

## Electronic Supplementary Information

# Ultra-low loading Pt atomic cluster electrode with Pt-O bond as an active site with the high hydrogen evolution reaction performance

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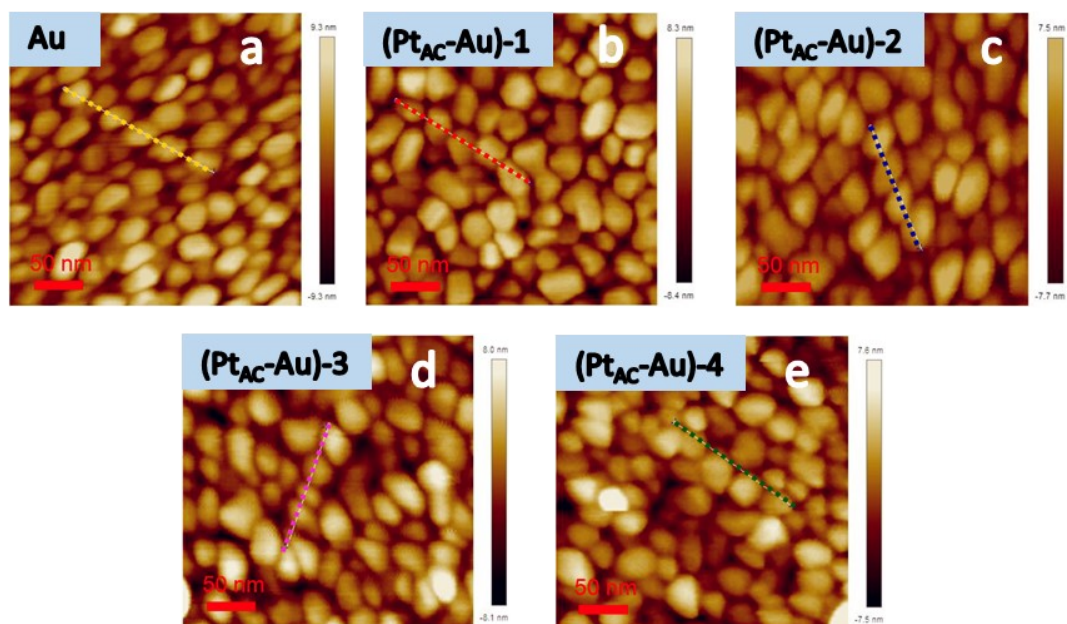
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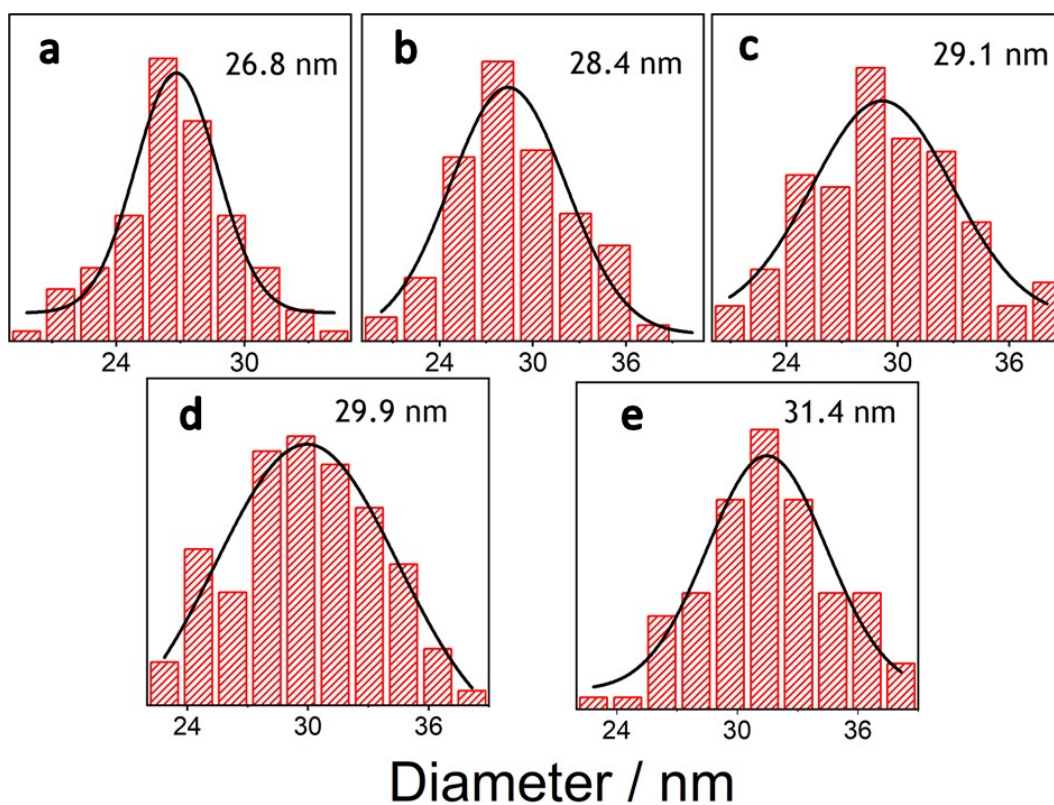
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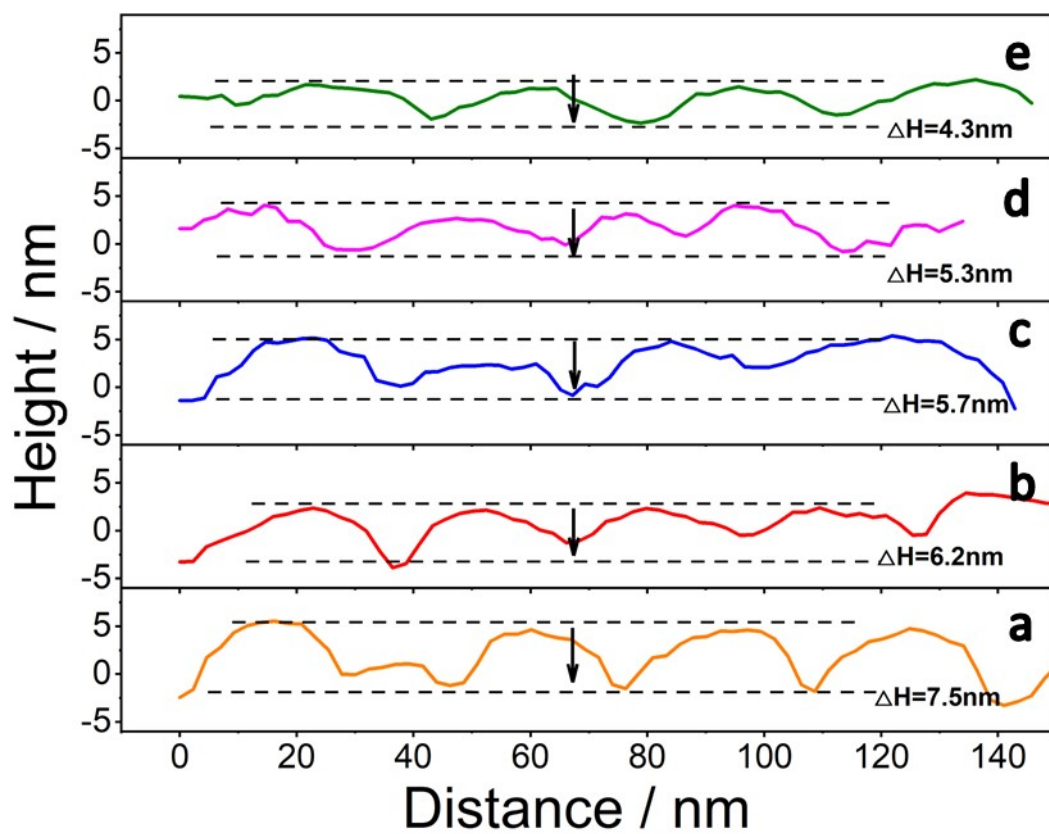
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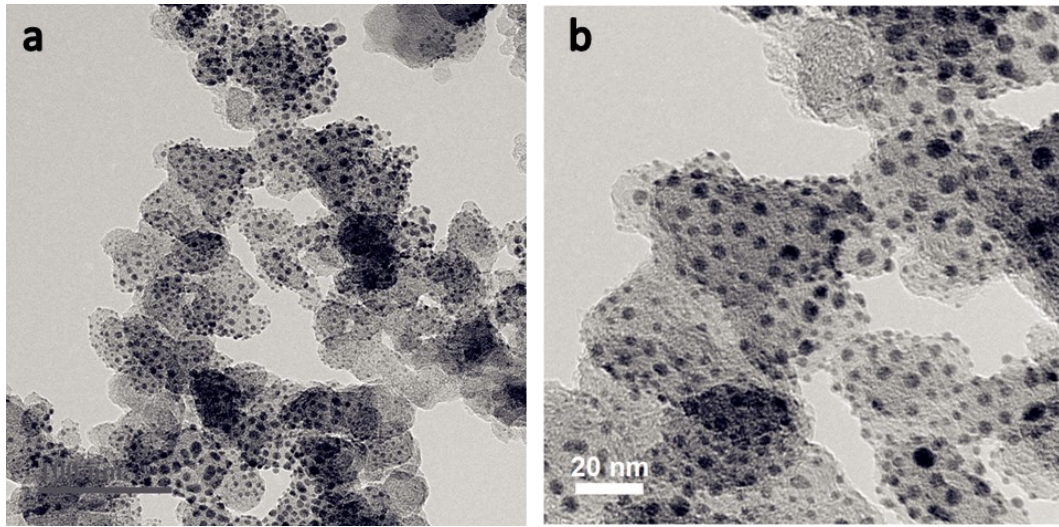
**Figure S1** AFM images of Au and (Pt<sub>AC</sub>-Au)-x (x=1, 2, 3, 4) with different Pt loading (Au (a), (Pt<sub>AC</sub>-Au)-1 (b), (Pt<sub>AC</sub>-Au)-2 (c), (Pt<sub>AC</sub>-Au)-3 (d) and (Pt<sub>AC</sub>-Au)-4 (e)).



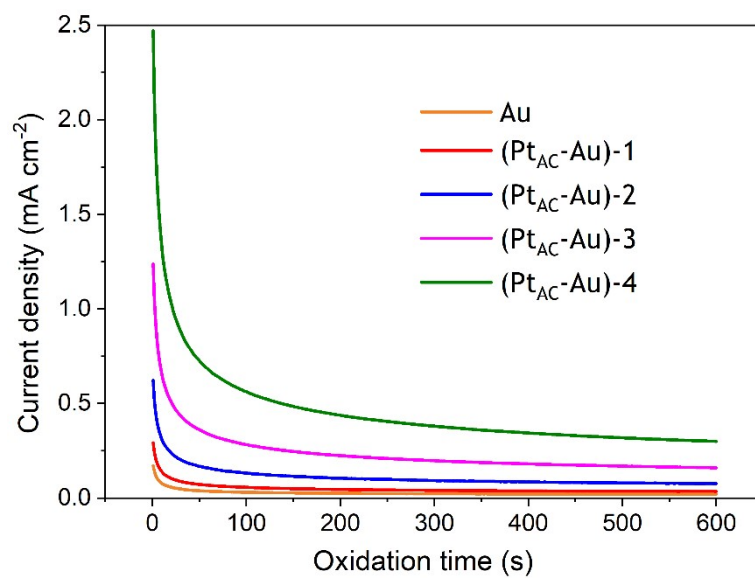
**Figure S2** The particle sizes of Au and (Pt<sub>AC</sub>-Au)-x (x=1, 2, 3, 4) with different Pt loading according to AFM image (Au (a), (Pt<sub>AC</sub>-Au)-1 (b), (Pt<sub>AC</sub>-Au)-2 (c), (Pt<sub>AC</sub>-Au)-3 (d) and (Pt<sub>AC</sub>-Au)-4 (e)).



**Figure S3** The height differences of Au and  $(\text{Pt}_{\text{AC}}\text{-Au})\text{-}x$  ( $x=1, 2, 3, 4$ ) with different Pt loading according to AFM image (Au (a),  $(\text{Pt}_{\text{AC}}\text{-Au})\text{-}1$  (b),  $(\text{Pt}_{\text{AC}}\text{-Au})\text{-}2$  (c),  $(\text{Pt}_{\text{AC}}\text{-Au})\text{-}3$  (d) and  $(\text{Pt}_{\text{AC}}\text{-Au})\text{-}4$  (e)).

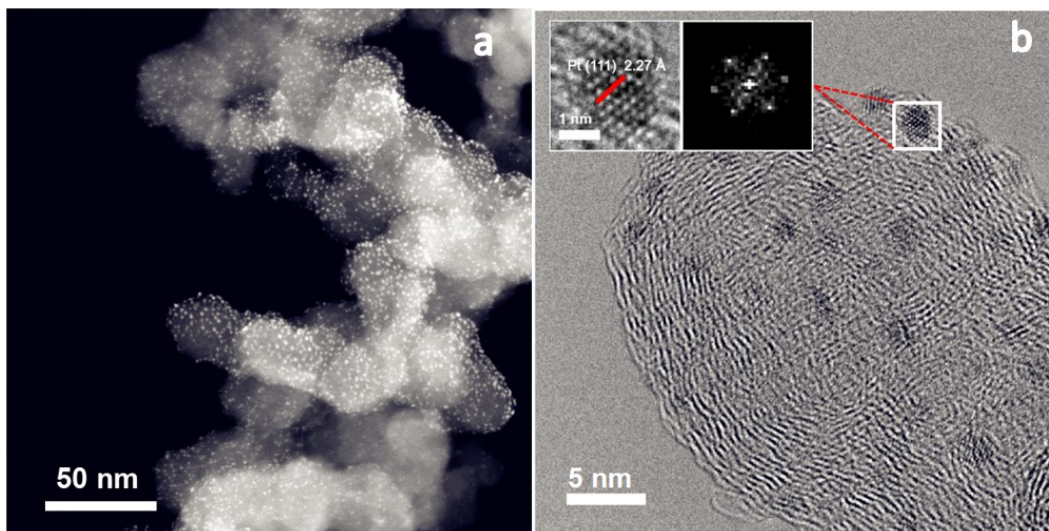


**Figure S4** The TEM images of the (Pt<sub>AC</sub>-Au)-1 electrode.

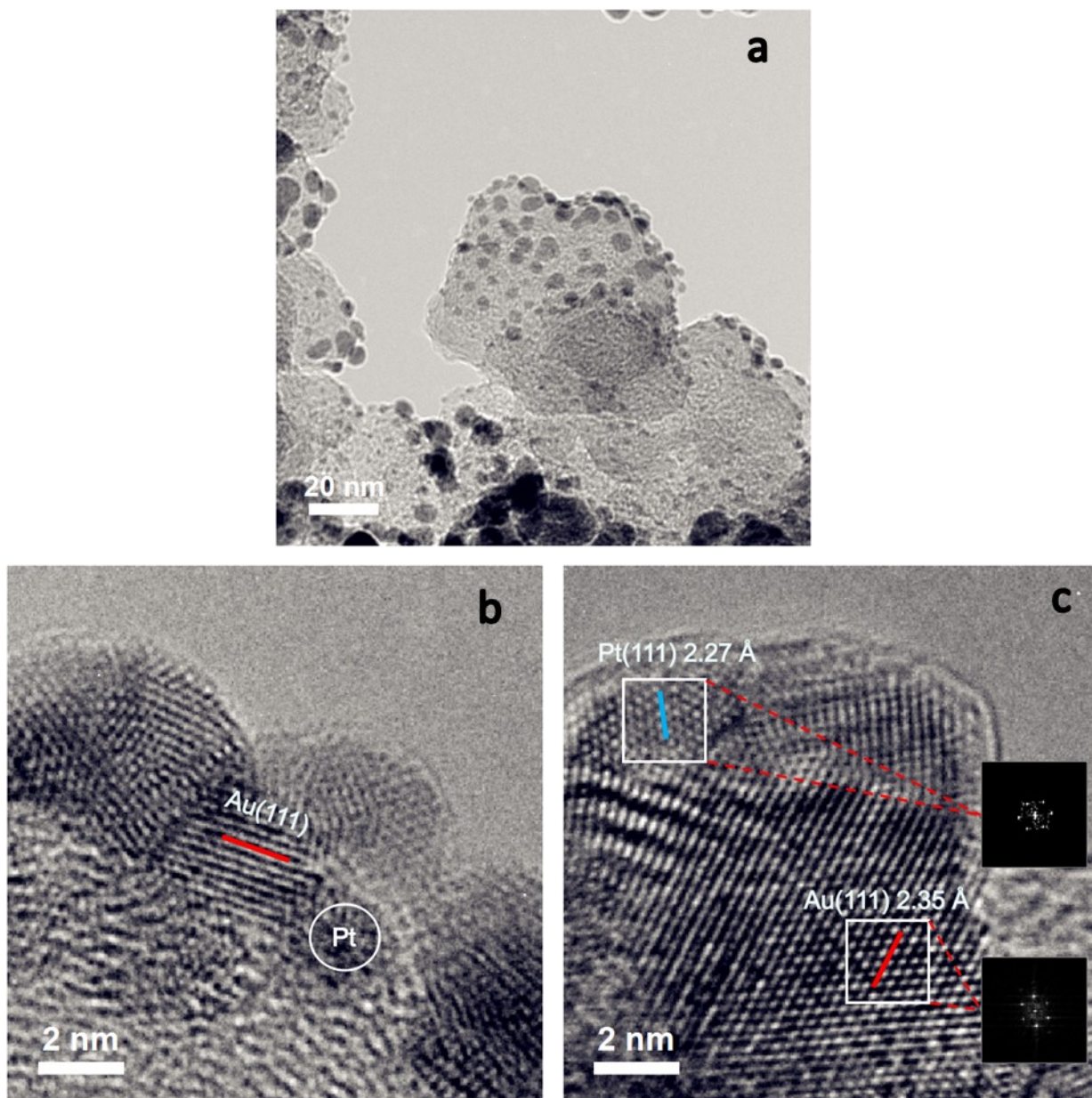


**Figure S5** The anode oxidation curves of Au and (Pt<sub>AC</sub>-Au)-x (x=1, 2, 3, 4) electrodes with different Pt loading at 1.7 V.



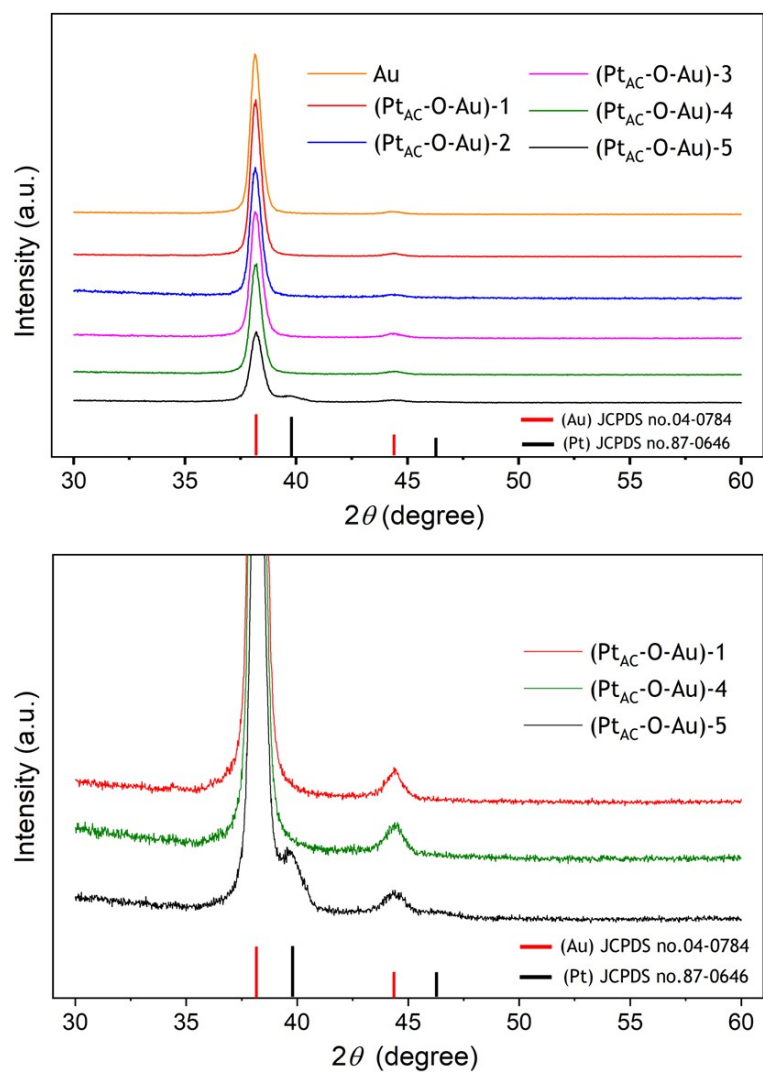


**Figure S6** The HAADF-STEM (a) and HRTEM (b) of the Pt-1 electrode.

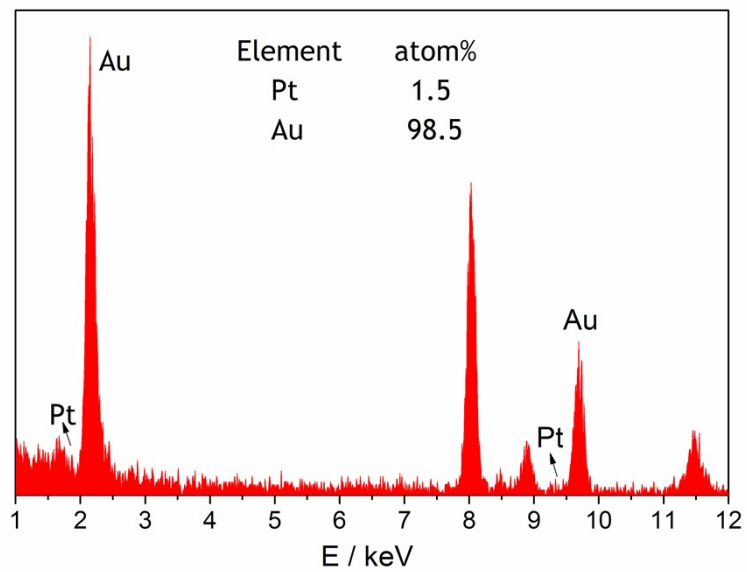


**Figure S7** The TEM (a) and HRTEM (b, c) images of the (Pt<sub>AC</sub>-Au)-4 electrode.

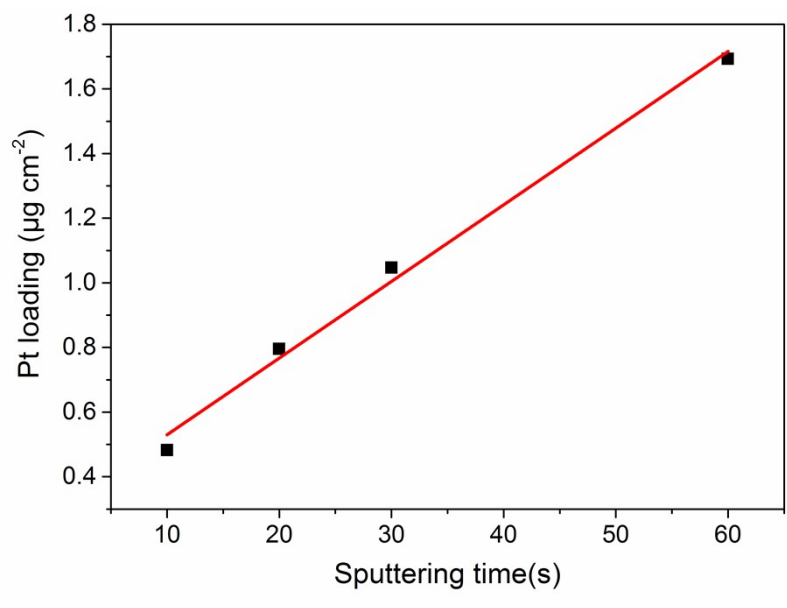




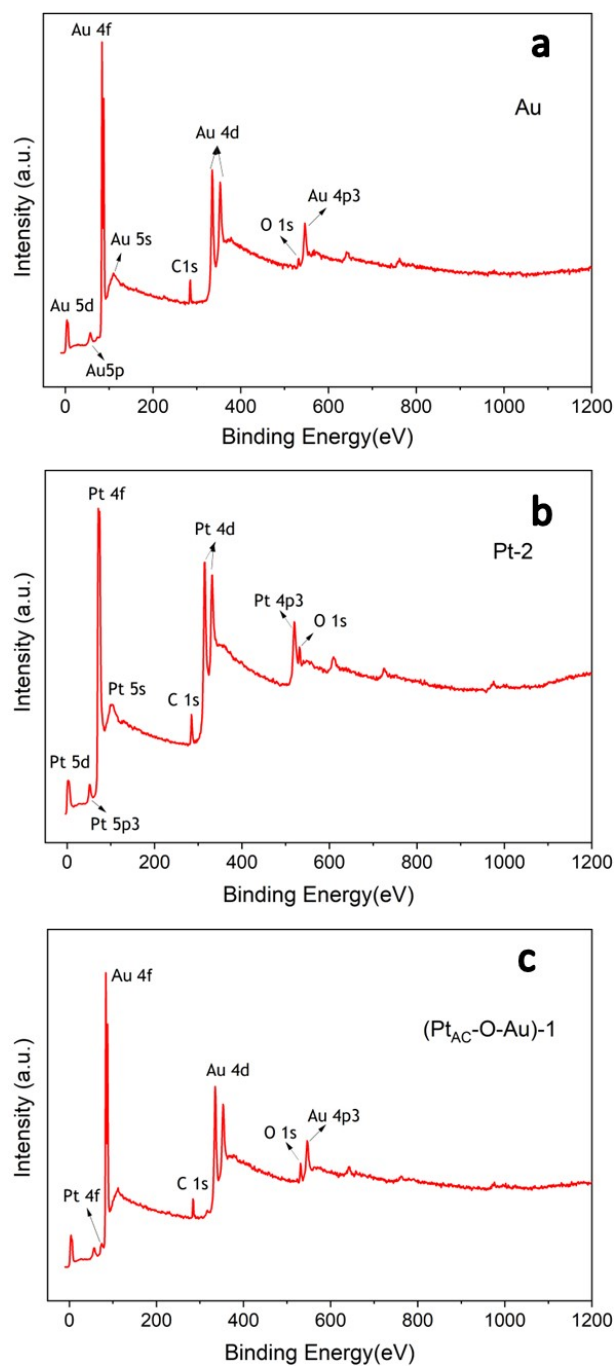
**Figure S8** XRD diagrams of Au and (Pt<sub>AC</sub>-O-Au)-x (x=1, 2, 3, 4, 5) with different Pt loading.



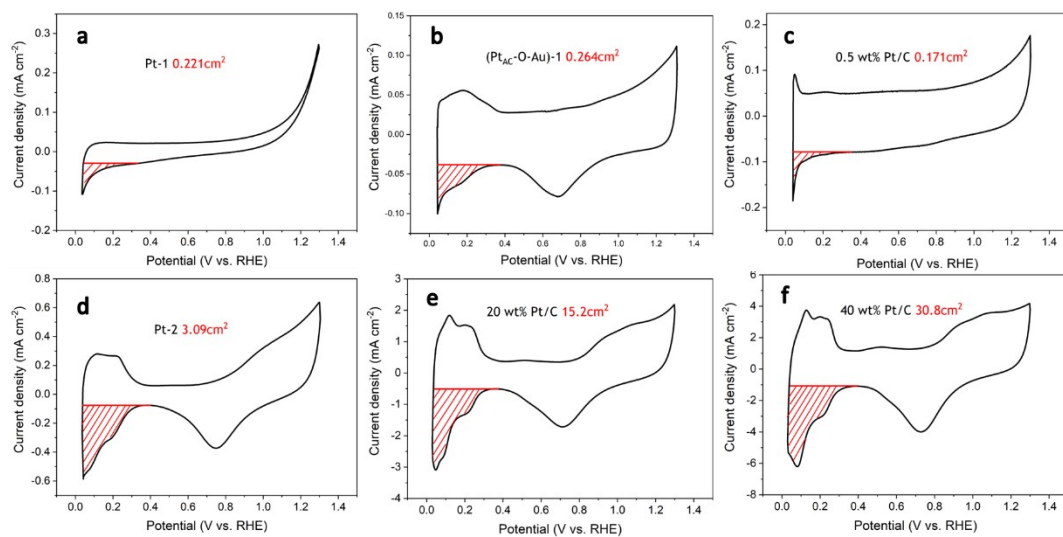
**Figure S9** The EDS element analysis of (Pt<sub>AC</sub>-Au)-1 electrode.



**Figure S10** Relationship between the sputtering time and Pt loading

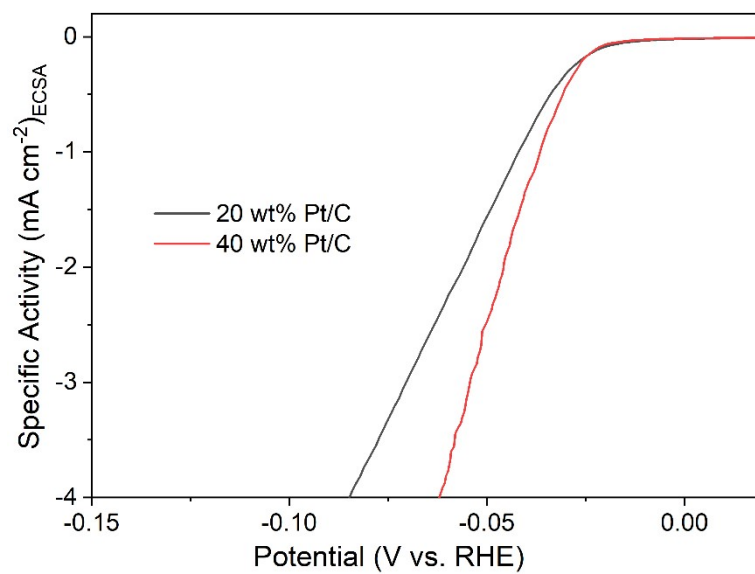


**Figure S11** The survey XPS analysis of the Au (a), Pt-2 (b) and (Pt<sub>AC</sub>-O-Au)-1 (c) electrodes.

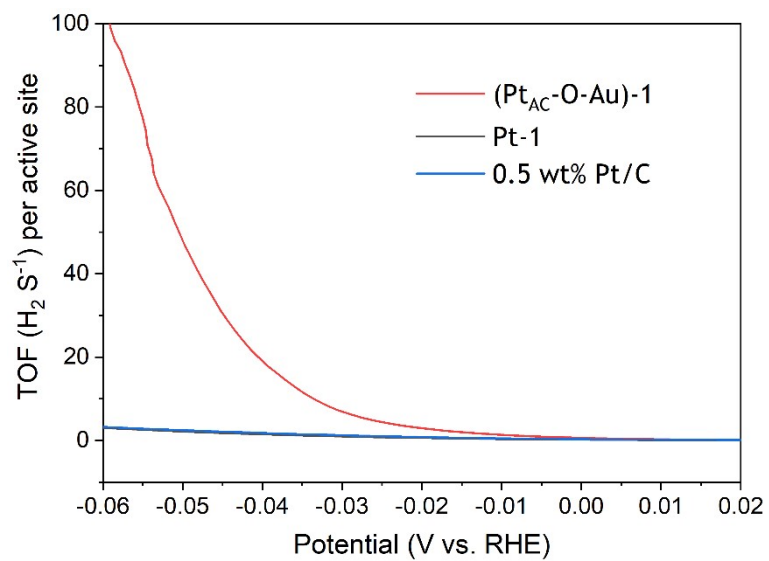


**Figure S12** Cyclic voltammograms of Pt-1 (a), (Pt<sub>AC</sub>-O-Au)-1 (b), 0.5 wt% Pt/C (c), Pt-2 (d), 40 wt% Pt/C (e) and 20 wt% Pt/C (f).

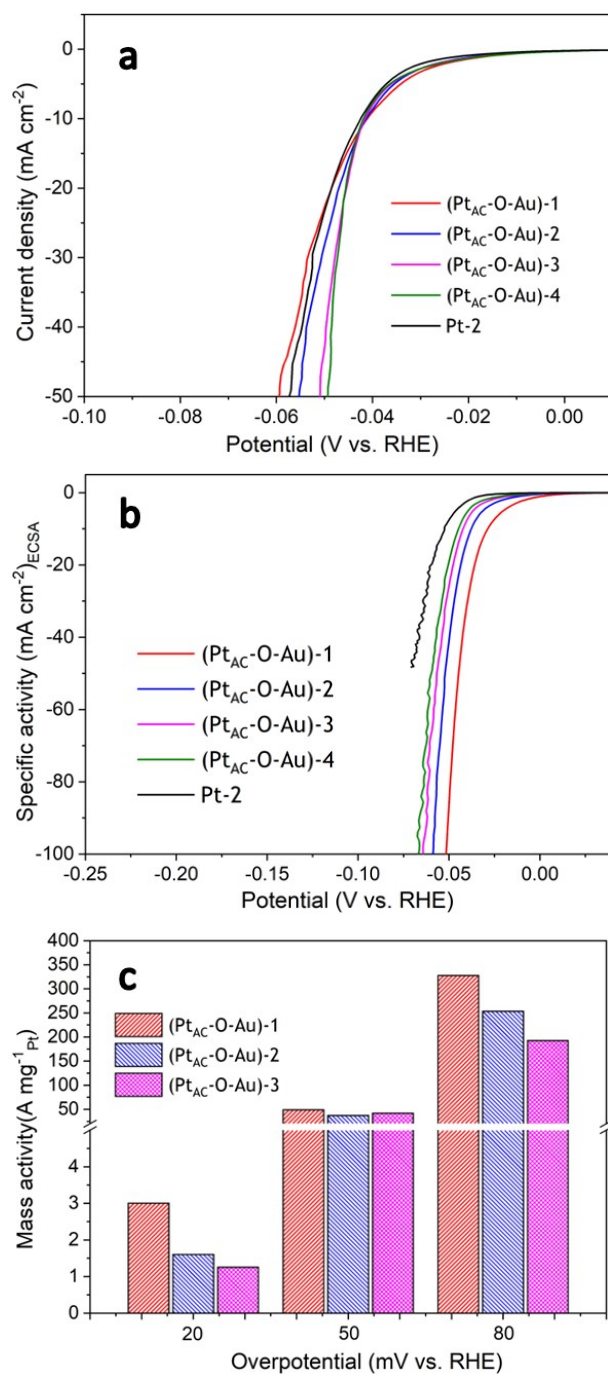




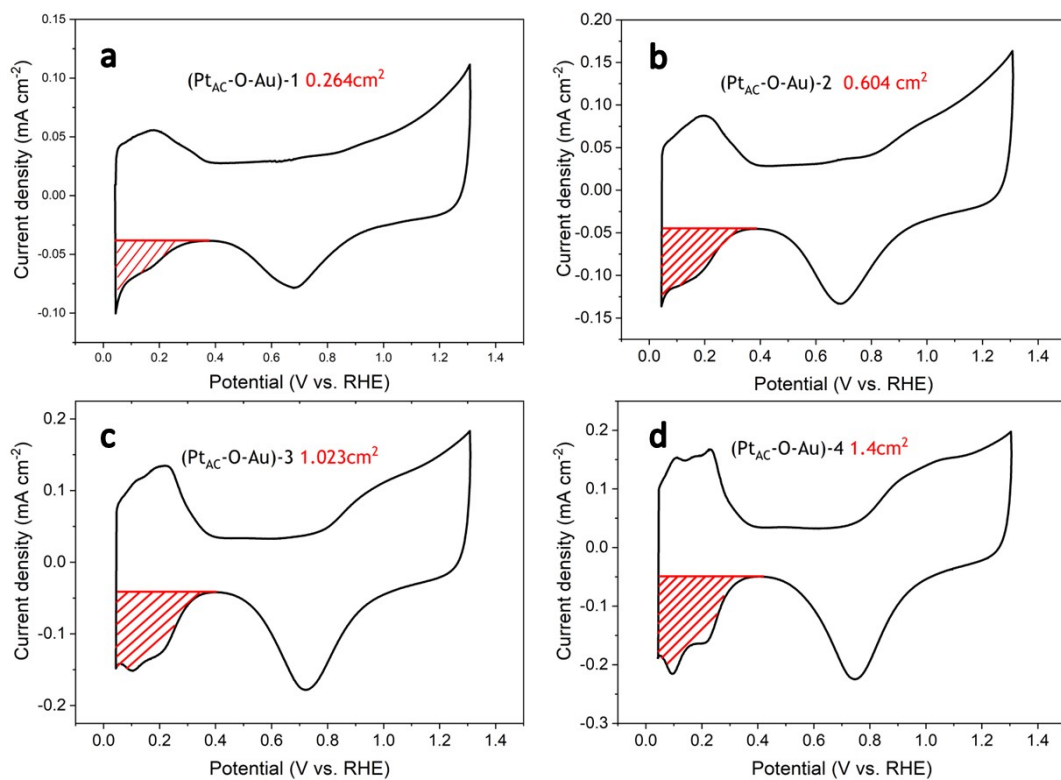
**Figure S13** The specific activities (SAs) of 20 wt% Pt/C and 40 wt% Pt/C.



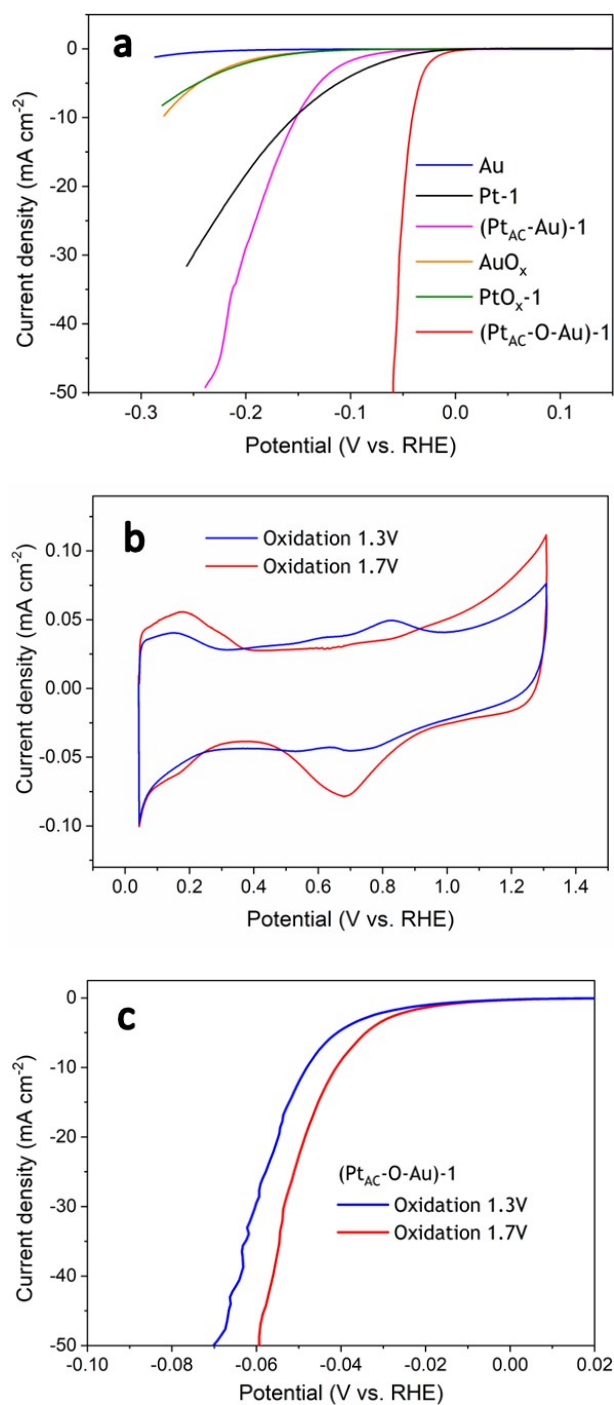
**Figure S14** The turnover frequency (TOF) of  $(\text{Pt}_{\text{AC}}\text{-O-Au})\text{-1}$ , Pt-1 and 0.5 wt% Pt/C.



**Figure S15** HER apparent activity (a), specific activity (b) and mass specific activity (c) of (Pt<sub>AC</sub>-O-Au)-x (x=1, 2, 3, 4) with different Pt loading.

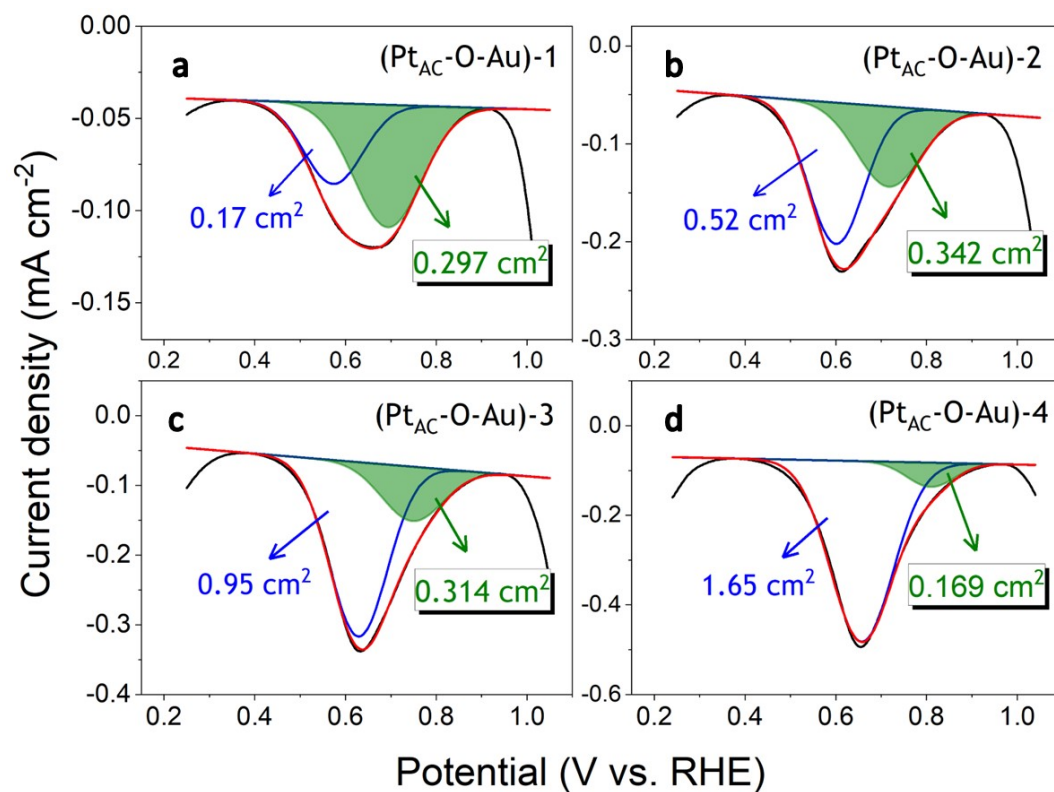


**Figure S16** Cyclic voltammograms of (Pt<sub>AC</sub>-O-Au)-1 (a), (Pt<sub>AC</sub>-O-Au)-2 (b), (Pt<sub>AC</sub>-O-Au)-3 (c) and (Pt<sub>AC</sub>-O-Au)-4 (d).

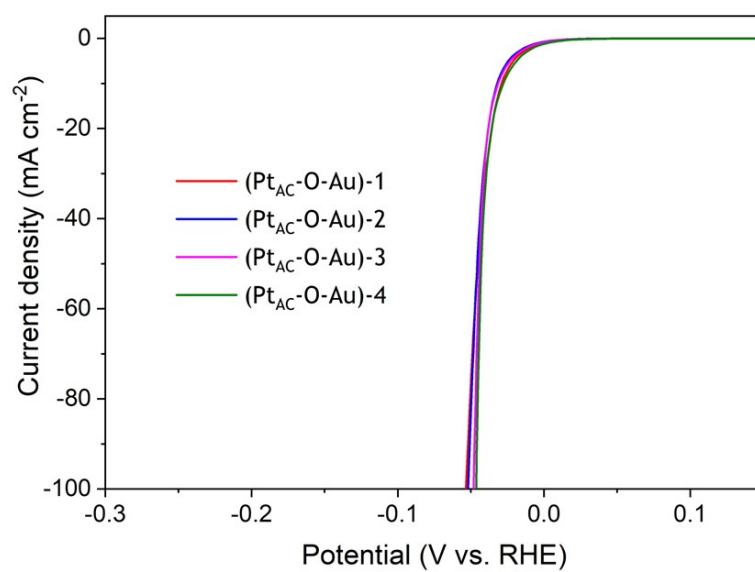


**Figure S17** The apparent activity of (Pt<sub>AC</sub>-O-Au)-1, (Pt<sub>AC</sub>-Au)-1, PtO<sub>x</sub>-1, Pt-1, AuO<sub>x</sub> and Au (a). Cyclic voltammograms (b) and linear sweep voltammetry curves (c) of (Pt<sub>AC</sub>-O-Au)-1 at anodic oxidation potentials of 1.3 and 1.7 V.

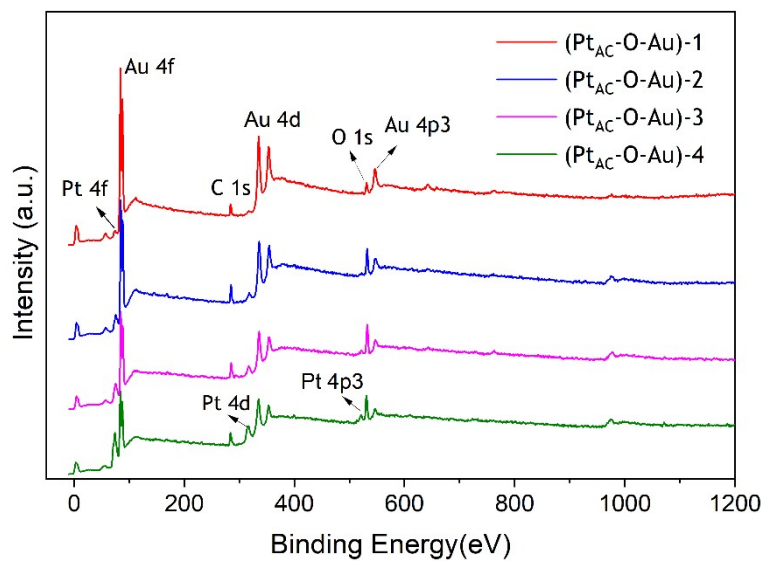




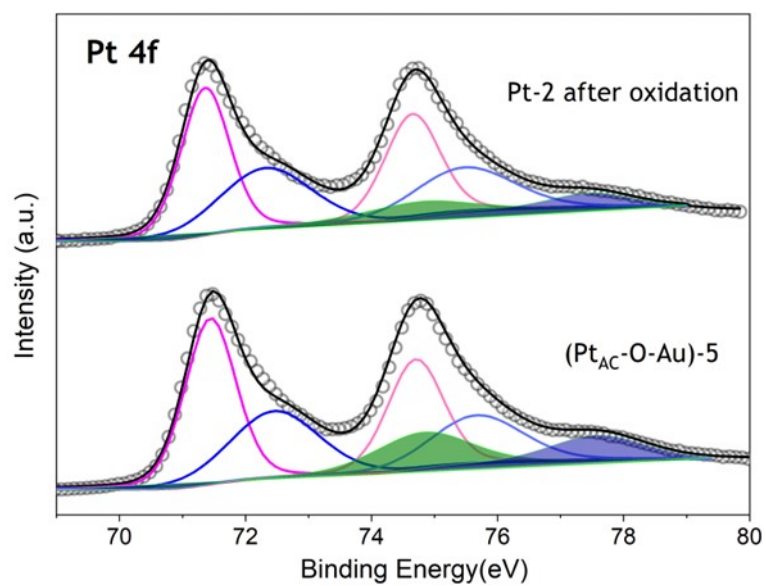
**Figure S18** Peak fitting of Au-O-Pt and PtO<sub>x</sub> reduction peaks of (Pt<sub>AC</sub>-O-Au)-1 (a), (Pt<sub>AC</sub>-O-Au)-2 (b), (Pt<sub>AC</sub>-O-Au)-3 (c) and (Pt<sub>AC</sub>-O-Au)-4 (d).



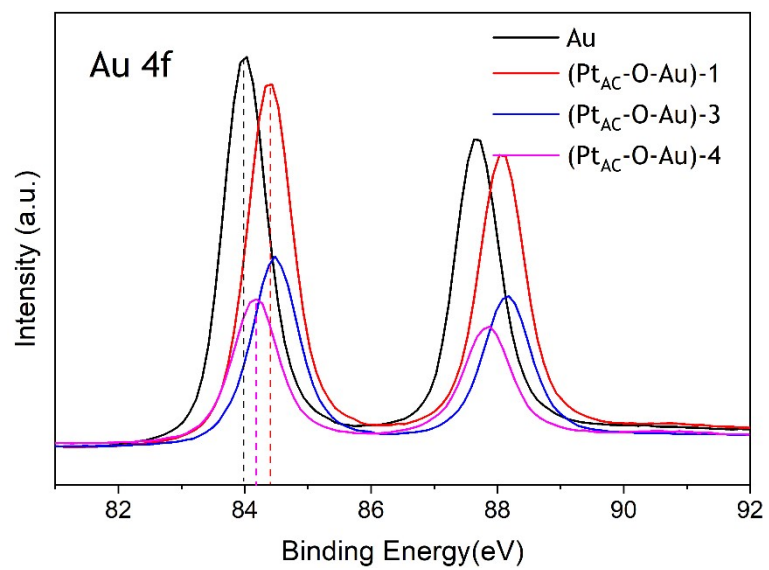
**Figure S19** The specific activity of (Pt<sub>AC</sub>-O-Au)-x (x=1, 2, 3, 4) with different Pt loading (The calculation of ECSA is based on the reduction peak area of Au-O-Pt at 0.72V).



**Figure S20** The survey XPS analysis of the (Pt<sub>AC</sub>-Au)-x (x=1, 2, 3, 4) electrodes with different Pt loading.

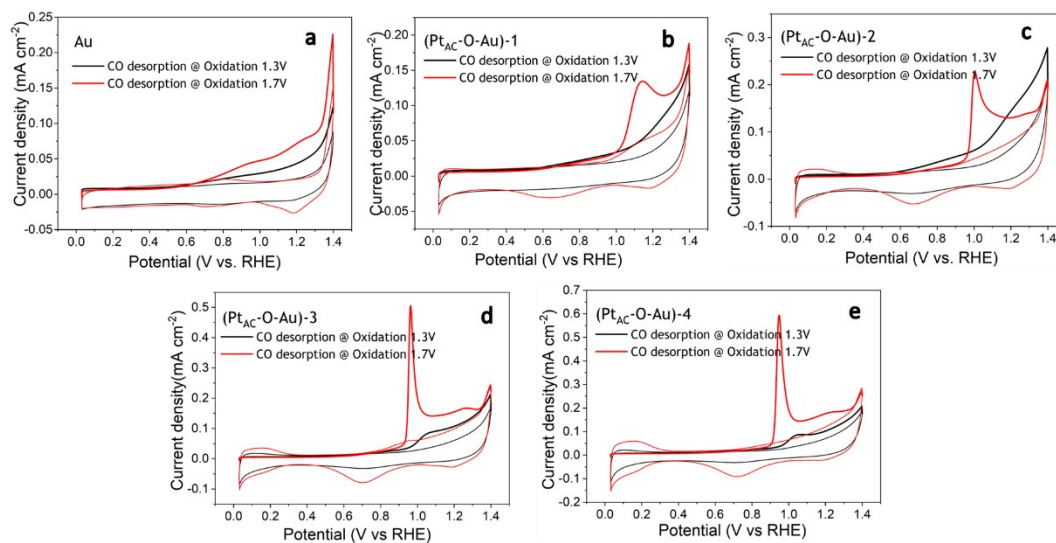


**Figure S21** The XPS core-level spectra of Pt 4f obtained from the (Pt<sub>AC</sub>-O-Au)-5 and Pt-2 (after oxidation) electrodes.



**Figure S22** The XPS core-level spectra of Au 4f obtained from Au and (Pt<sub>AC</sub>-Au)-x (x=1, 3, 4) electrodes with different Pt loading.





**Figure S23** The CO adsorption-stripping curves of (Pt<sub>AC</sub>-Au)-x (x=1, 2, 3, 4) electrodes with anodic oxidation potential at 1.3 V and 1.7 V (Au (a), (Pt<sub>AC</sub>-Au)-1 (b), (Pt<sub>AC</sub>-Au)-2 (c), (Pt<sub>AC</sub>-Au)-3 (d) and (Pt<sub>AC</sub>-Au)-4 (e)).

## Supplementary Tables

**Table S1** Pt loading with different sputtering times using ICP-OES

Sputteri ng time(s)	Electrode	Pt( $\mu\text{g cm}^{-2}$ )	Au( $\mu\text{g cm}^{-2}$ )	Pt(atom%)
10	(Pt <sub>AC</sub> -O-Au)-1	0.482	100	0.484
20	(Pt <sub>AC</sub> -O-Au)-2	0.795	104	0.766
30	(Pt <sub>AC</sub> -O-Au)-3	1.047	109	0.961
60	(Pt <sub>AC</sub> -O-Au)-4	1.693	101	1.664
600	(Pt <sub>AC</sub> -O-Au)-5	17.58	105	14.44

**Table S2** Pt loading with different sputtering times using XPS

Sputtering time(s)	Electrode	Pt(atom%)	Au(atom%)
	Au		100
10	(Pt <sub>AC</sub> -O-Au)-1	4.79	95.21
20	(Pt <sub>AC</sub> -O-Au)-2	15.3	84.7
30	(Pt <sub>AC</sub> -O-Au)-3	23.9	76.1
60	(Pt <sub>AC</sub> -O-Au)-4	42.4	57.6
600	(Pt <sub>AC</sub> -O-Au)-5	99	1

**Table S3** The Pt loading capacity, apparent activity and mass specific activity of different low Pt loading electrocatalysts in the references and this work.

Electrodes	Loading ( $\mu\text{g}_{\text{Pt}} \text{cm}^{-2}$ )	$\eta$ at 10 mA $\text{cm}^{-2}$ (mV)	Mass activity ( $\text{A mg}^{-1}$ )	Tafel slope ( $\text{mv dec}^{-1}$ )	Ref
Pt/VS <sub>2</sub> /CP	379.2	77	22.88 @ 200mV	39.46	1
Pt/Ni-Mo-N-O	243.75	89.5@100	10.05 @ 134mV	31.5	2
Pt-a-MoS <sub>3</sub> NDs	84.99	11.5	3.23 @ 20mV	25.4	3
Pt SASs/AG	31.1	12	22.4 @ 50mV	29.33	4
Pt/WO <sub>3</sub> -600	29.94	8	7.015 @ 50mV	35	5
Pt <sub>doped</sub> @WC <sub>x</sub>	22.9	2.4	14.28 @ 100mV	20	6
PtW NP s/C <sup>a</sup>	20.4	19.4	0.566 @ 20mV	27.8	7
Pt <sub>0.04</sub> /Ni-DA	17.72	19	2.13 @ 50mV	34	8
Pt/def-WO <sub>3</sub> @CFC	15.9	42	0.764 @ 50mV	61	9
Pt <sub>3</sub> Ni <sub>2</sub> NWs-S/C	15.0	~23	-----	-----	10
Pt@MoS <sub>2</sub> /NiS <sub>2</sub>	~13.68	34	-----	40	11
MXene/B-Pt	13.4	14	-----	78.6	12
Mo <sub>2</sub> TiC <sub>2</sub> T <sub>x</sub> -Pt <sub>SA</sub>	~12	30	8.3 @ 77mV	30	13
Pt-MoO <sub>2</sub> @PC	11.6	20	11.34 @ 50mV	22	14
Pt/TiO <sub>2</sub> -O <sub>V</sub>	10.45	18	14.82 @ 50mV	12	15
Pt/MOF-C	10.191	28	0.97 @ 50mV	24.4	16
Pt@VNC	9.802	5	-----	21	17
Pt <sub>1</sub> Ru <sub>1</sub> /NMHCS-A	9.58	22	3.49 @ 50mV	38	18
CDs/Pt-PANI	8.1	30	1.22 @ 30mV	41.7	19
Pt <sub>5</sub> /HMCS <sup>a</sup>	7.6	20.7	12.8 @ 30 mV	28.3	20
Pt <sub>1</sub> /N-C	6.25	19	-----	14.2	21
Pt <sub>at</sub> -CoP MNSs/CFC	5.89	13	-----	30.28	22
Pt@mh-3D MXene <sup>a</sup>	4.8	12	19.45 @ 100mV	24.2	23
Pt-GDY-2	4.65	~66	23.64 @ 100mV	46.6	24
Pt-HNCNT	~3.8	15	20.4 @ 50mV	29.1	25
0.66% Pt-WC/CNT	3.4	42	8.9 @ 50mV	-----	26
PtNC/S-C	2.55	11	26.1 @ 20mV	24	27
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> -Pt <sub>SA</sub>	2.38	38	23.21 @ 50mV	45	28

Pt Cs/MoO <sub>2</sub> NSs-L	1.79	47	7.43 @ 50mV	32.6	29
AL-Pt/Pd <sub>3</sub> Pb <sup>a</sup>	~1.6	13.8	7.834 @ 50mV	18	30
CNT/Pt@CdSe-OCPs	~1.58	-----	166.2 @ 150mV	61.3	31
Pt <sub>C</sub> (250)	1.41	37.4	107.7 @ 50mV	30	32
Pt-GT-1	1.4	15	-----	24	33
Pt-Ru dimer	1.38	50	23.1 @ 50mV	28.9	34
F-SnO <sub>2</sub> @Pt	1.2	42	≈10 @ 50mV	34	35
CoPt-Pt <sub>SA</sub> /NDPCF	1.18	31	74.31 @ 50mV	43.65	36
Pt-Ti <sub>x</sub> C	1	24.9	49.69 @ 50mV	22.3	37
Pt/MPNC	0.9	18.6±1.6	56±3 @ 50mV	-----	38
Pt SA/m-WO <sub>3-x</sub>	0.86	47	12.8 @ 50mV	45	39
Pt-SAs/CoNC	0.45	57	18.8 @ 50mV	64.5	40
Pt-SAs/WS <sub>2</sub>	0.415	32	54 @ 100 mV	28	41
PtO <sub>x</sub> /TiO <sub>2</sub>	~0.21	~110	8.68 @ 50mV	40	42
Pt/TiB <sub>x</sub> O <sub>y</sub>	0.191	-----	37.8 @ 50mV	32	43
(Pt <sub>AC</sub> -O-Au)-1	0.48	41	3.02 @ 20mV 49.2 @ 50mV 329 @ 80mV	28.6	This work

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