

SUPPORTING INFORMATION

**The synthesis and super capacitive characterization of microwave-assisted highly crystalline  $\alpha\text{-Fe}_2\text{O}_3/\text{Fe}_3\text{O}_4$  nanoheterostructure**

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Figure SI 1

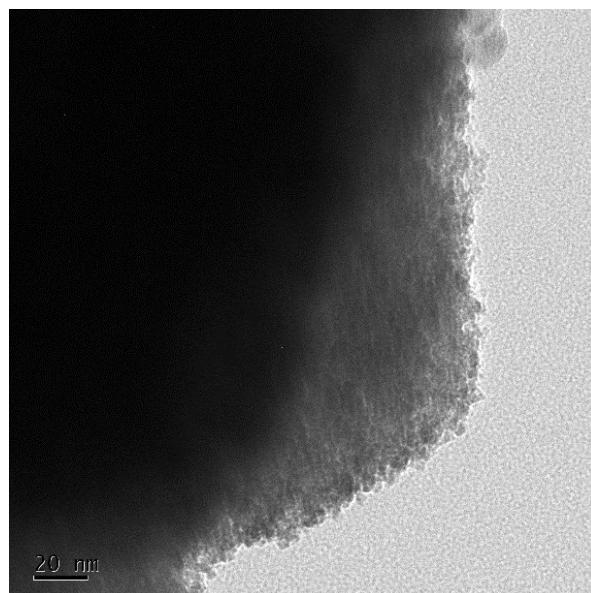


Figure SI 1 TEM image of  $\text{Fe}_2\text{O}_3/\text{Fe}_3\text{O}_4$  nanocomposite prepared at 1:3 Water to EG solvent ratios

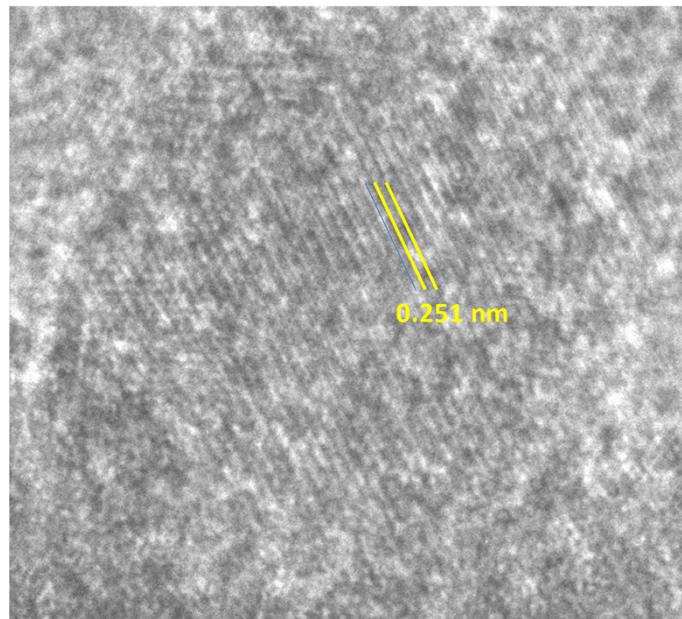


Figure SI 2 lattice resolved HRTEM of  $\text{Fe}_2\text{O}_3$  on  $\text{Fe}_2\text{O}_3$  nanoparticles.

Figure SI3

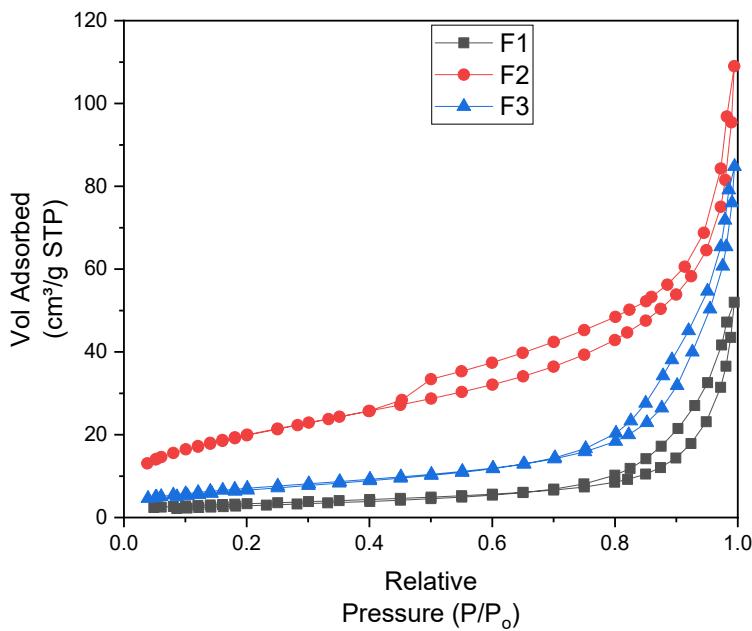


Figure SI3: Nitrogen adsorption and desorption isotherms of F1, F2 and F3.

Table SI1: Comparison of supercapacitance with reported values.

Material	Supercapacitance (F/g)
a-Fe <sub>2</sub> O <sub>3</sub> /Fe <sub>3</sub> O <sub>4</sub> heterostructure (ref 1)	150 @ 0.5mA/g
Fe <sub>2</sub> O <sub>3</sub> -Fe <sub>3</sub> O <sub>4</sub> /N-rGO (Ref 2)	120 @ 0.8 A/g
Fe <sub>3</sub> O <sub>4</sub> /Fe <sub>2</sub> O <sub>3</sub> heterostructures (This work)	165 @ 0.5mA/g
Fe <sub>3</sub> O <sub>4</sub> nanoparticles (This work)	143 @ 0.5mA/g
Fe <sub>3</sub> O <sub>4</sub> nanoparticles (Ref 3)	95.4 @ 1.0mA/g
Fe <sub>3</sub> O <sub>4</sub> nanoparticles (Ref 4)	145 @ 0.5mA/g

1. Dejian Chen et al. RSC Adv., 2016, 6, 45023
2. Sourav Mallick et al. ChemElectroChem 2018, 5, 2348 – 2356.
3. P.M.Anjan et al., Materials Science and Engineering: B, 2023, 290, 116313
4. T. Arun et al. *Applied Surface Science* 485 (2019) 147–157