

Supporting information

Enhanced electrocatalytic oxygen redox reactions of iron oxide nanorod films by combining oxygen vacancy formation and cobalt doping

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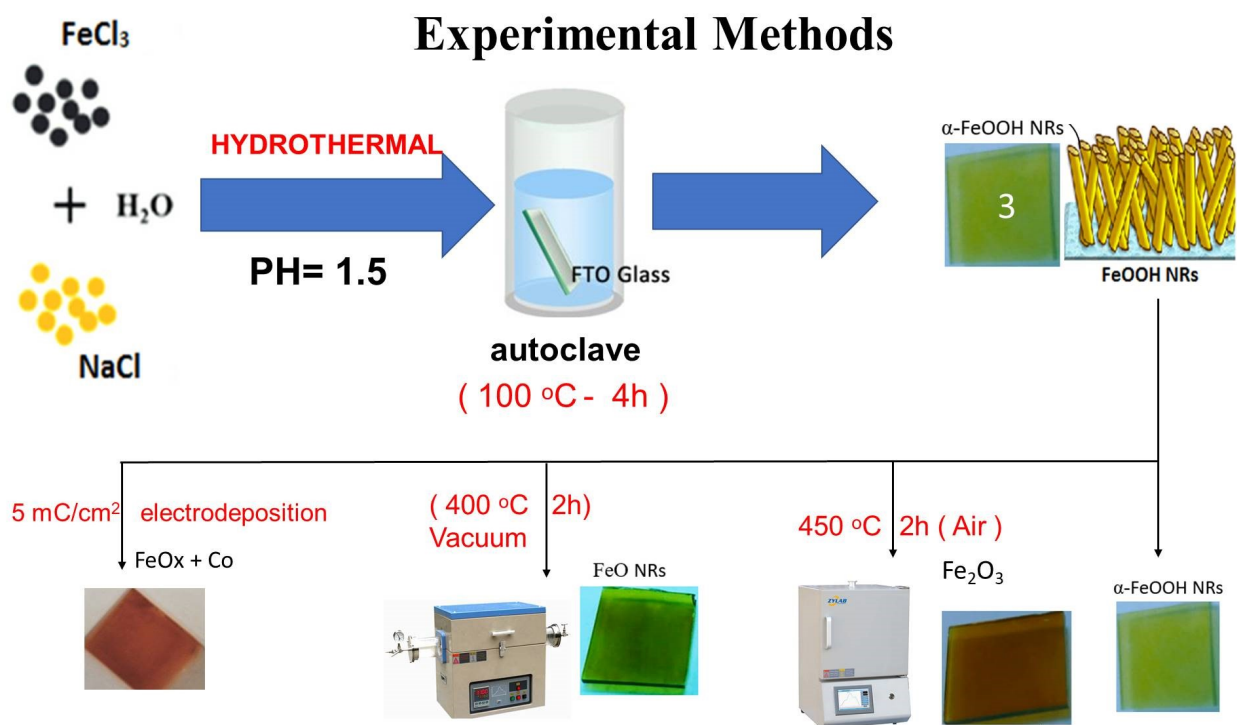


Figure S1. Schematic illustration of the synthesis procedures of fabricated $\alpha\text{-FeOOH}$, Fe_2O_3 , FeO, and Co-doped FeO NRs materials.

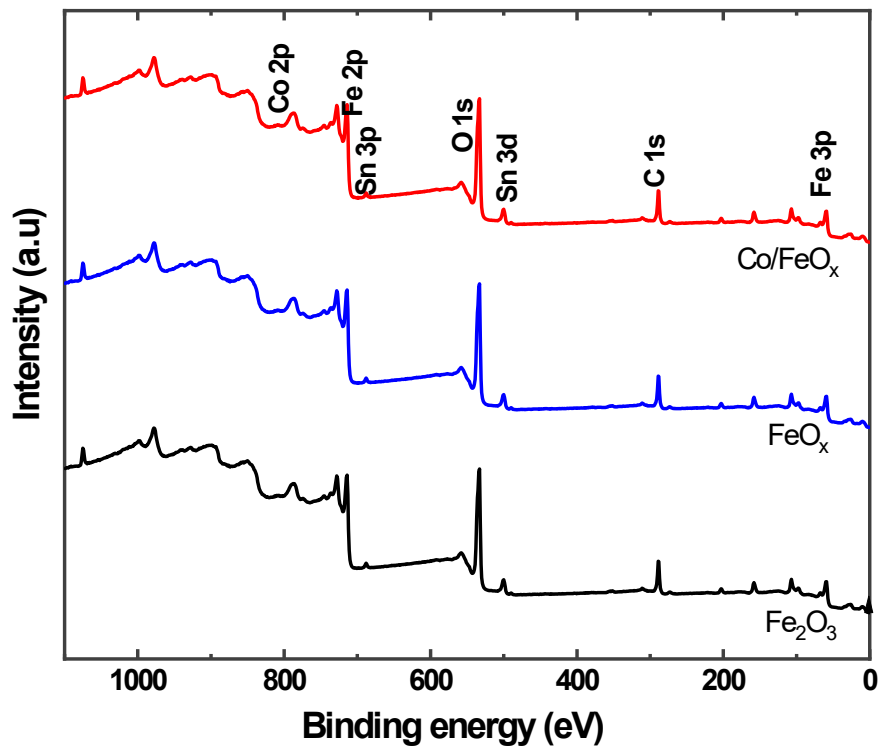


Figure S2. Comparative XPS survey spectra of fabricated Fe₂O₃, FeO_x, and Co-doped FeO_x electrodes fabricated by hydrothermal and electrodeposition approaches

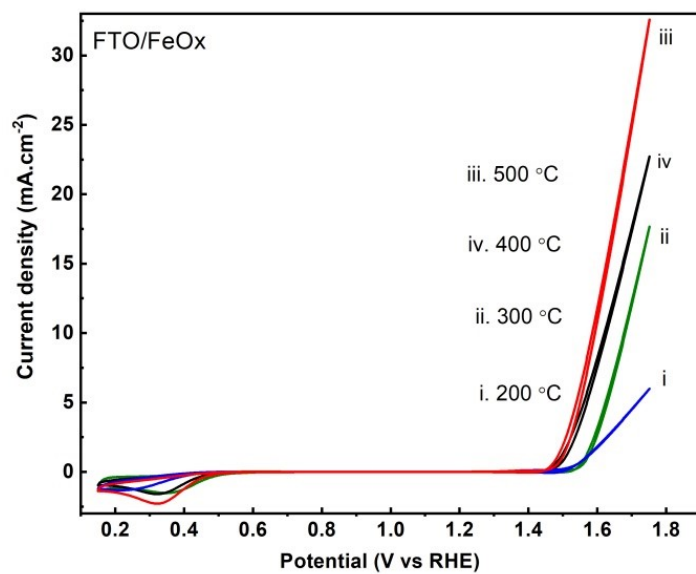


Figure S3. Cyclic voltammetric plots FeO_x/FTO electrode after annealing under Vacuum at different annealing temperatures (in 1.0 M KOH (pH 13.6) with scan rate 10 mVs⁻¹)

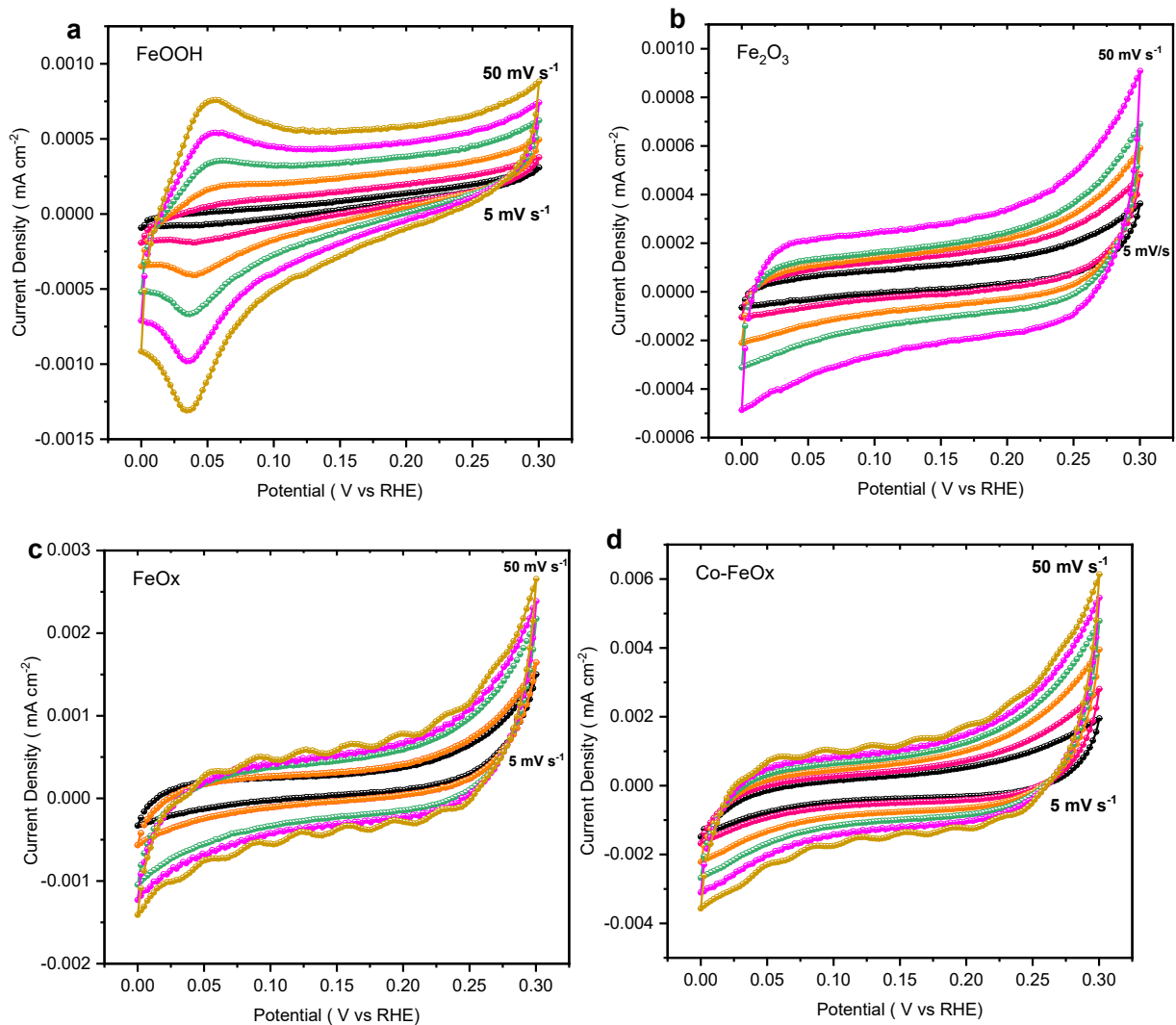


Figure S4. Electrochemical surface area measurements. Cyclic voltammograms of a) FeOOH, Fe₂O₃ (b), FeOx (c), Co-doped FeOx (d) electrodes at various scan rates.

Table S1. Impedance parameter values derived from the fitting to the equivalent circuit for the impedance spectra recorded in 1.0 M KOH solution with an applied potential of 1.6 V vs RHE. R_s = solution resistance, R_{ct} = charge-transfer resistance,.

Materials	R_s, Ω	CPE, μMho	R_{ct}, Ω	n
Fe_2O_3 NRs	35	6.02	55.6	0.587
$\alpha\text{-FeOOH}$ NRs	21	88.2	17.9	0.764
FeO_x NRs	17.5	2.84	6.65	0.700
Co/FeO_x NRs	11.5	10.4	2.61	0.65