

Electronic Supplementary Information (ESI)

**Computational Studies on Functionalized Janus MXenes
 $\text{MM}'\text{CT}_2$, ($\text{M, M}'=\text{Zr, Ti, Hf}$, $\text{M} \neq \text{M}'$; $\text{T} = -\text{O, -F, -OH}$):
Photoelectronic Properties and Potential Photocatalytic
Activities**

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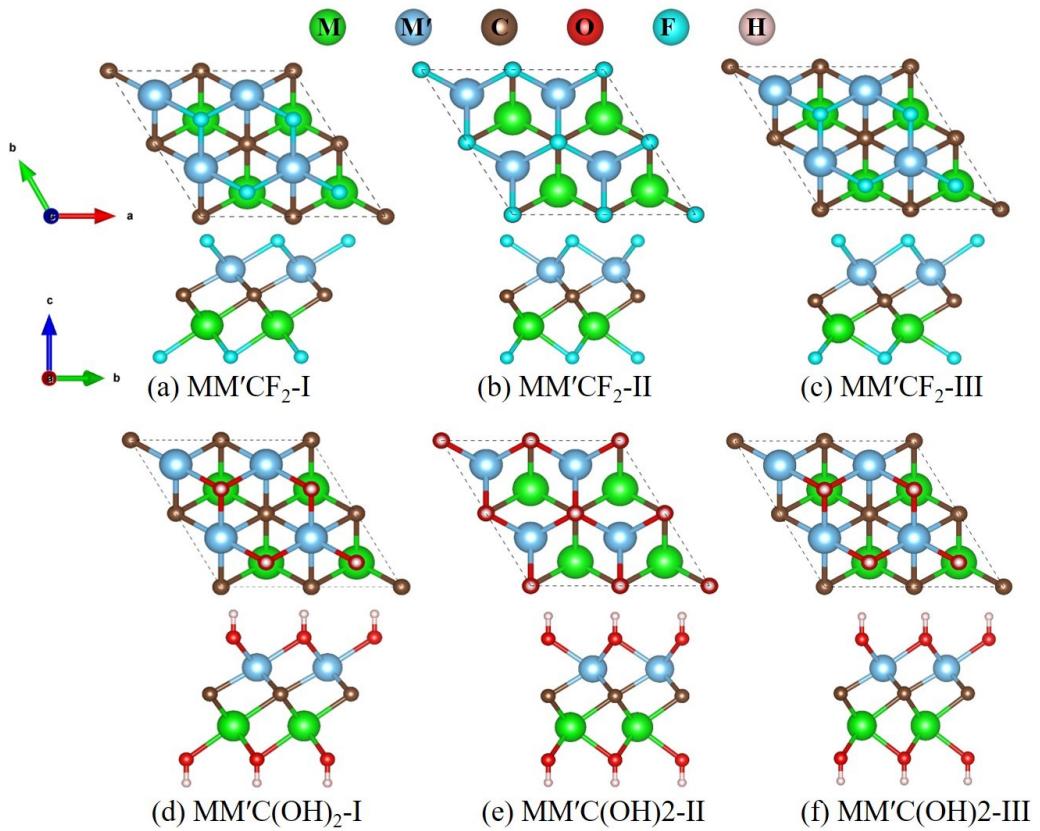


Fig. S1 Top view (top) and side view (bottom) of geometries of (a–c) $\text{MM}'\text{CF}_2$ -I, II, III; (d–f) $\text{MM}'\text{C}(\text{OH})_2$ -I, II, III. $\text{M}, \text{M}' = \text{Zr}, \text{Ti}, \text{Hf}; \text{M} \neq \text{M}'$.

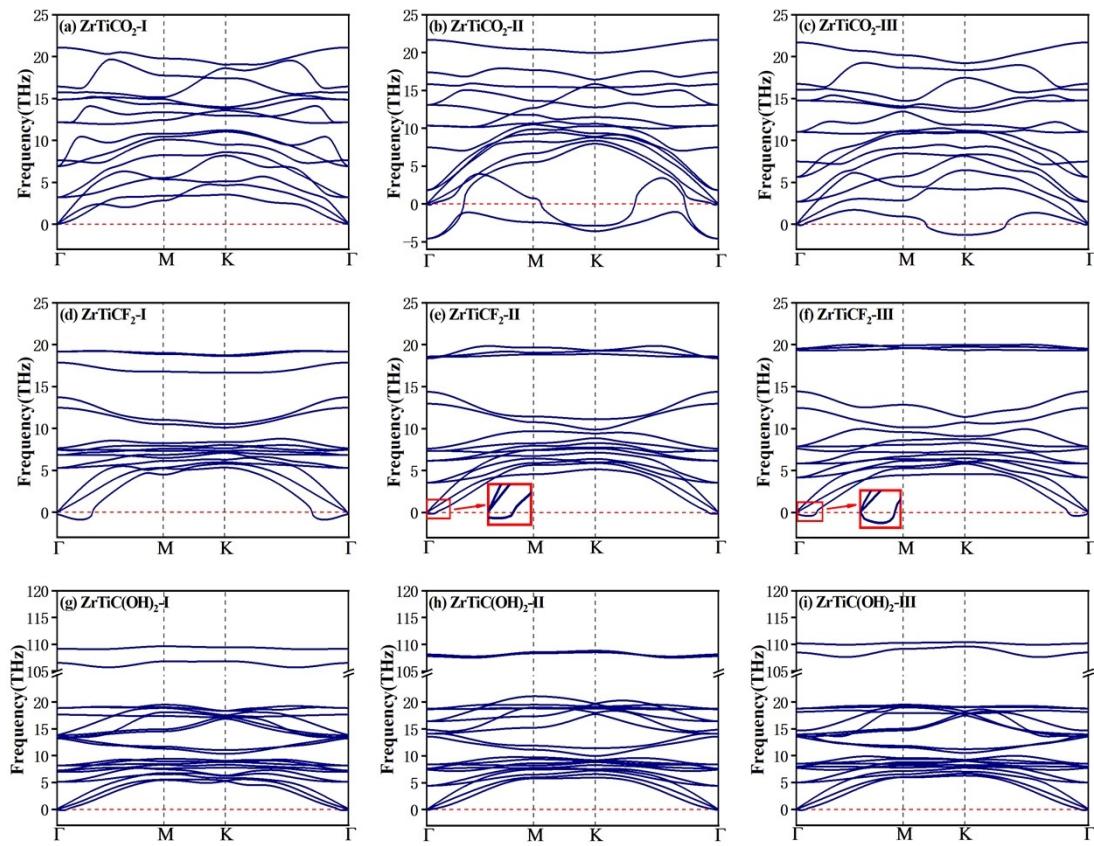


Fig. S2 Phonon dispersion of surface-functionalized ZrTiC MXenes with ZrTiCT₂-I, ZrTiCT₂-II, and ZrTiCT₂-III structures (T = -O, -F, and -OH). The insets in (e) and (f) display magnified images of the acoustic branches around the Γ -point.

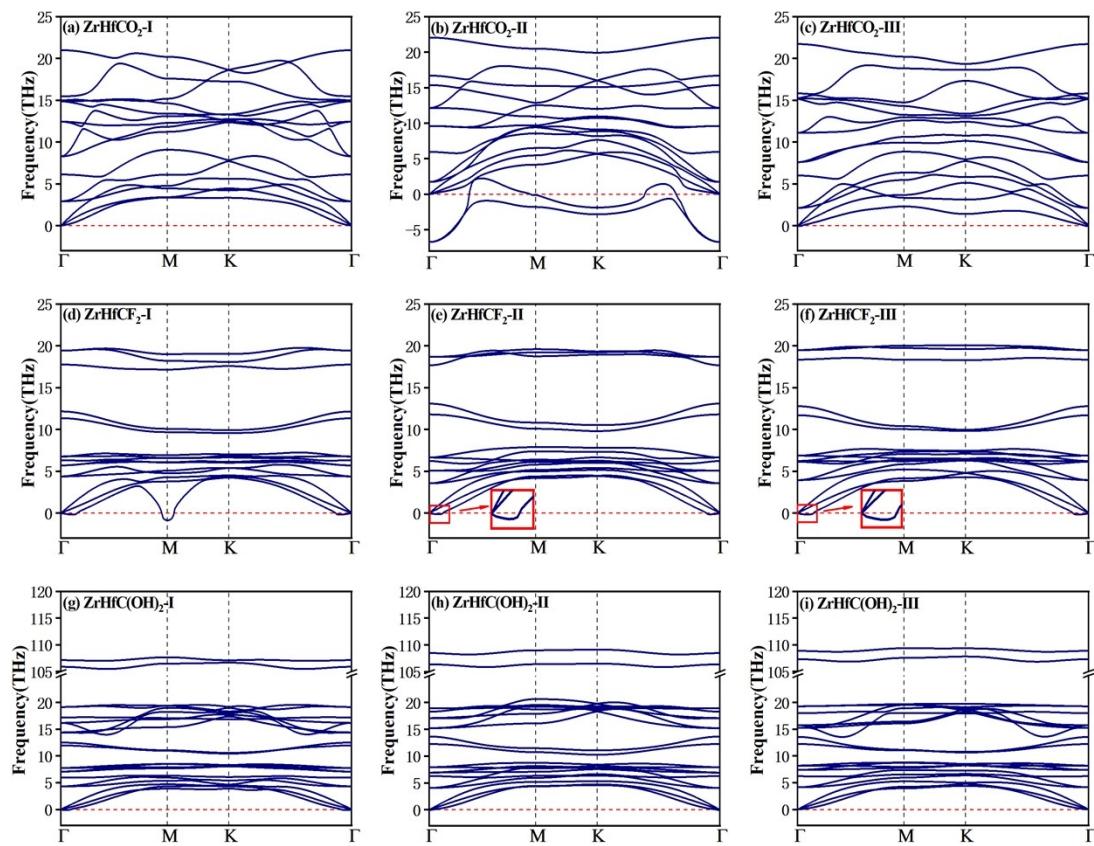


Fig. S3 Phonon dispersion of surface-functionalized ZrHfC MXenes with ZrHfCT₂-I, ZrHfCT₂-II, and ZrHfCT₂-III structures (T = -O, -F, and -OH). The insets in (e) and (f) display magnified images of the acoustic branches around the Γ -point.

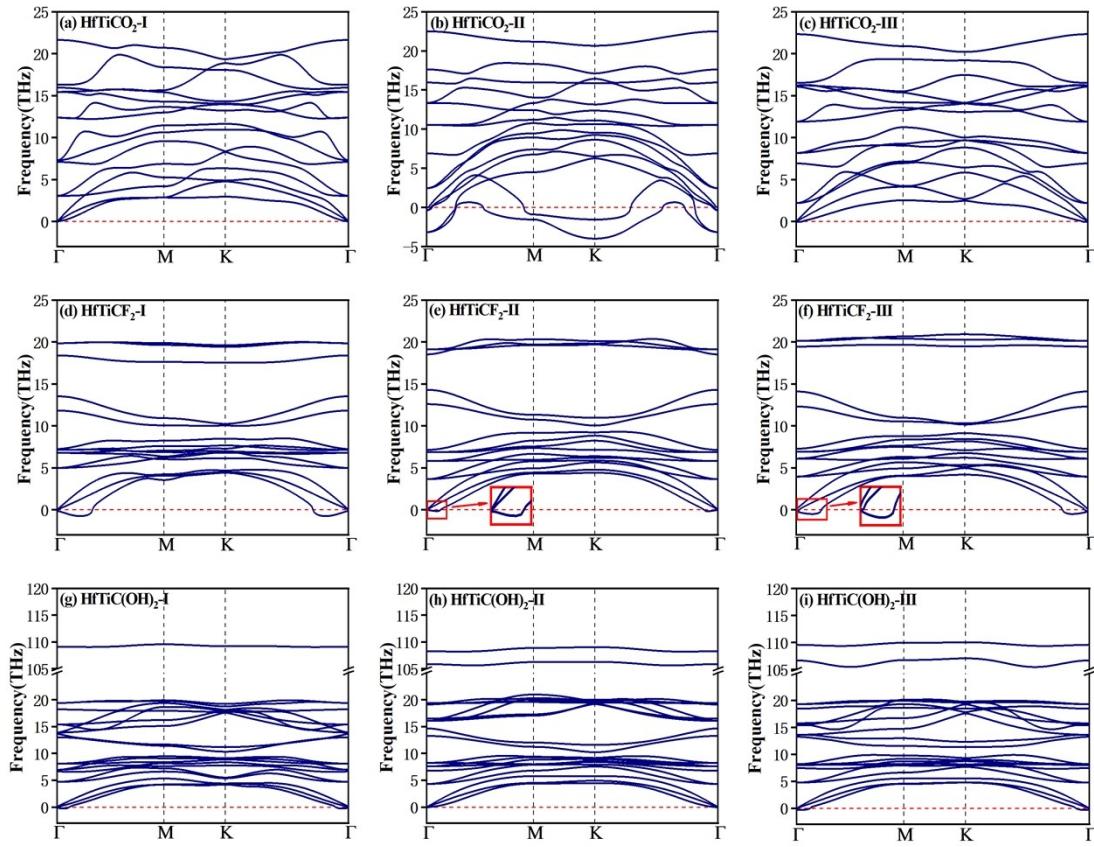


Fig. S4 Phonon dispersion of surface-functionalized HfTiC MXenes with HfTiCT₂-I, HfTiCT₂-II, and HfTiCT₂-III structures (T = -O, -F, and -OH). The insets in (e) and (f) display magnified images of the acoustic branches around the Γ -point.

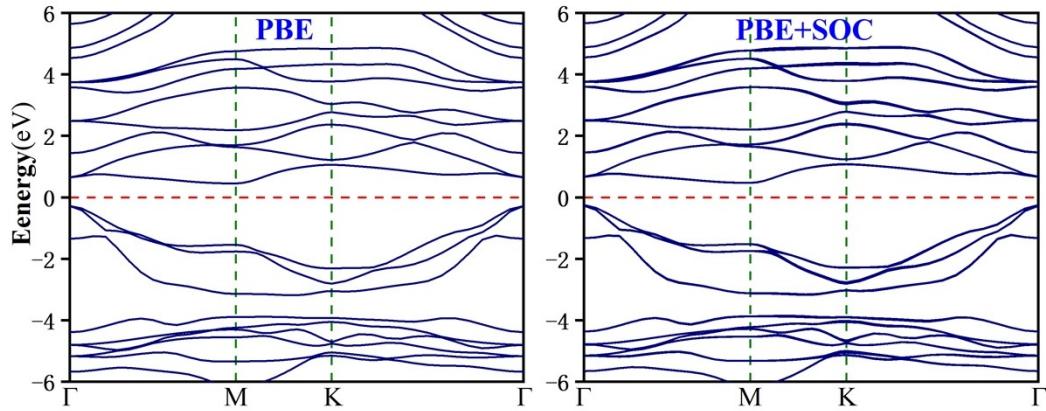


Fig. S5 Band structures of ZrTiCO₂-I with PBE and PBE+SOC methods. The red dashed lines mark the Fermi level at 0 eV.

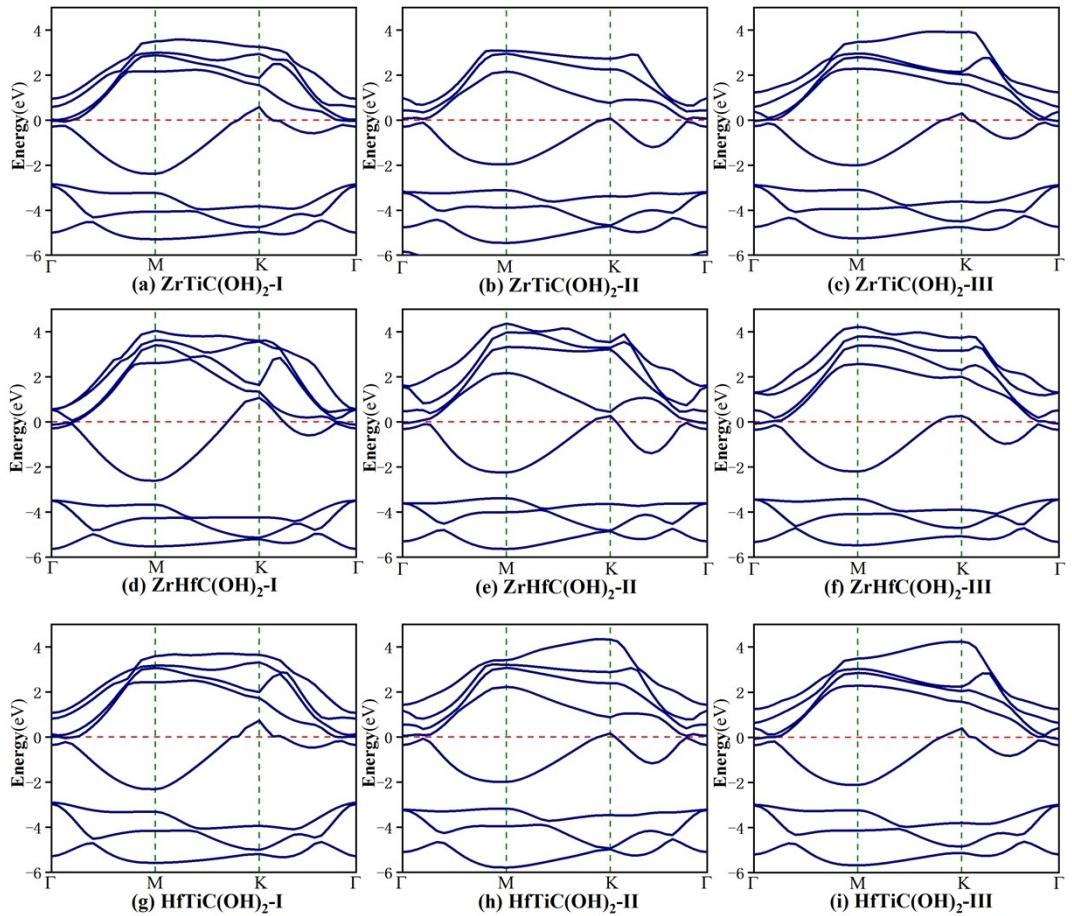


Fig. S6 Band structures of (a) $\text{ZrTiC(OH)}_2\text{-I}$, (b) $\text{ZrTiC(OH)}_2\text{-II}$, (c) $\text{ZrTiC(OH)}_2\text{-III}$, (d) $\text{ZrHfC(OH)}_2\text{-I}$, (e) $\text{ZrHfC(OH)}_2\text{-II}$, (f) $\text{ZrHfC(OH)}_2\text{-III}$, (g) $\text{HfTiC(OH)}_2\text{-I}$, (h) $\text{HfTiC(OH)}_2\text{-II}$, and (i) $\text{HfTiC(OH)}_2\text{-III}$. The red dashed lines mark the Fermi level at 0 eV.

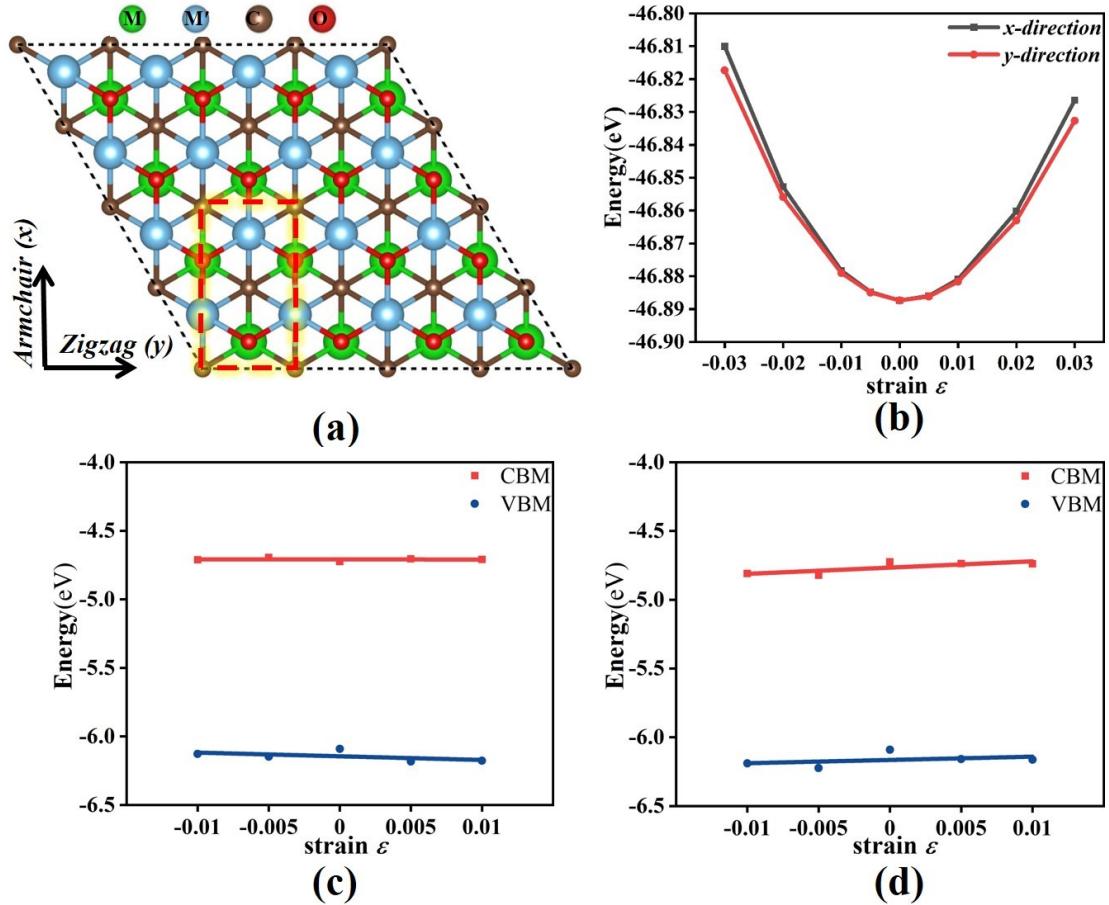


Fig. S7 (a) ZrTiCO₂-I Janus monolayer; the black and red dashed lines represent a 4×4 hexagonal supercell and the orthogonal primitive cell, respectively. (b) Relationship between total energy and uniaxial strain along the *x* and *y* directions. (c,d) Shift of band edge position with respect to the uniaxial strain along the (c) *x*- and, (d) *y*-direction.

Details of Carrier Mobility Calculations

As an example, we describe the calculations of the electron carrier mobility along the *x* direction of the ZrTiCO₂-I Janus monolayer. To obtain the elastic modulus *C* and deformation potential constant *E_I*, we calculated the total energy and band edge positions as a function of the uniaxial strain *ε*, as shown in Fig. S6(b,c). The calculated results are listed in Table S4.

To calculate the effective mass, we plotted the relationship between the energy *E(k)* and the wave vector *k* with 30 high-density *k*-points near the CBM, then employed a second-order polynomial function to fit the curve:

$$E(k) = Intercept + B_1 k + B_2 k^2$$

The second derivative of the fitting function is:

$$\left. \frac{\partial^2 E(k)}{\partial k^2} \right|_{k=M(0.5,0,0)} = 2B_2$$

According to the definition of carrier effective mass, the electron effective mass along the x -direction can be calculated as:

$$m^* = \hbar^2 / (\partial^2 E(k) / \partial k^2) = \hbar^2 / 2B_2$$

Table S1 Cohesive energies (E_{coh} , eV/atom) of functionalized Janus MXenes.

| | ZrTiCT ₂ | | | ZrHfCT ₂ | | | HfTiCT ₂ | | |
|--------------|---------------------|------|------|---------------------|------|------|---------------------|------|------|
| | T=O | T=F | T=OH | T=O | T=F | T=OH | T=O | T=F | T=OH |
| Geometry I | 7.59 | 6.88 | 5.99 | 8.14 | 7.25 | 6.28 | 7.78 | 6.99 | 6.09 |
| Geometry II | 7.20 | 6.79 | 5.95 | 7.71 | 7.19 | 6.25 | 7.38 | 6.90 | 6.05 |
| Geometry III | 7.42 | 6.82 | 6.00 | 7.96 | 7.26 | 6.29 | 7.60 | 6.94 | 6.08 |

Table S2 Band gaps (E_g , eV) of MM'CT₂ with differently functionalized geometries calculated at the HSE06 level. The M label denotes metallic properties of the corresponding MM'CT₂ material.

| | ZrTiCT ₂ | | | ZrHfCT ₂ | | | HfTiCT ₂ | | |
|-------------------------|---------------------|------|------|---------------------|------|------|---------------------|------|------|
| | T=O | T=F | T=OH | T=O | T=F | T=OH | T=O | T=F | T=OH |
| MM'CT ₂ -I | 1.37 | none | M | 1.71 | none | M | 1.43 | none | M |
| MM'CT ₂ -II | none | none | M | none | none | M | none | none | M |
| MM'CT ₂ -III | none | none | M | 1.77 | none | M | 1.94 | none | M |

Table S3 Vacuum level, CBM and VBM levels (eV) of ZrTiCO₂-I, ZrHfCO₂-I, ZrHfCO₂-III, HfTiCO₂-I, and ZrHfCO₂-III. E_{CBM} and E_{VBM} denote band edge positions, whose value are listed relative to the Fermi level, vacuum level and normal hydrogen electrode ($E_{NHE} = -4.5$ eV), respectively.

| | E_{Vacuum} | E_{Fermi} | E_{CBM} | | | E_{VBM} | | |
|--------------------------|--------------|-------------|--------------|---------------|------------|--------------|---------------|------------|
| | | | vs. Fermi | vs. vacuum | vs. NHE | vs. Fermi | vs. vacuum | vs. NHE |
| ZrTiCO ₂ -I | 4.24 | -1.77 | 1.29 | -4.73 | 0.23 | -0.08 | -6.09 | 1.59 |
| ZrHfCO ₂ -I | 3.99 | -1.61 | 1.59 | -4.01 | -0.49 | -0.12 | -5.72 | 1.22 |
| ZrHfCO ₂ -III | 4.21 | -2.47 | 1.52 | -5.16 | 0.66 | -0.25 | -6.93 | 2.43 |
| HfTiCO ₂ -I | 3.69 | -2.19 | 1.24 | -4.65 | 0.15 | -0.19 | -6.08 | 1.58 |
| HfTiCO ₂ -III | 3.92 | -3.35 | 1.83 | -5.43 | 0.93 | -0.10 | -7.37 | 2.87 |

Table S4 Effective mass $|m^*|$, deformation potential constant $|E_I|$, in-plane stiffness C , and carrier-mobility μ for electrons and holes along the x and y directions.

| | | C (N/m) | $ E_I $ (eV) | $ m^* (m_e)$ | μ (cm 2 ·V $^{-1}$ ·s $^{-1}$) |
|--------------------------|------------------|-----------|--------------|---------------|--|
| ZrTiCO ₂ -I | Electron (x) | 280 | 0.08 | 16.45 | 2296 |
| | Hole (x) | 280 | 2.68 | 0.16 | 21621 |
| | Electron (y) | 253 | 4.56 | 4.21 | 10 |
| | Hole (y) | 253 | 2.38 | 1.08 | 544 |
| ZrHfCO ₂ -I | Electron (x) | 289 | 7 | 1.92 | 23 |
| | Hole (x) | 289 | 3.84 | 0.18 | 8589 |
| | Electron (y) | 258 | 3.08 | 2.31 | 72 |
| | Hole (y) | 258 | 1.18 | 1.15 | 1989 |
| ZrHfCO ₂ -III | Electron (x) | 267 | 11.34 | 2.76 | 4 |
| | Hole (x) | 267 | 9.82 | 0.30 | 437 |
| | Electron (y) | 240 | 2.42 | 2.65 | 83 |
| | Hole (y) | 240 | 0.52 | 2.16 | 2701 |
| HfTiCO ₂ -I | Electron (x) | 267 | 3.58 | 2.99 | 33 |
| | Hole (x) | 267 | 0.58 | 13.23 | 64 |
| | Electron (y) | 240 | 8.04 | 3.93 | 3 |
| | Hole (y) | 240 | 10.22 | 1.16 | 24 |
| HfTiCO ₂ -III | Electron (x) | 262 | 4.02 | 2.81 | 29 |
| | Hole (x) | 262 | 6.3 | 0.69 | 197 |
| | Electron (y) | 234 | 0.16 | 2.57 | 19649 |
| | Hole (y) | 234 | 1.88 | 1.46 | 441 |