

## ESI

# **LaAeAl<sub>3</sub>S<sub>7</sub> (Ae = Ca, Sr): Cairo Pentagonal Layered thioaluminates Achieving Well-balance between Strong Second Harmonic Generation Response and Wide Bandgap**

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**Table S1.** Dimensional distribution of  $\text{AlS}_4$  units in the known thioaluminates.

Compound	Space group	Unit	Dimensionality
$(\text{Ag}_{0.8}\text{Cu}_{3.2})\text{Al}_4\text{S}_8$	<i>I-42d</i>	$\text{AlS}_4$	3D
$(\text{Ag}_{3.6}\text{Cu}_{0.4})\text{Al}_4\text{S}_8$	<i>I-42d</i>	$\text{AlS}_4$	3D
$\text{Ln}_3\text{AlMS}_7$ (Ln=Y, Lanthanide series; M=Fe, Co, Ni, Zn, Mn, Cr, V, Ti)	<i>P6_3</i>	$\text{AlS}_4$	0D
$\text{Ba}_2\text{AlSbS}_5$	<i>Pnma</i>	$\text{AlS}_4$	0D
$\text{LiBa}_2\text{AlS}_4$	<i>P2_1/m</i>	$\text{AlS}_4$	0D
$\text{Na}_{0.80}\text{Ba}_{0.80}(\text{AlS}_3)_{0.80}$	<i>Pnma</i>	$\text{AlS}_4$	0D
$(\text{Al}_{0.973}\text{Li}_{3.126})\text{S}_3$	<i>C2/c</i>	$\text{AlS}_4$	0D
$\text{Al}(\text{V}_4\text{S}_8)$	<i>F-43m</i>	$\text{AlS}_4$	0D
$\text{Ag}_9\text{AlS}_6$	<i>F-43m</i>	$\text{AlS}_4$	0D
$\text{Al}_{0.55}\text{Mo}_2\text{S}_4$	<i>F-43m</i>	$\text{AlS}_4$	0D
$\text{AlMo}_4\text{S}_8$	<i>F-43m</i>	$\text{AlS}_4$	0D
$\text{Al}_{0.75}\text{Mo}_2\text{S}_4$	<i>F-43m</i>	$\text{AlS}_4$	0D
$\text{Al}(\text{PS}_4)$	<i>I222</i>	$\text{AlS}_4$	0D
$\text{AlMo}_4\text{S}_8$	<i>R3mR</i>	$\text{AlS}_4$	0D
$\text{Na}_5(\text{AlS}_4)$	<i>Pbca</i>	$\text{AlS}_4$	0D
$\text{TaS}_2\text{Al}_{0.33}$	<i>P6_3/mmc</i>	$\text{AlS}_4$	0D
$\text{Na}_3\text{AlS}_3$	<i>P2_1/n</i>	$\text{AlS}_4$	0D
$\text{AlLi}_5\text{S}_4$	<i>P2_1/m</i>	$\text{AlS}_4$	0D
$\text{Na}_6\text{Al}_2\text{S}_6$	<i>P2_1/c</i>	$\text{AlS}_4$	0D
$\text{Rb}_4(\text{Al}_2\text{S}_5)$	<i>Pna2_1</i>	$\text{AlS}_4$	1D
$\text{MAl}_2\text{S}_4$ (M=Eu, Pb, Sr)	<i>Cccm</i>	$\text{AlS}_4$	1D
$\text{Bi}_2(\text{Al}_4\text{S}_8)$	<i>P4/nncZ</i>	$\text{AlS}_4$	1D
$\text{TlAlS}_2$	<i>C2/c</i>	$\text{AlS}_4$	2D

ZnAl <sub>2</sub> S <sub>4</sub>	Pna2 <sub>1</sub>	AlS <sub>4</sub>	2D
FeAl <sub>2</sub> S <sub>4</sub>	R3mH	AlS <sub>4</sub>	2D
MnAl <sub>2</sub> S <sub>4</sub>	R3mH	AlS <sub>4</sub>	2D
AeAl <sub>2</sub> S <sub>4</sub> (Ae=Ca, Sr)	Fddd	AlS <sub>4</sub>	2D
Tl <sub>3</sub> Al <sub>13</sub> S <sub>21</sub>	C1m1	AlS <sub>4</sub>	3D
CuAlS <sub>2</sub>	I-42d	AlS <sub>4</sub>	3D
CdAl <sub>2</sub> S <sub>4</sub>	I-4	AlS <sub>4</sub>	3D
HgAl <sub>2</sub> S <sub>4</sub>	I-4	AlS <sub>4</sub>	3D
Tl <sub>3</sub> (Al <sub>7</sub> S <sub>12</sub> )	P2 <sub>1</sub>	AlS <sub>4</sub>	3D
AlInS <sub>3</sub>	P6 <sub>1</sub>	AlS <sub>4</sub>	3D
BaAl <sub>4</sub> S <sub>7</sub>	Pmn2 <sub>1</sub>	AlS <sub>4</sub>	3D
Ln <sub>6</sub> Al <sub>3.3</sub> S <sub>14</sub> (Ln=Y, Lanthanide series)	P6 <sub>3</sub>	AlS <sub>6</sub> ; AlS <sub>4</sub>	AlS <sub>6</sub> 1D; AlS <sub>4</sub> 0D

**Table S2.** Comparison of the SHG effect and band gap among LaAeAl<sub>3</sub>S<sub>7</sub> and other reported NLO rare-earth chalcogenides.

Number	Compound	E <sub>g</sub> (eV)	SHG (×AgGaS <sub>2</sub> )	References
1	La <sub>2</sub> CuSbS <sub>5</sub>	2.06	0.5	S1
2	Sm <sub>4</sub> InSbS <sub>9</sub>	2.13	0.75	S2
3	ZnY <sub>8</sub> Si <sub>2</sub> S <sub>14</sub>	2.38	0.84	S3
4	Gd <sub>4</sub> GaSbS <sub>9</sub>	2.41	0.8	S2
5	Dy <sub>3</sub> GaS <sub>6</sub>	2.81	0.084	S4
6	Y <sub>3</sub> GaS <sub>6</sub>	2.88	0.21	S4
7	La <sub>3</sub> LiGeS <sub>7</sub>	3.02	0.7	S5
8	KYGeS <sub>4</sub>	3.15	1	S6
9	La <sub>4</sub> InSbS <sub>9</sub>	2.07	1.5	S7
10	EuHgSnS <sub>4</sub>	2.14	1.77	S8
11	La <sub>8</sub> Sb <sub>2</sub> S <sub>15</sub>	2.3	1.2	S9
12	La <sub>3</sub> LiSnS <sub>7</sub>	2.4	1.2	S5
13	La <sub>6</sub> In <sub>2</sub> GeS <sub>14</sub>	2.61	1.8	S10
14	Ba <sub>4</sub> Sm <sub>2</sub> Cd <sub>3</sub> S <sub>10</sub>	2.77	1.8	S11
15	LiSm <sub>3</sub> SiS <sub>7</sub>	2.83	1.5	S5
16	LaSrGa <sub>3</sub> S <sub>7</sub>	2.92	1.3	S12
17	LaCaGa <sub>3</sub> S <sub>7</sub>	2.96	1.3	S12
18	EuCdGeS <sub>4</sub>	2.5	2.6	S13
19	Sm <sub>4</sub> GaSbS <sub>9</sub>	2.23	3.8	S14
20	La <sub>6</sub> Ga <sub>2</sub> GeS <sub>14</sub>	2.54	4.8	S10

21	CsLaGeS <sub>4</sub>	3.6	0.18	S15
22	<b>LaCaAl<sub>3</sub>S<sub>7</sub></b>	<b>3.76</b>	<b>0.8</b>	<b>this work</b>
23	<b>LaSrAl<sub>3</sub>S<sub>7</sub></b>	<b>3.78</b>	<b>1.1</b>	<b>this work</b>

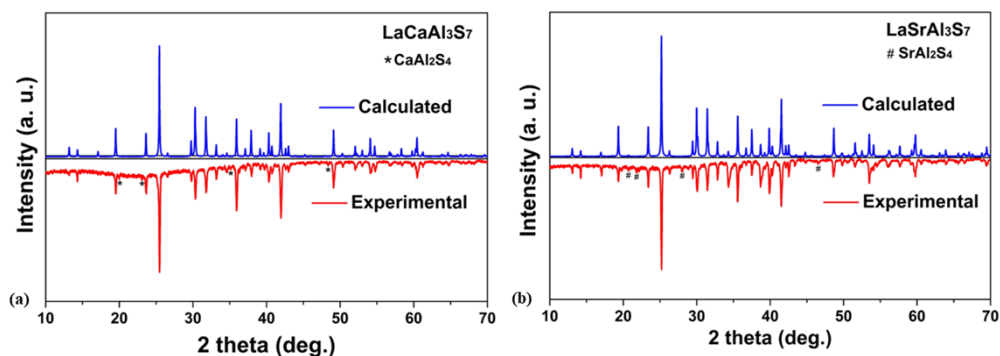
**Table S3.** Crystal data and structure refinement for LaAeAl<sub>3</sub>S<sub>7</sub>.

Empirical formula	LaCaAl <sub>3</sub> S <sub>7</sub>	LaSrAl <sub>3</sub> S <sub>7</sub>
Formula weight	484.35	531.89
crystal system	<i>Tetragonal</i>	
space group	<i>P-42<sub>1</sub>m</i>	
<i>a</i> (Å)	9.4861(16)	9.5854(16)
<i>c</i> (Å)	6.1824(13)	6.2319(18)
<i>Z</i> , <i>V</i> (Å <sup>3</sup> )	2, 556.3(2)	2, 572.6(3)
<i>D<sub>c</sub></i> (g/cm <sup>3</sup> )	2.891	3.085
<i>μ</i> (mm <sup>-1</sup> )	5.788	9.784
GOF on F <sup>2</sup>	1.063	1.059
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (I>2σ(I)) <sup>a</sup>	0.0399, 0.0941	0.0202, 0.0369
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all data)	0.0442, 0.0968	0.0234, 0.0375
<i>Flack parameter</i>	-0.04(3)	0.039(14)
Largest diff. peak and hole (e Å <sup>-3</sup> )	1.463, -1.009	0.672, -0.320

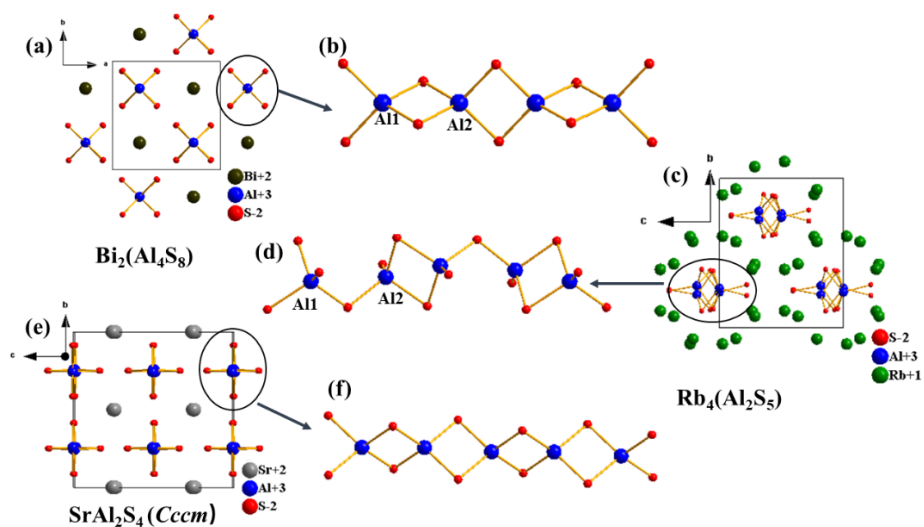
**Table S4.** Comparison on the distortion degrees ( $\Delta d/\%$ ) of AeS<sub>8</sub>/(La/Ae)S<sub>8</sub> units among LaAeAl<sub>3</sub>S<sub>7</sub> and AeAl<sub>2</sub>S<sub>4</sub>.

Compound	(La/Ae)S <sub>8</sub> /AeS <sub>8</sub>	(La/Ae)S <sub>8</sub> /AeS <sub>8</sub> ( $\Delta d/\%$ )
LaCaAl <sub>3</sub> S <sub>7</sub>	(La/Ca)S <sub>8</sub>	2.763
LaSrAl <sub>3</sub> S <sub>7</sub>	(La/Sr)S <sub>8</sub>	3.062
CaAl <sub>2</sub> S <sub>4</sub> ( <i>Fddd</i> )	Ca1S <sub>8</sub>	0.0007
	Ca2S <sub>8</sub>	0.0024
	Ca3S <sub>8</sub>	0.140
	Sr1S <sub>8</sub>	0.030
SrAl <sub>2</sub> S <sub>4</sub> ( <i>Fddd</i> )	Sr2S <sub>8</sub>	0.071
	Sr3S <sub>8</sub>	0.092
SrAl <sub>2</sub> S <sub>4</sub> ( <i>Cccm</i> )	SrS <sub>8</sub>	0.0073

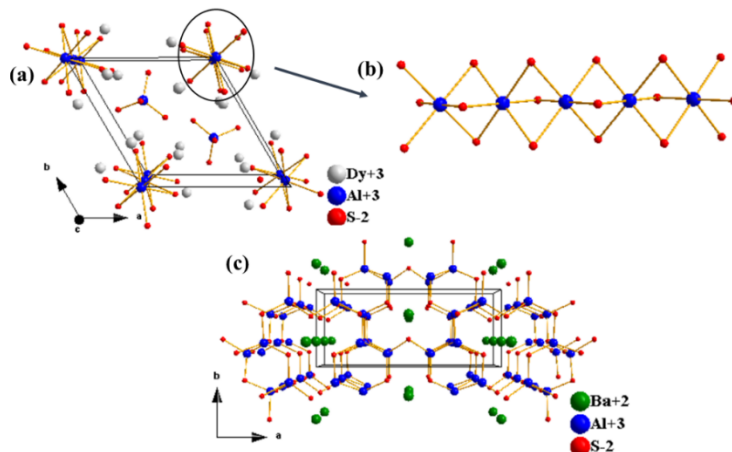
## 2. Figures



**Fig. S1.** Experimental XRD patterns for (a)  $\text{LaCaAl}_3\text{S}_7$  and (b)  $\text{LaSrAl}_3\text{S}_7$ .



**Fig. S2.** (a) Crystal structure of  $\text{Bi}_2(\text{Al}_4\text{S}_8)$  along the  $c$ -axis; (b) 1D chain is composed of  $\text{AlS}_4$  units in  $\text{Bi}_2(\text{Al}_4\text{S}_8)$ ; (c) Crystal structure of  $\text{Rb}_4(\text{Al}_2\text{S}_5)$  along the  $a$ -axis; (d) 1D chain is composed of  $\text{AlS}_4$  units in  $\text{Rb}_4(\text{Al}_2\text{S}_5)$ ; (e) Crystal structure of  $\text{SrAl}_2\text{S}_4$  along the  $a$ -axis; (f) 1D chain is composed of  $\text{AlS}_4$  units in  $\text{SrAl}_2\text{S}_4$ .



**Fig. S3.** (a) Crystal structure of  $\text{Dy}_6\text{Al}_{3.3}\text{S}_{14}$  along the  $c$ -axis; (b) 1D chain is composed of  $\text{AlS}_6$  units in  $\text{Dy}_6\text{Al}_{3.3}\text{S}_{14}$ ; (c) Crystal structure of  $\text{BaAl}_4\text{S}_7$  along the  $c$ -axis.

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