

Degradable Fe₃O₄-Based Nanocomposite for Cascade Reaction-Enhanced Anti-Tumor Therapy

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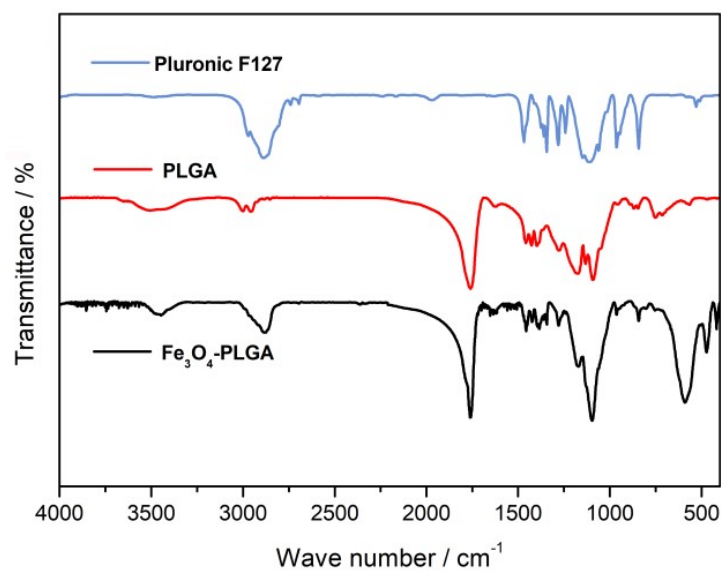


Fig. S1 FTIR spectra of Fe₃O₄-PLGA, PLGA and F127

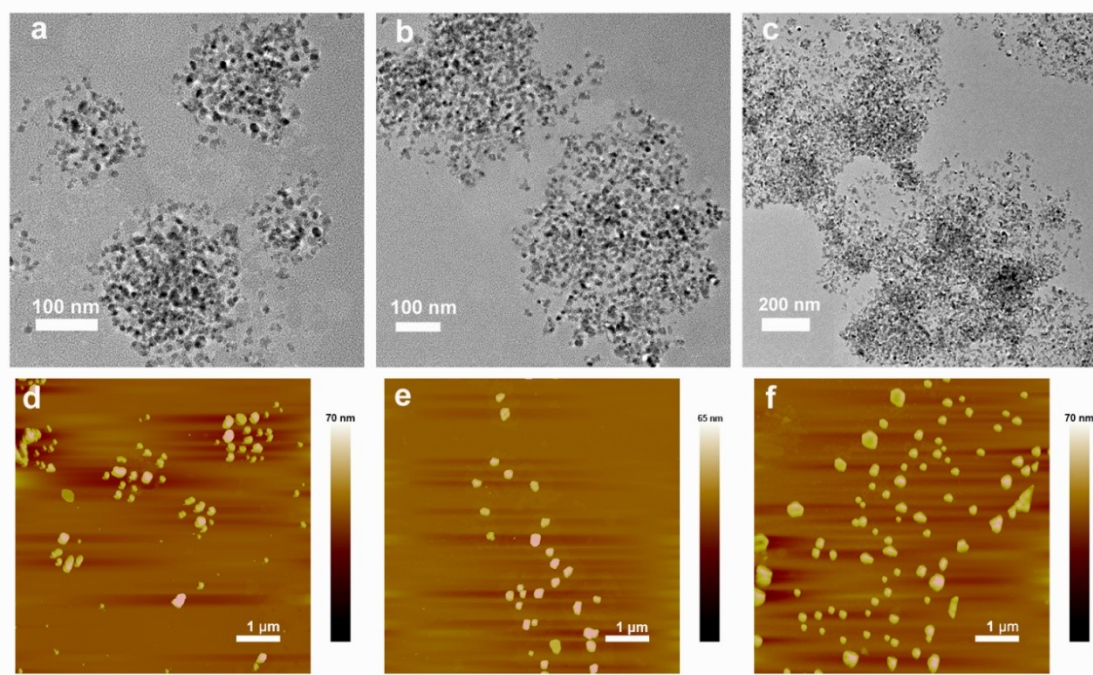


Fig. S2 TEM images (a-c) and AFM images (d-f) of Fe₃O₄-PLGA nanocomposites
with different PLGA/Fe₃O₄ mass ratio (2:1(a, d), 3:1(b, e) 4:1(c, f))

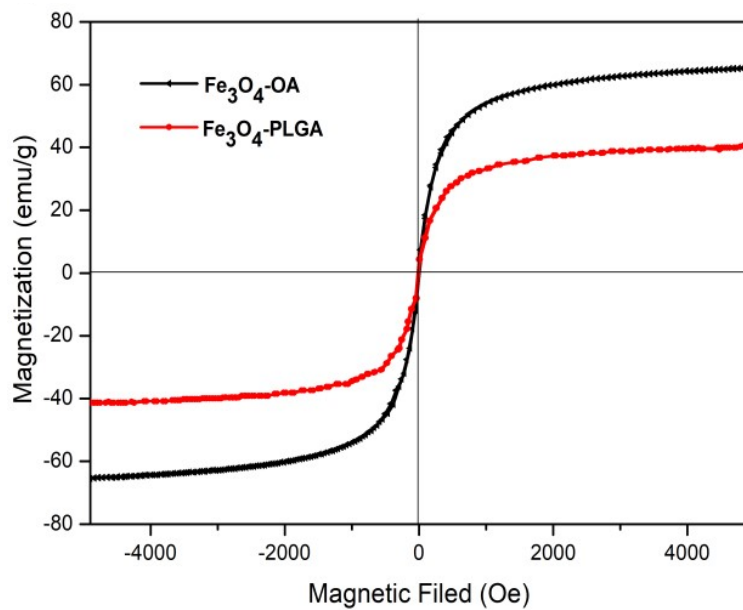


Fig. S3 VSM for Fe₃O₄-OA and Fe₃O₄-PLGA

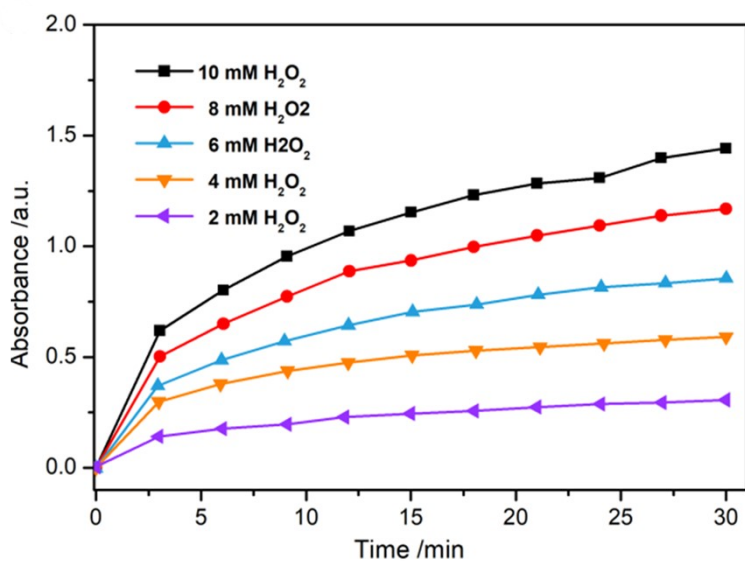


Fig. S4 Time-course absorbance of Fe₃O₄-PLGA-H₂O₂-TMB systems upon the addition of varied concentrations of H₂O₂

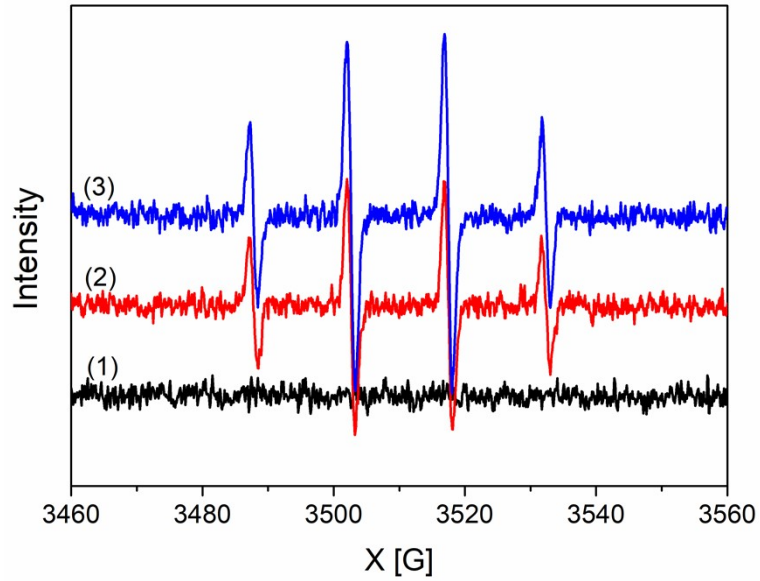


Fig. S5 EPR spectra of $\bullet\text{OH}$ trapped by DMPO under different conditions: (1) control, (2) Fe_3O_4 -PLGA nanocomposite with 10 mM H_2O_2 at pH 5.0, (3) Fe_3O_4 /GOx-PLGA nanocomposite with 5 mM β -D-glucose at pH 5.0

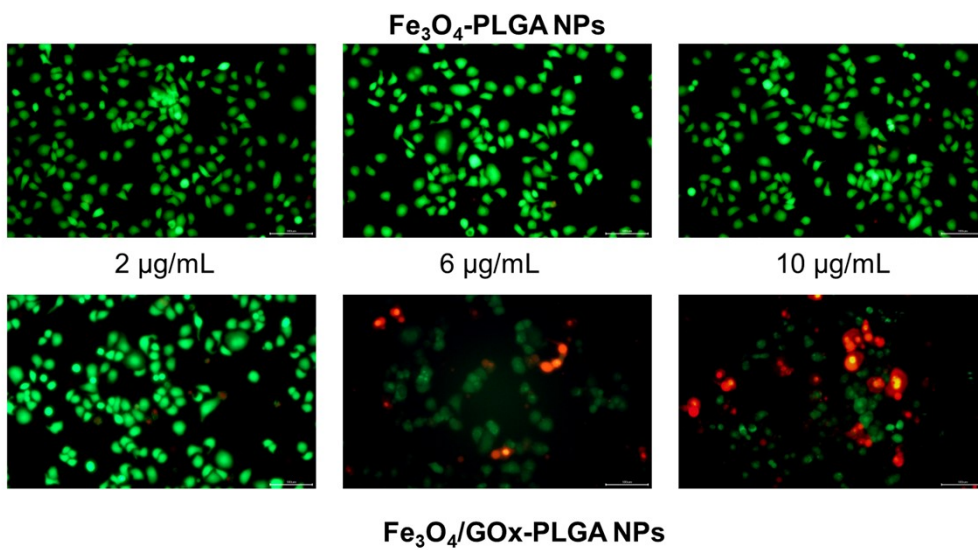


Fig. S6 Fluorescence images of viable and dead cell distributions after separate incubation with Fe_3O_4 -PLGA and Fe_3O_4 /GOx-PLGA nanocomposites under neutral (pH = 7.4) at varied concentrations for 6 h

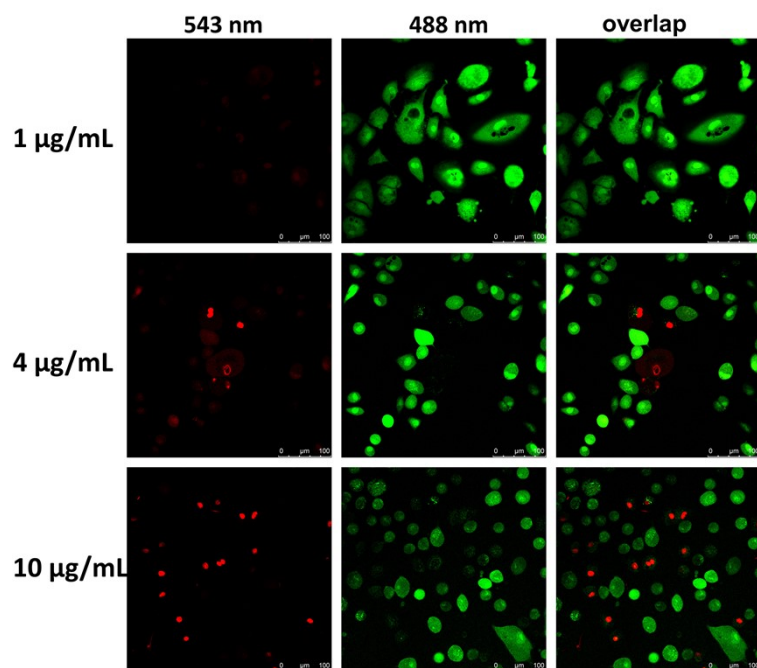


Fig. S7 CLSM images of HeLa cells after co-incubation with $\text{Fe}_3\text{O}_4/\text{GOx-PLGA}$ nanocomposite under neutral ($\text{pH} = 7.4$) conditions at varied concentrations for 6 h

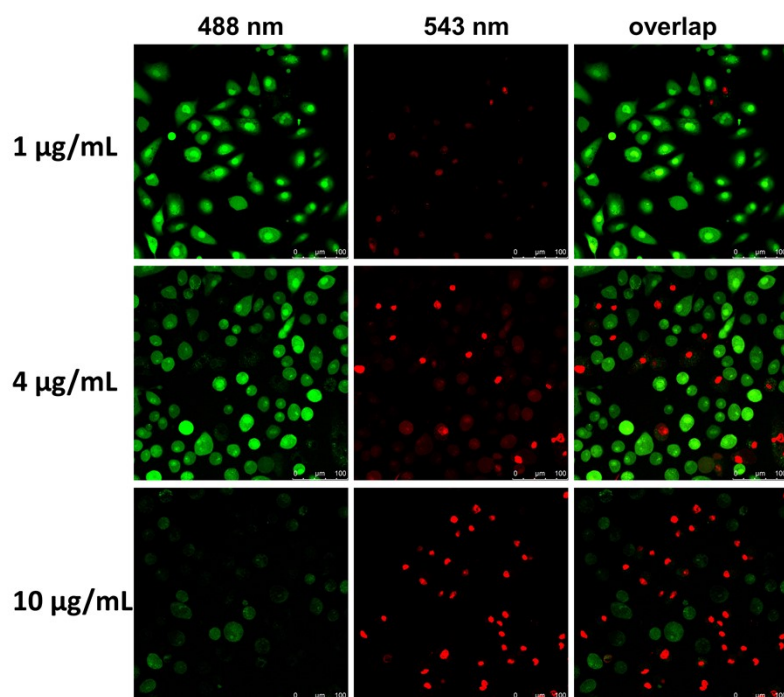


Fig. S8 CLSM images of HeLa cells after co-incubation with $\text{Fe}_3\text{O}_4/\text{GOx-PLGA}$ nanocomposite under acidic ($\text{pH} = 6.0$) conditions at varied concentrations for 6 h