## Supplementary Information

## Synergetic Effect between Non-metals and Dual Metal Catalysts for

## Nitrogen Reduction Reaction

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	Fe <sub>2</sub> @BN <sub>4</sub>		Fe <sub>2</sub> @SN <sub>4</sub>				
species	ΔΖΡΕ	$T\Delta S$	species	ΔΖΡΕ	$T\Delta S$		
N-N**	0.20	0.11	N-N**	0.18	0.07		
N-NH**	0.50	0.12	N-NH**	0.50	0.10		
NH-NH**	0.81	0.13	NH-NH**	0.82	0.12		
NH-NH2**	1.16	0.14	NH-NH2**	1.17	0.13		
NH2-NH2**	1.34	0.20	NH2-NH2**	1.35	0.19		
NH2**	0.70	0.07	NH2**	0.67	0.11		
NH3**	1.02	0.16	NH3**	1.02	0.18		
	Co <sub>2</sub> @BN4		Co <sub>2</sub> @PN4				
species	ΔΖΡΕ	$T\Delta S$	species	ΔΖΡΕ	$T\Delta S$		
N-N**	0.20	0.11	N-N*	0.21	0.14		
N-NH**	0.50	0.11	NNH*	0.48	0.15		
NH-NH**	0.82	0.10	NNH2*	0.81	0.15		
NH-NH2**	1.19	0.11	N*	0.09	0.03		
NH2-NH2**	1.49	0.16	NH*	0.38	0.05		
NH2**	0.73	0.07	NH2*	0.72	0.06		
NH3**	1.02	0.15	NH3*	1.03	0.13		
	Ni <sub>2</sub> @PN4		gas				
species	ΔΖΡΕ	$T\Delta S$	species	ΔΖΡΕ	$T\Delta S$		
N-N**	0.19	0.13	NH3	0.92	0.59		
N-NH**	0.51	0.10	H2	0.27	0.40		
NH-NH**	0.81	0.12	N2	0.15	0.59		
NH-NH2**	1.17	0.12					
NH2-NH2**	1.30	0.17					
NH2**	0.71	0.06					
NH3**	1.02	0.14					

Table S1. Calculated  $\triangle$ ZPE and T $\triangle$ S of gas molecules and intermediates on catalyst, the \* represent end-on mode, while \*\* represent side-on mode.

	Distal process								
Surface	N2	NNH	NNH2	N	NH	NH2	NH3		
Fe2@NN4	-0.84	0.15	0.24	-0.50	-0.82	-1.35	-1.63		
Fe2@BN4	-0.81	-0.48	-0.34	-1.03	-1.54	-2.42	-1.46		
Fe2@ON4	-1.02	-0.15	-0.55	-0.79	-1.58	-2.26	-1.65		
Fe2@SN4	-1.13	-0.84	-1.00	-1.69	-2.07	-2.71	-2.36		
Fe2@PN4	-0.72	-0.11	-0.41	-0.81	-1.48	-2.43	-1.46		
Co2@NN4	-0.84	0.23	0.25	-0.13	-0.13	-1.86	-1.51		
Co2@BN4	-0.73	-0.35	-0.26	-0.94	-1.31	-1.91	-1.52		
Co2@ON4	-1.13	-0.11	-0.24	-0.43	-1.25	-1.79	-1.69		
Co2@SN4	-0.68	-0.13	0.16	-0.58	-1.01	-1.32	-1.21		
Co2@PN4	-0.37	-0.03	-0.09	-0.47	-1.15	-1.40	-1.17		
Ni2@NN4	-0.79	0.45	0.88	0.98	0.05	-1.02	-1.54		
Ni2@BN4	-0.54	0.39	0.35	0.29	-0.41	-1.44	-1.53		
Ni2@ON4	-0.58	-0.28	-0.30	0.36	-0.87	-1.77	-1.71		
Ni2@SN4	-0.17	0.74	0.40	0.74	-0.13	-1.06	-0.78		
Ni2@PN4	-0.36	0.69	0.66	0.92	-0.21	-1.39	-1.11		
	Alternating process								
Surface	N2	NNH	NHNH	NHNH2	NH2NH2	NH2	NH3		
Fe2@NN4	-0.87	0.15	0.63	-0.38	-0.56	-1.29	-1.68		
Fe2@BN4	-0.81	-0.48	0.12	-0.31	0.30	-0.87	-1.51		
Fe2@ON4	-1.02	-0.15	-0.33	0.28	0.50	-2.25	-1.65		
Fe2@SN4	-1.13	-0.84	-0.03	-0.82	-0.41	-2.30	-2.36		
Fe2@PN4	-0.72	-0.11	0.04	-0.34	0.52	-1.98	-1.46		
Co2@NN4	-0.85	0.21	0.68	-0.32	-0.35	-1.87	-1.51		
Co2@BN4	-0.73	-0.35	0.40	-0.03	0.69	-0.91	-1.52		
Co2@ON4	-1.13	-0.11	-0.16	0.11	0.27	-1.76	-1.69		
Co2@SN4	-0.68	-0.14	0.18	0.37	0.39	-1.55	-1.21		
Co2@PN4	-0.37	-0.03	0.27	0.14	-0.39	-1.66	-1.17		
Ni2@NN4	-0.82	0.42	0.60	-0.05	0.39	-0.98	-1.57		
Ni2@BN4	-0.54	0.39	0.52	0.35	0.29	-0.76	-1.53		
Ni2@ON4	-0.58	-0.28	-0.26	0.20	0.31	-1.64	-1.71		
Ni2@SN4	-0.17	0.74	0.69	0.52	0.45	-1.10	-0.78		
Ni2@PN4	-0.36	0.69	0.99	0.43	0.48	-0.53	-1.11		
			Ensy	matic pro	ocess				
Surface	N2	NNH	NHNH	NHNH2	NH2NH2	NH2	NH3		

Table S2. The free energy of 15 catalysts via distal, alternating and enzymetic processes.

Fe2@NN4	0.05	0.23	0.28	-0.16	0.36	-1.53	-1.57
Fe2@BN4	0.07	0.18	-0.08	-0.19	-1.14	-0.87	-1.56
Fe2@ON4	-0.37	-0.15	-0.55	-0.53	-1.22	-2.25	-1.69
Fe2@SN4	-0.82	-0.62	-1.06	-1.15	-2.13	-2.37	-2.41
Fe2@PN4	-0.07	0.13	-0.24	0.08	-1.18	-2.33	-1.36
Co2@NN4	-0.12	0.49	0.49	-0.05	0.09	-1.47	-1.67
Co2@BN4	-0.25	0.11	0.10	0.19	0.45	-0.89	-1.38
Co2@ON4	-0.47	0.10	-0.38	-0.32	-0.98	-1.80	-1.79
Co2@SN4	-0.29	0.07	0.06	0.20	-0.57	-1.50	-1.50
Co2@PN4	-0.19	0.20	-0.03	-0.11	-0.68	-1.66	-1.21
Ni2@NN4	-0.55	0.74	0.36	-0.04	0.04	-1.45	-1.93
Ni2@BN4	0.02	0.44	0.53	0.22	0.75	-0.76	-1.62
Ni2@ON4	-0.53	-0.14	-0.25	-0.13	-0.38	-1.73	-1.73
Ni2@SN4	-0.66	0.55	0.55	0.51	0.27	-1.18	-1.41
Ni2@PN4	0.13	0.46	0.61	0.24	0.55	-1.31	-0.98

Table S3. The bader charge  $Q_{N2}$  and bond length  $d_{N-N}$  of  $N_2$  adsorbed on 15 catalysts with end-on and side-on configurations; the binding energy  $E_b$  of transition metal atom pairs anchored on the slab calculated by the energies of metal atoms in the vacuum and the bulk metal ( $E_{coh}$ ).

Surface	Q <sub>N2</sub> -end-on	d <sub>N-N-end-on</sub>	Q <sub>N2-side</sub> -on	d <sub>N-N-side-on</sub>	E <sub>b</sub> (vacuum)	E <sub>b</sub> (bulk)	E <sub>coh</sub>
Fe2@NN4	0.33	1.17	0.57	1.20	5.11	4.02	4.28
Fe2@BN4	0.51	1.17	0.71	1.21	5.46	4.36	4.28
Fe2@ON4	0.51	1.16	0.70	1.22	4.36	3.25	4.28
Fe2@SN4	0.58	1.17	0.67	1.22	3.03	1.94	4.28
Fe2@PN4	0.56	1.17	0.79	1.23	4.90	3.81	4.28
Co2@NN4	0.35	1.16	0.47	1.18	5.39	2.72	4.39
Co2@BN4	0.44	1.16	0.57	1.20	5.76	3.09	4.39
Co2@ON4	0.49	1.16	0.52	1.19	4.65	1.98	4.39
Co2@SN4	0.53	1.16	0.63	1.19	4.80	2.13	4.39
Co2@PN4	0.51	1.16	0.63	1.19	5.36	2.69	4.39
Ni2@NN4	0.23	1.16	0.42	1.14	5.45	1.33	4.44
Ni2@BN4	0.35	1.16	0.49	1.18	5.33	1.21	4.44
Ni2@ON4	0.46	1.16	0.47	1.18	4.79	0.67	4.44
Ni2@SN4	0.40	1.16	0.32	1.14	5.04	0.92	4.44
Ni2@PN4	0.41	1.16	0.48	1.19	5.51	1.39	4.44
Fe@SN4	0.32	1.14	0.33	1.17	5.39	4.31	4.28



Figure S1. The structures of  $N_2$  adsorbed on (a)  $Fe_2@SN_4$  and (b)  $Fe_2@SN_4$  with endon and side-on configurations.



Figure S2. Free energy diagram of DACs for HER.



Figure S3. The calculated free energy profiles for NRR through enzymatic mechanism and distal mechanism on  $Fe@SN_4$ .