

# **SUPPORTING INFORMATION**

**Solvothermal synthesis of organoclay/Cu-MOF composite and its application in film modified GCE for simultaneous electrochemical detection of deoxyepinephrine, acetaminophen and tyrosine**

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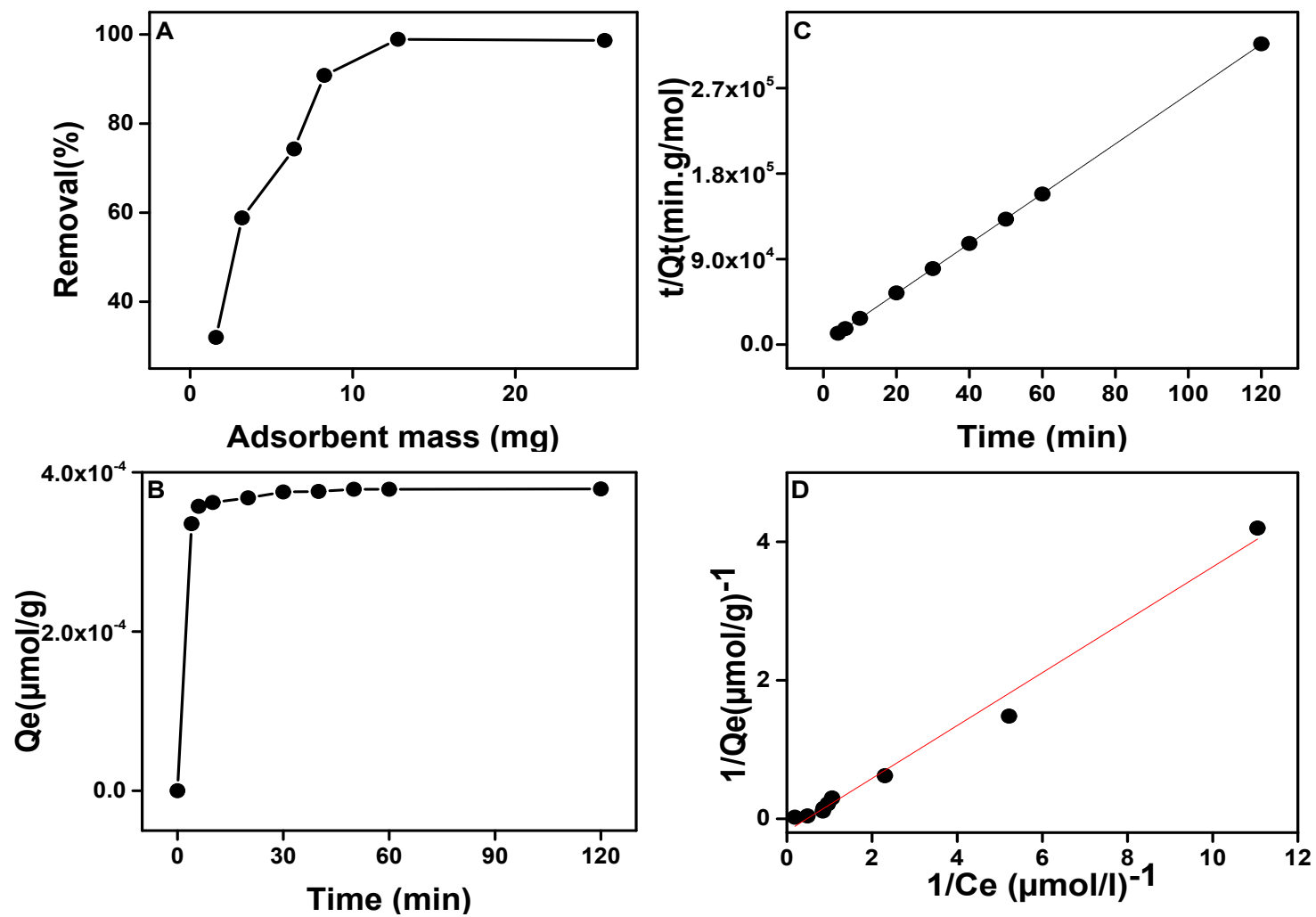
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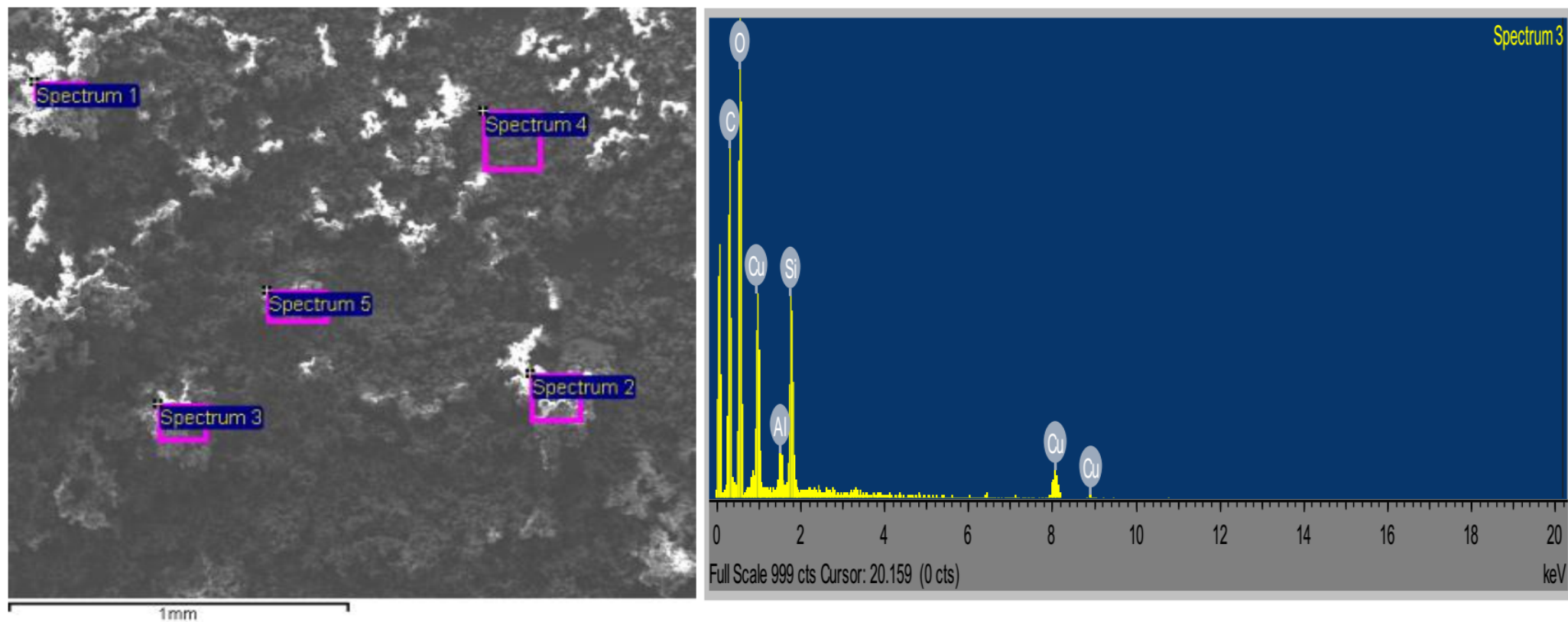
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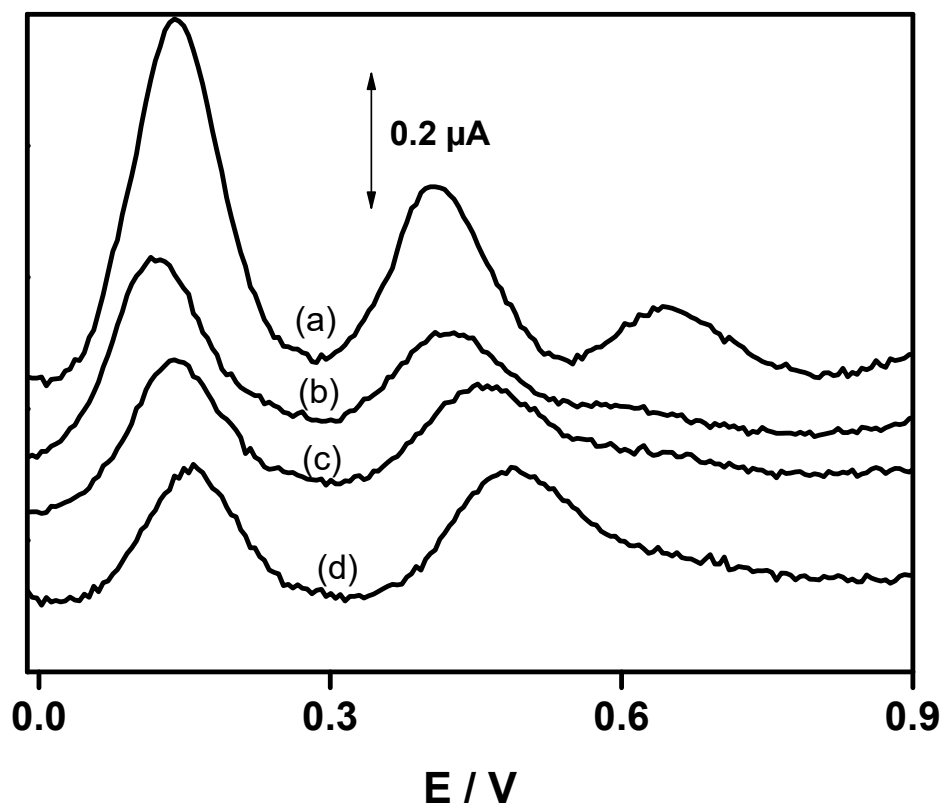
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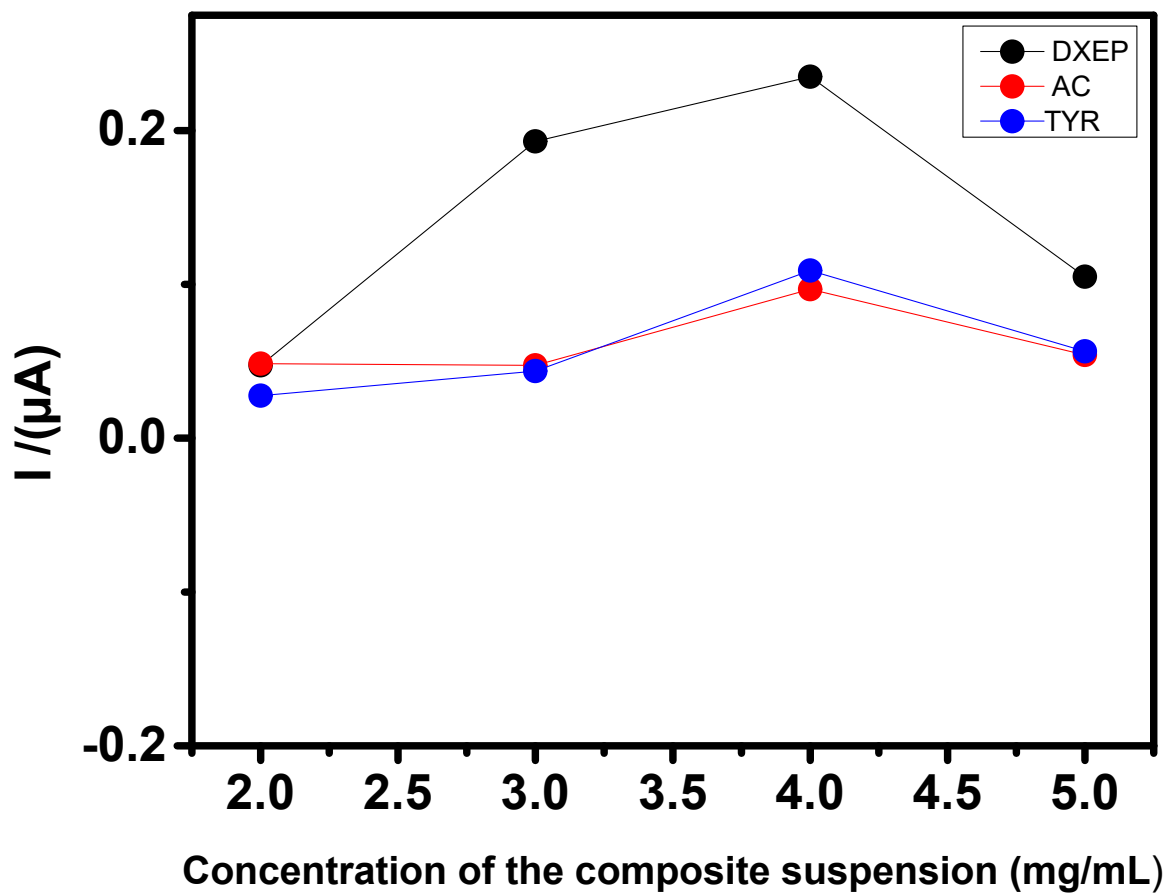
**Fig. S1.** Effect of the (A) mass of clay and (B) time on the dye adsorption, (C) Pseudo second order model and (D) Langmuir plots



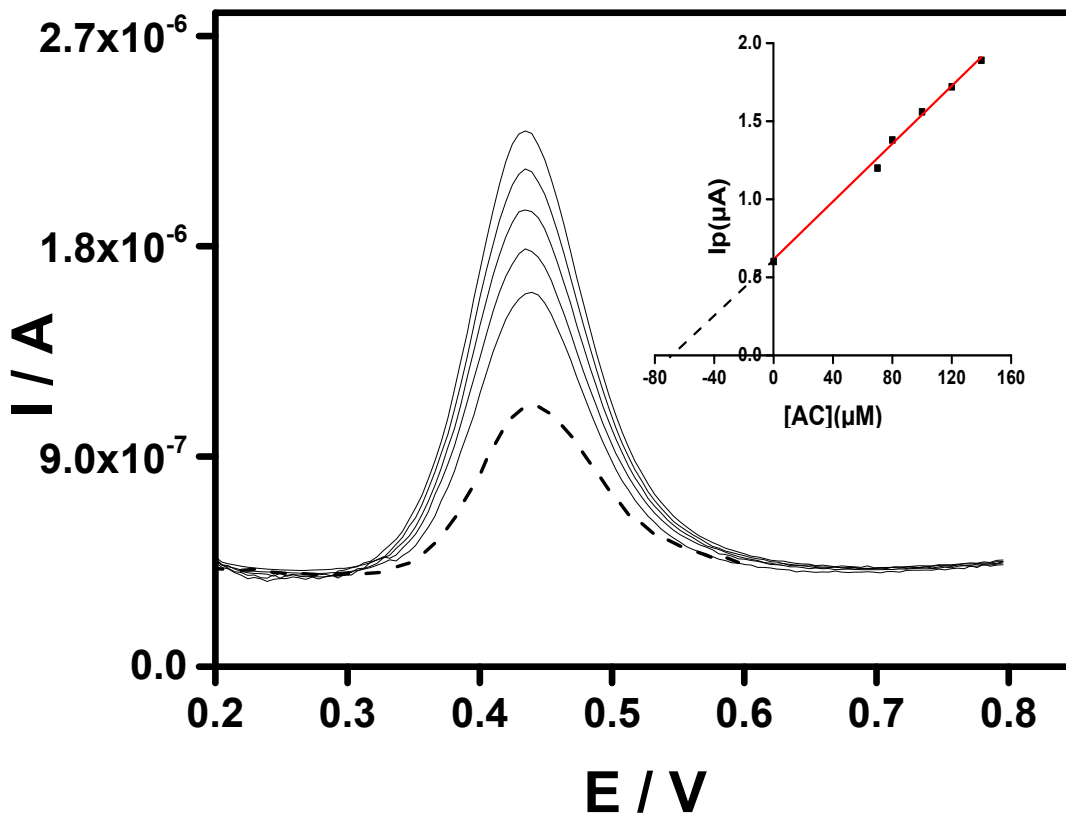
**Fig. S2.** EDS analysis and spectra of Sa-TN<sub>10</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>



**Fig. S3.** DPVs recorded at different scan rates at Sa-TN<sub>5</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>/GCE (a), Sa-TN<sub>10</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>/GCE, Sa-TN<sub>25</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>/GCE and Sa-TN<sub>50</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>/GCE in AcB pH 5 containing either: (A) 19.8 μM DXEP, (B) 23.8 μM AC or (C) 24.2 μM TYR.



**Fig. S4.** Dependence of electrochemical response on suspension concentration of Sa-TN<sub>50</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub> dropped coating on GCE



**Fig. S5.** DPVs recorded at Sa-TN<sub>50</sub>/Cu<sub>3</sub>(BTC)<sub>2</sub>/GCE in Ac buffer solution with pH 5 in which AC sample was added (dashed line) followed by 5 successive additions of AC standards (solid lines). Inset shows plot of peak current vs. AC concentration.

**Table S1.** Thionin acetate kinetic parameters obtained from the pseudo-second order models

<b>Experimental data</b>	<b>Value</b>
$q_{e(\text{exp})}$	379.1 $\mu\text{mol/g}$
<b>Pseudo second-order</b>	
$q_{e2}$	381 $\mu\text{mol/g}$
$h$	776 $\mu\text{mol/g.min}$
$k_2$	5344.295 $\text{g/mol.min}$
$R^2$	0.9999

**Table S2.** Determination of DXEP, AC and TYR contained in tap water.

Added ( $\mu\text{mol/L}$ )			Found ( $\mu\text{mol/L}$ )			Recovery %		
DXEP	AC	TYR	DXEP	AC	Tyr	DXEP	AC	TYR
2	1	2	1.95	0.90	1.95	97.5	90.0	97.5
4	2	4	4.33	1.99	3.65	108.2	99.5	91.3