## Supporting Information

# Visible-Light-Induced Electron-Donor-Acceptor (EDA) complex-Initiated Synthesis of Non-Anomeric S-Aryl 

## Glycosides

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## 1.General Information

Unless otherwise specified, materials were purchased from commercial suppliers and used without further purification. All manipulations were performed in a dried sealed tube equipped with a magnetic stir bar under an argon atmosphere. Except for the specially mentioned dry solvent, all the solvents were treated according to general methods. All the reactions were monitored by TLC and were visualized using UV light. The product purification was done using silica gel column chromatography. TLC characterization was performed with precoated silica gel GF254 ( 0.2 mm ), while column chromatography characterization was performed with silica gel (100-200 mesh). ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded with tetramethylsilane (TMS, $\delta=0.00 \mathrm{ppm}$ ) as the internal standard. ${ }^{1} \mathrm{H}$ NMR spectra were recorded at 400 or 600 MHz (Varian), and ${ }^{13} \mathrm{C}$ NMR spectra were recorded at 100 or 150 MHz (Varian). ${ }^{19} \mathrm{~F}$ NMR spectra were recorded at 376 MHz . Chemical shifts are reported in ppm downfield from $\mathrm{CDCl}_{3}(\delta=7.26 \mathrm{ppm})$ for ${ }^{1} \mathrm{H}$ NMR, and chemical shifts for ${ }^{13} \mathrm{C}$ NMR spectra are reported in ppm relative to the central $\mathrm{CDCl}_{3}(\delta=77.0$ ppm) or DMSO-d6 ( $\delta=39.6 \mathrm{ppm})$. Coupling constants were given in Hz . The following notations were used: br-broad, s-singlet, d-doublet, t -triplet, q-quartet, m-multiplet, dd-doublet of doublet, dt-doublet of triplet, tdtriplet of doublet, and ddd-doublet of doublet of doublet. Melting points
were measured with a YRT-3 melting point apparatus (Shantou Keyi Instrument \& Equipment Co., Ltd., Shantou, China). The blue light source (455nm) was provided by Shanghai 3S Technology Co., Ltd SSSTECH-LAL1CV 1.0 parallel reactor (Figure S1). The volume of the reaction tube is 10 ml .


Figure S1. Photoreactor and reaction tube in this study

## 2.Substrates synthesis

### 2.1 General procedure for the synthesis of carboxylic acid


heme S1. General procedure for the synthesis of ribose-derived carboxylic acid.

Step 1 ${ }^{[1]}$ : To a 250 mL round bottom flask was added D-ribose (10.0 g, $66.1 \mathrm{mmol}, 1$ equiv.), methanol ( 40 mL ), and acetone ( 40 mL ). Next, concentrated $\mathrm{HCl}(1 \mathrm{~mL})$ was added at room temperature, the flask was
heated to $75^{\circ} \mathrm{C}$ for 4 h . The solution was cooled to room temperature and filtered to remove the solid then sat. $\mathrm{Na}_{2} \mathrm{CO}_{3}$ was added to adjust pH above 7. The solution was extracted by ethyl acetate, washed by brine, dried by $\mathrm{Na}_{2} \mathrm{SO}_{4}$, combined the organic phase and concentrated with reduced pressure. The desired crude product was obtained as a faint yellow oil (10.8 g, 80\%)

Step $2^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\mathrm{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid in more than $89 \%$ yield.

cheme S2. General procedure for the synthesis of Bn-protected ribosederived carboxylic acid.

Step $1^{[3]}$ : To a solution of methyl beta D-ribofuranoside (1.0 equiv) in
pyridine ( 0.5 M ) was added trityl chloride (1.1 equiv) and the reaction mixture was stirred at room temperature overnight. The solvent was removed in vacuo and the residue was coevaporated with toluene, dissolve in EtOAc and washed with 1 M aqueous $\mathrm{CuSO}_{4}$ solution, brine dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated to afford desired product. Purification by flash column chromatography (cyclohexane/EtOAc) yielded product as a white foam ( $86 \%$ yield).

Step $2^{[4]}$ : The obtained compound from above step was dissolved in DMF ( 0.2 M ) and cooled to $0^{\circ} \mathrm{C}$. Sodium hydride ( $60 \%$ in mineral oil, 4.0 equiv) was added portion-wise. Then, $\mathrm{BnCl} / \mathrm{MeI}$ ( 3.0 equiv) was added dropwise. The mixture was stirred overnight and allowed to warm to room temperature. The reaction was quenched with MeOH and thiourea, and stirred for another 2 h at room temperature. The solvents were evaporated in vacuo and the residue was dissolved in EtOAc, and then washed with water. The aqueous phase was extracted with EtOAc, the combined organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo to obtain crude product Purification by flash column chromatography (cyclohexane $/ \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc}$ ) yielded product as a white foam (about $80 \%$ yield).

Step $3^{[2]}$ : The obtained compound from above step was dissolved in $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}(10: 1: 0.1, \mathrm{v} / \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ followed by addition of TsOH (0.5 equiv). After being stirred for 20 min at room temperature, the
resultant mixture was extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluent to get the alcohol.

Step $4^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\operatorname{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid more than 78\% yield.



Scheme S3. General procedure for the synthesis of arabinose-derived carboxylic acid.

Step $1^{[5]}$ : To a solution of methyl beta $D$-arabinose (1.0 equiv) in $\operatorname{DMF}(0.1 \mathrm{M})$ was added imidazole ( 1.5 equiv) and cooled to $0^{\circ} \mathrm{C}$. Then TBDPSCl (1.5 equiv) was added dropwise. The mixture was stirred
overnight and allowed to warm to room temperature. The solvents were evaporated in vacuo and the residue was dissolved in EtOAc, and then washed with water. The aqueous phase was extracted with EtOAc, the combined organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo to obtain crude product Purification by flash column chromatography (cyclohexane/ EtOAc) yielded product as a colorless oil (83\% yield).

Step $2^{[6]}$ : The obtained compound from above step was dissolved in acetone and 2, 2-dimethoxypropane ( 10 mL ) and $p$-toluenesulfonic acid was added. The reaction was neutralized by triethylamine and concentrated after stired 30 min at room temperature. The solvents were evaporated in vacuo, purification by flash column chromatography (cyclohexane / EtOAc). yielded product as a colorless oil (84\% yield).

Step 3 ${ }^{[4]}$ : The obtained compound from above step was dissolved in DMF ( 0.1 M ) and cooled to $0{ }^{\circ} \mathrm{C}$. Sodium hydride ( $60 \%$ in mineral oil, 2.0 equiv) was added portion wise. Then, BnCl ( 2.0 equiv) was added dropwise. The mixture was stirred overnight and allowed to warm to room temperature. The reaction was quenched with MeOH and thiourea, and stirred for another 2 h at room temperature. The solvents were evaporated in vacuo and the residue was dissolved in EtOAc, and then washed with water. The aqueous phase was extracted with EtOAc, the combined organic phases were washed with water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$,
filtered, and concentrated in vacuo to obtain crude product as an oil. Purification by flash column chromatography (cyclohexane / EtOAc). yielded product as a white solid ( $87 \%$ yield).

Step $4^{[5]}$ : The obtained compound from above step was dissolved in anhydrous THF ( 0.5 M ), was reacted with TBAF ( 1 M in THF, 1.2 equiv). The solution was stirred at room temperature for 2 h . The solvents were evaporated in vacuo, purification by flash column chromatography (cyclohexane / EtOAc). yielded product as a white solid (79\% yield).

Step $5^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\mathrm{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid 78\% yield.


Scheme S4. General procedure for the synthesis of xylose-derived
carboxylic acid.
Step 1 ${ }^{[3]}$ : To a solution of 1,2-O-Isopropylidene-alpha-D-xylofuranose (1.0 equiv) in pyridine ( 0.5 M ) was added trityl chloride (1.1 equiv) and the reaction mixture was stirred at room temperature overnight. The solvent was removed in vacuo and the residue was coevaporated with toluene, dissolve in EtOAc and washed with 1 M aqueous $\mathrm{CuSO}_{4}$ solution, brine dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated to afford desired product. Purification by flash column chromatography (cyclohexane/EtOAc) yielded product as a white foam ( $83 \%$ yield).

Step 2 ${ }^{[4]}$ : The obtained compound from above step was dissolved in DMF ( 0.2 M ) and cooled to $0^{\circ} \mathrm{C}$. Sodium hydride ( $60 \%$ in mineral oil, 2.0 equiv) was added portion-wise. Then, $\mathrm{BnCl}(1.5$ equiv) was added dropwise. The mixture was stirred overnight and allowed to warm to room temperature. The reaction was quenched with MeOH and thiourea, and stirred for another 2 h at room temperature. The solvents were evaporated in vacuo and the residue was dissolved in EtOAc, and then washed with water. The aqueous phase was extracted with EtOAc, the combined organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo to obtain crude product Purification by flash column chromatography (cyclohexane $/ \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc}$ ) yielded product as a white foam ( $81 \%$ yield).

Step $3^{32]}$ : The obtained compound from above step was dissolved in
$\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}(10: 1: 0.1, \mathrm{v} / \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ followed by addition of TsOH ( 0.5 equiv). After being stirred for 20 min at room temperature, the resultant mixture was extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluent to get the alcohol.

Step $4^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\mathrm{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid 78\% yield.



Scheme S5. General procedure for the synthesis of $\mathrm{Bn} / \mathrm{Me}$-protected hexose-derived carboxylic acid.

Step $1^{[3]}$ :To a solution of methyl- $\alpha$-D-glucopyranoside (arbutin) (1.0
equiv) in pyridine ( 0.5 M ) was added trityl chloride (1.1 equiv) and the reaction mixture was stirred at room temperature overnight. The solvent was removed in vacuo and the residue was coevaporated with toluene, dissolve in EtOAc and washed with 1 M aqueous $\mathrm{CuSO}_{4}$ solution, brine dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated to afford desired product. Purification by flash column chromatography (cyclohexane/EtOAc) yielded product as a white foam ( $80 \%$ yield).

Step. $2^{[1,4]}$ : The obtained compound from above step was dissolved in DMF ( 0.2 M ) and cooled to $0^{\circ} \mathrm{C}$. Sodium hydride ( $60 \%$ in mineral oil, 6.0 equiv) was added portion-wise. Then, $\mathrm{BnCl} / \mathrm{MeI}$ (4.5 equiv) was added dropwise. The mixture was stirred overnight and allowed to warm to room temperature. The reaction was quenched with MeOH and thiourea, and stirred for another 2 h at room temperature. The solvents were evaporated in vacuo and the residue was dissolved in EtOAc, and then washed with water. The aqueous phase was extracted with EtOAc, the combined organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo to obtain crude product Purification by flash column chromatography (cyclohexane $/ \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc}$ ) yielded product as a white foam (about 78\% yield).

Step. $3^{[2]}$ : The obtained compound from above step was dissolved in $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}(10: 1: 0.1, \mathrm{v} / \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ followed by addition of TsOH ( 0.5 equiv). After being stirred for 20 min at room temperature, the
resultant mixture was extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluent to get the alcohol.

Step. $4^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\operatorname{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid more than $80 \%$ yield.


Scheme S6. General procedure for the synthesis of fructose-derived carboxylic acid.

## (3aR,5aR,8aR,8bS)-2,2,7,7-tetramethyltetrahydro-3aH-

bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran-3a-carboxylic acid ${ }^{[2]}$ : To a 50 mL flask, the diacetonefructose ( $3 \mathrm{~g}, 11.5 \mathrm{mmol}, 1.0$ equiv) was treated with TEMPO ( $360 \mathrm{mg}, 2.3 \mathrm{mmol}, 0.2$ equiv.) and $\mathrm{PhI}(\mathrm{OAc})_{2}(11.1 \mathrm{~g}$,
$34.5 \mathrm{mmol}, 3$ equiv. $)$ in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(60 \mathrm{ml}, 1: 1, \mathrm{v} / \mathrm{v})$ at room temperature for 9 h , and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid(2.6 g, 83\%).


Scheme S7. General procedure for the synthesis of galactose-derived carboxylic acid.
((3aR,5R,5aS,8aS,8bR)-2,2,7,7-tetramethyltetrahydro-5H-bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran-5-yl)methanol ${ }^{[7]}$ :To a 250 mL round bottom flask was added D-galactose ( $7.0 \mathrm{~g}, 38.8 \mathrm{mmol}, 1$ equiv.), acetone $(250 \mathrm{~mL})$. Next, concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}(7.7 \mathrm{~mL})$ was added at $0{ }^{\circ} \mathrm{C}$. The reaction mixtures were stirred at room temperature for $3-5 \mathrm{~h}$ and then neutralized by the addition of sat. $\mathrm{Na}_{2} \mathrm{CO}_{3}$ until $\mathrm{pH}=7$. The precipitate was removed by filtration and the filtrates were combined and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluent to get the product as a colorless oil. $(8.5 \mathrm{~g}, 84 \%)$.
bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran-5-carboxylic acid ${ }^{[2]}$ : The acid was prepared according to the reported paper. TEMPO ( 0.2 equiv.) and $\mathrm{PhI}(\mathrm{OAc})_{2}$ (3 equiv.) were added to a solution of alcohol obtained from above step (1.0 equiv.) in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(1: 1, \mathrm{v} / \mathrm{v}, 0.2 \mathrm{M})$ at room temperature. The reaction mixture was stirred for 9 h at room temperature, and then saturated aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(5 \mathrm{~mL})$ was added. After being stirred for 20 min at room temperature, the resultant mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluant to afford carboxylic acid $87 \%$ yield.

### 2.2 Synthetic procedure of N -hydroxyphthalimide esters



A round-bottom flask or culture tube was charged with carboxylic acid derived from sugar (1.0 equiv), N -hydroxyphthalimide (1.1 equiv) and 4-dimethylaminopyridine ( 0.05 equiv). Dichloromethane was added (0.1- 0.2 M ), and the mixture was stirred vigorously. Dicyclohexylcarbodiimide (DCC, 1.1 equiv) was added, and then the mixture was allowed to stir until the acid was consumed (determined by TLC). Typical reaction times were between 0.5 h to 12 h . The mixture was filtered through a thin pad of Celite and rinsed with additional
$\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The solvent was removed under reduced pressure, and purification of the crude mixture by column chromatography (DCM/hexane/ethyl acetate as eluent) afforded the desired N hydroxyphthalimide esters.


1,3-dioxoisoindolin-2-yl (3aS,4S,6R,6aR)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxole-4-carboxylate (1a): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(\mathbf{4 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 7.89$ (dd, $J=5.5$, $3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.80(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.36(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~s}, 1 \mathrm{H}), 5.00(\mathrm{~s}, 1 \mathrm{H})$, $4.66(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.50(\mathrm{~s}, 3 \mathrm{H}), 1.52(\mathrm{~s}, 3 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (2S,3S,4R,5R)-3,4-bis(benzyloxy)-5-methoxytetrahydrofuran-2carboxylate (1b): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.90(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.80$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.42-7.28(\mathrm{~m}, 10 \mathrm{H}), 5.00(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.79(\mathrm{~d}, J=11.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.71-4.60(\mathrm{~m}, 4 \mathrm{H}), 3.87(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (2S,3S,4R,5R)-3,4,5-trimethoxytetrahydrofuran-2-carboxylate (1c): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(600 \mathrm{MHz}$, Chloroform-d) $\delta 7.93-7.85$ (m, 2H), $7.85-7.76$ (m, 2H), 5.05 $(\mathrm{s}, 1 \mathrm{H}), 4.85(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.47(\mathrm{~s}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 3 \mathrm{H}), 3.53(\mathrm{~s}, 3 \mathrm{H}), 3.46(\mathrm{~s}$, $3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (3aR,5S,6R,6aR)-6-(benzyloxy)-2,2-dimethyltetrahydrofuro[2,3-
d] [1,3]dioxole-5-carboxylate (1d): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathbf{~ M H z}$, Chloroform- $\boldsymbol{d}) \delta 7.91$ (dd, $J=5.4$, $3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.80(\mathrm{dd}, J=5.4,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.46-7.27(\mathrm{~m}, 5 \mathrm{H}), 6.11(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H})$, $5.20(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.92(\mathrm{~d}, J=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{~d}, J=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{~d}, J=3.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.46(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (3aS,5S,6S,6aS)-6-(benzyloxy)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole-5-carboxylate (1e): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.90$ (dd, $J=5.5$, $3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.80(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.28(\mathrm{~m}, 5 \mathrm{H}), 6.07(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H})$, $5.07(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.78-4.72(\mathrm{~m}, 2 \mathrm{H}), 4.71-4.65(\mathrm{~m}, 2 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13} \mathbf{C}-$ NMR ( 100 MHz , Chloroform- $\boldsymbol{d}$ ) $\delta 166.3,161.4,136.5,134.9,128.8,128.7,128.3$, $128.0,124.1,113.8,107.0,84.4,82.9,80.7,72.5,25.9,25.7$ HRMS (ESI) m/z [M + Na] ${ }^{+}$ calculated for 462.1159 , found 462.1161 .


1,3-dioxoisoindolin-2-yl
(3aR,5S,5aR,8aS,8bR)-2,2,7,7-tetramethyltetrahydro-5H-bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran-5-carboxylate (1f): ${ }^{1} \mathrm{H}-\mathrm{NMR} \quad(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.90(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.70(\mathrm{~d}, J=$ $5.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.84(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{qd}, J=7.5,2.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.46(\mathrm{dd}, J=5.0,2.6 \mathrm{~Hz}$, $1 \mathrm{H}), 1.61(\mathrm{~s}, 3 \mathrm{H}), 1.54(\mathrm{~s}, 3 \mathrm{H}), 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.37(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl
(3aR,5aR,8aR,8bS)-2,2,7,7-tetramethyltetrahydro-3aH-bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran-3a-carboxylate (1g): ${ }^{1} \mathbf{H}-\mathrm{NMR} \quad(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.89(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.88(\mathrm{~d}, J=$ $2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{dd}, J=7.9,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.31(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 2 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H})$, $1.57(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 1.37(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (2S,3S,4S,5R,6S)-3,4,5-tris(benzyloxy)-6-methoxytetrahydro-2H-pyran-2-carboxylate (1h): ${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}(400 \mathbf{~ M H z}$, Chloroform-d) $\delta 7.90$ (dd, $J=5.5,3.1$ $\mathrm{Hz}, 2 \mathrm{H}), 7.80(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.43-7.26(\mathrm{~m}, 15 \mathrm{H}), 4.99(\mathrm{~d}, J=10.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.93$ $(\mathrm{d}, J=10.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.86(\mathrm{~d}, J=11.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.82(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=3.4 \mathrm{~Hz}$, $1 \mathrm{H}), 4.66(\mathrm{~d}, J=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.58(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.07(\mathrm{t}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.96-3.90$ $(\mathrm{m}, 1 \mathrm{H}), 3.63(\mathrm{dd}, J=9.6,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.49(\mathrm{~s}, 3 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl (2S,3S,4S,5R,6S)-3,4,5,6-tetramethoxytetrahydro-2H-pyran-2carboxylate (1i): ${ }^{1} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.90$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.81$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.95(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.65(\mathrm{~d}, J=1.7$ $\mathrm{Hz}, 6 \mathrm{H}), 3.57(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{~s}, 3 \mathrm{H}), 3.52(\mathrm{~s}, 3 \mathrm{H}), 3.50(\mathrm{~d}, J=2.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.30(\mathrm{dd}$, $J=9.5,3.5 \mathrm{~Hz}, 1 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$


1,3-dioxoisoindolin-2-yl
(2S,3S,4S,5R,6S)-3,4,5-tris(benzyloxy)-6-(4-(benzyloxy)phenoxy)tetrahydro-2H-pyran-2-carboxylate (1j): ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.91$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.81 (dd, $J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.48-7.27$ (m, $20 \mathrm{H}), 7.13-7.06(\mathrm{~m}, 2 \mathrm{H}), 6.98-6.92(\mathrm{~m}, 2 \mathrm{H}), 5.08-5.02(\mathrm{~m}, 3 \mathrm{H}), 5.00-4.93(\mathrm{~m}, 3 \mathrm{H})$, $4.85(\mathrm{dd}, J=10.7,7.2 \mathrm{~Hz}, 3 \mathrm{H}), 4.38(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.12-4.05(\mathrm{~m}, 1 \mathrm{H}), 3.85-3.76(\mathrm{~m}$, $2 \mathrm{H})$. The compound was identified by spectral comparison with literature data. ${ }^{[4]}$

### 2.3 Synthetic procedure of ibuprofen derivant



Sept 1: To a solution of ibuprofen ( 1.0 equiv) in DCM ( 0.5 M ) w as added oxalyl chloride ( 3.0 equiv) and the reaction mixture was s tirred at room temperature for 6 h . The solvent was removed in va cuo to get the crude product.

Sept 2: To a solution of 4,4'-dithiodiphenol (1.0 equiv) in DCM ( 0.2 M ) was added pyridine ( 2.0 equiv) and cooled to $0{ }^{\circ} \mathrm{C}$.Then the obtained compound from above step dissolved in DCM ( 0.2 M ) was added dropwise. The mixture was stirred 6 h and allowed to warm to room temperature. The solvents were evaporated in vacuo and the residue was purified by flash column chromatography (cyclohexane/ EtOAc) to afford the product more than $95 \%$ yield.

Sept 3: The obtained compound from above step was dissolved in acetic acid ( 0.2 M ) followed by addition of Zn powder (8.5 equiv). After refluxed for 4 h , the resultant mixture was filtered and the filtrate was extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel using hexane/EtOAc as eluent to get the product. ( $93 \%$ yield).


4-mercaptophenyl 2-(4-isobutylphenyl)propanoate (2z): Pale oil ${ }^{\mathbf{1}} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.28(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.26-7.21(\mathrm{~m}, 2 \mathrm{H}), 7.14(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.93$ $-6.82(\mathrm{~m}, 2 \mathrm{H}), 3.91(\mathrm{q}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 1 \mathrm{H}), 2.47(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.86(\mathrm{dp}, J=$ $13.6,6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.59(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.91(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(100 \mathrm{MHz}$, Chloroform-d) $\delta 173.2,149.2,140.9,137.1,130.8,129.5,127.5,127.2$ 122.2, 45.2, 45.1, 30.2, 22.4, 18.5. HRMS (ESI) m/z [M + Na] calculated for 649.2417, found 649.2419.

## 3. General Procedures and Optimization of Reaction Conditions

3.1 General procedure (standard condition): To an oven-dried 10 mL glass tube equipped with a stir bar, was added glycosyl NHP ester 1 (0.15 mmol), thiophenol 2 ( 0.1 mmol ), $i \operatorname{Pr}_{2} \operatorname{NEt}(0.1 \mathrm{mmol})$. The tube was evacuated and back-filled with $\mathrm{N}_{2}$ (three times), then sealed with rubber stopper and parafilm. Then, anhydrous DMSO ( 1 mL ) was added using a syringe. The solution was then stirred at room temperature under the irradiation of 12 W Blue LEDs for 24 h using electronic fan to cool the tube (reaction setup shown in Figure ). After completion of the reaction, 5 mL water was added and extracted by ethyl acetate $(3 \times 5 \mathrm{~mL})$. The combined organic layer was washed with brine $(5 \mathrm{~mL})$ and then dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated in vacuum. The desired products were obtained in the corresponding yields after purification by flash chromatography on silica gel eluting with hexane/ethyl acetate or hexane/dichloromethane.

### 3.2 Optimization of Reaction Conditions

Table S1. Base screening.

|  <br> 1a |  |  |  |
| :---: | :---: | :---: | :---: |
| entry ${ }^{\text {a }}$ | base | yield ${ }^{\text {b }}$ (\%) | d.r. |
| 1 | DIPEA | 85 | $>20: 1$ |
| 2 | TMEDA | 73 | $>20: 1$ |
| 3 | DABCO | trace | -- |
| 4 | $\mathrm{Et}_{3} \mathrm{~N}$ | 62 | $>20: 1$ |
| 5 | DMAP | 75 | $>20: 1$ |
| 6 | $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ | trace | -- |
| 7 | $\mathrm{Na}_{2} \mathrm{CO}_{3}$ | 52 | $>20: 1$ |
| 8 | CsF | 39 | $>20: 1$ |

${ }^{\text {a }}$ Reaction conditions: A mixture of 1a ( 0.15 mmol$)$, 2a $(0.10 \mathrm{mmol})$, Base $(0.10 \mathrm{mmol})$ in DMA ( 1.0 mL ) was irradiated with 12 W blue LEDs at room temperature under $\mathrm{N}_{2}$ for 24 h ; ${ }^{\text {b }}$ Isolated yields.

Table S2. Light source screening.


| 3 | Blue LED | 85 | $>20: 1$ |
| :---: | :---: | :---: | :---: |
| 4 | Green LED | 77 | $>20: 1$ |
| 5 | White LED | 67 | $>20: 1$ |

${ }^{\text {a }}$ Reaction conditions: A mixture of $1 \mathrm{a}(0.15 \mathrm{mmol})$, $2 \mathrm{a}(0.10 \mathrm{mmol})$, DIPEA $(0.10 \mathrm{mmol})$ in DMA ( 1.0 mL ) was irradiated with 12 W LEDs at room temperature under $\mathrm{N}_{2}$ for 24 h ; ${ }^{\mathrm{b}}$ Isolated yields.

Table S3. Solvent screening.

|  <br> 1a |  <br> $2 a$ |  |  |
| :---: | :---: | :---: | :---: |
| entry ${ }^{\text {a }}$ | solvent | yield ${ }^{\text {b }}$ (\%) | d.r. |
| 1 | 1,4-Dioxane | N.R. ${ }^{\text {c }}$ | -- |
| 2 | DMA | 85 | $>20: 1$ |
| 3 | $\mathrm{CH}_{3} \mathrm{CN}$ | N.R. | -- |
| 4 | THF | trace | -- |
| 5 | DMF | 76 | $>20: 1$ |
| 6 | DMSO | 91 | $>20: 1$ |
| 7 | N-Methyl-2-pyrrolidone | 83 | $>20: 1$ |
| 8 | Cyclohexanol | N.R. | -- |

[^0]Table S4. Control experiments under standard reaction condition.

|  <br> $1 a$ |  |  |  |
| :---: | :---: | :---: | :---: |
| entry ${ }^{\text {a }}$ | change from the standard condition | yield ${ }^{\text {b }}$ (\%) | d.r. |
| 1 | standard condition | 91 | $>20: 1$ |
| 2 | without $h v$ | N.R. ${ }^{\text {c }}$ | -- |
| 3 | without DIPEA | N.R. | -- |
| 4 | air condition | 42 | $>20: 1$ |

${ }^{\text {a }}$ Reaction conditions: A mixture of $1 \mathrm{a}(0.15 \mathrm{mmol})$, $2 \mathrm{a}(0.10 \mathrm{mmol})$, DIPEA $(0.10 \mathrm{mmol})$ in DMSO ( 1.0 mL ) was irradiated with 12 W blue LEDs at room temperature under $\mathrm{N}_{2}$ for 24 h ; ${ }^{\mathrm{b}}$ Isolated yields; ${ }^{\mathrm{c}}$ N.R. $=$ No reaction.

## 4. Scale-up reaction:



To a 100 mL round bottom flask with magnetic stirrer, glycosyl NHP ester 1a (1.09 g, 3 mmol ), thiophenol 2a (378.1 mg, 2 mmol ), $i \operatorname{Pr}_{2} \mathrm{NEt}(348 \mathrm{ml}, 2 \mathrm{mmol})$, and DMSO ( 30 ml ) were added. Then, the bottle was evacuated and back-filled with $\mathrm{N}_{2}$ (three times). The solution was then stirred at room temperature under the irradiation of Blue LEDs for 24 h using electronic fan to cool the flask. (Figure S2). After completion of the reaction, 50 mL water was added and extracted by ethyl acetate $(3 \times 50 \mathrm{~mL})$. The combined organic layer was washed with
brine $(100 \mathrm{~mL})$ and then dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated in vacuum. The desired products were obtained in the corresponding yields after purification by flash chromatography on silica gel eluting with hexane/ethyl acetate to afford the desired product $\mathbf{3 a}$ in $90 \%$ yield. It should be noted that two Kessil PR160L-456 nm lamps were used in the gram-scale reaction to provide the blue light source.


Figure S2 The photoreactor of gram-scale reaction

## 5.The mechanistic studies

### 5.1 Radical trapping experiment



To an oven-dried 10 mL glass tube equipped with a stir bar, was added glycosyl NHP ester 1a ( 0.15 mmol ), thiophenol 2a ( 0.1 mmol ), $i \operatorname{Pr}_{2} \mathrm{NEt}(0.1 \mathrm{mmol}), \mathrm{TEMPO}(0.45 \mathrm{mmol})$. The tube was evacuated and back-filled with $\mathrm{N}_{2}$ (three times), then sealed with rubber stopper and
parafilm. Then, anhydrous DMSO ( 1 mL ) was added using a syringe. The solution was then stirred at room temperature under the irradiation of 12 W Blue LEDs for 24 h using electronic fan to cool the tube. After 24 h , no corresponding product 3a was formed by TLC analysis, suggesting that the in-situ formed glycosyl radical might act as a key intermediate during this transformation.


To an oven-dried 10 mL glass tube equipped with a stir bar, was added glycosyl NHP ester 1a ( 0.15 mmol ), thiophenol 2a ( 0.1 mmol ), $i \mathrm{Pr}_{2} \mathrm{NEt}$ ( 0.1 mmol ) , 1,1-diphenylethylene ( 0.45 mmol ),. The tube was evacuated and back-filled with $\mathrm{N}_{2}$ (three times), then sealed with rubber stopper and parafilm. Then, anhydrous DMSO ( 1 mL ) was added using a syringe. The solution was then stirred at room temperature under the irradiation of 12 W Blue LEDs for 24 h using electronic fan to cool the tube. After completion of the reaction, corresponding product 3a was trace and the heck-type product 4 was isolated in $67 \%$ yield, indicating that this photo-induced thioglycosidation protocol may proceed through a radical-based mechanism. ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ ( $\mathbf{4 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 7.47$ - 7.23 (m, $10 \mathrm{H}), 6.09(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~s}, 1 \mathrm{H}), 4.79-4.68(\mathrm{~m}, 3 \mathrm{H}), 3.40(\mathrm{~s}, 3 \mathrm{H}), 1.42(\mathrm{~s}, 3 \mathrm{H})$, 1.31 (s, 3H).


### 5.2 UV-vis absorption spectra

The UV-vis absorption spectra of glycosyl NHP ester (0.15 M), thiophenol $(0.1 \mathrm{M}), i \operatorname{Pr}_{2} \mathrm{NEt}(0.1 \mathrm{M})$ in DMSO were recored in 1 cm path quartz cuvettes by using a GENESYS UV-Visible spectrophotometer (thermo scientific).


## 5.3 ${ }^{1} \mathbf{H}$ NMR spectra of 1 a and 2 a with or without DIPEA

The ${ }^{1} \mathrm{H}$ NMR spectra ( 400 MHz ) of $\mathbf{1 a}\left(0.06 \mathrm{mmol}\right.$ in $0.8 \mathrm{ml} d_{6}$-DMSO)





The ${ }^{1} \mathrm{H}$ NMR spectra $(400 \mathrm{MHz})$ of $\mathbf{1 a}\left(0.06 \mathrm{mmol}\right.$ in $0.8 \mathrm{ml} d_{6}$-DMSO)

+ DIPEA (0.04 mmol)


The ${ }^{1} \mathrm{H}$ NMR spectra $(400 \mathrm{MHz})$ of $\mathbf{2 a}\left(0.04 \mathrm{mmol}\right.$ in $0.8 \mathrm{ml} d_{6}$-DMSO)


The ${ }^{1} \mathrm{H}$ NMR spectra $(400 \mathrm{MHz})$ of $\mathbf{2 a}\left(0.04 \mathrm{mmol}\right.$ in $0.8 \mathrm{ml} d_{6}$-DMSO) + DIPEA ( 0.04 mmol )


The ${ }^{1} \mathrm{H}$ NMR spectra $(400 \mathrm{MHz})$ of $\mathbf{1 a}\left(0.06 \mathrm{mmol}\right.$ in $0.8 \mathrm{ml} d_{6}$-DMSO)
$+\mathbf{2 a}(0.04 \mathrm{mmol})+$ DIPEA $(0.04 \mathrm{mmol})$


## 6. Configuration determination

### 6.1 Plausible conformation of glycosyl radical


Ribose

Ribose $\mathrm{R}=\mathrm{Me}, \mathrm{Bn}$

Arabinose

Xylose

Glucose $\mathrm{R}=\mathrm{Me}, \mathrm{Bn}$

Galactose


The stereochemical outcome of glycosyl radical could be explained by a consequence of a combination of stereoelectronic and steric factors.

### 6.2 H-H coupling constant analysis


(3aS,4R,6R,6aR)-4-((4-bromophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3a): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(\mathbf{4 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 7.44$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.35 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), \mathbf{5 . 5 7}(\mathbf{s}, \mathbf{1 H}), \mathbf{5 . 1 1}(\mathrm{s}, \mathbf{1 H}), \mathbf{4 . 8 9}(\mathbf{d}, \boldsymbol{J}=\mathbf{5 . 7} \mathbf{H z}, \mathbf{1 H}), \mathbf{4 . 7 1}(\mathbf{d}, \boldsymbol{J}=\mathbf{5 . 8} \mathbf{H z}, \mathbf{1 H})$, $3.38(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H})$.

For furanose, the hydrogens on the same side of the sugar are coupled.
It is easy to assign different hydrogens of the sugar ring based on the chemical shift value: $5.57\left(\mathrm{~s}, \mathbf{H}^{1}\right.$ or $\left.\mathbf{H}^{4}\right), 5.11\left(\mathrm{~s}, \mathbf{H}^{1}\right.$ or $\left.\mathbf{H}^{4}\right), 4.89(\mathrm{~d}, J=5.7 \mathrm{~Hz}, \mathbf{1 H})$, $4.71(\mathrm{~d}, \boldsymbol{J}=5.8 \mathrm{~Hz}, \mathbf{1 H})$. Therefore, we can conclude that $\mathrm{H}^{1}$ and $\mathrm{H}^{4}$ are on the same side of the sugar ring based on the $\mathrm{H}-\mathrm{H}$ coupling constants.


(3aR,5R,6S,6aS)-6-(benzyloxy)-5-((4-bromophenyl)thio)-2,2-
dimethyltetrahydrofuro[2,3-d][1,3]dioxole (3ad): ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 400 MHz , Chloroform-d) $\delta$ $7.44-7.28(\mathrm{~m}, 9 \mathrm{H}), \mathbf{6 . 0 1}\left(\mathrm{d}, \boldsymbol{J}=\mathbf{3 . 8} \mathbf{~ H z}, \mathbf{H}^{\mathbf{1}}\right), \mathbf{5 . 4 8}\left(\mathrm{s}, \mathbf{H}^{4}\right), \mathbf{4 . 6 7}\left(\mathbf{d}, \boldsymbol{J}=\mathbf{3 . 8} \mathbf{~ H z}, \mathbf{H}^{2}\right), 4.60(\mathrm{~s}$, $2 \mathrm{H}), 4.23\left(\mathrm{~s}, \mathbf{H}^{3}\right), 1.67(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H})$.
(3aR,5S,6S,6aS)-6-(benzyloxy)-5-((4-bromophenyl)thio)-2,2-dimethyltetrahydrofuro[2,3d] [1,3]dioxole (3ad'): ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ ( $\mathbf{4 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta 7.45-7.27(\mathrm{~m}, 9 \mathrm{H}), \mathbf{6 . 0 8}$ (d, J $\left.=\mathbf{3 . 8} \mathbf{H z}, \mathbf{H}^{1}\right), \mathbf{5 . 5 5}\left(\mathbf{d}, \boldsymbol{J}=\mathbf{3 . 4} \mathbf{H z}, \mathbf{H}^{4}\right), 4.79-4.66(\mathrm{~m}, 2 \mathrm{H}), \mathbf{4 . 6 4}\left(\mathrm{d}, \boldsymbol{J}=\mathbf{3 . 8} \mathbf{H z}, \mathbf{H}^{\mathbf{2}}\right), 4.17(\mathrm{~d}$, $\left.\boldsymbol{J}=\mathbf{3 . 4} \mathbf{H z}, \mathbf{H}^{3}\right), 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H})$.

By comparing the $1 \mathrm{H}-\mathrm{NMR}$ data, we easily distinguish the configurations of the two isomers. It is the $\beta$-isomer when $\mathrm{H}^{3}$ and $\mathrm{H}^{4}$ are all singlet or the $\alpha$-isomer when $\mathrm{H}^{3}$ and $\mathrm{H}^{4}$ are coupled to each other.

There is one compound ${ }^{[8]}$ that has been reported that is very similar to our product 3ac. There is almost no difference between the chemical shift value and coupling constant of the hydrogen in the sugar ring between these two compounds. And they are all beta-isomer which proves our peak assignments and coupling constant analysis are correct.




(2R,3S,4R,5R,6S)-2-((4-bromophenyl)thio)-3,4,5,6-tetramethoxytetrahydro-2H-pyran (3ah): ${ }^{1} \mathbf{H}-N M R(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.56-7.34(\mathrm{~m}, 4 \mathrm{H}), \mathbf{4 . 7 9}$ (d, J=3.5 Hz, H ), 4.72 (d, $\boldsymbol{J}=\mathbf{1 0 . 0} \mathbf{H z}, \mathbf{H}^{\mathbf{5}}$ ), $3.60(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H}), 3.50(\mathrm{~s}, 3 \mathrm{H}, 1 \mathrm{H}), 3.27(\mathrm{~s}, 3 \mathrm{H}), \mathbf{3 . 1 8}$ (dd, J $=9.7,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.02-2.94(\mathrm{~m}, 1 \mathrm{H})$.
(2S,3S,4R,5R,6S)-2-((4-bromophenyl)thio)-3,4,5,6-tetramethoxytetrahydro-2H-pyran
 $\boldsymbol{J}=\mathbf{3 . 5} \mathbf{H z}, \mathbf{H}^{\mathbf{1}}$ ), $\mathbf{3 . 7 0}$ (t, $\boldsymbol{J}=\mathbf{8 . 5} \mathbf{H z}, \mathbf{1 H}$ ), $3.62(\mathrm{~s}, 3 \mathrm{H}), 3.53(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~s}, 3 \mathrm{H}), \mathbf{3 . 4 7}$ (dd, $\boldsymbol{J}$ $=\mathbf{8 . 5}, \mathbf{5 . 2} \mathrm{Hz}, \mathbf{1 H}), 3.40(\mathrm{~s}, 3 \mathrm{H}), \mathbf{3 . 2 4}(\mathbf{d d}, \boldsymbol{J}=\mathbf{8 . 5}, \mathbf{3 . 5} \mathbf{~ H z}, \mathbf{1 H})$.

We identified $\mathrm{H}^{1}$ and $\mathrm{H}^{5}$ through the chemical shift because they had the largest chemical shift among the hydrogens in the sugar ring at first. When $\mathrm{H}^{4}$ and $\mathrm{H}^{5}$ are on the same side, the chemical shift of $\mathrm{H}^{5}$ will be redshifted obviously according to the previous work about thioglycoside. ${ }^{[9]}$ What's more, it is when the dihedral angle of $\mathrm{H}^{4}$ and $\mathrm{H}^{5}$ is about 180 degrees that the coupling constant of these two hydrogens
could be 10.0 Hz . So we concluded that 3ah was $\alpha$-isomer.

(3aS,5S,5aR,8aR,8bR)-5-((4-bromophenyl)thio)-2,2,7,7-tetramethyltetrahydro-5H-
bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran (3af): ${ }^{\mathbf{1}} \mathbf{H}-N M R(400 ~ M H z, ~ C h l o r o f o r m-d) ~ \delta 7.49-$ $7.38(\mathrm{~m}, 4 \mathrm{H}), \mathbf{5 . 3 1}\left(\mathrm{d}, \boldsymbol{J}=2.4 \mathrm{~Hz}, \mathrm{H}^{\mathbf{1}}\right), 4.54\left(\mathrm{~d}, \boldsymbol{J}=\mathbf{5 . 4} \mathrm{Hz}, \mathbf{H}^{\mathbf{5}}\right), 4.43(\mathrm{~d}, \boldsymbol{J}=\mathbf{9 . 7} \mathrm{Hz}, \mathbf{1 H})$, 4.24 (d, $\boldsymbol{J}=\mathbf{1 . 7} \mathbf{H z}, \mathbf{1 H}), 4.00$ (dd, $\boldsymbol{J}=\mathbf{9 . 6}, \mathbf{5 . 4} \mathbf{H z}, \mathbf{1 H}$ ), 1.52 (s, 3H), 1.47 (s, 3H), 1.38 (s, $3 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H})$.
(3aS,5R,5aR,8aR,8bR)-5-((4-bromophenyl)thio)-2,2,7,7-tetramethyltetrahydro-5H-
bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran (3af'): ${ }^{1} \mathbf{H}-N M R(400 ~ M H z, ~ C h l o r o f o r m-d) ~ \delta ~ 7.40(s, ~$ 4H), $5.68\left(\mathrm{~d}, \boldsymbol{J}=5.0 \mathrm{~Hz}, \mathrm{H}^{5}\right), 5.20\left(\mathrm{~d}, \boldsymbol{J}=1.8 \mathrm{~Hz}, \mathrm{H}^{1}\right), 4.67\left(\mathrm{dd}, \boldsymbol{J}=7.6,2.7 \mathrm{~Hz}, \mathrm{H}^{3}\right), 4.48$ (dd, $\left.\boldsymbol{J}=\mathbf{7 . 6}, 1.9 \mathrm{~Hz}, \mathbf{H}^{2}\right), 4.38\left(\mathbf{d d}, \boldsymbol{J}=\mathbf{5 . 0}, \mathbf{2 . 7} \mathbf{~ H z}, \mathbf{H}^{4}\right), 1.55(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.39(\mathrm{~s}$, $3 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H})$.

We determined the two isomers by judging the redshift which was similar to the compounds $3 \mathrm{ah} / 3 \mathrm{ah}$ '. In this case, the coupling constants of $H^{5}$ of two isomers are basically the same because the $H^{4}$ is on the equatorial bond (e bond). The coupling constants of the hydrogens of the sugar ring of compound 3 af' are clear and easy to assign. $5.68(\mathbf{d}, \boldsymbol{J}=\mathbf{5 . 0} \mathrm{Hz}$, $\left.H^{5}\right), 5.20\left(\mathrm{~d}, J=1.8 \mathrm{~Hz}, \mathrm{H}^{1}\right), 4.67\left(\mathrm{dd}, J=7.6,2.7 \mathrm{~Hz}, \mathrm{H}^{3}\right), 4.48\left(\mathrm{dd}, J=7.6,1.9 \mathrm{~Hz}, \mathrm{H}^{2}\right)$, $4.38\left(\mathrm{dd}, J=5.0,2.7 \mathrm{~Hz}, \mathrm{H}^{4}\right)$

### 6.3 NOE experiments










## 7. References

1. Ji, P.; Zhang, Y.; Wei, Y.; Huang, H.; Hu, W.; Mariano, P. A.; Wang, W., Visible-Light-Mediated, Chemo- and Stereoselective Radical Process for the Synthesis of C-Glycoamino Acids. Org Lett 2019, 21 (9), 3086-3092.
2. Masuda, K.; Nagatomo, M.; Inoue, M., Direct assembly of multiply oxygenated carbon chains by decarbonylative radical-radical coupling reactions. Nat Chem 2017, 9 (3), 207212.
3. Agnihotri, G.; Misra, A. K., Synthesis of a di- and a trisaccharide related to the O-antigen of Escherichia coli O83:K24:H31. Carbohydr Res 2006, 341 (14), 2420-5.
4. Qi, R.; ang, C.; Ma, Z.; Wang, H.; Chen, Q.; Liu, L.; Pan, D.; Ren, X.; Wang, R.; Xu, Z., Visible-Light-Promoted Stereoselective C(sp(3) )-H Glycosylation for the Synthesis of C-Glycoamino Acids and C-Glycopeptides. Angew Chem Int Ed Engl 2022, 61 (24), e202200822.
5. Takashima, K.; Sakano, M.; Kinouchi, E.; Nakamura, S.; Marumoto, S.; Ishikawa, F.; Ninomiya, K.; Nakanishi, I.; Morikawa, T.; Tanabe, G., Elongation of the side chain by linear alkyl groups increases the potency of salacinol, a potent alpha-glucosidase inhibitor from the Ayurvedic traditional medicine "Salacia," against human intestinal maltase. Bioorg Med Chem Lett 2021, 33, 127751.
6. Yamada, K.; Hayakawa, H.; Sakata, S.; Ashida, N.; Yoshimura, Y., Synthesis and antiviral evaluation of alpha-D-2',3'-didehydro-2',3'-dideoxy-3'-C-hydroxymethyl nucleosides. Bioorg Med Chem Lett 2010, 20 (20), 6013-6.
7. Annika, G., Martin, Hiersemann., (-)-Lytophilippine A: Synthesis of a C1-C18 Building Block. Org Lett 2010, 12 (22), 5258-5261.
8. Li, J.; Y, X.; Wang, S.; Zhang, L.; Zhou, X.; Wang, S.; Ji, S., Visible-LightPromoted Cross-Coupling Reactions of 4-Alkyl-1,4- dihydropyridines with Thiosulfonate or Selenium Sulfonate: A Unified Approach to Sulfides, Selenides, and Sulfoxides. Org Lett 2020, 22 (12), 4908-4913.
9. Feng, G.; Luo, T.; Guo, Y.; Liu, C.; Dong, H., Concise Synthesis of 1-Thioalkyl Glycoside Donors by Reaction of Per-O-acetylated Sugars with Sodium Alkanethiolates under Solvent-Free Conditions. J. Org. Chem. 2022, 87 (5), 3638-3646.

## 8. Characterization of products


(3aS,4R,6R,6aR)-4-((4-bromophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3a): White solid; 32.8 mg ; $91 \%$ yield; $>20: 1$ d.r.; M.p. $79-80{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.44(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.57(\mathrm{~s}, 1 \mathrm{H})$, $5.11(\mathrm{~s}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.32$ ( $\mathrm{s}, 3 \mathrm{H}$ ). ${ }^{13} \mathbf{C}-\mathrm{NMR}(100 \mathrm{MHz}$, Chloroform-d) $\delta 134.6,132.2,132.1,121.2,113.1,110.5$, 93.1, 85.7, 84.6, 55.3, 26.5, 25.1. HRMS (ESI) m/z [M + Na ${ }^{+}$calculated for 382.9923 , found 382.9925.

(3aS,4R,6R,6aR)-4-((4-fluorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3b): White solid; 25.3 mg ; $84 \%$ yield; $>20: 1$ d.r.; M.p. $56-57{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$
( $\mathbf{4 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 7.49$ (ddt, $J=8.3,5.3,2.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.06-7.00$ (m, 2H), 5.50 (s, $1 \mathrm{H}), 5.11(\mathrm{~s}, 1 \mathrm{H}), 4.91(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H})$, 1.32 (s, 3H). ${ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform- $\boldsymbol{d}$ ) $\delta 162.4$ (d, $J=248.4 \mathrm{~Hz}$ ), 133.7 (d, $J=$ $8.1 \mathrm{~Hz}), 130.1$ (d, $J=3.0 \mathrm{~Hz}), 116.3$ (d, $J=22.2 \mathrm{~Hz}$ ), 113.0, 110.0, 94.1, 85.8, 84.7, 55.3, 26.4, 25.1. ${ }^{19}$ F-NMR ( $\mathbf{3 7 6} \mathbf{~ M H z}$, Chloroform- $\boldsymbol{d}$ ) $\delta-114.11$ ( $\mathrm{tt}, J=8.6,5.3 \mathrm{~Hz}$ ). HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 323.0724, found 323.0723.

(3aS,4R,6R,6aR)-4-((4-chlorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3c): White solid; 27.2 mg ; $85 \%$ yield; $>20: 1$ d.r.; M.p. $64-65^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ HMR ( 400 MHz , Chloroform-d) $\delta 7.44-7.39$ (m, 2H), $7.31-7.27$ (m, 2H), 5.56 (s, 1H), 5.11 (s, $1 \mathrm{H}), 4.90$ (d, $J=6.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.71 (d, $J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.39$ (s, 3H), 1.46 ( $\mathrm{s}, 3 \mathrm{H}), 1.32$ (s, 3H). ${ }^{13}$ C-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 133.9,133.3,132.0,129.3,113.1,110.5,93.3,85.8$, 84.6, 55.3, 26.5, 25.1. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 339.0428, found 339.0431.

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-((4-
(trifluoromethyl)phenyl)thio)tetrahydrofuro[3,4-d][1,3]dioxole (3d): White solid; 34.1 mg ; 97\% yield; >20:1 d.r.; M.p.81-82 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}-$ NMR ( 400 MHz , Chloroform-d) $\delta 7.55$ (s, 4H), $5.70(\mathrm{~s}, 1 \mathrm{H}), 5.15(\mathrm{~s}, 1 \mathrm{H}), 4.91(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.36(\mathrm{~s}, 3 \mathrm{H}), 1.49$ $(\mathrm{s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}){ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 141.0$ (d, $J=1.0 \mathrm{~Hz}$ ), 129.0, 128.5 (q, $J=32.0 \mathrm{~Hz}$ ), 125.9 (q, $J=4.0 \mathrm{~Hz}$ ), 124.1 (q, $J=270.0 \mathrm{~Hz}$ ), 113.25, 110.46, 92.07, 85.70, 84.56, 55.22, 26.45, 25.07. ${ }^{19}$ F-NMR ( $\mathbf{3 7 6} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta-62.55$. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 373.0692, found 373.0691.


## 4-(((3aS,4R,6R,6aR)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-

yl)thio)benzonitrile (3e): White solid; $22.6 \mathrm{mg} ; 73 \%$ yield; $>20: 1$ d.r.; M.p.61- $62^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-$ NMR ( 400 MHz , Chloroform-d) $\delta 7.59$ - 7.54 (m, 2H), 7.52 - 7.48 (m, 2H), $5.72(\mathrm{~s}, 1 \mathrm{H})$, $5.15(\mathrm{~s}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.33$ (s, 3H). ${ }^{13}$ C-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta 143.4,132.5,128.3,118.7$, 113.4, 110.5, 109.4, 91.4, 85.5, 84.5, 55.2, 26.4, 25.0. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 330.0770, found 330.0773 .

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-(p-tolylthio)tetrahydrofuro[3,4-
d][1,3]dioxole (3f): White solid; 14.0 mg ; $47 \%$ yield; $>20: 1$ d.r.; M.p. $75-76{ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.40(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.13$ (d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 5.54(\mathrm{~s}, 1 \mathrm{H})$, $5.10(\mathrm{~s}, 1 \mathrm{H}), 4.92(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H}), 1.45$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.32(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathrm{NMR}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 137.4,131.6,131.4,129.9$, $112.9,110.4,93.8,85.9,84.7,55.2,26.5,25.2,21.1 . \operatorname{HRMS}(E S I) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 319.0974, found 319.0977.

(3aR,4R,6R,6aS)-4-methoxy-6-((4-methoxyphenyl)thio)-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3g): White solid; 24.5 mg ; 78\% yield; $>20: 1$ d.r.; M.p. $85-86{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.46(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.44(\mathrm{~s}, 1 \mathrm{H})$, $5.09(\mathrm{~s}, 1 \mathrm{H}), 4.91(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.44(\mathrm{~s}, 3 \mathrm{H}), 1.44$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.31 ( $\mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathrm{NMR}(100 \mathrm{MHz}$, Chloroform-d) $\delta 159.6,134.4,125.2,114.8$, 112.9, 110.5, 94.6, 85.8, 84.7, 55.3, 55.3, 26.5, 25.2. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 335.0923 , found 335.0921 .

(3aS,4R,6R,6aR)-4-((4-(tert-butyl)phenyl)thio)-6-methoxy-2,2-
dimethyltetrahydrofuro[3,4-d][1,3]dioxole (3h): Pale oil; $26.3 \mathrm{mg} ; 77 \%$ yield; $>20: 1$ d.r.. ${ }^{1} \mathbf{H}-N M R(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.46-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 2 \mathrm{H}), 5.56(\mathrm{~s}, 1 \mathrm{H})$, $5.11(\mathrm{~s}, 1 \mathrm{H}), 4.93(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.32$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.31(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 150.6,131.5,131.2,126.2$, $112.9,110.4,93.7,86.0,84.7,55.2,34.6,31.3,26.5,25.2 . \operatorname{HRMS}(E S I) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$ calculated for 361.1444, found 361.1442.

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-(phenylthio)tetrahydrofuro[3,4-
d][1,3]dioxole (3i): Pale oil; $17.9 \mathrm{mg} ; 63 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.52-7.48(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.28-7.26(\mathrm{~m}, 1 \mathrm{H}), 5.63(\mathrm{~s}, 1 \mathrm{H})$, $5.13(\mathrm{~s}, 1 \mathrm{H}), 4.94(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}), 1.33$ ( $\mathrm{s}, 3 \mathrm{H}$ ). ${ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta 135.3,130.6,129.1,127.1,113.0,110.4$, 93.3, 85.9, 84.7, 55.2, 26.7, 25.1. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 305.0818 , found 305.0820 .

(3aS,4R,6R,6aR)-4-((2-fluorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3j): Pale oil; 25.3 mg ; 84\% yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathrm{H}-\mathrm{NMR}$ ( 400 MHz , Chloroform-d) $\delta 7.54(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.23(\mathrm{~m}, 1 \mathrm{H}), 7.10(\mathrm{q}, J=8.8,8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $5.61(\mathrm{~s}, 1 \mathrm{H}), 5.09(\mathrm{~s}, 1 \mathrm{H}), 4.96(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{~s}, 3 \mathrm{H}), 1.46$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.32 (s, 3H). ${ }^{\mathbf{1 3}} \mathbf{C - N M R}(100 \mathrm{MHz}$, Chloroform-d) $\delta 160.5(\mathrm{~d}, J=245.0 \mathrm{~Hz}$ ), 132.6 $(\mathrm{d}, J=1.0 \mathrm{~Hz}), 128.5(\mathrm{~d}, J=8.0 \mathrm{~Hz}), 123.7(\mathrm{~d}, J=4.0 \mathrm{~Hz}), 120.8(\mathrm{~d}, J=18 \mathrm{~Hz}), 114.9(\mathrm{~d}, J=$ 22 Hz ), 112.0, 109.6, $91.3(\mathrm{~d}, J=2 \mathrm{~Hz}), 85.1,83.6,54.3,25.4,24.1 .{ }^{19}$ F-NMR (376 MHz, Chloroform-d) $\delta-108.10$ (ddd, $J=9.3,7.4,5.2 \mathrm{~Hz}$ ). HRMS (ESI) m/z $[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 323.0724, found 323.0723.

(3aS,4R,6R,6aR)-4-((2-chlorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3k): Pale oil; $25.6 \mathrm{mg} ; 81 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathrm{H}-\mathrm{NMR}$ ( $\mathbf{4 0 0} \mathrm{MHz}$, Chloroform-d) $\delta 7.60(\mathrm{dd}, J=7.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{dd}, J=7.9,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{td}, J=$ $7.6,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{td}, J=7.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~s}, 1 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=5.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.74(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 134.9,134.3,130.6,129.8,127.7,127.4,113.1,110.5,91.6,85.9,84.6,55.2$, 26.5, 25.1. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 339.0428, found 339.0427.

(3aS,4R,6R,6aR)-4-((2-bromophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (31): Pale oil; $28.1 \mathrm{mg} ; 77 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathrm{H}-\mathrm{NMR}$ ( $\mathbf{6 0 0} \mathrm{MHz}$, Chloroform-d) $\delta 7.59$ (dd, $J=17.7,7.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.30(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{t}, J=7.6 \mathrm{~Hz}$, $1 \mathrm{H}), 5.68(\mathrm{~s}, 1 \mathrm{H}), 5.14(\mathrm{~s}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~d}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H})$, 1.49 (s, 3H), 1.34 (s, 3H). ${ }^{13} \mathbf{C}-$ NMR ( $\left.150 ~ M H z, ~ C h l o r o f o r m-d\right) ~ \delta 137.1, ~ 133.1, ~ 130.1, ~ 128.0, ~$ $127.6,124.4,113.1,110.4,91.8,85.7,84.6,55.2,26.4,25.1$. HRMS (ESI) m/z [M + Na] ${ }^{+}$ calculated for 382.9923 , found 382.9921 .

(3aR,4R,6R,6aS)-4-methoxy-6-((2-methoxyphenyl)thio)-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3m): Pale oil; $19.4 \mathrm{mg} ; 62 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.50(\mathrm{dd}, J=7.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.29-7.22(\mathrm{~m}, 1 \mathrm{H}), 6.93(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H})$, $6.88(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.69(\mathrm{~s}, 1 \mathrm{H}), 5.08(\mathrm{~s}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~d}, J=5.8$ $\mathrm{Hz}, 1 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.32$ ( $\mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$-NMR ( 100 MHz , Chloroform-d) $\delta 156.9,131.3,127.7,121.8,120.2,111.8,109.8,109.6,90.0,85.1,83.7,54.8$, 54.3, 25.4, 24.1. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 335.0923, found225.0926.

(3aS,4R,6R,6aR)-4-((3-fluorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3n): Pale oil; $26.3 \mathrm{mg} ; 87 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.30-7.21(\mathrm{~m}, 3 \mathrm{H}), 6.98-6.91(\mathrm{~m}, 1 \mathrm{H}), 5.64(\mathrm{~s}, 1 \mathrm{H}), 5.14(\mathrm{~s}, 1 \mathrm{H}), 4.91(\mathrm{~d}$, $J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}$ $(100 \mathrm{MHz}$, Chloroform-d) $\delta 162.8(\mathrm{~d}, J=247.0 \mathrm{~Hz}), 137.8(\mathrm{~d}, J=8.0 \mathrm{~Hz}), 130.0(\mathrm{~d}, J=8.0$ $\mathrm{Hz}), 125.6(\mathrm{~d}, J=3 \mathrm{~Hz}), 116.9(\mathrm{~d}, J=23 \mathrm{~Hz}), 113.9(\mathrm{~d}, J=21 \mathrm{~Hz}), 113.15,110.44,92.81$, 85.76, 84.60, 55.22, 26.46, 25.11. ${ }^{\mathbf{1 9}} \mathbf{F}$-NMR ( $\mathbf{3 7 6} \mathbf{~ M H z}$, Chloroform-d) $\delta-111.78--111.88$ (m). HRMS (ESI) m/z [M+Na] calculated for 323.0724, found 323.0729.

(3aS,4R,6R,6aR)-4-((3-chlorophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (30): White solid; $26.8 \mathrm{mg} ; 84 \%$ yield; $>20: 1$ d.r.; M.p.57-58 ${ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.49(\mathrm{t}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{dt}, J=7.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-$ $7.19(\mathrm{~m}, 2 \mathrm{H}), 5.62(\mathrm{~s}, 1 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 4.90(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H})$, $3.39(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C - N M R}(100 \mathrm{MHz}$, Chloroform-d) $\delta 137.5,134.8$, 130.1, 130.0, 128.3, 127.1, 113.1, 110.5, 92.9, 85.7, 84.6, 55.2, 26.7, 25.1. HRMS (ESI) m/z $[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 339.0428, found 339.0427.

(3aS,4R,6R,6aR)-4-((3-bromophenyl)thio)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3p): White solid; $33.7 \mathrm{mg} ; 93 \%$ yield; $>20: 1$ d.r.; M.p. $58-59{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$
(400 MHz, Chloroform-d) $\delta 7.64(\mathrm{t}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.42-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.18(\mathrm{t}, J=7.9 \mathrm{~Hz}$, $1 \mathrm{H}), 5.61(\mathrm{~s}, 1 \mathrm{H}), 5.12(\mathrm{~s}, 1 \mathrm{H}), 4.90(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{~s}, 3 \mathrm{H})$, $1.48(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 137.8,132.9,130.4,130.0$, $128.8,122.9,113.1,110.5,93.0,85.7,84.6,55.3,26.5,25.1$. HRMS (ESI) m/z [M + Na] ${ }^{+}$ calculated for 382.9923 , found 382.9927 .

(3aR,4R,6R,6aS)-4-methoxy-6-((3-methoxyphenyl)thio)-2,2-dimethyltetrahydrofuro[3,4d][1,3]dioxole (3q): Pale oil; $27.7 \mathrm{mg} ; 88 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.23$ (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.10-7.03(\mathrm{~m}, 2 \mathrm{H}), 6.82-6.77(\mathrm{~m}, 1 \mathrm{H}), 5.63$ (s, $1 \mathrm{H}), 5.11(\mathrm{~s}, 1 \mathrm{H}), 4.92(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.40(\mathrm{~s}, 3 \mathrm{H})$, 1.47 ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.32 ( $\mathrm{s}, 3 \mathrm{H}$ ). ${ }^{13} \mathbf{C}-$ NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 159.9,136.5,129.9,122.8$, $116.0,113.0,112.9,110.4,93.1,85.9,84.7,55.3,55.2,26.5,25.1$. HRMS (ESI) m/z [M + $\mathrm{Na}]^{+}$calculated for 335.0923, found 335.0924

(3aS,4R,6R,6aR)-4-((2,5-difluorophenyl)thio)-6-methoxy-2,2-
dimethyltetrahydrofuro[3,4-d][1,3]dioxole (3r): faint yellow oil; $26.8 \mathrm{mg} ; 84 \%$ yield; $>20: 1$ d.r.. ${ }^{1} \mathbf{H}-N M R(400 ~ M H z, ~ C h l o r o f o r m-d) ~ \delta 7.31-7.26(m, 1 H), 7.04(t d, J=8.8,4.6$ $\mathrm{Hz}, 1 \mathrm{H}), 6.97-6.90(\mathrm{~m}, 1 \mathrm{H}), 5.62(\mathrm{~s}, 1 \mathrm{H}), 5.11(\mathrm{~s}, 1 \mathrm{H}), 4.94(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=$ $5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ 159.1 (dd, $J=139.4 \mathrm{~Hz}, J=2.0 \mathrm{~Hz}$ ), 156.7 (dd, $J=133.3 \mathrm{~Hz}, J=3.0 \mathrm{~Hz}$ ), $124.0(\mathrm{dd}, J=20.2$ $\mathrm{Hz}, J=8.1 \mathrm{~Hz}), 118.8(\mathrm{dd}, J=25.2 \mathrm{~Hz}, J=2.0 \mathrm{~Hz}), 116.5(\mathrm{dd}, J=26.3 \mathrm{~Hz}, J=9.1 \mathrm{~Hz})$, 115.3 (dd, $J=24.2 \mathrm{~Hz}, J=8.1 \mathrm{~Hz}), 113.2,110.7,91.9,85.9,84.5,55.3,26.4,25.1 .{ }^{19}$ F-NMR ( 376 MHz , Chloroform-d) $\delta-114.95--115.07$ (m), -117.78 (dtd, $J=16.2,7.9,4.7 \mathrm{~Hz}$ ). HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 341.0629, found 341.0630.

(3aS,4R,6R,6aR)-4-((3,5-dichlorophenyl)thio)-6-methoxy-2,2-
dimethyltetrahydrofuro[3,4-d][1,3]dioxole (3s): White solid; $31.3 \mathrm{mg} ; 89 \%$ yield; $>20: 1$ d.r.; M.p. $46-47{ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.35$ (d, $J=1.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.22(\mathrm{t}, J$ $=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.62(\mathrm{~s}, 1 \mathrm{H}), 5.14(\mathrm{~s}, 1 \mathrm{H}), 4.88(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H})$, $3.37(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}-\mathbf{N M R}(\mathbf{1 0 0} \mathbf{~ M H z}, \mathbf{C h l o r o f o r m}-\boldsymbol{d}) \delta 139.2,135.3$,
$127.6,126.9,113.3,110.5,92.5,85.6,84.5,55.3,26.4,25.1 . \operatorname{HRMS}(E S I) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$ calculated for 373.0038 , found 373.0036 .

dimethyltetrahydrofuro[3,4-d][1,3]dioxole (3t): faint yellow oil; $33.0 \mathrm{mg} ; 94 \%$ yield; $>20: 1$ d.r. ${ }^{1} \mathbf{H}-N M R(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.54(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.41(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.23(\mathrm{dd}, J=8.5,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.63(\mathrm{~s}, 1 \mathrm{H}), 5.12(\mathrm{~s}, 1 \mathrm{H}), 4.94(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=$ $5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.35(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ $135.0,133.6,133.0,131.3,129.6,127.7,113.2,110.5,91.6,85.7,84.5,55.3,26.4,25.1$. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 373.0038, found 373.0039.

(3aS,4R,6R,6aR)-4-((2,4-dimethylphenyl)thio)-6-methoxy-2,2-
dimethyltetrahydrofuro[3,4-d][1,3]dioxole (3u): Pale oil; $13.4 \mathrm{mg} ; 47 \%$ yield; $>20: 1$ d.r.. ${ }^{1} \mathbf{H}-N M R(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.44(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.03(\mathrm{~s}, 1 \mathrm{H}), 7.01-6.97(\mathrm{~m}$, $1 \mathrm{H}), 5.48(\mathrm{~s}, 1 \mathrm{H}), 5.10(\mathrm{~s}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H})$, $2.40(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ $138.9,137.3,131.6,131.3,131.0,127.5,112.9,110.4,93.1,86.2,84.7,55.3,26.5,25.2,21.0$, 20.7. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 333.1131, found 333.1132 .


4-(((3aS,4R,6R,6aR)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-
yl)thio)pyridine (3v): White solid; 23.7 mg ; 83\% yield; >20:1 d.r.; M.p. 88-89 ${ }^{\circ} \mathrm{C} . \mathbf{}^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ ( 400 MHz, Chloroform-d) $\delta 8.44$ (d, $J=5.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.29 (dd, $J=4.7,1.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.79 (s, $1 \mathrm{H}), 5.16(\mathrm{~s}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H})$, $1.34(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}$-NMR ( 100 MHz , Chloroform-d) $\delta 149.5,148.0,121.6,113.4,110.4,90.0$, 85.5, 84.5, 55.2, 26.4, 25.0. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 306.0770 , found 306.0773.

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-(thiophen-2-ylthio)tetrahydrofuro[3,4-
d][1,3]dioxole (3w): White solid; $18.7 \mathrm{mg} ; 65 \%$ yield; $>20: 1$ d.r.; M.p. $69-70{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.43-7.37(\mathrm{~m}, 1 \mathrm{H}), 7.23-7.19(\mathrm{~m}, 1 \mathrm{H}), 7.01(\mathrm{dd}, J=5.3,3.6$ $\mathrm{Hz}, 1 \mathrm{H}), 5.38(\mathrm{~s}, 1 \mathrm{H}), 5.10(\mathrm{~s}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.48(\mathrm{~s}$, $3 \mathrm{H}), 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathrm{NMR}(100 \mathbf{~ M H z}$, Chloroform-d) $\delta$ 133.5, 131.1, 129.3, $126.8,112.0,109.6,94.6,84.5,83.6,54.4,25.4,24.1$. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 311.0382, found 311.0381.
MeO,

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-((2-methylfuran-3-
yl)thio)tetrahydrofuro[3,4-d][1,3]dioxole (3x): Pale oil; 20.3 mg ; 71\% yield; >20:1 d.r.. ${ }^{\mathbf{1}} \mathbf{H}$ NMR (400 MHz, Chloroform-d) $\delta 7.29(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.42(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.24$ (s, $1 \mathrm{H}), 5.06(\mathrm{~s}, 1 \mathrm{H}), 4.87(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.45(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 3 \mathrm{H})$, 1.43 (s, 3H), 1.31 (s, 3H). ${ }^{\mathbf{1 3}} \mathbf{C - N M R}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 155.1,140.9,114.8,112.9$, $110.5,109.3,93.8,86.0,84.7,55.4,26.4,25.2,11.9$. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 309.0767, found 309.0769.

(3aR,4R,6R,6aS)-4-methoxy-2,2-dimethyl-6-(naphthalen-2-ylthio)tetrahydrofuro[3,4d][1,3]dioxole (3y): White solid; $26.1 \mathrm{mg} ; 78 \%$ yield; $>20: 1$ d.r.; M.p. $70-71{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ (400 MHz, Chloroform-d) $\delta 7.97$ (d, $J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 7.56$ (dd, $J=$ $8.6,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.52-7.44(\mathrm{~m}, 2 \mathrm{H}), 5.75(\mathrm{~s}, 1 \mathrm{H}), 5.16(\mathrm{~s}, 1 \mathrm{H}), 4.99(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.76$ $(\mathrm{d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 133.8,132.8,132.2,129.1,128.7,128.3,127.7,127.5,126.6,126.1,113.0$, $110.5,93.1,85.9,84.7,55.3,26.5,25.1$. HRMS (ESI) m/z [M + Na $]^{+}$calculated for 355.0974, found 355.0977 .


4-(((3aS,4R,6R,6aR)-6-methoxy-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-
yl)thio)phenyl 2-(4-isobutylphenyl)propanoate (3z): Pale oil; $26.3 \mathrm{mg} ; 54 \%$ yield; $>20: 1$ d.r. ${ }^{1} \mathbf{H}-N M R(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.46(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H})$, $7.14(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.96(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.54(\mathrm{~s}, 1 \mathrm{H}), 5.10(\mathrm{~s}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=5.7$ $\mathrm{Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{q}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{~s}, 3 \mathrm{H}), 2.47(\mathrm{~d}, J=7.1 \mathrm{~Hz}$, $2 \mathrm{H}), 1.86(\mathrm{dp}, J=13.3,6.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.60(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 4 \mathrm{H}), 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}), 0.92$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 0.90 ( $\mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathrm{NMR}(150 \mathrm{MHz}$, Chloroform-d) $\delta 173.0,150.2,140.9,137.0$,
$132.3,132.1,129.5,127.2,122.2,113.0,110.4,93.5,85.8,84.6,55.2,45.2,45.0,30.2,26.4$, 25.1, 22.4, 18.4. HRMS (ESI) m/z [M+Na] calculated for 509.1968, found 509.1970.

(2R,3S,4R,5R)-3,4-bis(benzyloxy)-2-((4-bromophenyl)thio)-5-methoxytetrahydrofuran (3aa): White solid; $43.5 \mathrm{mg} ; 86 \%$ yield; $>20: 1$ d.r.; M.p. $47-48{ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.43-7.32(\mathrm{~m}, 14 \mathrm{H}), 5.52(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{~s}, 1 \mathrm{H}), 4.65(\mathrm{~d}, J=1.2$ $\mathrm{Hz}, 2 \mathrm{H}), 4.63(\mathrm{~s}, 2 \mathrm{H}), 4.15(\mathrm{dd}, J=6.3,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.92-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-$ NMR (100 MHz, Chloroform-d) $\delta$ 137.5, 137.4, 133.9, 132.8, 132.0, 128.5, 128.05, 128.0, 127.9, 121.4, 107.4, 88.9, 81.8, 79.4, 72.9, 72.7, 55.4. HRMS (ESI) m/z [M+Na]+ calculated for 523.0549, found 523.0550.

(2R,3S,4R,5R)-2-((4-bromophenyl)thio)-3,4,5-trimethoxytetrahydrofuran (3ab): Pale oil; 31.7 mg ; 91\% yield; >20:1 d.r.. ${ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}$ ( $\mathbf{4 0 0} \mathbf{~ M H z}$, Chloroform- $\boldsymbol{d}$ ) $\delta 7.43-7.40(\mathrm{~m}, 2 \mathrm{H})$, $7.39-7.36(\mathrm{~m}, 2 \mathrm{H}), 5.40(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{~s}, 1 \mathrm{H}), 3.99(\mathrm{dd}, J=6.2,4.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.79(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.49(\mathrm{~s}, 6 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ 134.0, 132.6, 132.0, 121.4, 106.5, 88.5, 84.1, 81.6, 58.9, 58.6, 55.5. HRMS (ESI) m/z [M + $\mathrm{Na}]^{+}$calculated for 370.9923 , found 370.9925 .

(3aS,5S,6R,6aR)-6-(benzyloxy)-5-((4-bromophenyl)thio)-2,2-
dimethyltetrahydrofuro[2,3-d][1,3]dioxole (3ac): Pale oil; $34.2 \mathrm{mg} ; 78 \%$ yield; d.r. $=6: 1$. ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.46-7.28(\mathrm{~m}, 9 \mathrm{H}), 6.02(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.48(\mathrm{~s}$, $1 \mathrm{H}), 4.67(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.59(\mathrm{~s}, 2 \mathrm{H}), 4.23(\mathrm{~s}, 1 \mathrm{H}), 1.67(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}$ ( 100 MHz , Chloroform-d) $\delta 136.7,134.6,132.9,132.0,128.6,128.2,127.95,121.5,113.6$, 107.1, 91.4, 86.5, 83.4, 72.2, 26.4, 26.1. HRMS (ESI) m/z [M + Na] calculated for 459.0236, found 459.0238 .

(3aS,5R,6R,6aR)-6-(benzyloxy)-5-((4-bromophenyl)thio)-2,2-
dimethyltetrahydrofuro[2,3-d][1,3]dioxole (3ac'): Pale oil; $5.7 \mathrm{mg} ; 13 \%$ yield; d.r. $=6: 1$.
${ }^{1} \mathrm{H}$-NMR ( $\mathbf{4 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta 7.46-7.27$ (m, 9H), 6.08 (d, $J=3.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.55 (d, $J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.78-4.66(\mathrm{~m}, 2 \mathrm{H}), 4.64(\mathrm{~d}, J=3.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.45$ (s, 3H), 1.33 (s, 3H). ${ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta$ 136.9, 134.6, 132.2, 132.1, 128.5, 128.1, 127.9, 121.1, 112.6, 105.5, 89.8, 83.5, 82.0, 72.8, 27.1, 26.5. HRMS (ESI) m/z $[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 459.0236, found 459.0237.

(3aR,5R,6S,6aS)-6-(benzyloxy)-5-((4-bromophenyl)thio)-2,2-
dimethyltetrahydrofuro[2,3-d][1,3]dioxole (3ad): Pale oil; $34.3 \mathrm{mg} ; 78 \%$ yield; d.r. $=6: 1$. ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.44-7.28(\mathrm{~m}, 9 \mathrm{H}), 6.01(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.48$ (s, $1 \mathrm{H}), 4.67(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.60(\mathrm{~s}, 2 \mathrm{H}), 4.23(\mathrm{~s}, 1 \mathrm{H}), 1.67(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ 136.7, 134.6, 132.9, 132.0, 128.6 128.2, 127.94, 121.5, 113.6, 107.1, 91.4, 86.5, 83.4, 72.2, 26.4, 26.1. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 459.0236, found 459.0239.

(3aR,5S,6S,6aS)-6-(benzyloxy)-5-((4-bromopheny)thio)-2,2-dimethyltetrahydrofuro[2,3d][1,3]dioxole (3ad'): Pale oil; $5.8 \mathrm{mg} ; 13 \%$ yield; d.r. $=6: 1 .{ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.45-7.27(\mathrm{~m}, 9 \mathrm{H}), 6.08(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.55(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79$ $-4.66(\mathrm{~m}, 2 \mathrm{H}), 4.64(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13}$ C-NMR ( 100 MHz , Chloroform-d) $\delta$ 136.9, 134.6, 132.2, 132.1, 128.5, 128.1, 127.9, 121.1, 112.6, 105.5, 89.8, 83.5, 82.0, 72.8, 27.1, 26.5. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 459.0236 , found 459.0237 .

(3aS,5aS,8aS,8bS)-5a-((4-bromophenyl)thio)-2,2,7,7-tetramethyltetrahydro-5H-bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran (3ae): Pale oil; $28.7 \mathrm{mg} ; 68 \%$ yield; $>20: 1$ d.r.. ${ }^{\mathbf{1}} \mathbf{H}-$ NMR ( 400 MHz , Chloroform-d) $\delta 7.51$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.44 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 4.55 (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.33(\mathrm{~s}, 1 \mathrm{H}), 4.22(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H})$, 1.34 (s, 3H), 1.22 (s, 3H). ${ }^{13} \mathbf{C}-$ NMR ( $\mathbf{1 0 0} \mathbf{~ M H z , ~ C h l o r o f o r m - d ) ~} \delta$ 137.1, 130.7, 128.2, 122.7, 109.0, 108.3, 108.3, 74.2, 70.0, 68.8, 61.8, 25.3, 24.9, 23.4, 23.1. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$ calculated for 439.0185, found 439.0184.

(3aS,5S,5aR,8aR,8bR)-5-((4-bromophenyl)thio)-2,2,7,7-tetramethyltetrahydro-5H-
bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran (3af): Pale oil; $24.4 \mathrm{mg} ; 58 \%$ yield; d.r. $=3: 1 .{ }^{\mathbf{1}} \mathbf{H}$ NMR (400 MHz, Chloroform-d) $\delta 7.48-7.38(\mathrm{~m}, 4 \mathrm{H}), 5.31(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{~d}, J=$ $5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{~d}, J=9.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{dd}, J=9.6,5.4 \mathrm{~Hz}, 1 \mathrm{H})$, $1.52(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.38(\mathrm{~s}, 3 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta$ $134.8,132.0,131.2,122.5,111.4,109.3,97.2,81.5,75.5,74.4,71.5,28.0,27.8,25.8,25.4$. HRMS (ESI) m/z [M+Na]+ calculated for 439.0185, found 439.0183.

(3aS,5R,5aR,8aR,8bR)-5-((4-bromophenyl)thio)-2,2,7,7-tetramethyltetrahydro-5H-bis([1,3]dioxolo)[4,5-b:4',5'-d]pyran (3af'): Yellow solid; $9.7 \mathrm{mg} ; 23 \%$ yield; d.r. $=3: 1$. M.p. $87-88^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.40(\mathrm{~s}, 4 \mathrm{H}), 5.68(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H})$, $5.20(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.67(\mathrm{dd}, J=7.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.48(\mathrm{dd}, J=7.6,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.38$ (dd, $J=5.0,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.55(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.39(\mathrm{~s}, 3 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(100$ MHz, Chloroform-d) $\delta 135.3,131.9,130.3,120.4,110.5,109.3,97.5,78.9,73.6,71.7,70.0$, 26.0, 25.8, 24.9, 24.7. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 439.0185, found 439.0184 .

(2R,3S,4R,5R,6S)-3,4,5-tris(benzyloxy)-2-((4-bromophenyl)thio)-6-methoxytetrahydro-2H-pyran (3ag): White solid; $24.5 \mathrm{mg} ; 39 \%$ yield; d.r. $=1: 1$. M.p. $87-88^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}(400$ MHz, Chloroform-d) $\delta 7.56-7.27(\mathrm{~m}, 19 \mathrm{H}), 4.93(\mathrm{~d}, J=10.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.88(\mathrm{~d}, J=10.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.85-4.75(\mathrm{~m}, 4 \mathrm{H}), 4.64(\mathrm{~d}, J=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.03-3.94(\mathrm{~m}$, $1 \mathrm{H}), 3.52(\mathrm{dd}, J=9.7,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{dd}, J=10.0,8.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.24(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}$ ( 100 MHz , Chloroform-d) $\delta 138.5,138.0,138.0,134.7,132.1,132.0,128.5,128.4,128.4$, $128.1,128.1,128.0,127.9,127.7,122.4,98.9,81.6,81.0,80.8,79.3,76.1,75.6,73.6,55.7$. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 643.1124, found 643.1122.

(2S,3S,4R,5R,6S)-3,4,5-tris(benzyloxy)-2-((4-bromophenyl)thio)-6-methoxytetrahydro$\mathbf{2 H}-p y r a n(3 a g '):$ White solid; $24.1 \mathrm{mg} ; 39 \%$ yield; d.r. $=1: 1$. M.p. $86-87^{\circ} \mathrm{C} . \mathbf{1}^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}(400$

MHz, Chloroform-d) $\delta 7.52-7.28(\mathrm{~m}, 19 \mathrm{H}), 5.36(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{~s}, 2 \mathrm{H}), 4.87(\mathrm{~d}$, $J=12.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.78-4.67(\mathrm{~m}, 3 \mathrm{H}), 4.56(\mathrm{~d}, J=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{t}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.80(\mathrm{dd}, J=8.2,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~d}, J=3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(100 \mathrm{MHz}$, Chloroform-d) $\delta 138.5,138.3,137.6,137.1,132.4,131.9,128.4,128.4,128.3,128.1,128.0$, $127.9,127.8,127.7,120.6,100.9,86.0,79.1,78.3,75.4,73.7,73.4,57.2$. HRMS (ESI) m/z $[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 643.1124 , found 643.1123 .

(2R,3S,4R,5R,6S)-2-((4-bromophenyl)thio)-3,4,5,6-tetramethoxytetrahydro-2H-pyran (3ah): Pale oil; 17.3 mg ; 44\% yield; d.r. = 1:1. ${ }^{\mathbf{1}} \mathbf{H}$ NMR (400 MHz, Chloroform-d) $\delta 7.56$ $7.34(\mathrm{~m}, 4 \mathrm{H}), 4.79(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.60(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H})$, $3.50(\mathrm{~s}, 4 \mathrm{H}), 3.27(\mathrm{~s}, 3 \mathrm{H}), 3.18(\mathrm{dd}, J=9.7,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.03-2.94(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathbf{N M R}(100$ MHz, Chloroform- $\boldsymbol{d}$ ) $\delta 133.9,131.0,130.8,121.5,97.2,82.2,81.3,80.1,79.5,60.0,59.8$, 58.2, 54.7. HRMS (ESI) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calculated for 415.0185, found 415.0183 .

(2S,3S,4R,5R,6S)-2-((4-bromophenyl)thio)-3,4,5,6-tetramethoxytetrahydro-2H-pyran (3ah'): White solid; 16.9 mg ; $43 \%$ yield; d.r. $=1: 1$. M.p. $122-123{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}$, Chloroform-d) $\delta 7.40(\mathrm{~s}, 4 \mathrm{H}), 5.40(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.87(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.70(\mathrm{t}, J=$ $8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.62(\mathrm{~s}, 3 \mathrm{H}), 3.53(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~s}, 3 \mathrm{H}), 3.47(\mathrm{dd}, J=8.5,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.40(\mathrm{~s}$, $3 \mathrm{H}), 3.24(\mathrm{dd}, J=8.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathbf{C}-\mathrm{NMR}(100 \mathrm{MHz}$, Chloroform-d) $\delta 137.0,132.1$, $131.9,120.6,100.3,85.3,81.2,80.3,78.2,60.7,59.3,59.0,57.1 . \operatorname{HRMS}(E S I) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$ calculated for 415.0185 , found 415.0184 .

(2R,3R,4R,5S,6R)-3,4,5-tris(benzyloxy)-2-(4-(benzyloxy)phenoxy)-6-((4-
bromophenyl)thio)tetrahydro-2H-pyran (3ai): White solid; $66.4 \mathrm{mg} ; 84 \%$ yield; $>20: 1$ d.r.; M.p.148-149 ${ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}-\mathrm{NMR}(400 \mathrm{MHz}, \mathbf{C h l o r o f o r m - d}) \delta 7.53-7.27(\mathrm{~m}, 22 \mathrm{H}), 7.18-7.12$ (m, 2H), $6.79-6.70(\mathrm{~m}, 4 \mathrm{H}), 5.61(\mathrm{~d}, J=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.59(\mathrm{~s}, 1 \mathrm{H}), 5.03 \mathrm{~s}, 2 \mathrm{H}), 4.96(\mathrm{~d}, J=$ $10.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.94-4.84(\mathrm{~m}, 2 \mathrm{H}), 4.81(\mathrm{~d}, J=10.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.78-4.70(\mathrm{~m}, 2 \mathrm{H}), 3.99-3.89$ (m, 2H), $3.68(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathbf{C}$-NMR ( $\mathbf{1 0 0} \mathbf{~ M H z}$, Chloroform-d) $\delta 154.6,150.9,138.3$, $138.2,137.5,137.1,133.0,132.6,132.1,128.7,128.6,128.5,128.4,128.3,128.2,128.1$, $128.0,127.9,127.8,127.7,127.5,127.2,118.6,115.5,98.7,84.1,81.4,80.4,79.0,75.9,75.2$, 73.2, 70.5. HRMS (ESI) m/z [M + Na] ${ }^{+}$calculated for 811.1699, found 811.1700

## 9. NMR spectra




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| 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 |  |  | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |












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| 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 |  | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -10 |








| 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 |  | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |
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| 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 |


[^0]:    ${ }^{\text {a }}$ Reaction conditions: A mixture of 1a ( 0.15 mmol$)$, $2 \mathrm{a}(0.10 \mathrm{mmol})$, DIPEA $(0.10 \mathrm{mmol})$ in solvent ( 1.0 mL ) was irradiated with 12 W blue LEDs at room temperature under $\mathrm{N}_{2}$ for 24 h ;
    ${ }^{\mathrm{b}}$ Isolated yields; ${ }^{\mathrm{c}}$ N.R. $=$ No reaction.

