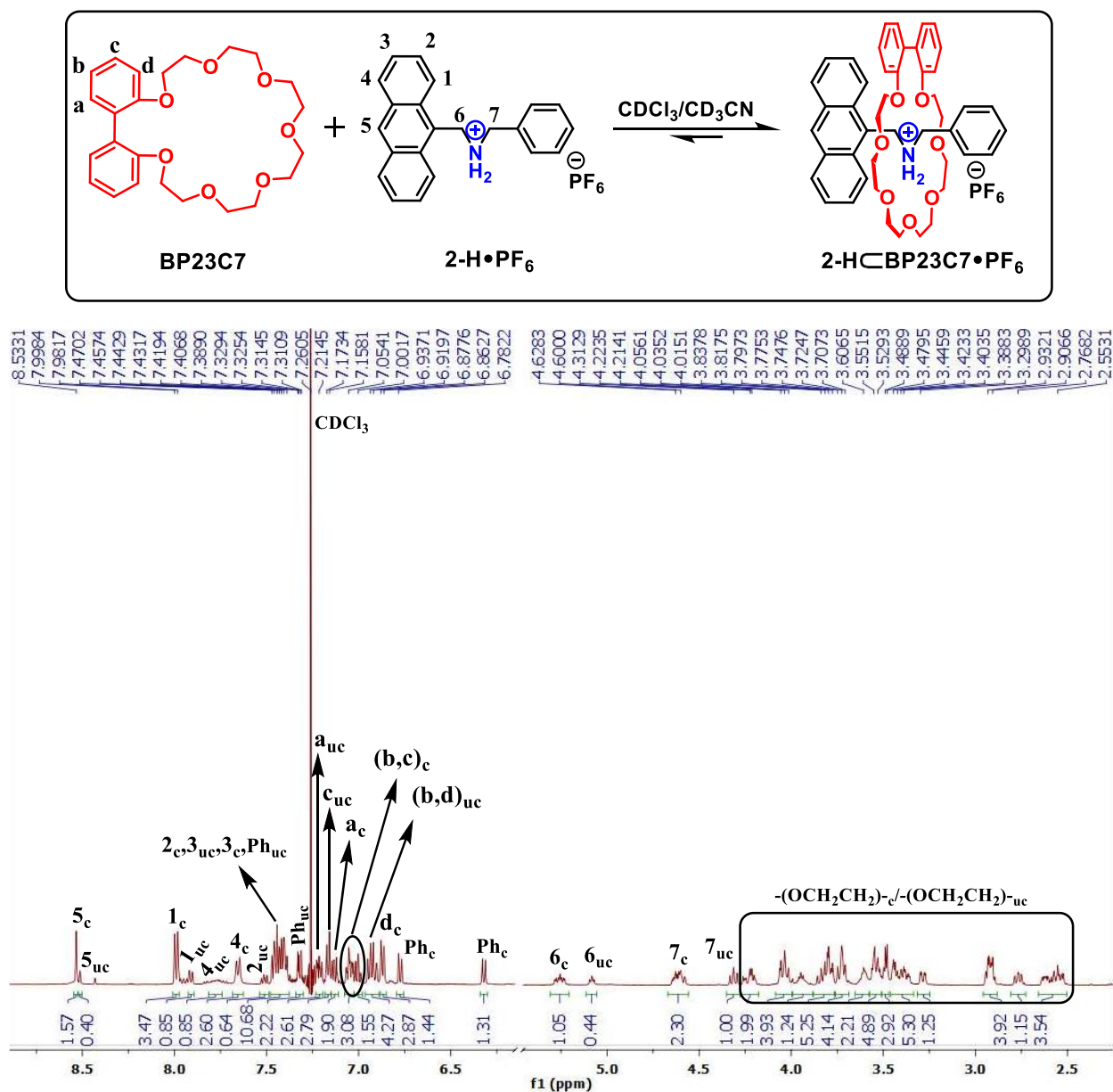


Figure S7. ^1H NMR spectrum (500 MHz, CDCl₃/CD₃CN = 11:1) of a 5mM equimolar mixture of $2\text{-H}\cdot\text{PF}_6$ & BP23C7. K_a value calculated from integrations of the complexed and the uncomplexed peaks “6” of $2\text{-H}\cdot\text{PF}_6$ is $[(1.05/1.49) \times 5.00 \times 10^{-3}] / [(1 - 1.05/1.49) \times 5.00 \times 10^{-3}]^2 = 1.6 \times 10^3 \text{ M}^{-1}$.

9. Optical data of the mixture of 2-H•PF₆ & BP23C7

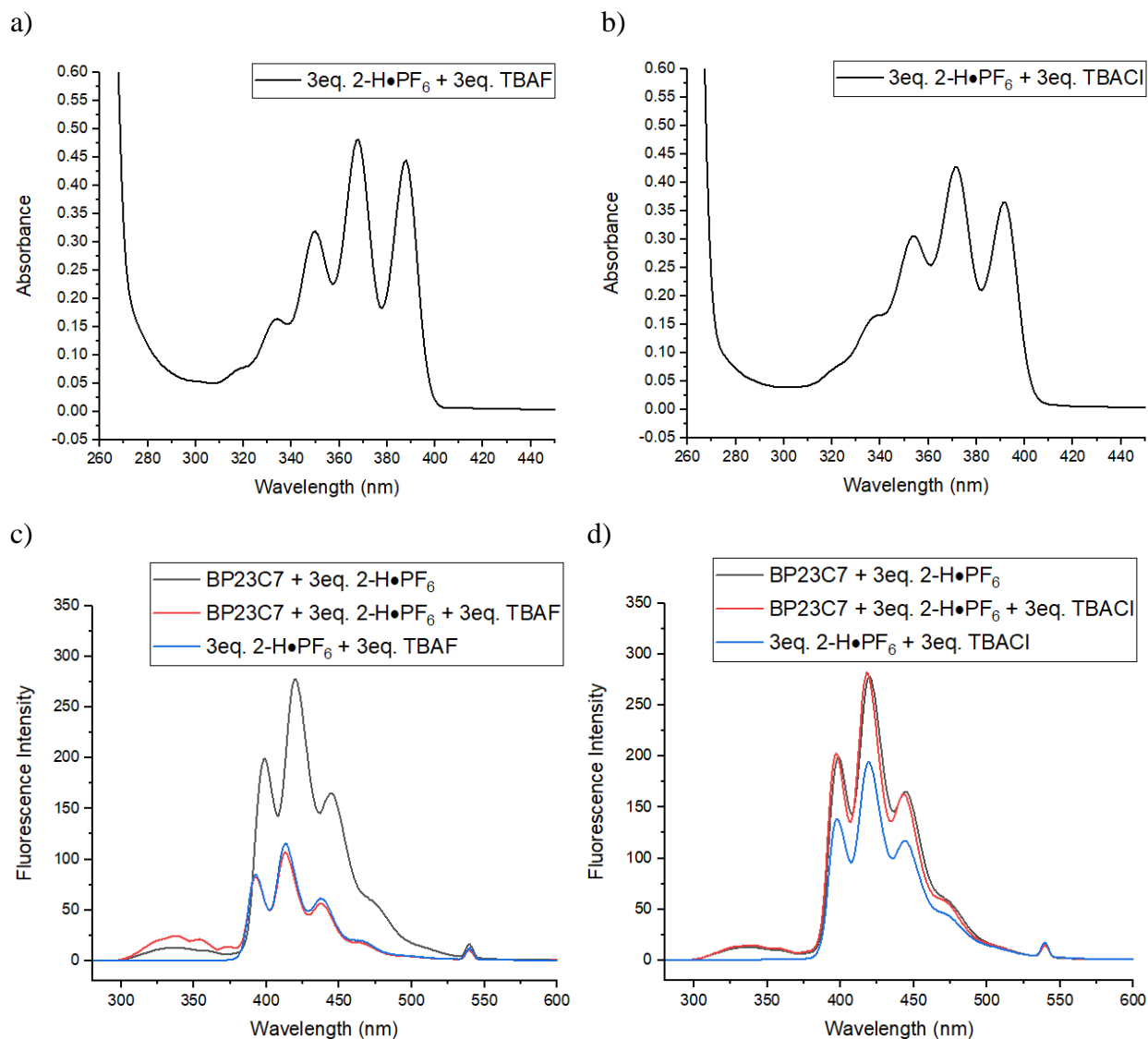
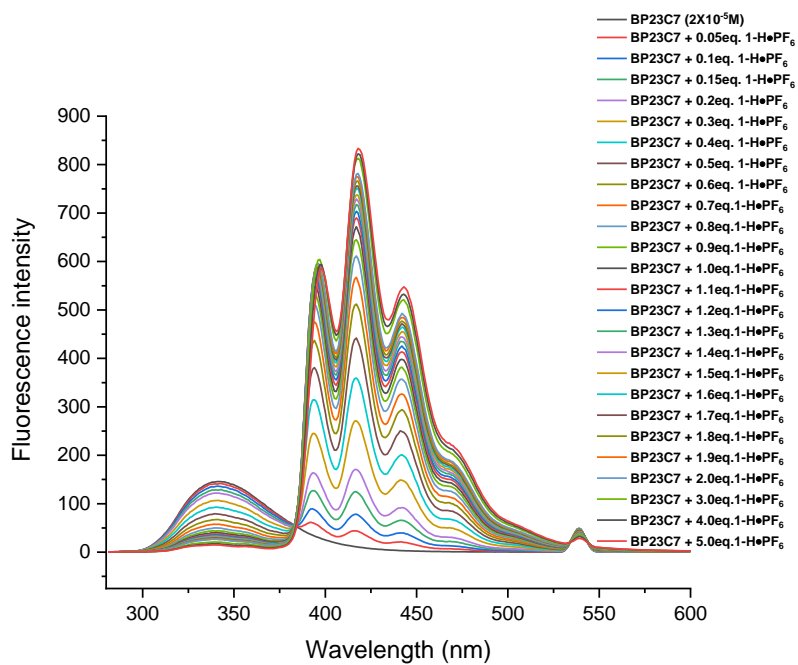


Figure S8. Absorption spectra (CH₂Cl₂, 298 K) of a) 3eq. 2-H•PF₆ ($6 \times 10^{-5} \text{ M}^{-1}$) + 3eq. TBAF; and b) 3eq. 2-H•PF₆ ($6 \times 10^{-5} \text{ M}^{-1}$) + 3eq. TBACl.

Fluorescence spectra (CH₂Cl₂, $\lambda_{\text{exc}} = 270 \text{ nm}$, 298 K) of c) BP23C7 ($2 \times 10^{-5} \text{ M}^{-1}$) + 3eq. 2-H•PF₆ (black), BP23C7 + 3eq. 2-H•PF₆ + 3eq. TBAF (red), 3eq. 2-H•PF₆ + 3eq. TBAF (blue); and d) BP23C7 ($2 \times 10^{-5} \text{ M}^{-1}$) + 3eq. 2-H•PF₆ (black), BP23C7 + 3eq. 2-H•PF₆ + 3eq. TBACl (red), 3eq. 2-H•PF₆ + 3eq. TBACl (blue).

10. Fluorescence titration of BP23C7 with 1-H·PF₆ & 1-H·Cl

a)



b)

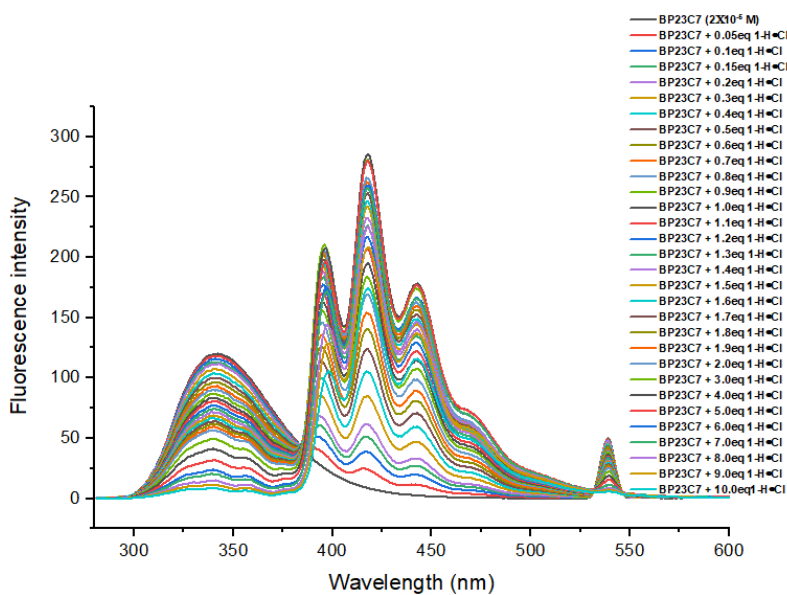
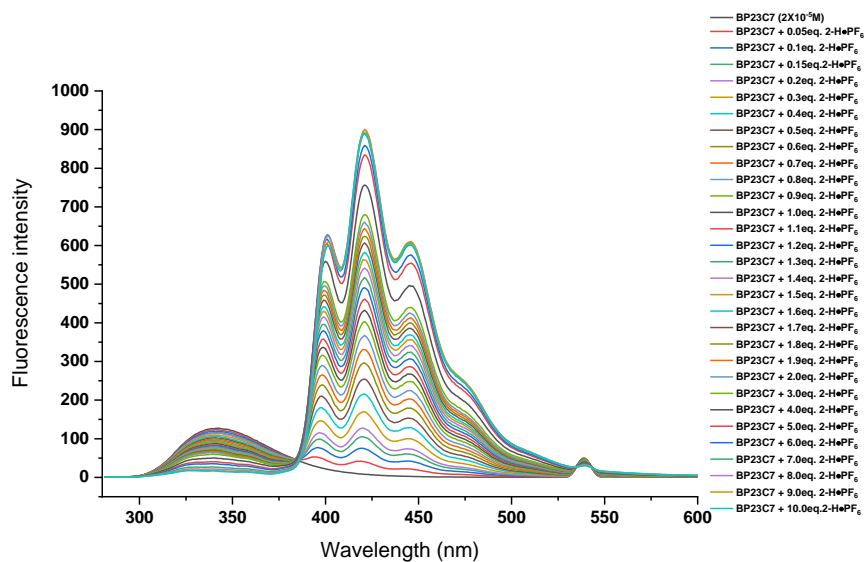


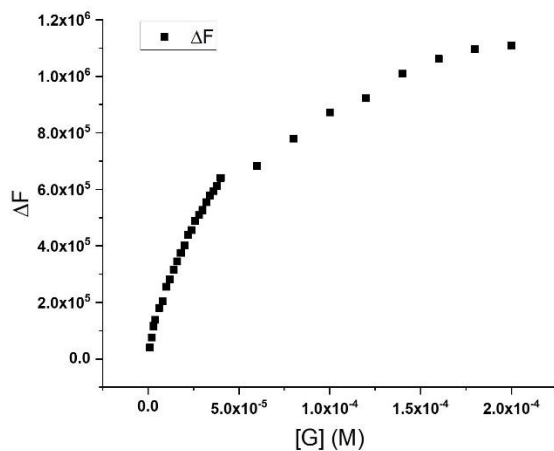
Figure S9. Fluorescence titration experiment of BP23C7 (2×10^{-5} M) with a) 1-H·PF₆ and b) 1-H·Cl.

11. Fluorescence titration of BP23C7 with 2-H·PF₆

a)



b)



c)

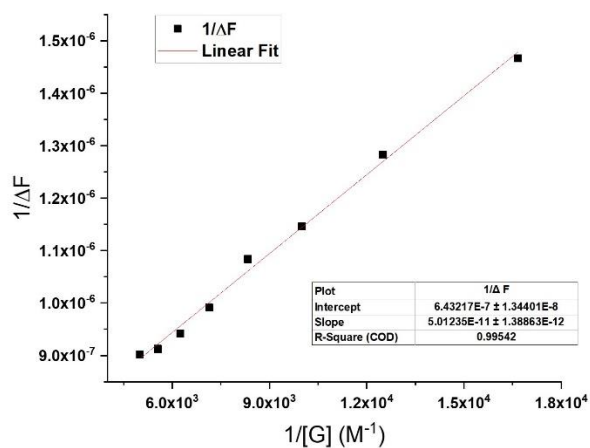


Figure S10. a) Fluorescence titration experiment of BP23C7 (2×10^{-5} M) with 2-H·PF₆, b) Plot of change in fluorescence quenching of BP23C7 with concentration of 2-H·PF₆; c) Benesi-Hildebrand plot from fluorescence titration data of (a), $K_a = \text{Intercept}/\text{Slope} = (1.283 \pm 0.008) \times 10^4 \text{ M}^{-1}$.

12. 2D NOESY NMR spectrum of the [3]Rotaxane 10-H₂•2PF₆

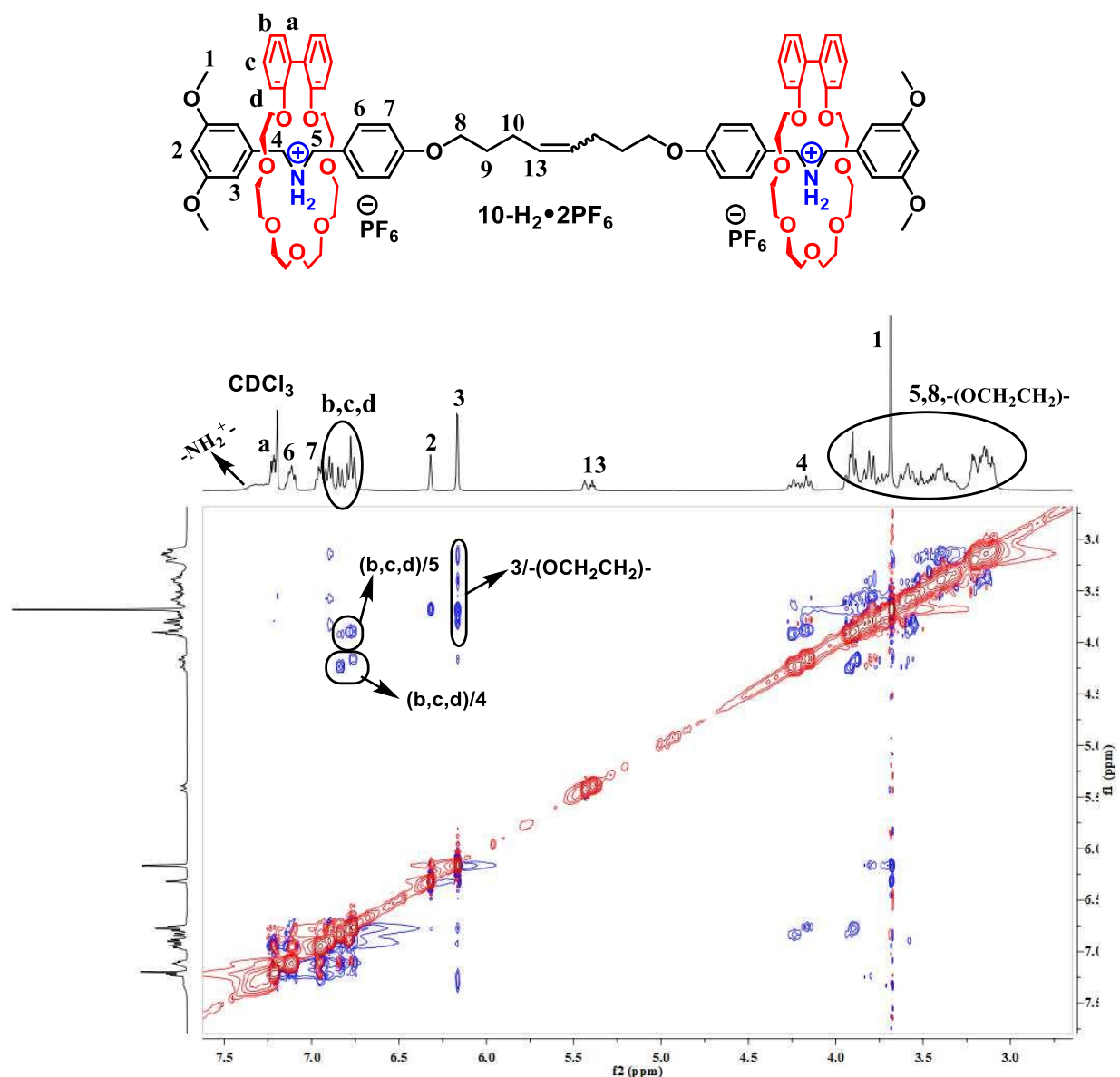


Figure S11. ¹H-¹H NOESY NMR (400 MHz, CDCl₃) of 10-H₂•2PF₆. In this case, a strong correlation between (-OCH₂CH₂-) protons of **BP23C7** with the aromatic protons H₃ of the stopper group is seen. Moreover, strong NOE cross peaks between **BP23C7** aromatic protons H_b, H_c, H_d and the benzylic protons H₄ and H₅ of the axle corroborates the interlocking geometry.

13. HR ESI-MS spectrum of the [3]Rotaxane 10-H₂•2PF₆

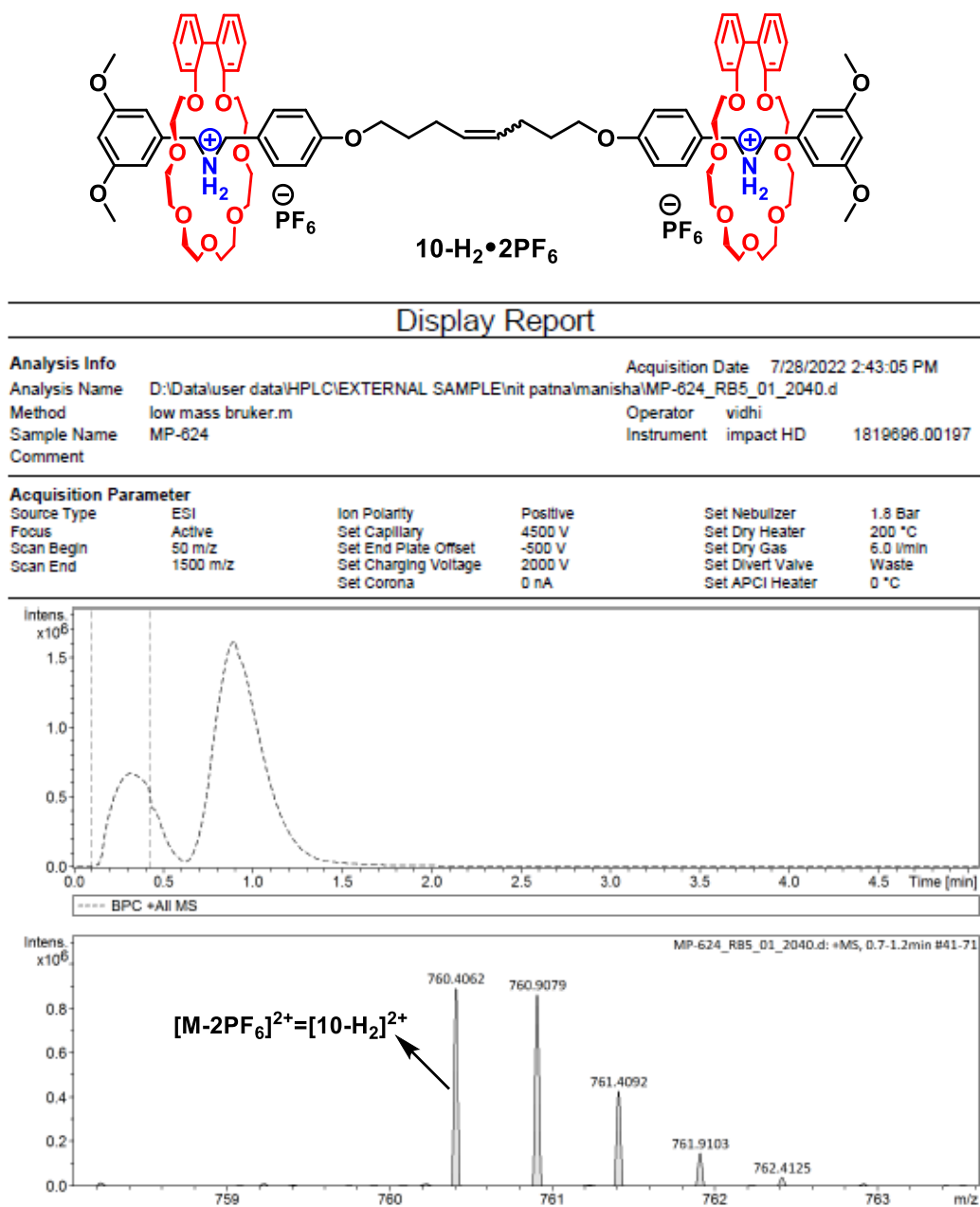


Figure S12. HR ESI-MS spectrum of the [3]rotaxane **10-H₂•2PF₆**. Peak at $m/z = 760.4062$ corresponds to $[M-2PF_6]^{2+}$ species. Isotopic distribution aptly reaffirms the effective species as the dicationic species, $[10-H_2]^{2+}$. HR MS (ESI): m/z Calcd for $C_{88}H_{116}N_2O_{20}^{2+}$ ($M-2PF_6$)²⁺: 760.4055 (100%), 760.9072 (95.2%).

14. 2D NOESY NMR spectrum of the [3]Rotaxane 11-H₂•2PF₆

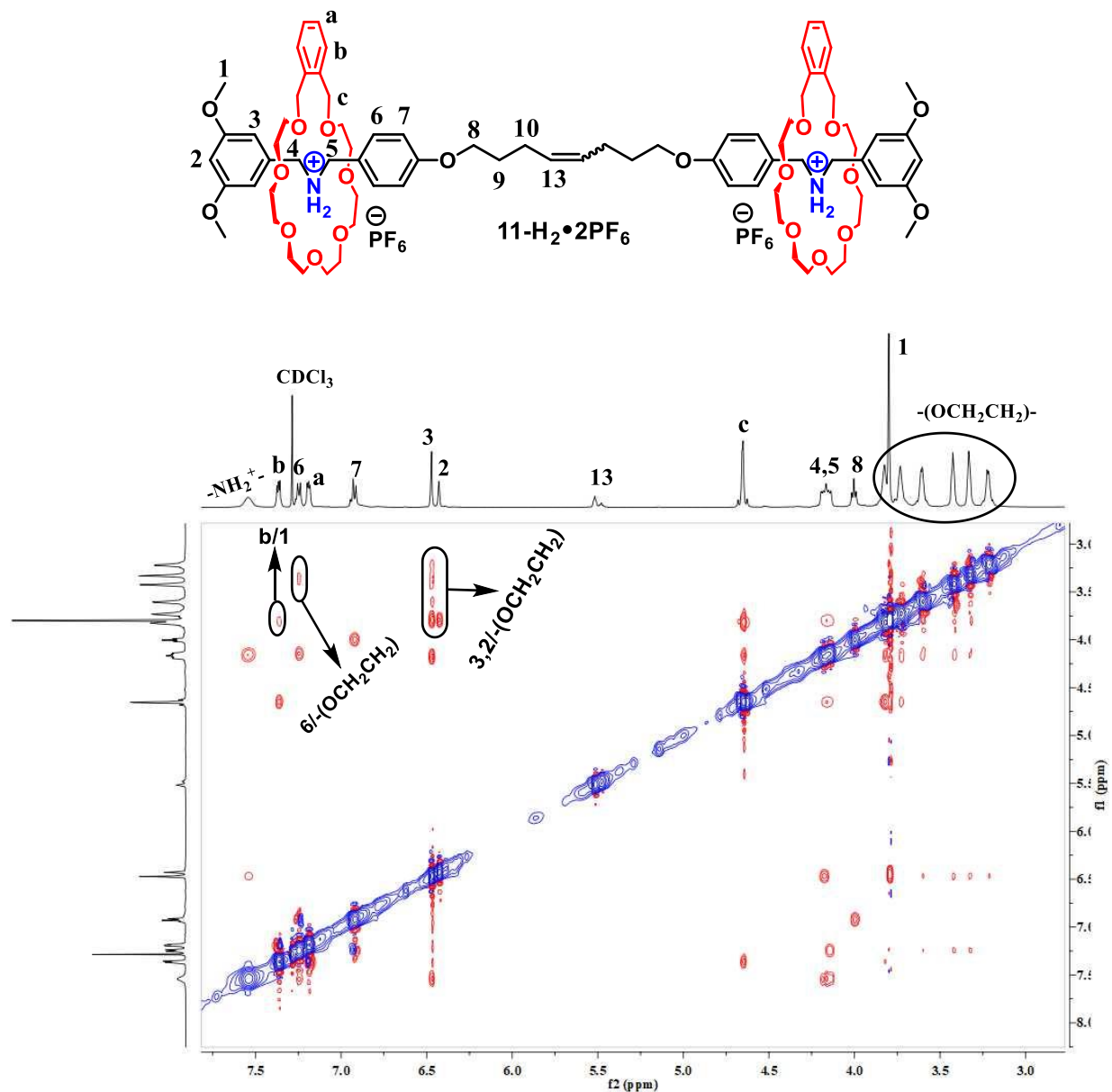


Figure S13. ¹H-¹H NOESY NMR (400 MHz, CDCl₃) of 11-H₂•2PF₆. In this case, strong correlations between (-OCH₂CH₂-) protons of X23C7 with the aromatic protons H₂/H₃/H₆ of the axle are observed. Additional correlations such as “H_b/H₁” unambiguously indicates the interlocked geometry.

15. HR ESI-MS spectrum of the [3]Rotaxane **11-H₂·2PF₆**

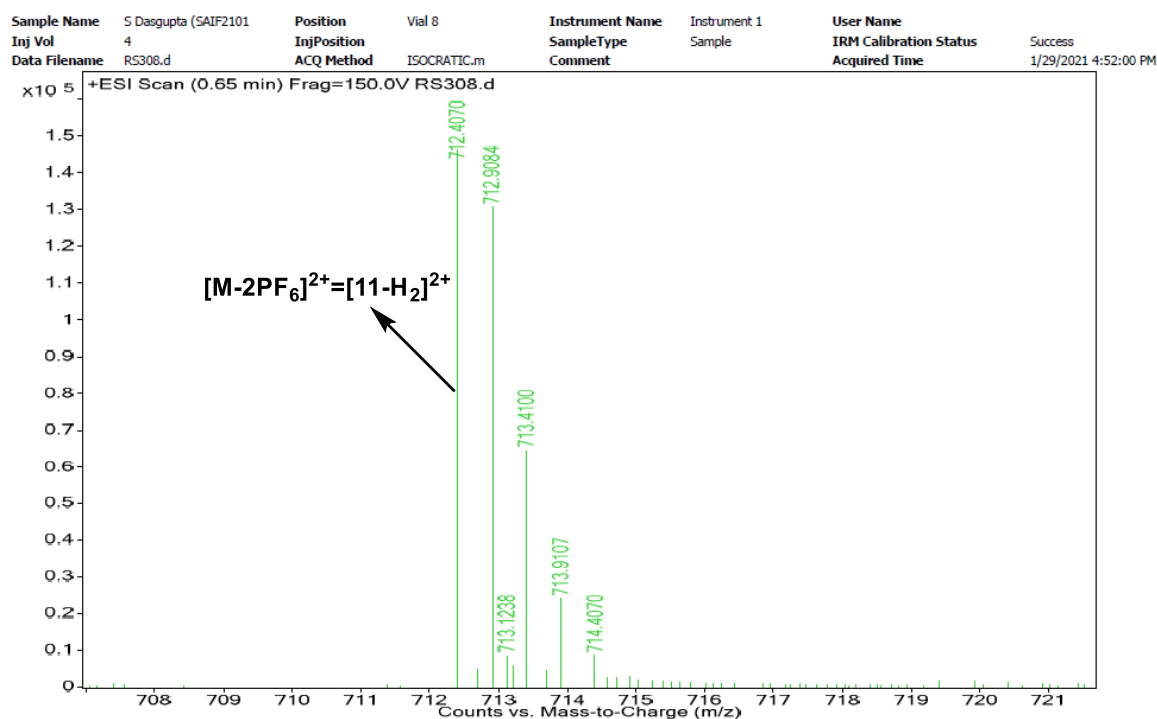
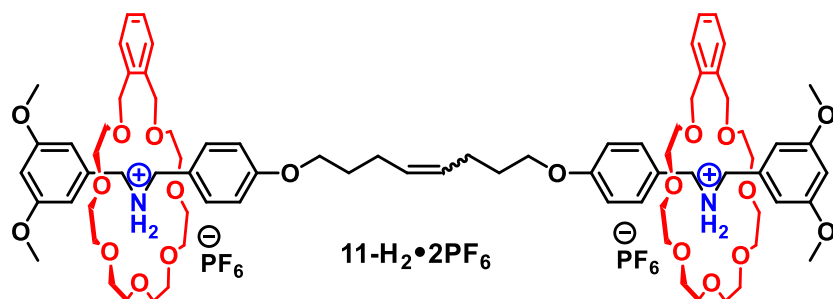
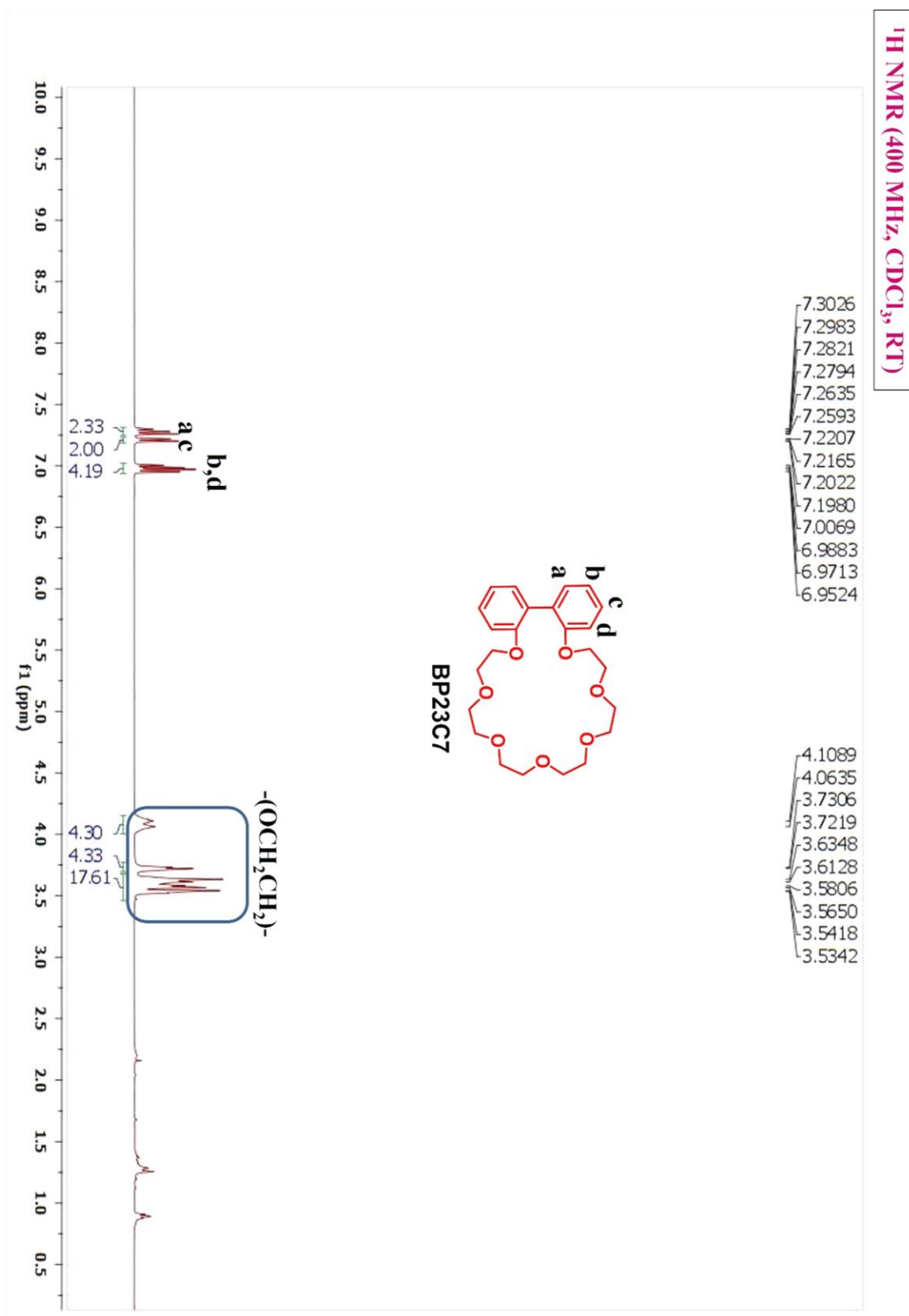
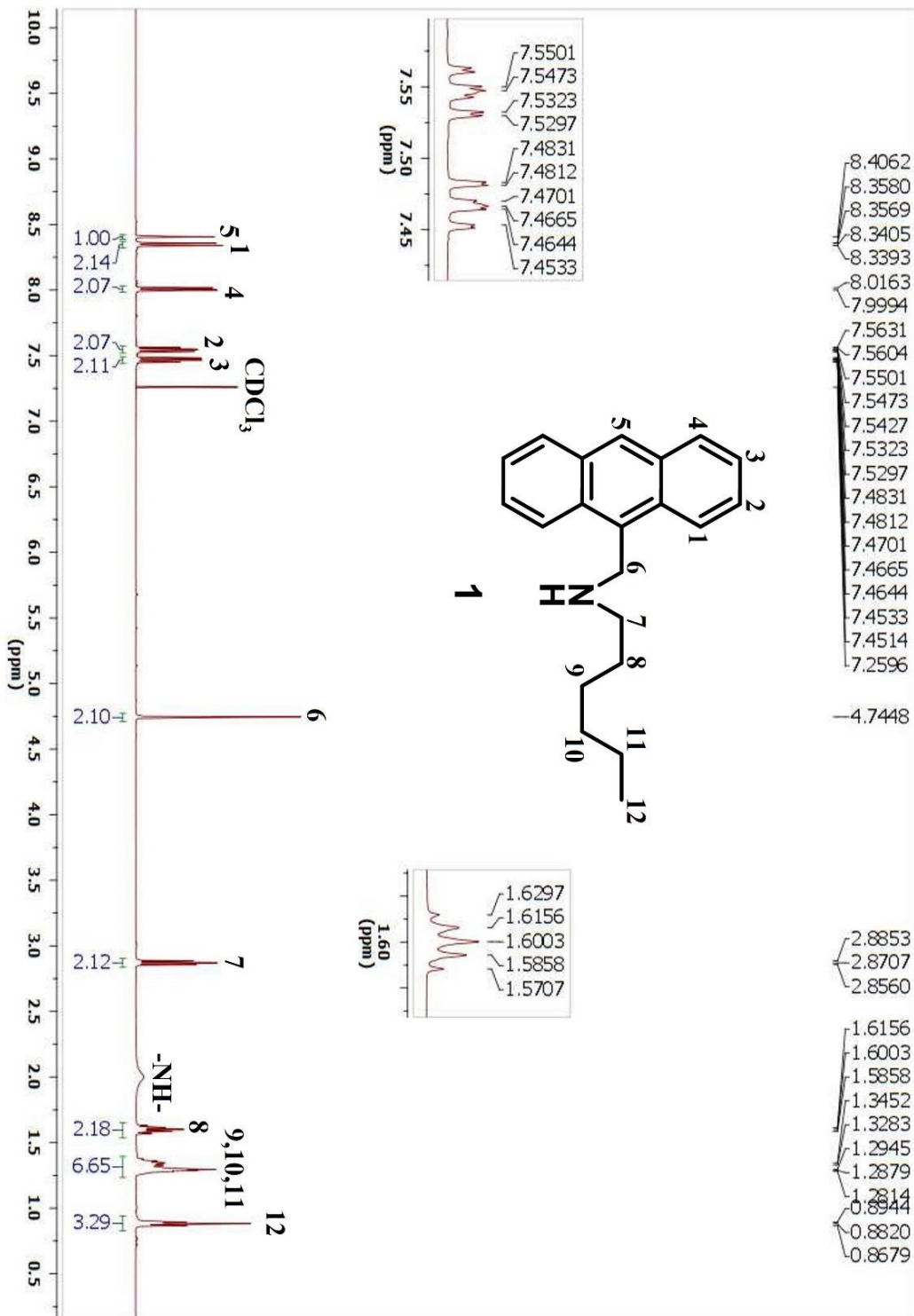


Figure S14. HR ESI-MS spectrum of the [3]rotaxane **11-H₂·2PF₆**. Peak at $m/z = 712.4070$ corresponds to the $[M-2PF_6]^{2+}$ species. Isotopic distribution clearly suggests the effective species as the dicationic species, $[11-H_2]^{2+}$. HR MS (ESI): m/z Calcd for $C_{80}H_{116}N_2O_{20}^{2+}$ ($M-2PF_6$)²⁺: 712.4055 (100%), 712.9072 (86.5%).

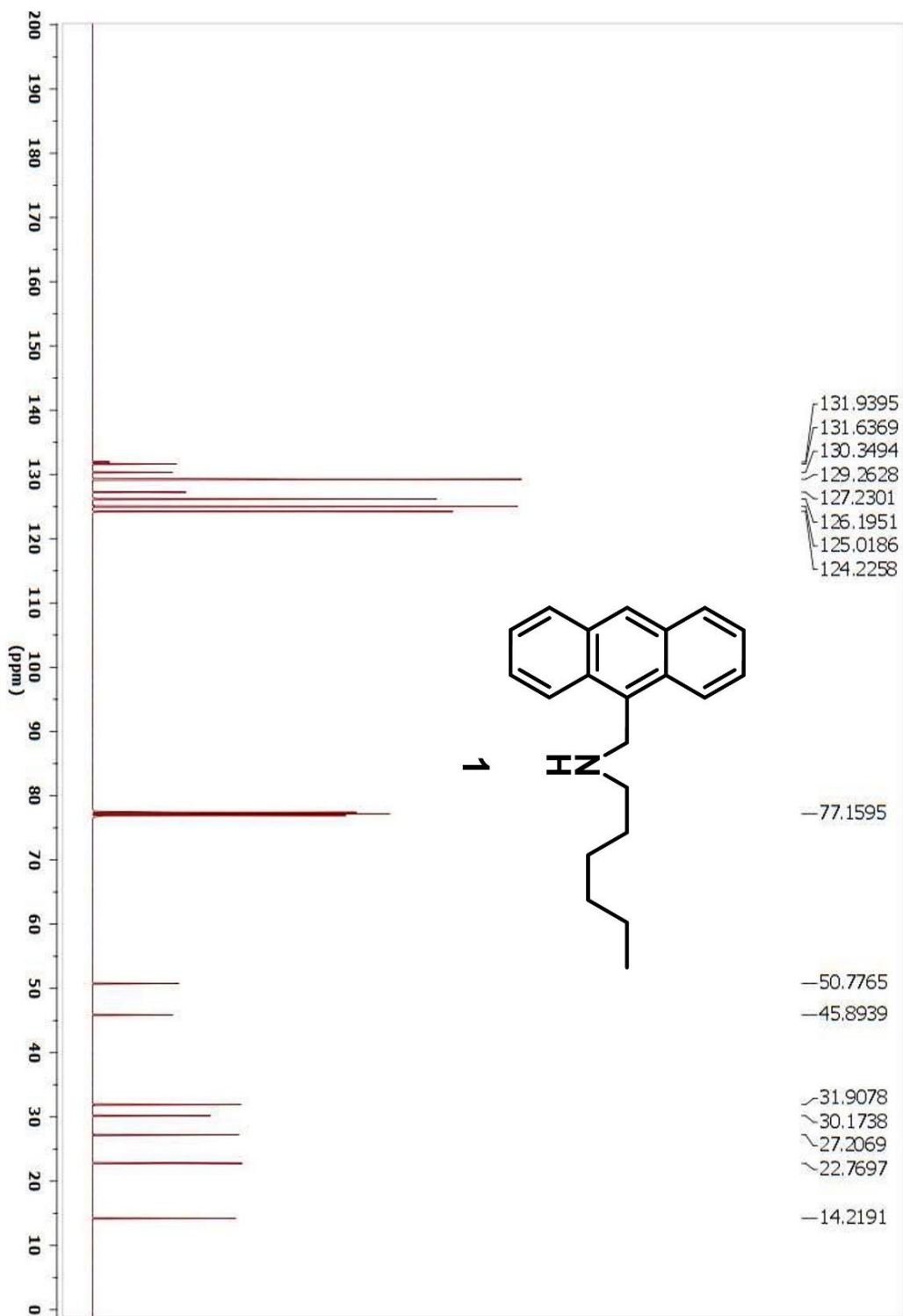
16. ^1H NMR & ^{13}C NMR spectra of all synthesized compounds



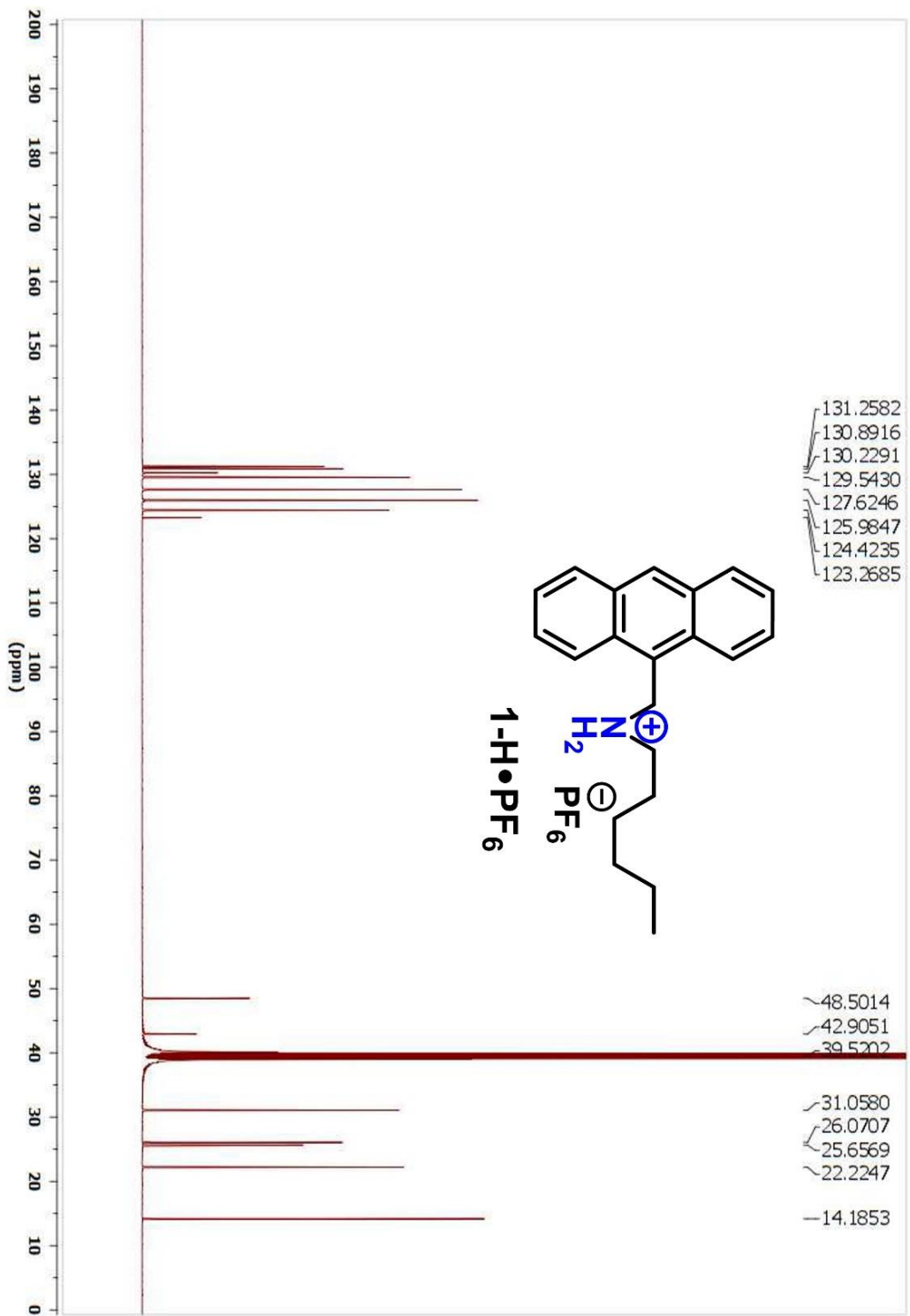
¹H NMR (500 MHz, CDCl₃, RT)



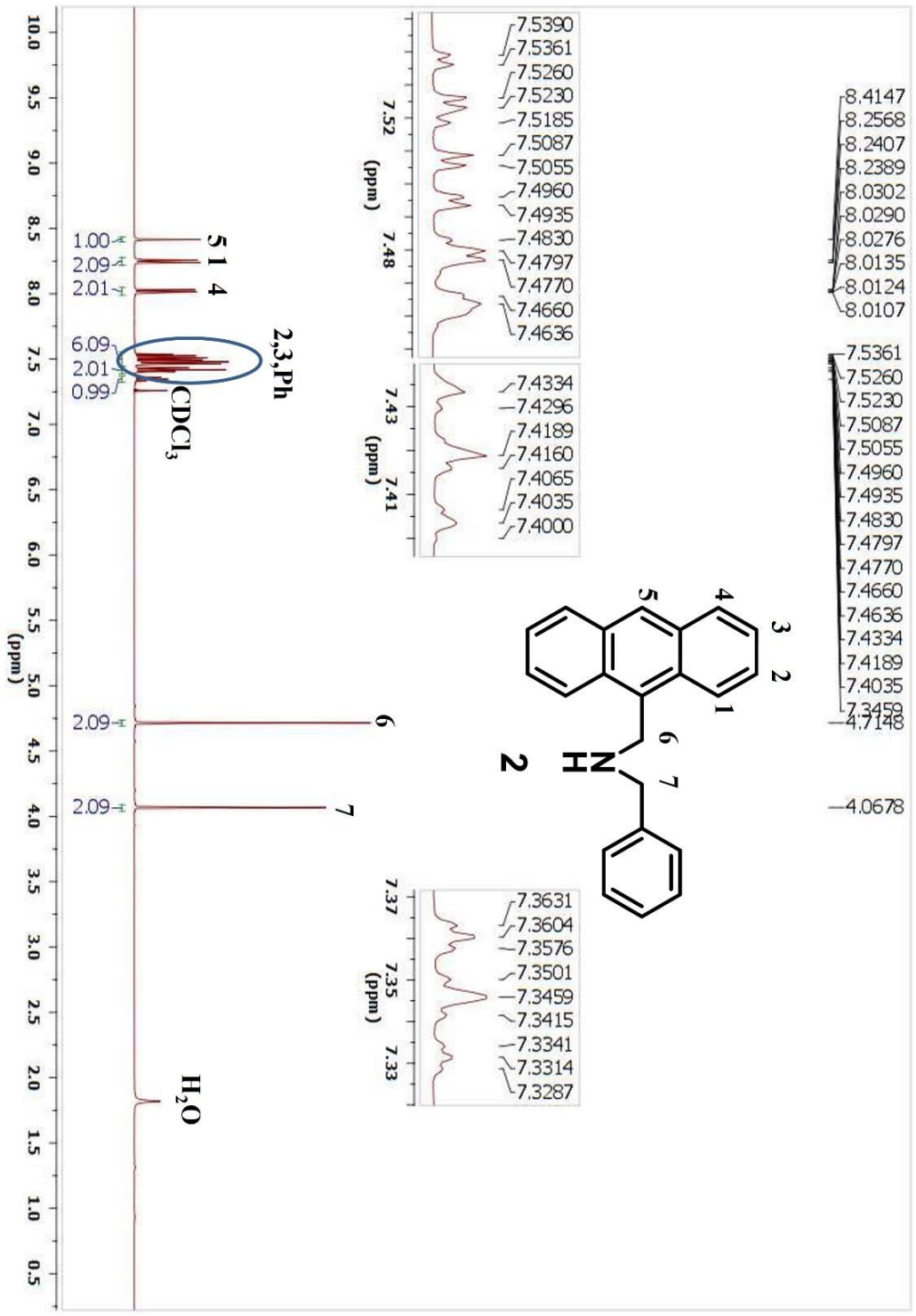
$^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3 , RT)



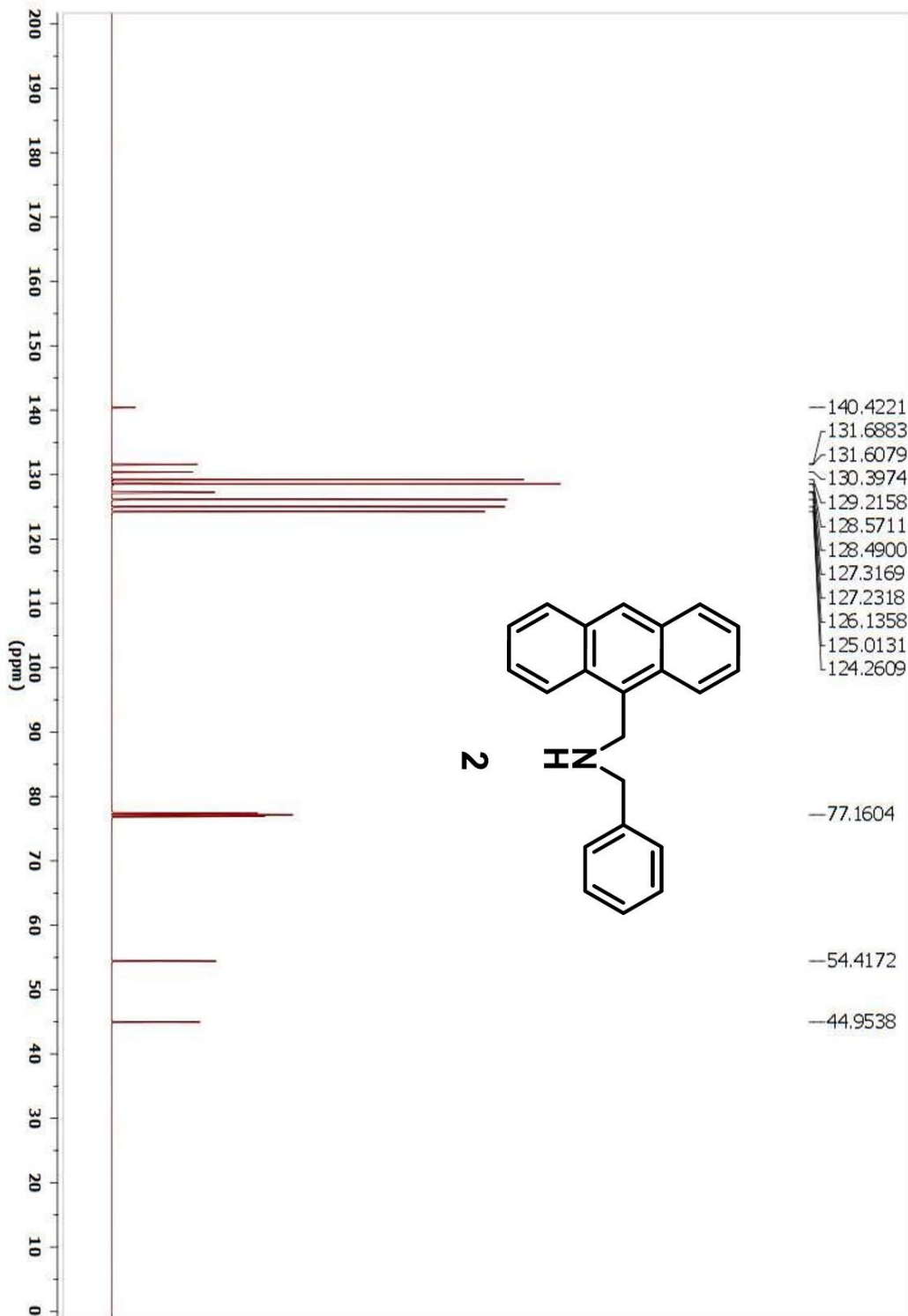
$^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, DMSO- d_6 , RT)



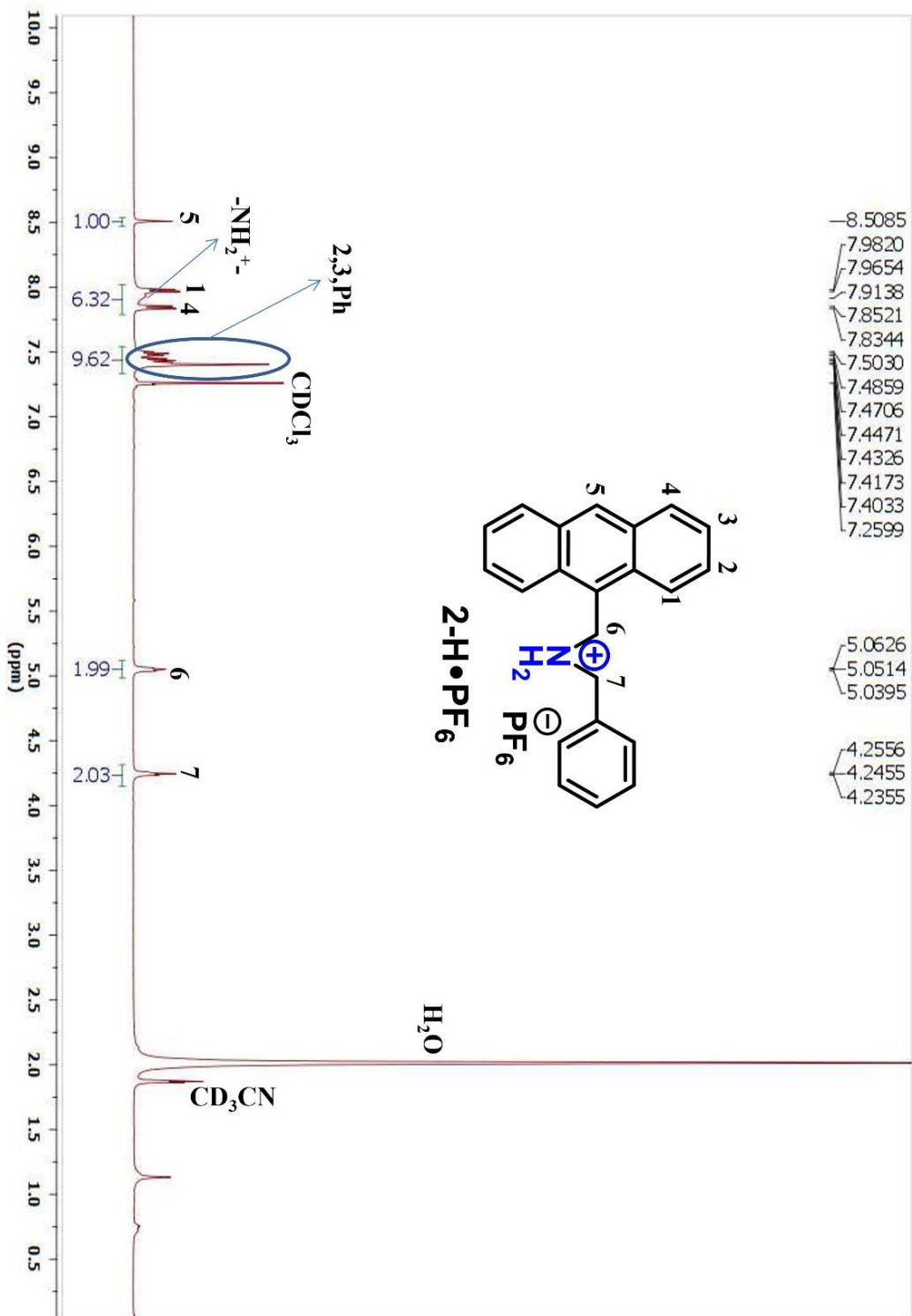
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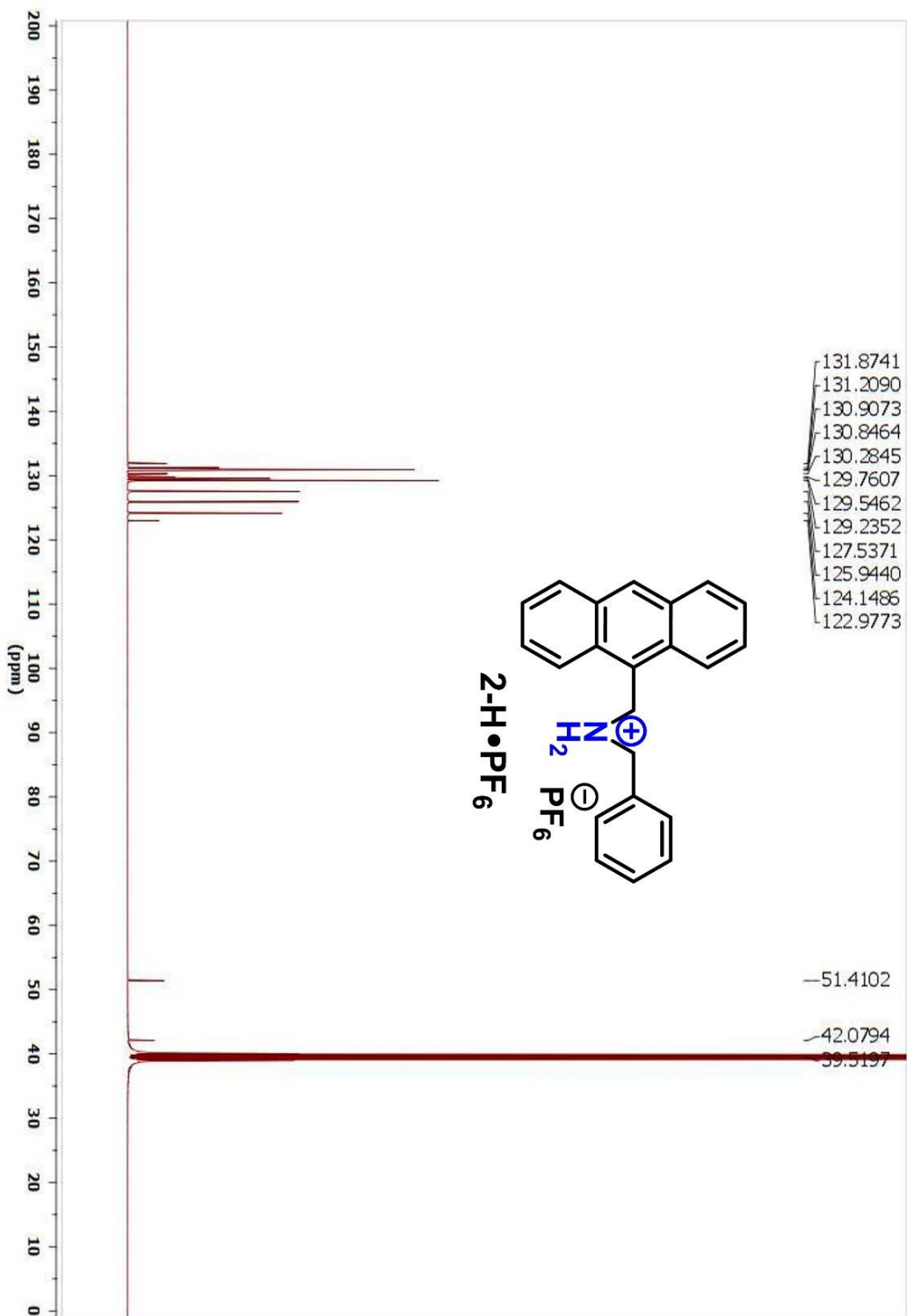
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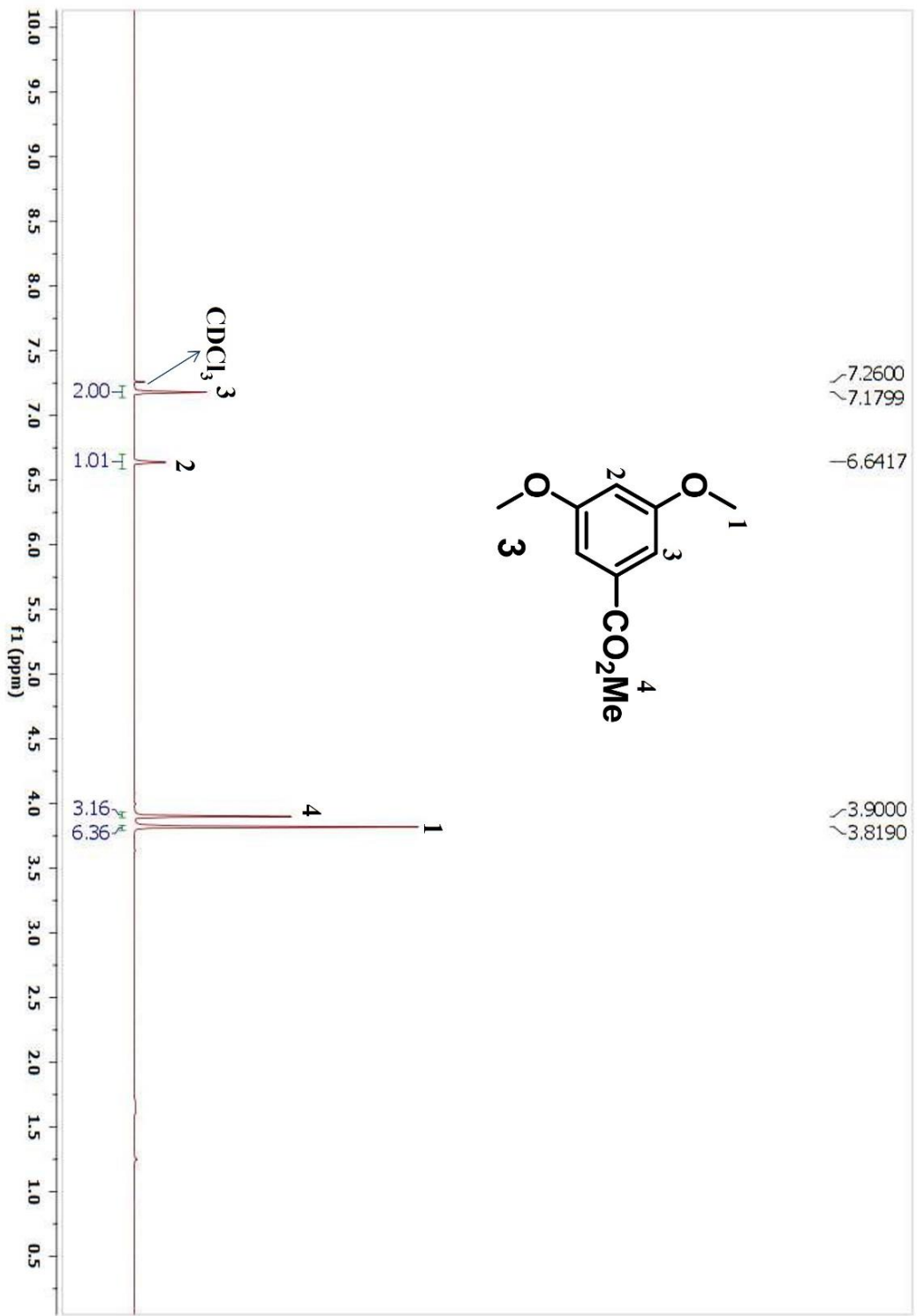
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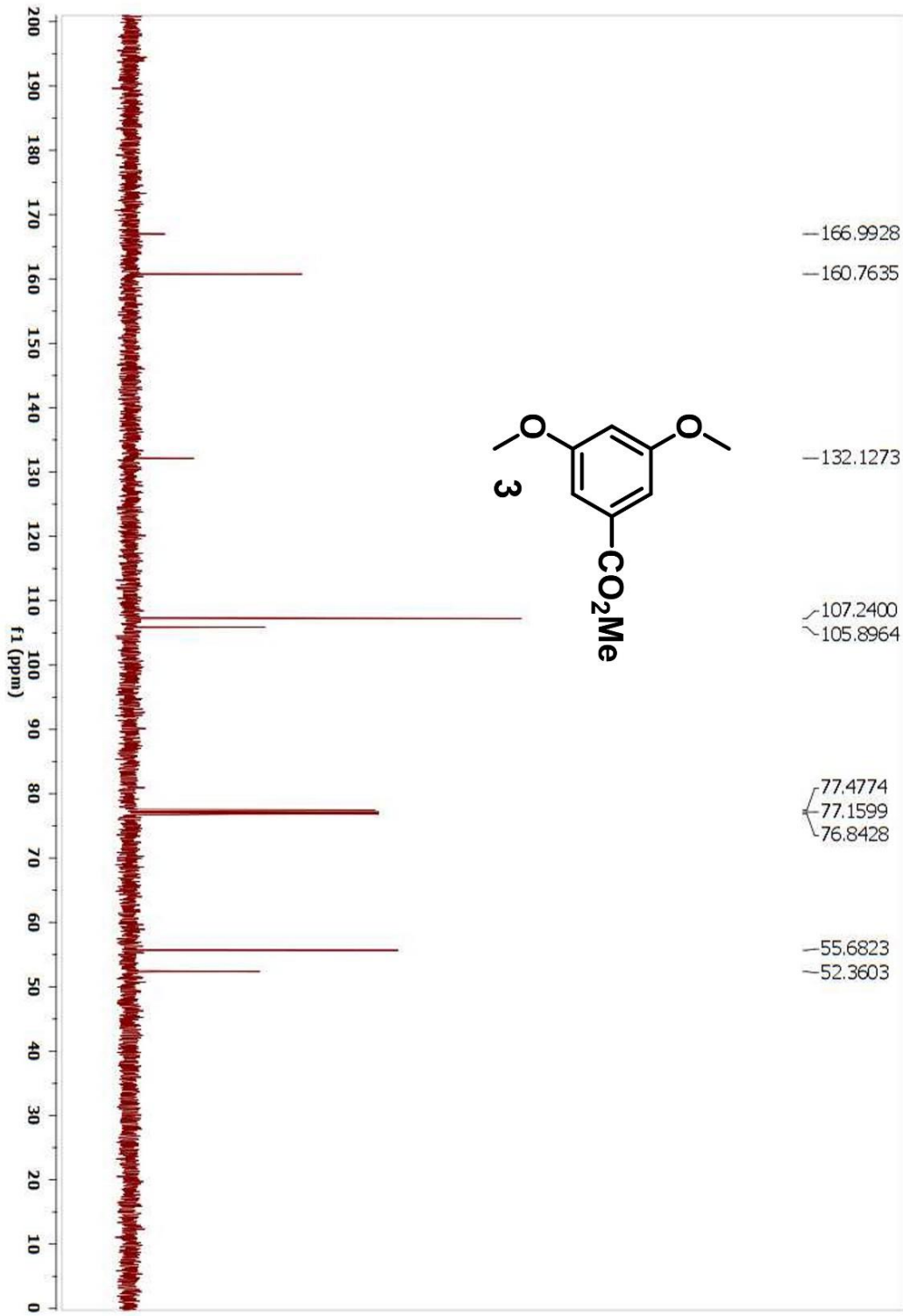
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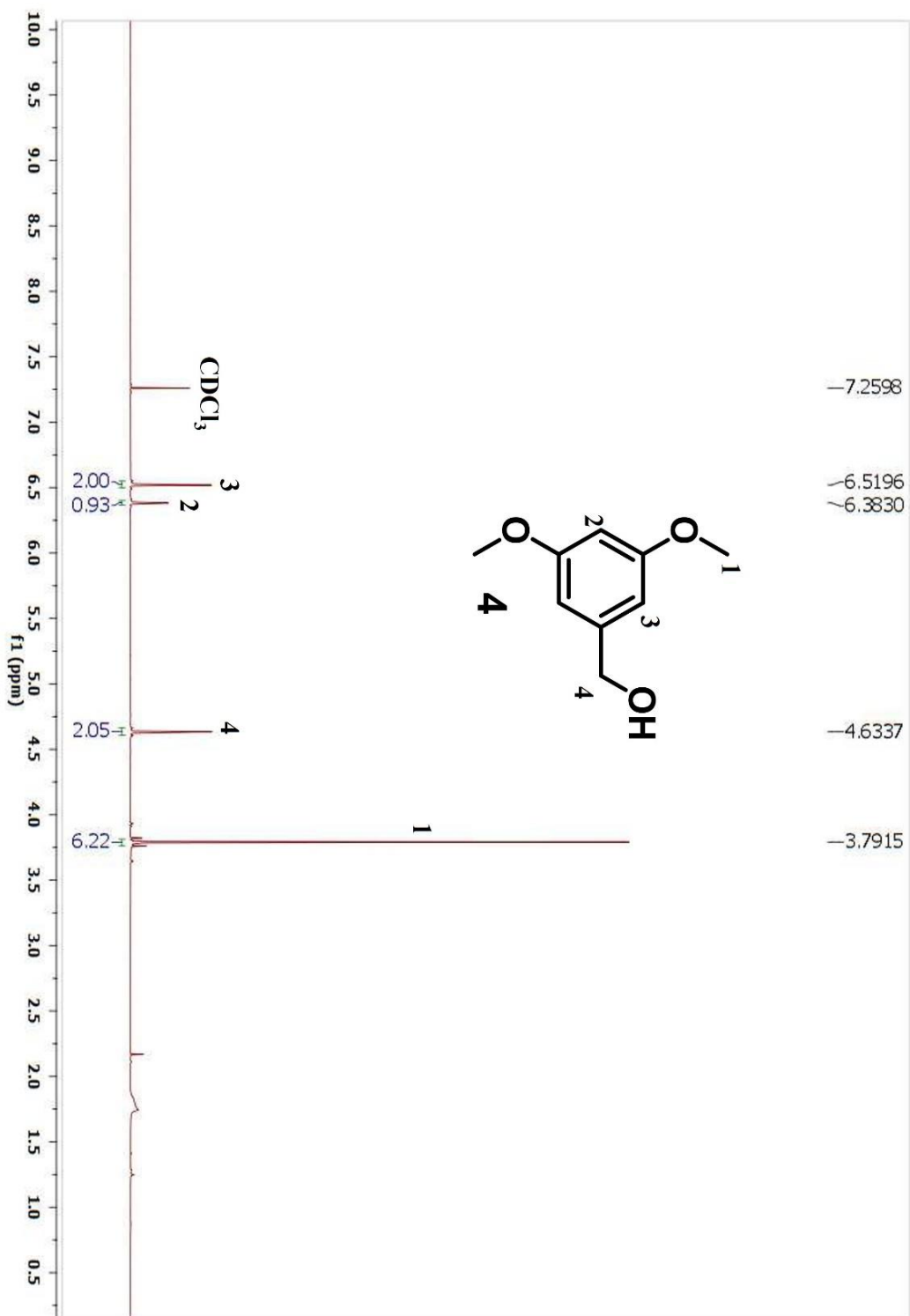
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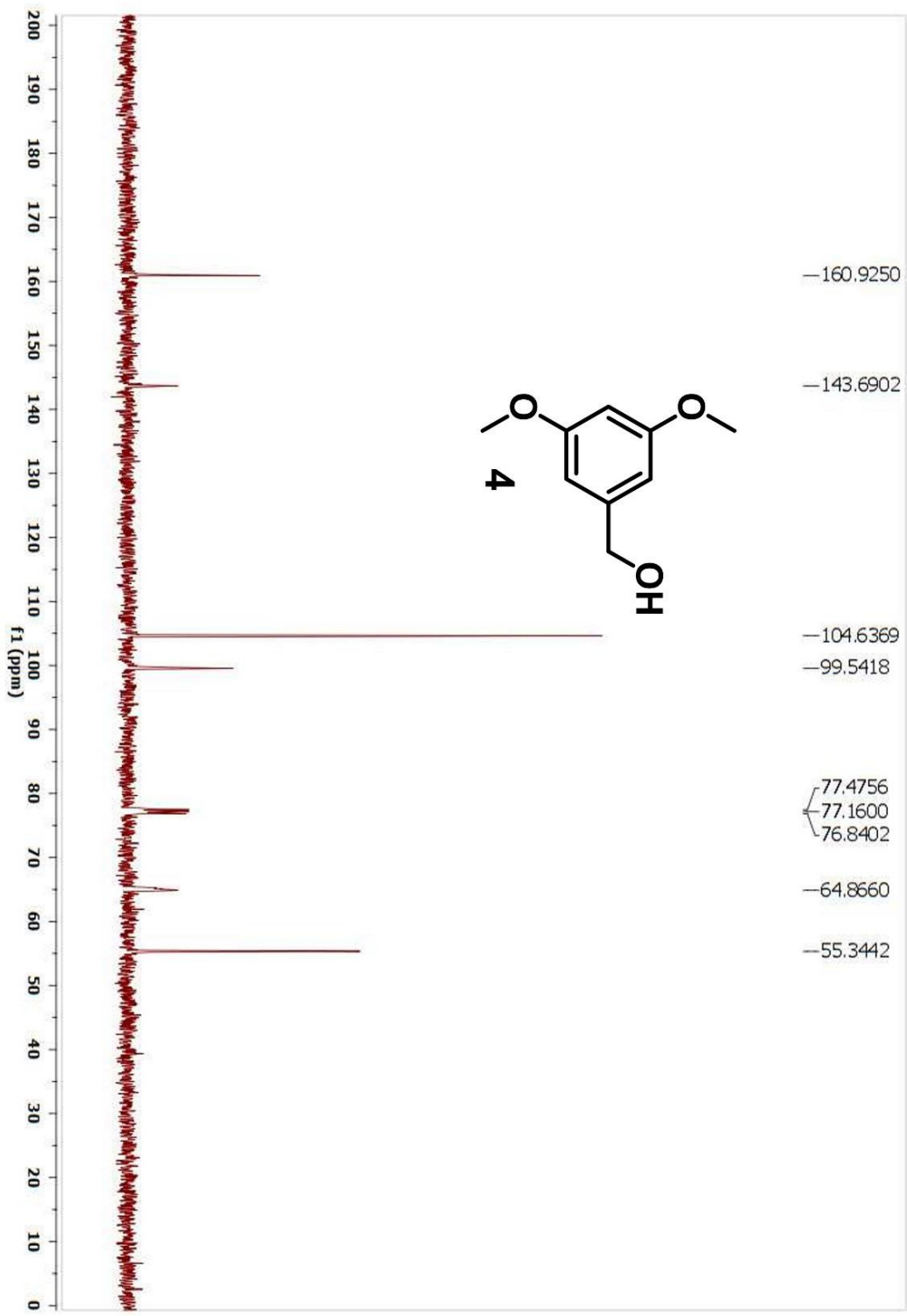
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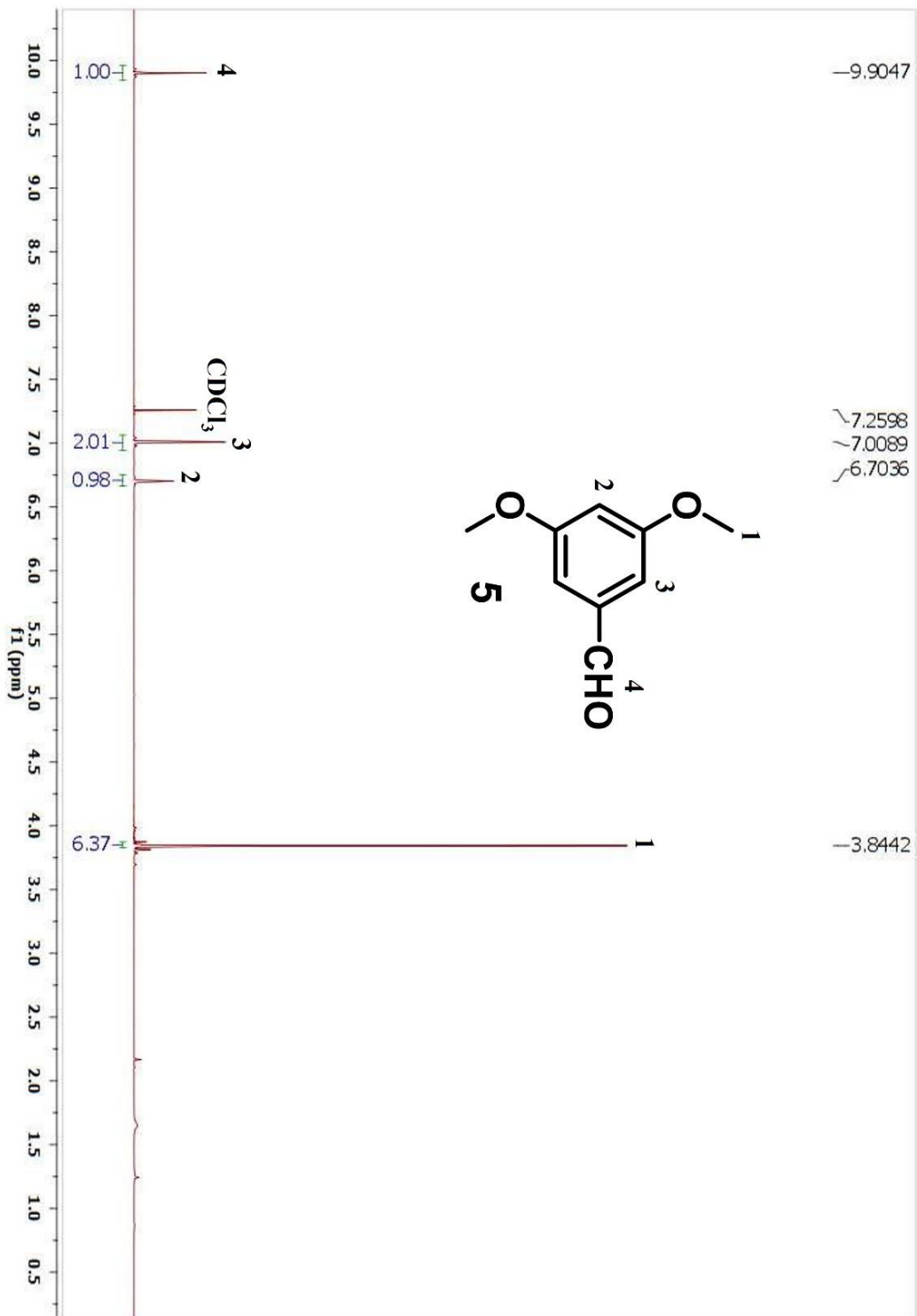
¹H NMR (500 MHz, CDCl₃, RT)



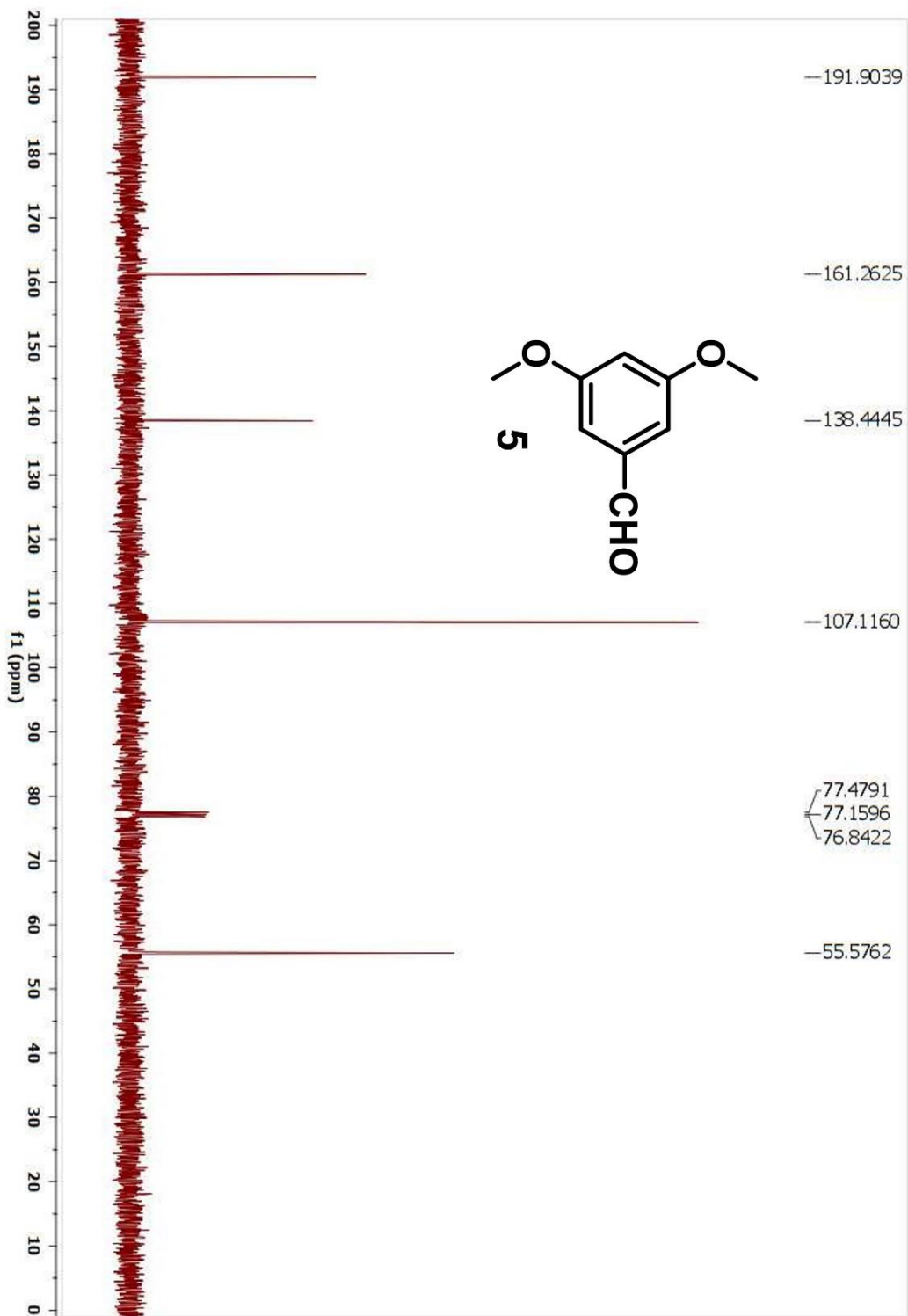
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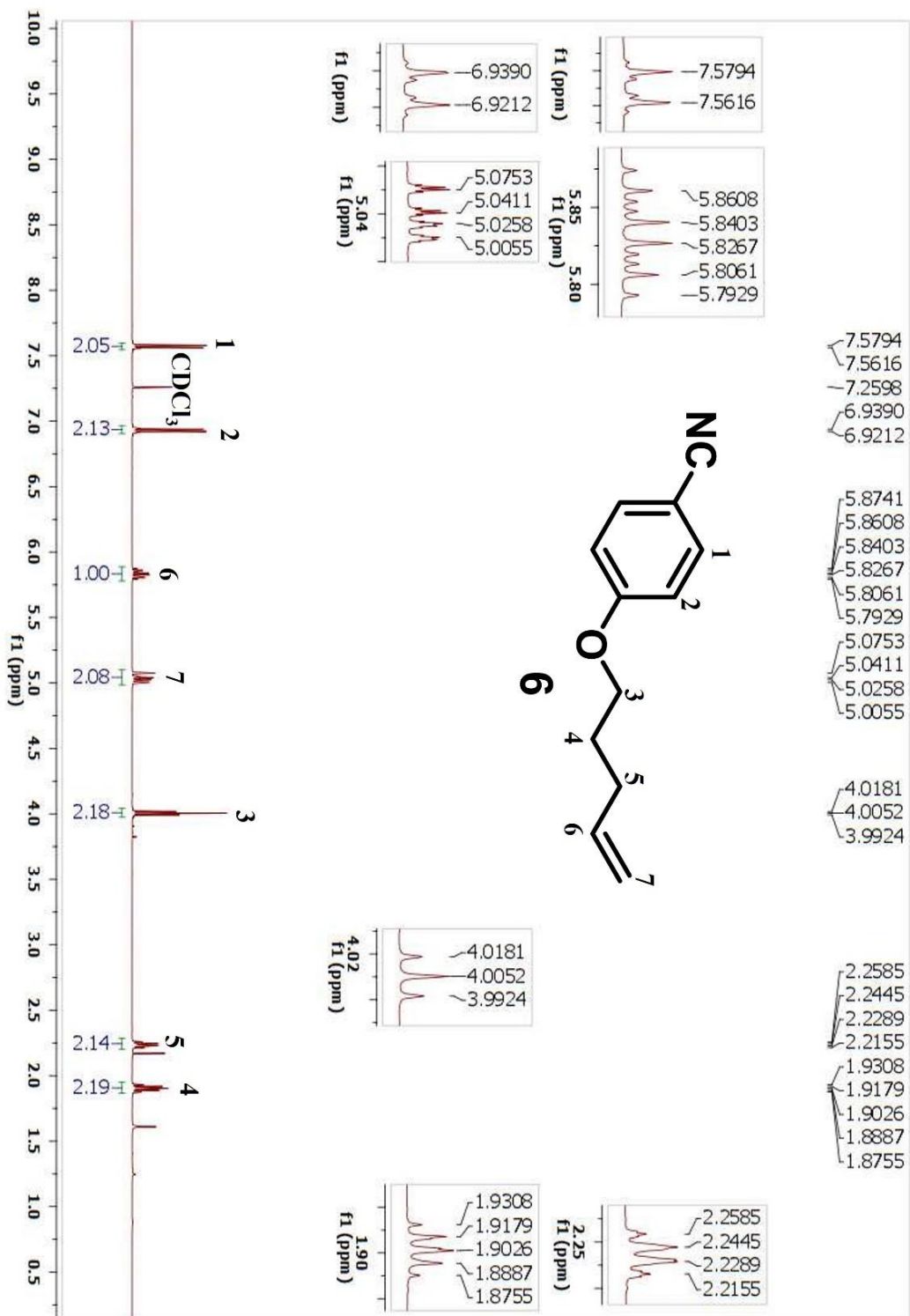
¹H NMR (500 MHz, CDCl₃, RT)



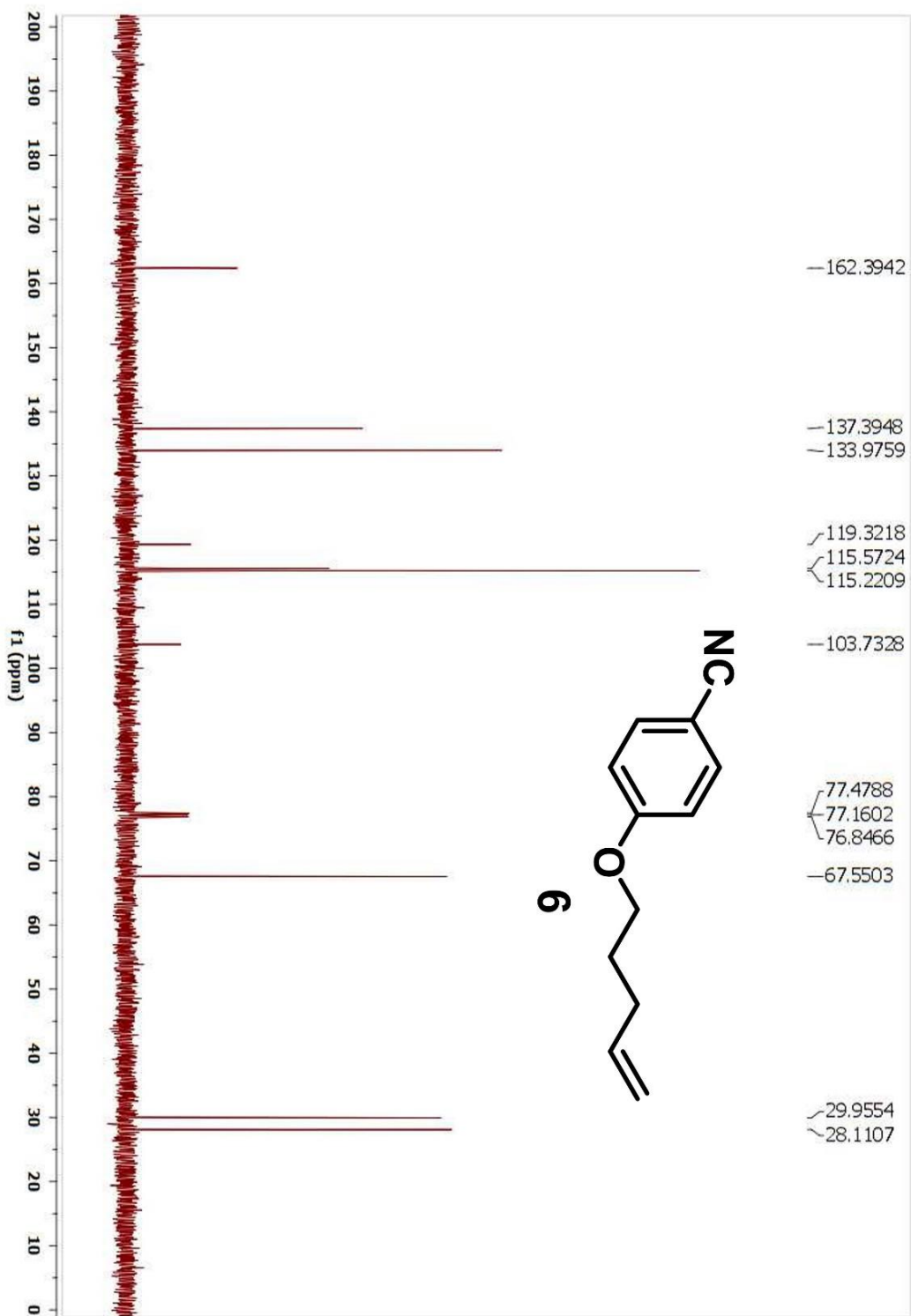
$^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , RT)



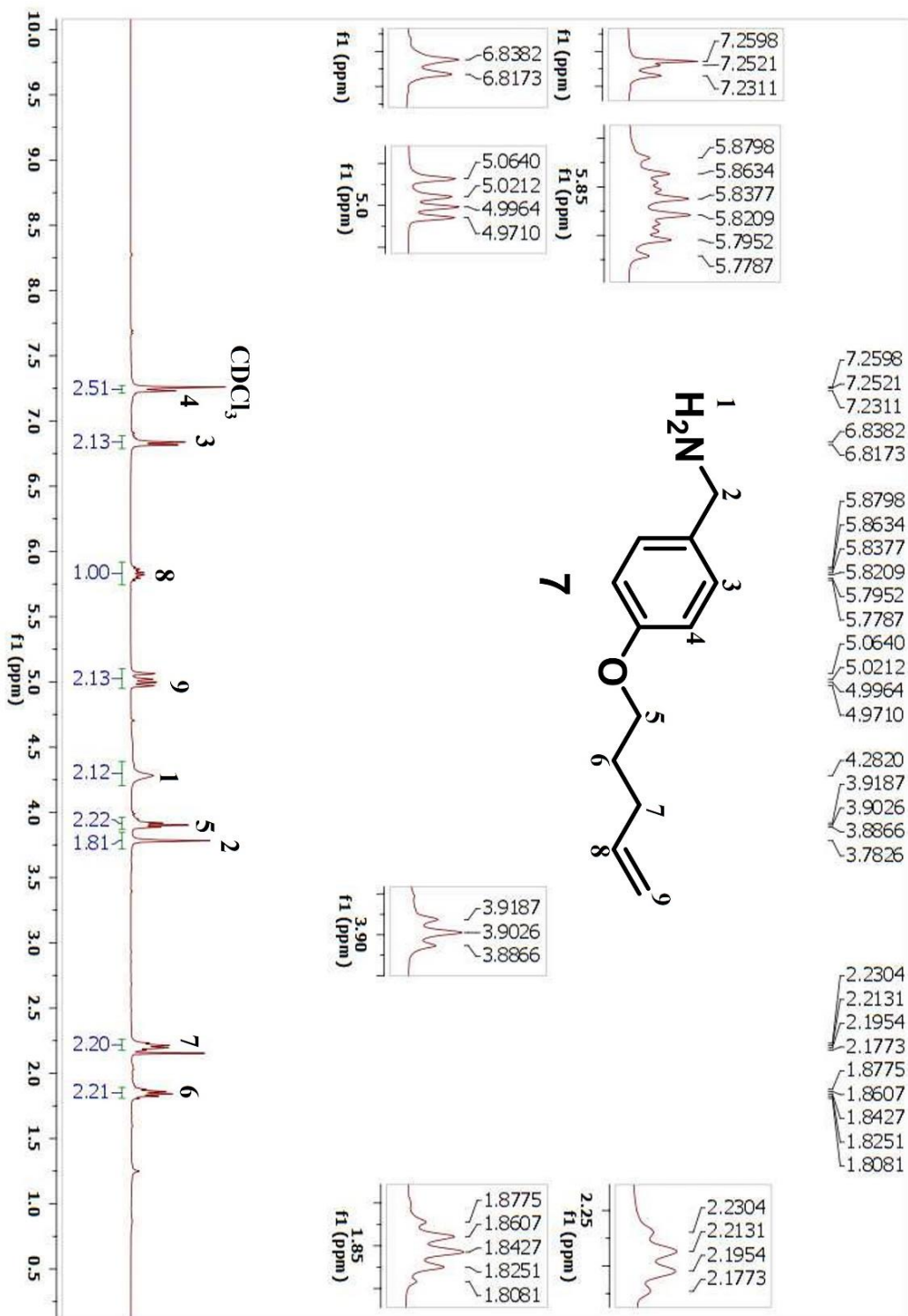
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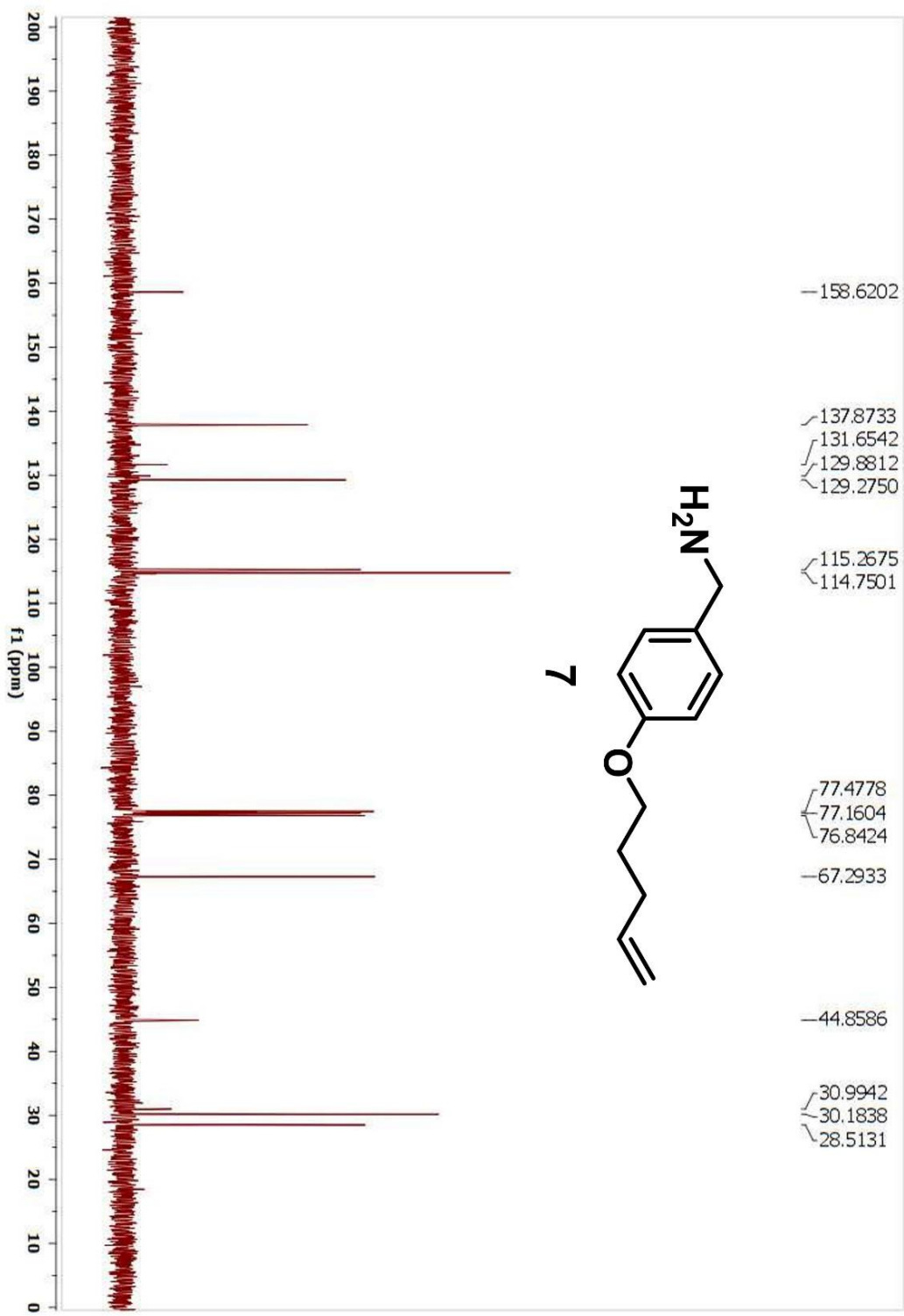
$^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , RT)



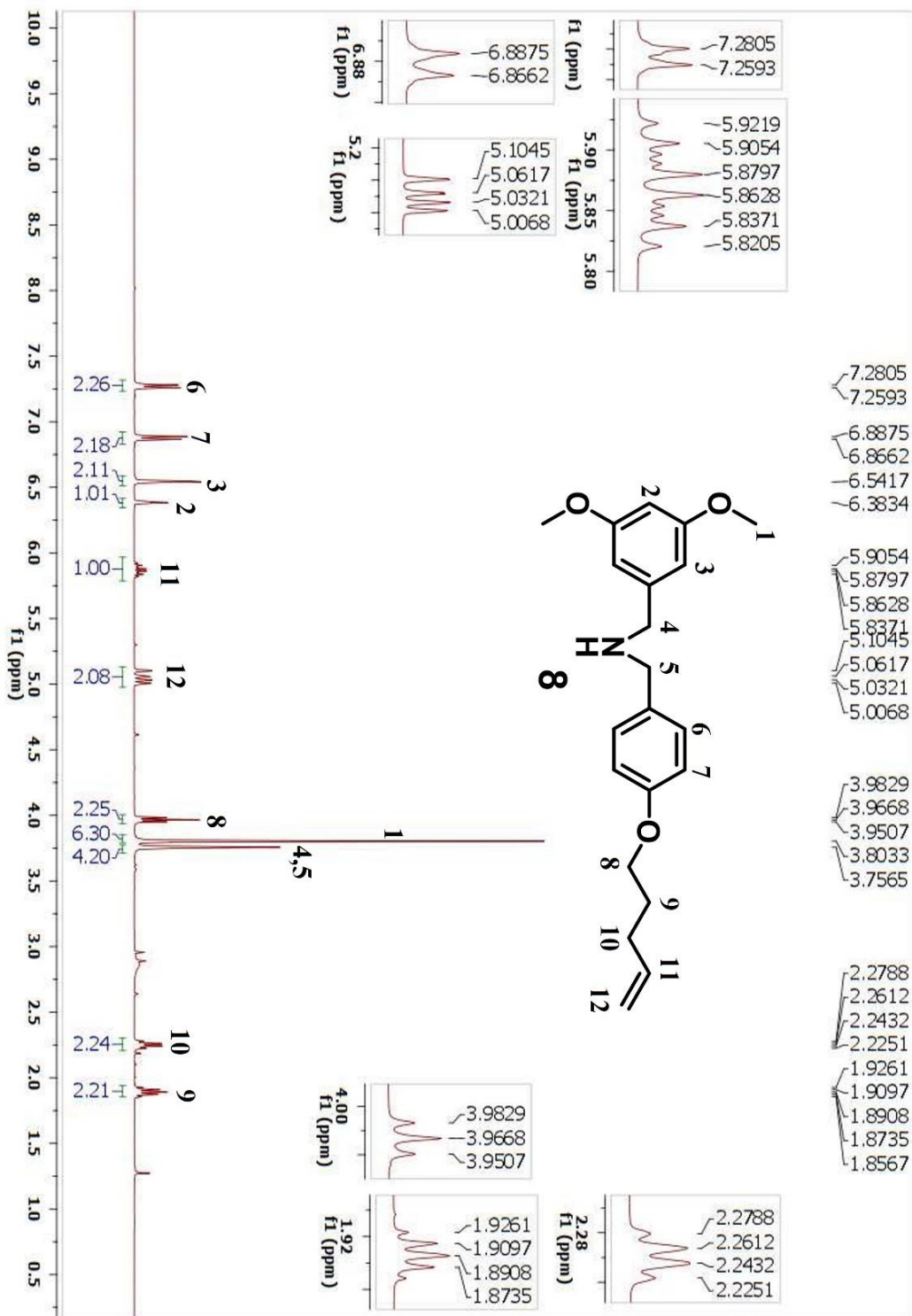
¹H NMR (400 MHz, CDCl₃, RT)



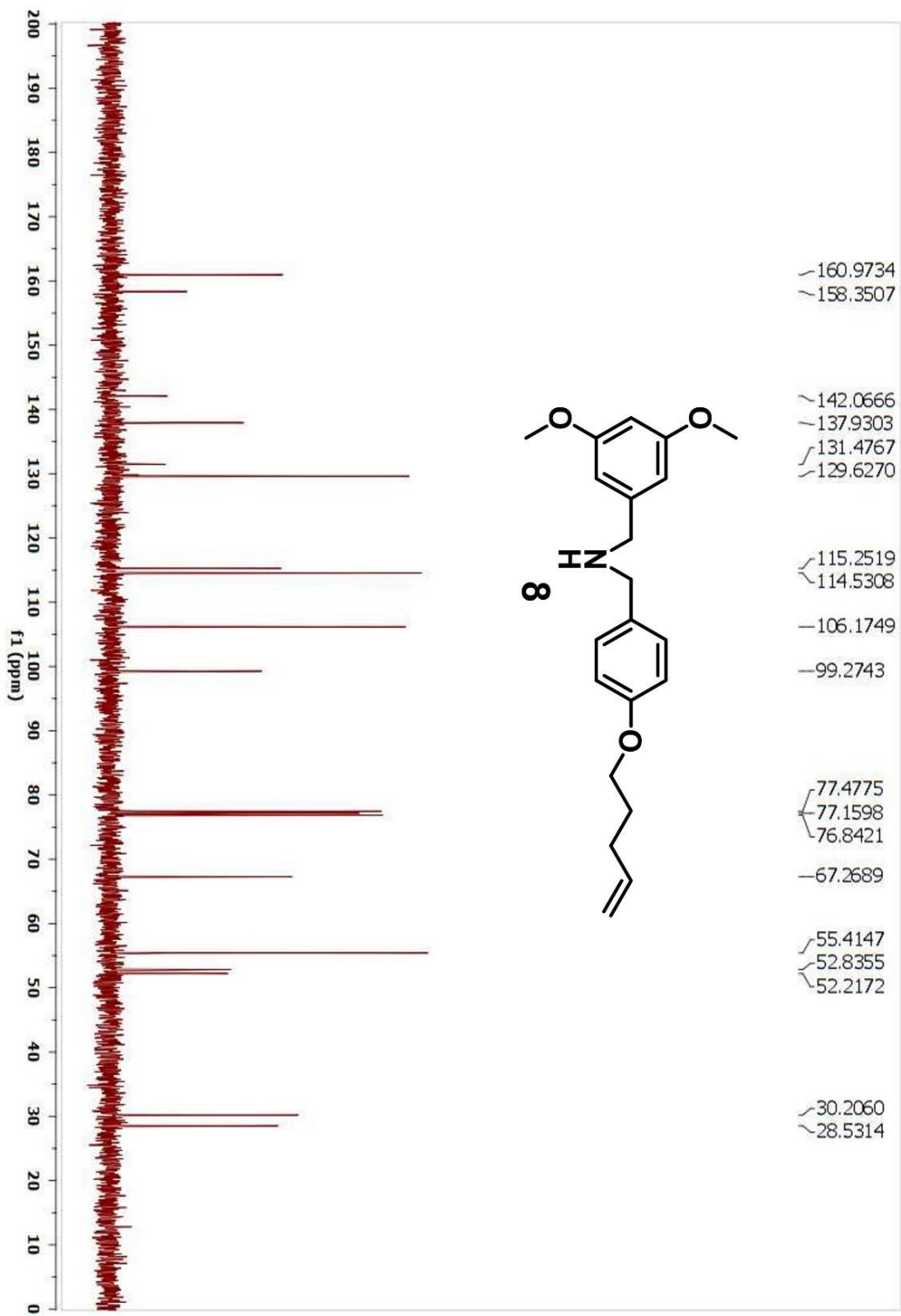
$^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , RT)



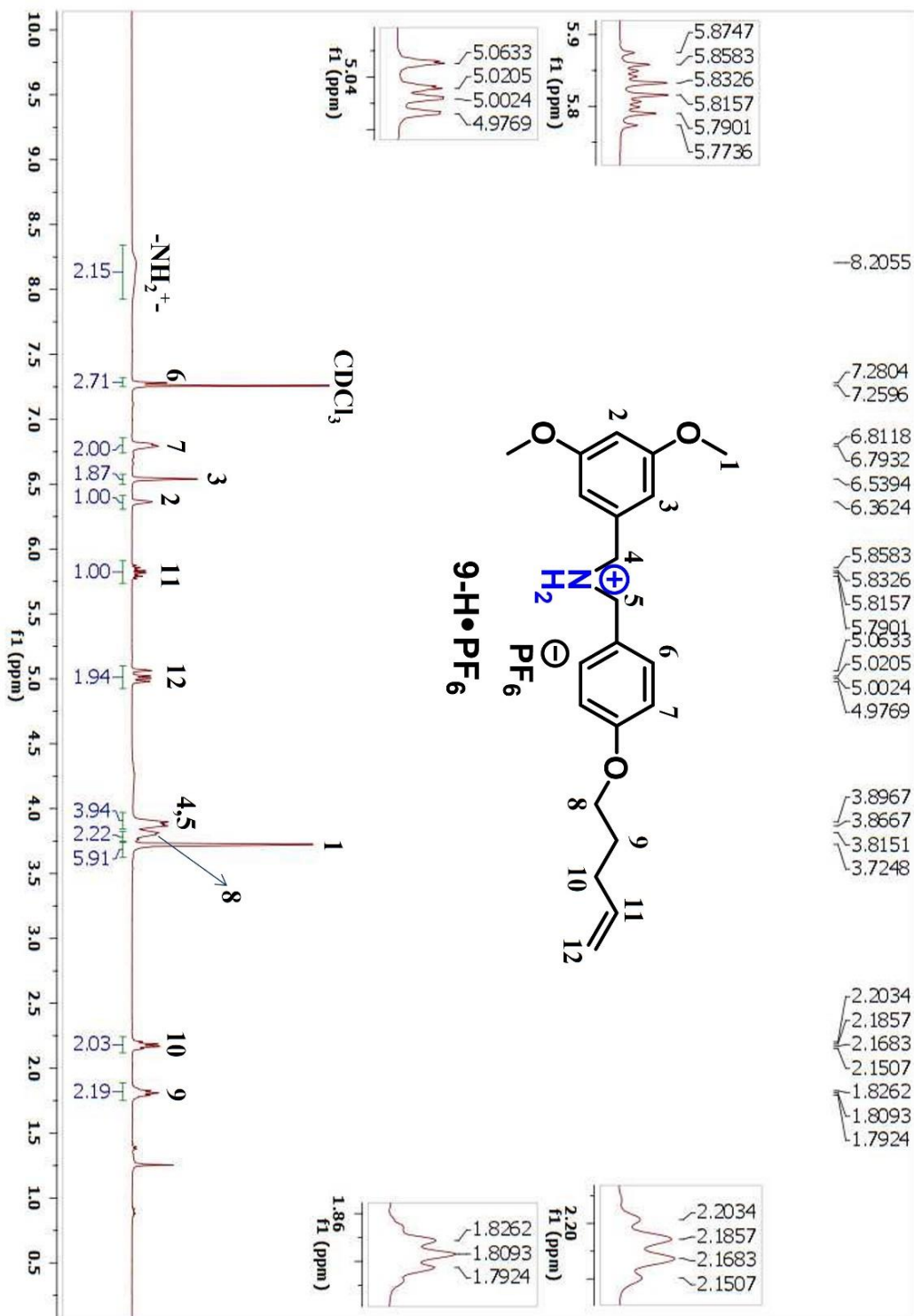
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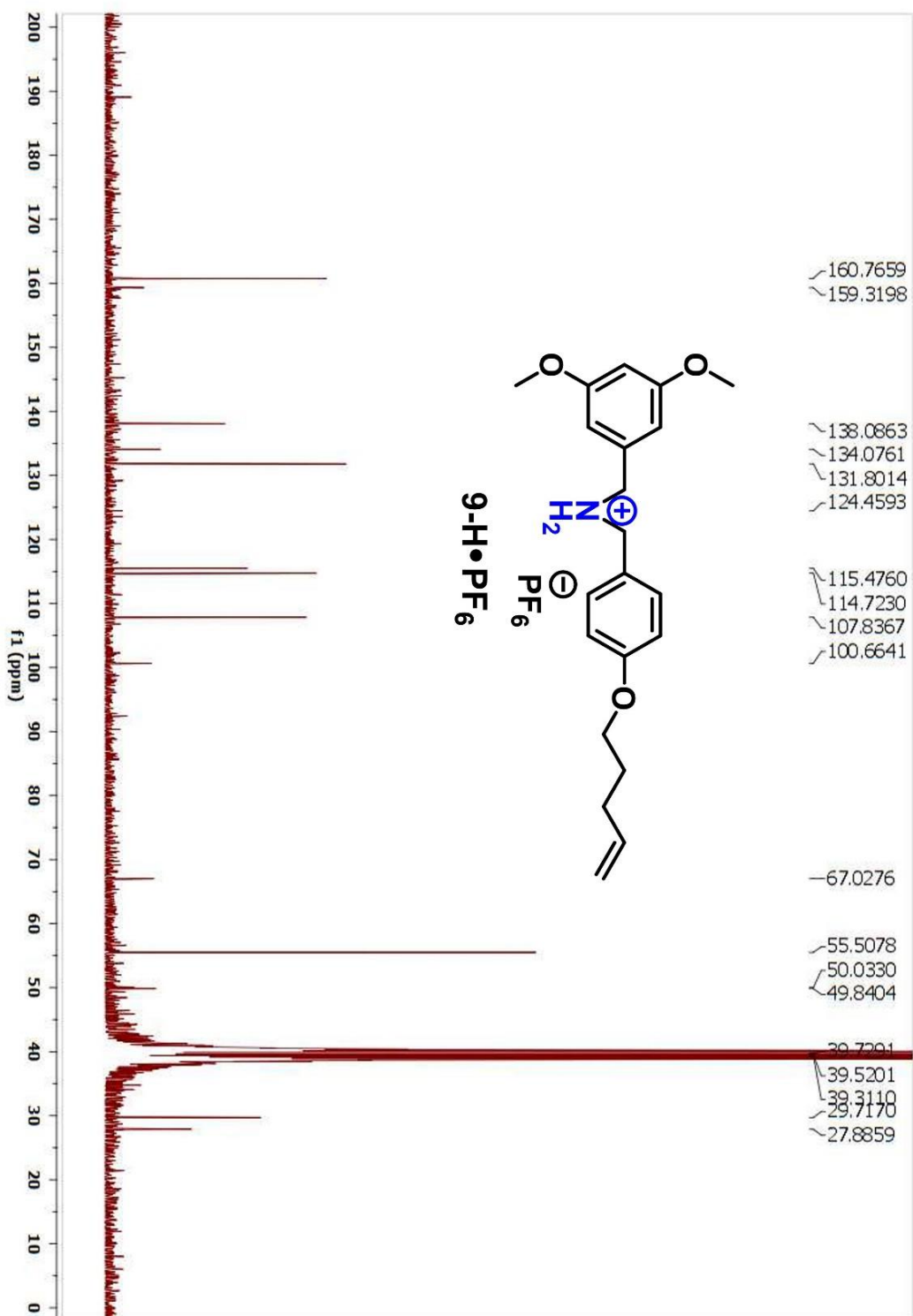
$^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , RT)



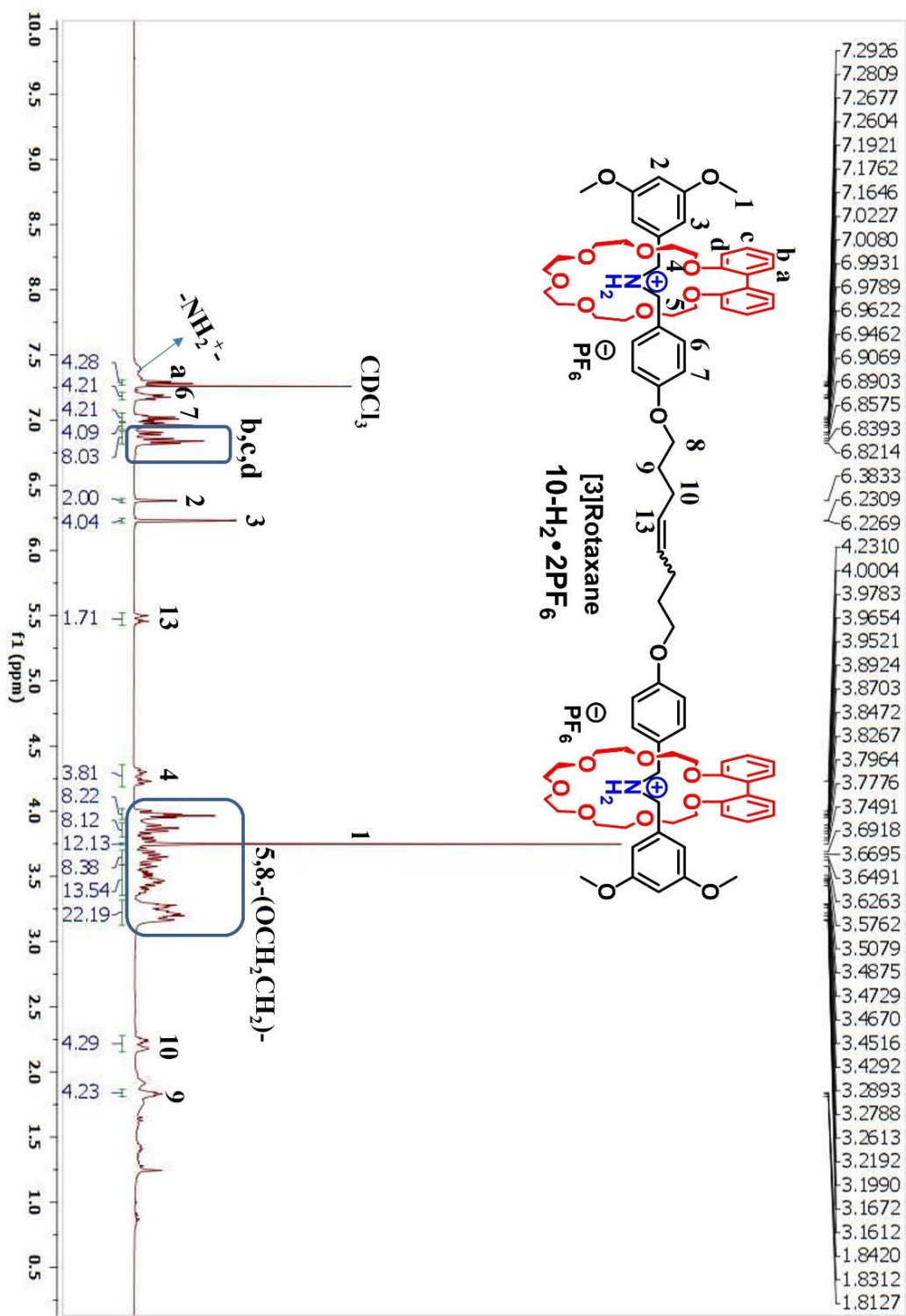
¹H NMR (400 MHz, CDCl₃, RT)



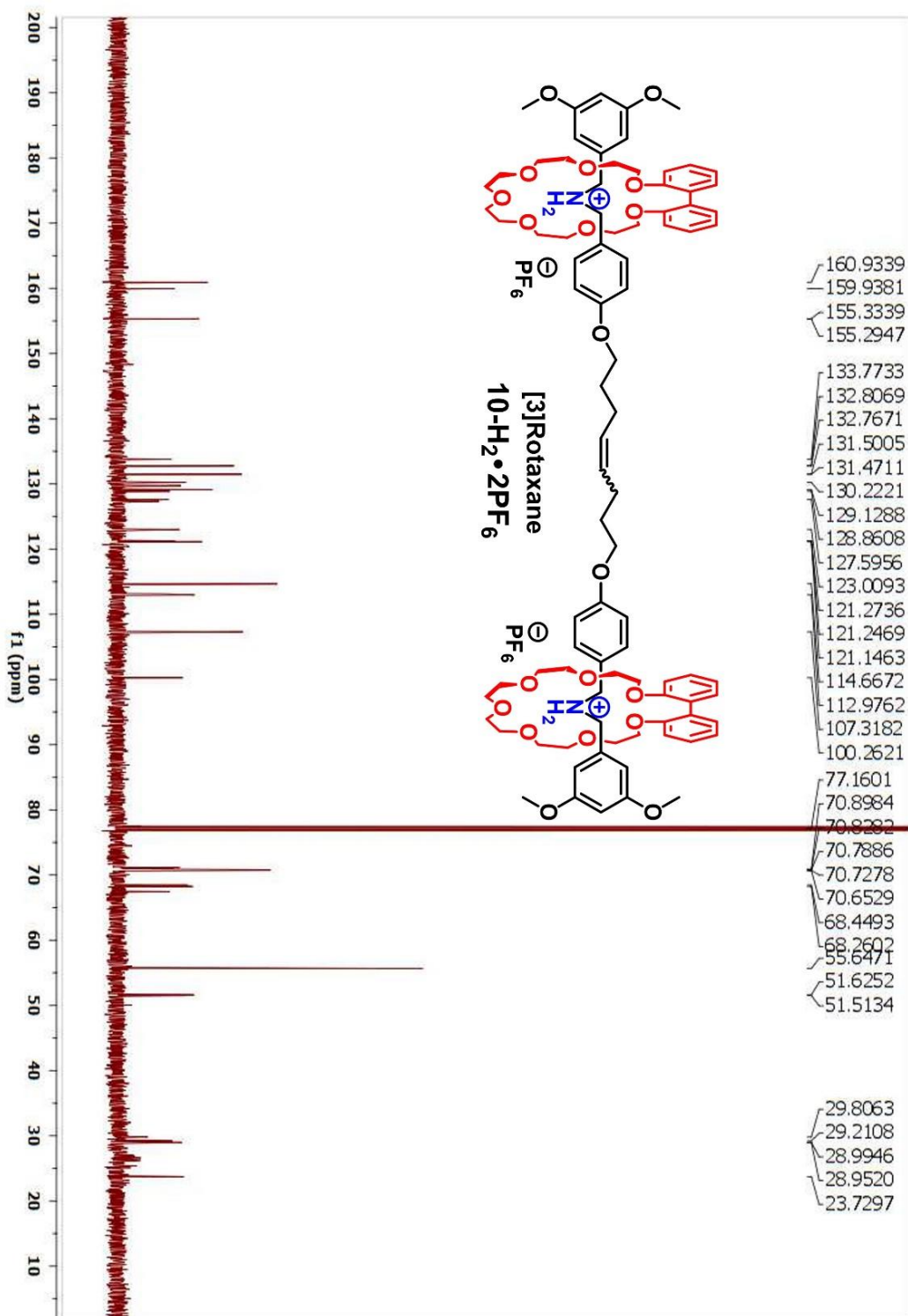
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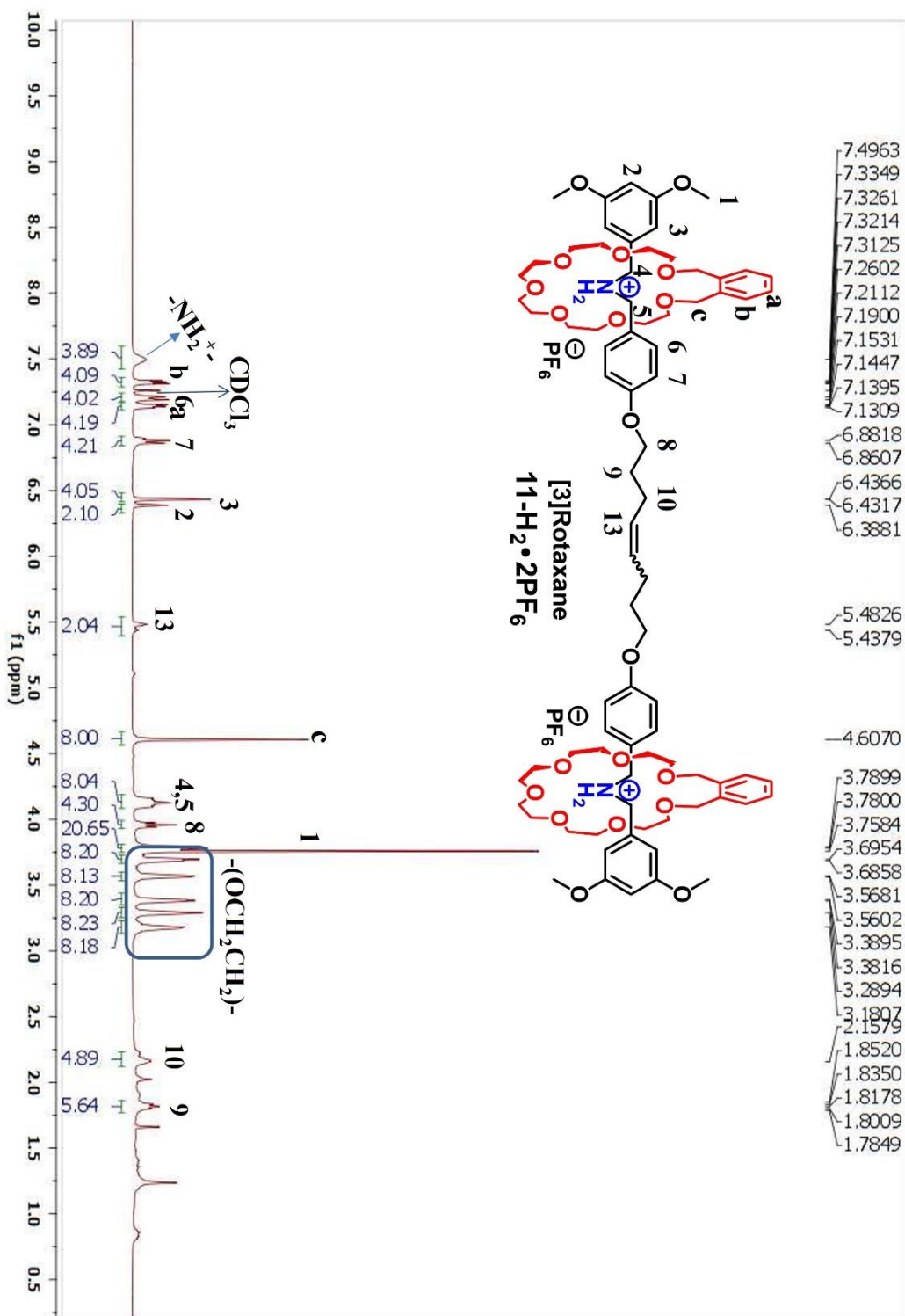
¹H NMR (400 MHz, CDCl₃, RT)



$^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3 , RT)



¹H NMR (400 MHz, CDCl₃, RT)



$^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3 , RT)

