

Supplementary Information

Neutron Diffraction and DFT Studies of Oxygen Defect and Transport in Higher-Order Ruddlesden-Popper Phase Materials

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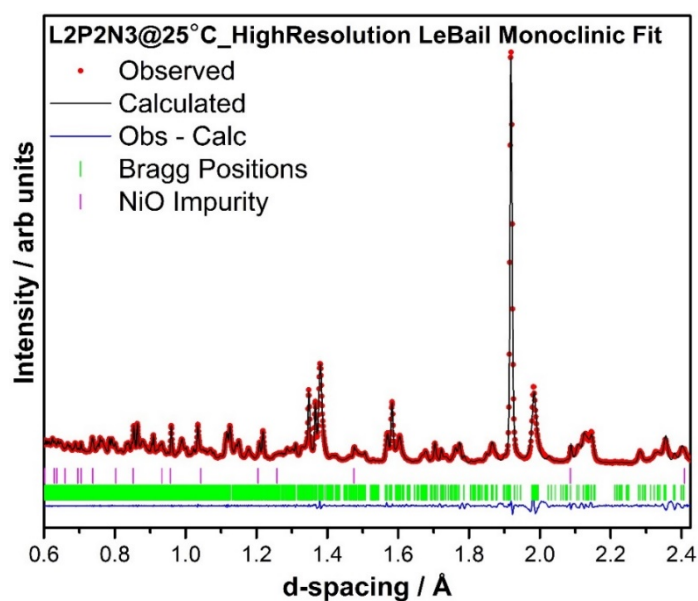


Figure S1. Le Bail refinement of the room-temperature high-resolution (bank 5) L2P2N3 data set in monoclinic $P12_1/a1$ space group ($\chi^2 = 4.10$; $R_p = 1.7\%$ and $wR_p = 1.1\%$).

Table S1. Fractional atomic coordinates and thermal parameters obtained after Rietveld refinement of

A	x	y	z	Uiso
La/Pr(1)	0.0	-0.0043(2)	0.4328(2)	0.008(3)
La/Pr(2)	0.0	-0.0122(2)	0.2990(2)	0.013(6)
Ni(1)	0.0	0.0	0.0	0.008 (1)
Ni(2)	0.0	-0.0015(1)	0.1380(1)	0.009(8)
O(1)	0.25	0.25	0.0067(7)	
O(2)	0.0	0.9549(1)	0.0687(3)	
O(3)	0.25	0.25	0.135(3)	
O(4)	0.0	0.0377(2)	0.2134(3)	
O(5)	0.25	0.75	0.1457(3)	

L3P1N3 ND patterns recorded at 250 °C using orthorhombic model (*Bmab*).

Table S2. Fractional atomic coordinates and thermal parameters obtained after Rietveld refinement of L3P1N3 ND patterns recorded at 800 °C using tetragonal model (*I4/mmm*).

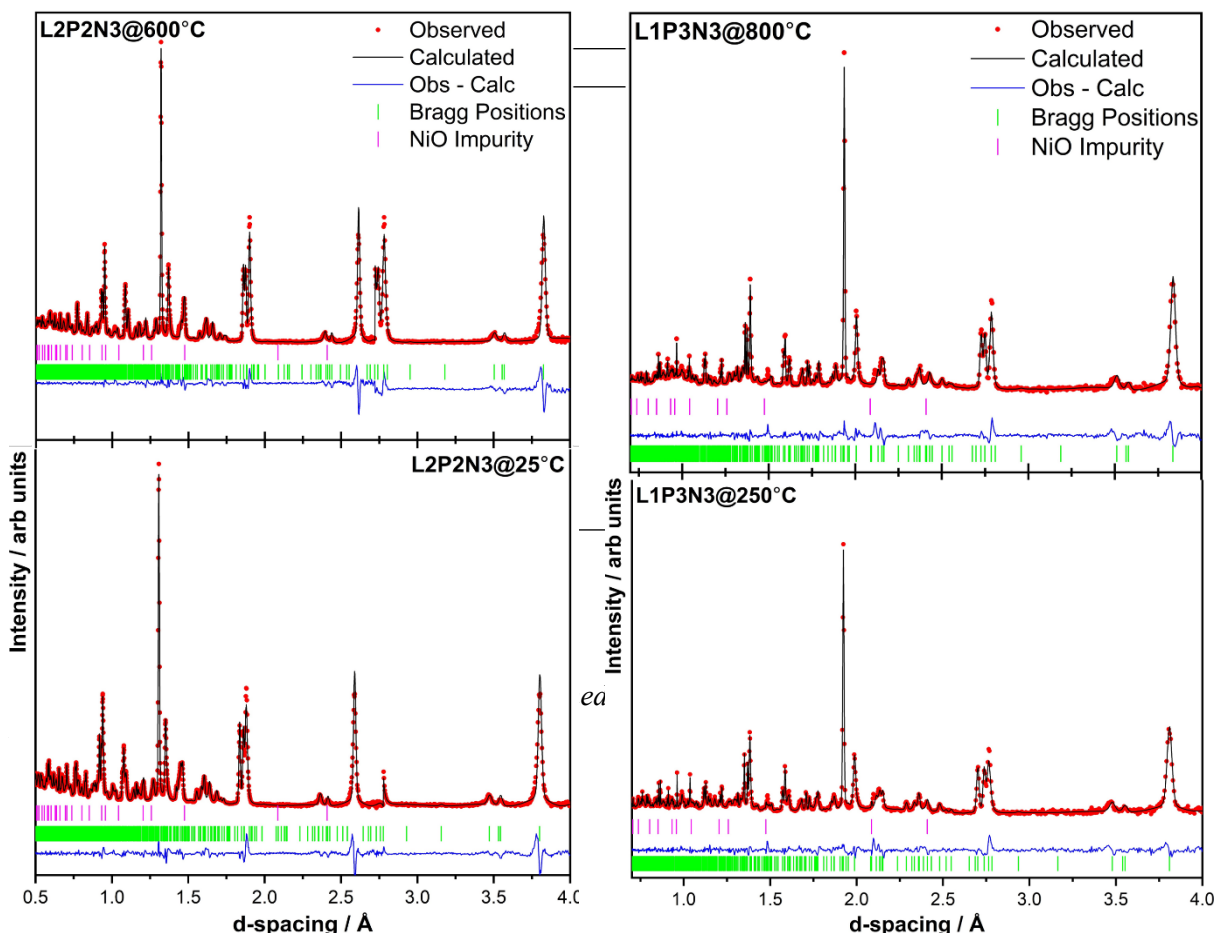


Figure S2. Rietveld refinement of the two data sets obtained at 25 °C ($\chi^2 = 1.91$; $R_p = 6.5\%$ and $wR_p = 3.0\%$) and 600 °C ($\chi^2 = 1.52$; $R_p = 5.7\%$ and $wR_p = 2.7\%$). A ~2% impurity of NiO was detected and was refined in cubic *Fm3m* space symmetry.

Figure S3. Rietveld refinement of the two data sets obtained at 250 °C ($\chi^2 = 11.47$; $R_p = 8.7\%$ and $wR_p = 7.6\%$) and 800 °C ($\chi^2 = 14.36$; $R_p = 6.9\%$ and $wR_p = 5.9\%$). A ~2% impurity of NiO was detected and was refined in cubic *Fm3m* space symmetry.

Atom	Atomic displacement parameters (\AA^2) $\times 100$			Fractional occupancy
	U_{11}/U_{iso}	U_{22}	U_{33}	
La(1)	1.9(3)			0.77(3)
Pr(1)	1.9(3)			0.33(3)
La(2)	1.8(7)			0.86(5)
Pr(2)	1.8(7)			0.14(5)
O(1)	0.7(6)	6.6(3)	11.0(4)	0.97(3)
O(2)	5.4(8)	5.4(8)	3.8(9)	0.97(3)
O(3)	0.6(9)	3.5(1)	3.5(1)	0.93(2)
O(4)	6.3(3)	6.3(3)	1.6(1)	1.0

Table S4. Rietveld refinement parameters obtained for the LIP3N3 composition recorded at 800°C

Atom	Atomic displacement parameters (\AA^2) $\times 100$					Fractional occupancy
	U_{11}/U_{iso}	U_{22}	U_{33}	U_{12}	U_{23}	
La(1)	3.5(3)					0.32(6)
Pr(1)	3.5(3)					0.68(6)
La(2)	1.7(4)					0.39(7)
Pr(2)	1.7(4)					0.61(7)
O(1)	6.8(1)	5.6(1)	1.7(1)	6.1(8)		0.97(5)
O(2)	5.3(8)	0.8(7)	3.9(8)		1.0(8)	0.96(3)
O(3)	0.5(6)	3.1(1)	2.6(1)	0.6(6)		0.83(5)
O(4)	3.5(1)	11.2(1)	3.5(7)		3.3(1)	1.0
O(5)	1.3(7)	1.9(5)	2.1(7)	0.3(4)		1.0

with orthorhombic model (*Bmab*).

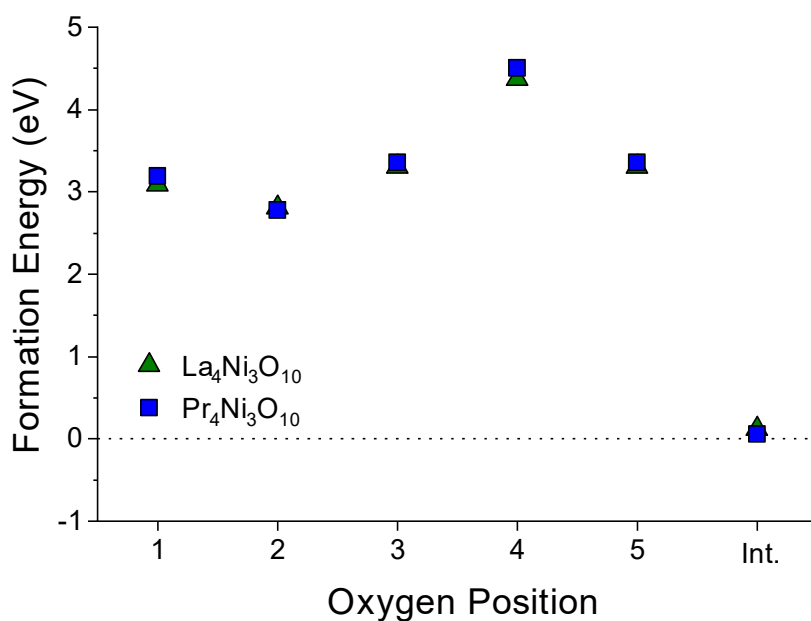


Figure S4: Plot of formation energy for oxygen vacancies at the O1–O5 sites in the Bmab primitive cell structures of $\text{La}_4\text{Ni}_3\text{O}_{10}$ and $\text{Pr}_4\text{Ni}_3\text{O}_{10}$, with 2 formula units per cell ($\text{M}_8\text{Ni}_6\text{O}_{20\pm 1}$) at $T = 0$ K without a U correction on Ni.

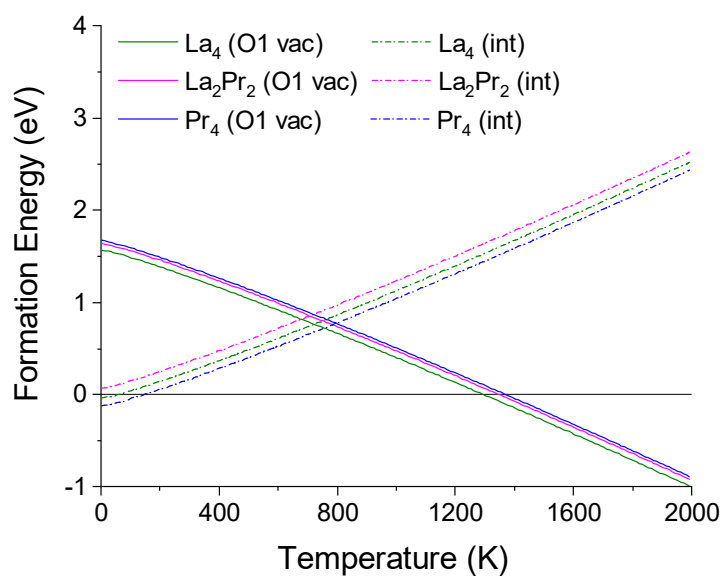


Figure S5. Variation in oxygen interstitial (int.) and O1 vacancy formation energy as a function of temperature for lowest energy $\text{La}_4\text{Ni}_3\text{O}_{10}$, $\text{La}_2\text{Pr}_2\text{Ni}_3\text{O}_{10}$ and $\text{Pr}_4\text{Ni}_3\text{O}_{10}$ structures.