

Supporting Information

Structural confinement toward suppressing concentration and thermal quenching for improving near-infrared luminescence of Fe³⁺

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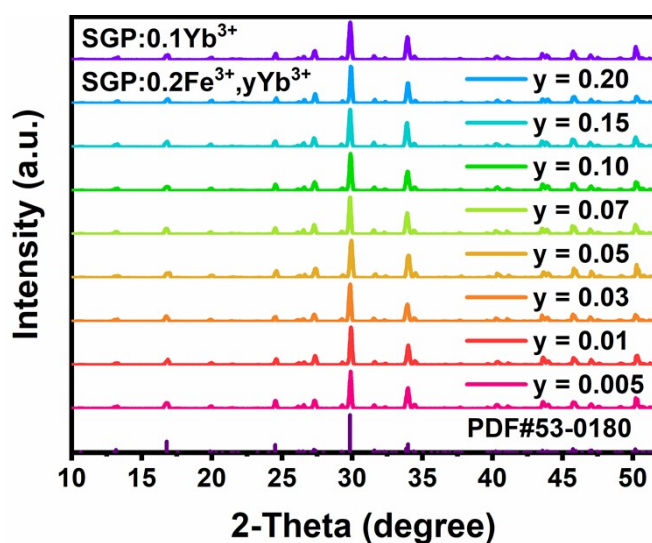


Fig. S1 XRD patterns of SGP:0.2Fe³⁺,yYb³⁺ (0.005 ≤ y ≤ 0.20) and SGP:0.1Yb³⁺.

Table S1. Main parameters of processing and refinement of SGP:0.2Fe³⁺ and Sr₉Fe(PO₄)₇ samples

Compound	SGP:0.2Fe ³⁺	Sr ₉ Fe(PO ₄) ₇
Space Group	<i>I</i> 12/ <i>a</i> 1	<i>I</i> 12/ <i>a</i> 1
Symmetry	Monoclinic	Monoclinic
<i>a</i> (Å)	17.9674(9)	17.9749(8)
<i>b</i> (Å)	10.5955(5)	10.6087(5)
<i>c</i> (Å)	18.2615(8)	18.2795(8)
$\alpha = \gamma$ (°)	90.00	90.00
β (°)	132.778(2)	132.843(2)
<i>V</i> (Å ³)	2551.7(2)	2555.8(2)
<i>R_p</i> (%)	12.6	14.8
<i>R_{wp}</i> (%)	16.0	19.4
<i>R_{exp}</i> (%)	10.03	11.20
χ^2	2.54	3.01

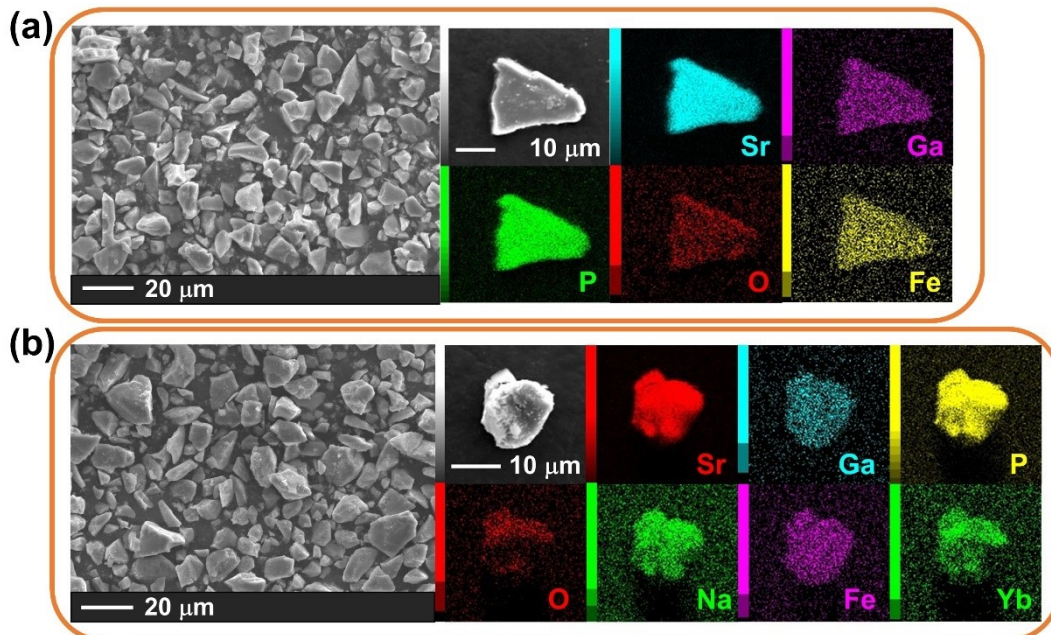


Fig. S2 SEM images and element mapping images of SGP:0.2Fe³⁺ and SGP:0.2Fe³⁺,0.07Yb³⁺.

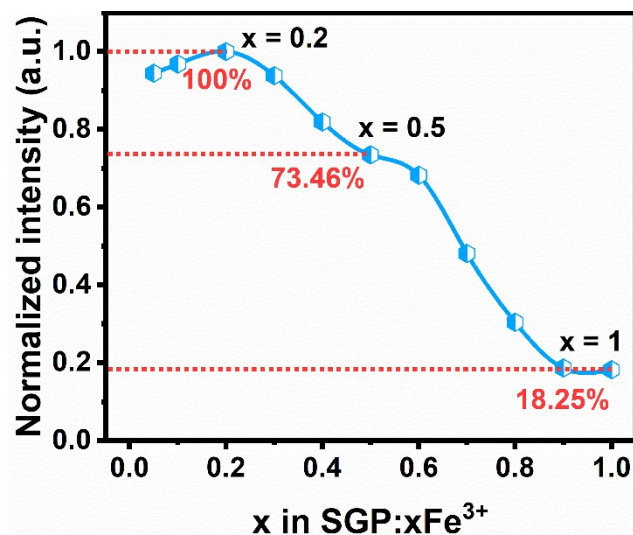


Fig. S3 Relationship between integrated PL intensity and Fe³⁺ concentration in SGP:xFe³⁺.

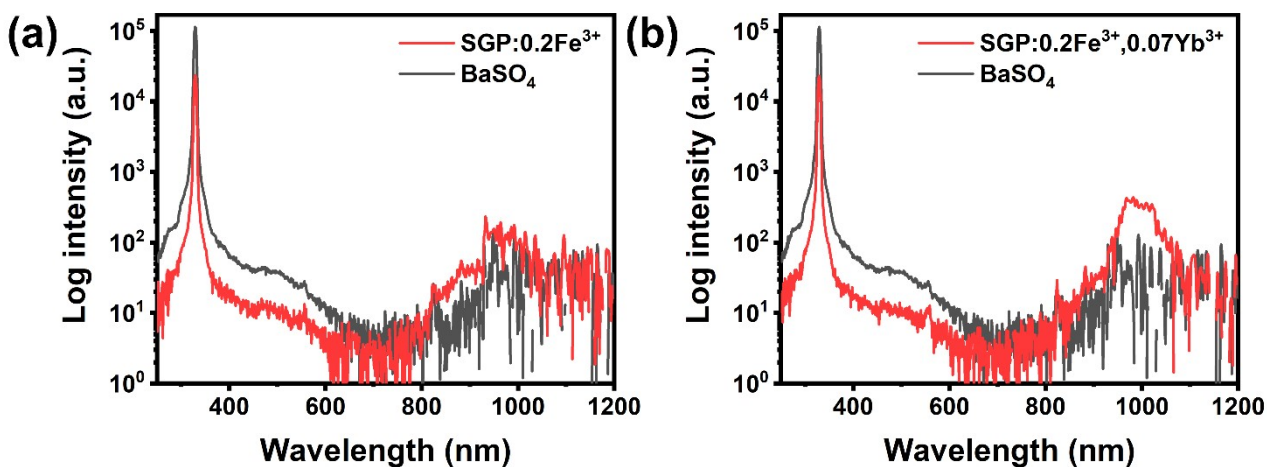


Fig. S4 (a) Spectra of SGP:0.2Fe³⁺ and BaSO₄ to determine the IQE and EQE values. (b) Spectra of SGP:0.2Fe³⁺,0.07Yb³⁺ and BaSO₄ to determine the IQE and EQE values.

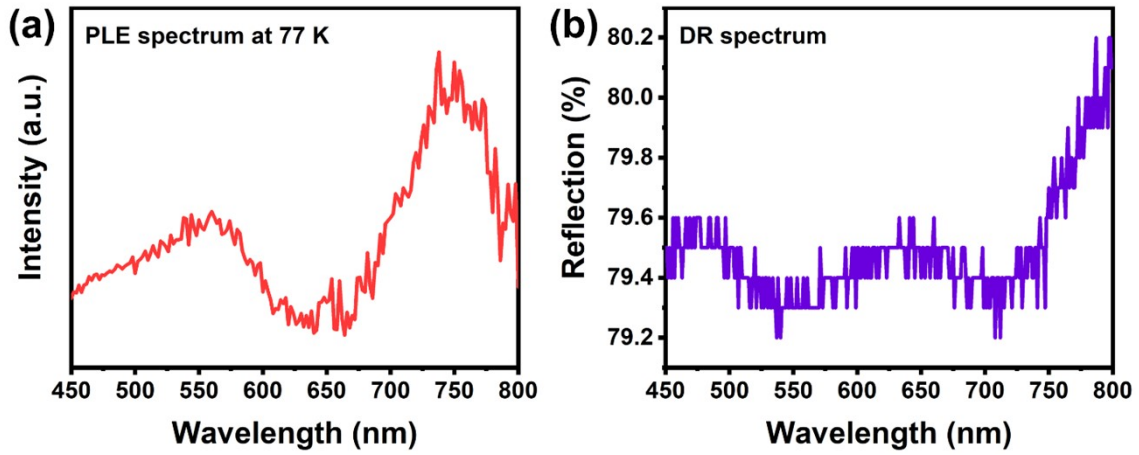


Fig. S5 Magnified (a) PLE spectrum measured at 77 K and (b) DR spectrum ranging from 450 to 800 nm of SGP:0.2Fe³⁺.

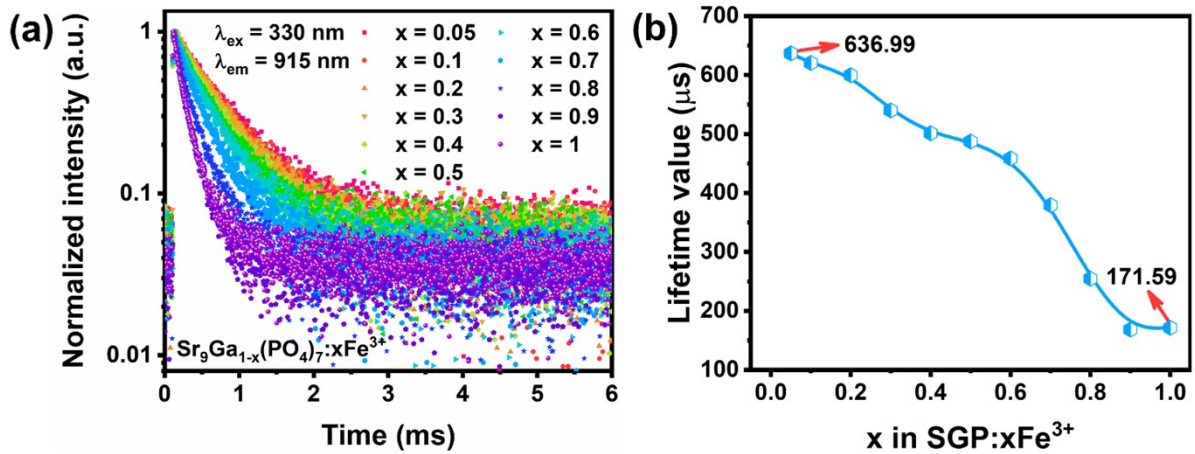


Fig. S6 (a) Luminescence decay curves of SGP:xFe³⁺ ($0.05 \leq x \leq 1$) excited by 330 nm and monitoring at 915 nm. (b) Relationship between lifetime values and Fe³⁺ concentration in SGP:xFe³⁺.

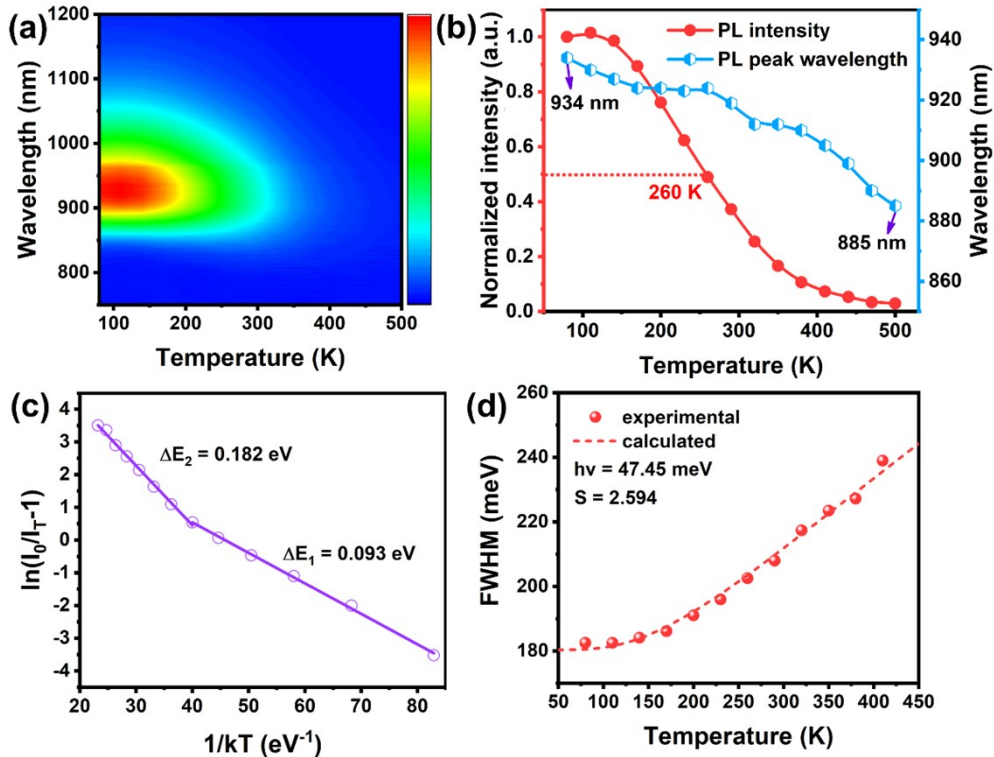


Fig. S7 (a) Temperature-dependent PL spectra at 80-500 K, (b) dependence of PL intensity and PL peak wavelength on temperature, (c) activation energy plots using the Arrhenius equation, (d) fitting of the FWHM as a function of temperature for SGP:0.2Fe³⁺.

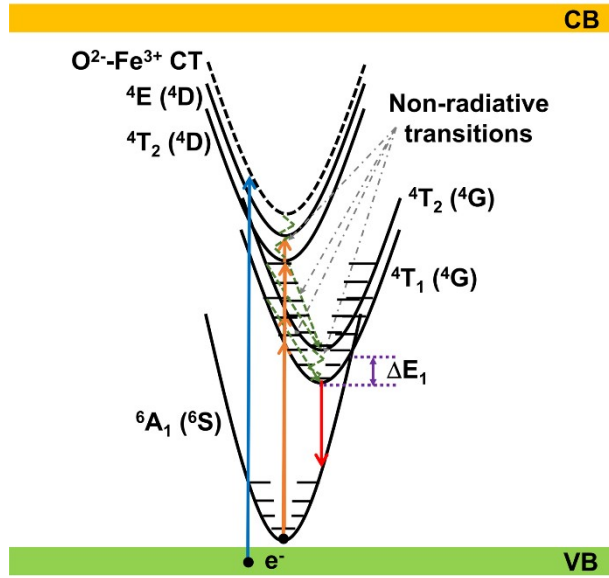


Fig. S8 Configurational coordinate diagram to illustrate thermal quenching behavior.

The temperature-dependence of PL intensity can be described by a modified Arrhenius equation:¹

$$I_T = \frac{I_0}{1 + A \exp(-\Delta E / kT)} \quad (S1)$$

where I_0 is the initial intensity, I_T is the intensity at a given temperature, A is a constant, k is the Boltzmann constant and ΔE is the activation energy for the thermal quenching.

The widening of FWHM can be explained qualitatively based on the following Equation:²

$$FWHM = 2.36 \times hv \times \sqrt{S} \times \sqrt{\coth\left(\frac{hv}{2kT}\right)} \quad (S2)$$

where hv , S , k and T represent the energy of lattice vibration, Huang-Rhys parameter, Boltzmann constant and Kelvin temperature, respectively.

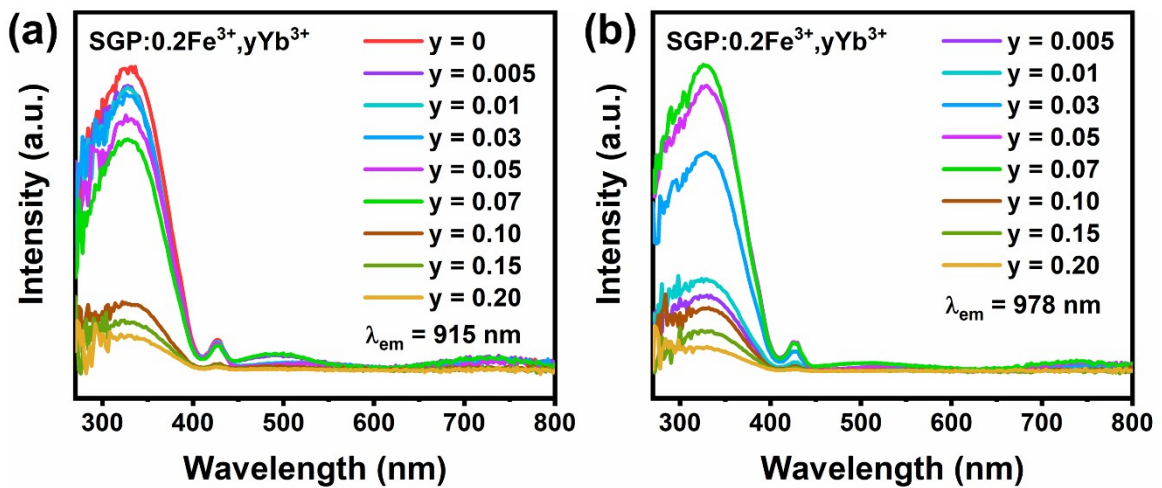


Fig. S9 (a) PLE spectra of SGP:0.2Fe³⁺,yYb³⁺ (0 ≤ y ≤ 0.20) monitoring at 915 nm. (b) PLE spectra of SGP:0.2Fe³⁺,yYb³⁺ (0.005 ≤ y ≤ 0.20) monitoring at 978 nm.

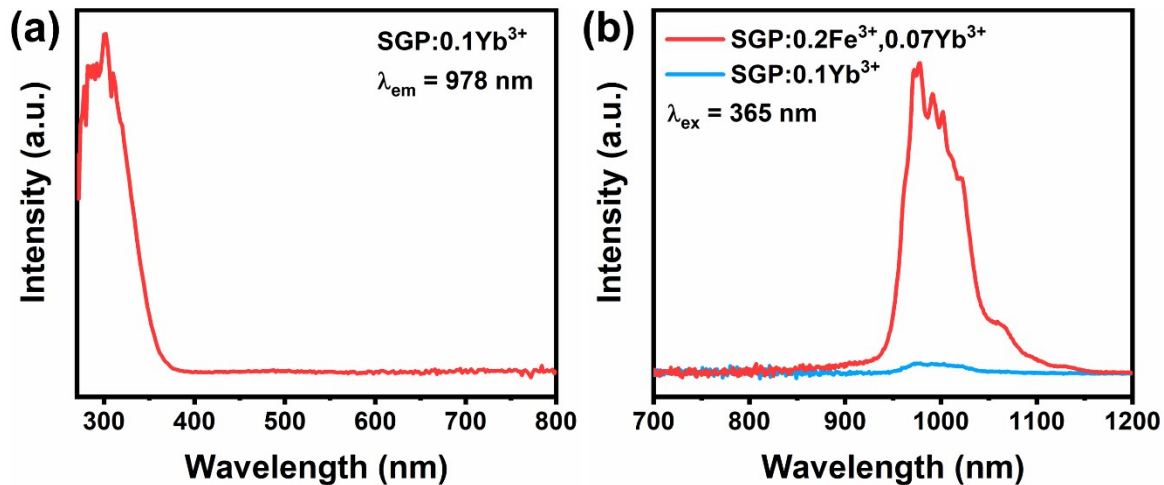


Fig. S10 (a) PLE spectrum of SGP:0.1Yb³⁺ monitoring at 978 nm. (b) PL spectra of SGP:0.1Yb³⁺ and SGP:0.2Fe³⁺,0.07Yb³⁺ excited by 365 nm.

The ET efficiency (η_{ET}) can be calculated by the attenuation of lifetime based on the following equation:³

$$\eta_{ET} = 1 - \frac{\tau_{Fe}}{\tau_{Fe_0}} \quad (S3)$$

where τ_{Fe} is the lifetime of Fe³⁺ in Yb³⁺-doped samples, and τ_{Fe_0} represents the lifetime of Fe³⁺ in Yb³⁺-free samples.

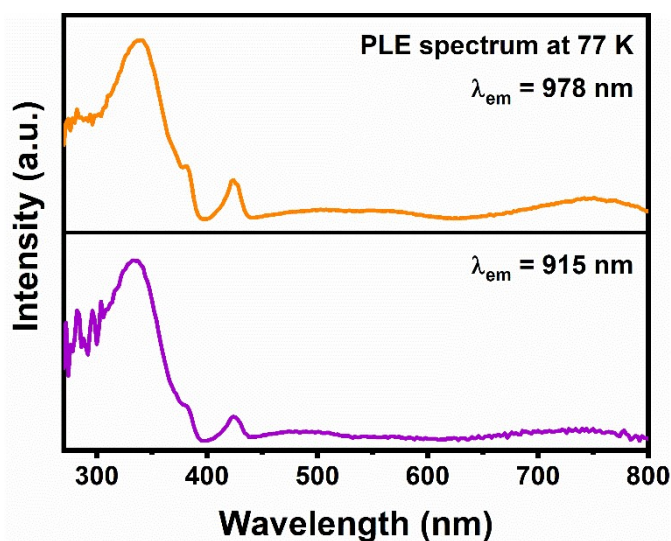


Fig. S11 PLE spectra of SGP:0.2Fe³⁺,0.07Yb³⁺ monitoring at 915 and 978 nm measured at 77 K.

References

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