Supporting Information

Stable Self-Trapped Broadband Emission from an Organolead Halide Coordination Polymer with Strong Layer Corrugation and High Chemical Robustness

Ruonan Xi, Yilin Jiang, Yukong Li, Jinlin Yin, Honghan Fei*

School of Chemical Science and Engineering, Shanghai Key Laboratory of Chemical Assessment and Sustainability, Tongji University, Shanghai 200092, P. R. China

Corresponding Author: fei@tongji.edu.cn

Identification code	TJU-19			
Empirical formula	Pb ₂ Cl ₂ C ₅ H ₂ N ₂ O ₄			
Formula weight	639.37			
Temperature/K	297.0			
Crystal system	triclinic			
Space group	<i>P</i> -1			
a/Å	7.0068(8)			
b/Å	12.0274(16)			
$c/\text{\AA}$	12.3700(15)			
$\alpha/^{\circ}$	65.906(4)			
β^{\prime}	89.857(4)			
γ^{\prime}	87.582(4)			
V/Å ³	950.7(2)			
Z	4			
$\rho_{calc}g/cm^3$	4.467			
μ/mm^{-1}	35.926			
<i>F</i> (000)	1104.0			
Crystal size/mm ³	$0.12\times0.11\times0.11$			
Radiation	MoKa ($\lambda = 0.71073$)			
2θ range for data collection/°	5.82 to 55.04			
Index ranges	$-9 \le h \le 8, -15 \le k \le 15, -16 \le l \le 15$			
Reflections collected	22509			
Independent reflections	$4369 \left[R_{\rm int} = 0.0666, R_{\rm sigma} = 0.0511 \right]$			
Data/restraints/parameters	4369/0/271			
Goodness-of-fit on F ²	1.119			
Final <i>R</i> indexes [I>= 2σ (I)]	$R_1 = 0.0343, wR_2 = 0.0648$			
Final R indexes [all data]	$R_1 = 0.0531, wR_2 = 0.0707$			
Largest diff. peak/hole / e Å ⁻³	1.50/-2.03			
$\overline{R_{I} = \sum(F_{o} - F_{c}) / \sum F_{o} ; wR_{2} = \{\sum[w(F_{o}^{2} - F_{c}^{2})] / \sum[w(F_{o}^{2})]^{2}\}^{1/2}}$				

 Table S1. Crystallographic data and structure refinement for TJU-19.

Material	$\lambda_{abs}{}^a$	$\lambda_{em}{}^b$	(x, y) ^c	PLQY ^d	
ТПΙ 10	247 nm	520 nm	(0.23, 0.45)	12 0%	

 Table S2. Photophysical properties of TJU-19

TJU-19347 nm520 nm(0.23, 0.45)12.9%^a λ_{abs} is the absorption edge; ^b λ_{em} is the maximum emission wavelength; ^c (x, y) is the chromatic coordinates; ^d PLQY is photoluminescence quantum yield.



Figure S1. Simulated and experimental PXRD of TJU-19.



Figure S2. Crystallographic top-view of a single $[Pb_2Cl_2]^{2+}$ layer.



Figure S3. Layer corrugation of a single $[Pb_2Cl_2]^{2^+}$ layer of TJU-19 and a single layer of (110)-oriented perovskite **1**.



Figure S4. Simulated and experimental PXRD of perovskite 1.



Figure S5. Thermogravimetric analysis of TJU-19 in N_2 flow.





h.



Figure S7. Excitation spectra of TJU-19 at room temperature.



Figure S8. UV-vis diffuse reflectance spectroscopy and photoluminescence spectra of TJU-19 measured at room temperature.



Figure S9. (a) Tauc plot of TJU-19 fitted by direct bandgap. (b) UV-vis diffuse reflectance spectroscopy of TJU-19. (c) Tauc plot of TJU-19 fitted by indirect bandgap.



Figure S10. Excitation-dependent emission spectra of TJU-19 from 300 to 330 nm.



Figure S11. UV-vis absorption spectra of the free imdc²⁻ ligands.



Figure S12. The coordination environment of the μ_3/μ_4 -bridging chloride species in TJU-19.



Figure S13. Calculated DOS for the ligand (orange), Cl (blue) and Pb (yellow) in TJU-19.



Figure S14. TA kinetics of TJU-19 ranging from 400 to 550 nm.



Figure S15. PXRD and emission spectra of TJU-19 before and after continuous

illumination in air from a 300 W Xe lamp for 12 h.