

Supporting Information

Successive redox modulation in an iron(II) spin-crossover framework

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Contents

Thermogravimetric analysis.....	3
Crystal data.....	5
Powder X-Ray Diffraction	10
Infrared spectra.....	12
XPS Measurements	15
Magnetization measurements.....	19

Thermogravimetric analysis

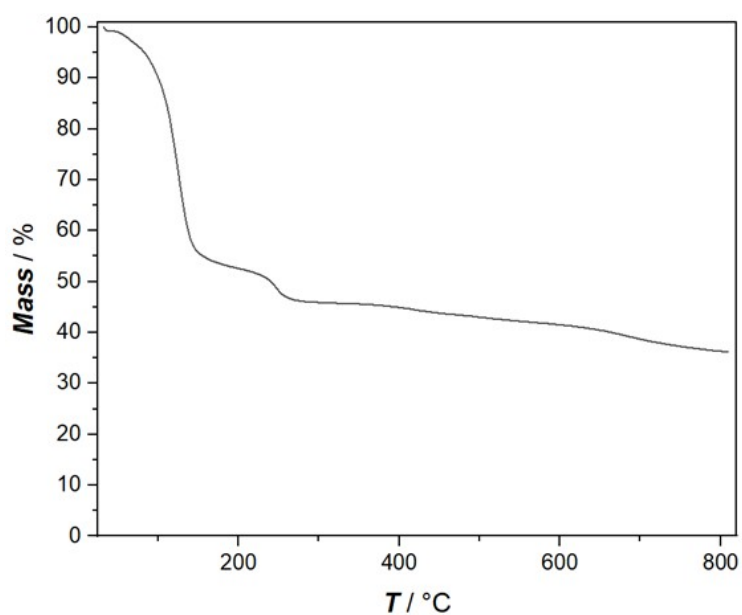


Figure S1. Thermogravimetric analysis curve of **1**. The weight loss of 53.4% is close to the theoretical value of 53.0% corresponding to the escape of 2 H₂O and 7 TCE molecules.

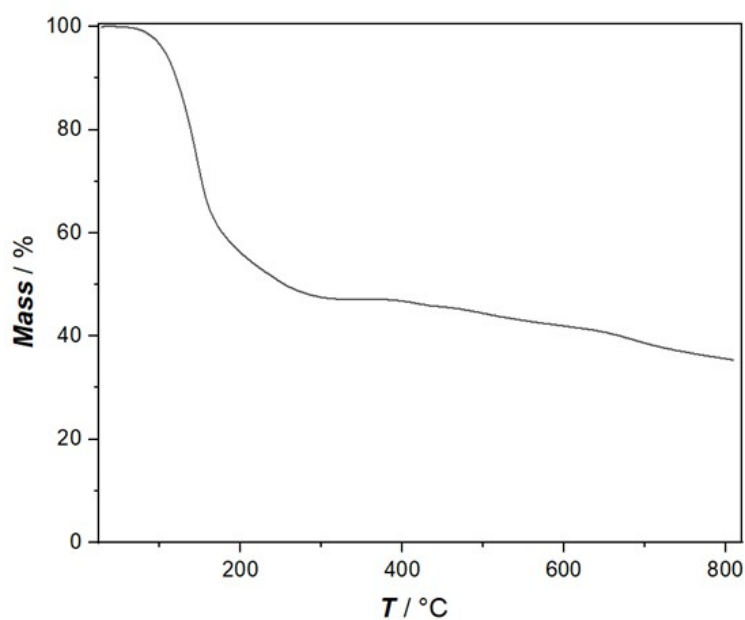


Figure S2. Thermogravimetric analysis curve of **2**. The weight loss of 52.6% is close to the theoretical value of 51.9% corresponding to the escape of 1 I₂ and 7 TCE molecules.

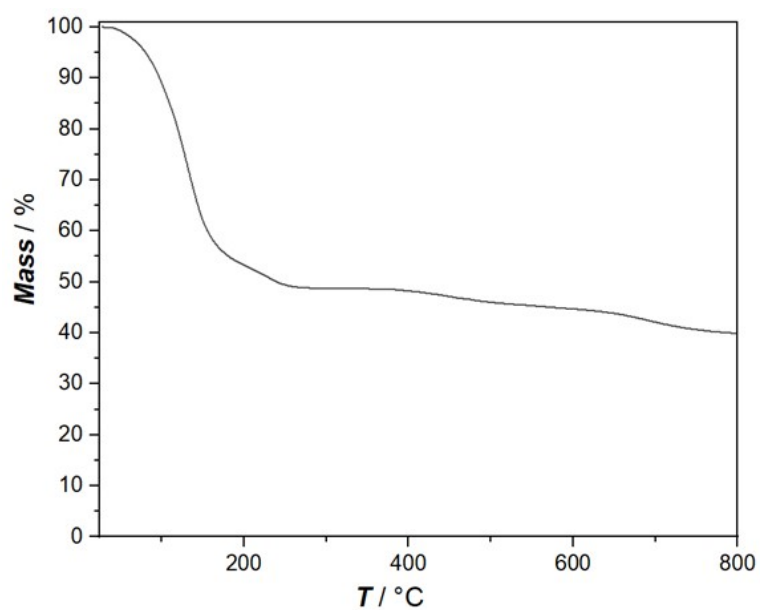


Figure S3. Thermogravimetric analysis curve of **3**. The weight loss of 51.1% is close to the theoretical value of 50.3% corresponding to the escape of 1 H₂O and 8 TCE molecules.

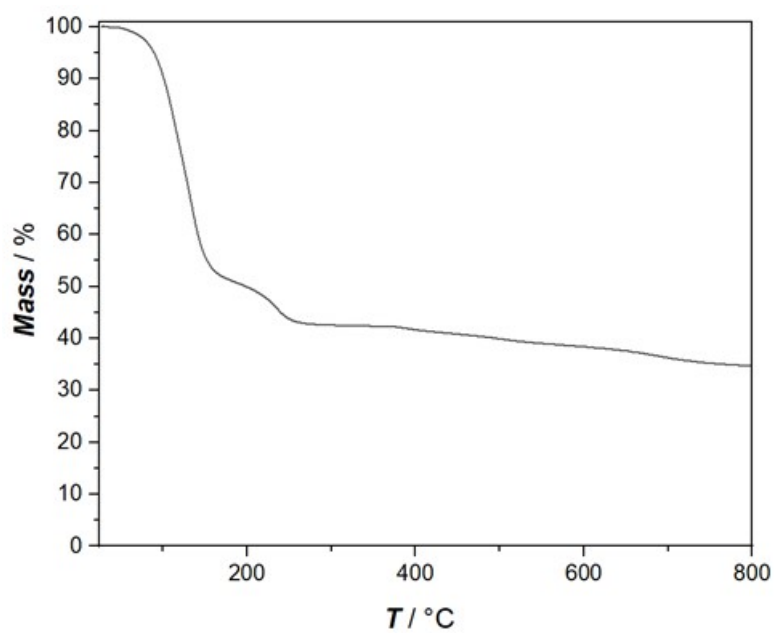


Figure S4. Thermogravimetric analysis curve of **4**. The weight loss of 57.2% is close to the theoretical value of 57.7% corresponding to the escape of 3 H₂O and 8 TCE molecules.

Crystal data

Table S1. Crystallographic data for **1**.

Empirical formula	$C_{62}H_{50}AuCl_{28}O_2FeIN_6$	
Formula weight	2283.39	
Temperature/K	300.00	120.00
Crystal system	orthorhombic	orthorhombic
Space group	<i>Pmmm</i>	<i>Pmmm</i>
$a / \text{Å}$	10.5117(10)	10.0549(7)
$b / \text{Å}$	14.7107(14)	14.5520(10)
$c / \text{Å}$	16.9860(17)	16.5361(12)
$\alpha / ^\circ$	90	90
$\beta / ^\circ$	90	90
$\gamma / ^\circ$	90	90
Volume / Å^3	2626.6(4)	2419.5(3)
Z	1	1
$\rho_{\text{calc}} / \text{g/cm}^3$	1.444	1.567
μ / mm^{-1}	2.575	2.795
$F(000)$	1114.0	1114.0
Crystal size / mm^3	$0.226 \times 0.215 \times 0.205$	$0.226 \times 0.215 \times 0.205$
Reflections collected	24957	36892
Independent reflections	3532	3383
Goodness-of-fit on F^2	1.081	1.185
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0829, wR_2 = 0.2287$	$R_1 = 0.0519, wR_2 = 0.1533$
Final R indexes [all data]	$R_1 = 0.0882, wR_2 = 0.2337$	$R_1 = 0.0539, wR_2 = 0.1556$
Largest diff. peak/hole / $e \text{ Å}^{-3}$	3.57/−1.28	2.52/−0.96
CCDC	2249183	2249182

$${}^a R_1 = \frac{\sum |F_o| - |F_c|}{\sum |F_o|}; {}^b wR_2 = \left\{ \frac{\sum w(F_o^2 - F_c^2)^2}{\sum [w(F_o^2)^2]} \right\}^{1/2}$$

Table S2. Crystallographic data for **2**.

Empirical formula	$C_{62}H_{46}AuCl_{28}FeI_5N_6$	
Formula weight	2754.96	
Temperature/K	320.0	102.00
Crystal system	orthorhombic	orthorhombic
Space group	<i>Pmmm</i>	<i>Fmmm</i>
<i>a</i> / Å	10.4944(9)	19.9965(11)
<i>b</i> / Å	14.6760(13)	28.2630(17)
<i>c</i> / Å	17.0345(15)	33.579(2)
<i>α</i> / °	90	90
<i>β</i> / °	90	90
<i>γ</i> / °	90	90
Volume / Å ³	2623.6(4)	18977.4(19)
<i>Z</i>	1	8
ρ_{calc} g/cm ³	1.744	1.928
μ /mm ⁻¹	3.754	4.152
<i>F</i> (000)	1306.0	10448.0
Crystal size / mm ³	0.397 × 0.346 × 0.279	0.234 × 0.217 × 0.186
Reflections collected	23389	44070
Independent reflections	3085	6286
Goodness-of-fit on <i>F</i> ²	1.120	1.066
Final <i>R</i> indexes [<i>I</i> ≥ 2σ (<i>I</i>)]	<i>R</i> ₁ = 0.1028, <i>wR</i> ₂ = 0.2803	<i>R</i> ₁ = 0.0869, <i>wR</i> ₂ = 0.2724
Final <i>R</i> indexes [all data]	<i>R</i> ₁ = 0.1059, <i>wR</i> ₂ = 0.2843	<i>R</i> ₁ = 0.1082, <i>wR</i> ₂ = 0.2919
Largest diff. peak/hole / e Å ⁻³	3.75/−2.17	2.85/−3.67
CCDC	2249191	2249190

$${}^a R_1 = \frac{\sum |F_o| - |F_c|}{\sum |F_o|}; {}^b wR_2 = \left\{ \frac{[\sum w(F_o^2 - F_c^2)^2]}{\sum [w(F_o^2)^2]} \right\}^{1/2}$$

Table S3. Crystallographic data for **3**.

Empirical formula	C ₆₄ H ₅₀ AuCl ₃₂ OBr ₅ FeN ₆
Formula weight	2705.86
Temperature/K	120.0
Crystal system	orthorhombic
Space group	<i>Pmmm</i>
<i>a</i> / Å	10.006(3)
<i>b</i> / Å	14.398(5)
<i>c</i> / Å	16.704(5)
α / °	90
β / °	90
γ / °	90
Volume / Å ³	2406.5(13)
<i>Z</i>	1
ρ_{calc} g/cm ³	1.867
μ /mm ⁻¹	4.679
<i>F</i> (000)	1308.0
Crystal size / mm ³	0.143 × 0.121 × 0.116
Reflections collected	24481
Independent reflections	2257
Goodness-of-fit on <i>F</i> ²	1.024
Final <i>R</i> indexes [<i>I</i> ≥ 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0923, <i>wR</i> ₂ = 0.2550
Final <i>R</i> indexes [all data]	<i>R</i> ₁ = 0.1326, <i>wR</i> ₂ = 0.2826
Largest diff. peak/hole / e Å ⁻³	2.78/−1.21
CCDC	2249194

$${}^a R_1 = \sum |F_o| - |F_c| / \sum |F_o|; {}^b wR_2 = \{[\sum w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$$

Table S4. Selected bond lengths for **1**

T / K	300	120
Fe1–N1 / Å	2.124(9)	1.906(6)
Fe1–N2 / Å	2.235(5)	2.002(3)
<Fe–N> ^a / Å	2.198	1.970
ΣFe ^b / °	4.78	4.19
Au1–C1	1.958(10)	1.964(7)
∠Fe1–N1–C1 / °	180	180
∠Au1–C1–N1 / °	180	180

^aaverage Fe–N bond length; ^bOctahedral distortion parameters.

Table S5. Selected bond lengths for **2**

T / K	320	102
Fe1–N1 / Å	2.143(13)	1.892(10)
Fe1–N2 / Å	2.227(9)	1.997(6)
<Fe–N> ^a / Å	2.199	1.962
ΣFe ^b / °	3.58	14.5
Au1–C1	1.949(14)	1.943(12)
Au1–I1	2.699(11)	2.617(3)
Au1–I1A	2.658(14)	2.607(5)
Au1–I1B	-	2.586(9)
∠Fe1–N1–C1 / °	180	180
∠Au1–C1–N1 / °	180	180

^aaverage Fe–N bond length; ^bOctahedral distortion parameters.

Table S6. Selected bond lengths for **3**

T / K	120
Fe1–N1 / Å	1.86(2)
Fe1–N2 / Å	2.055(12)
<Fe–N> ^a / Å	1.990
ΣFe ^b / °	10.6
Au1–C1	1.96(3)
Au1–Br1	2.405(9)
Au1–Br2	2.405(13)
Au1–Br2A	2.394(12)
∠Fe1–N1–C1 / °	180
∠Au1–C1–N1 / °	180

^aaverage Fe–N bond length; ^bOctahedral distortion parameters.

Powder X-Ray Diffraction

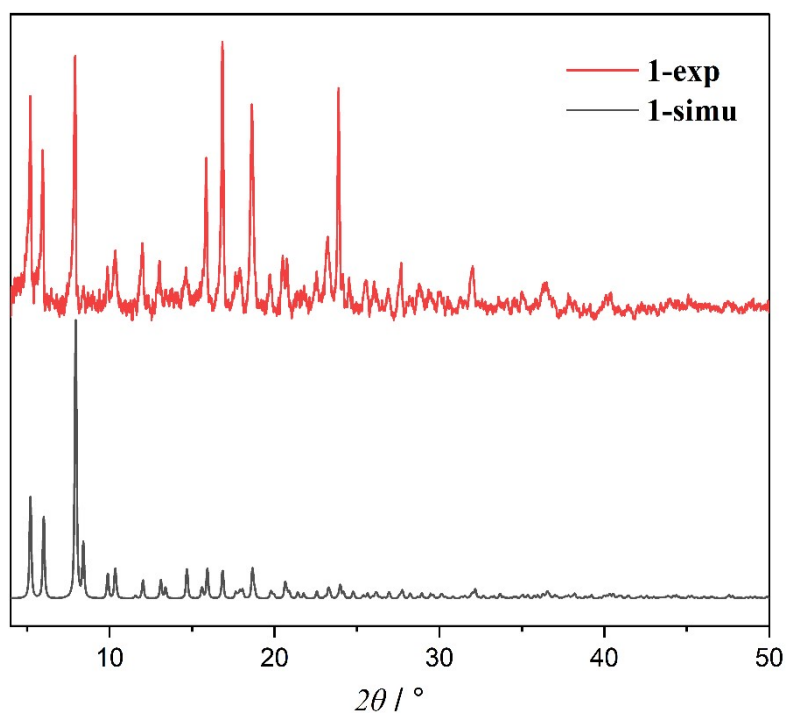


Figure S5. Powder X-ray diffraction (PXRD) patterns for **1**. The simulated PXRD pattern was obtained from crystal data at 300 K.

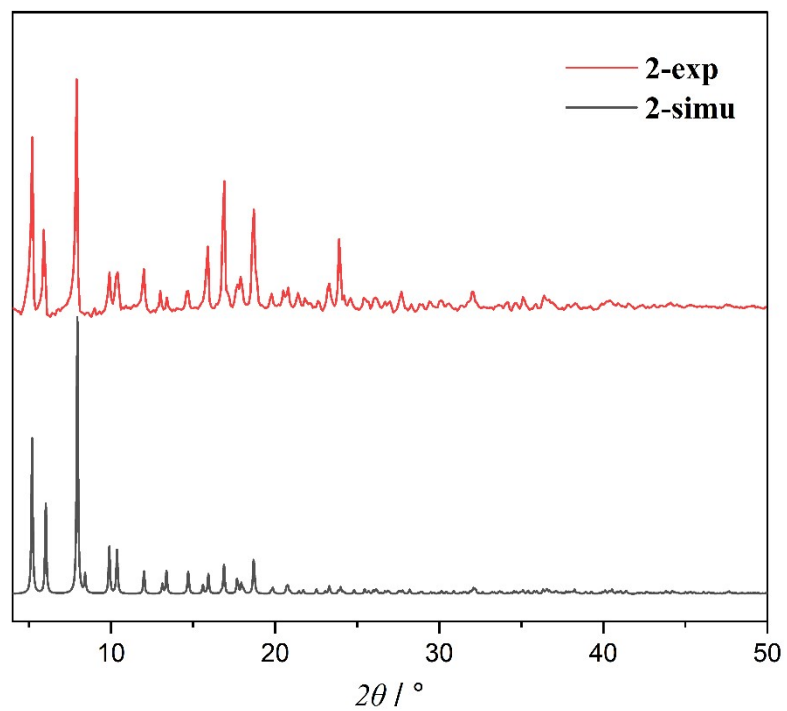


Figure S6. Powder X-ray diffraction (PXRD) patterns for **2**. The simulated PXRD pattern was obtained from crystal data at 320 K.

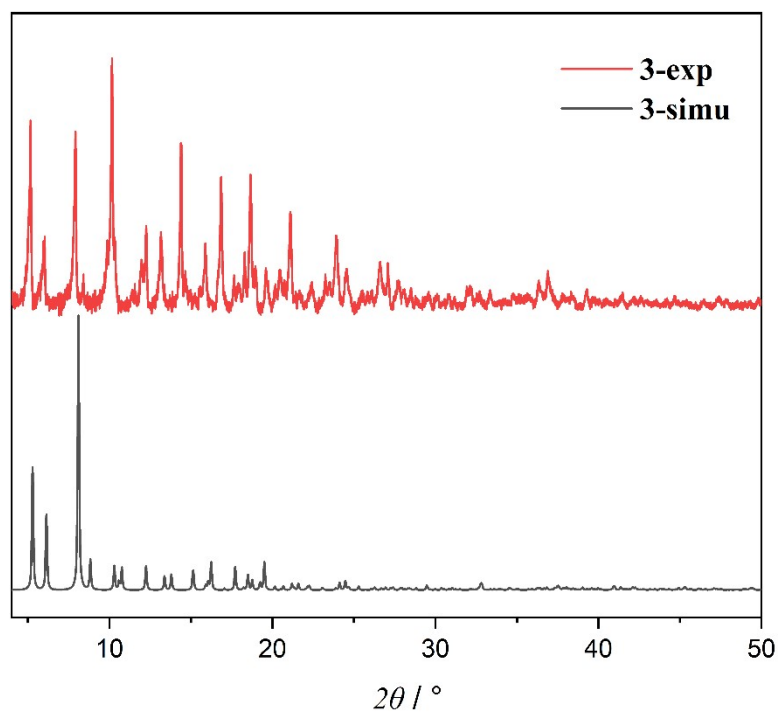


Figure S7. Powder X-ray diffraction (PXRD) patterns for **3**. The simulated PXRD pattern was obtained from crystal data at 120 K.

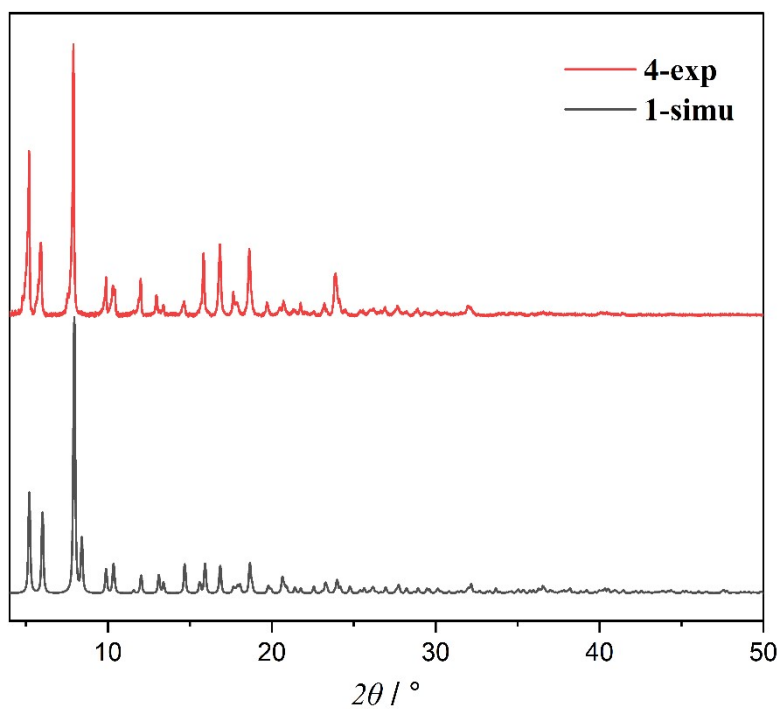


Figure S8. Powder X-ray diffraction (PXRD) patterns for **4**. The simulated PXRD pattern was obtained from crystal data of **1** at 300 K.

Infrared spectra

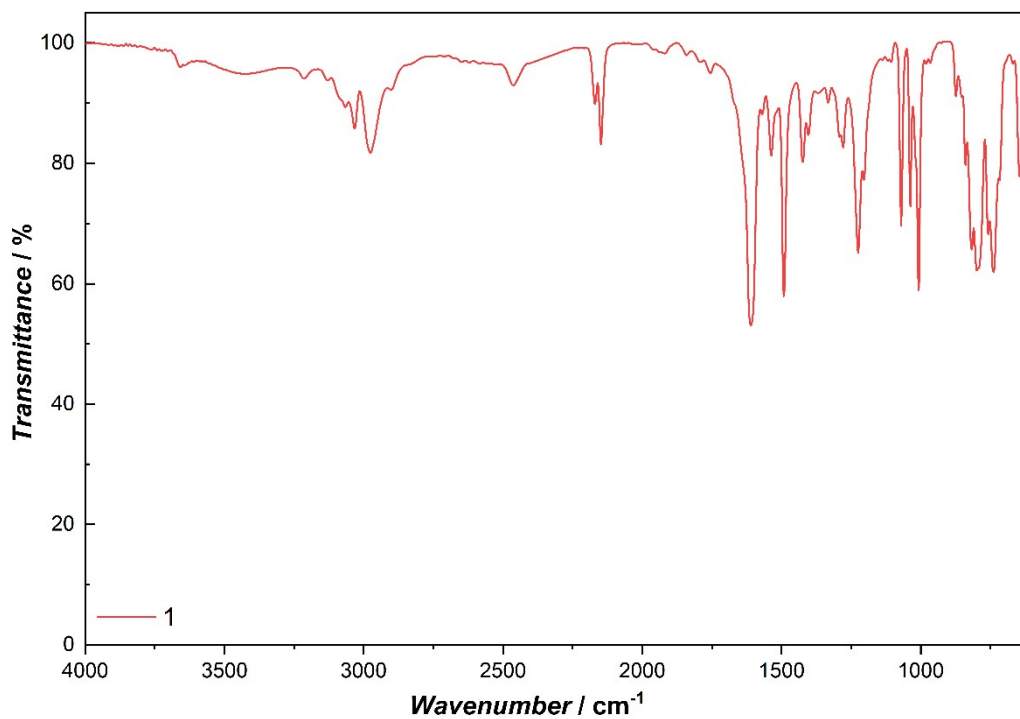


Figure S9. The infrared spectrum of **1**.

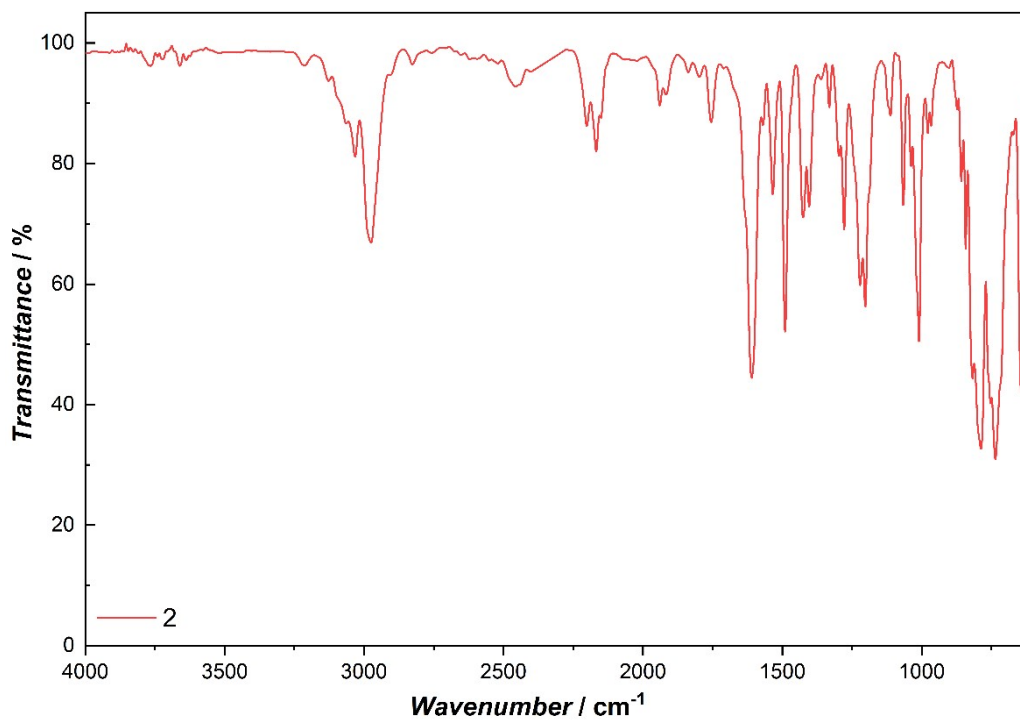


Figure S10. The infrared spectrum of **2**.

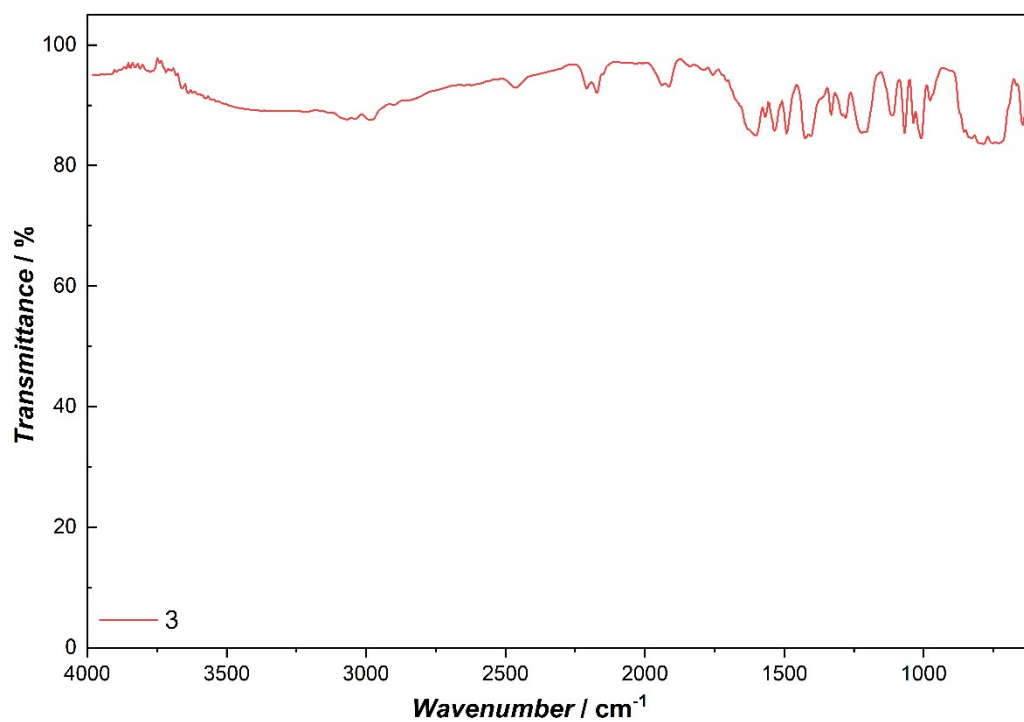


Figure S11. The infrared spectrum of **3**.

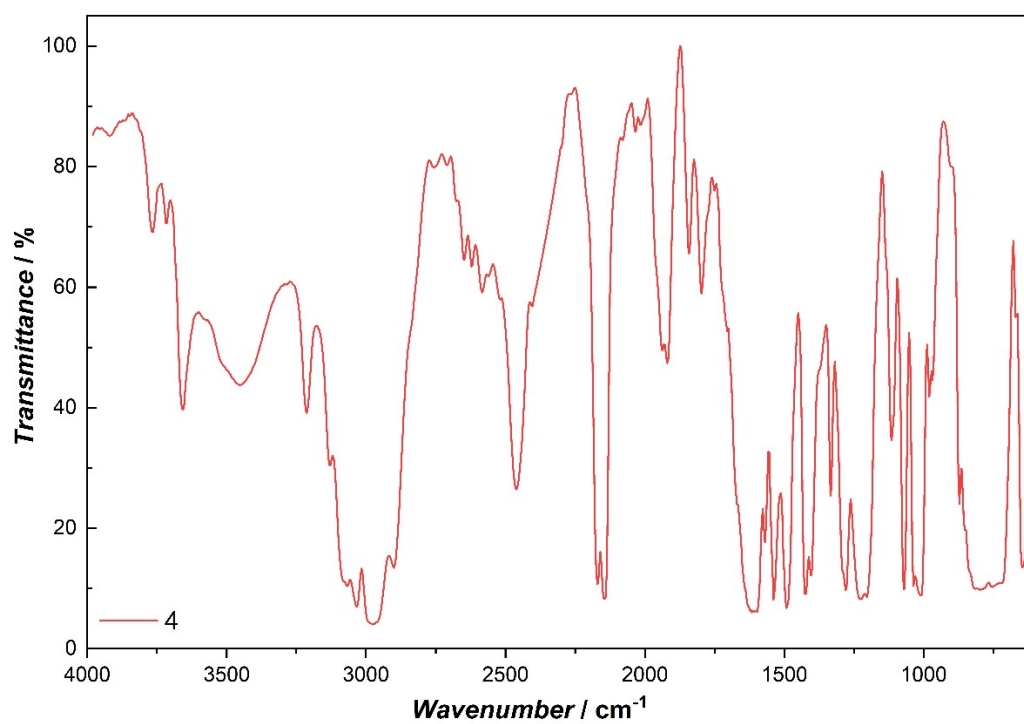


Figure S12. The infrared spectrum of **4**.

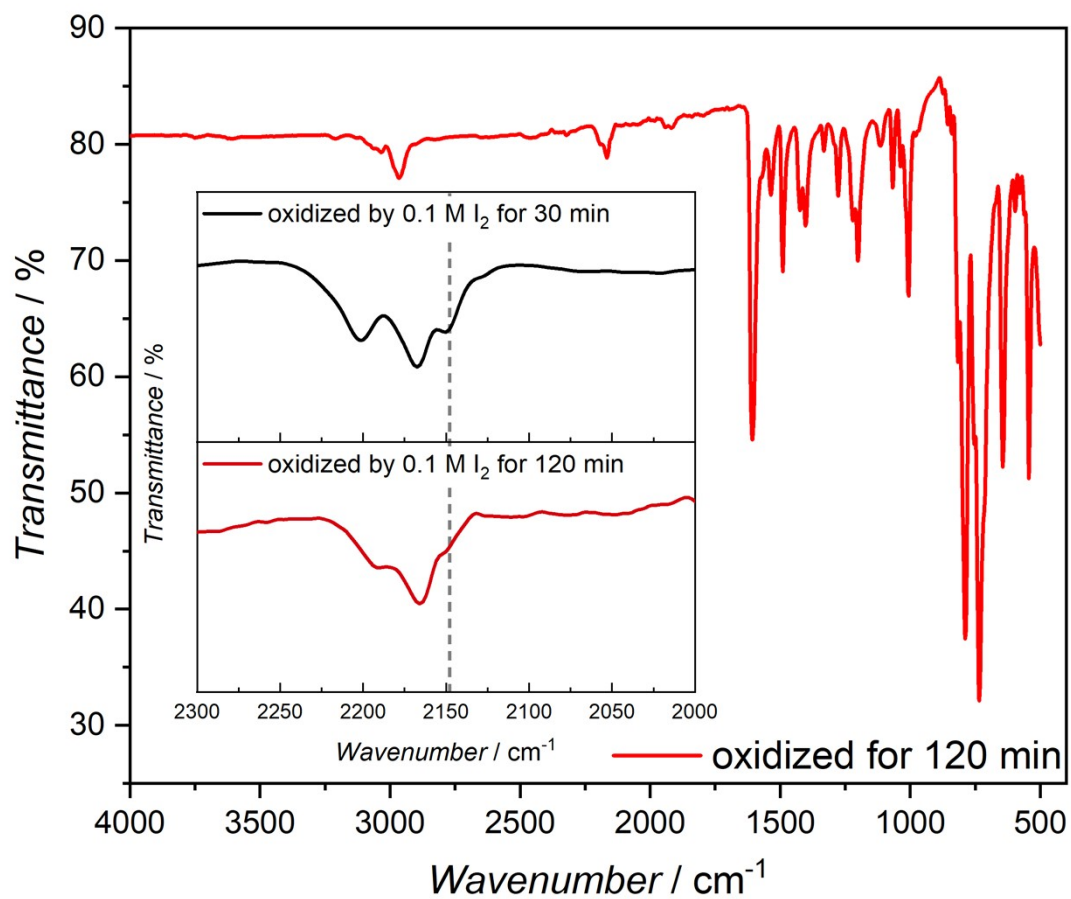


Figure S13. The infrared spectrum of the oxidation product obtained by reacting **1** with 0.1 M I₂ for 120 min. Inset: the enlarged view of the cyanide vibration signals for the products after being oxidized by 0.1 M I₂ for 30 and 120 min.

XPS Measurements

X-ray photoelectron spectroscopy (XPS) measurements were carried out with the Thermo Fisher Scientific ESCALAB Xi+ system ($Al-K_{\alpha}$ source). Sequential XPS spectra were recorded by scanning the electron analyzer from 100 to 75 eV (increment: 0.05 V, pass energy: 20.0 eV). The total acquisition times for Au 4f scan were about 3.5 min for **1**, about 0.5 min for **2** and **3** and about 1 min for **4**. The peak analysis processes were carried out with the XPSPEAK41 package. The electron-scattering background was deducted using a Shirley method without changing the intrinsic profile of raw data. The intervals and area ratios of $4f_{7/2}$ and $4f_{5/2}$ were restricted to 3.67 eV and 4:3 according to the literature. The full-width-at-half-maximum of the $4f_{7/2}$ and $4f_{5/2}$ doublets was fitted equally. The best fitted results were obtained by using a pseudo-Voigt function (20% Gaussian and 80% Lorentzian).

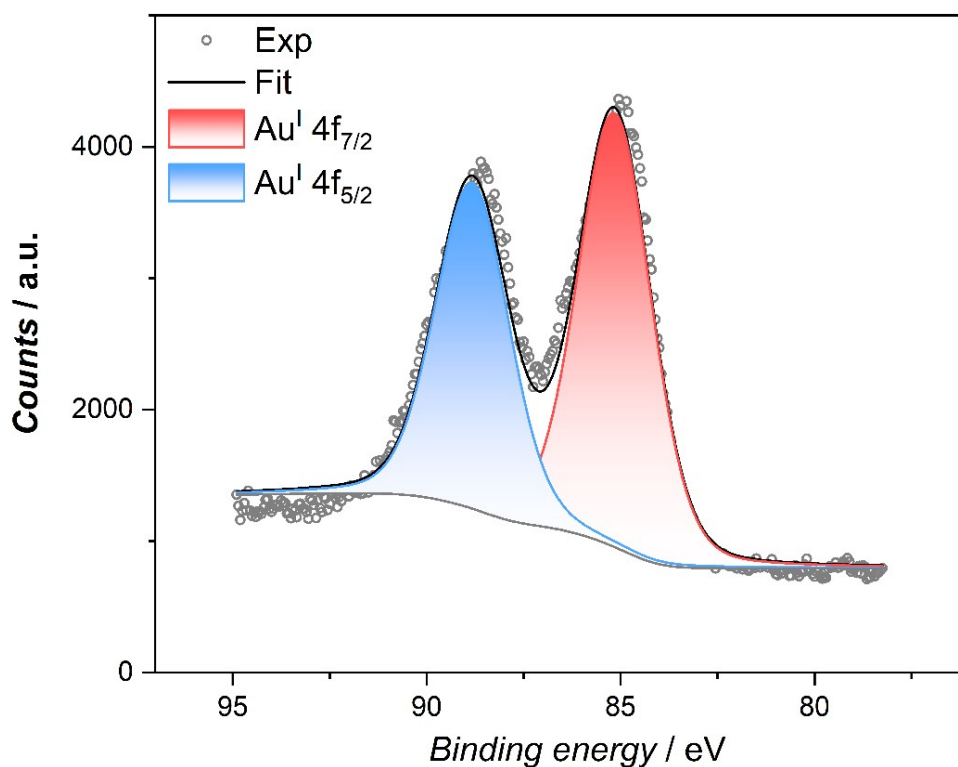


Figure S14. X-ray photoelectron spectroscopy of Au 4f in **1**.

Table S7. The parameters of best fit results of Au 4f in **1**.

Peak	Position / eV	Area	FWHM / eV
Au ^I 4f _{7/2}	85.154	8634.142	2.243
Au ^I 4f _{5/2}	88.824	6475.606	2.243

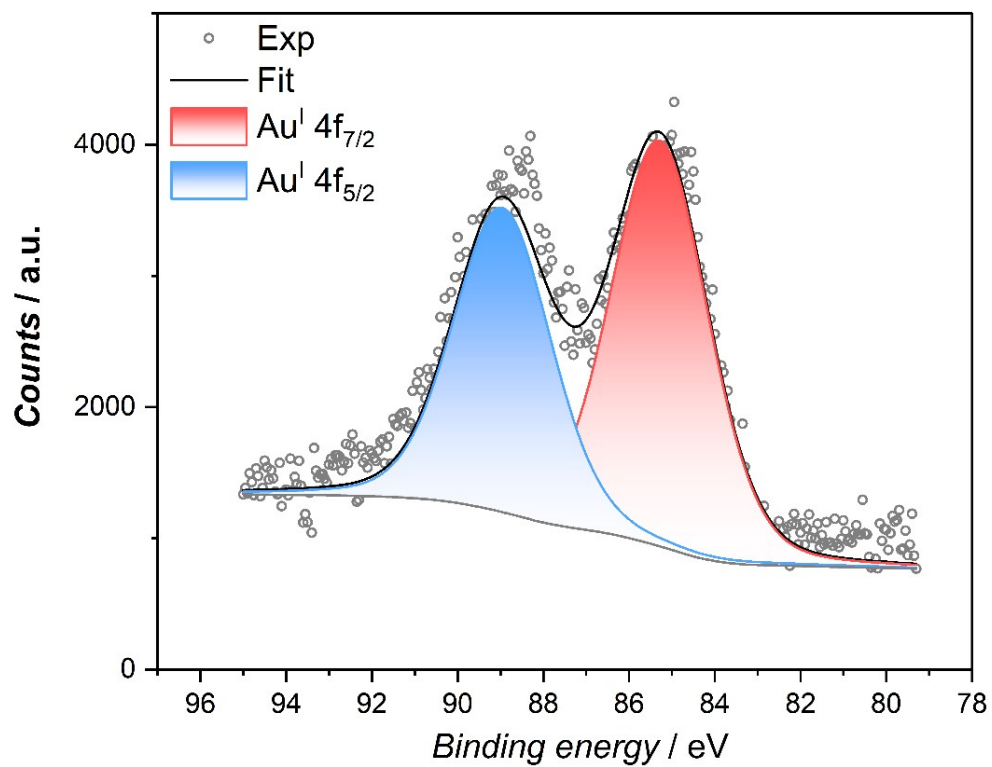


Figure S15. X-ray photoelectron spectroscopy of Au 4f in **2**.

Table S8. The parameters of best fit results of Au 4f in **2**.

Peak	Position / eV	Area	FWHM / eV
Au ^I 4f _{7/2}	85.287	9663.406	2.680
Au ^I 4f _{5/2}	88.957	7247.555	2.680

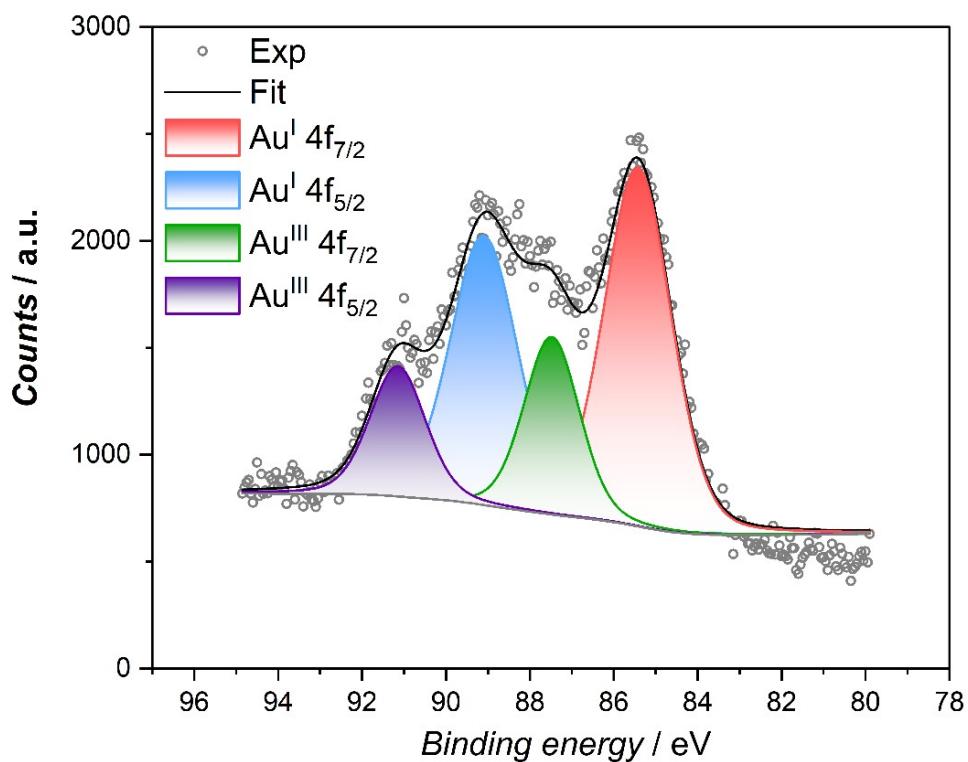


Figure S16. X-ray photoelectron spectroscopy of Au 4f in **3**.

Table S9. The parameters of best fit results of Au 4f in **3**.

Peak	Position / eV	Area	FWHM / eV
Au ^I 4f _{7/2}	85.427	3694.506	1.887
Au ^I 4f _{5/2}	89.097	2770.879	1.887
Au ^{III} 4f _{7/2}	87.479	1510.974	1.560
Au ^{III} 4f _{5/2}	91.149	1133.231	1.560

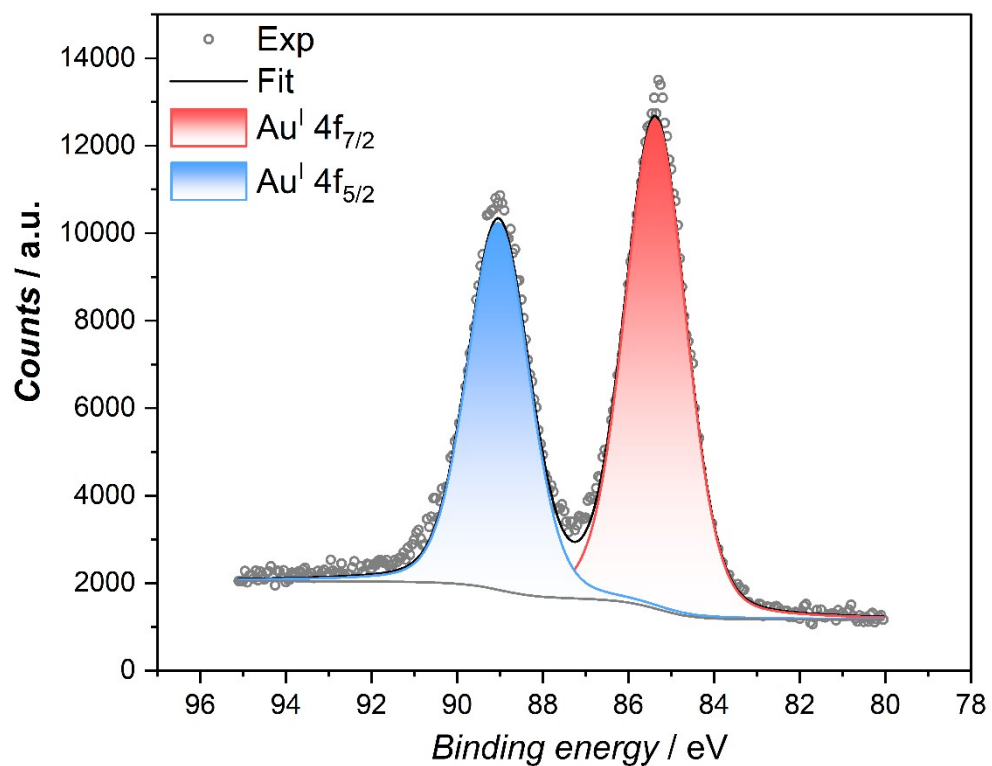


Figure S17. X-ray photoelectron spectroscopy of Au 4f in **4**.

Table S10. The parameters of best fit results of Au 4f in **4**.

Peak	Position / eV	Area	FWHM / eV
Au ^I 4f _{7/2}	85.368	22125.31	1.697
Au ^I 4f _{5/2}	89.038	16593.98	1.697

Magnetization measurements

Magnetic susceptibility measurements were performed on a Quantum Design PPMS3 SQUID magnetometer with a sweep rate of 2 K min^{-1} under applied field of 5 kOe. Polycrystalline samples were packed in plastic film with a small amount of mother liquor. Data were corrected for the signal of the sample holder and diamagnetic contribution calculated from Pascal's constants.

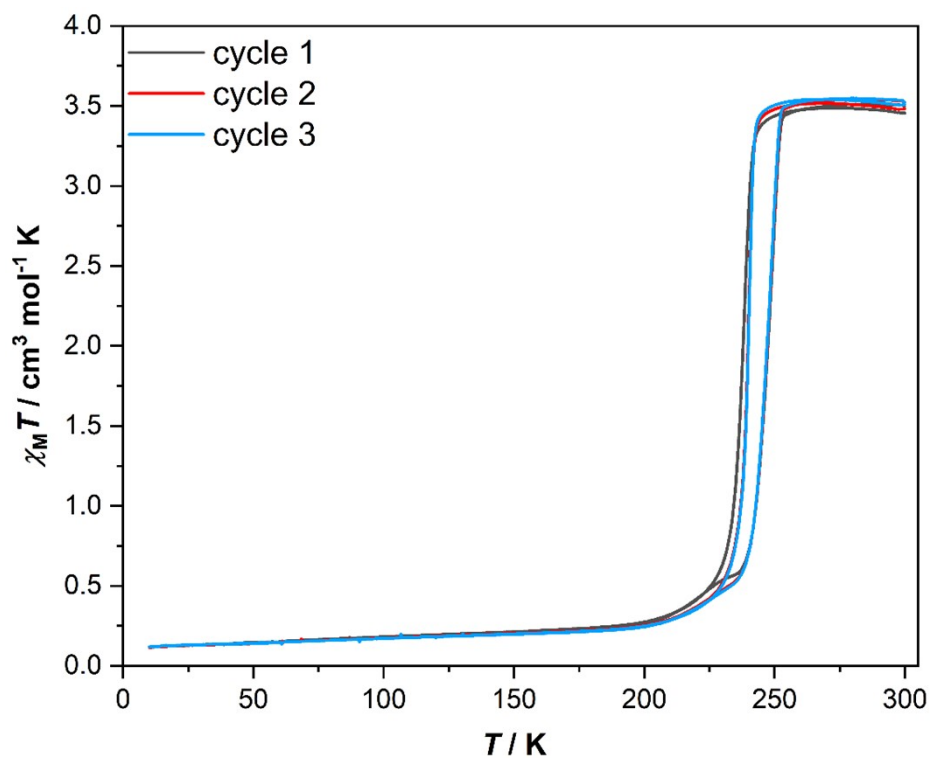


Figure S18. The temperature-dependent magnetic susceptibility data of **1** with a sweep rate of 2 K min^{-1} .

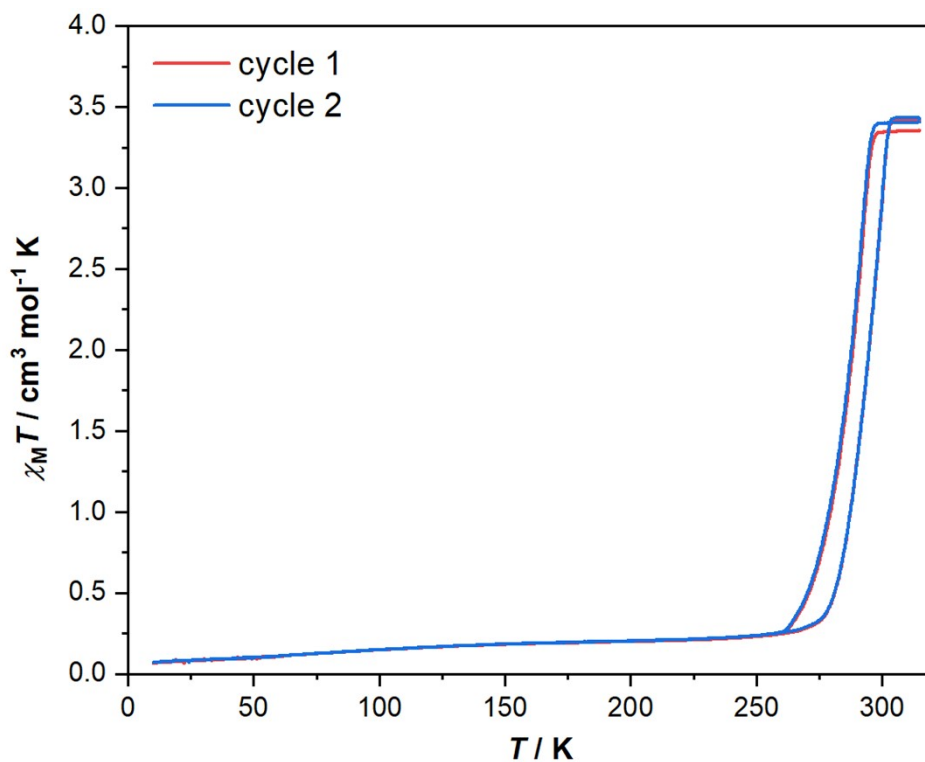


Figure S19. The temperature-dependent magnetic susceptibility data of **2** with a sweep rate of 2 K min⁻¹.

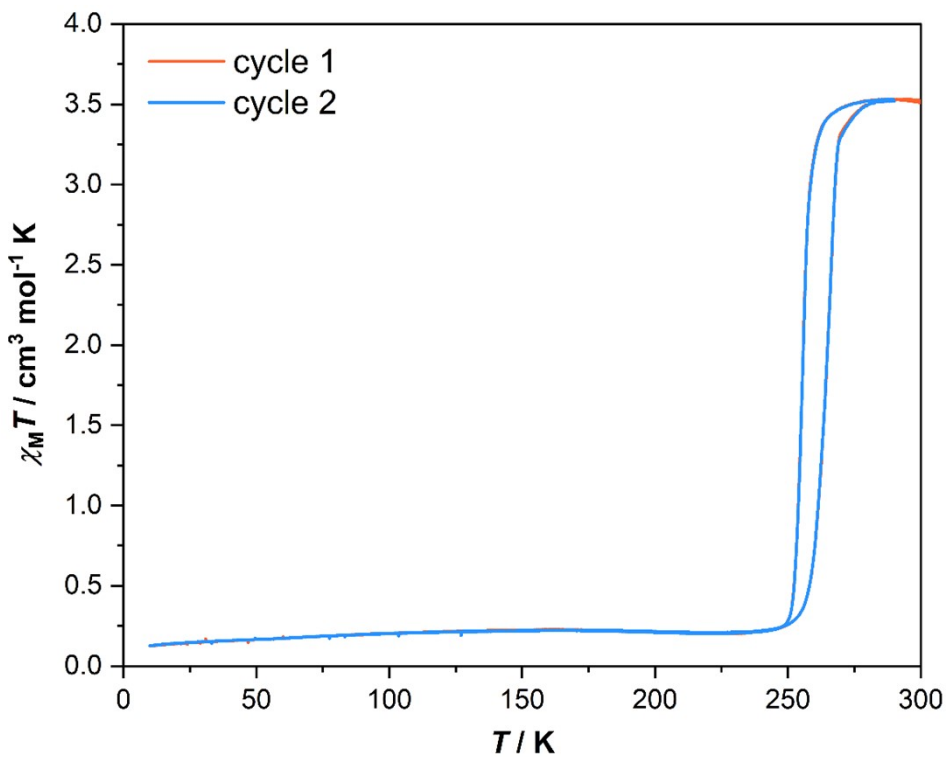


Figure S20. The temperature-dependent magnetic susceptibility data of **3** with a sweep rate of 2 K min⁻¹.

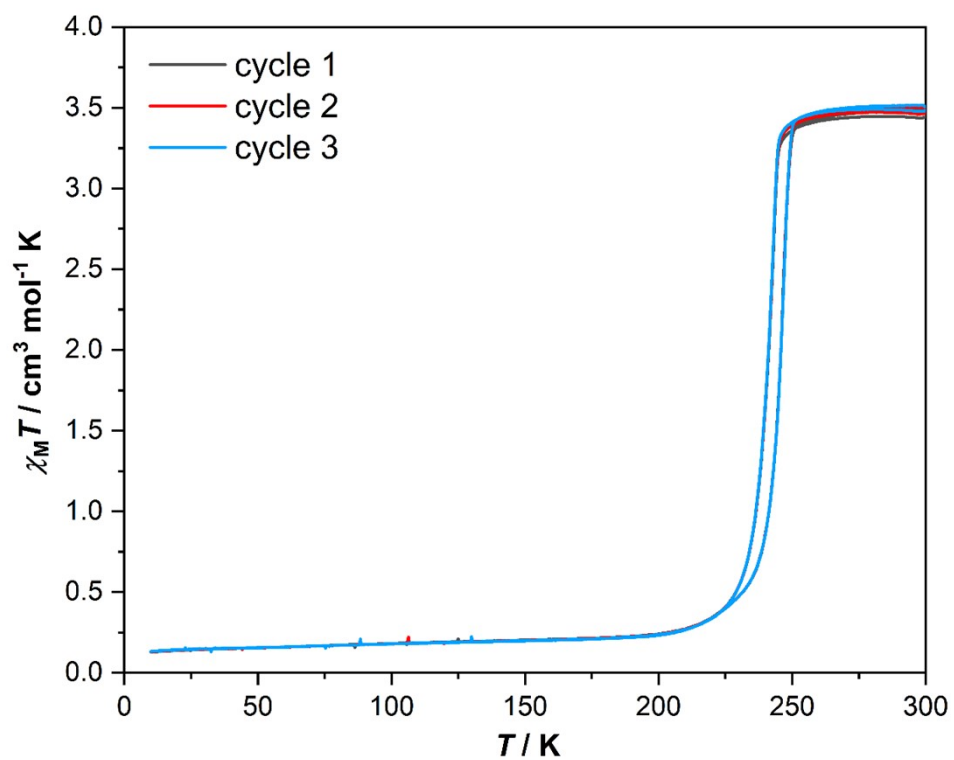


Figure S21. The temperature-dependent magnetic susceptibility data of **4** with a sweep rate of 2 K min^{-1} .