

## Supporting Information

### From $(\text{NH}_4)_3[\text{Zr}(\text{PO}_4)_2\text{F}]$ to $(\text{NH}_4)_3[\text{Sn}_2(\text{PO}_4)_2]\text{Cl}$ : Rational Design of a Tin-based Short-Wave Ultraviolet Phosphate with Large Optical Anisotropy

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**Table S1.** Atomic coordinates and equivalent isotropic displacement parameters for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

Atom	x/a	y/b	z/c	Ueq [Å <sup>2</sup> ]	S.O.F.	Wyckoff
Sn1	0.61623(4)	0.34466(4)	0.37526(4)	0.02020(17)	1	4d
P1	0.34477(16)	0.36179(15)	0.53080(14)	0.0189(4)	1	4d
Cl1	0.5	0.0386(9)	0.25	0.118(3)	0.802(13)	8d
O1	0.4253(5)	0.3038(5)	0.4422(4)	0.0314(12)	1	4d
O3	0.3261(5)	0.5137(4)	0.5146(4)	0.0257(11)	1	4d
O2	0.2000(4)	0.2976(4)	0.5259(4)	0.0264(11)	1	4d
O4	0.4144(6)	0.3331(5)	0.6278(4)	0.0402(14)	1	4d
N2	0.6962(7)	0.3401(7)	0.6766(5)	0.048(2)	1	4d
H2A	0.746423	0.341314	0.62118	0.058	1	4d
H2B	0.714013	0.413514	0.71205	0.058	1	4d
H2C	0.717753	0.267654	0.71205	0.058	1	4d
H2D	0.604683	0.337914	0.66075	0.058	1	4d
N1	0	0.5	0.5	0.080(5)	1	8d
H1A	-0.01353	0.51373	0.43541	0.096	0.5	4d
H1B	-0.07918	0.47475	0.52815	0.096	0.5	4d
H1C	0.02955	0.57559	0.52815	0.096	0.5	4d
H1D	0.06458	0.43449	0.50847	0.096	0.5	4d
Cl2	0.916(5)	0.571(3)	0.694(3)	0.14(2)	0.099(6)	4d

**Table S2.** Anisotropic displacement parameters for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ , in  $\text{\AA}^2$ .

Atom	U11	U22	U33	U12	U13	U23
Sn1	0.0203(2)	0.0209(3)	0.0194(3)	0.00118(17)	0.00197(18)	-0.00050(19)
P1	0.0156(7)	0.0164(7)	0.0247(10)	-0.0016(6)	0.0008(7)	0.0006(7)
Cl1	0.087(4)	0.188(8)	0.077(5)	0.00000	0.029(4)	0.00000
O1	0.018(2)	0.033(3)	0.042(3)	-0.005(2)	0.008(2)	-0.012(3)
O3	0.034(3)	0.014(2)	0.030(3)	-0.001(2)	0.001(2)	0.0009(19)
O2	0.019(2)	0.021(2)	0.039(3)	-0.0048(19)	0.003(2)	0.004(2)
O4	0.032(3)	0.056(4)	0.033(3)	0.000(3)	-0.010(2)	0.015(3)
N2	0.047(4)	0.070(5)	0.028(4)	0.023(4)	-0.007(3)	-0.001(4)
N1	0.022(5)	0.056(7)	0.161(15)	0.005(5)	0.006(7)	0.052(8)
Cl2	0.24(5)	0.052(17)	0.14(4)	0.09(2)	-0.13(4)	-0.05(2)

**Table S3.** Selected bond lengths for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

Atom 1,2	d 1,2 [Å]
Sn1-O1	2.093(5)
Sn1-O3	2.122(5)
Sn1-O2	2.106(5)
P1-O1	1.537(5)
P1-O3	1.540(4)
P1-O2	1.539(4)
P1-O4	1.497(6)

**Table S4.** Selected bond angles for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

<b>Atom 1,2,3</b>	<b>Angle 1,2 ,3[°]</b>
O1-Sn1-O3	93.41(18)
O1-Sn1-O2	86.32(19)
O2-Sn1-O3	84.52(18)
O1-P1-O3	108.6(3)
O1-P1-O2	105.7(3)
O2-P1-O3	107.2(3)
O4-P1-O1	112.2(3)
O4-P1-O3	111.4(3)
O4-P1-O2	111.5(3)
P1-O1-Sn1	135.0(3)
P1-O3-Sn1	121.6(3)
P1-O2-Sn1	127.0(3)

**Table S5.** Bond valence sums (BVS) for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

<b>Atom</b>	<b>BVS</b>
Sn1	2.15
P1	5.07
O1	-1.99
O2	-1.95
O3	-1.92

**Figure S1.** Measured and calculated PXRD patterns for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

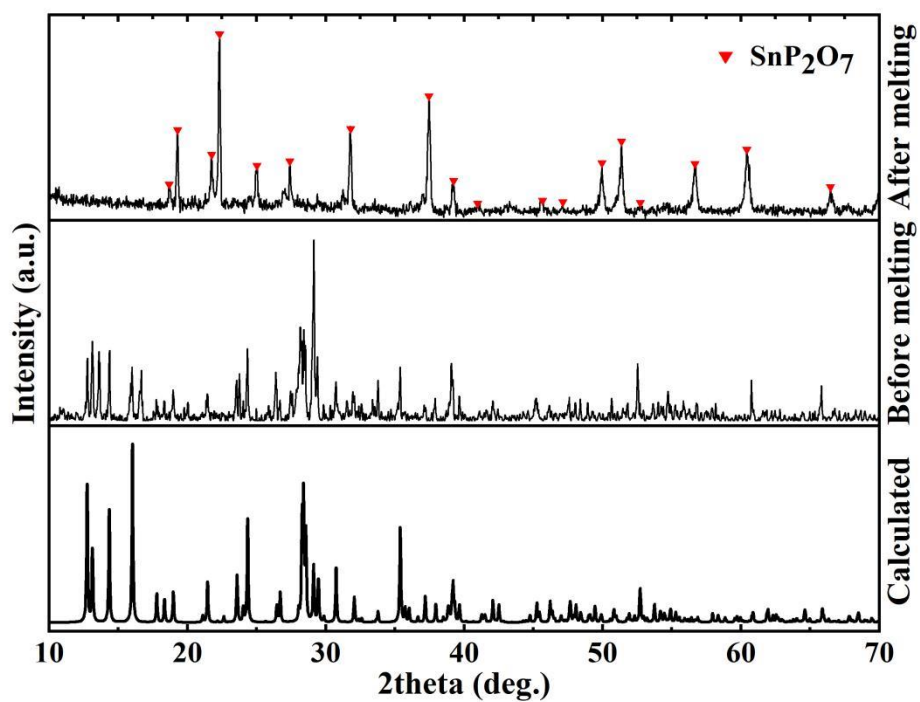


Figure S2. Measured TG curve for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

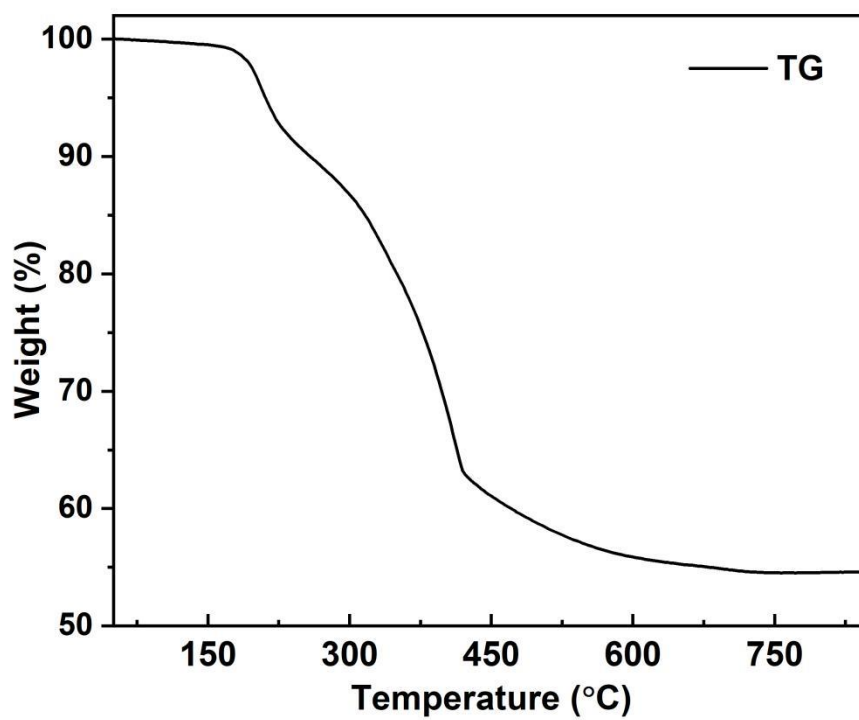
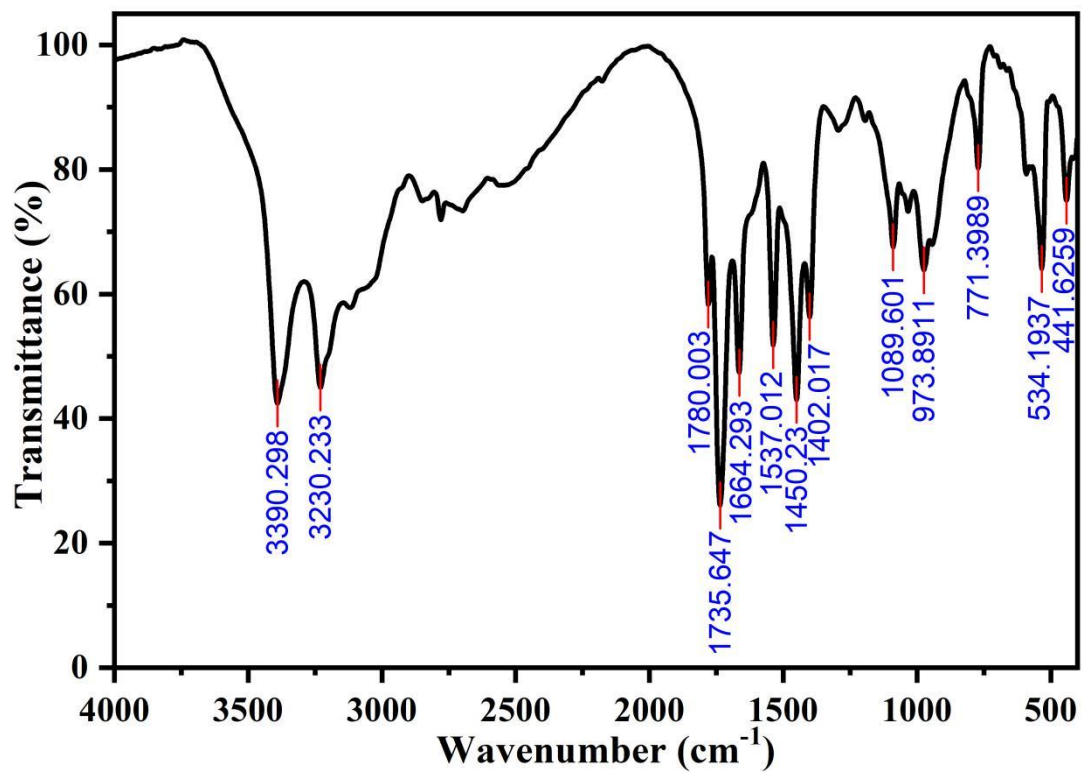




Figure S3. IR spectrum for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .



**Figure S4.** EDS map for  $(\text{NH}_4)_3\text{Sn}_2(\text{PO}_4)_2\text{Cl}$ .

