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Supplementary Information

A catalyst-free cross-coupling of isocyanates and triarylboranes for

secondary amide synthesis

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1. General methods

Experiments involving moisture and/or air sensitive components were performed in oven-dried glassware by using Schlenk line techniques with a four-port dual-bank manifold under a nitrogen atmosphere. Anhydrous solvents such as 1,2-dichloroethane, benzotrifluoride, chlorobenzene, chloroform, 1,1,2,2-tetrachloroethane, and 1,1,2-trichloroethane were purchased in a septum-sealed bottle with 4 Å molecular sieve beads. Triarylboranes including triphenylborane, tris(4-methylphenyl)borane, tris(2-methylphenyl) borane, tris(4-fluorophenyl)borane, tris(3,5-dimethylphenyl)borane, tri-1-naphthalenylborane, tributylborane and tribenzylborane, as well as all isocyanates used are commercial available, which were purchased from Energy-Chemical, Aladdin, J&K Scientific, Macklin, Adamas, Bidepharm, Heowns, Chemieliva, etc without further purification. Other triarylboranes were freshly prepared according to the literature reports.¹ Diphenylborinic acid was prepared according to the known procedure.²

Analytical thin layer chromatography (TLC) was performed using Merck 60 F254 precoated silica gel plate (0.2 mm thickness). Subsequent to elution, plates were visualized using UV radiation (254 nm) on Spectroline Model ENF-24061/F 254 nm. Further visualization was possible by staining with basic solution of potassium permanganate or acidic solution of ceric molybdate, followed by heating on a hot plate. Flash chromatography was performed using NUOTAI silica gel (200 – 300 mesh) with distilled solvents. Columns were typically packed as slurry and equilibrated with petroleum ether prior to use.

Proton nuclear magnetic resonance (¹H NMR) and carbon nuclear magnetic resonance (¹³C NMR) spectroscopy were performed on a Bruker Advance 400 MHz and JEOL 400 MHz spectrometers. Chemical shifts for ¹H NMR spectra are reported as in units of parts per million (ppm) downfield from SiMe₄ (δ 0.0) and relative to the signal of chloroform-*d* (*J* = 7.264, singlet), methanol-*d*₄ (*J* = 3.310, quintet), and dimethylsulfoxide-*d*₆ (*J* = 2.500, quintet). Multiplicities were given as: s (singlet); d (doublet); t (triplet); q (quartet); dd (doublet of doublets); td (triplet of doublets); qd (quartet of doublets); m (multiplet), etc. The number of protons (n) for a given resonance is indicated by nH. Coupling constants are reported as a J value in Hz. Carbon nuclear magnetic resonance spectra (¹³C NMR) are reported as δ in units of parts per million (ppm) downfield from SiMe₄ (δ = 0.0) and relative to the signal of chloroform-*d* (δ = 77.16, triplet), methanol-*d*₄ (*J* = 49.00, septet), and dimethylsulfoxide-*d*₆ (*J* = 39.52, septet). To clarify the complete

signal assignments, "× number" indicates the multiple carbons due to the superposition of chemical shifts.

High resolution mass spectral analysis (HRMS) was performed on Water Q-TOF Premier mass spectrometer (Thermo Electron Corporation).

Table S1 Reaction optimization^a

Ì	N ² C ²⁰ +	BPh ₃	cat. solvent	→) 3a	N Ph
Entry	Catalyst	Solvent	t (° C)	Time (h)	$\mathbf{Yield} \ (\%)^b$
1	Ni(cod) ₂ , KO ^t Bu	CH ₂ ClCH ₂ Cl	120	24	20
2	KO'Bu	CH ₂ ClCH ₂ Cl	120	24	21
3	None	CH ₂ ClCH ₂ Cl	120	24	53
4	None	PhCF ₃	120	24	47
5	None	PhCl	120	24	24
6	None	CHCl ₃	120	24	51
7	None	CHCl ₂ CHCl ₂	120	24	51
8	None	CH ₂ ClCHCl ₂	120	24	76
9	None	CH ₂ ClCHCl ₂	120	16	84
10	None	CH ₂ ClCHCl ₂	120	12	71
11	None	CH ₂ ClCHCl ₂	110	16	72
12	None	CH ₂ ClCHCl ₂	130	16	69
13 ^c	None	CH ₂ ClCHCl ₂	120	16	83
14^d	None	CH ₂ ClCHCl ₂	120	16	76
15 ^{c,e}	None	CH ₂ ClCHCl ₂	120	16	73

^{*a*} Unless other specified, reactions were conducted with **1a** (0.3 mmol), **2a** (0.36 mmol), and solvent (2.0 mL) under N₂ for 24 h. ^{*b*} Isolated yields. ^{*c*} 1.5 mL solvent was used. ^{*d*} 1.0 mL solvent was used. ^{*e*} The reaction was performed under air. cod = 1,5-cyclooctadiene. n.d. = not detected.

2. General procedure for the synthesis of the products 3a-3ah

$$\begin{array}{cccc} R_{N} = C^{-0} & + & Ar_{3}B & \underbrace{CH_{2}CICHCl_{2}(0.2 \text{ M})}_{N_{2},120 \text{ °C}, 16 \text{ h}} & R_{N} & Ar \\ 1 & 2 & & 3 \end{array}$$

A 10 mL screw-capped Schlenk tube equipped with a stir bar was charged with isocyanate **1** (0.30 mmol, 1.0 equiv.), triarylborane **2** (0.36 mmol, 1.2 equiv.) and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C in oil bath for 16 h under N₂ atmosphere. After full conversion, the reaction mixture was cooled down to room temperature, quenched with sat. aq. NaHCO₃ (10 mL), and then extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were washed with sat. NaCl, dried over Na₂SO₄ and concentrated under vacuum. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate) to give the corresponding amide **3a–3ah**.

3. Scale-up synthesis of the product 3z



A 100 mL screw-capped Schlenk tube equipped with a stir bar was charged with 3,5dimethylphenyl isocyanate 1z (0.88 g, 6.0 mmol, 1.0 equiv.), triphenylborane 2a (1.74 g, 7.2 mmol, 1.2 equiv.), and 1,1,2-trichloroethane (30 mL). The mixture was stirred at 120 °C in oil bath for 16 h under N₂ atmosphere. After full conversion, the reaction mixture was cooled down to room temperature, quenched with sat. aq. NaHCO₃ (50 mL), and then extracted with CH₂Cl₂ (50 mL × 3). The combined organic layers were washed with sat. NaCl, dried over Na₂SO₄ and concentrated under vacuum. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 - 12:1) to give 3z (1.07 g, 4.75 mmol, 79%).

4. Sequential modular synthesis of the products 3ai-3ak^{3,4}



To a stirred CH₂Cl₂ (1 mL) solution of triphosgene (0.1 mmol, 29.7 mg, 0.33 equiv.) was added

a CH_2Cl_2 (0.5 mL) solution of 3-amino-9-ethylcarbazole (0.3 mmol, 63.1 mg, 1.0 equiv.) dropwise. After 30 min, the solution was cooled to -35 °C. Et₃N (0.1 mL) was added dropwise. The mixture was warmed to room temperature slowly and stirred for 2 h. The mixture was concentrated under vacuum to give the isocyanate **1ai**, which was used in the next step immediately.

A 10 mL screw-capped Schlenk tube was charged with the above isocyanate **1ai**, triphenylborane (0.36 mmol, 87.2 mg, 1.2 equiv.), and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C in oil bath for 16 h under N₂ atmosphere. After that, the reaction mixture was cooled down to room temperature, quenched with sat. aq. NaHCO₃ (10 mL), and then extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were washed with sat. NaCl, dried over Na₂SO₄ and concentrated under vacuum. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 - 10:1) to give the desired product **3ai** (73.2 mg, 0.233 mmol, 78% over two steps).



A 10 mL round-bottom flask was charged with amino acid ester hydrochloride (0.3 mmol, 1.0 equiv.), sat. aq. NaHCO₃ (1.2 mL) and CH₂Cl₂ (1.2 mL). The solution was stirred at 0 °C for 20 min. Whereafter, triphosgene (0.1 mmol, 29.7 mg, 0.33 equiv.) was added into the solution. The mixture was stirred at 0 °C for 1 h. After that, the reaction mixture was warmed to room temperature and then extracted with CH₂Cl₂ (10 mL \times 3). The combined organic layers were washed with sat. NaCl, dried over Na₂SO₄ and concentrated under vacuum to give the isocyanate **1aj** or **1ak**, which was used in the next step immediately.

A 10 mL screw-capped Schlenk tube equipped with a stir bar was charged with the above isocyanate **1aj** or **1ak**, triphenylborane (0.36 mmol, 87.2 mg, 1.2 equiv.), and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C in oil bath for 16 h under N₂ atmosphere. After that, the reaction mixture was cooled down to room temperature, quenched with sat. aq. NaHCO₃ (10 mL), and then extracted with CH_2Cl_2 (10 mL × 3). The combined organic layers were washed with sat.

NaCl, dried over Na₂SO₄ and concentrated under vacuum. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate) to give the product **3aj** (33.1 mg, 0.133 mmol, 44% over two steps) or **3ak** (35.4 mg, 0.119, 40% over two steps).

5. Mechanistic exploration experiments

5.1 Radical trapping experiments



An oven-dried 10 mL Schlenk tube equipped with a stir bar was charged with 4-methylphenyl isocyanate **1a** (40.3 mg, 0.30 mmol, 1.0 equiv.), triphenylborane **2a** (87.2 mg, 0.36 mmol, 1.2 equiv.), and radical scavenger (1.0 equiv.): TEMPO (0.3 mmol, 46.9 mg), BHT (0.3 mmol, 66.1 mg) or 1,1-diphenylethylene (0.3 mmol, 54.5 mg), and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C for 16 h under N₂ atmosphere. After cooling to room temperature, the mixture was quenched with sat. aq. NaHCO₃ (10 mL). The layer was separated and the aqueous layer was extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were rinsed with sat. NaCl (10 mL), dried over Na₂SO₄, and concentrated under vacuo. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 - 12:1) to give the product **3a** (with TEMPO, **3a**: 23.2 mg, 37%; with BHT, **3a**: 49.2 mg, 78%; with 1,1-diphenylethylene, **3a**: 47.9 mg, 76%).

5.2 Competition experiments



An oven-dried 10 mL Schlenk tube equipped with a stir bar was charged with 4-(trifluoromethyl)phenyl isocyanate **1n** (45.7 mg, 0.30 mmol, 1.0 equiv.), 4-methoxyphenyl isocyanate **1ab** (56.7 mg 0.30 mmol, 1.0 equiv.), triphenylborane **2a** (72.6 mg, 0.30 mmol, 1.0 equiv.), and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C for 4 h under N₂ atmosphere. After cooling to room temperature, the mixture was quenched with sat. aq. NaHCO₃ (10 mL). The layer was separated and the aqueous layer was extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were rinsed with sat. NaCl, dried over Na₂SO₄, and concentrated under vacuo. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 - 6:1) to give the product **3n** (46.6 mg, 0.176 mmol, 59%) and **3ab** (19.4 mg, 0.085 mmol, 28%). The ratio of **3n** and **3ab** was determined to be 2.1:1.



An oven-dried 10 mL Schlenk tube equipped with a stir bar was charged with 4-methylphenyl isocyanate **1a** (40.3 mg, 0.30 mmol, 1.0 equiv.), tris(4-methoxyphenyl)borane **2d** (99.6 mg, 0.30 mmol, 1.0 equiv.), tris(4-fluorophenyl)borane **2g** (88.8 mg, 0.30 mmol, 1.0 equiv.) and 1,1,2-trichloroethane (1.5 mL). The mixture was stirred at 120 °C for 4 h under N₂ atmosphere. After cooling to room temperature, the mixture was quenched with sat. aq. NaHCO₃ (10 mL). The layer was separated and the aqueous layer was extracted with CH_2Cl_2 (10 mL × 3). The combined organic layers were rinsed with sat. NaCl, dried over Na₂SO₄, and concentrated under vacuo. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 – 6:1) to give the product **3d** (19.3 mg, 0.080 mmol, 27%) and **3g** (8.7 mg, 0.038 mmol, 13%). The ratio of **3d** and **3g** was determined to be 2.1:1.

5.3 Control experiments



A 10 mL screw-capped Schlenk tube equipped with a stir bar was charged with 4-methylphenyl isocyanate **1a** (0.30 mmol, 1.0 equiv.), specific borane (x equiv.), and 1,1,2-trichloroethane (1.5 mL).

The mixture was stirred at 120 °C in oil bath for 16 h under N₂ atmosphere. After that, the reaction mixture was cooled down to room temperature, quenched with sat. aq. NaHCO₃ (10 mL), and then extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were washed with sat. NaCl, dried over Na₂SO₄ and concentrated under vacuum. The resultant residue was purified by flash column chromatography (petroleum ether/ethyl acetate = 20:1 - 12:1) to give the product **3a** (with 0.34 equiv. Ph₃B, **3a**: 19.0 mg, 30%; with 0.67 equiv. Ph₃B, **3a**: 36.2 mg, 57%; with 1.2 equiv. NaBPh₄, **3a**: 33.2 mg, 52%; with 1.2 equiv. Ph₂BOH, **3a**: 4.5 mg, 7%; with 1.2 equiv. PhB(OH)₂, **3a**: 0%).

Note: Under the standard reaction conditions, both diphenylborinic acid and phenylboronic acid could be isolated by flash column chromatography (petroleum ether/ethyl acetate). ¹H NMR data and spectra of two compounds are provided below.

Ph₂BOH: ¹**H NMR** (400 MHz, DMSO-*d*₆) δ 9.99 (brs, 1H), 7.70 (dd, *J* = 7.6, 1.5 Hz, 4H), 7.50 - 7.45 (m, 2H), 7.44 - 7.38 (m, 4H). The data of diphenylborinic acid is consistent with the literature report.⁵





¹H NMR spectrum for diphenylborinic acid (DMSO-d₆)

PhB(OH)₂: ¹H NMR (400 MHz, Chloroform-*d*) δ 8.33 – 8.19 (m, 2H), 7.65 – 7.57 (m, 1H),

7.55 - 7.49 (m, 2H). The data of phenylboronic acid is consistent with the literature report.⁶



¹H NMR spectrum for phenylboronic acid (CDCl₃)

6. NMR data of the products

N-(p-Tolyl)benzamide (3a)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (52.7 mg, 0.249 mmol, 83%). M.p.: 155.8 - 157.4 °C. ¹H NMR (400 MHz, Chloroform-d) δ 8.01 (brs, 1H), 7.86 – 7.84 (m, 2H), 7.54 - 7.50 (m, 3H), 7.46 - 7.42 (m, 2H), 7.15 (d, J = 8.2 Hz, 2H), 2.34 (s, 3H). ¹³C NMR

(101 MHz, Chloroform-d) δ 165.9, 135.5, 135.1, 134.3, 131.8, 129.6 \times 2, 128.8 \times 2, 127.1 \times 2, $120.5 \times 2, 21.0.$ HRMS (ESI): m/z calculated for $C_{14}H_{14}NO [M + H]^+: 212.1070$; Found: 212.1068.

4-Methyl-*N*-(*p*-tolyl)benzamide (3b)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (41.9 mg, 0.186 mmol, 62%). M.p.: 165.4 -166.6 °C. ¹H NMR (400 MHz, Chloroform-d) δ 8.14 (brs, 1H), 7.74 (d, J = 8.2 Hz, 2H), 7.53 (d, J = 8.5 Hz, 2H), 7.20 (d, J = 7.9 Hz, 2H), 7.13 (d, J = 8.3 Hz, 2H), 2.39 (s, 3H), 2.33 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 165.9, 142.1, 135.6, 134.0, 132.2, 129.5 \times 2, 129.3 \times 2, 127.2 \times 2, 120.6 \times 2, 21.5, 21.0. HRMS (ESI) m/z calculated for C₁₅H₁₆NO [M + H]⁺: 226.1226; Found: 226.1228.

2-Methyl-N-(p-tolyl)benzamide (3c)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (45.2 mg, 0.201 mmol, 67%). M.p.: 145.4 - 146.6 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.54 (brs, 1H), 7.50 (d, J = 8.1

Hz, 2H), 7.45 (d, J = 7.6 Hz, 1H), 7.35 (t, J = 7.4 Hz, 1H), 7.25 – 7.21 (m, 2H), 7.17 (d, J = 8.0 Hz, 2H), 2.49 (s, 3H), 2.35 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 168.1, 136.7, 136.5, 135.6, 134.3, 131.3, 130.3, 129.7 × 2, 126.7, 126.0 × 2, 120.1, 21.0, 19.9. HRMS (ESI) m/z calculated for $C_{15}H_{16}NO [M + H]^+: 226.1226;$ Found: 226.1228.

4-Methoxy-N-(p-tolyl)benzamide (3d)⁷



The title compound was prepared according to the general procedure and isolated as a yellow solid (56.3 mg, 0.233 mmol, 78%). M.p.: 147.0 – 147.8 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.03 (brs,

1H), 7.82 (d, J = 8.1 Hz, 2H), 7.51 (d, J = 8.4 Hz, 2H), 7.13 (d, J = 8.1 Hz, 2H), 6.90 (d, J = 8.8 Hz, 2H), 3.84 (s, 3H), 2.33 (s, 3H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 165.5, 162.4, 135.6, 134.0, 129.6 × 2, 129.0 × 2, 127.2, 120.5 × 2, 113.9 × 2, 55.5, 21.0. **HRMS** (ESI) m/z calculated for C₁₅H₁₆NO₂ [M + H]⁺: 242.1176; Found: 242.1178.

3-Methoxy-*N*-(*p*-tolyl)benzamide (3e)⁸



The title compound was prepared according to the general procedure and isolated as a brown solid (51.2 mg, 0.212 mmol, 71%). M.p.: 122.5 – 125.7 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.86 (brs, 1H), 7.52 (d, *J* = 8.2 Hz, 2H), 7.45 – 7.40 (m, 1H), 7.38 – 7.36 (m, 2H), 7.17 (d, *J* = 8.1 Hz,

2H), 7.08 – 7.05 (m, 1H), 3.85 (s, 3H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.7, 160.0, 136.7, 135.4, 134.4, 129.8 × 2, 129.7 × 2, 120.4, 118.8, 118.1, 112.5, 55.6, 21.0. **HRMS** (ESI) m/z calculated for C₁₅H₁₆NO₂ [M + H]⁺: 242.1176; Found: 242.1186.

4-(tert-Butyl)-N-(p-tolyl)benzamide (3f)⁸



The title compound was prepared according to the general procedure and isolated as a white solid (51.3 mg, 0.192 mmol, 64%). M.p.: 125.6 -127.9 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.83 – 7.79 (m, 3H),

7.52 (d, J = 8.4 Hz, 2H), 7.48 (d, J = 8.5 Hz, 2H), 7.16 (d, J = 8.2 Hz, 2H), 2.34 (s, 3H), 1.35 (s, 9H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 165.7, 155.4, 135.6, 134.2, 132.3, 129.7 × 2, 127.0 × 2, 125.8 × 2, 120.3 × 2, 35.1, 31.3 × 3, 21.0. **HRMS** (ESI) m/z calculated for C₁₈H₂₂NO [M + H]⁺: 268.1696; Found: 268.1700.

4-Fluoro-N-(p-tolyl)benzamide (3g)⁸



The title compound was prepared according to the general procedure and isolated as a white solid (39.9 mg, 0.174 mmol, 58%). M.p.: 174.0 - 174.6 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.88 – 7.85 (m, 3H), 7.49 (d, J = 8.4 Hz, 2H), 7.17 – 7.11 (m, 4H), 2.34 (s, 3H). ¹⁹F NMR (377 MHz, Chloroform-d) δ -107.56. ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.0 (d, J = 253.1 Hz), 164.8, 135.3, 134.5, 131.3 $(d, J = 3.3 \text{ Hz}), 129.7 \times 2, 129.6 (d, J = 9.0 \text{ Hz}) \times 2, 120.6 \times 2, 115.9 (d, J = 21.9 \text{ Hz}) \times 2, 21.0.$ **HRMS** (ESI) m/z calculated for C₁₄H₁₃FNO [M + H]⁺: 230.0976; Found: 230.0983.

3,5-Dimethyl-N-(p-tolyl)benzamide (3h)9



The title compound was prepared according to the general procedure and isolated as a yellow solid (44.5 mg, 0.186 mmol, 62%). M.p.: 123.9 -125.6 °C. ¹H NMR (400 MHz, Chloroform-d) δ 7.77 (brs, 1H), 7.53 (d, J = 8.4 Hz, 2H), 7.46 (d, J = 1.5 Hz, 2H), 7.18 – 7.16 (m, 3H), 2.38 (s, 6H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 166.2, 138.6 × 2, 135.6, 135.2, 134.2, 133.4, 129.7 × 2, 124.9 × 2, $120.3 \times 2, 21.4 \times 2, 21.0$. **HRMS** (ESI) m/z calculated for C₁₆H₁₈NO [M + H]⁺: 240.1383; Found:

240.1389.

4-Methoxy-3-methyl-*N*-(*p*-tolyl)benzamide (3i)



The title compound was prepared according to the general procedure and isolated as a yellow solid (51.8 mg, 0.203 mmol, 68%). M.p.: 124.5 – 125.2 °C. ¹**H NMR** (400 MHz, Chloroform-*d*) δ 7.83 – 7.63

(m, 3H), 7.57 - 7.48 (m, 2H), 7.23 - 7.12 (m, 2H), 6.87 (d, J = 8.5 Hz, 1H), 3.89 (s, 3H), 2.34 (s, 3H), 2.27 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 165.5, 160.7, 135.7, 134.0, 129.7 × 2, 129.6, 127.1, 126.8, 126.5, 120.3 × 2, 109.6, 55.6, 21.0, 16.5. **IR** (KBr): v 3282, 2972, 1646, 1592, 1568, 1519, 1393, 1047, 868, 508 cm⁻¹. **HRMS** (ESI) m/z calculated for $C_{16}H_{18}NO_2 [M + H]^+$: 256.1332; Found: 256.1334.

4-Methoxy-2-methyl-*N*-(*p*-tolyl)benzamide (3j)



The title compound was prepared according to the general procedure and isolated as a white solid (49.1 mg, 0.192 mmol, 64%). M.p.: 161.4 – 163.5 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.49 – 7.45

(m, 3H), 7.39 (brs, 1H), 7.17 (d, J = 8.1 Hz, 2H), 6.79 - 6.75 (m, 2H), 3.84 (s, 3H), 2.52 (s, 3H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 166.3, 159.9, 139.1, 136.3, 134.0, 129.6 × 2, 129.0, 127.5, 120.8, 117.3, 111.8 × 2, 56.1, 21.0, 20.5. **IR** (KBr): v 3279, 2972, 1732, 1644, 1607, 1520, 1398, 1263, 824, 506 cm⁻¹. **HRMS** (ESI) m/z calculated for C₁₆H₁₈NO₂ [M + H]⁺: 256.1332; Found: 256.1337.

N-(*p*-Tolyl)-1-naphthamide (3k)¹⁰



The title compound was prepared according to the general procedure and isolated as a yellow solid (47.9 mg, 0.183 mmol, 61%). M.p.: 192.8 – 194.2 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.36 – 8.34 (m, 1H),

7.94 (d, J = 8.2 Hz, 1H), 7.90 – 7.88 (m, 1H), 7.74 (brs, 1H), 7.70 (d, J = 7.0 Hz, 1H), 7.58 – 7.54 (m, 4H), 7.49 – 7.45 (m, 1H), 7.20 (d, J = 8.0 Hz, 2H), 2.37 (s, 3H). ¹³**C** NMR (101 MHz, Chloroform-*d*) δ 167.6, 135.6, 134.7, 134.4, 133.8, 131.0, 130.2, 129.7 × 2, 128.5 × 2, 127.4, 126.7, 125.4, 125.2, 124.8, 120.2, 21.1. HRMS (ESI) m/z calculated for C₁₈H₁₆NO [M + H]⁺: 262.1226; Found: 262.1216.

N-(*p*-Tolyl)benzo[*d*][1,3]dioxole-5-carboxamide (31)¹⁰



The title compound was prepared according to the general procedure and isolated as a yellow solid (61.4 mg, 0.241 mmol, 80%). M.p.: 137.2 – 139.2 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.71 (brs, 1H), 7.49 (d, *J* = 8.4 Hz, 2H), 7.43 – 7.34 (m, 1H), 7.20 – 7.13 (m, 2H), 6.86 (d, *J*

= 8.1 Hz, 1H), 6.05 (s, 2H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.1, 150.7, 148.2, 135.4, 134.2, 129.7 × 2, 129.3, 121.8, 120.4 × 2, 108.2, 107.8, 101.9, 21.0. HRMS (ESI) m/z calculated for C₁₅H₁₄NO₃ [M + H]⁺: 256.0968; Found: 256.0972.

N-(*p*-Tolyl)thiophene-2-carboxamide (3m)¹⁰



The title compound was prepared according to the general procedure and isolated as a yellow solid (53.4 mg, 0.246 mmol, 82%). M.p.: 203.8 – 205.3 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.70 (brs, 1H), 7.62 (dd, *J* = 3.7,

1.2 Hz, 1H), 7.53 (dd, J = 5.0, 1.2 Hz, 1H), 7.51 – 7.47 (m, 2H), 7.20 – 7.15 (m, 2H), 7.12 (dd, J = 5.0, 3.7 Hz, 1H), 2.34 (s, 3H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 160.2, 139.6, 135.1, 134.4,

130.7, 129.6 \times 2, 128.5, 127.9, 120.6 \times 2, 21.0. **HRMS** (ESI) m/z calculated for C₁₂H₁₂NOS [M + H]+: 218.0634; Found: 218.0641.

N-(4-(Trifluoromethyl)phenyl)benzamide (3n)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (59.2 mg, 0.223 mmol, 74%). M.p.: 206.6 -208.2 °C. ¹**H NMR** (400 MHz, DMSO- d_6) δ 10.59 (brs, 1H), 8.03 (d,

J = 8.5 Hz, 2H), 7.98 (d, *J* = 7.1 Hz, 2H), 7.72 (d, *J* = 8.4 Hz, 2H), 7.64 - 7.60 (m, 1H), 7.57 - 7.53 (m, 2H). ¹⁹F NMR (376 MHz, DMSO-d₆) δ -60.24. ¹³C NMR (101 MHz, DMSO-d₆) δ 166.1, 142.9 $(q, J = 1.5 \text{ Hz}), 134.5, 132.0, 128.5 \times 2, 127.8 \times 2, 126.0 \times 2 (q, J = 4.0 \text{ Hz}), 124.4 (q, J = 271.3 \text{ Hz})$ Hz), 123.6 (q, J = 32.0 Hz), 120.1 × 2. HRMS (ESI) m/z calculated for C₁₄H₁₁F₃NO [M + H]⁺: 266.0787; Found: 266.0788.

N-(4-Cyanophenyl)benzamide (30)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (42.1 mg, 0.189 mmol, 63%). M.p.: 171.8 -173.4 °C. ¹H NMR (400 MHz, Chloroform-d) δ 8.23 (brs, 1H), 7.87 (d, J = 7.6 Hz, 2H), 7.81 (d, J = 8.8 Hz, 2H), 7.64 – 7.57 (m, 3H), 7.49 (t, J = 7.6 Hz, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 166.1, 142.2, 134.2, 133.4 × 2, 132.6, 129.1 × 2, 127.3 × 2, 120.1 × 2, 119.0, 107.4. **HRMS** (ESI) m/z calculated for C₁₄H₁₁N₂O [M + H]⁺: 223.0866; Found: 223.0869.

N-(4-Nitrophenyl)benzamide (3p)¹¹



The title compound was prepared according to the general procedure and isolated as a yellow solid (54.8 mg, 0.226 mmol, 75%). M.p.: 199.8 -200.9 °C. ¹**H NMR** (400 MHz, Methanol-*d*₄) δ 8.28 (d, *J* = 9.1 Hz, 2H),

8.08 (brs, 1H), 7.91 – 7.89 (m, 2H), 7.86 (d, J = 9.2 Hz, 2H), 7.64 – 7.60 (m, 1H), 7.56 – 7.52 (m, 2H). ¹³C NMR (101 MHz, Methanol- d_4) δ 169.0, 146.4, 144.8, 135.8, 133.4, 129.7 × 2, 128.8 × 2, 125.7×2 , 121.3×2 . **HRMS** (ESI) m/z calculated for C₁₃H₁₁N₂O₃ [M + H]⁺: 243.0764; Found: 243.0762.

N-(4-Fluorophenyl)benzamide (3q)¹¹



The title compound was prepared according to the general procedure and isolated as a white solid (40.7 mg, 0.189 mmol, 63%). M.p.: 184.5 – 187.3 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.87 (d, *J* = 7.2 Hz, 2H), 7.82

(brs, 1H), 7.62 – 7.55 (m, 3H), 7.52 – 7.48 (m, 2H), 7.07 (t, J = 8.6 Hz, 2H). ¹⁹F NMR (377 MHz, Chloroform-*d*) δ -117.48. ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.9, 159.7 (d, J = 243.9 Hz), 134.9, 134.0 (d, J = 3.1 Hz), 132.1, 129.0 × 2, 127.1 × 2, 122.3 × 2 (d, J = 8.2 Hz), 115.9 × 2 (d, J = 22.6 Hz). HRMS (ESI) m/z calculated for C₁₃H₁₁FNO [M + H]⁺: 216.0819; Found: 216.0828.

N-(4-Chlorophenyl)benzamide (3r)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (50.3 mg, 0.217 mmol, 72%). M.p.: 188.2 - 190.7 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.38 (brs, 1H), 7.96 - 7.93 (m,

2H), 7.82 (d, J = 8.9 Hz, 2H), 7.62 – 7.58 (m, 1H), 7.56 – 7.51 (m, 2H), 7.41 (d, J = 8.9 Hz, 2H). ¹³C NMR (101 MHz, DMSO- d_6) δ 165.7, 138.2, 134.7, 131.7, 128.5 × 2, 128.4 × 2, 127.7 × 2, 127.3, 121.8 × 2. HRMS (ESI) m/z calculated for C₁₃H₁₁³⁵CINO [M + H]⁺: 232.0524; Found: 232.0531; C₁₃H₁₁³⁷CINO [M + H]⁺: 234.0494; Found: 234.0502.

N-(2-Chlorophenyl)benzamide (3s)¹¹



The title compound was prepared according to the general procedure and isolated as a white solid (62.6 mg, 0.270 mmol, 90%). M.p.: 120.2 - 122.8 °C.

¹**H NMR** (400 MHz, Chloroform-*d*) δ 8.58 (dd, J = 8.3, 1.5 Hz, 1H), 8.46 (brs,

1H), 7.94 – 7.92 (m, 2H), 7.61 – 7.57 (m, 1H), 7.55 – 7.51 (m, 2H), 7.44 – 7.41 (m, 1H), 7.37 – 7.32 (m, 1H), 7.11 – 7.07 (m, 1H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 165.4, 134.9, 134.8, 132.3, 129.2, 129.1 × 2, 128.0, 127.2 × 2, 124.9, 123.2, 121.6. **HRMS** (ESI) m/z calculated for C₁₃H₁₁³⁵CINO [M + H]⁺: 232.0524; Found: 232.0529; C₁₃H₁₁³⁷CINO [M + H]⁺: 234.0494; Found: 234.0495.

N-(3,4-Dichlorophenyl)benzamide (3t)¹²



The title compound was prepared according to the general procedure and isolated as a white solid (53.1 mg, 0.200 mmol, 67%). M.p.: 148.6 – 149.3 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.03 (s, 1H), 7.88 – 7.87 (m,

1H), 7.84 – 7.82 (m, 2H), 7.58 – 7.55 (m, 1H), 7.48 – 7.44 (m, 3H), 7.38 (d, J = 8.7 Hz, 1H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 166.0, 137.5, 134.3, 133.0, 132.4, 130.7, 129.0 × 2, 127.9, 127.2 × 2, 122.1, 119.6. HRMS (ESI) m/z calculated for C₁₃H₁₀³⁵Cl₂NO [M + H]⁺: 266.0134; Found: 266.0138; C₁₃H₁₀³⁷Cl₂NO [M + H]⁺: 268.0104; Found: 268.0105.

N-(3-Bromophenyl)benzamide (3u)¹³

The title compound was prepared according to the general procedure and isolated as a white solid (57.3 mg, 0.208 mmol, 69%). M.p.: 138.7 – 140.1 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.15 (brs, 1H), 7.88 (t, *J* = 2.0 Hz, 1H), 7.83 – 7.80 (m, 2H), 7.55 – 7.50 (m, 2H), 7.44 – 7.40 (m, 2H), 7.26 – 7.24 (m, 1H), 7.17 (t, *J* = 8.0 Hz, 1H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 166.1, 139.3, 134.5, 132.2, 130.4, 128.9 ×2, 127.6, 127.2 ×2, 123.4, 122.7, 119.0. HRMS (ESI) m/z calculated for C₁₃H₁₁⁷⁹BrNO [M + H]⁺: 276.0019; Found: 276.0018; C₁₃H₁₁⁸¹BrNO [M + H]⁺: 277.9998; Found: 277.9994.

N-(4-(Trifluoromethoxy)phenyl)benzamide (3v)¹⁴

The title compound was prepared according to the general procedure and isolated as a white solid (64.8 mg, 0.230 mmol, 77%). M.p.: 187.2 $-190.0 \degree C. \ ^{1}H \ NMR \ (400 \ MHz, \ Methanol-d_4) \ \delta \ 7.94 - 7.92 \ (m, \ 2H),$ 7.81 (d, J = 9.1 Hz, 2H), 7.61 - 7.56 (m, 1H), 7.53 - 7.49 (m, 2H), 7.27 (d, J = 8.6 Hz, 2H). ^{19}F NMR (376 MHz, Methanol-d_4) $\delta \ -59.62. \ ^{13}C \ NMR \ (101 \ MHz, \ Methanol-d_4) \ \delta \ 168.9, 146.7 \ (q, J = 2.1 \ Hz), 139.1, 136.0, 133.0, 129.7 \times 2, 128.7 \times 2, 123.4 \times 2, 122.6 \times 2, 122.0 \ (q, J = 254.7 \ Hz).$ HRMS (ESI) m/z calculated for C₁₄H₁₁F₃NO₂ [M + H]⁺: 282.0736; Found: 282.0740.

N-Phenylbenzamide (3w)⁷



The title compound was prepared according to the general procedure and isolated as a white solid (42.0 mg, 0.213 mmol, 71%). M.p.: 161.6 - 162.4 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.97 – 7.83 (m, 3H), 7.70 – 7.61 (m,

2H), 7.58 – 7.53 (m, 1H), 7.50 – 7.46 (m, 2H), 7.40 – 7.36 (m, 2H), 7.16 (td, J = 7.5, 1.1 Hz, 1H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.9, 138.0, 135.1, 132.0, 129.2 × 2, 128.9 × 2, 127.2 × 2, 124.7, 120.3 × 2. HRMS (ESI) m/z calculated for C₁₃H₁₂NO [M + H]⁺: 198.0913; Found: 198.0914.

N-(*o*-Tolyl)benzamide (3x)⁷

The title compound was prepared according to the general procedure and isolated as a white solid (37.6 mg, 0.178 mmol, 59%). M.p.: 145.2 – 146.5 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 9.89 (brs, 1H), 8.00 – 7.98 (m, 2H), 7.61 – 7.51 (m, 3H), 7.36 – 7.34 (m, 1H), 7.29 – 7.27 (m, 1H), 7.24 – 7.15 (m, 2H), 2.24 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 165.3, 136.4, 134.5, 133.7, 131.5, 130.3, 128.4 ×2, 127.6 ×2, 126.6, 126.0 ×2, 17.9. HRMS (ESI) m/z calculated for C₁₄H₁₄NO [M + H]⁺: 212.1070; Found: 212.1069.

N-(*m*-Tolyl)benzamide $(3y)^7$



The title compound was prepared according to the general procedure and isolated as a yellow solid (44.3 mg, 0.210 mmol, 70%). M.p.: 124.9 - 125.6 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.91 (brs, 1H), 7.86 – 7.83 (m, 2H),

7.54 – 7.50 (m, 2H), 7.47 – 7.40 (m, 3H), 7.25 – 7.21 (m, 1H), 6.95 (dd, J = 7.6, 1.4 Hz, 1H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.9, 139.1, 138.0, 135.2, 131.9, 129.0, 128.9 × 2, 127.1 × 2, 125.5, 121.0, 117.4, 21.6. **HRMS** (ESI) m/z calculated for C₁₄H₁₄NO [M + H]⁺: 212.1070; Found: 212.1068.

N-(3,5-Dimethylphenyl)benzamide (3z)¹⁵



The title compound was prepared according to the general procedure and isolated as a yellow solid (50.9 mg, 0.226 mmol, 75%). M.p.: 141.3 – 143.0

°C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.03 (d, *J* = 11.1 Hz, 1H), 7.87 –

7.84 (m, 2H), 7.53 – 7.49 (m, 1H), 7.46 – 7.41 (m, 2H), 7.29 (s, 2H), 6.79 (brs, 1H), 2.30 (s, 6H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.9, 138.8, 137.9, 135.1, 131.8 × 2, 128.8 × 2, 127.1 × 2, 126.4, 118.2 × 2, 21.4 × 2. HRMS (ESI) m/z calculated for C₁₅H₁₆NO [M + H]⁺: 226.1226; Found: 226.1231.

N-(3-Chloro-4-methylphenyl)benzamide (3aa)¹⁶



The title compound was prepared according to the general procedure and isolated as a yellow solid (53.4 mg, 0.217 mmol, 72%). M.p.: 124.4 – 125.9 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.99 (brs, 1H), 7.85 –

7.82 (m, 2H), 7.72 (d, J = 2.3 Hz, 1H), 7.55 – 7.51(m, 1H), 7.47 – 7.39 (m, 3H), 7.17 (d, J = 8.2 Hz, 1H), 2.34 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 165.9, 136.8, 134.7, 134.6, 132.3, 132.1, 131.1, 128.9 × 2, 127.2 × 2, 121.1, 118.8, 19.6. **HRMS** (ESI) m/z calculated for C₁₄H₁₃³⁵ClNO [M + H]⁺: 246.0680; Found: 246.0686; C₁₄H₁₃³⁷ClNO [M + H]⁺: 248.0651; Found: 248.0658.

N-(4-Methoxyphenyl)benzamide (3ab)⁷



The title compound was prepared according to the general procedure and isolated as a yellow solid (41.5 mg, 0.183 mmol, 61%). M.p.: 149.0 - 151.1 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.86 (d, *J* = 7.6 Hz,

2H), 7.81 (brs, 1H), 7.55 – 7.52 (m, 3H), 7.49 –7.45 (m, 2H), 6.92 – 6.89 (m, 2H), 3.82 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 156.7, 135.1, 131.8, 131.1, 128.8 × 2, 127.1 × 2, 122.3, 114.3 × 4, 55.6. HRMS (ESI) m/z calculated for C₁₄H₁₄NO₂ [M + H]⁺: 228.1019; Found: 228.1027.

N-(Naphthalen-1-yl)benzamide (3ac)⁷



The title compound was prepared according to the general procedure and isolated as a yellow solid (39.9 mg, 0.161 mmol, 54%). M.p.: 160.0 – 161.5 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.21 (brs, 1H), 8.07 (d, *J* = 7.4 Hz,

1H), 8.01 (d, J = 7.5 Hz, 2H), 7.92 (t, J = 5.7 Hz, 2H), 7.76 (d, J = 8.3 Hz, 1H), 7.61 – 7.52 (m, 6H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 166.4, 135.0, 134.3, 132.5, 132.1, 129.0 × 2, 129.0, 127.6, 127.3 × 2, 126.6, 126.3, 126.2, 125.9, 121.4, 120.8. HRMS (ESI) m/z calculated for C₁₇H₁₄NO [M + H]⁺: 248.1070; Found: 248.1078.

N-Benzylbenzamide (3ad)¹¹



The title compound was prepared according to the general procedure and isolated as a white solid (38.6 mg, 0.183 mmol, 61%). M.p.: 120.2 - 121.2 °C. ¹**H NMR** (400 MHz, Chloroform-*d*) δ 7.72 (dd, *J* = 8.4, 1.4 Hz, 2H),

7.42 - 7.40 (m, 1H), 7.33 (td, J = 7.4, 1.4 Hz, 2H), 7.27 - 7.25 (m, 4H), 7.23 - 7.19 (m, 1H), 6.60(brs, 1H), 4.55 (dd, J = 5.7, 1.8 Hz, 2H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 167.5, 138.3, 134.4, 131.6, 128.8 × 2, 128.6 × 2, 127.9 × 2, 127.6, 127.1 × 2, 44.1. HRMS (ESI) m/z calculated for C₁₄H₁₄NO [M + H]⁺: 212.1070; Found: 212.1073.

N-Ethylbenzamide (3ae)¹¹

The title compound was prepared according to the general procedure and isolated as a white solid (24.5 mg, 0.164 mmol, 55%). M.p.: 119.9 – 120.2 °C. ¹H NMR (400 MHz, Chloroform-d) δ 7.78 – 7.75 (m, 2H), 7.51 – 7.47 (m, 1H), 7.44 –

7.38 (m, 2H), 6.22 (brs, 1H), 3.50 (qd, J = 7.3, 5.6 Hz, 2H), 1.25 (t, J = 7.3 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 167.7, 134.9, 131.5, 128.7 × 2, 126.9 × 2, 35.1, 15.0. **HRMS** (ESI) m/z calculated for C₉H₁₂NO [M + H]⁺: 150.0913; Found: 150.0913.

N-Butylbenzamide (3af)¹⁷



The title compound was prepared according to the general procedure and isolated as a yellow oil (28.2 mg, 0.159 mmol, 53%). ¹H NMR (400 MHz, Chloroform-*d*) δ 7.76 (d, J = 7.4 Hz, 2H), 7.49 – 7.40 (m, 3H), 6.20 (brs, 1H), 3.46 (q, J = 6.8 Hz, 2H), 1.62 - 1.58 (m, 2H), 1.44 - 1.41 (m, 2H), 0.96 (t, J = 7.3 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 167.7, 135.0, 131.4, 128.7 × 2, 126.9 × 2, 39.9, 31.9, 20.3, 13.9. **HRMS** (ESI) m/z: calculated for $C_{11}H_{16}NO [M + H]^+$: 178.1226; Found: 178.1229.

N-Octylbenzamide (3ag)¹⁷



The title compound was prepared according to the general procedure and isolated as a white solid (41.2 mg, 0.177 mmol, 59%). M.p.: 119.8 – 119.9 °C. ¹H NMR (400 MHz, Chloroform-

d) δ 7.78 – 7.75 (m, 2H), 7.51 – 7.46 (m, 1H), 7.44 – 7.39 (m, 2H), 6.23 (brs, 1H), 3.44 (td, J = 7.3, 5.7 Hz, 2H), 1.64 – 1.59 (m, 2H), 1.33 – 1.26 (m, 10H), 0.90 – 0.86 (m, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 167.7, 134.9, 131.4, 128.6 × 2, 127.0 × 2, 40.2, 31.9, 29.8, 29.4, 29.3, 27.1, 22.7, 14.2. **HRMS** (ESI) m/z calculated for $C_{15}H_{24}NO [M + H]^+$: 234.1852; Found: 234.1854.

N-Cyclohexylbenzamide (3ah)¹⁸



The title compound was prepared according to the general procedure and isolated as a white solid (32.3 mg, 0.159 mmol, 53%). M.p.: 137.8 – 139.6 °C. ¹**H NMR** (400 MHz, Chloroform-d) δ 7.76 – 7.74 (m, 2H), 7.51 – 7.47 (m,

1H), 7.45 – 7.40 (m, 2H), 5.96 (brs, 1H), 4.03 – 3.94 (m, 1H), 2.06 – 2.02 (m, 2H), 1.79 – 1.72 (m, 2H), 1.69 – 1.64 (m, 1H), 1.49 – 1.38 (m, 2H), 1.29 – 1.19 (m, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 166.8, 135.3, 131.4, 128.7 × 2, 126.9 × 2, 48.8, 33.4 × 2, 25.7 × 2, 25.1. **HRMS** (ESI) m/z calculated for $C_{13}H_{18}NO [M + H]^+$: 204.1383; Found: 204.1382.

N-(9-Ethyl-9H-carbazol-3-yl)benzamide (3ai)¹⁹



The title compound was prepared according to the typical procedure and isolated as a yellow oil (73.2 mg, 0.233 mmol, 78%). M.p.: 184.8 -185.8 °C. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.44 (brs, 1H), 8.12 (s, 1H), 8.06 (d, J = 7.8 Hz, 1H), 7.94 (d, J = 7.3 Hz, 2H), 7.63 (d, J = 8.4 Hz, 1H), 7.55 (t, J = 7.4 Hz,

1H), 7.50 - 7.46 (m, 3H), 7.40 (d, J = 8.1 Hz, 1H), 7.34 (d, J = 8.9 Hz, 1H), 7.21 (t, J = 7.4 Hz, 1H), 4.33 (q, J = 7.3 Hz, 2H), 1.41 (t, J = 7.2 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 166.0, 140.5, 137.5, 135.3, 131.7, 129.8, 128.8 × 2, 127.2 × 2, 126.0, 123.1, 122.9, 120.8, 119.9, 118.9, 113.3, 108.7 × 2, 37.7, 13.9. **HRMS** (ESI) m/z calculated for $C_{21}H_{19}N_2O[M+H]^+: 315.1492$; Found: 315.1498.

Methyl benzoyl-L-leucinate (3aj)²⁰



The title compound was prepared according to the typical procedure and isolated as a yellow oil (33.1 mg, 0.133 mmol, 44%). ¹H NMR (400 MHz, Chloroform-d) δ 7.83 – 7.79 (m, 2H), 7.54 – 7.50 (m, 1H), 7.47 – 7.43 (m, 1H), 6.54 (d, J = 8.3 Hz, 1H), 4.90 - 4.84 (m, 1H), 3.77 (s, 3H), 1.79 - 1.67 (m, 3H),

1.01 - 0.97 (m, 6H). ¹³C NMR (101 MHz, Chloroform-d) δ 173.9, 167.2, 134.0, 131.9, 128.8 × 2, $127.2 \times 2, 52.6, 51.2, 42.0, 25.1, 23.0, 22.2$. **HRMS** (ESI) m/z calculated for C₁₄H₂₀NO₃ [M + H]⁺: 250.1438; Found: 250.1437.

Ethyl benzoyl-L-phenylalaninate (3ak)²¹



The title compound was prepared according to the typical procedure and isolated as a yellow oil (35.4 mg, 0.119 mmol, 40%). ¹H NMR (400 MHz, Chloroform-*d*) δ 7.75 – 7.72 (m, 2H), 7.53 – 7.49 (m, 1H), 7.45 – 7.41 (m, 2H),

7.32 – 7.25 (m, 2H), 7.16 – 7.14 (m, 2H), 6.64 (d, J = 7.6 Hz, 1H), 5.10 – 5.05 (m, 1H), 4.22 (q, J = 7.1 Hz, 2H), 3.32 – 3.22 (m, 2H), 1.28 (t, J = 7.2 Hz, 3H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 171.7, 166.9, 136.0, 134.0, 131.9, 129.5 × 2, 128.7 × 4, 127.3, 127.1 × 2, 61.8, 53.6, 38.0, 14.3. **HRMS** (ESI) m/z calculated for C₁₈H₂₀NO₃ [M + H]⁺: 298.1438; Found: 298.1440.

7. DFT calculations for the reaction mechanism

DFT calculations were performed to gain insight into the mechanisms of the catalyst-free crosscoupling of isocyanates and triarylboranes.

Computational details

All theoretical calculations were carried out using density functional theory with the Gaussian 16 program package,²² and the structures were illustrated by CYLview (**Figure S1**).²³ The calculations were carried out for all molecules using the B3LYP functional²⁴⁻²⁶ and 6-31G**²⁷ basis set with Grimme's D3 dispersion corrections and Becke-Johnson damping.²⁸ The effect of solvation in 1,1,2-trichloroethane were accounted for implicitly using the SMD polarizable continuum model.²⁹ Intrinsic reaction coordinate (IRC)³⁰ calculations were conducted to verify the critical reaction steps. The energetic results were then improved by the single-point calculations at the B3LYP+D3BJ/6-311++G**^{31,32}/SMD. For comparison, the performances of several popular DFT functionals (e.g., B3LYP, BP86,^{33,34} M06-2X³⁵), and basis sets (6-31G**, 6-311++G**, def2-SVP,³⁶ def2-TZVPP³⁷) were studied (see Table S2), implying that our method used in this study is reliable.

Method	$\Delta\Delta$ ‡G(TS1)
B3LYP+D3BJ/6-311++G**/SMD//B3LYP+D3BJ/6-31G**/SMD	23.1
B3LYP+D3BJ/def2-TZVPP/SMD//B3LYP+D3BJ/def2-SVP/SMD	25.8
M06-2X+D3/6-311++G**/SMD// M06-2X+D3/6-31G**/SMD	21.5
M06-2X+D3/def2-TZVPP/SMD// M06-2X+D3/def2-SVP/SMD	23.8
BP86+D3BJ/def2-TZVPP/SMD//BP86+D3BJ/def2-SVP/SMD	16.2

Table S2. Computed free energy barriers of transition states (in kcal/mol) using different levels of theories.





Figure S1. Computed structures of 1a, 2a, IM1, IM2, IM3, 3a, Part of the hydrogen atoms are omitted for clarity, selected bond distance is given in Å (color code, C: grey, N: blue, O: red, H: white, B: pink).

1	~
I	a

Gibbs Free Energy = -439.1123207

2a	
Cibbo Erros Errorary	710.96

Gibbs Free Energy = -719.86755

51005 11	ree Energy = -+3).1123207		В	-0.52682	-2.674	0.000023
N	0 2002	2 62078	0 24120	С	1.036393	-2.6709	-0.0016
N C	0.2092	-2.03976	-0.34139	С	1.772825	-3.67728	-0.6616
C	1.002466	-3.52237	-0.62533	С	1.770181	-1.66155	0.656814
0	1.897827	-4.26698	-0.87872	С	3.1666	-3.67041	-0.67714
С	-1.17831	-2.47521	-0.21291	Н	1 240632	-4 46841	-1 1809
С	-1.66898	-1.20895	0.153035	C III	3 164004	-1 66275	0 660267
С	-2.07806	-3.53923	-0.4366		1 225021	-1.00275	1 177259
С	-3.04962	-1.01292	0.291306	п	1.235931	-0.87257	1.1//238
Н	-0.96389	-0.3914	0.328009	C	3.865591	-2.66515	-0.004/1
С	-3.45195	-3.32346	-0.29351	Н	3.708465	-4.44891	-1.20646
н	-1 69153	-4 52397	-0 71799	Η	3.703864	-0.88205	1.197401
C C	3 96629	2 05061	0.070242	Н	4.951887	-2.66295	-0.0059
С П	-3.90029	-2.03901	0.070242	С	-1.31116	-1.3218	0.014774
п	-3.42209	-0.02391	0.378739	С	-2.55611	-1.20115	0.667664
H	-4.14306	-4.15582	-0.46/35	С	-0.80383	-0.17072	-0.62428
С	-5.45532	-1.83908	0.19723	С	-3.25217	0.006237	0.694479
Н	-5.93682	-1.78924	-0.79682	Н	-2.97545	-2.06628	1.172151
Η	-5.93976	-2.66388	0.747443	C	-1 50656	1 033097	-0.6253
Η	-5.67974	-0.89663	0.722186	ч	0 150202	0.22006	1 12004
				п	0.150395	-0.22600	-1.13904
				C	-2.73073	1.125401	0.041144

Н	-4.20129	0.075394	1.218158	С	-1.98469	-1.73483	-5.83461
Н	-1.10017	1.899723	-1.13879	Н	-1.75	-0.37519	-4.19393
Н	-3.276	2.064834	0.051101	С	-2.62183	-3.99366	-5.25779
С	-1.30576	-4.02931	-0.01315	Н	-2.853	-4.3997	-3.16533
С	-2.55177	-4.15475	-0.66311	С	-2.30935	-3.03589	-6.22534
С	-0.79229	-5.17851	0.624372	Н	-1.73621	-0.98806	-6.58309
С	-3.24304	-5.36491	-0.68861	Н	-2.88083	-5.0052	-5.55704
Н	-2.97577	-3.29117	-1.16636	Н	-2.31917	-3.30173	-7.2785
С	-1.49017	-6.38514	0.626696	Ν	-5.76601	-3.02427	-3.6293
Н	0.162927	-5.11744	1.136857	С	-5.69666	-2.11902	-4.421
С	-2.71553	-6.48216	-0.03687	Ο	-5.64555	-1.33089	-5.29974
Н	-4.19313	-5.43775	-1.21003	С	-5.75766	-3.26428	-2.25241
Н	-1.07906	-7.25028	1.138941	С	-5.98865	-2.23833	-1.32733
Н	-3.25706	-7.42377	-0.04582	С	-5.51316	-4.56563	-1.80373
				С	-5.9739	-2.5255	0.033971
IM1				Н	-6.16689	-1.22788	-1.67787
Gibbs Fi	ree Energy = -11	58.975603		С	-5.49541	-4.83235	-0.43759
				Н	-5.33011	-5.35107	-2.52872
В	-2.32189	-1.9472	-1.9693	С	-5.72962	-3.82285	0.504481
С	-1.91083	-2.99568	-0.88489	Н	-6.14386	-1.72174	0.744892
С	-0.97607	-4.01456	-1.1682	Н	-5.2881	-5.84267	-0.09694
С	-2.45389	-2.96771	0.416859	С	-5.72197	-4.12393	1.980615
С	-0.59935	-4.95044	-0.20671	Н	-6.72154	-4.41172	2.330459
Н	-0.5299	-4.06346	-2.15655	Н	-5.04485	-4.95048	2.213051
С	-2.09694	-3.91089	1.37922	Н	-5.41333	-3.25124	2.563575
Н	-3.18295	-2.20644	0.668397				
С	-1.16548	-4.90409	1.069628	TS1			
Н	0.131284	-5.71644	-0.45024	Gibbs	Free Energy = -115	58.943045	
Н	-2.54162	-3.87123	2.369064				
Н	-0.88009	-5.63671	1.819229				
С	-2.74761	-0.50139	-1.54705	В	-2.93889	-1.8858	-1.89246
С	-3.6832	0.238635	-2.30148	С	-1.92066	-2.36189	-0.73069
С	-2.22534	0.116163	-0.39148	С	-0.67574	-2.95077	-1.00978
С	-4.09767	1.510568	-1.91089	С	-2.25219	-2.20868	0.630148
Н	-4.09806	-0.19394	-3.20526	С	0.197697	-3.35004	0.004921
С	-2.61584	1.397938	-0.00596	Н	-0.38361	-3.11444	-2.0419
Н	-1.49227	-0.41507	0.207364	С	-1.38932	-2.60253	1.653264
С	-3.56193	2.094841	-0.76055	Н	-3.21252	-1.7782	0.902398
Н	-4.83144	2.048738	-2.50408	С	-0.15399	-3.17508	1.343545
Н	-2.18733	1.852183	0.882829	Н	1.151176	-3.80404	-0.25209
Н	-3.87653	3.089234	-0.45692	Н	-1.68332	-2.46882	2.690939
С	-2.30012	-2.33861	-3.48275	Н	0.521058	-3.48698	2.13543
С	-1.99339	-1.39276	-4.48354	С	-3.06535	-0.19676	-2.01701
С	-2.6018	-3.64962	-3.90761	С	-3.51524	0.39232	-3.23306

С	-2.52194	0.68119	-1.03968	С	1.145263	-0.63196	2.276991
С	-3.47047	1.763769	-3.43976	С	3.370377	-2.20935	1.706488
Н	-3.92178	-0.25418	-4.00447	Н	2.819962	-1.97406	-0.35489
С	-2.46763	2.052028	-1.24557	С	1.953213	-1.09095	3.313992
Н	-2.15133	0.264171	-0.10928	Н	0.275089	-0.02708	2.511875
С	-2.94968	2.594956	-2.44271	С	3.066007	-1.88801	3.03035
Н	-3.83429	2.190558	-4.36906	Н	4.232756	-2.83005	1.481295
Н	-2.05166	2.704065	-0.48385	Н	1.715472	-0.83498	4.342395
Н	-2.91399	3.668934	-2.6004	Н	3.692135	-2.25561	3.838231
С	-2.72112	-2.52678	-3.35082	С	-0.94381	2.069359	-1.08533
С	-1.57339	-2.1904	-4.09562	С	-1.34182	2.74855	-2.24361
С	-3.63682	-3.3977	-3.96253	С	0.032797	2.632127	-0.25305
С	-1.33616	-2.71446	-5.36568	С	-0.74202	3.95641	-2.58678
Н	-0.85678	-1.48826	-3.67607	Н	-2.10777	2.308895	-2.87309
С	-3.41307	-3.92719	-5.23739	С	0.618552	3.851486	-0.58849
Н	-4.54965	-3.67035	-3.44281	Н	0.325273	2.12633	0.661089
С	-2.25787	-3.59246	-5.9429	С	0.239938	4.509637	-1.75968
Н	-0.4382	-2.43324	-5.90953	Н	-1.03769	4.468132	-3.49739
Н	-4.14486	-4.59883	-5.6786	Н	1.369829	4.286135	0.06337
Н	-2.08028	-4.00169	-6.93338	Н	0.705181	5.453724	-2.02642
Ν	-4.44699	-2.04845	-1.36732	С	1.191834	-0.05743	-1.63079
С	-4.8184	-0.81322	-1.27938	С	2.492847	0.461181	-1.76801
0	-5.64218	-0.01864	-0.97412	С	0.472708	-0.33928	-2.80816
С	-5.20946	-3.15032	-0.88228	С	3.04265	0.714321	-3.02335
С	-6.53957	-3.01364	-0.47703	Н	3.071499	0.685938	-0.87701
С	-4.58273	-4.4003	-0.81713	С	1.024049	-0.11015	-4.06746
С	-7.23255	-4.12946	-0.00948	Н	-0.53113	-0.74868	-2.7356
Н	-7.03425	-2.05046	-0.52595	С	2.309188	0.425435	-4.17691
С	-5.29259	-5.50089	-0.3497	Н	4.041549	1.132908	-3.1051
Н	-3.55056	-4.50304	-1.12725	Н	0.453294	-0.34358	-4.96154
С	-6.62767	-5.38918	0.064634	Н	2.738399	0.615088	-5.15649
Н	-8.26673	-4.01431	0.302128	Ν	-0.86604	-0.16216	-0.0326
Н	-4.79685	-6.46663	-0.30378	С	-1.62861	0.78731	-0.74327
С	-7.37732	-6.59159	0.577738	Ο	-2.81023	0.607151	-1.00855
Н	-7.36083	-7.41029	-0.15034	С	-1.6099	-1.07927	0.785838
Н	-6.926	-6.97491	1.500434	С	-2.47538	-0.60456	1.777606
Н	-8.42172	-6.349	0.790915	С	-1.4436	-2.45351	0.621259
				С	-3.15217	-1.5009	2.596529
IM2				Н	-2.60591	0.464744	1.908174
Gibbs Fr	ee Energy = -115	9.010212		С	-2.11733	-3.34515	1.45706
				Н	-0.78223	-2.8238	-0.1557
В	0.565165	-0.36241	-0.23331	С	-2.98249	-2.88769	2.455634
С	1.434404	-0.93417	0.931003	Н	-3.81806	-1.12045	3.36665
С	2.56892	-1.72655	0.671973	Н	-1.97111	-4.41336	1.323818

С	-3.72301	-3.84603	3.353022	0	1.510135	1.980396	1.4249
Н	-3.41618	-4.87954	3.171635	С	-0.44648	-0.69082	2.86963
Н	-3.5441	-3.61957	4.410136	С	0.557029	-1.64681	3.080981
Н	-4.80572	-3.784	3.191273	С	-1.78708	-1.03345	3.057711
				С	0.206477	-2.93599	3.467846
IM3				Н	1.596441	-1.37535	2.928211
Gibbs Fr	ee Energy = -115	8.975056		С	-2.11899	-2.33009	3.44555
				Н	-2.5541	-0.28917	2.877311
В	0.32696	0.890331	-1.48047	С	-1.13457	-3.30271	3.658006
С	0.840315	2.217539	-2.12749	Н	0.98978	-3.67286	3.623864
С	-0.02815	3.053726	-2.85916	Н	-3.16479	-2.59125	3.580415
С	2.181443	2.634452	-1.99799	С	-1.49626	-4.69824	4.097286
С	0.420674	4.234194	-3.44976	Н	-0.90789	-5.45004	3.561324
Н	-1.06866	2.7649	-2.97416	Н	-1.29988	-4.83861	5.167516
С	2.631333	3.827434	-2.56114	Н	-2.55576	-4.90736	3.926221
Н	2.873443	2.021132	-1.42966				
С	1.752197	4.626733	-3.29595	TS2			
Н	-0.26603	4.851666	-4.02175	Gibbs	s Free Energy = -115	8.913464	
Н	3.665451	4.133628	-2.43152				
Н	2.102771	5.551913	-3.74449	В	-3.76829	1.555979	1.686063
С	-1.12009	0.832729	-0.89006	С	-3.24336	3.051493	1.507267
С	-1.8876	-0.35065	-0.90271	С	-3.70303	3.862333	0.45545
С	-1.69937	1.970409	-0.28944	С	-2.25168	3.583917	2.346559
С	-3.16622	-0.39726	-0.35023	С	-3.18916	5.142438	0.240258
Н	-1.47732	-1.24443	-1.3623	Н	-4.48429	3.488734	-0.20262
С	-2.96597	1.925052	0.290392	С	-1.73674	4.86625	2.146234
Н	-1.13341	2.896194	-0.25702	Н	-1.87942	2.986416	3.174652
С	-3.70536	0.740414	0.255522	С	-2.20255	5.64989	1.088482
Н	-3.73915	-1.31937	-0.38263	Н	-3.56245	5.746144	-0.58266
Н	-3.37932	2.810811	0.763979	Н	-0.97312	5.254658	2.8149
Н	-4.69715	0.703807	0.697208	Н	-1.80451	6.648105	0.929765
С	1.257687	-0.36385	-1.41107	С	-5.42281	1.39793	1.62646
С	2.287376	-0.57312	-2.35302	С	-6.0596	0.286283	1.003534
С	1.112663	-1.33147	-0.39417	С	-6.25786	2.476789	2.041639
С	3.122616	-1.68718	-2.28705	С	-7.43562	0.222696	0.878606
Н	2.425512	0.144948	-3.15527	Н	-5.44407	-0.53625	0.654244
С	1.958004	-2.43558	-0.30607	С	-7.63931	2.413114	1.921735
Н	0.329339	-1.21258	0.344271	Н	-5.79458	3.348375	2.493588
С	2.964446	-2.61805	-1.25719	С	-8.22672	1.280744	1.352229
Н	3.89934	-1.82892	-3.03303	Н	-7.9072	-0.63985	0.418579
Н	1.827871	-3.15248	0.499401	Н	-8.25953	3.233045	2.269032
Н	3.62113	-3.4814	-1.19829	Н	-9.30772	1.221956	1.264992
Ν	-0.14093	0.609465	2.457087	С	-2.97491	0.424166	0.880779
С	0.739554	1.241632	1.931659	С	-2.53289	0.661533	-0.43264

С	-2.72753	-0.85122	1.418148	Η	3.326387	-0.31791	0.474243
С	-1.88411	-0.32354	-1.17792	С	2.948041	-2.99755	2.524473
Н	-2.69236	1.640085	-0.8775	Н	1.110239	-4.08964	2.810754
С	-2.07372	-1.84301	0.683436	Н	4.616775	-1.71971	2.044684
Н	-3.04427	-1.07625	2.432794	Н	3.516583	-3.61667	3.212629
С	-1.65141	-1.5832	-0.62112	С	1.248907	0.368235	-1.39074
Н	-1.55403	-0.10718	-2.19049	С	0.660069	1.596792	-1.74757
Н	-1.8937	-2.8173	1.13004	С	2.383318	-0.04432	-2.11446
Н	-1.14271	-2.35169	-1.19621	С	1.183073	2.384957	-2.77184
Ν	-5.74057	0.291855	4.107843	Н	-0.21399	1.951276	-1.20486
С	-4.87873	0.813858	3.411954	С	2.903131	0.730129	-3.15178
0	-3.67589	1.17931	3.284782	Н	2.856748	-0.9902	-1.86881
С	-5.59849	-0.38914	5.335819	С	2.306413	1.949647	-3.4787
С	-4.35676	-0.55838	5.967859	Н	0.718108	3.334779	-3.01964
С	-6.75285	-0.91172	5.924601	Н	3.773195	0.385725	-3.70329
С	-4.28859	-1.24606	7.174634	Н	2.714582	2.558273	-4.28038
Н	-3.45824	-0.1496	5.517889	0	-0.72883	-0.53535	-0.06181
С	-6.667	-1.59614	7.135251	Н	-1.20573	-0.03429	-0.73956
Н	-7.70625	-0.77319	5.425561				
С	-5.43764	-1.77824	7.780306	3a			
Н	-3.32389	-1.37144	7.659088	Gibbs	Free Energy = -671	.3860386	
Н	-7.57077	-1.99613	7.58617				
С	-5.33934	-2.53658	9.078945	С	-1.00351	2.227066	-0.97432
Н	-4.87645	-3.51946	8.927405	С	-1.57168	3.120689	-1.89212
Н	-4.72155	-2.00055	9.807139	С	0.379501	2.265723	-0.74522
Н	-6.32559	-2.69903	9.521917	С	-0.7694	4.019365	-2.58985
				Н	-2.64456	3.094575	-2.04692
H_2O				С	1.180256	3.172542	-1.43867
Gibbs Fre	e Energy = -76.4	46416249		Н	0.83749	1.616674	-0.00469
				С	0.609352	4.045957	-2.36581
0	-0.79718	-0.20389	0	Н	-1.21794	4.700481	-3.30644
Н	0.168678	-0.14992	0	Н	2.248429	3.202004	-1.24754
Н	-1.06875	0.724623	0	Н	1.235191	4.749727	-2.90606
				Ν	-1.36043	0.143768	0.211128
Ph ₂ BOH				С	-1.93497	1.29681	-0.25382
Gibbs Fre	e Energy = -564	.1051826		0	-3.12851	1.570585	-0.11598
				С	-1.9468	-0.89824	0.955017
В	0.627514	-0.51062	-0.24842	С	-3.27863	-0.89637	1.392287
С	1.471392	-1.39197	0.731609	С	-1.12524	-1.98984	1.279082
С	0.86836	-2.46418	1.417678	С	-3.75887	-1.97552	2.134787
С	2.834916	-1.14514	0.977253	Н	-3.91862	-0.06111	1.150057
С	1.594528	-3.26299	2.298714	С	-1.62468	-3.05526	2.017966
Н	-0.1848	-2.67047	1.250338	Н	-0.08894	-1.99296	0.951173
С	3.566901	-1.93337	1.865843	С	-2.95501	-3.07287	2.460836

Н	-4.79314	-1.95792	2.468401
Н	-0.96832	-3.88772	2.257533
С	-3.49929	-4.24441	3.237837
Н	-3.75021	-5.08069	2.572942
Н	-2.76834	-4.61856	3.962087
Н	-4.40925	-3.97383	3.780985
Н	-0.40098	-0.023	-0.05985

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 $^{13}\mathrm{C}$ NMR spectrum for compound 3a (CDCl_3)





¹³C NMR spectrum for compound **3b** (CDCl₃)

- 2.49



 $^1\mathrm{H}$ NMR spectrum for compound $3c~(\mathrm{CDCl}_3)$



 ^{13}C NMR spectrum for compound 3c (CDCl_3)







¹³C NMR spectrum for compound **3d** (CDCl₃)







 ^{13}C NMR spectrum for compound 3e (CDCl_3)



¹³C NMR spectrum for compound **3f** (CDCl₃)







^{0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 -220} f1 (ppm)

 ^{19}F NMR spectrum for compound 3g (CDCl_3)





 $^1\mathrm{H}$ NMR spectrum for compound **3h** (CDCl₃)







¹H NMR spectrum for compound **3**j (CDCl₃)











 $^1\mathrm{H}\,\mathrm{NMR}$ spectrum for compound 3m (CDCl_3)



 $^{13}\mathrm{C}$ NMR spectrum for compound 3m (CDCl_3)



¹H NMR spectrum for compound **3n** (DMSO- d_6)



¹³C NMR spectrum for compound **3n** (DMSO- d_6)





90 80 fl (ppm)



¹³C NMR spectrum for compound **3p** (CD₃OD)



¹H NMR spectrum for compound **3q** (CDCl₃)



 ^{19}F NMR spectrum for compound 3q (CDCl_3)



¹³C NMR spectrum for compound **3q** (CDCl₃)







¹³C NMR spectrum for compound 3r (DMSO- d_6)



¹H NMR spectrum for compound **3s** (CDCl₃)



 $^{13}\mathrm{C}$ NMR spectrum for compound 3s (CDCl₃)

803 803 804 805







 ^{13}C NMR spectrum for compound **3t** (CDCl₃)



¹H NMR spectrum for compound **3u** (CDCl₃)



 ^{13}C NMR spectrum for compound $\boldsymbol{3u}~(\text{CDCl}_3)$



¹H NMR spectrum for compound **3v** (CD₃OD)



 ^{13}C NMR spectrum for compound $3v\,(\text{CD}_3\text{OD})$



 ^{13}C NMR spectrum for compound 3w (CDCl_3)

90 80 fl (ppm)


¹H NMR spectrum for compound 3x (DMSO- d_6)



¹³C NMR spectrum for compound 3x (DMSO- d_6)







 ^{13}C NMR spectrum for compound $3y~(\text{CDCl}_3)$



 ^{13}C NMR spectrum for compound $3z\,(\text{CDCl}_3)$



 $^{13}\mathrm{C}$ NMR spectrum for compound $\textbf{3aa}~(\mathrm{CDCl}_3)$



 $^{13}\mathrm{C}$ NMR spectrum for compound $\boldsymbol{3ab}$ (CDCl_3)





 $^{13}\mathrm{C}$ NMR spectrum for compound 3ac (CDCl_3)





 $^{13}\mathrm{C}$ NMR spectrum for compound $\boldsymbol{3ad}~(\mathrm{CDCl}_3)$



 $^1\mathrm{H}$ NMR spectrum for compound $3ae~(\mathrm{CDCl}_3)$



 ^{13}C NMR spectrum for compound $\textbf{3ae}~(\text{CDCl}_3)$



 $^{13}\mathrm{C}$ NMR spectrum for compound $3af\,(\mathrm{CDCl}_3)$



¹H NMR spectrum for compound **3ag** (CDCl₃)



 $^{13}\mathrm{C}$ NMR spectrum for compound $\boldsymbol{3ag}~(\mathrm{CDCl}_3)$



 $^1\mathrm{H}\,\mathrm{NMR}$ spectrum for compound $\mathbf{3ah}\;\mathrm{(CDCl_3)}$



¹³C NMR spectrum for compound **3ah** (CDCl₃)



¹³C NMR spectrum for compound **3ai** (CDCl₃)



¹³C NMR spectrum for compound **3aj** (CDCl₃)





¹³C NMR spectrum for compound **3ak** (CDCl₃)