

## *Supporting Information for*

### **Quantitative evaluation of H-donating abilities of C(sp<sup>3</sup>)-H bonds of nitrogen-containing heterocycles in hydrogen atom transfer reaction**

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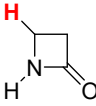
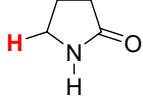
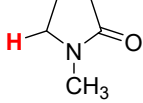
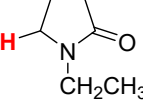
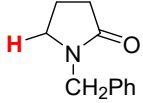
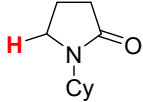
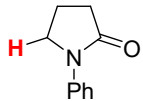
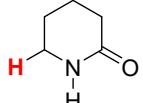
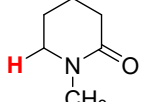
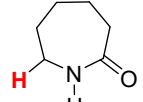
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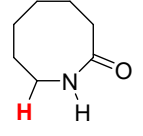
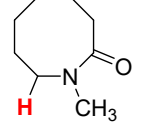
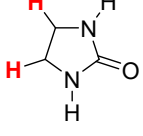
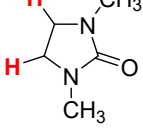
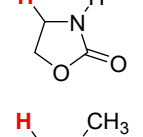
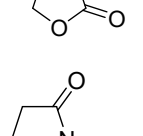
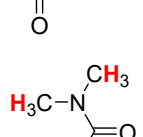
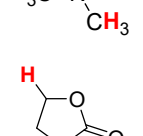
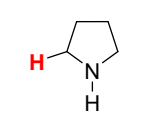
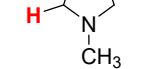

SI	Kinetic studies of the HAT reactions from C(sp <sup>3</sup> )-H bonds of nitrogen-containing heterocycles and other H-donors to CumO• radical in literatures	S2
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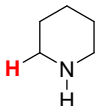
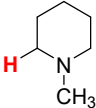




## SI. Kinetic studies of the HAT reactions from C(sp<sup>3</sup>)-H bonds of nitrogen-containing heterocycles and other H-donors to CumO<sup>•</sup> radical in literatures

The second-order rate constants  $k_2$  and activation free energies  $\Delta G^\ddagger_{\text{XH/Y}}$  which were obtained by Eyring equation [ $k_2 = (k_B T/h)\exp(-\Delta G^\ddagger/RT)$ ] of HAT reactions from C(sp<sup>3</sup>)-H bonds of nitrogen-containing heterocycles and other H-donors to CumO<sup>•</sup> radical in acetonitrile (MeCN) at 298 K are listed in Table S1.

**Table S1.** Second-order rate constants ( $k_2$ ), activation free energies ( $\Delta G^\ddagger_{\text{XH/CumO}^\bullet}$ ) of HAT reactions (XH/CumO<sup>•</sup>) and the references in MeCN at 298 K.

Entry	XH	$k_2$ (M <sup>-1</sup> s <sup>-1</sup> ) <sup>a</sup>	$\Delta G^\ddagger_{\text{XH/CumO}^\bullet}$ (kcal/mol) <sup>b</sup>	Ref.
1H		$3.60 \times 10^5$	9.87	S1
2H		$2.27 \times 10^6$	8.78	S1
3H		$4.53 \times 10^6$	8.37	S1
4H		$4.60 \times 10^6$	8.36	S1
5H		$2.34 \times 10^6$	8.76	S1
6H		$3.80 \times 10^6$	8.47	S1
7H		$1.90 \times 10^6$	8.88	S1
8H		$2.76 \times 10^6$	8.66	S1
9H		$3.75 \times 10^6$	8.48	S1
10H		$8.90 \times 10^5$	9.33	S1

11H		$2.45 \times 10^6$	8.73	S1
12H		$6.44 \times 10^5$	9.52	S1
13H		$1.14 \times 10^6$	9.19	S1
14H		$9.20 \times 10^6$	7.95	S1
15H		$3.20 \times 10^7$	7.21	S1
16H		$8.90 \times 10^5$	9.33	S1
17H		$2.02 \times 10^6$	8.85	S1
18H		$1.00 \times 10^5$	10.63	S1
19H		$6.00 \times 10^6$	8.20	S1
20H		$4.00 \times 10^4$	11.17	S1
21H		$1.24 \times 10^8$	6.41	S2
22H		$1.91 \times 10^8$	6.15	S2

23H		$1.07 \times 10^8$	6.50	S2
24H		$1.22 \times 10^8$	6.42	S2
25H		$9.54 \times 10^5$	9.29	S3
26H		$1.10 \times 10^6$	9.21	S4
27H		$2.20 \times 10^6$	8.80	S5
28H		$3.20 \times 10^6$	8.57	S6

<sup>a</sup> The uncertainty of date is smaller than 5%. <sup>b</sup> The date of  $\Delta G^\ddagger_{\text{XH/Y}}$  are derived from Eyring equation  $k_2 = (k_B T/h) \exp(-\Delta G^\ddagger/RT)$ .

## SII. The brief introduction of Applied Photophysics SX.18MV-R stopped-flow

SX.18MV-R stopped-flow is a fast dynamic measuring device with high sensitivity produced by Applied Photophysics Limited Company. The system has a detection wavelength range of 200 ~ 900 nm and a movable sample cell of 20ul, 5ul and 2ul to suit different determination needs. The Dead-Time of the system is less than 500 us (far less than the Dead-Time of the general stopped-flow device about 4 ms), and the rate constant can be measured up to  $3000 \text{ s}^{-1}$ ; High sensitivity, low noise (0.0001AU); Temperature control range (-20 to 60 °C). RISC OS3.11 operating system, fast and simple, and can directly analyze the dynamics curve to give the corresponding dynamics parameters. The picture of the Applied Photophysics SX.18MV-R stopped-flow is shown below.



## References

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