First-principles study on the structure and electronic properties of  $M_2CS_x$  (M=Sc, Ti, Y, Zr and Hf, x= 1, 2)

Huaijin Zhu<sup>a,b</sup>, Nianxiang Qiu<sup>b\*</sup>, Gang Fang<sup>b</sup>, Shiyu Du<sup>b,c,d,e\*</sup>

<sup>a</sup>Faculty of Electrical Engineering and Computer Science, Ningbo University, Ningbo, Zhejiang 351201, China

<sup>b</sup>Engineering Laboratory of Advanced Energy Materials, Ningbo Institute of Materials Technology and Engineering, Chinese Academy of Sciences, Ningbo, Zhejiang 315201, P. R. China.

<sup>c</sup>School of Materials Science and Engineering, China University of Petroleum (East China), Qingdao, 266580, China

<sup>d</sup>School of Computer Science, China University of Petroleum (East China), Qingdao, 266580, China <sup>e</sup>Milky-Way Sustainable Energy Ltd, Zhuhai, China

\*Corresponding Authors: E-mail: <u>giunianxiang@nimte.ac.cn</u> and <u>dushiyu@nimte.ac.cn</u>



**Figure S1.** The top and side views of six possible configurations (**A-D**) of stoichiometric  $M_2CS_2$  (M=Sc, Ti, Y, Zr and Hf). Cyan, yellow and brown spheres represent M, S and C atoms, respectively. **A:** The S layers on both sides are directly projected to the middle C layer, **B:** The S layers on both sides are directly projected to the bottom M layer, **C:** The S layer on one side is directly projected to the middle C layer and the S layer on the other side is directly projected to the bottom M layer, **D:** The S atoms on both sides are located on the top of M atoms.

_					
		А	В	С	D
_	Sc <sub>2</sub> CS <sub>2</sub>	-35.83	-35.02	-36.17	-32.27
	$Ti_2CS_2$	-38.88	-40.07	-39.38	-36.01
	$Y_2CS_2$	-36.39	-36.05	-36.46	-35.99
	Zr2CS	-41.52	-41.96	-41.49	-30.53
	Hf <sub>2</sub> CS <sub>2</sub>	-43.68	-44.88	-44.36	-40.10

Table S1. The calculated total energies of the four possible configurations of M<sub>2</sub>CS<sub>2</sub>.

**Table S2**. The calculated total energies (in eV) of fully optimized  $2 \times 2 \times 1$  supercells of the most stable configurations of Sc<sub>2</sub>CS<sub>2</sub> and Y<sub>2</sub>CS<sub>2</sub> with and without dipole correction in this study, as well

(14/10/06/4/2 2020) 12 (12/) 2220 2/10/ 00/11/01/06/11						
	Configuration	Total energy in this	Dipole-corrected	Dipole-corrected		
		study	total energy in	total energy by		
			this study	Zhang <i>et al</i> .		
$Sc_2CS_2$	type-B	-144.68	144.69	142.16		
	distorted type-B	-144.70				
Y <sub>2</sub> CS <sub>2</sub>	type-B	-145.83	145.83	-143.99		
	distorted type-B	-145.84				

as the dipole-corrected total energies of the corresponding structures reported by Zhang *et al.* (*Nanoscale* 2020, 12 (41), 21291) for comparison.



Figure S2. Distorted type-B Sc<sub>2</sub>CS<sub>2</sub> structure before and after optimization.



**Figure S3.** The top and side views of six possible configurations (**A-F**) of nonstoichiometric M<sub>2</sub>CS (M=Ti, Zr, and Hf) created by *disorder* code based on the stable configuration of M<sub>2</sub>CS<sub>2</sub> in **Figure S1 B**. The bright blue, yellow and brown balls indicate M, S and C atoms, respectively. **A:** All four S atoms on one side of the M<sub>2</sub>CS supercell are removed, **B** and **C**: One S atom is removed from one side of the M<sub>2</sub>CS supercell, while three S atoms are removed from the other side. It leaves the only S atom on one side close to the only S vacancy on the other side in configuration **B**, and far from the vacancy in configuration **C**. **D**, **E** and **F**: The four S atoms on both sides of the M<sub>2</sub>CS supercell are uniformly removed. The remaining S atoms on both sides are arranged in stripes. The stripe-like sulfur terminations on one side are directly below those on the other side, otherwise it is configuration **F**. In configuration **E**, the strip-like sulfur groups on both sides appear to cross.

	А	В	С	D	E	F	
Ti <sub>2</sub> CS	-131.11	-133.62	-133.48	-134.86	-134.3	-134.30	
$Zr_2CS$	-137.50	-139.73	-139.59	-140.58	-140.40	-140.27	

**Table S3**. The calculated total energies of the six possible configurations of nonstoichiometric  $M_2CS$  (M=Ti, Zr, and Hf).



**Figure S4**. The top and side views of nine possible configurations (**A-I**) of nonstoichiometric Sc<sub>2</sub>CS and Y<sub>2</sub>CS based on the stable configuration of Sc<sub>2</sub>CS<sub>2</sub> and Y<sub>2</sub>CS<sub>2</sub> in **Figure S1 C**. The bright blue, yellow and brown balls indicate M, S and C atoms, respectively. **A**: All the four S atoms on the side of the M<sub>2</sub>CS supercell projected to the bottom M layer are removed, **B** and **C**: One S atom on the side of the M<sub>2</sub>CS supercell projected to the middle C layer is removed, while three S atoms on the side of the M<sub>2</sub>CS supercell projected to the bottom M layer are removed. It leaves the only S atom on one side close to the only S vacancy on the other side in configuration **B**, and far from the vacancy in configuration **C**. **D**, **E** and **F**: The four S atoms on both sides of the M<sub>2</sub>CS supercell are uniformly removed. The remaining S atoms on both sides are arranged in stripes. The stripe-like sulfur terminations on both sides are parallel, and in configuration **F**, the distance between the two stripe-like S groups is the shortest in the xy plane, otherwise it is configuration **D**. In configuration **E**, the strip-like sulfur groups on both sides appear to cross. **G** and **H**: Three S atoms on the side of the M<sub>2</sub>CS supercell projected to the middle C layer is removed. It leaves the only S atom on the side of the M<sub>2</sub>CS supercell projected to the middle C layer is removed.

vacancy in configuration H. I: All the four S atoms on the side of the  $M_2CS$  supercell projected to the middle C layer are removed.

**Table S4**. The calculated total energies of the nine possible configurations of nonstoichiometric  $Y_2CS$  and  $Y_2CS$ .

	А	В	С	D	E	F	G	Н	I
Sc <sub>2</sub> CS	-117.24	-119.88	-119.52	-120.75	-120.88	-121.32	-120.31	-118.72	-118.57
Y <sub>2</sub> CS	-117.94	-120.05	-119.83	-120.55	-120.85	-121.28	-118.95	-118.97	-119.12



**Figure S5.** The most stable configuration of  $M_2CS$  (M=Sc, Y, Ti, Zr, and Hf) derived from  $M_2CS_2$  configurations A, B, and C in **Figure S1**, respectively, with strip-like sulfur terminations on both sides.

**Table S5**. The calculated total energies of the most stable configurations of  $M_2CS$  (M=Sc, Y, Ti, Zr, and Hf) derived from  $M_2CS_2$  configurations C, A, and B in **Figure S1**, respectively.

	А	В	С
Sc <sub>2</sub> CS	-117.63	-121.42	-121.31
Ti <sub>2</sub> CS	-132.86	-134.86	-128.41
Y2CS	-117.28	-121.30	-121.28
Zr <sub>2</sub> CS	-139.44	-140.58	-133.48
Hf <sub>2</sub> CS	-149.79	-151.13	-150.33





**Figure S6.** The calculated phonon dispersions of the  $M_2CS_x$  (M=Sc, Ti, Y, Zr, and Hf, x=1, 2).

**Figure S7.** The charge density differences of  $M_2CS_x$  (M=Sc, Y, Ti, Zr and Hf, x=1, 2).