

**Supporting Information  
for  
Sulfur monoxide complex of platinum fluoride with a positively  
charged ligand**

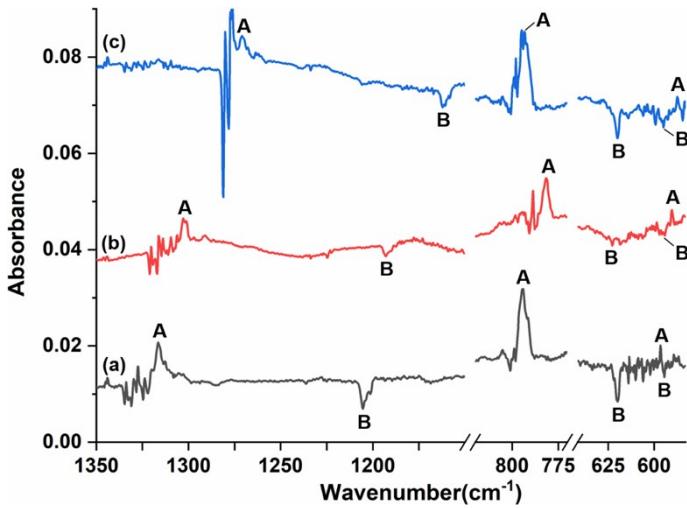
Deji Leng,<sup>a,b</sup> Zhixin Xiong,<sup>a,b</sup> Jingwen Hu,<sup>a,b</sup> Tiejian Zhu,<sup>a</sup> Xiuting Chen,<sup>a</sup> Yu Gong<sup>\*a</sup>

<sup>a</sup> Department of Radiochemistry, Shanghai Institute of Applied Physics, Chinese Academy of Sciences, Shanghai 201800, China

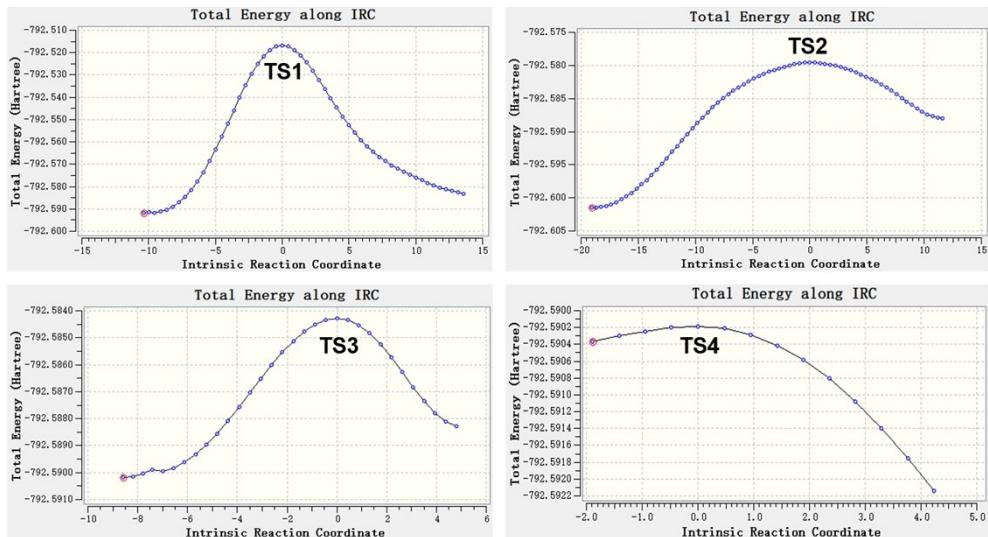
<sup>b</sup> School of Nuclear Science and Technology, University of Chinese Academy of Sciences, Beijing 100049, China

\*E-mail: [gongyu@sinap.ac.cn](mailto:gongyu@sinap.ac.cn).

## Additional Experimental and Theoretical Results



**Figure S1.** Infrared difference spectra in the 1350-1150, 820-770 and 640-585 cm<sup>-1</sup> regions from codeposition of laser-ablated platinum atoms and isotopically labeled SOF<sub>2</sub> in solid argon at 4 K (spectra taken after 25 K annealing minus spectra taken after  $\lambda > 220$  nm irradiation): (a) 0.5% SOF<sub>2</sub>, (b) 0.5% <sup>34</sup>SOF<sub>2</sub>, (c) 0.5% S<sup>18</sup>OF<sub>2</sub>. Labels A and B denote the Pt(SOF<sub>2</sub>) and PtF<sub>2</sub>( $\eta^1$ -SO) products, respectively.



**Figure S2.** IRC plots of TS1-4 associated with the structural transformation as shown in the potential energy profile.

**Table S1.** Experimental infrared absorptions ( $\text{cm}^{-1}$ ) of  $\text{Pt}(\text{SOF}_2)$  and  $\text{PtF}_2(\eta^1\text{-SO})$  in solid argon.

	mode	$\text{SOF}_2$	$^{34}\text{SOF}_2$	$^{38}\text{SOF}_2$
Pt( $\text{SOF}_2$ )	S-O str.	1316.4 <sup>a</sup>	1302.8 <sup>a</sup>	1270.8 <sup>a</sup>
	sym.F-S-F str.	794.3 <sup>a</sup>	781.5 <sup>a</sup>	793.4 <sup>a</sup>
	wagging	596.6	590.6	587.5
$\text{PtF}_2(\eta^1\text{-SO})$	S-O str.	1205.4	1193.0	1162.0
	antisym.F-Pt-F str.	619.8	622.1	619.6
	sym.F-Pt-F str.	594.9	594.7	594.8

<sup>a</sup> Determined from the infrared difference spectrum as they are partially covered by the  $\text{SOF}_2$  precursor bands.

**Table S2.** Orbital composition of the S–O and Pt–S bonds in  $\text{PtF}_2(\eta^1\text{-SO})$  from NBO calculations at the B3LYP/6-311+G(3df)/SDD level.

	bond type	% of NBO on each atom	% of each hybrid in the NBO from each atom			
			s	p	d	f
S-O	$\pi$	O (68.54)	0.00	99.25	0.74	0.01
		S (31.46)	0.00	98.05	1.86	0.09
S-O	$\sigma$	O (68.63)	25.92	72.51	1.55	0.02
		S (31.37)	16.87	80.95	2.04	0.14
Pt-S	$\sigma$	S (47.90)	6.23	91.88	1.84	0.06
		Pt (52.10)	21.51	0.21	78.28	-
S-O	$\pi^*$	O (31.46)	0.00	99.25	0.74	0.01
		S (68.54)	0.00	98.05	1.86	0.09
Pt-S	$\sigma^*$	S (52.10)	6.23	91.88	1.84	0.06
		Pt (47.90)	21.51	0.21	78.28	-

**Table S3.** NPA charges of SO and Pt in the  $\text{PtF}_2(\eta^1\text{-SO})$  complex calculated at different level of theory.

functional	SO	Pt
B3LYP	0.37	0.82
M06	0.39	0.84
M06-L	0.30	0.84
PBE0	0.38	0.82
TPSSh	0.33	0.81

**Table S4.** Vibrational frequencies and bond lengths of free SO and the SO ligand in  $\text{PtF}_2(\eta^1\text{-SO})$  calculated at different level of theory.

functional	SO in $\text{PtF}_2(\eta^1\text{-SO})$		free SO	
	frequency ( $\text{cm}^{-1}$ )	bond length ( $\text{\AA}$ )	frequency ( $\text{cm}^{-1}$ )	bond length ( $\text{\AA}$ )
B3LYP	1245.7	1.449	1157.3	1.488
M06	1309.3	1.438	1198.7	1.479
M06-L	1250.5	1.453	1167.5	1.487
PBE0	1286.6	1.441	1200.7	1.478
TPSSh	1224.3	1.457	1153.8	1.491

**Cartesian coordinates for PtF<sub>2</sub>(η<sup>1</sup>-SO), Pt(SOF<sub>2</sub>) as well as the transition states and intermediates involved in the isomerization process obtained at the B3LYP/6-311+G(3df)/SDD level of theory. All of these species possess singlet ground states.**

**PtF<sub>2</sub>(η<sup>1</sup>-SO)**

F	0.05974800	0.85788800	1.87175700
O	-1.21955600	-2.36141200	0.00000000
F	0.05974800	0.85788800	-1.87175700
S	0.05974800	-1.68038900	0.00000000
Pt	0.09903900	0.38891700	0.00000000

**Pt(SOF<sub>2</sub>)**

O	-1.83055800	-1.49863500	0.00000000
F	0.22970800	-1.97387700	1.15088100
F	0.22970800	-1.97387700	-1.15088100
S	-0.46296900	-1.11177300	0.00000000
Pt	0.22970800	0.83727200	0.00000000

**TS1 (-371.30 cm<sup>-1</sup>)**

O	2.04892700	-1.36713100	-0.44702700
F	1.98916600	0.43737700	1.09510100
F	0.76315500	1.75206600	-0.60024100
S	1.33087100	-0.13122500	-0.25971200

Pt -0.80072100 -0.08549200 0.04202400

### **IM1**

O -1.41087700 -1.20191500 -0.24929500  
F -2.29375800 1.12967000 -0.45153600  
F 2.12868600 0.84452200 0.14756700  
S -1.56792000 0.07511800 0.52532000  
Pt 0.48537700 -0.11992700 -0.04711600

### **TS2 (-80.26 cm<sup>-1</sup>)**

O 2.11617800 0.87071000 -0.90142500  
F 2.02451400 -1.02074200 0.66588600  
F -1.64364700 1.14689900 0.72603300  
S 1.46568000 0.50170500 0.33548600  
Pt -0.56164300 -0.20677400 -0.13697000

### **IM2**

O 2.24033700 1.14508600 -0.54553400  
F 1.38664500 -1.35864800 -0.09149500  
F -1.90785200 1.10677800 0.07933600  
S 1.54305500 0.37222500 0.44526000  
Pt -0.48616300 -0.16473700 -0.03398000

**TS3 (-128.38 cm<sup>-1</sup>)**

O	-2.21540700	-1.27977400	-0.43672200
F	-0.88049100	1.72255100	-0.13520000
F	1.76653900	-1.25997100	0.06303500
S	-1.58659900	-0.25355700	0.35836300
Pt	0.45044100	0.12989600	-0.02039200

**IM3**

O	2.49809300	-0.76561200	-0.00053800
F	-0.46868300	1.98213200	-0.00006800
F	-0.80987100	-1.90626200	0.00026600
S	1.60691700	0.37136400	0.00041300
Pt	-0.43831300	-0.00640700	-0.00005200

**TS4 (-61.87 cm<sup>-1</sup>)**

F	-0.64264200	1.95503100	-0.07262500
O	2.54928600	0.56778600	0.19670400
F	-0.69826500	-1.93134300	0.04050700
S	1.58168800	-0.44898200	-0.14689800
Pt	-0.43119400	0.03113100	0.01366400