

## Constructing Interface Engineering and Tailoring Nanoflower-like FeP/CoP

### Heterostructure Enhance Oxygen Evolution Reaction

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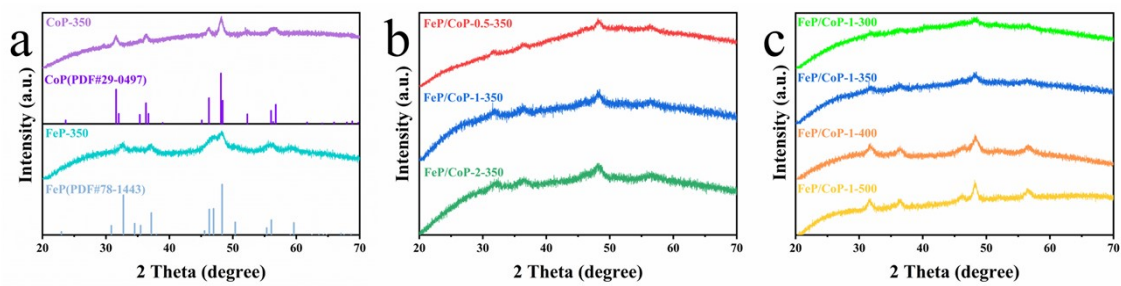
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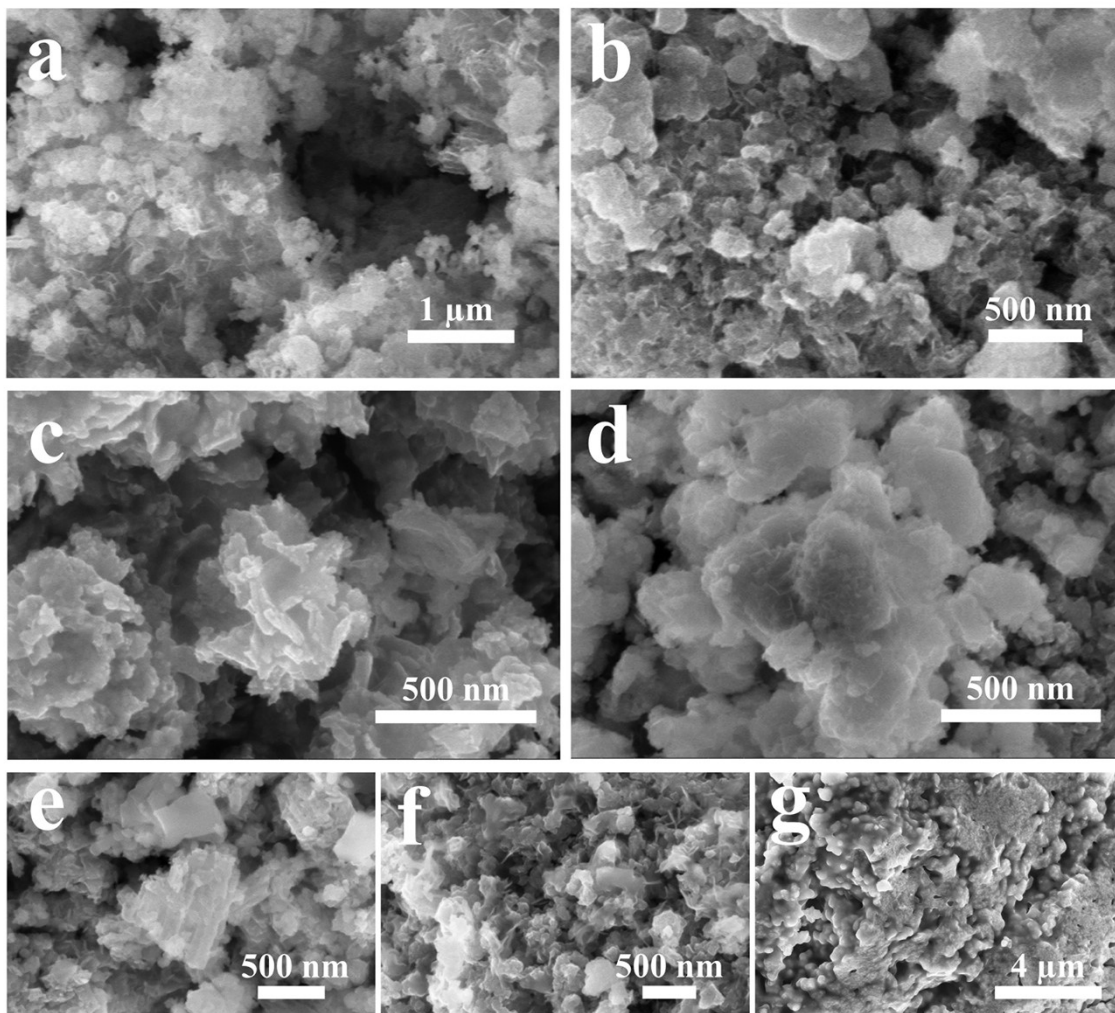
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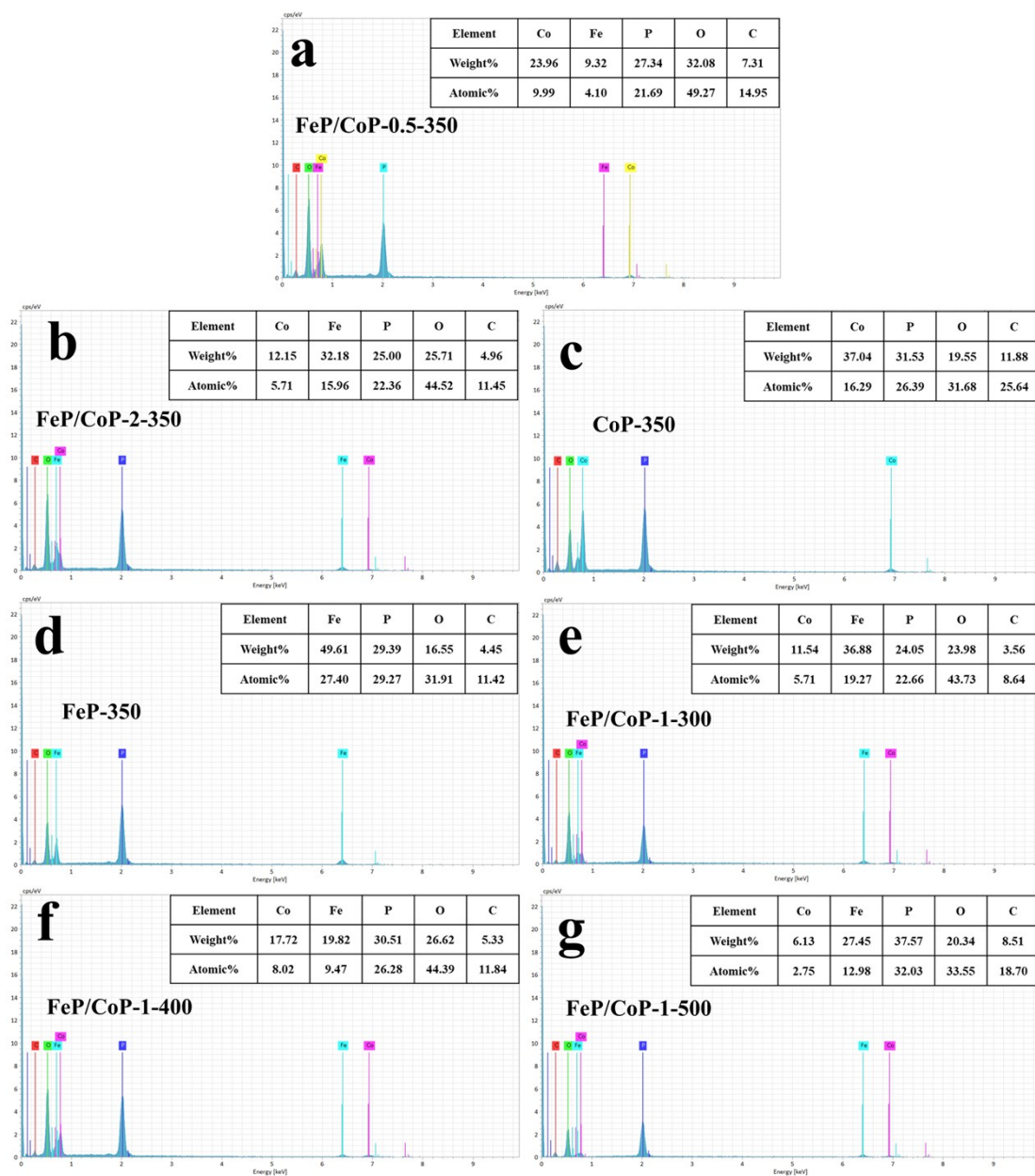
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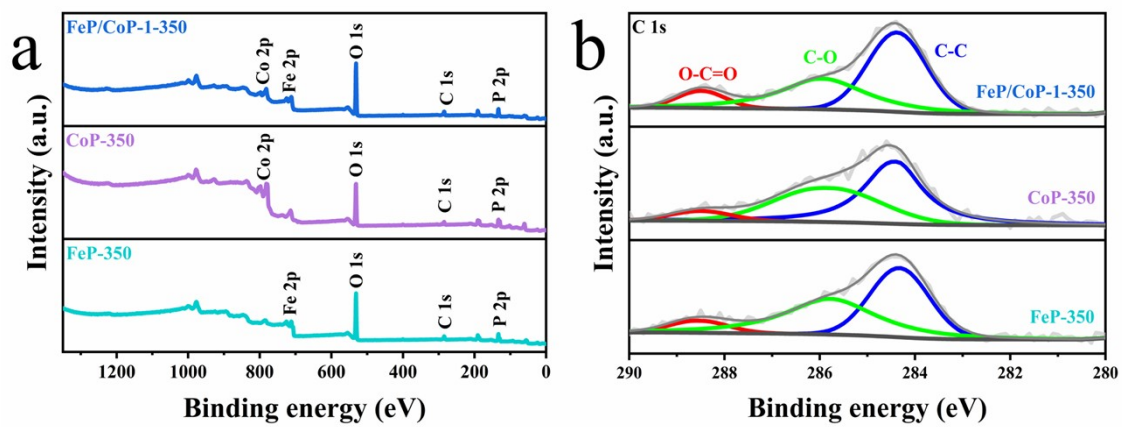
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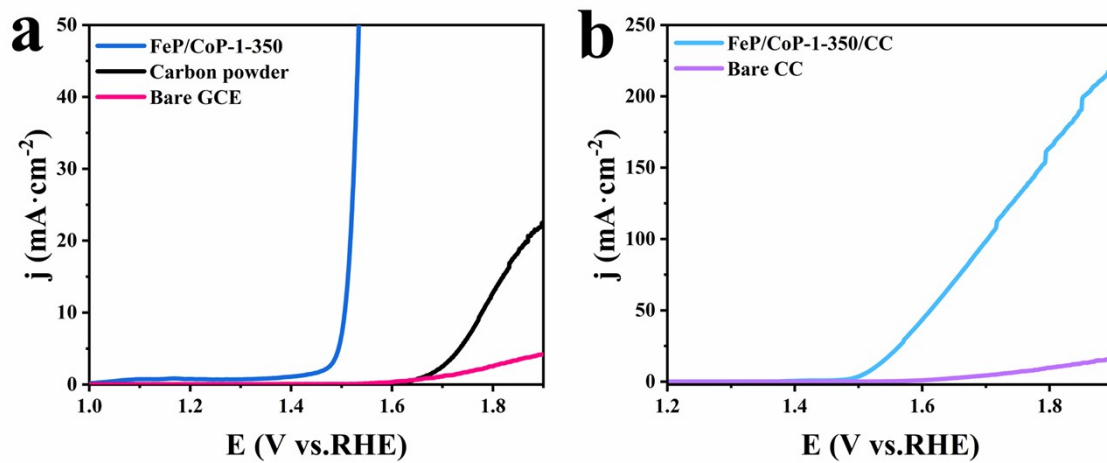


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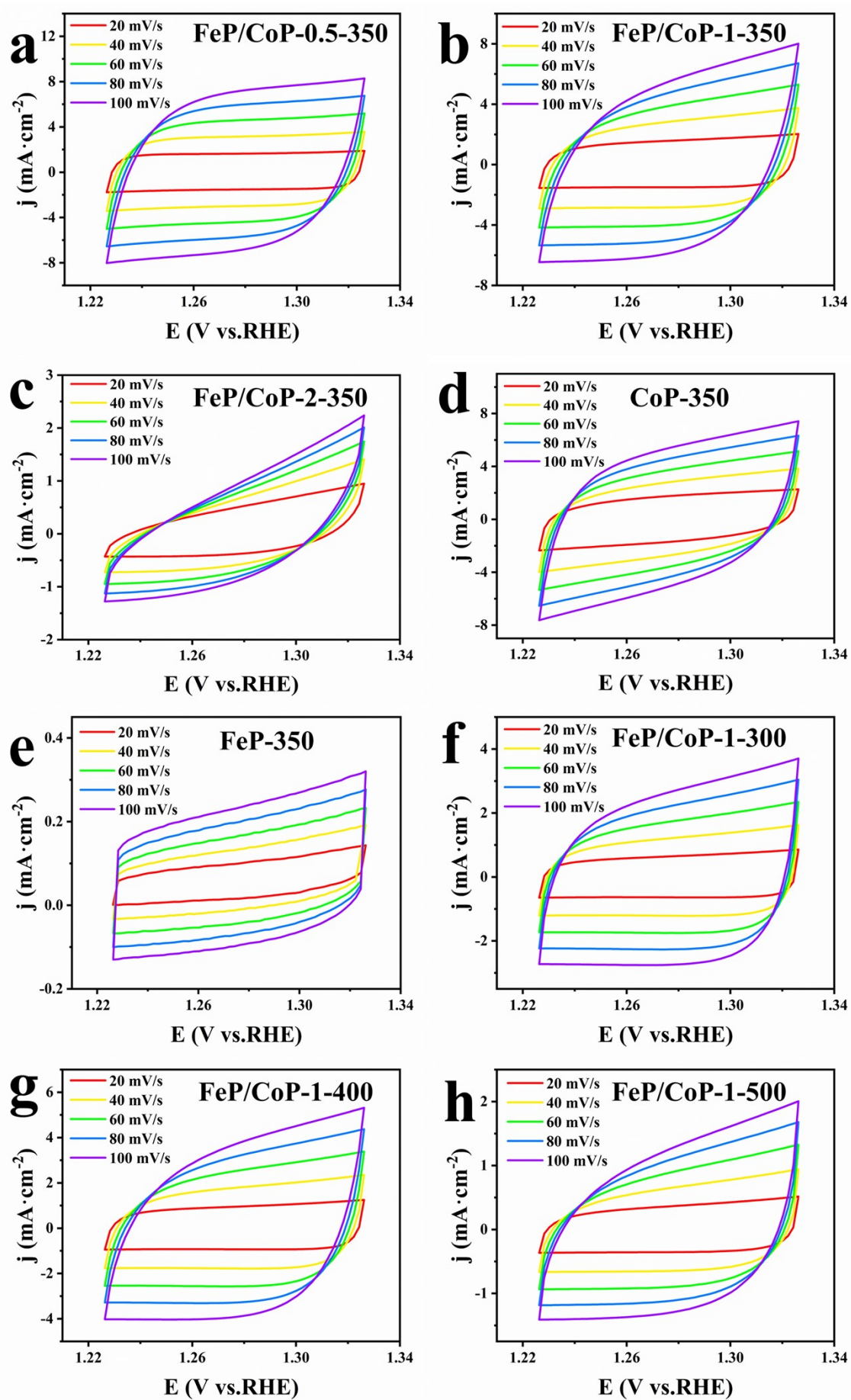


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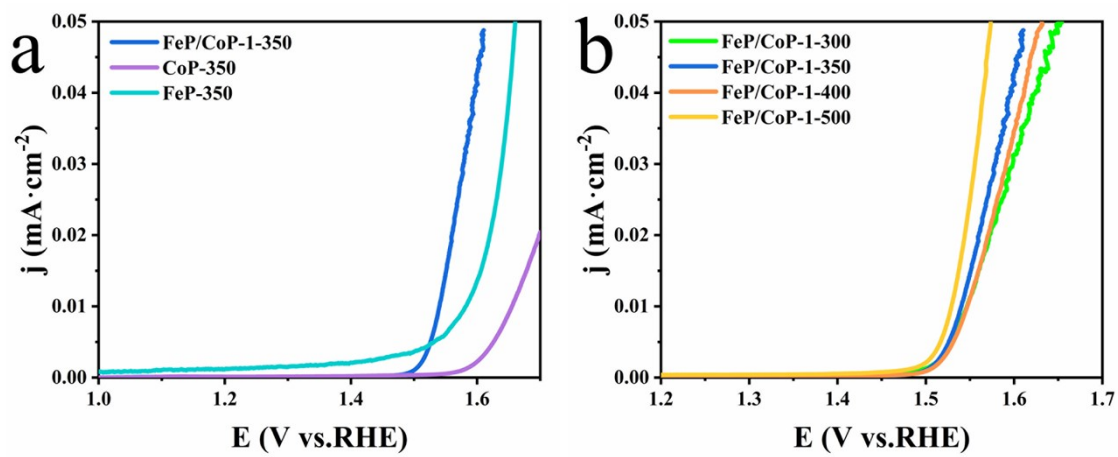
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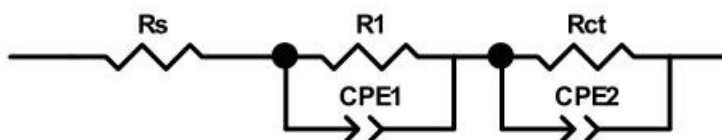


**Figure S7.** LSV curves normalized by ECSA.



**Table S1.** The values of charge transfer resistance ( $R_{ct}$ ) and resistance ( $R_s$ ) for different samples.

	$R_s$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )
FeP/CoP-0.5-350	5.206	10.080
FeP/CoP-1-350	4.924	8.857
FeP/CoP-2-350	5.047	12.220
CoP-350	4.949	15.980
FeP-350	5.388	25.080
FeP/CoP-1-300	5.011	9.500
FeP/CoP-1-400	4.813	9.647
FeP/CoP-1-500	4.855	11.77



In the simplified equivalent electrical circuit,  $R_s$  is the overall series resistance,  $CPE_1$  and  $R_1$  are the constant phase element and resistance representing electron transport at CoFe-P catalyst/glassy carbon interface and between CoFe-P catalyst, respectively.  $CPE_2$  is the constant phase element of the CoFe-P catalyst/electrolyte interface, and  $R_{ct}$  is the charge transfer resistance at CoFe-P catalyst/electrolyte interface related to the OER electrocatalysis process.

**Table S2.** The values of ECSA for different samples.

	ECSA (cm <sup>2</sup> )
FeP/CoP-0.5-350	1709
FeP/CoP-1-350	1292.25
FeP/CoP-2-350	144
CoP-350	1084.5
FeP-350	38.25
FeP/CoP-1-300	642
FeP/CoP-1-400	896.75
FeP/CoP-1-500	286.25

**Table S3.** The comparison of OER performance between this work and other reported transition metal phosphide catalysts.

Electrocatalysts	Overpotential (mV)	Tafel slop ( $\text{mV} \cdot \text{dec}^{-1}$ )	Reference
FeP/CoP-1-350	276	37.71	This Work
NiCoFeP hollow nanoprism	294	50.5	S1
Cu-CoP nanosheets	411	101.4	S2
P@pCoPc/Co <sub>3</sub> O <sub>4</sub> nanosheets	320	57.4	S3
CoP/NCNHP	310	70	S4
NiCoFeP films	300	124	S5
CoFeP	350	59	S6
Ce <sub>0.5</sub> -CoP	365	96	S7
CoP/rGO	340	66	S8
CoP NFs	323	49.6	S9
Hollow Mo-CoP nanoboxes	305	56	S10
CoP-TiOx	337	72.1	S11
Ni <sub>0.6</sub> Co <sub>1.4</sub> P	300	80	S12

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