

Supporting Information

CdS-POM nanosheet for highly efficient visible-light-driven H₂ evolution

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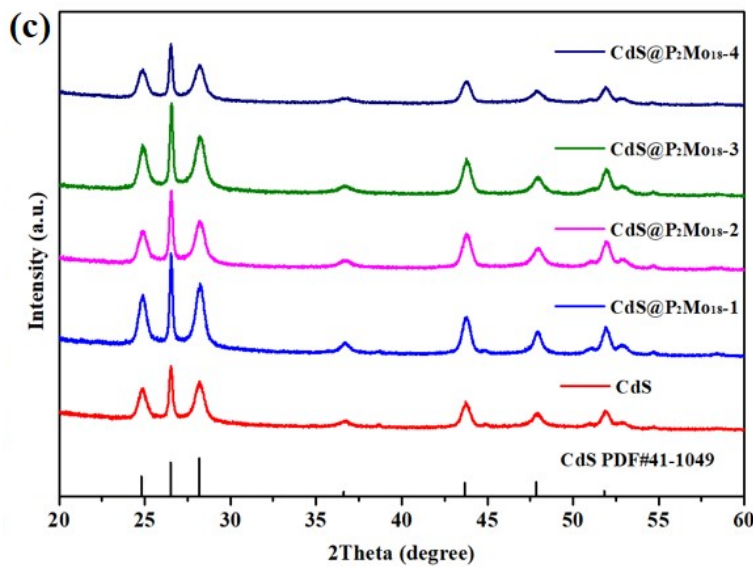
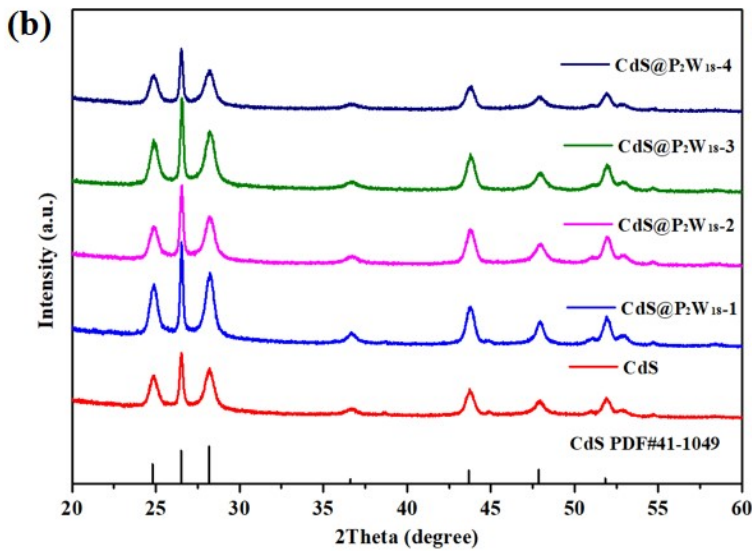
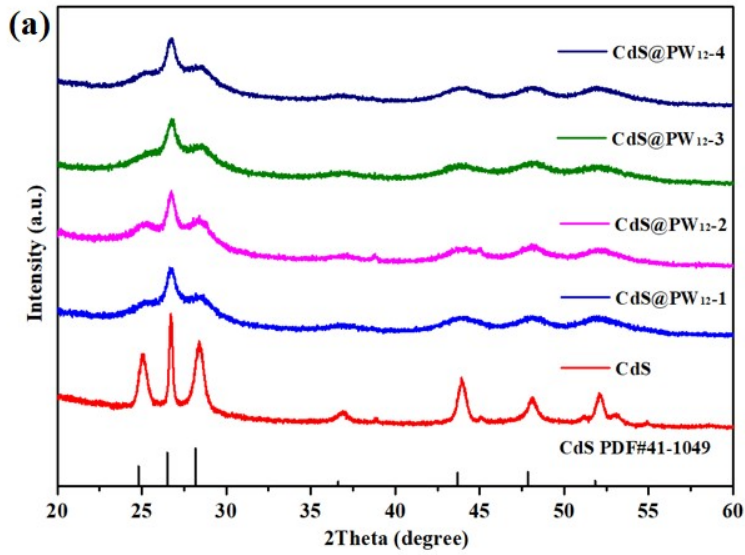


Figure S1. XRD patterns of the different POM doped into CdS. (a) XRD patterns of CdS@PW₁₂ composite photocatalysts. (b) XRD patterns of CdS@P₂W₁₈ composite photocatalysts. (c) XRD patterns of CdS@P₂Mo₁₈ composite photocatalysts.

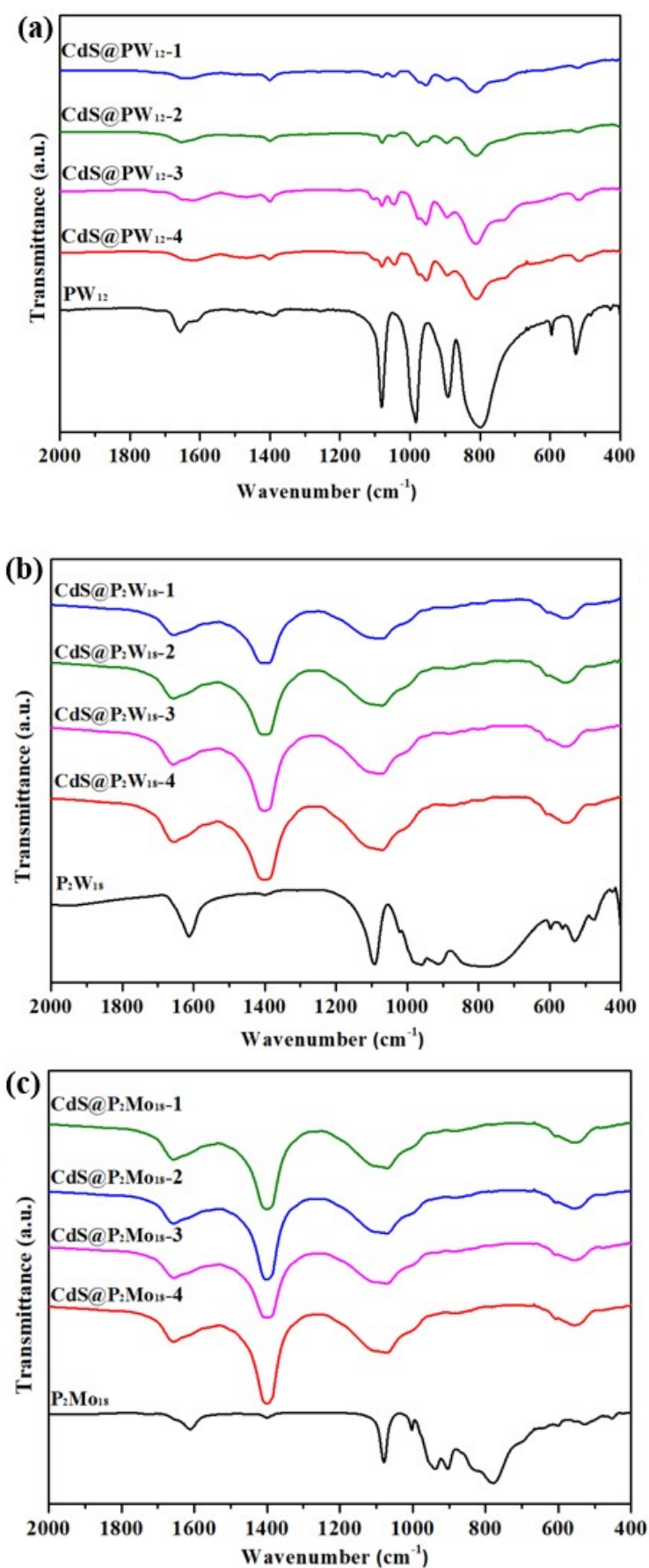


Figure S2. (a) FT-IR spectra of CdS@PW₁₂ composite photocatalysts. (b) FT-IR spectra of CdS@P₂W₁₈ composite photocatalysts. (c) FT-IR spectra of CdS@P₂Mo₁₈ composite photocatalysts.

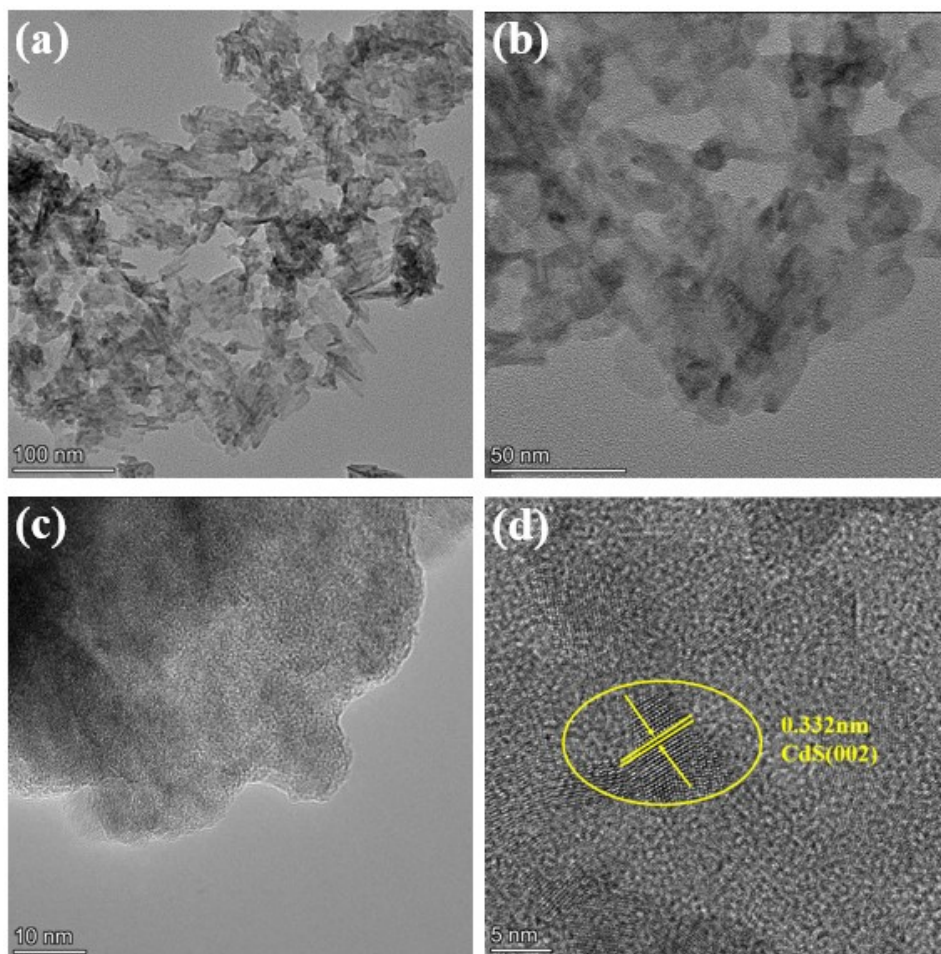
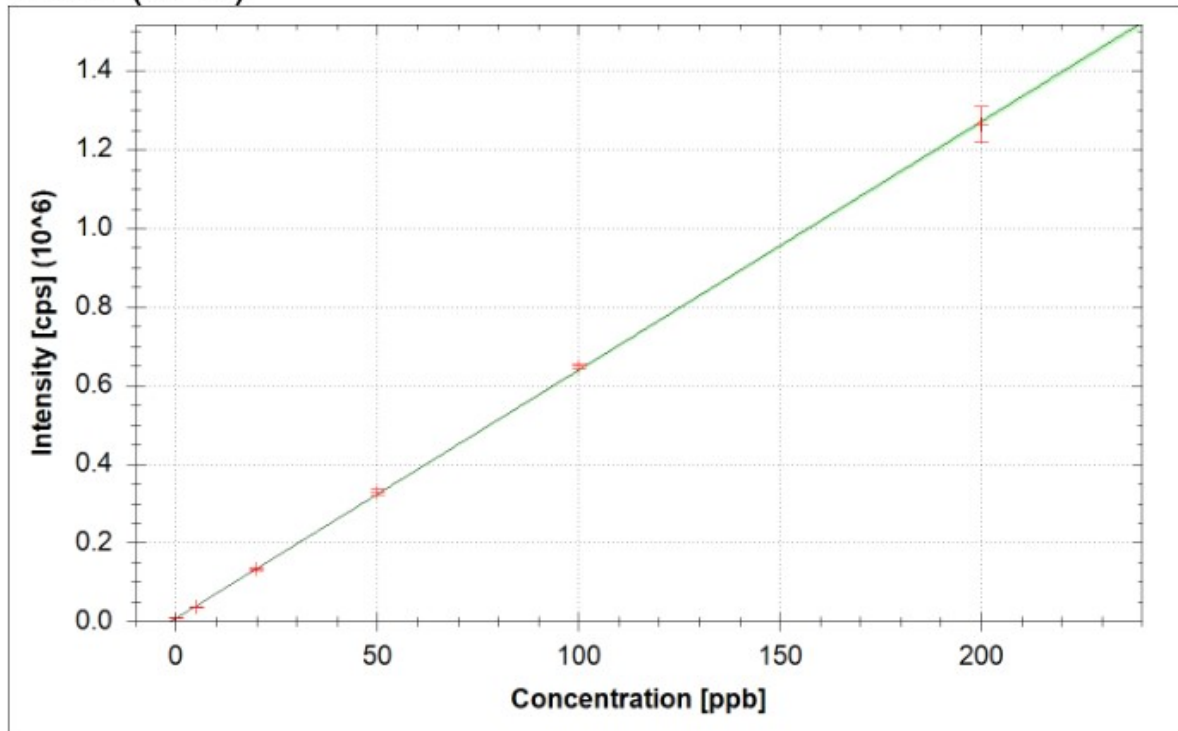


Figure S3. (a) TEM of CdS and (b-d), HRTEM images of CdS.

95Mo (KED)



$$f(x) = 6314.5539 \cdot x + 7972.5991$$

$$R^2 = 0.9999$$

$$\text{BEC} = 1.263 \text{ ppb}$$

$$\text{LoD} = 0.2354 \text{ ppb}$$

Figure S4. Calibration curves of Mo elements for ICP-MS.

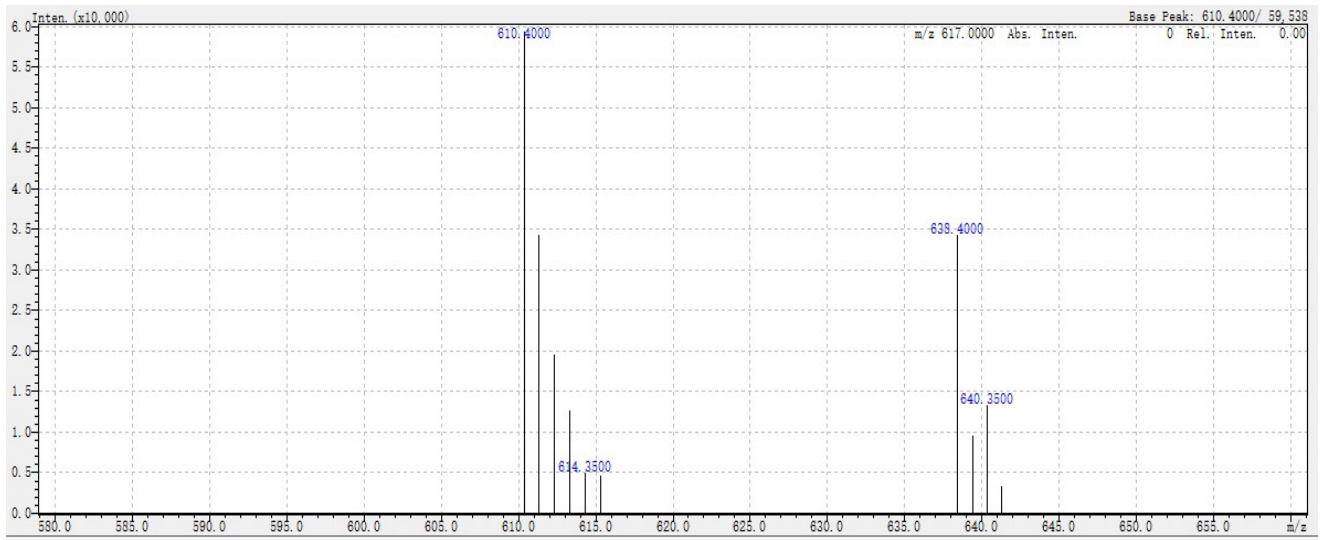


Figure S5. ESI-MS spectrum of PMo_{12} .

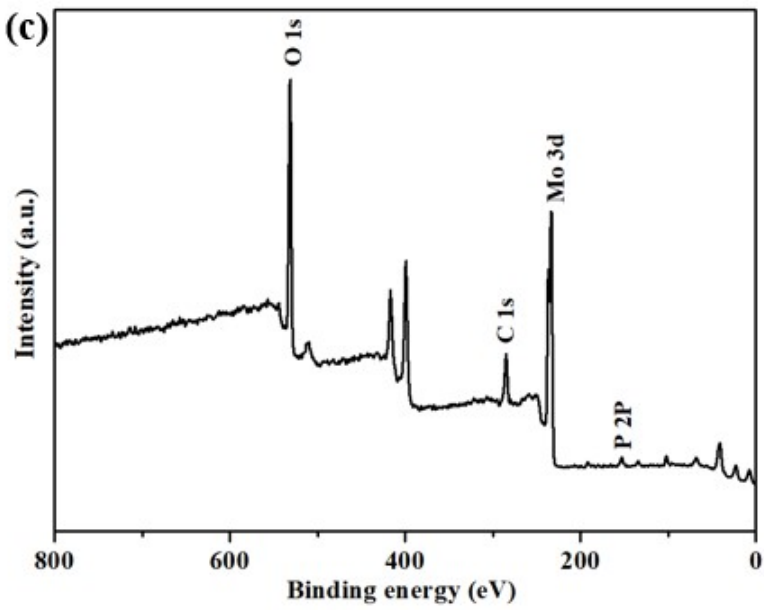
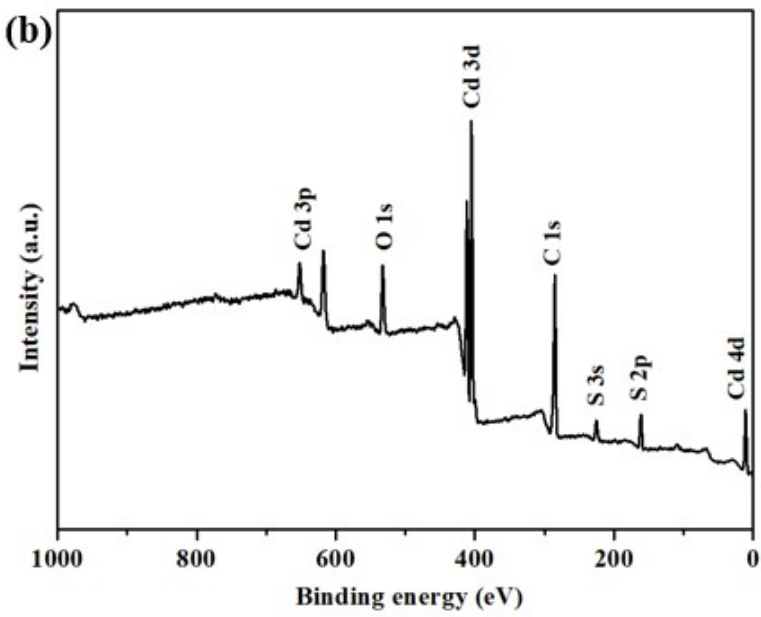
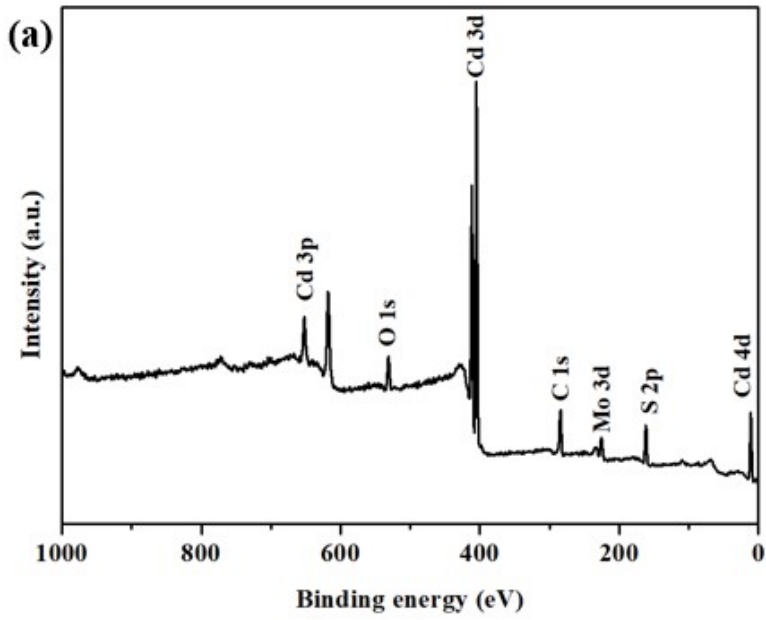


Figure S6. (a) The XPS survey spectrum of CdS@PMo₁₂₋₃. (b) The XPS survey spectrum of CdS. (c) The XPS survey spectrum of PMo₁₂.

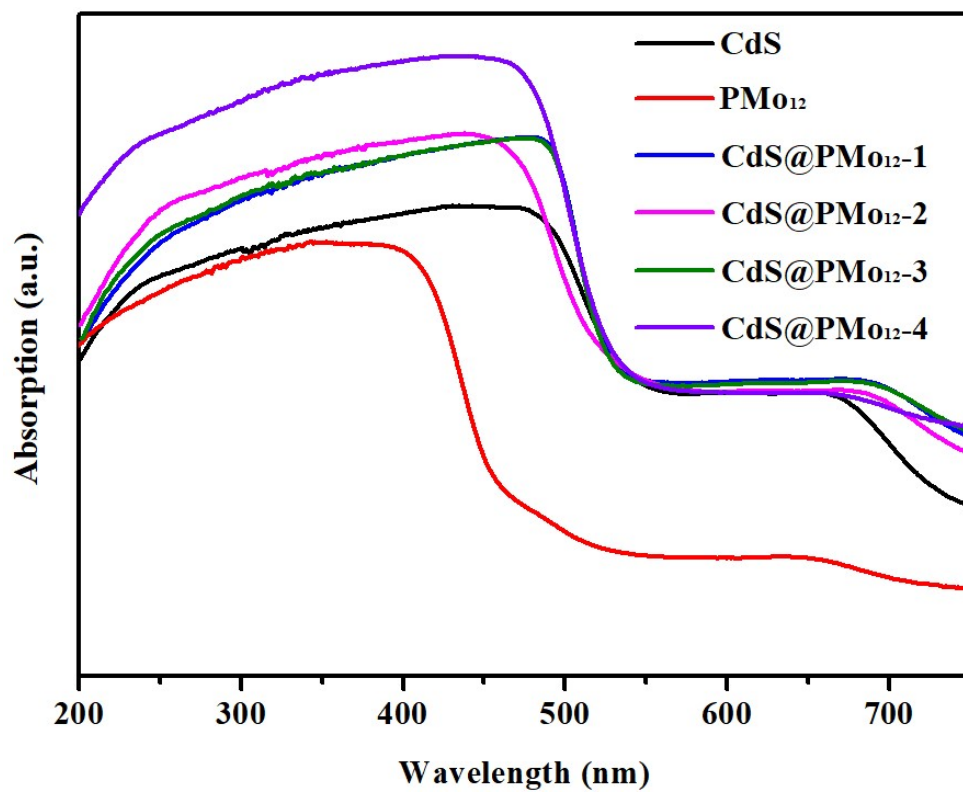


Figure S7. Ultraviolet-visible diffuse reflectance spectra of CdS, PMo₁₂ and CdS@PMo₁₂.

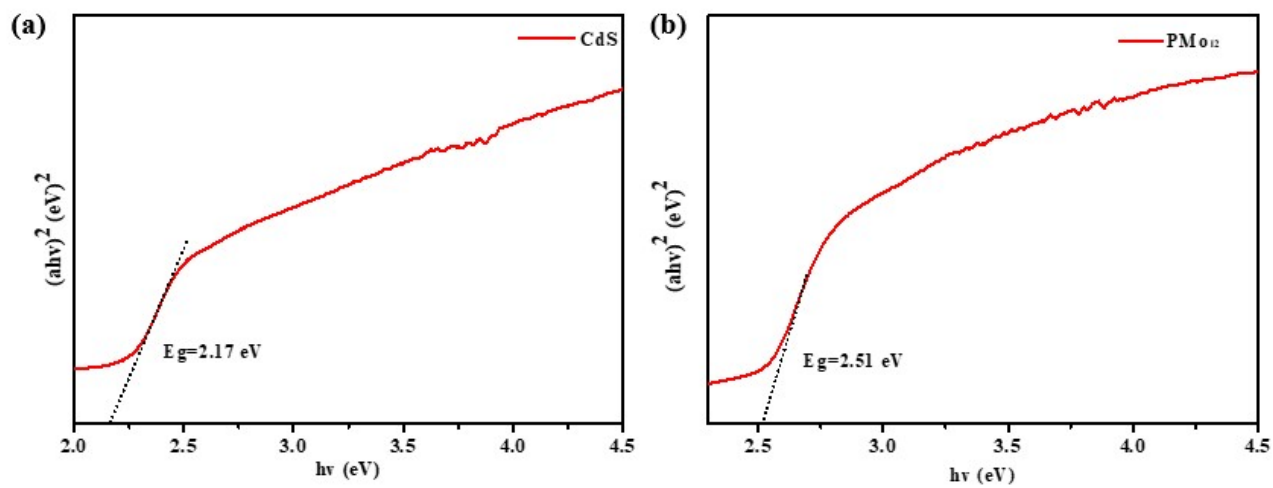


Figure S8. (a) Band-gap calculation of CdS. (b) Band-gap calculation of PMo₁₂.

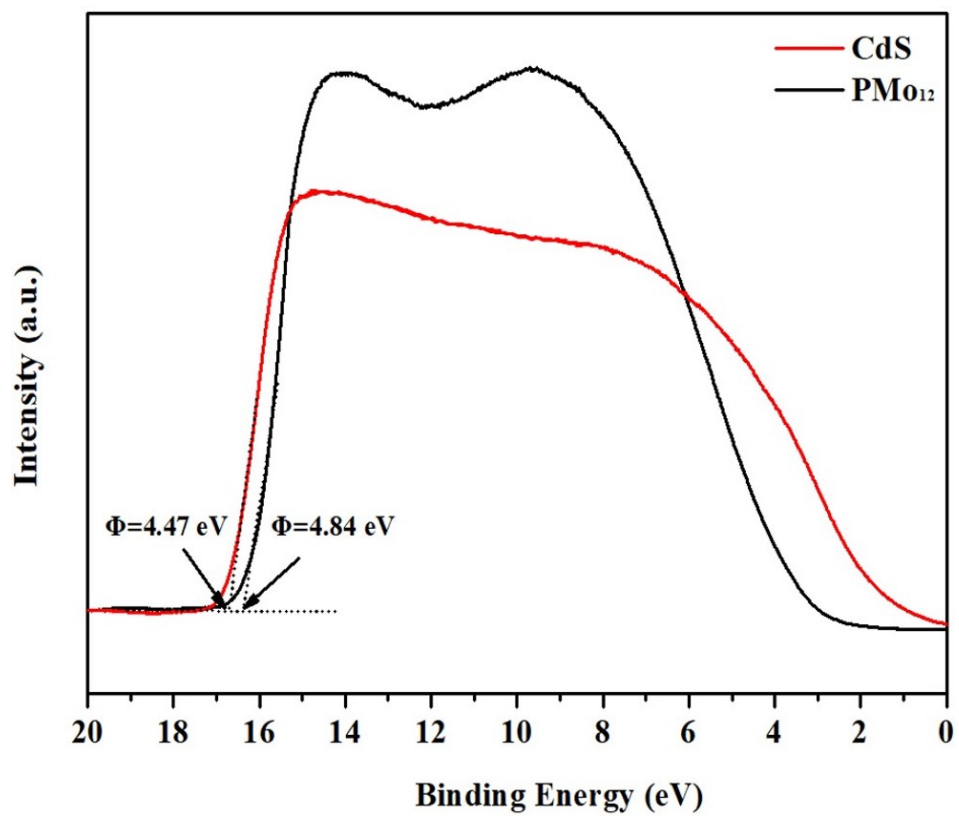


Figure S9. UPS spectra of CdS and PMo₁₂.

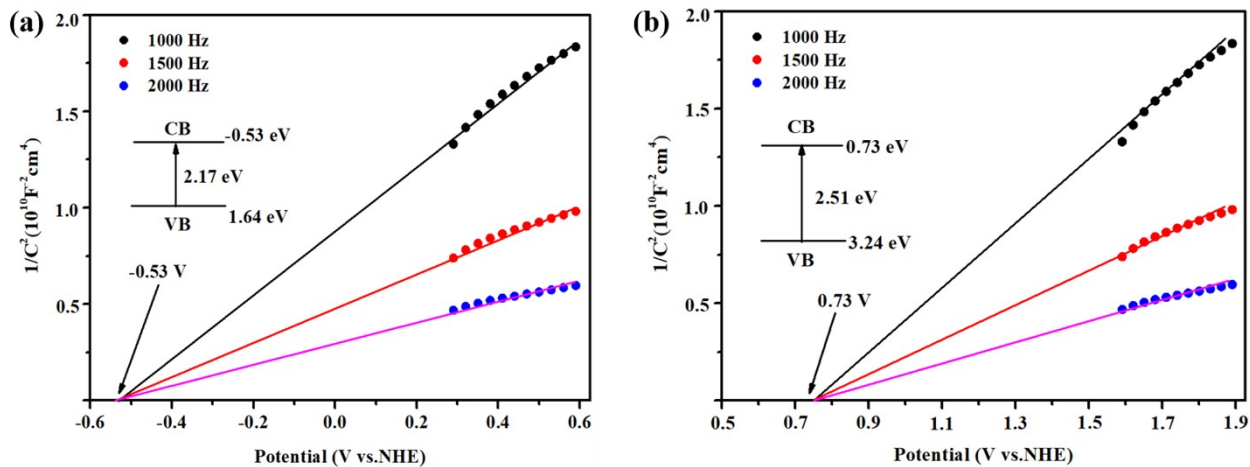


Figure S10. (a) Mott-Schottky diagram of CdS. (b) Mott-Schottky diagram of PMo₁₂.

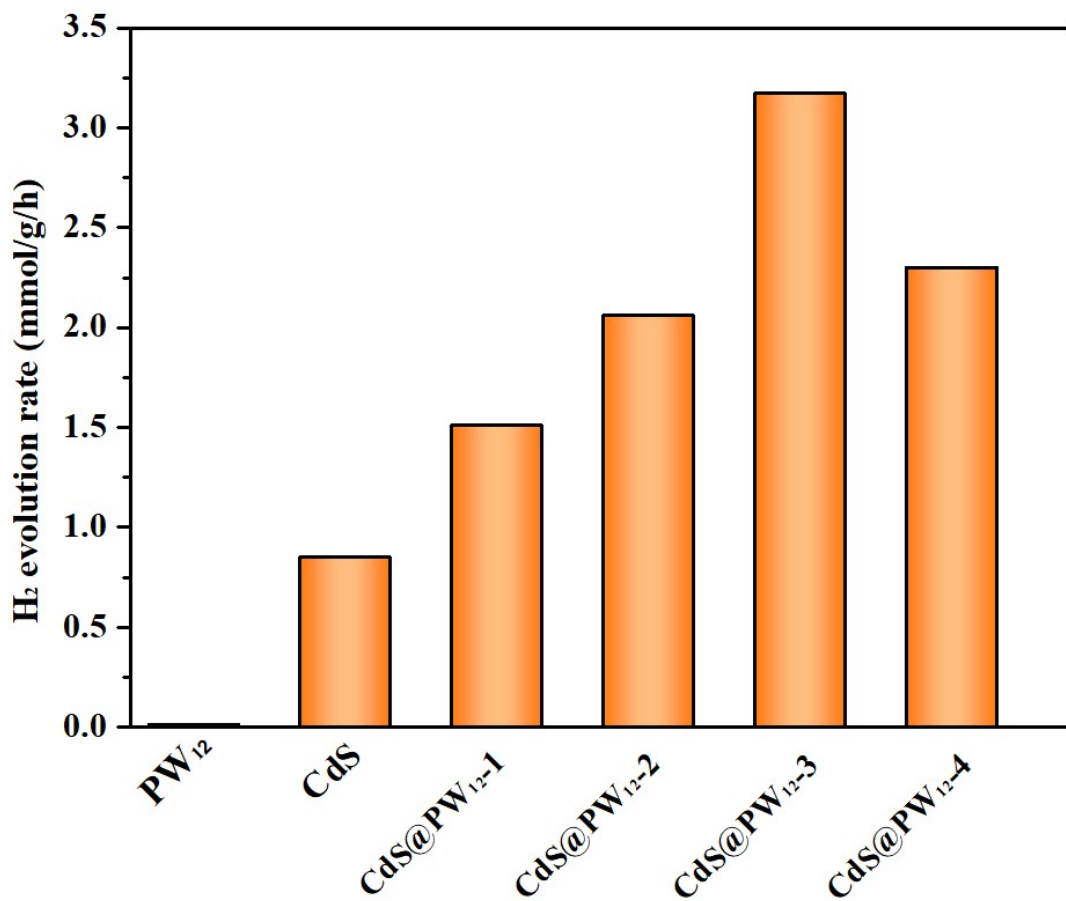


Figure S11. A comparison of the visible-light-driven H₂ evolution rates based on CdS, PW₁₂ and CdS@PW₁₂ as photocatalysts ($\lambda > 420$ nm).

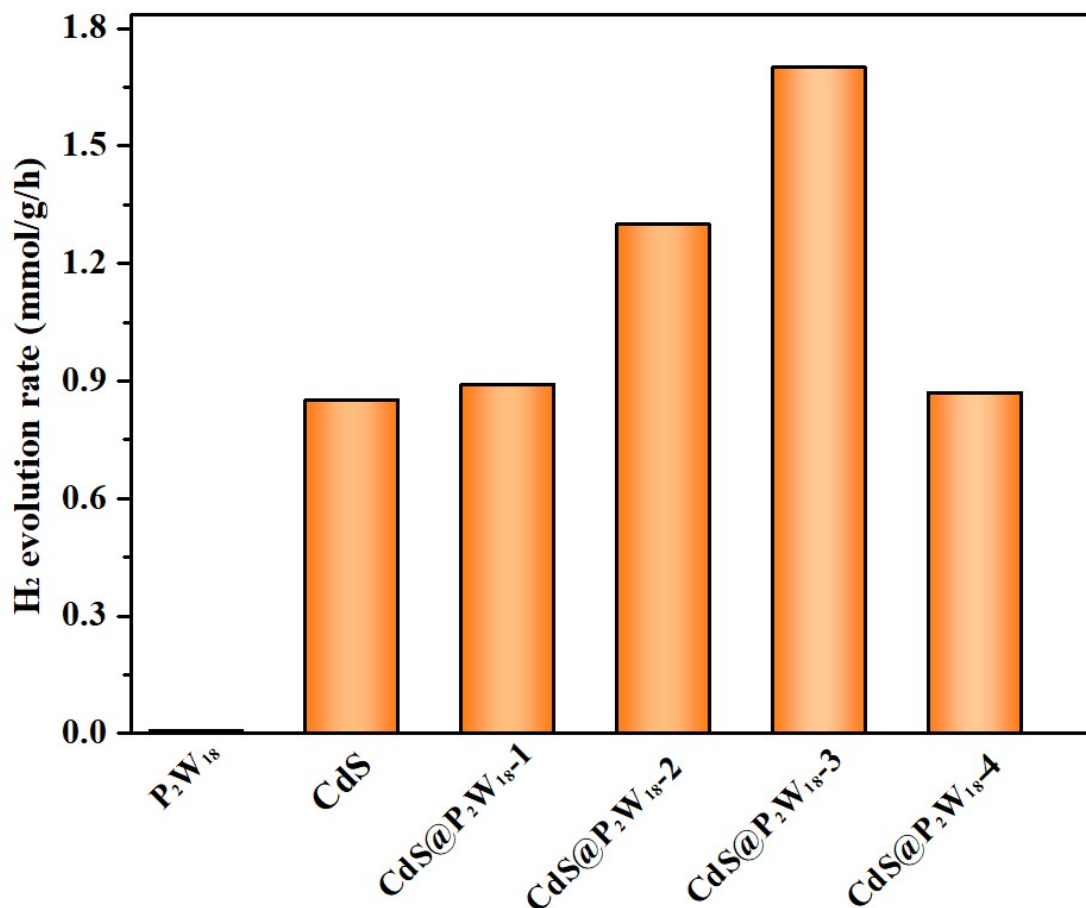


Figure S12. A comparison of the visible-light-driven H₂ evolution rates based on CdS, P₂W₁₈ and CdS@P₂W₁₈ as photocatalysts ($\lambda > 420$ nm).

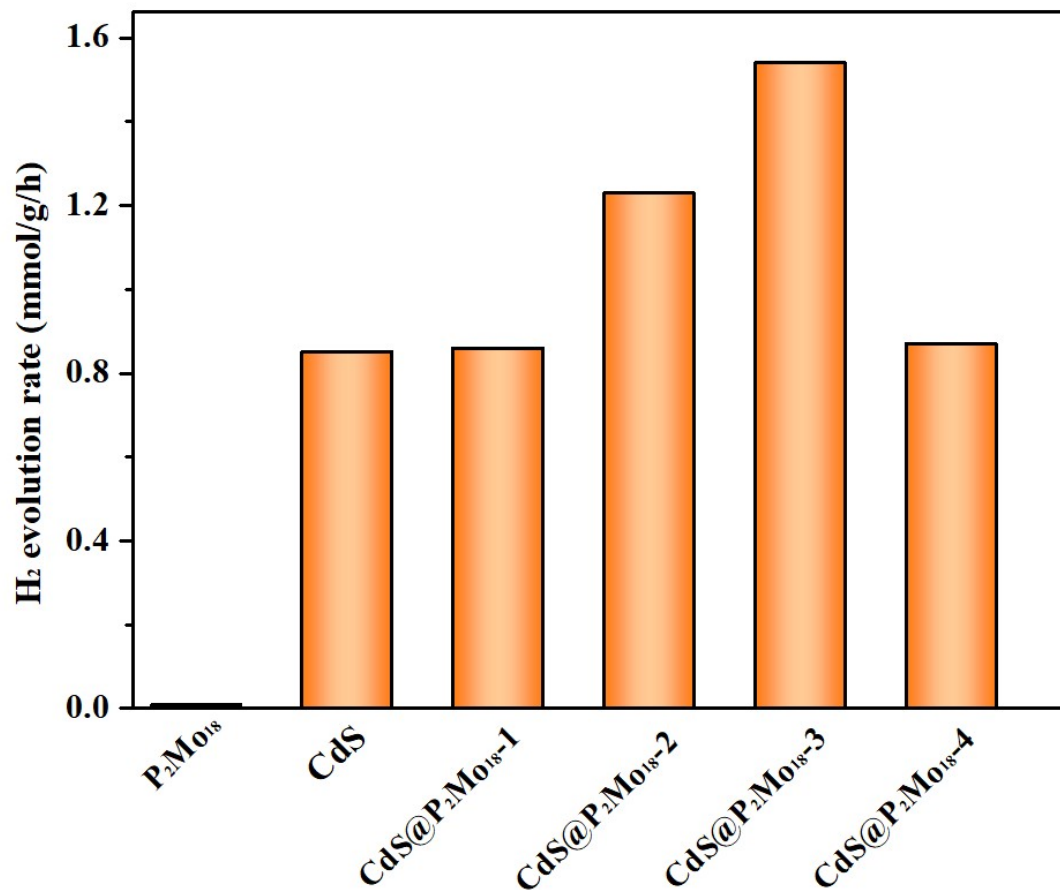


Figure S13. A comparison of the visible-light-driven H₂ evolution rates based on CdS, P₂Mo₁₈ and CdS@P₂Mo₁₈ as photocatalysts ($\lambda > 420$ nm).

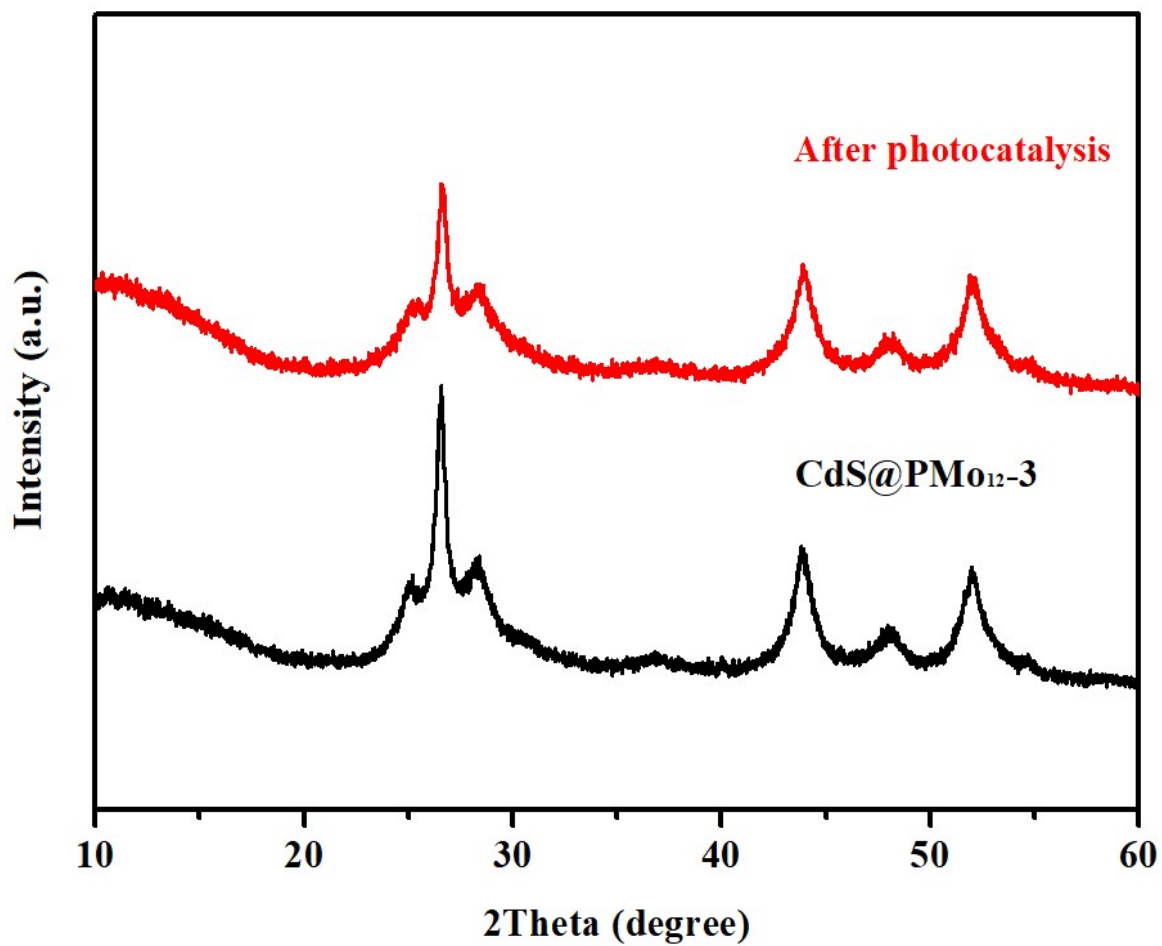


Figure S14. PXRD patterns of CdS@PMo₁₂₋₃ before and after the photocatalytic H₂ evolution.

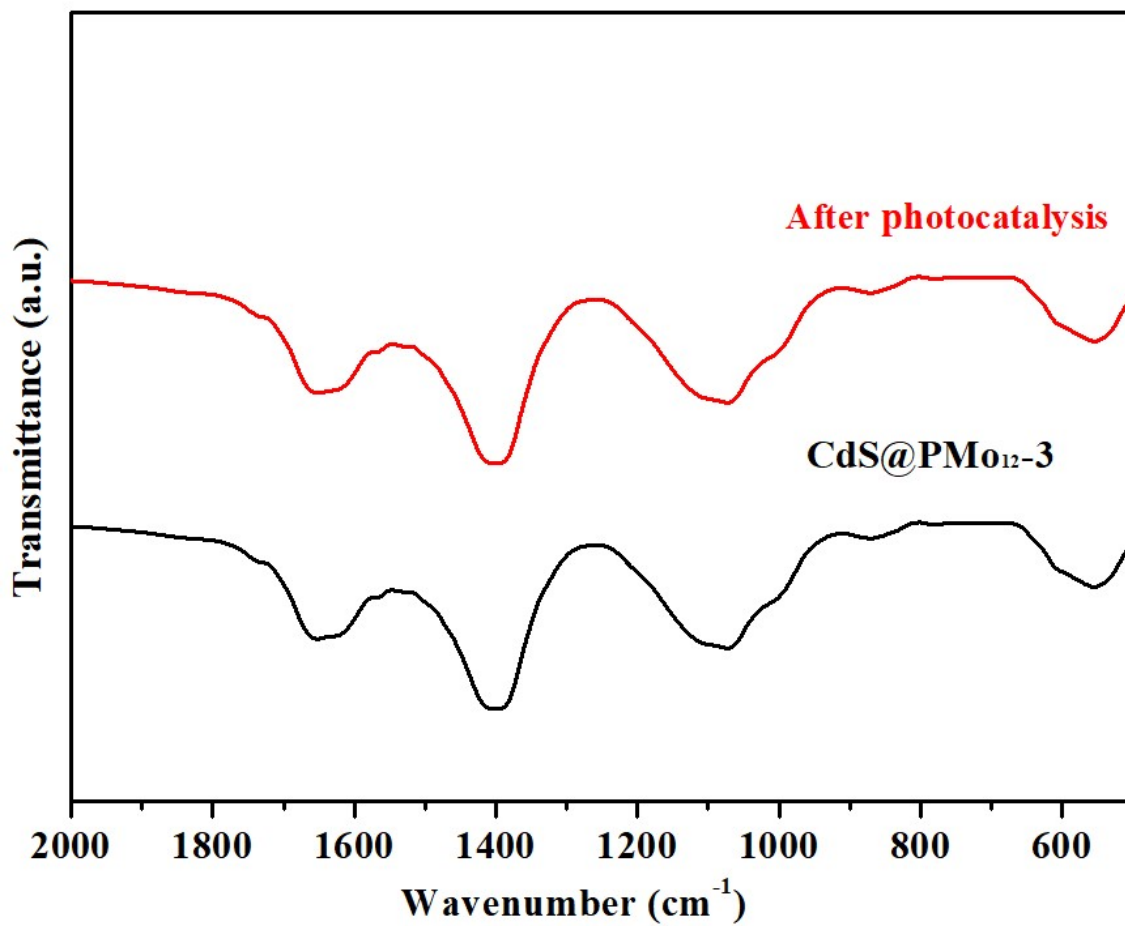


Figure S15. FT-IR spectra of CdS@PMo₁₂-3 before and after the photocatalytic H₂ evolution.

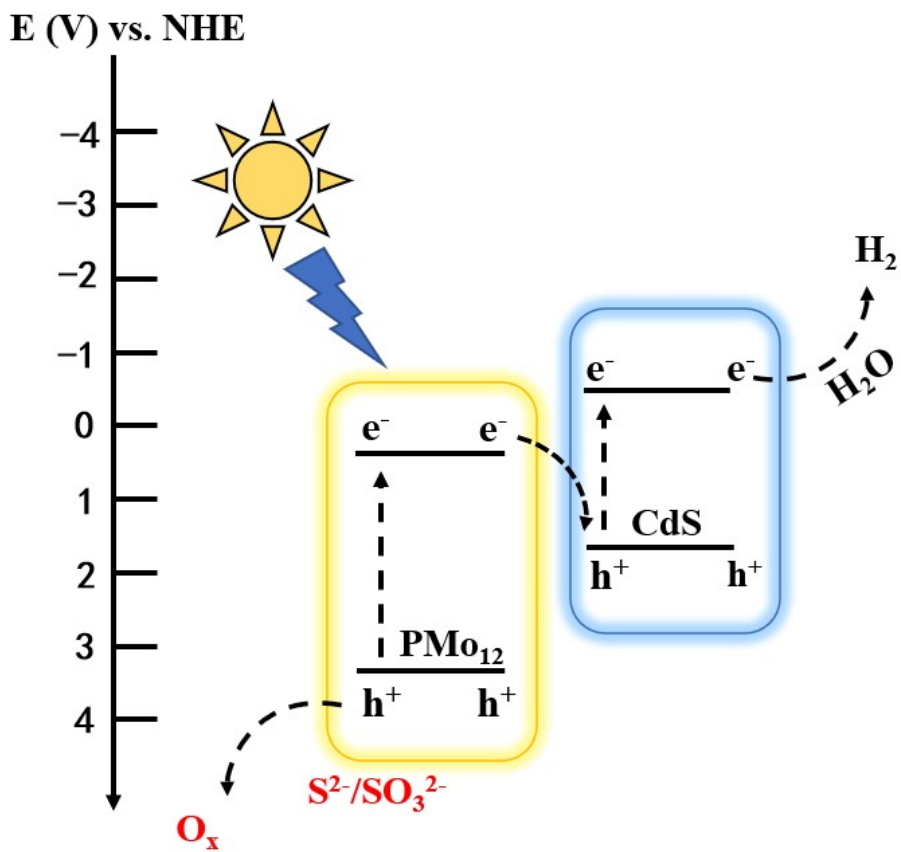


Figure S16. Proposed possible mechanism schematic of CdS@PMo₁₂ composite.

Table S1. ICP-MS results of CdS@PMo₁₂.

| Compounds | PMo ₁₂ (mg) | Elements Mo (wt%) |
|--------------------------|------------------------|----------------------|
| CdS@PMo ₁₂ -1 | 10 | 0.7 |
| CdS@PMo ₁₂ -2 | 30 | 1.1 |
| CdS@PMo ₁₂ -3 | 50 | 1.8 |
| CdS@PMo ₁₂ -4 | 70 | 2.7 |

Table S2. A comparison of the H₂-production rates of the POM-based photocatalysts and most CdS-based photocatalysts.

| Photocatalyst | Amounts of catalyst(mg) | Activity (mmol h ⁻¹ g ⁻¹) | Refs |
|--|-------------------------|--|-----------|
| CdS@PMo ₁₂ -3 | 2 | 5.7 | This Work |
| CdS-P ₅ W ₃₀ | 50 | 1.9 | 1 |
| Ti ₃ C ₂ T _x /CdS | 20 | 0.8 | 2 |
| CdS-NiE-350 | 100 | 4.3 | 3 |
| NiS/CdS | 30 | 2.2 | 4 |
| M-G/CdS | 100 | 2.3 | 5 |
| Au-Pt-CdS | 50 | 0.8 | 6 |
| CdS-NRs/NMOF-Ni | 50 | 4.5 | 7 |
| Ag ₂ S/CdS | 50 | 0.7 | 8 |
| MnO ₂ @CdS | 5 | 3.9 | 9 |
| P ₂ W ₁₅ V ₃ @MIL-101 | 3 | 0.9 | 10 |
| Co-POM | 1 | 2.0 | 11 |

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