## Supporting Information



Fig. S1 TC degradation images of  $CoMn_2O_4(P)$  prepared by NaBH<sub>4</sub> at different conditions: (a) temperature, (b) concentration, (c) time; (d) TC degradation images of  $CoMn_2O_4(C)$  prepared by NaH<sub>2</sub>PO<sub>2</sub> at different conditions.



**Fig. S2** SEM images of (a)  $CoMn_2O_4(P)$ , (b)  $CoMn_2O_4(S)$ , (c)  $CoMn_2O_4(C)$  and TEM images of (d)  $CoMn_2O_4(P)$ , (e)  $CoMn_2O_4(S)$ , (f)  $CoMn_2O_4(C)$ 



Fig. S3 Preparation of  $CoMn_2O_4$  spinel activated PDS for tetracycline degradation by reduction of NaBH<sub>4</sub> under different conditions:(a) Temperature (b) Concentration (c) Reaction time. Reaction conditions: [catalyst]= 0.25 g, [PMS]= 10 mL, 0.1 mol L<sup>-1</sup>, [TC] = 250 mL, 50 mg L<sup>-1</sup>, [temperature] = 30 °C.



Fig. S4 (a)~(i): The m/z value of  $CoMn_2O_4(P)$  after 30 minutes of activating PDS to degrade tetracycline.



Fig. S5 (a)~(i) The m/z value of  $CoMn_2O_4(S)$  after 30 minutes of activating PDS to degrade tetracycline.



Fig. S6 (a)~(g) The m/z value of  $CoMn_2O_4(C)$  after 30 minutes of activating PDS to degrade tetracycline.



Fig. S7 (a)~(h) The m/z value of CoMn<sub>2</sub>O<sub>4</sub>(P) after 120 minutes of activating PDS to degrade tetracycline.



Fig. S8 (a)~(h) The m/z value of CoMn<sub>2</sub>O<sub>4</sub>(S) after 120 minutes of activating PDS to degrade tetracycline.



Fig. S9 (a)~(h) The m/z value of  $CoMn_2O_4(C)$  after 120 minutes of activating PDS to degrade tetracycline.

Sample	SSA (m²/g)	V <sub>total</sub> (cm <sup>3</sup> /g)	V <sub>micro</sub> (cm <sup>3</sup> /g)	V <sub>meso</sub> (cm <sup>3</sup> /g)	Pore size (nm)
CoMn <sub>2</sub> O <sub>4</sub> (P)	79.40	0.248	0.003	0.245	11.094
$CoMn_2O_4(S)$	24.24	0.098	0.001	0.097	17.095
$CoMn_2O_4(C)$	242.52	0.694	0.000	0.694	26.105

Table S1 The pore properties of catalysts

Sample -	Co 2p		Mn 2p			O 1s	
	Co <sup>2+</sup>	Co <sup>3+</sup>	Mn <sup>2+</sup>	Mn <sup>3+</sup>	Mn <sup>4+</sup>	O <sub>A</sub>	$O_L$
CoMn <sub>2</sub> O <sub>4</sub> (P)	18.0%	32.9%	16.0%	24.7%	17.6%	28.8%	71.2%
CoMn <sub>2</sub> O <sub>4</sub> (S)	17.5%	16.0%	11.8%	35.3%	19.8%	55.1%	44.9%
$CoMn_2O_4(C)$	14.9%	24.3%	24.2%	25.8%	14.4%	38.2%	61.8%

Table S2 XPS valence content of the synthesized catalyst before use

				5	2		
Commite	Co 2p		Mn 2p			O 1s	
Sample	Co <sup>2+</sup>	Co <sup>3+</sup>	Mn <sup>2+</sup>	Mn <sup>3+</sup>	Mn <sup>4+</sup>	O <sub>A</sub>	O <sub>L</sub>
CoMn <sub>2</sub> O <sub>4</sub> (P)	27.5%	19.0%	19.6%	22.2%	25.4%	31.7%	68.3%
CoMn <sub>2</sub> O <sub>4</sub> (S)	18.3%	18.4%	11.9%	37.5%	17.5%	49.2%	50.8%
CoMn <sub>2</sub> O <sub>4</sub> (C)	19.2%	22.7%	15.7%	26.2%	13.5%	25.8%	74.2%

Table S3 XPS valence content of the synthesized catalysts after use

No.	weight	intermediates	Samples					
TC	445	2 H <sub>3</sub> C N CH <sub>3</sub> 4 H <sub>0</sub> CH <sub>3</sub> 5 4 OH 9 10 11 12 12 NH <sub>2</sub> 1 OH O OH 3 O O	TC					
P1	397	CH3 +OH H OH OH OH OH	$\begin{array}{c} \text{CoMn}_2\text{O}_4 \ (\text{P}) & (30\text{min} \\ 120\text{min}) \\ \\ \text{CoMn}_2\text{O}_4 \ (\text{S}) & (30\text{min} \\ 120\text{min}) \\ \\ \\ \text{CoMn}_2\text{O}_4 \ (\text{C}) & (30\text{min} \\ 120\text{min}) \end{array}$					
P2	354	CH <sub>3</sub> +OH H OH OH OH OH OH	$CoMn_2O_4 (P) (120min) \\ CoMn_2O_4 (S) (30min \\ 120min) \\ CoMn_2O_4 (C) (120min) \\ CoMn_2O_4 (C) (30min) \\ CoMn_2O_4 (P) (30min) \\ CoMn_2O_4 (P) (20min) \\ CoMn_2O_4 (P) (P) (20min) \\ CoMn_2O_4 (P) (P) (P) (20min) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) (P) (P) \\ CoMn_2O_4 (P) (P) (P) (P) (P) (P) (P) (P) (P) (P)$					
Р3	340 (1)		$\begin{array}{c} \text{CoMn}_2\text{O}_4 \ (\text{F}) & (30\text{min} \\ 120\text{min}) \\ \text{CoMn}_2\text{O}_4 \ (\text{S}) & (30\text{min} \\ 120\text{min}) \\ \text{CoMn}_2\text{O}_4 \ (\text{C}) & (30\text{min} \\ 120\text{min}) \end{array}$					
P4	340 (2)	CH3 OH OH OH	$CoMn_2O_4$ (P) (30min)					
Р5	274	CH <sub>2</sub> OH OH O OH	$\begin{array}{c} CoMn_2O_4 \ (P) \ (30min \\ 120min) \\ \\ CoMn_2O_4 \ (S) \ (30min \\ 120min) \\ \\ CoMn_2O_4 \ (C) \ (30min \\ 120min) \end{array}$					
P6	183	OH	$\begin{array}{c} \text{CoMn}_2\text{O}_4 \ (\text{P}) & (30\text{min} \\ 120\text{min}) \\ \\ \text{CoMn}_2\text{O}_4 \ (\text{S}) & (30\text{min} \\ 120\text{min}) \\ \\ \\ \text{CoMn}_2\text{O}_4 \ (\text{C}) & (30\text{min} \\ 120\text{min}) \end{array}$					

## Table S4 Identification of CoMn<sub>2</sub>O<sub>4</sub>(S), CoMn<sub>2</sub>O<sub>4</sub>(P), CoMn<sub>2</sub>O<sub>4</sub>(C) activated PDS to degrade TC and its possible intermediates under visible light irradiation

			$CoMn_2O_4$ (P) (30min
D7		сн₂ 	120min)
	246		$CoMn_2O_4$ (S) (30min
Γ/	240		120min)
		 он о он	$CoMn_2O_4$ (C) (30min
			120min)
			$CoMn_2O_4$ (P) (30min
Р8		сн <sub>а</sub>	120min)
	202		$CoMn_2O_4$ (S) (30min
	283		120min)
		``o он о	$CoMn_2O_4$ (C) (30min
			120min)
Р9		~	$CoMn_2O_4$ (P) (30min
			120min)
	140	Соон	$CoMn_2O_4$ (S) (30min
	149		120min)
			$CoMn_2O_4$ (C) (30min
			120min)