# Supporting Information

# Functional Group Pair Distance based Descriptor for Isomerisation in Porous Molecular Framework Materials

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#### S0: Motivation

Functional Group Isomerism in Porous Molecular Materials: Describing Everything, Everywhere, All At Once

Predicting something (such as an adsorbate binding energy) for a 0-3D porous material typically involves the selection and modelling of a single isomer. Depending on the size and complexity of the adsorbate, the isomer(s) chosen may not be ideal or representative. Therefore, we would like to model adsorption for all structures in a topology (e.g. all possible isomers of a tetrahedral pore). This is our motivation for this work. If we consider two cases:

#### Case 1: 3D-periodic MOFs/COFs

From the crystal structure (Refcode: RUBTAK), the chemical formula for UiO-66 is  $C_{48}H_{28}O_{32}Zr_6$  and Z=4, so the molecular weight of one unit cell of UiO-66 is 6656.24 g/mol  $\therefore$  1g of UiO-66 = 1.5023 x 10<sup>4</sup> mol  $\therefore$  9.047 x 10<sup>19</sup> unit cells / g. 24 linkers per unit cell:  $\therefore$  2.171 x 10<sup>21</sup> linkers / g

Each UiO-66 unit cell contains 4 octahedral pores (which can be generated by coordinates (0.5, 0.5, 0.5)) and 8 tetrahedral pores (coordinates (0.25, 0.25, 0.25)).

This means that in 1g of UiO-66, there are:

And  $4 * 9.047 \times 10^{19} = 3.169 \times 10^{20}$  octahedral pores 8 \* 9.047 x 10<sup>19</sup> = 7.238 x 10<sup>20</sup> tetrahedral pores

Dividing these (very large!) numbers by the number of isomers of the octahedral (351976) and tetrahedral pores (176) gives the number of times we'd expect to see each pore isomer occurring in 1g of UiO-66:

=  $1.028 \times 10^{15}$  times for the octahedral pore =  $4.112 \times 10^{18}$  times for the tetrahedral pore

So in 1g of material, which seems like a fairly modest amount if we are considering an actual application, our guest molecule / adsorbate will see **every** pore isomer **many, many** times and therefore to model adsorption of some molecule in UiO-66 requires somehow accessing the "average" pore, somehow summed over every possible pore isomer – which is what we attempt here.

• Quick order-of-magnitude check / aside / reference: The original Lillerud UiO-66 paper (DOI:<u>10.1021/ja8057953</u>) used 0.227 mmol of both BDC and ZrCl<sub>4</sub> in their synthesis vs 0.15 mmol of MOF assumed above.)

#### Case 2: 0D-periodic cages (MOPs/POCs)

The analysis above applies similarly to 0D cages, with the following caveats:

- For the single and simple type of isomerism considered here (i.e. a linker flip), it is hard to see any entropic reason to end up with only one isomer, especially if the cages were in dilute solution.
- In a crystallisation process there could be a possibility of say, forcing (e.g. the limiting cases of) an "all FG in" or "all FG out" isomer, but this would still leave two choices (out of the original four) for each functional group, and therefore a significant (topology dependent) number of isomers w.r.t the inside of the pore/cage, even where the outside is fully defined.

We intend our descriptor to apply equally whether the pore is discrete (i.e. a MOP/POC) or part of a network (i.e. a MOF/COF), and also regardless of the identity of the connector.

#### Uniqueness:

The *x* (e.g. for tetrahedral 176) individual histograms are, by definition, unique – if every FG-FG distance was the same in two isomers *n* and *m*, then some symmetry operation(s) would map *n* onto *m*.

For most applications (e.g. adsorption) it is probably the blue / In-In part of our histograms, describing the environment strictly inside the pore, that is most relevant.

We have chosen not to delete the violet (In-Out) and red (Out-Out) parts of our histograms. By so doing, we are in essence leaving the machine learning algorithm to decide what is important. We think this is the safest choice because at this point, we have not defined our chemical use case:

- If the chemical use case involves 0D cages (POCs/MOPs), the In-In histogram should indeed define the pore for an adsorption problem, but if you were considering crystallization or aggregation as well, then the "out" parts may become important
- For the 3D (COF/MOF) case, it is arguably more subtle: We can imagine that, for example, UiO-66 could be represented as the (possibly weighted) sum of tetrahedral (Tri4Di6) and octahedral (Tet6Di12) histograms, but we know that each pore is not independent of its neighbours (i.e. a FG pointing in to one pore, must be out of the neighbouring pore). Given the incredibly large numbers of pores however (i.e. ~ 10<sup>20</sup> for 1g of material, worked above), we contend that this *local* effect can be ignored.

If we did decide to consider the In-In FG-FG distances only... then it is clear that there are some number of isomers (when considering all FG), that yield the same inner pore structure – which is probably what a guest molecule would consider to be the "same environment".

To illustrate, and to test whether the histogram defines the environment, we've worked this through in full for a  $D_{4h}$  square, deliberately ignoring symmetry / point group operations and exhaustively building up each isomer and from there testing uniqueness. The python code (as a jupyter notebook), and the output (data as xlsx, structures as xyz and figures/tables as pdf) are included in this SI.

#### In brief:

One substitution of each of the four linkers, yields  $4^4 = 256$  possible isomers, of which 39 are symmetry unique. If one considers only the inside pore environment, then this number reduces to 15.

Number of FG		Total symmetry-	Unique inner-			
inside pore		unique isomers	pore isomers			
	0	4	1			
	1	8	1			
	2	15	5			
	3	8	4			
	4	4	4			

## S1 Number of isomers of each pore topology arising from different functional group arrangements

Pore topology	Number of Isomers
$Tri^2Di^3$	10
$Tri^4Di^6$	176
$Tri_2^4Di^6$	568
$Tri^6Di^9$	21856
$Tri^8Di^{12}$	351976
$Tri^{20}Di^{30}$	9607679214180672
$Tet^2Di^4$	31
$Tet_3^3Di^3$	366
$Tet_4^4Di^8$	4244
$Tet^5Di^{10}$	52740
$Tet^6Di^{12}$	354024
$Tet^8Di^{16}$	268439588
$Tet^{16}Di^{32}$	17592195482624
$Tet^{24}Di^{48}$	1.650586719047232e27

Table S1: Number of isomers arising from different arrangement of functional group position of pore cage structure, functionalisation is based on one functional group per linker.

## S2 Example of Unique Isomer Calculation Using Group Theory

we consider  $Tet^2Di^4$ , the pore has point group  $D_{4h}$  with 16 symmetry elements:

$$G = E, 2 \cdot C_4, C_2, 2 \cdot C'_2, 2 \cdot C''_2$$
$$i, 2 \cdot S_4, \sigma_h, 2 \cdot \sigma_v, 2 \cdot \sigma_d$$

and the number of possible structures is:

$$H = 4^{n_{linkers}} = 4^4 = 256$$

Then the total number, n of unique structures is calculated as:

$$n = \frac{1}{16} [\chi(E) + 2 \bullet \chi(C_4) + \chi(C_2) + 2 \bullet \chi(C'_2) + 2 \bullet \chi(C''_2) + \chi(i) + 2 \bullet \chi(S_4) + \chi(\sigma_h) + 2 \bullet \chi(\sigma_v) + 2 \bullet \chi(\sigma_d)] = 31$$
(S1)

with each term of  $\chi(x)$  of the symmetry elements equal to:

$$\begin{array}{ll} \chi(E) = 256 & \chi(C_4) = 4 \\ \chi(C_2) = 16 & \chi(C_2') = 16 \\ \chi(C_2'') = 4 & \chi(i) = 16 \\ \chi(S_4) = 16 & \chi(\sigma_h) = 0 \\ \chi(\sigma_v) = 0 & \chi(\sigma_d) = 64 \end{array}$$

#### S3 Algorithm of Isomer Enumeration

#### Step 1: Preparation

Step 1a: Labelling each functional group slot with an index.

In this step, the pore structures are installed with full functionalisation, which means all hydrogen atoms in the benzene linker are replaced by a functional group (FG). Then each functional group slot is labelled by an index named as FG\_id. FG\_id in the same linker need to be in a sequence of 4 consecutive numbers, as shown in Figure S1.



Figure S1: Pore  $Tet^2Di^4$  with assigned index for each functional group, functional group slot is shown in blue circles.

Step 1b: Generating a transformation library.

After indexing each FG slot, symmetry operations are applied to the pore. Again,  $Tet^2Di^4$  is used as an example,  $Tet^2Di^4$  has point group of  $D_{4h}$  with 16 symmetry elements. Therefore, the pore is transformed 16 times. After every transformation, the new structure is compared to the original pore structure to list the FG\_id position of the new structure, this information is stored in a table which we refer to as the transformation library (will be used in the next step). In the transformation library, the first column represents the initial position of FG. The columns after are the transformed (and projected) FG\_id for each symmetry operation. For the Tet2Di4 pore, the transformation library will be a table of dimension 16x16 (Figure S2). The 16 rows correspond to the 16 possible FG position, and the 16 columns correspond to the 16 symmetry elements.



Figure S2: (right) Symmetry elements of pore  $Tet^2Di^4$  (only 5 from the 16 are shown) (left) Transformation library, library of each functional group index and the transformed index, the first column represents the initial position, each subsequent column represent a specified symmetry operation, double lines represent hidden rows.

Generating a transformation library beforehand is useful for tracking the possible symmetry transformations of the pore, specifically in cutting down repetitive transformations (of the pore) each of which involve multiplication of the xyz coordinate matrices and transformation matrices which can became highly computationally costly.

#### Step 2: Enumeration of all possible functional group arrangement.

Using the example of  $Tet^2Di^4$  pore, as each linker is installed with one functional group, the pore isomers will have 4 functional groups. The isomers are encoded to a string of 4 FG\_id and also encoded to an equivalent single integer isomer\_id, defined by Equation S2. All possible combinations of FG\_id are listed (an example is shown in S3). Along with that, a unique flag list is prepared to mark the unique isomers Figure 5.10.

$$isomer\_id = \sum_{FG\_id} (FG\_id\%4) \bullet 4^{int\left(\frac{FG\_id}{4}\right)}$$
(S2)

lsomer_Id	FG_id					Unique?				
0	0	4	8	12		yes/no				
1	1	1 4 8 12			yes/no					
2	2	4	8	8 ¦ 12	12		yes/no			
3	3	4	8	12		yes/no				
4	0	5	8	12		yes/no				
5	1	¦ 5	8	12		yes/no				
6	2	5	8	12		yes/no				
255	3	7	11	¦ 15		yes/no				

Figure S3: Example of functional group enumeration unique list

Step 2a: Finding equivalent symmetry related isomers.

For each isomer, equivalent symmetry related isomers are determined by looking at the transformation library generated in step 2. For example, as illustrated in Figure S4, in the first isomer (isomer\_id=0), functionalisation is located at FG\_id 0-4-8-12, so the FG\_id rows are highlighted in green. Then, the highlighted green rows are copied into a new matrix and transposed. Each of the rows from the transpose matrix are the equivalent isomers.

E	<b>C</b> <sub>4</sub>	C2	<b>C</b> 2′		C <sub>2</sub> "	$\sigma_{h}$					
0	14	10	1	1	5	1					
1	15	11	0	1 1 1	4	0		Equiva	lent iso	omers:	
2	12	8	3		7	3		14	0	6	8
3	13	9	2	i I	6	2	┝→	10	12	2	6
4	0	12	15		1	5		1	15	9	7
8	6	2	9	   	13	9					
12	8	6	7		9	13		5	1	13	9
15	11	5	4	1 1 1	10	14		1	5	9	13

Figure S4: Illustration of equivalent isomers determined from transformation list

Step 2b: Marking the unique flag list.

As illustrated in Figure S5, Isomer\_id 0 will be marked as unique in the unique flag list, while the equivalent isomers are marked as a duplicate. To search for the Ids of the equivalent isomers, the string of 4 FG\_ids is converted back to its isomer\_Id by Equation S2. Steps 2a and 2b are repeated for the next unsigned unique flag list isomer until all isomer\_id in the unique flag list are marked.

					lsomer_ld	FG_id				Unique?
					0	0	4	8	12	yes
					1	1	4	8	12	
					2	2	4	8	12	
					3	3	4	8	12	
					4	0	¦ 5	8	¦ 12	
					5	1	5	8	12	
Equivalent isomers:			lsomer_id	6	2	5	8	12		
14	0	6	8	136	42	2	6	10	12	no
10	12	2	6	42	85	1	5	9	13	no
1	15	9	7	221	136	0	6	8	¦ 14	no
					221	1	7	9	15	no
5	1	13	9	85						
1	5	9	13	85	255	3	¦ 7	, 11	¦ 15	

Figure S5: Example of functional group enumeration unique list

In huge pores with more than 12 ditopic linkers, the number of isomers exceeds the maximum size of a python list. Encoding and decoding the isomers has significantly reduced the computational load. The encoding also cuts the computational time that would be required to search for string containing  $n_{FG}$  numbers in a list of  $4^{nlinker}$  isomers.

# S4 Enumeration and analysis of 2D $D_{4h}$ cage































Isomer 17



## Isomer number 0 Number of FG in pore: 4 Isomer FG-FG distances (Å):

5.8919 8.3324 5.8919 5.8919 8.3324 5.8919



#### Isomer number 1 Number of FG in pore: 4 Isomer FG-FG distances (Å):

5.8919 8.3324 7.5595 5.8919 7.8236 3.5048





### Isomer number 2 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 8.3324 11.6104 5.8919 12.7870 7.8447





### Isomer number 3 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 8.3324 10.6034 5.8919 13.1037 9.1602





## Isomer number 5 Number of FG in pore: 4 Isomer FG-FG distances (Å):

5.8919 7.8236 7.5595 3.5048 7.8236 5.8919





## Isomer number 6 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 7.8236 11.6104 3.5048 12.7870 10.6034





## Isomer number 7 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 7.8236 10.6034 3.5048 13.1037 11.6104





## Isomer number 9 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 12.7870 7.5595 7.8447 7.8236 9.1602





## Isomer number 10 Number of FG in pore: 2 Isomer FG-FG distances (Å):

5.8919 12.7870 11.6104 7.8447 12.7870 12.7133





## Isomer number 11 Number of FG in pore: 2 Isomer FG-FG distances (Å):

5.8919 12.7870 10.6034 7.8447 13.1037 14.5733





### Isomer number 13 Number of FG in pore: 3 Isomer FG-FG distances (Å):

5.8919 13.1037 7.5595 9.1602 7.8236 7.8447





## Isomer number 14 Number of FG in pore: 2 Isomer FG-FG distances (Å):

5.8919 13.1037 11.6104 9.1602 12.7870 10.5297





## Isomer number 15 Number of FG in pore: 2 Isomer FG-FG distances (Å):

5.8919 13.1037 10.6034 9.1602 13.1037 12.7133





## Isomer number 17 Number of FG in pore: 4 Isomer FG-FG distances (Å):

3.5048 8.3324 7.5595 7.5595 8.3324 3.5048



## Isomer number 18 Number of FG in pore: 3 Isomer FG-FG distances (Å):

3.5048 8.3324 11.6104 7.5595 13.1037 7.8447





## Isomer number 19 Number of FG in pore: 3 Isomer FG-FG distances (Å):

3.5048 8.3324 10.6034 7.5595 12.7870 9.1602





## Isomer number 26 Number of FG in pore: 2 Isomer FG-FG distances (Å):

3.5048 12.7870 11.6104 10.6034 13.1037 12.7133




## Isomer number 27 Number of FG in pore: 2 Isomer FG-FG distances (Å):

3.5048 12.7870 10.6034 10.6034 12.7870 14.5733





## Isomer number 30 Number of FG in pore: 2 Isomer FG-FG distances (Å):

3.5048 13.1037 11.6104 11.6104 13.1037 10.5297





## Isomer number 34 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 8.3324 11.6104 11.6104 17.9793 7.8447





#### Isomer number 35 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 8.3324 10.6034 11.6104 17.7505 9.1602





#### Isomer number 38 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 7.8236 11.6104 9.1602 17.9793 10.6034





#### Isomer number 39 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 7.8236 10.6034 9.1602 17.7505 11.6104





## Isomer number 41 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 12.7870 7.5595 12.7133 13.1037 9.1602





### Isomer number 42 Number of FG in pore: 1 Isomer FG-FG distances (Å):

7.8447 12.7870 11.6104 12.7133 17.9793 12.7133





### Isomer number 43 Number of FG in pore: 1 Isomer FG-FG distances (Å):

7.8447 12.7870 10.6034 12.7133 17.7505 14.5733





## Isomer number 45 Number of FG in pore: 2 Isomer FG-FG distances (Å):

7.8447 13.1037 7.5595 14.5733 13.1037 7.8447





### Isomer number 46 Number of FG in pore: 1 Isomer FG-FG distances (Å):

7.8447 13.1037 11.6104 14.5733 17.9793 10.5297





### Isomer number 47 Number of FG in pore: 1 Isomer FG-FG distances (Å):

7.8447 13.1037 10.6034 14.5733 17.7505 12.7133





## Isomer number 51 Number of FG in pore: 2 Isomer FG-FG distances (Å):

9.1602 8.3324 10.6034 10.6034 17.9793 9.1602





## Isomer number 57 Number of FG in pore: 2 Isomer FG-FG distances (Å):

9.1602 12.7870 7.5595 10.5297 12.7870 9.1602





#### Isomer number 58 Number of FG in pore: 1 Isomer FG-FG distances (Å):

9.1602 12.7870 11.6104 10.5297 17.7505 12.7133





#### Isomer number 59 Number of FG in pore: 1 Isomer FG-FG distances (Å):

9.1602 12.7870 10.6034 10.5297 17.9793 14.5733





#### Isomer number 62 Number of FG in pore: 1 Isomer FG-FG distances (Å):

9.1602 13.1037 11.6104 12.7133 17.7505 10.5297





#### Isomer number 63 Number of FG in pore: 1 Isomer FG-FG distances (Å):

9.1602 13.1037 10.6034 12.7133 17.9793 12.7133





## Isomer number 170 Number of FG in pore: 0 Isomer FG-FG distances (Å):

12.7133 17.9793 12.7133 12.7133 17.9793 12.7133



### Isomer number 171 Number of FG in pore: 0 Isomer FG-FG distances (Å):

12.7133 17.9793 10.5297 12.7133 17.7505 14.5733





## Isomer number 175 Number of FG in pore: 0 Isomer FG-FG distances (Å):

12.7133 17.7505 10.5297 14.5733 17.7505 12.7133





## Isomer number 187 Number of FG in pore: 0 Isomer FG-FG distances (Å):

14.5733 17.9793 10.5297 10.5297 17.9793 14.5733



Unique Isomers with 4 FG inside the pore

#### Isomer FG-FG distances (Å): 5.8919 8.3324 5.8919 5.8919

- 8.3324
- 5.8919



# 5.8919 7.8236 3.5048

. 10

'14

Isomer FG-FG distances (Å): 5.8919

- 8.3324 7.5595

Isomer FG-FG distances (Å):

# 5.8919 7.8236 7.5595 3.5048 7.8236 5.8919



#### Isomer FG-FG distances (Å): 3.5048 8.3324 7.5595 7.5595

8.3324 3.5048



Unique Isomers with 3 FG inside the pore

## Isomer FG-FG distances (Å): 5.8919

8.3324 11.6104 5.8919 12.7870 7.8447



Isomer FG-FG distances (Å): 5.8919

8.3324 10.6034 5.8919 13.1037 9.1602



## Isomer FG-FG distances (Å):

5.8919 7.8236 11.6104 3.5048 12.7870 10.6034



## Isomer FG-FG distances (Å):

5.8919 7.8236 10.6034 3.5048 13.1037 11.6104







Isomer FG-FG distances (Å):

3.5048 8.3324 11.6104 7.5595 13.1037 7.8447



Isomer FG-FG distances (Å):

3.5048 8.3324 10.6034 7.5595 12.7870 9.1602


Unique Isomers with 2 FG inside the pore

#### Isomer FG-FG distances (Å):

5.8919 12.7870 11.6104 7.8447 12.7870 12.7133



#### Isomer FG-FG distances (Å):

5.8919 12.7870 10.6034 7.8447 13.1037 14.5733



# Isomer FG-FG distances (Å): 5.8919

13.1037 11.6104 9.1602 12.7870 10.5297



#### Isomer FG-FG distances (Å):

5.8919 13.1037 10.6034 9.1602 13.1037 12.7133



#### Isomer FG-FG distances (Å):

3.5048 12.7870 11.6104 10.6034 13.1037 12.7133



#### Isomer FG-FG distances (Å):

3.5048 12.7870 10.6034 10.6034 12.7870 14.5733



#### Isomer FG-FG distances (Å):

3.5048 13.1037 11.6104 11.6104 13.1037 10.5297



Isomer FG-FG distances (Å):

7.8447 8.3324 11.6104 11.6104 17.9793 7.8447



Isomer FG-FG distances (Å):

7.8447 8.3324 10.6034 11.6104 17.7505 9.1602



Isomer FG-FG distances (Å):

7.8447 7.8236 11.6104 9.1602 17.9793 10.6034



Isomer FG-FG distances (Å):

7.8447 7.8236 10.6034 9.1602 17.7505 11.6104







Isomer FG-FG distances (Å):

9.1602 8.3324 10.6034 10.6034 17.9793 9.1602





Unique Isomers with 1 FG inside the pore

#### Isomer FG-FG distances (Å):

7.8447 12.7870 11.6104 12.7133 17.9793 12.7133



#### Isomer FG-FG distances (Å):

7.8447 12.7870 10.6034 12.7133 17.7505 14.5733



#### Isomer FG-FG distances (Å):

7.8447 13.1037 11.6104 14.5733 17.9793 10.5297



#### Isomer FG-FG distances (Å):

7.8447 13.1037 10.6034 14.5733 17.7505 12.7133



#### Isomer FG-FG distances (Å):

9.1602 12.7870 11.6104 10.5297 17.7505 12.7133



#### Isomer FG-FG distances (Å):

9.1602 12.7870 10.6034 10.5297 17.9793 14.5733



#### Isomer FG-FG distances (Å):

9.1602 13.1037 11.6104 12.7133 17.7505 10.5297



#### Isomer FG-FG distances (Å):

9.1602 13.1037 10.6034 12.7133 17.9793 12.7133



Unique Isomers with 0 FG inside the pore

#### Isomer FG-FG distances (Å):

12.7133 17.9793 12.7133 12.7133 17.9793

12.7133



#### Isomer FG-FG distances (Å):

12.7133 17.9793 10.5297 12.7133 17.7505



#### Isomer FG-FG distances (Å):

12.7133 17.7505 10.5297 14.5733 17.7505



#### Isomer FG-FG distances (Å):

14.5733 17.9793 10.5297 10.5297 17.9793 14.5733



S5 Histograms of the functional group - functional group distances of every pore topology.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S6: Histogram of FG-FG distances in pore  $Tri^2Di^3$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.





Figure S7: Histogram of FG-FG distances in pore  $Tri^4Di^6$ , constructed from a node with radius 5Å and two different linker size, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S8: Histogram of FG-FG distances in pore  $Tri_2^4Di^6$ , constructed from a node with radius 5Å and two different linker size, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S9: Histogram of FG-FG distances in pore  $Tri^6Di^9$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.





Figure S10: Histogram of FG-FG distances in pore  $Tri^8Di^{12}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.





Figure S11: Histogram of FG-FG distances in pore  $Tri^{20}Di^{30}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.


(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S12: Histogram of FG-FG distances in pore  $Tet^2Di^4$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S13: Histogram of FG-FG distances in pore  $Tet_3^3Di^3$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S14: Histogram of FG-FG distances in pore  $Tet_4^4Di^8$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S15: Histogram of FG-FG distances in pore  $Tet^5Di^{1}0$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S16: Histogram of FG-FG distances in pore  $Tet^6Di^{12}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.





Figure S17: Histogram of FG-FG distances in pore  $Tet^8Di^{16}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S18: Histogram of FG-FG distances in pore  $Tet^{16}Di^{32}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.





Figure S19: Histogram of FG-FG distances in pore  $Tet^{24}Di^{48}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4-benzenedicarboxylic acid) inner functional group positions.

S6 Functional Group - Functional Group distance Histogram of Tet<sup>2</sup>Di<sup>4</sup> Isomers.





(b)







Figure S6.1. (a) A Tet<sup>2</sup>Di<sup>4</sup> cage is constructed from a node with radius 5Å and linker benzene-1,4dicarboxylic acid (bdc). The functionalisation based on one per-bdc linker would generate 31 unique isomers. (b) The functional Group - functional group distance histogram of Tet<sup>2</sup>Di<sup>4</sup> Isomers.

# S7 The effect of linker rotation on the functional group- functional group distances

Table S2: Distances found in  $Tet^2Di^4$  pore. The linker rotation deviates the distances from its symmetric position, the maximum and minimum distance is tabulated. Colour code blue represents In-In distances, violet In-out distances, red out-out distances.

$\begin{array}{c} {\rm FG}\text{-}{\rm FG}  {\rm dis}\text{-}\\ {\rm tance}({\rm \AA}) \end{array}$	Shortest Distance(Å)	$\begin{array}{l} \text{Maximum} \\ \text{distance}(\text{\AA}) \end{array}$
3.68	2.44	5.82
4.70	3.81	6.52
5.20	5.20	6.31
5.96	5.96	6.96
7.91	6.18	9.32
8.43	6.82	9.76
10.07	9.80	10.48
10.48	10.22	10.88
10.57	8.34	11.85
10.95	8.83	12.19
14.94	14.28	14.94
15.22	14.57	15.22

Table S3: Distances found in  $Tri^4Di^6$  pore. The linker rotation deviates the distances from its symmetric position, the maximum and minimum distance is tabulated. Distances are colour-coded: blue represents In-In distances, violet In-out distances, red out-out distances.

$\begin{array}{c} \text{FG-FG}  \text{dis-} \\ \text{tance}(\text{\AA}) \end{array}$	Shortest Distance $(Å)$	$\begin{array}{l} {\rm Maximum} \\ {\rm distance}({\rm \AA}) \end{array}$
3.14	2.39	4.79
5.26	4.63	6.75
6.08	5.33	7.73
6.84	6.75	8.09
7.41	6.13	8.39
8.51	7.27	9.63
9.63	8.40	10.49
10.02	8.33	10.78
10.09	8.80	11.13
11.54	11.11	12.09
11.73	10.08	12.47
12.90	11.21	13.66
16.33	15.55	16.35

FG-FG dis-	Shortest Dis-	Maximum
tance(Å)	tance(Å)	distance(Å)
3.43	2.01	4.15
4.85	4.85	5.86
5.46	4.24	5.59
6.3	4.88	8.38
7.17	7.17	8.01
7.29	5.57	8.5
7.95	7.22	9.92
8.37	6.29	9.79
8.44	6.71	9.81
8.54	8.24	9.003
8.91	8.91	9.7
9.05	8.82	9.26
9.24	7.57	10.6
9.54	9.54	10.87
9.73	9.73	10.68
9.76	7.94	11.15
10.12	8.12	11.5
10.14	10.14	11.06
10.48	10.28	10.71
11.24	9.16	12.66
11.75	11.75	13.11
11.84	11.37	11.84
12.35	11.11	13.6
12.75	11.8	13.81
12.89	12.66	13.11
13.29	12.31	14.28
13.95	13.25	14.72
14.02	13.55	14.02
14.67	14.39	14.87
14.95	14.68	15.15
15.9	15.43	15.9
16.33	14.6	16.69
17.05	15.5	17.4
17.97	18.27	16.66
19.61	18.95	19.61
19.82	19.33	19.82

Table S4: Distances found in  $Tet^6Di^{12}$  pore. The linker rotation deviates the distances from its symmetric position, the maximum and minimum distance is tabulated. Colour code blue represents In-In distances, violet In-out distances, red out-out distances.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S20: KDE applied on FG-FG distance histograms of pore  $Tri^2Di^3$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.





Figure S21: KDE applied on FG-FG distance histograms of pore  $Tri^4Di^6$ , constructed from a node with radius 5Å and two different linker size, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S22: KDE applied on FG-FG distance histograms of pore  $Tri_2^4Di^6$ , constructed from a node with radius 5Å and two different linker size, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S23: KDE applied on FG-FG distance histograms of pore  $Tri^6Di^9$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S24: KDE applied on FG-FG distance histograms of pore  $Tri^8Di^{12}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S25: KDE applied on FG-FG distance histograms of pore  $Tri^{20}Di^{30}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.





Figure S26: KDE applied on FG-FG distance histograms of pore  $Tet^2Di^4$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S27: KDE applied on FG-FG distance histograms of pore  $Tet_2^3Di^3$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S28: KDE applied on FG-FG distance histograms of pore  $Tet_4^4Di^8$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S29: KDE applied on FG-FG distance histograms of pore  $Tet^5Di^{10}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S30: KDE applied on FG-FG distance histograms of pore  $Tet^6Di^{12}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner  $\operatorname{poly}(1,4\text{-benzenedicarboxylic acid})$  linker

Figure S31: KDE applied on FG-FG distance histograms of pore  $Tet^8Di^{16}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner poly(1,4-benzenedicarboxylic acid) linker

Figure S32: KDE applied on FG-FG distance histograms of pore  $Tet^{16}Di^{32}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.



(c) Inner  $\operatorname{poly}(1,4\text{-benzenedicarboxylic acid})$  linker

Figure S33: KDE applied on FG-FG distance histograms of pore  $Tet^{24}Di^{48}$ , constructed from a node with radius 5Å and two different linker sizes, (a) benzene-1,4-dicarboxylic acid, (b) poly(1,4-benzenedicarboxylic acid) outer functional group positions (c) poly(1,4benzenedicarboxylic acid) inner functional group positions.

## S9 Principal Component Analysis.

10 isomers of each topology were used for the input data of the Principal Component Analysis (PCA). The histogram of each isomer was generated and kernel density estimation was then applied on the histograms using gaussian kernel and bandwidth 0.1. The histograms were converted to a matrix with upper limit of 60Å and 1Å wide bins.

# S10. Representative Isomers of NH2-UiO-66 Oh Cage for Machine Learning Input.

We carefully selected a number of representative structures based on four features of geometrical properties we are particularly interested in:

- The number of functional groups pointing into the centre of the pore (Section S10.1).
- The number of functional groups on the windows of the pore (Section 0).
- The number of functional groups adjacent to one node (Section 0).
- The functional group functional group distance set (Section 0).

The final choice of representative isomers are computed by characterising each feature by a string of numbers, every unique combination of the four features is one representative isomer.

S10.1. The number of FGs pointing into the centre of the pore.



*Figure S10.1. (right) Octahedral pore of UiO-66. (left) benzene linkers of octahedral pore, blue rectangle showing functional group slots that is pointing into the centre of the pore.* 

In the UiO-66 octahedral pore, the bdc planar ring is oriented such that the functional group is either facing into or outwards the pore, as shown in Figure S10.1. The functional group that is facing into the pore is more important to the guest molecule, while the outer FG only gives a small electronic effect (through electron withdrawal / donation with respect to the pristine linker) to the guest molecule inside the pore. Thus we care about how many functional groups are inside the pore. This feature is characterised by a string of one single number, the number of FG pointing into the pore, as shown in Figure S10.2., an example is presented in Figure S10.3.



*Figure S10.2. Descriptor to represent the number of functional groups pointing inside the pore.* 

Example:



*Figure S10.3. (a) An example of functionalised UiO-66 octahedral pore. (b). There are 7 functional groups pointing into the pore, thus the descriptor is ''7''.* 

S10.2. The number of functional groups on the windows of the pore.



*Figure S10.4. A window of UiO-66 octahedral pore. The window is bordered by three linkers, the blue rectangle shows FG slots that is pointing into the pore centre.* 

The number of FG located on a window is important for the transport property of the guest molecules. The octahedral pore has 8 windows, every window is bordered by 3 linkers (Figure 4.). As one window has 3 linkers thus the number of functional groups on one window is in between 0-3. Figure 5. shows a string of 4 numbers, with each of the numbers representing the total number of FGs. An example of how the descriptor is calculated is in Figure S10.6.

"	Number of windows with 0 FG	Number of windows with 1 FG	Number of windows with 2 FGs	Number of windows with 3 FGs	"

Figure S10.5. Descriptor to represent the arrangement of functional groups on the window.

Example:



Figure S10.6. (a) An example of functionalised UiO-66 octahedral pore (b) the table of windows and the corresponding number of FG in the window (c) the descriptor used in choosing representative structures.

#### S10.3. The number of FG adjacent to one node



## Figure S10.7. One corner of UiO octahedral pore. Blue squares showing the FG slot where functional group is adjacent to the node.

In the UiO-66 octahedral pore, there are 6 Zr<sub>6</sub>O nodes, and every node is connected to 4 linkers, as shown in Figure 7. In many absorption situations, binding pockets are preferable for a given guest. We considered the functional groups adjacent to a node to be important because the functional groups in a pore corner are the most likely to be situated closely enough to form a binding pocket. The total number of FG adjacent to one node is in between 0-4, and similarly to the previous implementation, this feature is characterised by a string of 5 numbers, with each of the number representing the number node with the following conditions, as shown in Figure S10.8, an example is presented in Figure S10.9.

"	Number of nodes with 0 adjacent FG	Number of nodes with 1 adjacent FG	Number of nodes with 2 adjacent FGs	Number of nodes with 3 adjacent FGs	Number of nodes with 4 adjacent FGs	"
		na norme e		inaj net me e o o	najiri na s	

Figure S10.8. Descriptor to represent the arrangement of functional group adjacent to each node.



*Figure S10.9. (a) An example of functionalised UiO-66 octahedral pore (b) the list of nodes and the corresponding number of FG adjacent to the node (c) the descriptor used in choosing representative structures* 

### S10.4. The functional group - functional group distance set

Every structure has its own unique FG-FG distance set. In total, there are three types of FG-FG distances, shown in Figure 11, the first type is the distance which occurs between two FG located inside the pore (inin), the second type is two FG located outside the pore (out-out), and the third type is where one inside with one outside the pore (in-out).

This feature is characterised by a string of its FG-FG distance set, shown in Figure 10.

"	FG-FG distance 1	FG-FG distance 2	FG-FG distance 3	and so on	"
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Figure S10.10. Descriptor to	o represent the arrangen	ent of functional	l group l	based on fi	unctional	group-
	functional gro	oup distances.				

So to eliminate similar cage isomers, we rounded the FG-FG distance and only consider the FG-FG distance that occur inside the pore as they are a more prominent influence to the guest molecule. The FG-FG distance we consider is shown by the blue line in the figure below, while the violet and red dotted line are ignored.



*Figure S10.11. An example of functionalised UiO-66 octahedral pore. Blue line shows the distance between two FG that occur inside the pore, violet dotted line shows the distance between two FG that occur across the boundary of the pore, and red dotted line shows the distance that occur outside the pore.* 

### S11 Machine Learning Method

### S11.1. Dataset preparation

Before calculating the binding energy, the cages (with propranolol molecule inside) are relaxed to their stable position using GFN-xTB. During the relaxation, the  $Zr_6O$  node geometries are constrained, while the carboxylate oxygen atoms are allowed to move but the bond length to their neighbouring Zr atoms are restrained with a harmonic restraint in order to prevent bond breaking, while the rest of the atoms are free to move.

The molecule and cage single point energies are calculated from the relaxed structure. Finally, the binding energies are calculated by subtracting the individual cage and propranolol energies from the total energy (equation S11.1). The geometry optimisation and binding energy calculation are performed using the AMS 2020 software package.

$$E_{BE} = E_{cage \& propanolol} - (E_{cage} + E_{propanolol}) \quad (S11.1)$$

Each isomer has 10 binding energies corresponding to 10 random propranolol positions, standard error is calculated for each isomer. The isomers with standard error of the binding energies more than 8 kJ/mol are discarded from the machine learning dataset, the total number of isomers become 3127.

### S11.2. Machine learning model

The dataset is then trained on a supervised neural network, consisting of one layer with 6 neurons. The machine learning calculation is performed using scikit-learn python package.

The dataset is divided to 80:20 ratio for training data and testing data. The input of the machine learning is the cage isomer functional group pair distance histogram after KDE has been applied (gaussian kernel, bandwidth = 0.1Å, bin size of 0.2Å) and the response value f(x) of each cage isomer is the average binding energies.

After training using 80% of the data, the trained model is tested using the remaining 20% data. Finally, the binding energy of test data is compared, binding energy from machine learning versus binding energy from GFN-xTB calculation.

The dataset and code is available at: https://github.com/maryamnhd/FG-Pair-Distance-Descriptor https://github.com/maryamnhd/Cage-Isomers