

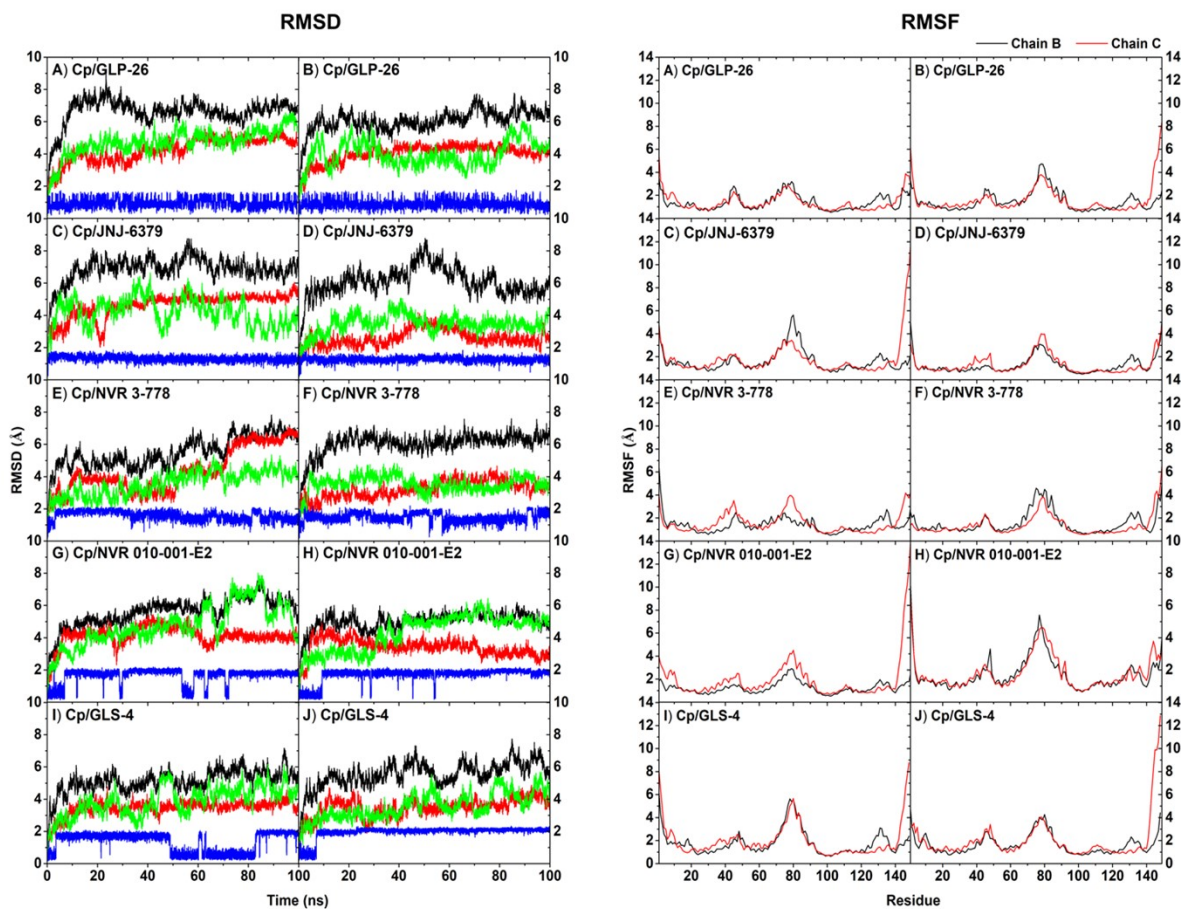
# Binding Characteristics of Pyrrole-scaffold Hepatitis B Virus Capsid Inhibitors and Identification of Novel Potent Compounds

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Pudhom<sup>4</sup>, and Thomayant Prueksaritanont<sup>1</sup>\*□□*

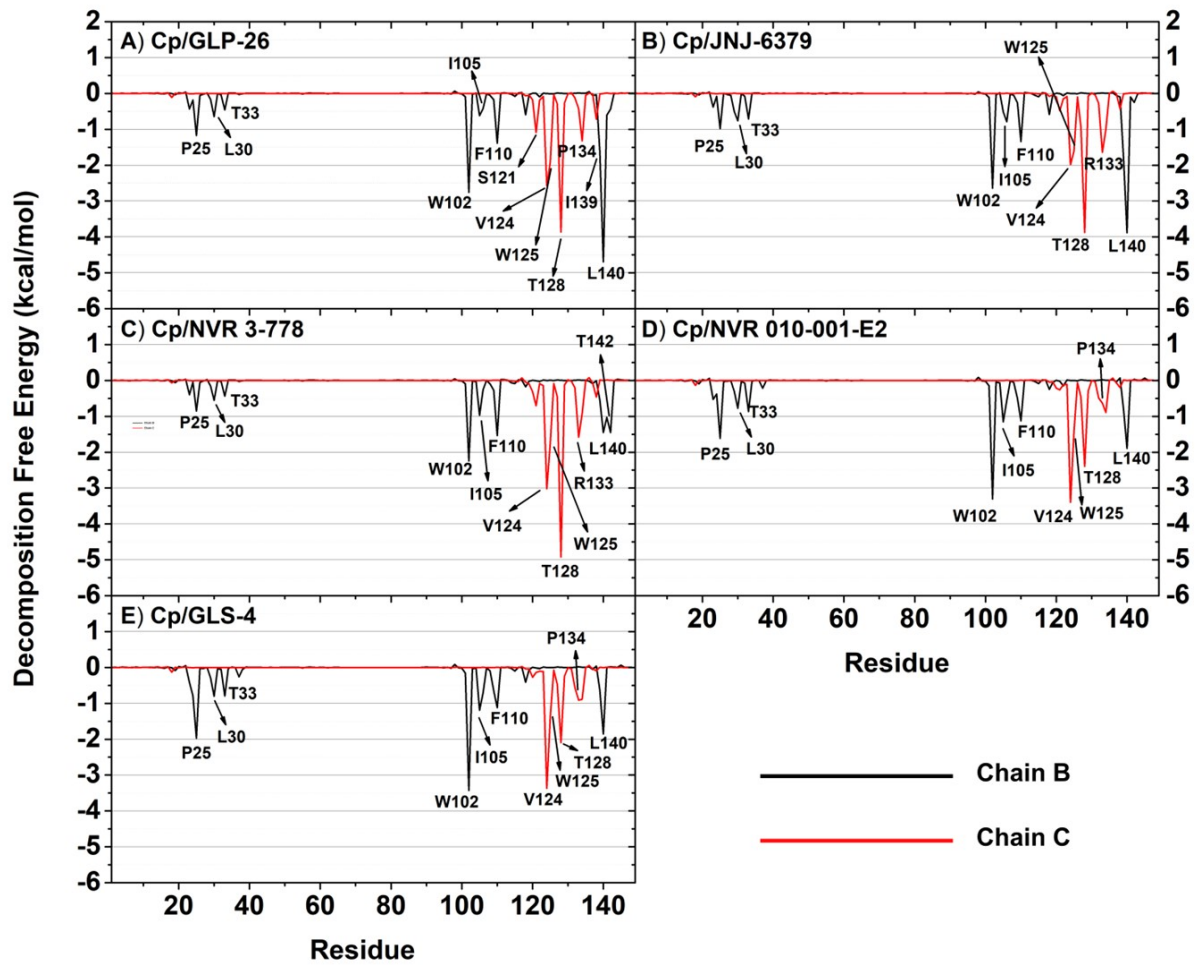
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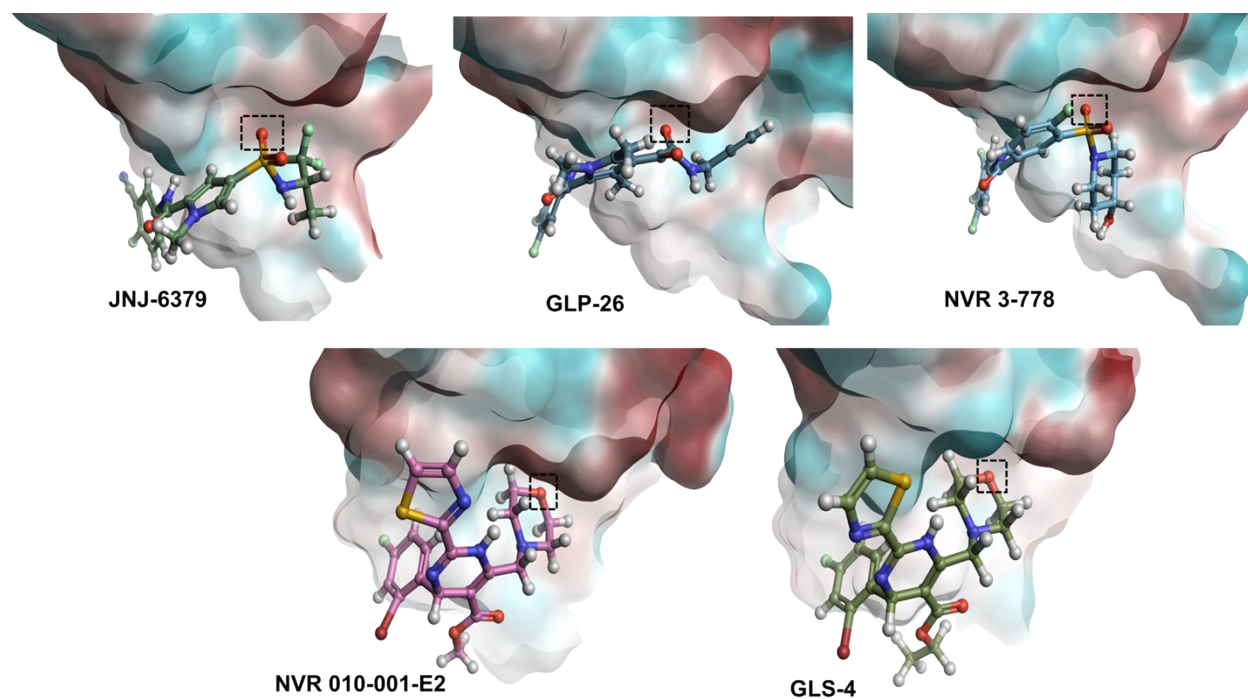
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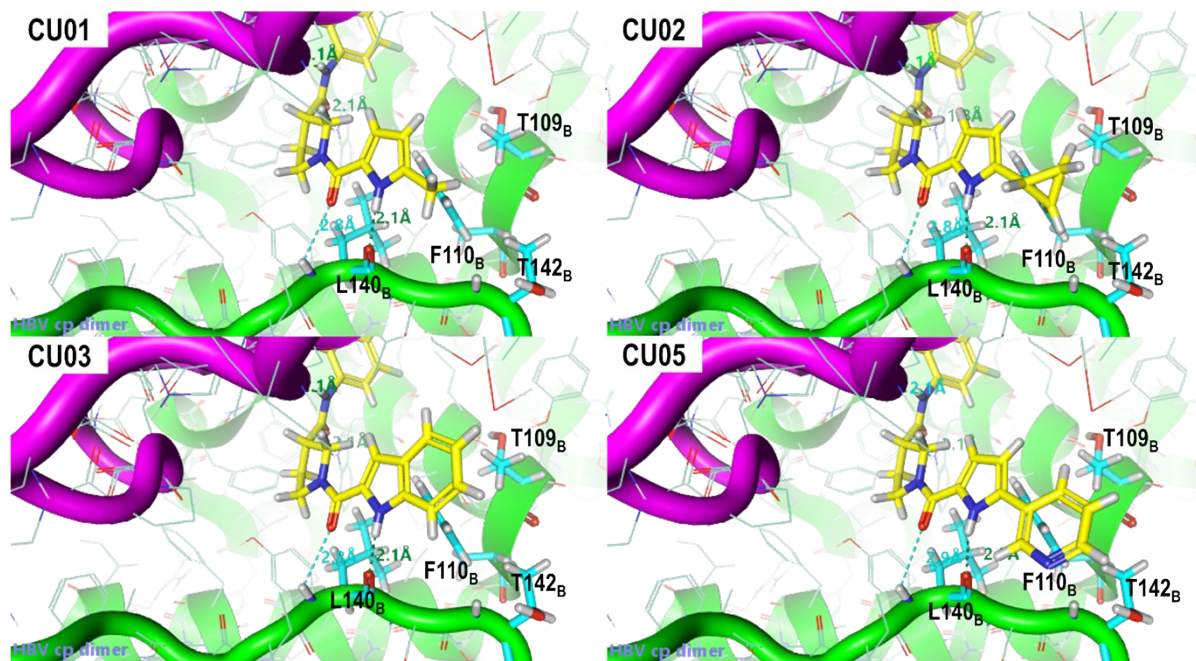
**Figure S1.** (Left) RMSDs for the heavy atom compared to their starting structures of the HBV Cp dimer complexed with A-B) GLP-26, C-D) JNJ-6379, E-F) NVR 3-778, G-H) NVR 010-001-E2, and I-J) GLS-4 (black = global structure, red = chain B backbone, green = chain C backbone, and blue = ligand). (Right) RMSFs of alpha-carbons in the HBV Cp dimer (black = chain B, red = chain C) complexed with A-B) GLP-26, C-D) JNJ-6379, E-F) NVR 3-778, G-H) NVR 010-001-E2, and I-J) GLS-4. Two independent simulations are illustrated for each protein-ligand complex.



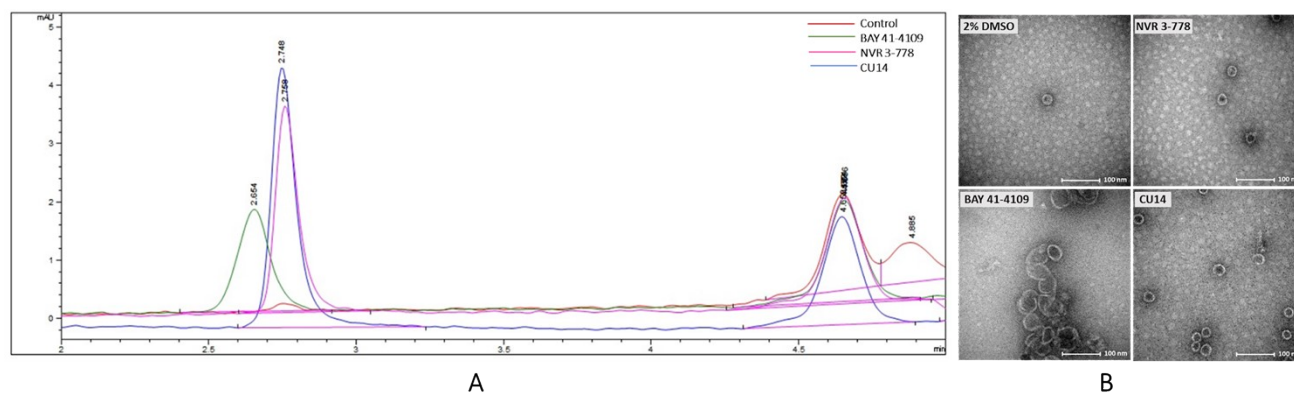
**Figure S2.**  $\Delta G_{bind}^{residue}$  of the HBV Cp dimer complexed with A) GLP-26, B) JNJ-6379, C) NVR 3-778, D) NVR 010-001-E2, and E) GLS-4.



**Figure S3.** ESP surface of five representative ligand-EBV capsid protein (chain C) complexes (red surface = positive, blue surface = negative). The black rectangle highlights the electronegative atom of the ligands that contacted the positive ESP surface (residues W125<sub>C</sub>, R133<sub>C</sub>, and P134<sub>C</sub>).



**Figure S4.** The predicted conformation of CU01-03 and CU05 in HBV capsid dimer interface



**Figure S5.** Effects of representative compound CU14 on HBV capsid assembly, determined by size exclusion chromatography (SEC, A) and transmission electron microscopy (TEM, B).

**Table S1.**  $\Delta G_{bind}^{residue}$  (kcal/mol) of the essential binding residues in HBV Cp dimer in complexation with five representative inhibitors.

Residues	GLP-26		JNJ-6379		NVR 3-778		NVR 010-001-E2		GLS-4		Binding region
	Nonpolar	Polar	Nonpolar	Polar	Nonpolar	Polar	Nonpolar	Polar	Nonpolar	Polar	
<b>P25<sub>B</sub></b>	-1.288	0.117	-1.086	0.104	-1.025	0.169	-1.715	0.104	-2.064	0.092	L1
<b>W102<sub>B</sub></b>	-1.320	-1.439	-1.783	-0.857	-1.499	-0.747	-2.323	-0.977	-2.269	-1.155	L1-L2
<b>I105<sub>B</sub></b>	-0.674	0.053	-0.682	0.164	-0.736	-0.231	-1.040	-0.113	-1.141	-0.053	L1
<b>F110<sub>B</sub></b>	-1.539	0.155	-1.724	0.385	-1.839	0.306	-1.347	0.221	-1.372	0.251	L2
<b>S121<sub>C</sub></b>	-1.105	0.021	-0.908	0.425	-1.131	0.425	-0.605	0.344	-0.369	0.233	L3
<b>V124<sub>C</sub></b>	-2.625	0.055	-2.232	0.246	-2.839	-0.177	-3.373	-0.023	-3.427	0.062	L1
<b>W125<sub>C</sub></b>	-1.630	-0.285	-1.584	-0.024	-1.843	0.102	-1.389	-0.029	-1.321	-0.054	L3
<b>T128<sub>C</sub></b>	-2.753	-1.107	-2.152	-1.726	-2.337	-2.581	-2.356	-0.037	-2.167	0.072	L1-L2
<b>R133<sub>C</sub></b>	-0.743	0.315	-1.107	-0.533	-1.013	-0.565	-0.815	0.183	-1.022	0.112	L3
<b>P134<sub>C</sub></b>	-1.536	0.218	-1.114	0.114	-0.984	0.059	-0.981	0.086	-0.942	0.055	L3
<b>I139<sub>B</sub></b>	-1.293	-0.339	-1.206	-0.606	-0.571	0.025	-0.900	0.094	-0.758	0.120	L3
<b>L140<sub>B</sub></b>	-3.075	-1.617	-2.719	-1.160	-2.704	1.254	-2.567	0.671	-2.594	0.738	L2-L3
<b>S141<sub>B</sub></b>	-0.920	0.311	-0.768	0.671	-0.915	-0.115	-0.166	0.149	-0.161	0.164	L3
<b>T142<sub>B</sub></b>	-0.472	0.033	-0.467	0.219	-0.199	-1.251	-0.028	0.059	-0.040	0.061	L3
<b>W125<sub>C</sub>, R133<sub>C</sub>, and P134<sub>C</sub></b>	-3.91	0.25	-3.81	-0.443	-3.84	-0.40	-3.19	0.24	-3.29	0.113	L3 small pocket

Polar ( $\Delta E_{ele} + \Delta G_{polar}$ ) and nonpolar ( $\Delta E_{vdW} + \Delta G_{nonpolar}$ ) energy contributions were analyzed for selected residues.

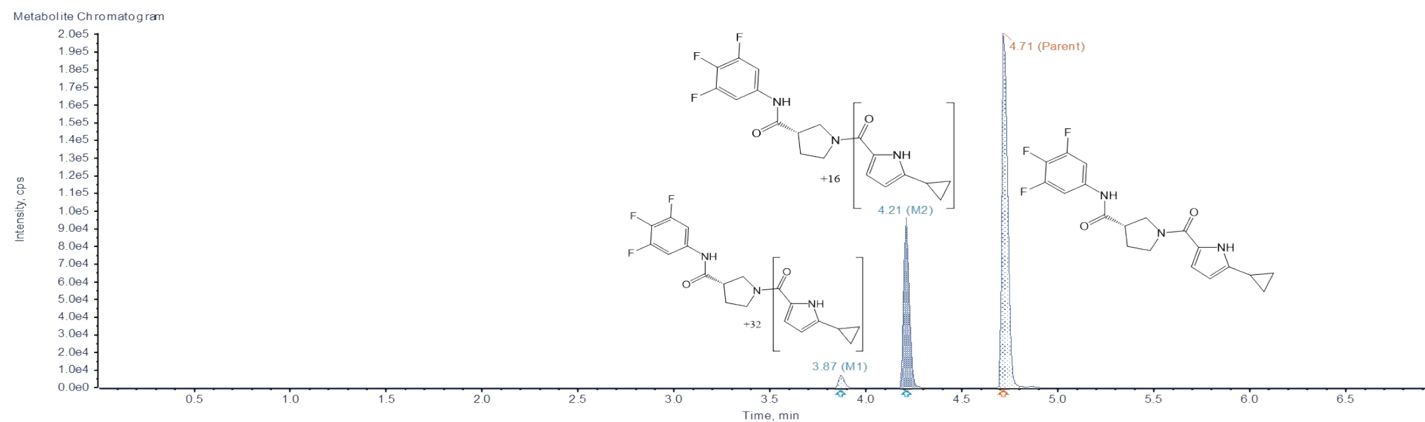
## **METABOLITE IDENTIFICATION**

### *Identification of metabolic soft spots using high resolution mass spectrometry*

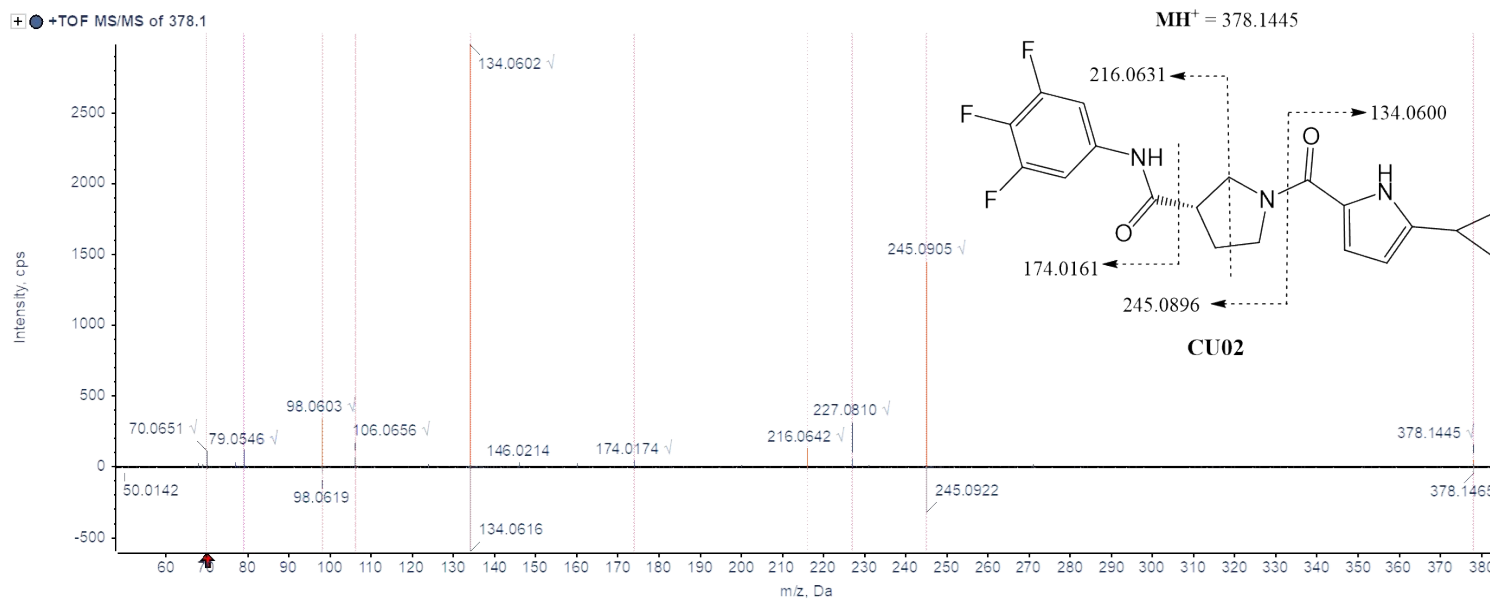
The metabolite identification was conducted using the AB Sciex X500B Q-TOF LC-MS/MS system. The samples were chromatographically separated using the Exion LC AD100 system with a Phenomenex Kinetex® C-18, 1.7  $\mu\text{m}$ , 50 $\times$ 2.1 mm analytical column. The mobile phases consisted of water and acetonitrile, both fortified with 0.1% formic acid, and were used as phase A and B, respectively. The flow rate was set at 0.4 mL/min using a 2-step linear gradient from 2% to 65% B for 4 min, followed by ramping to 95% for 0.5 min. The Turbo V™ ion source was operated in the electrospray positive ionization mode under following conditions: curtain gas at 40 psi, heating gas 1 at 40 psi, nebulizing gas 2 at 40 psi, temperature at 450 °C, ion spray voltage floating at 4.5 kV, and declustering potential at 80 V. Mass calibration was performed every 5 sample injections using a standard calibration solution. The TOF MS scan and product ion (MS/MS) spectra were acquired for the m/z range 50-800.

Dynamic background subtraction and the mass defect filter (MDF) function were enabled. The MDF window was set at  $\pm 40$  mDa, and the mass range was set as  $\pm 40$  Da around the m/z value of the parent ion. An information-dependent data acquisition (IDA) method was used for triggering MS/MS acquisition for the top 6 intense ions using CE spread at  $35 \pm 15$  V. The data were processed and interpreted using SCIEX OS version 1.6 and MetabolitePilot™ version 2.0 software (AB SCIEX, MA, USA).

## Ion chromatograms of parent CU02 and its metabolites

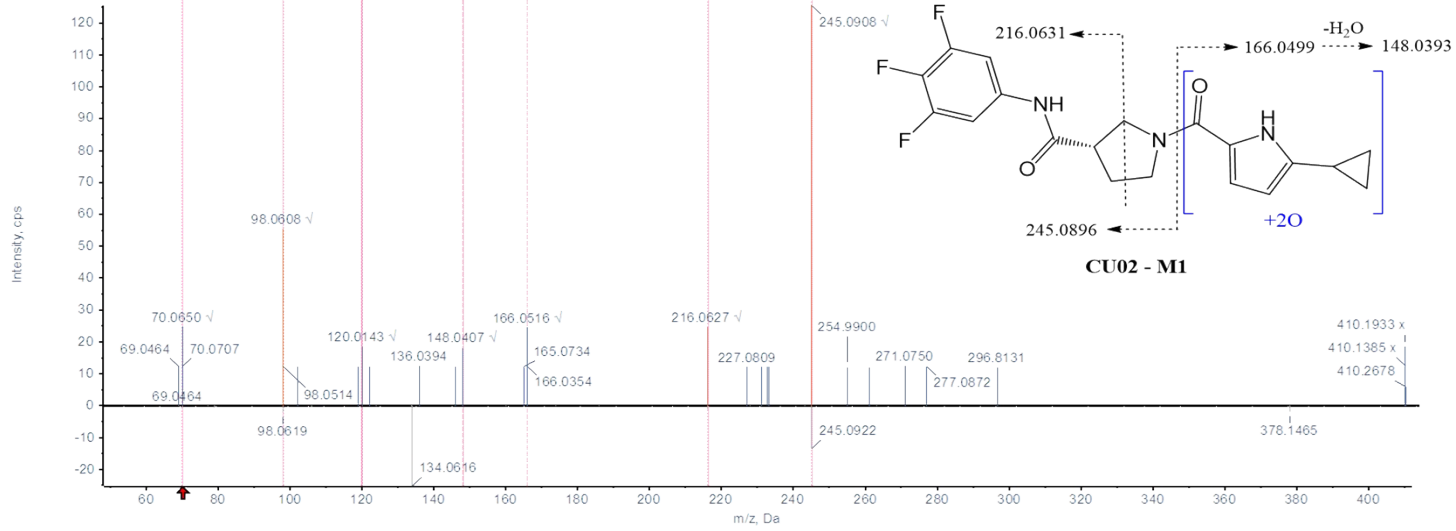


## Structures and fragmentation spectra of parent CU02 and its metabolites

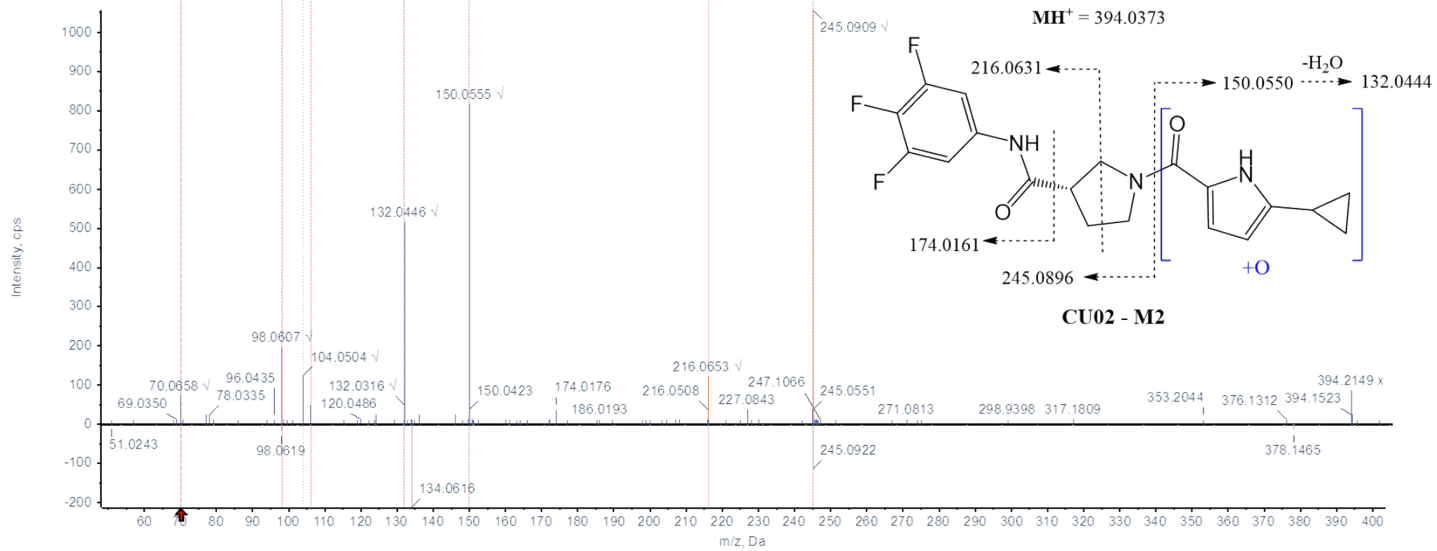




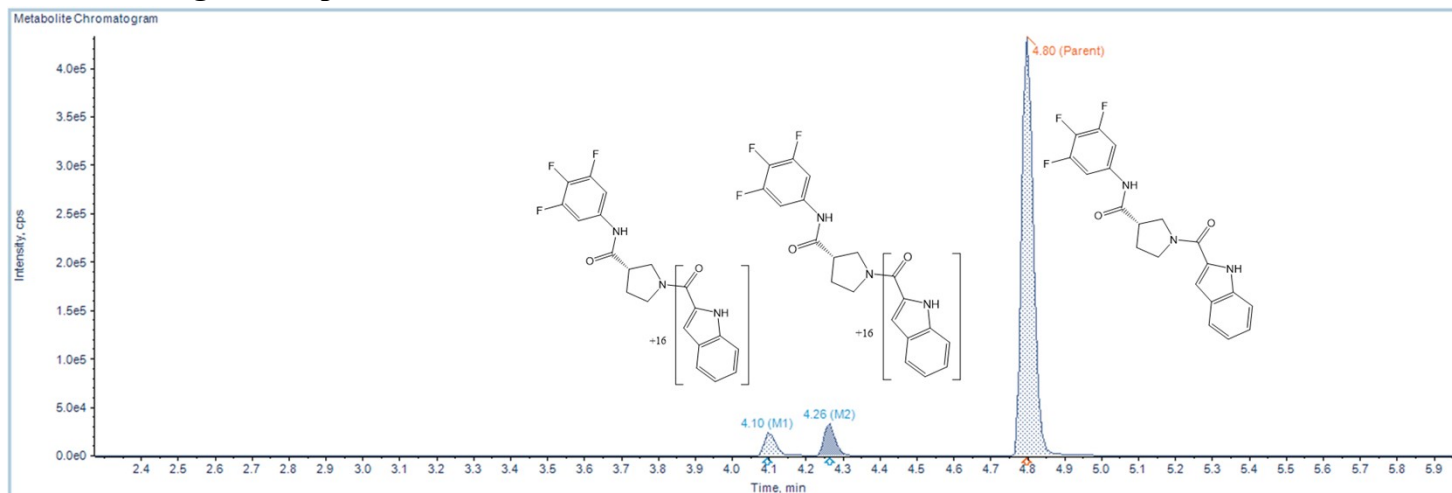
+●+TOF MS/MS of 410.1



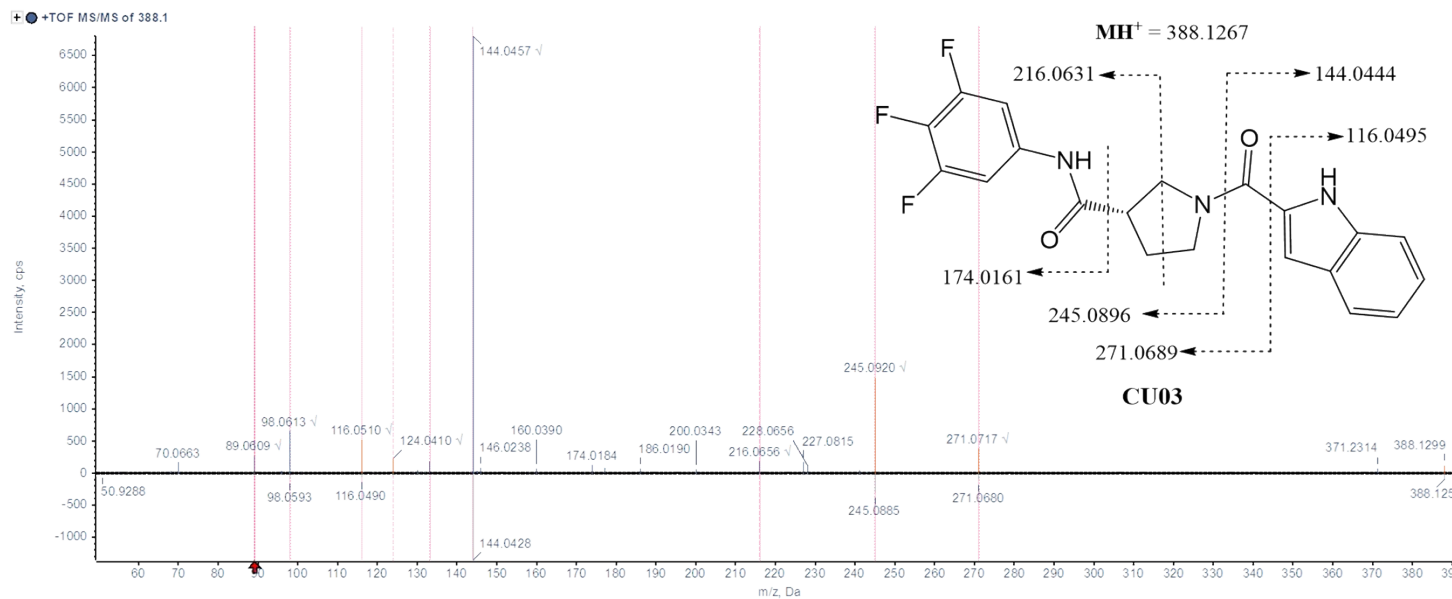
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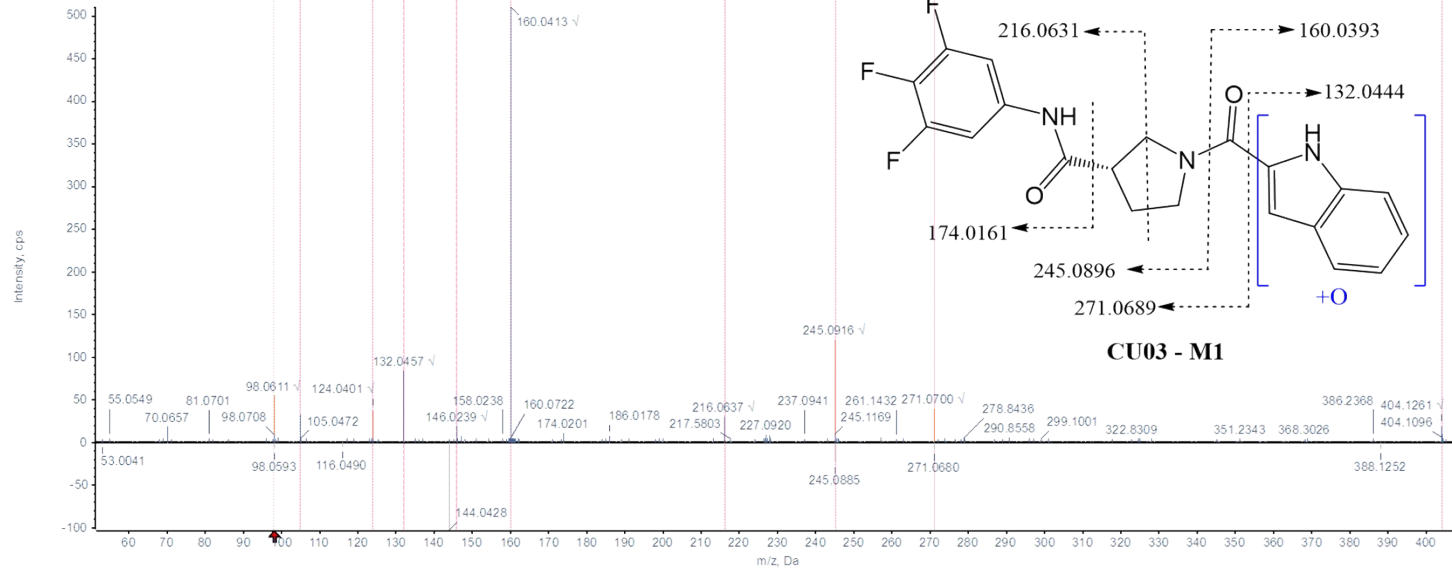
## Ion chromatograms of parent CU03 and its metabolites



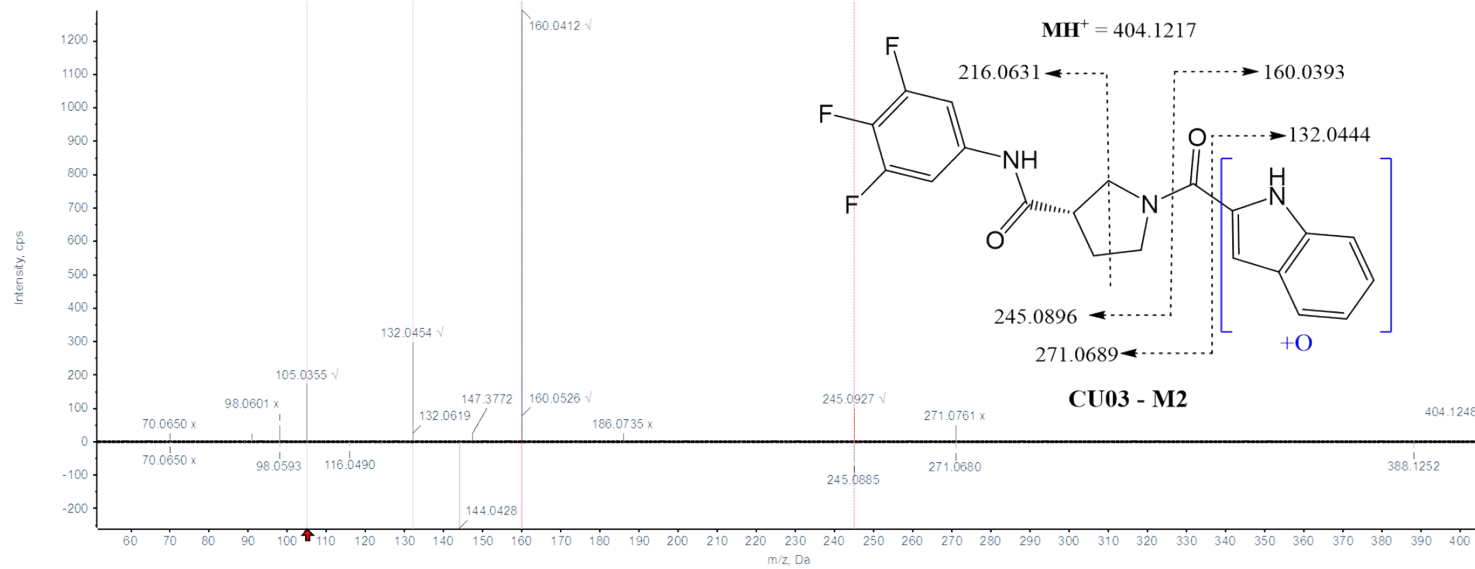
## Structures and fragmentation spectra of parent CU03 and its metabolites



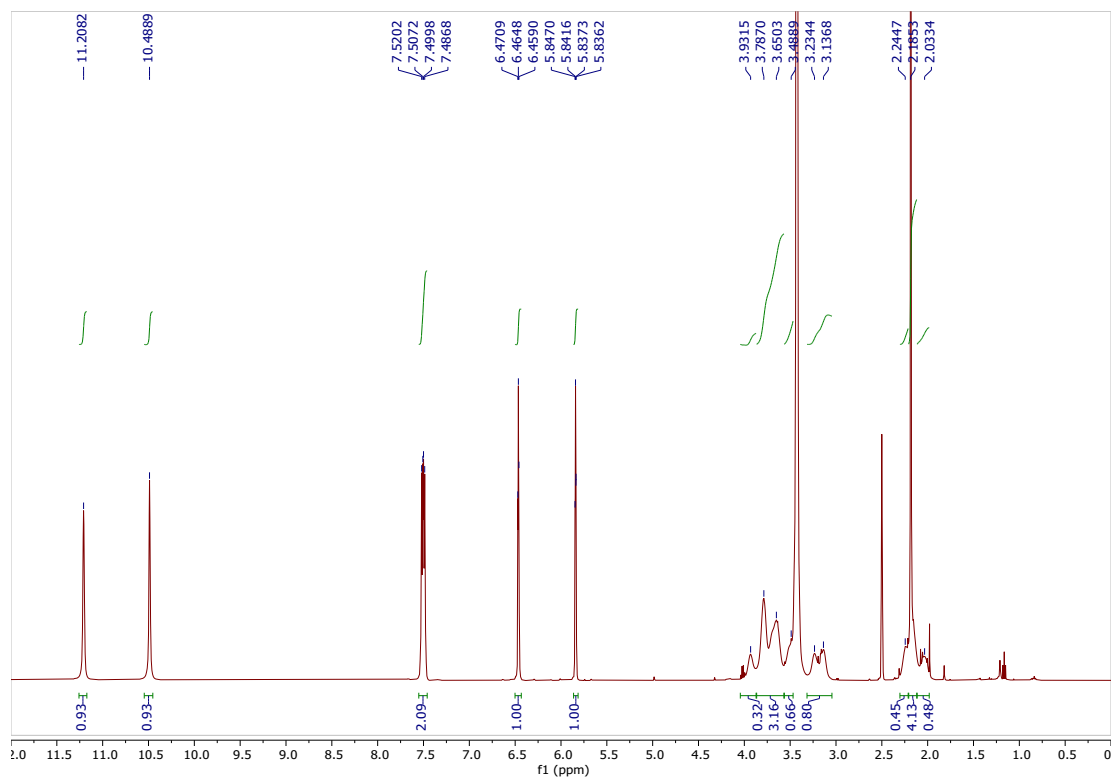
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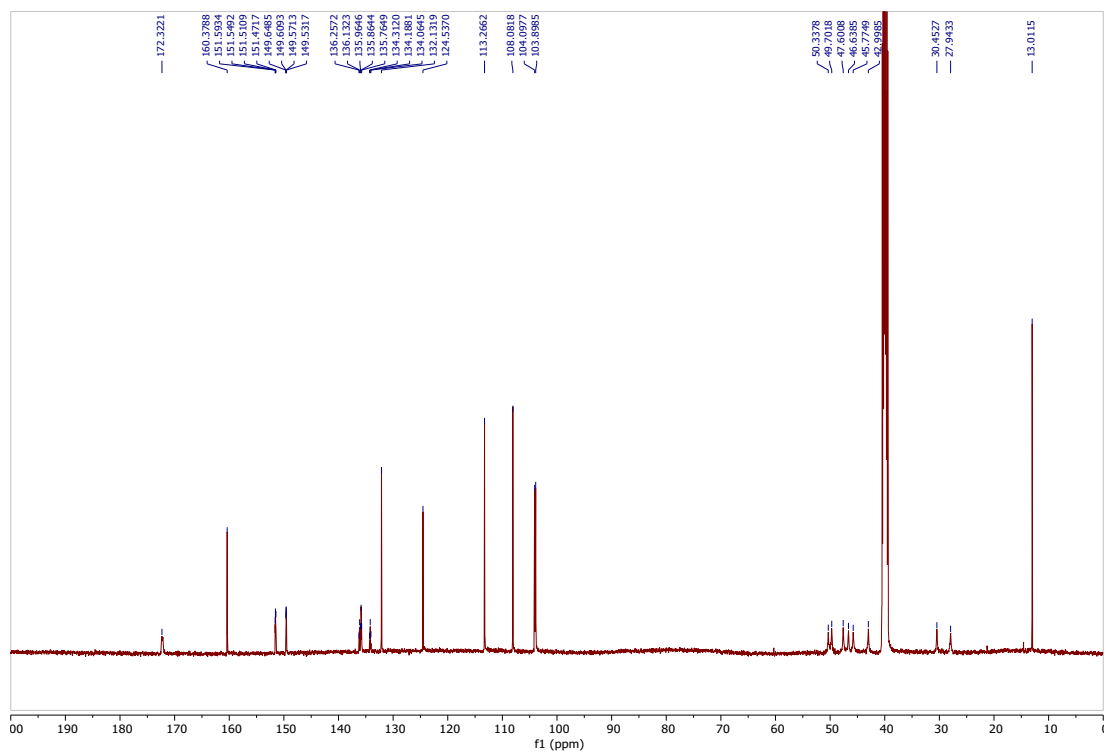
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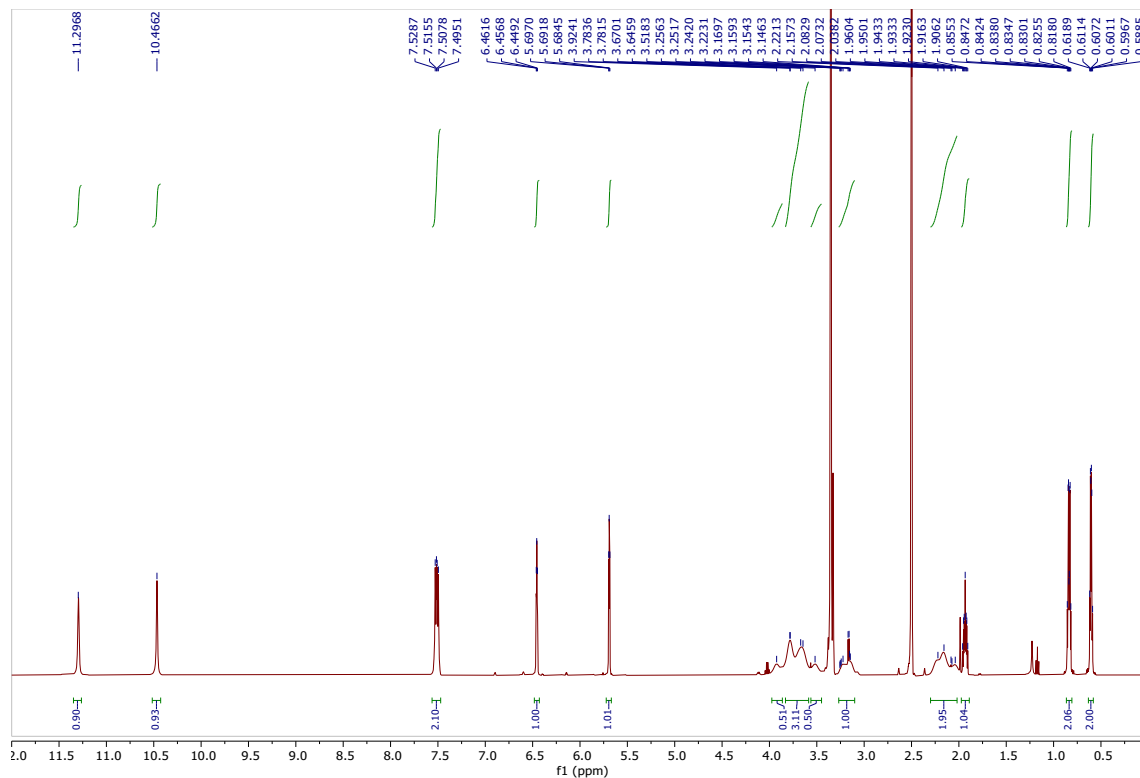
# NMR SPECTRUM OF SYNTHESIZED COMPOUNDS



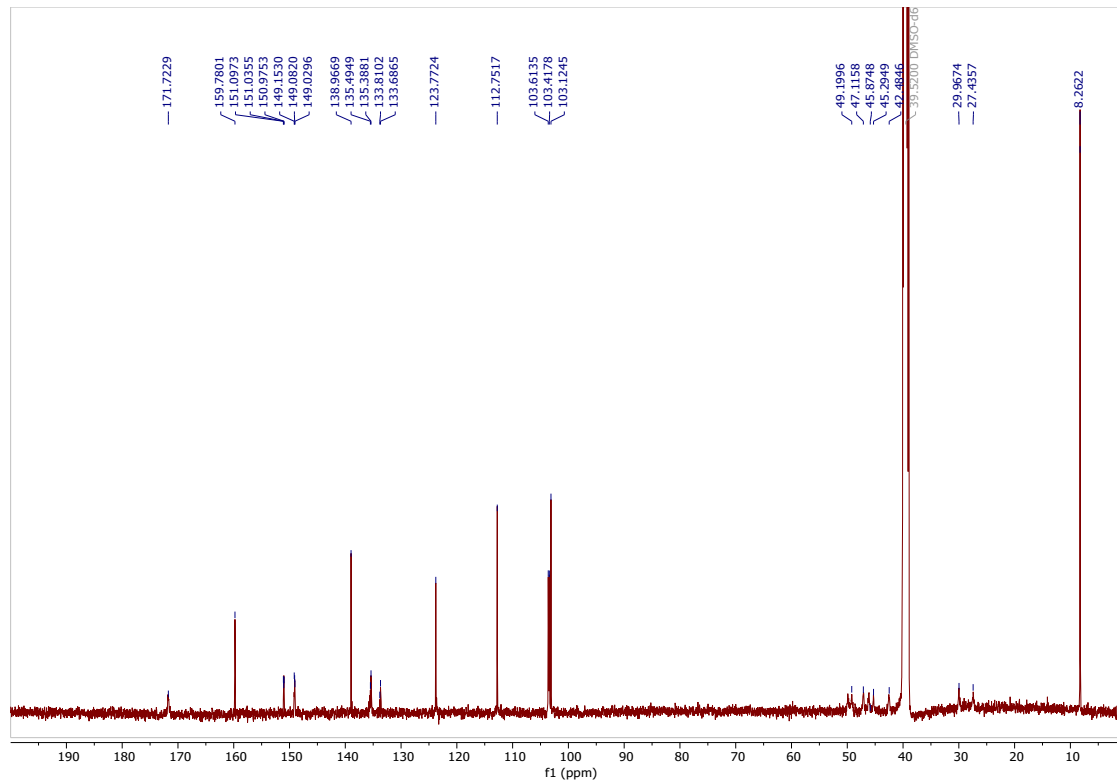
<sup>1</sup>H NMR spectrum of CU01 (DMSO-d<sub>6</sub>)



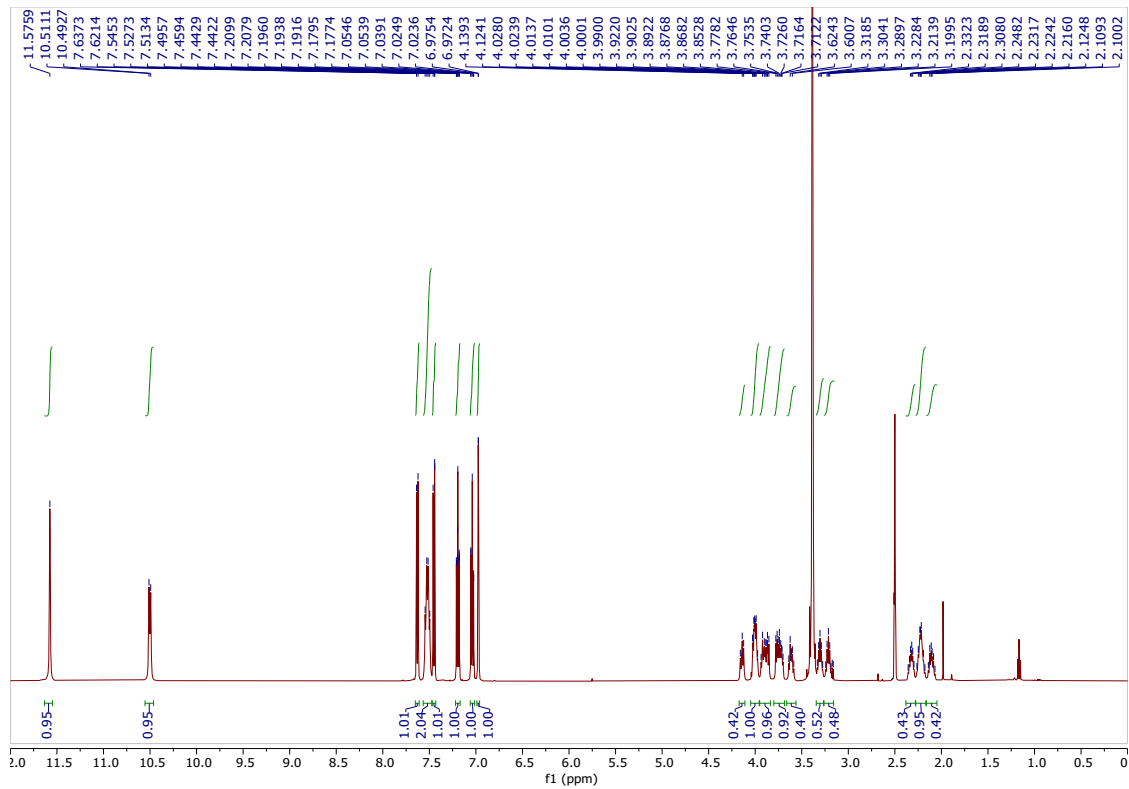
<sup>13</sup>C NMR spectrum of CU01 (DMSO-d<sub>6</sub>)



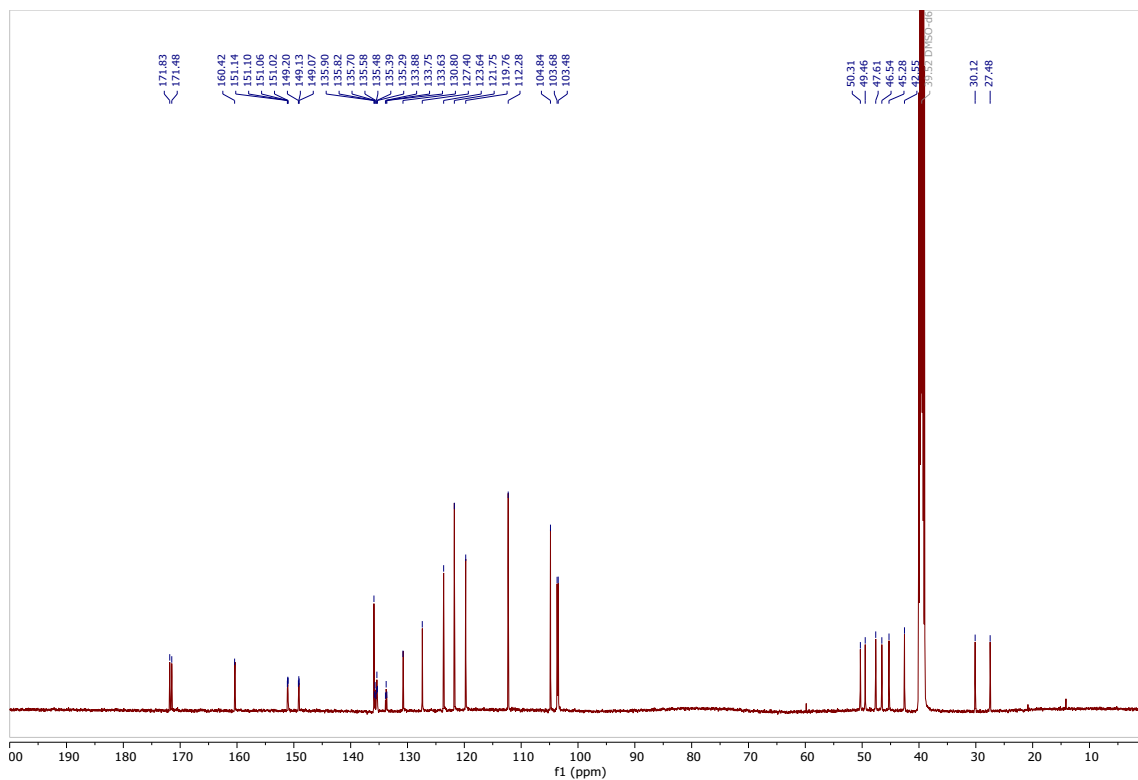
<sup>1</sup>H NMR spectrum of CU02 (DMSO-d<sub>6</sub>)



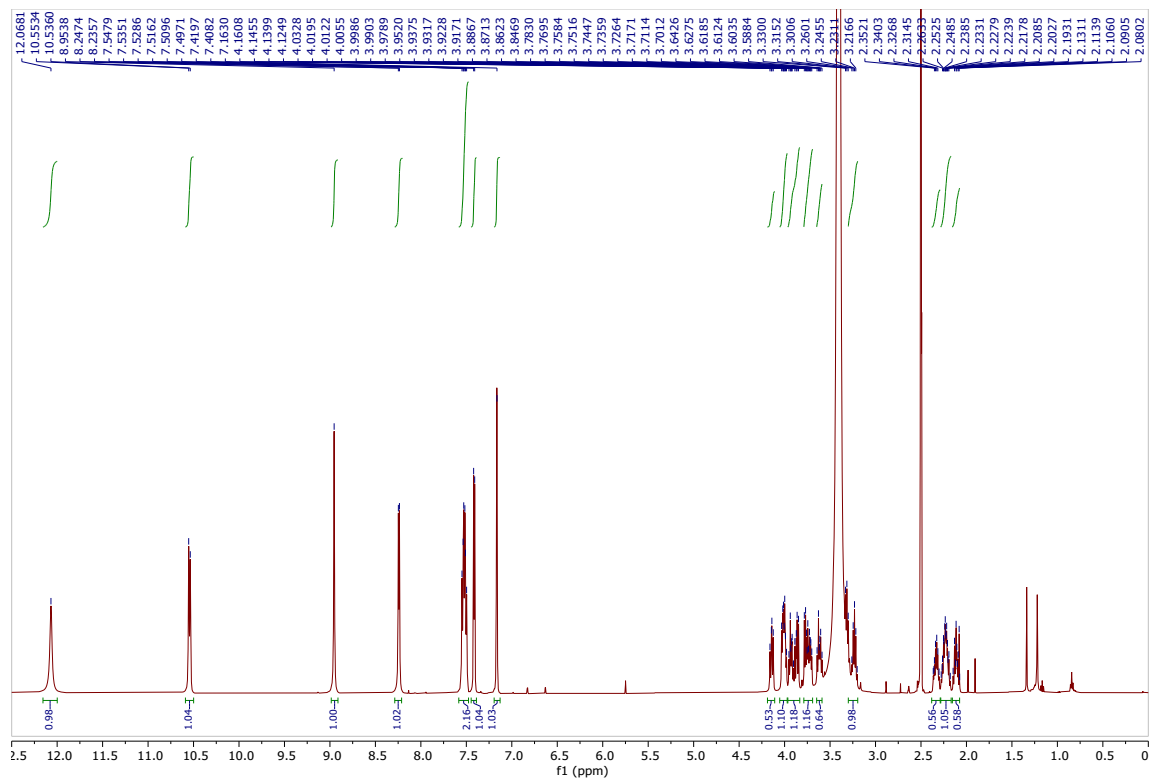
<sup>13</sup>C NMR spectrum of CU02 (DMSO-d<sub>6</sub>)



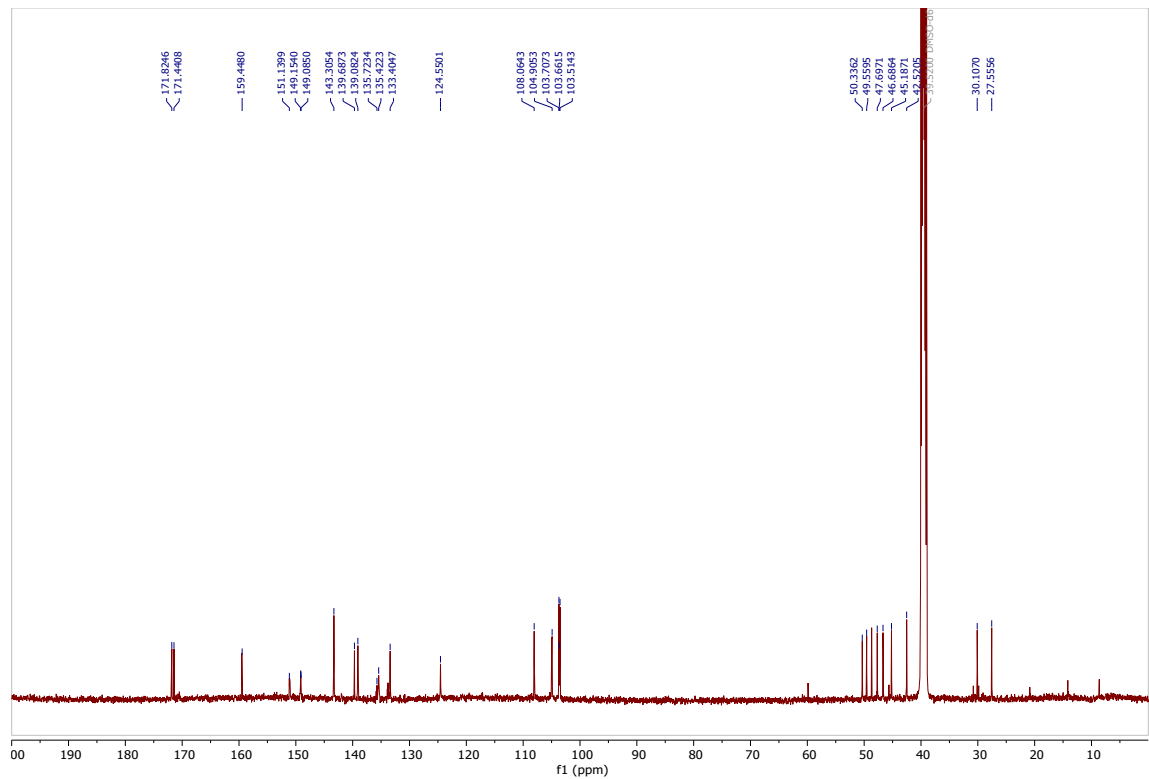
<sup>1</sup>H NMR spectrum of CU03 (DMSO-d<sub>6</sub>)



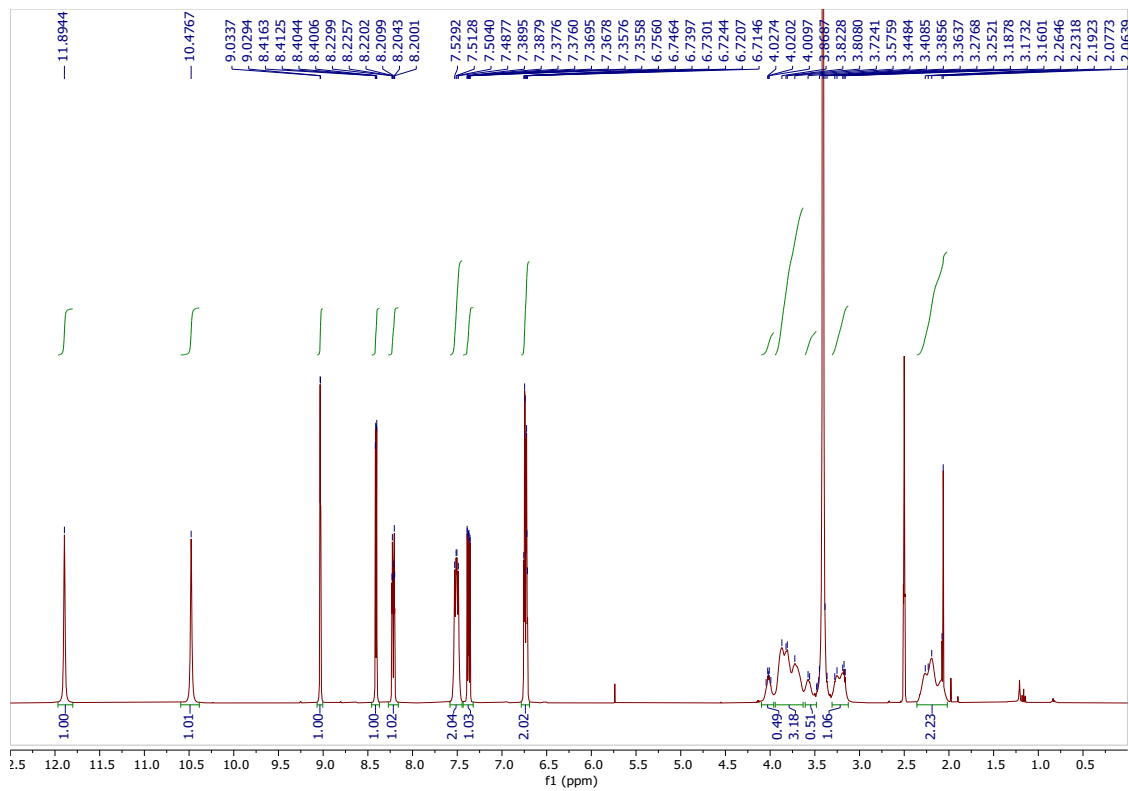
<sup>13</sup>C NMR spectrum of CU03 (DMSO-d<sub>6</sub>)



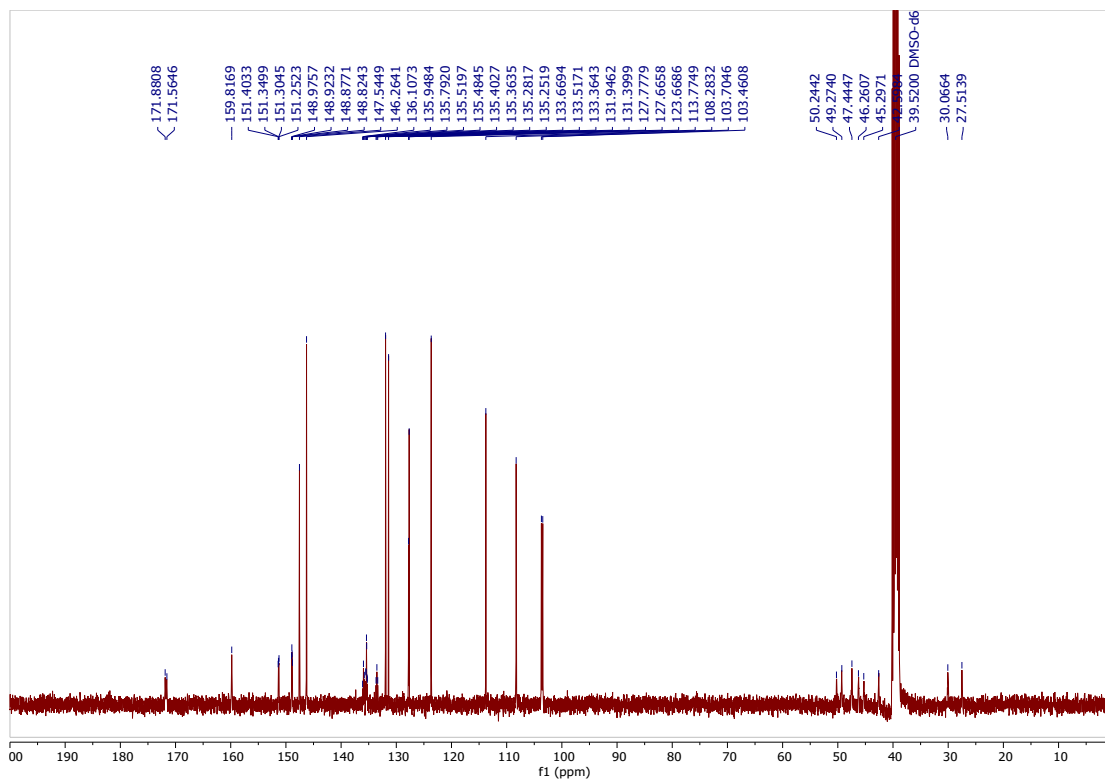
<sup>1</sup>H NMR spectrum of CU04 (DMSO-*d*<sub>6</sub>)



<sup>13</sup>C NMR spectrum of CU04 (DMSO-*d*<sub>6</sub>)

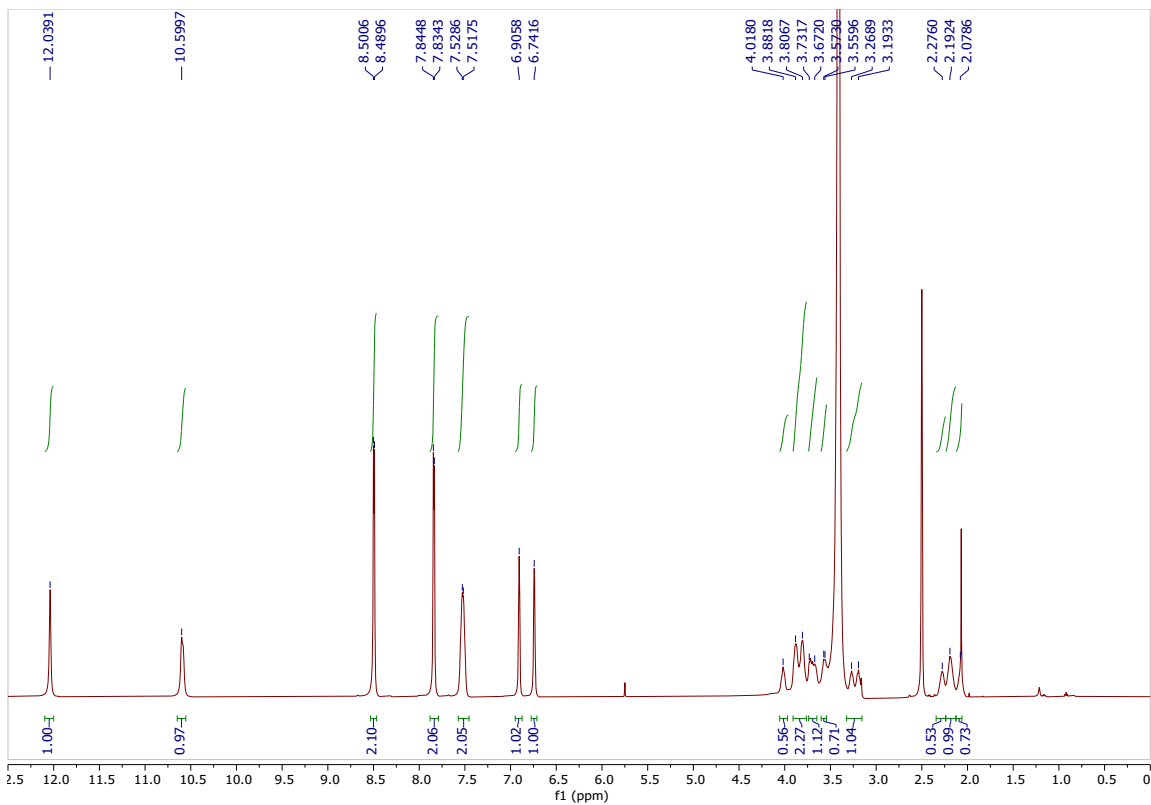


<sup>1</sup>H NMR spectrum of CU05 (DMSO-*d*<sub>6</sub>)

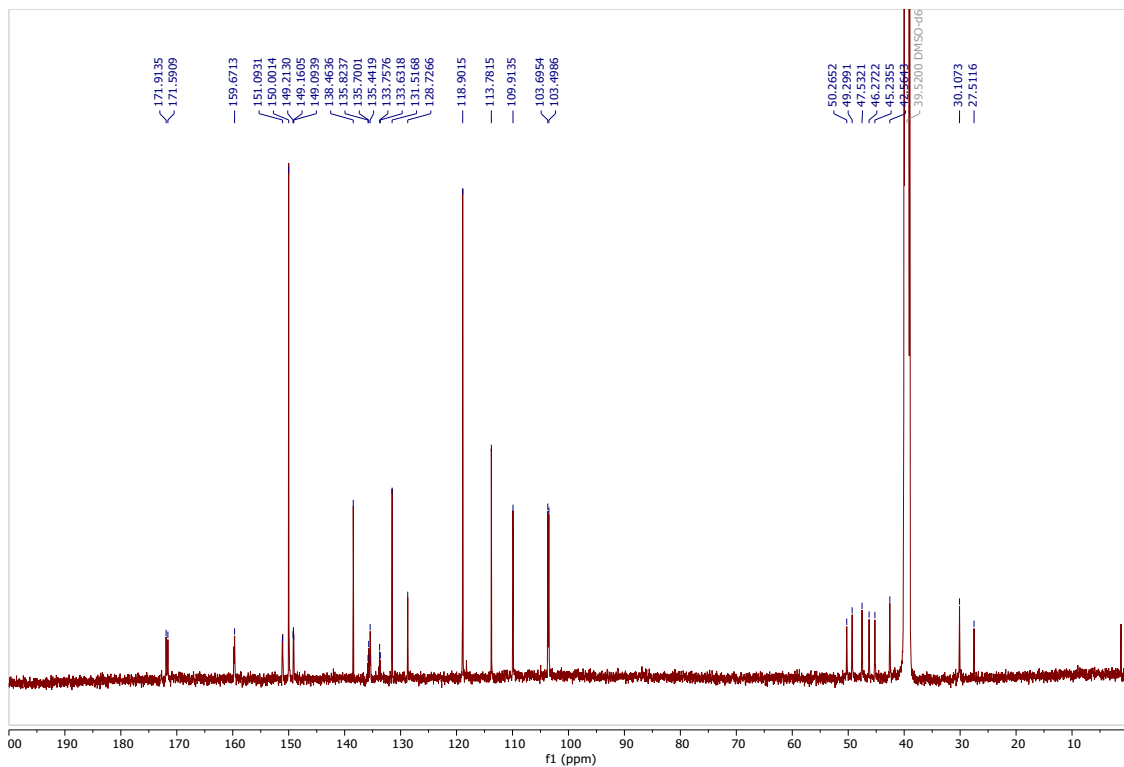


<sup>13</sup>C NMR spectrum of CU05 (DMSO-*d*<sub>6</sub>)

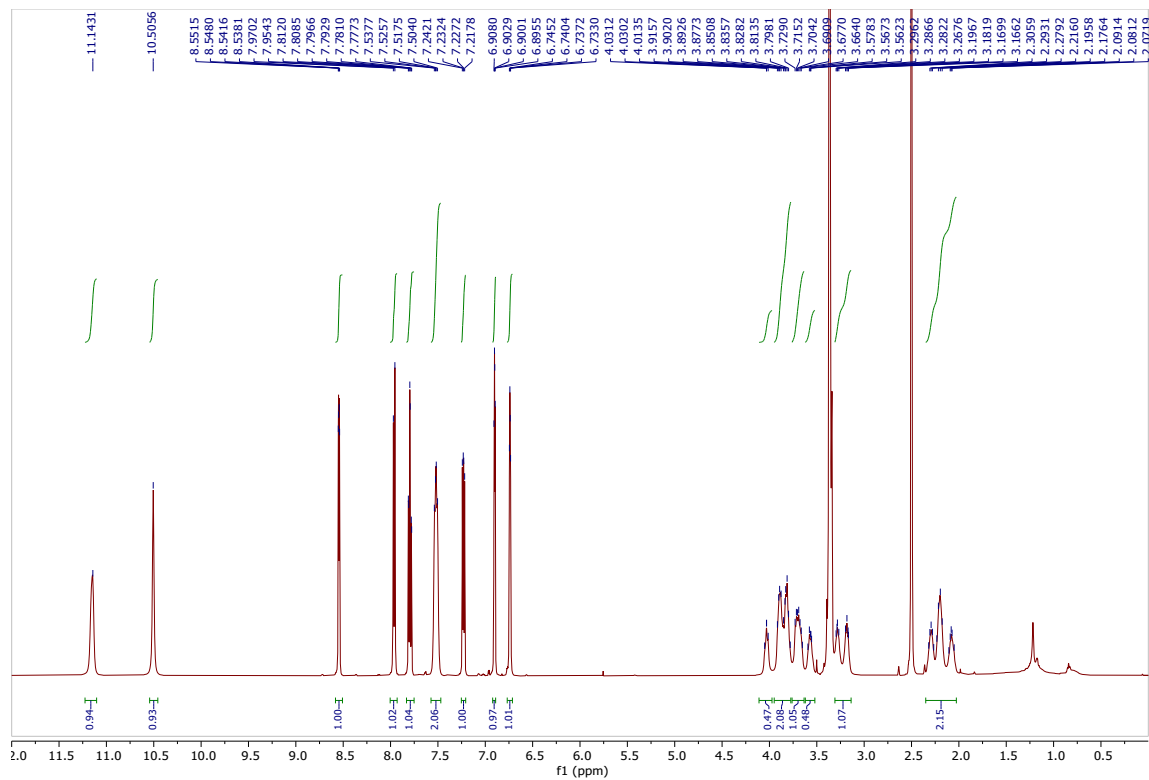




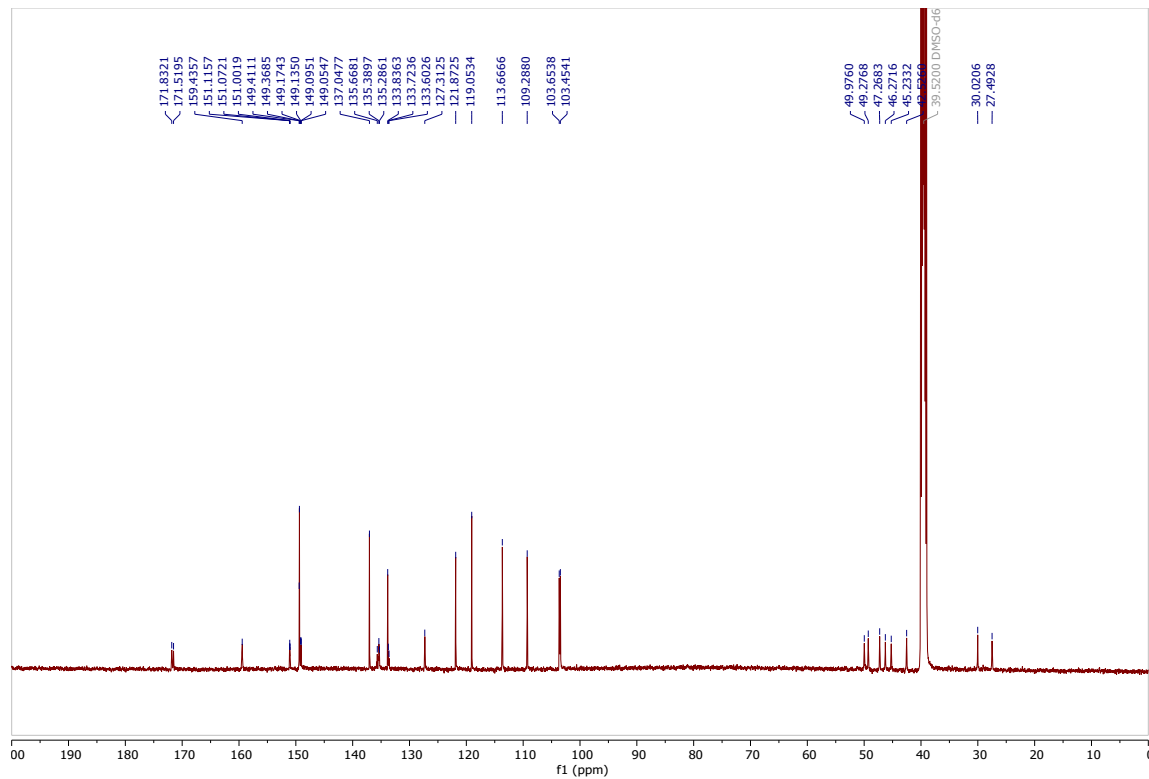
<sup>1</sup>H NMR spectrum of CU06 (DMSO-d<sub>6</sub>)



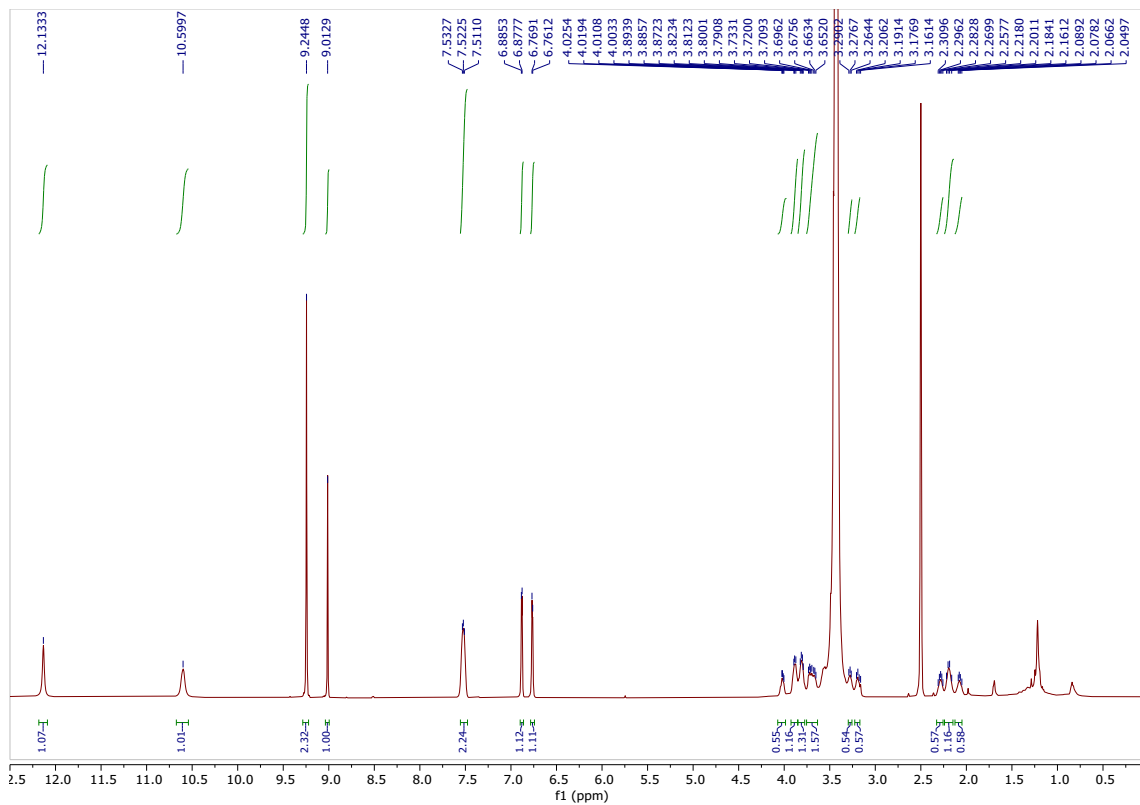
<sup>13</sup>C NMR spectrum of CU06 (DMSO-d<sub>6</sub>)



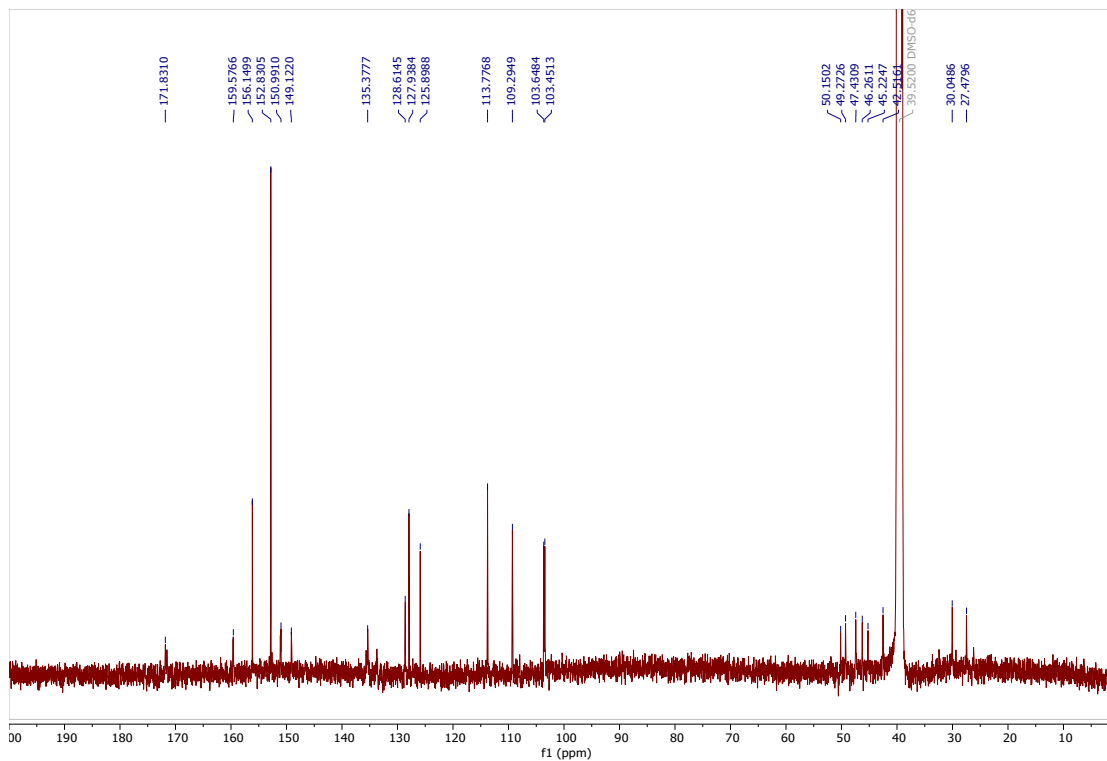
<sup>1</sup>H NMR spectrum of CU07 (DMSO-*d*<sub>6</sub>)



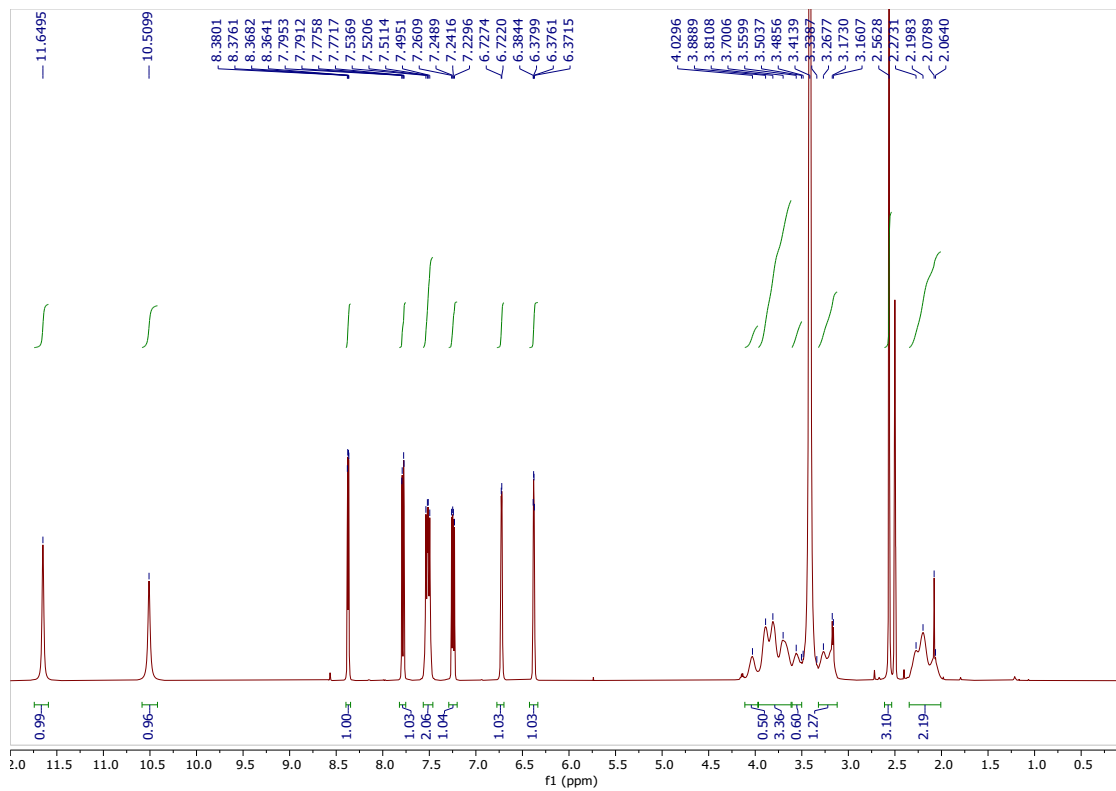
<sup>13</sup>C NMR spectrum of CU07 (DMSO-*d*<sub>6</sub>)



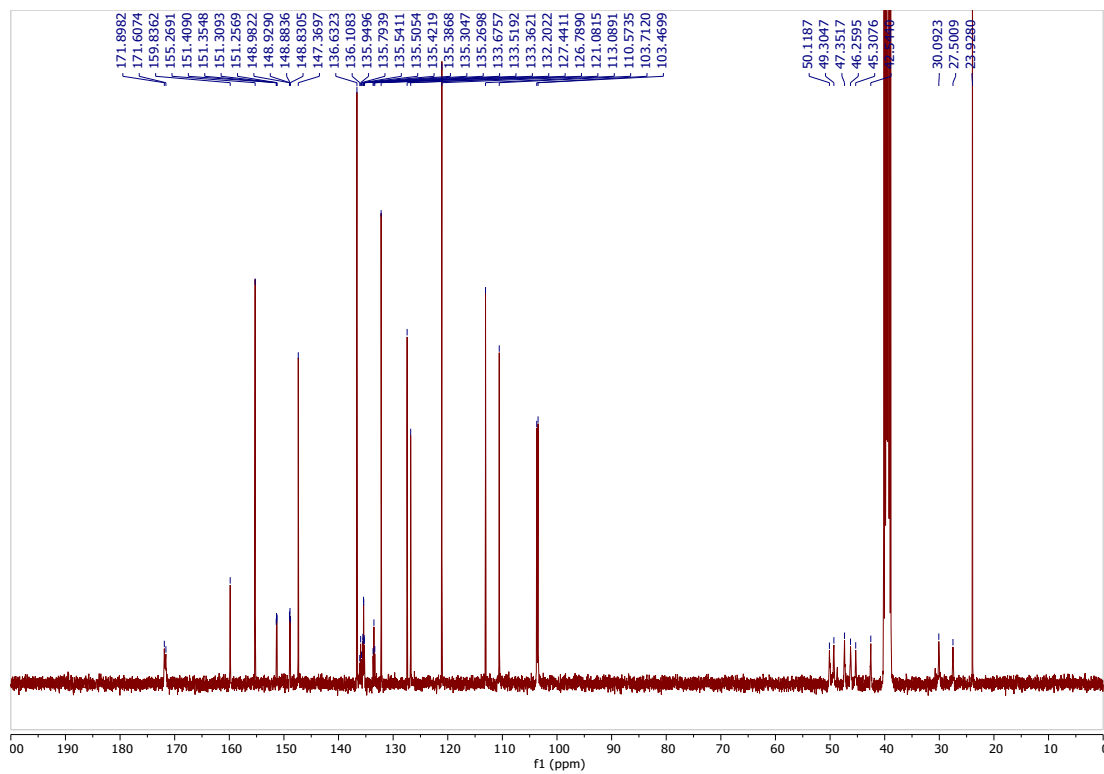
<sup>1</sup>H NMR spectrum of CU08 (DMSO-*d*<sub>6</sub>)



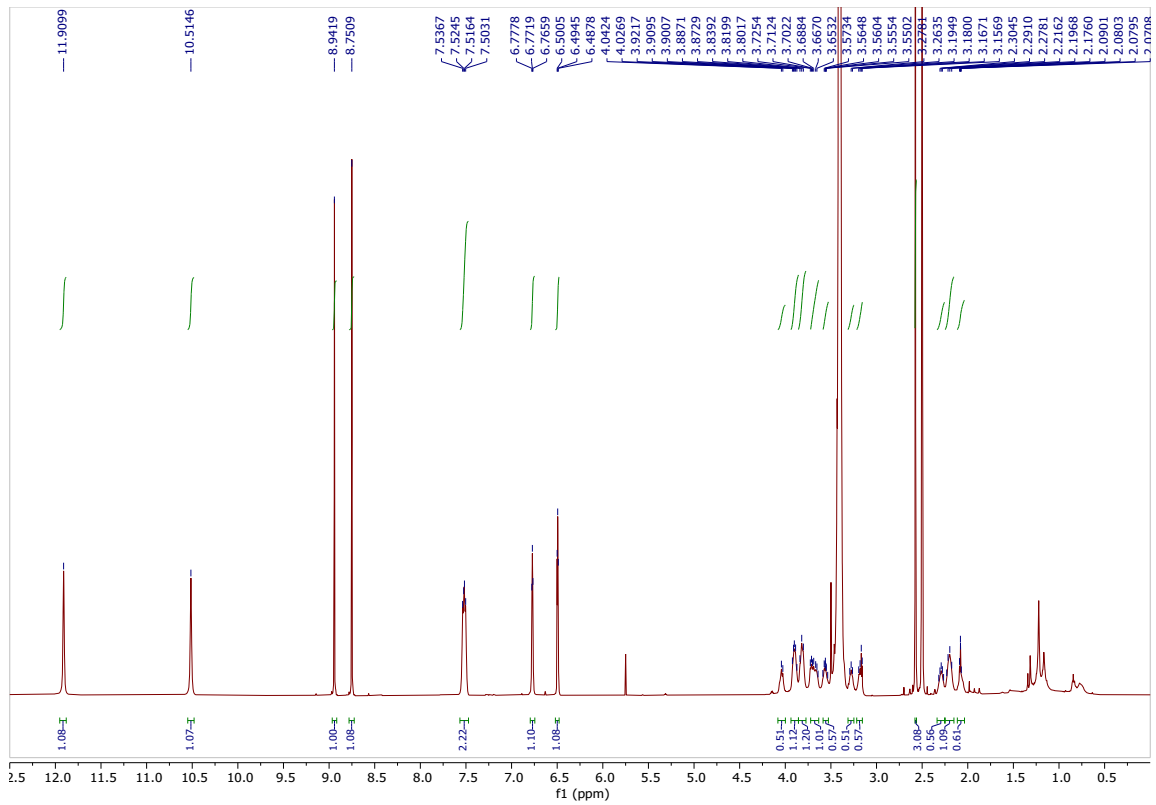
<sup>13</sup>C NMR spectrum of CU08 (DMSO-*d*<sub>6</sub>)



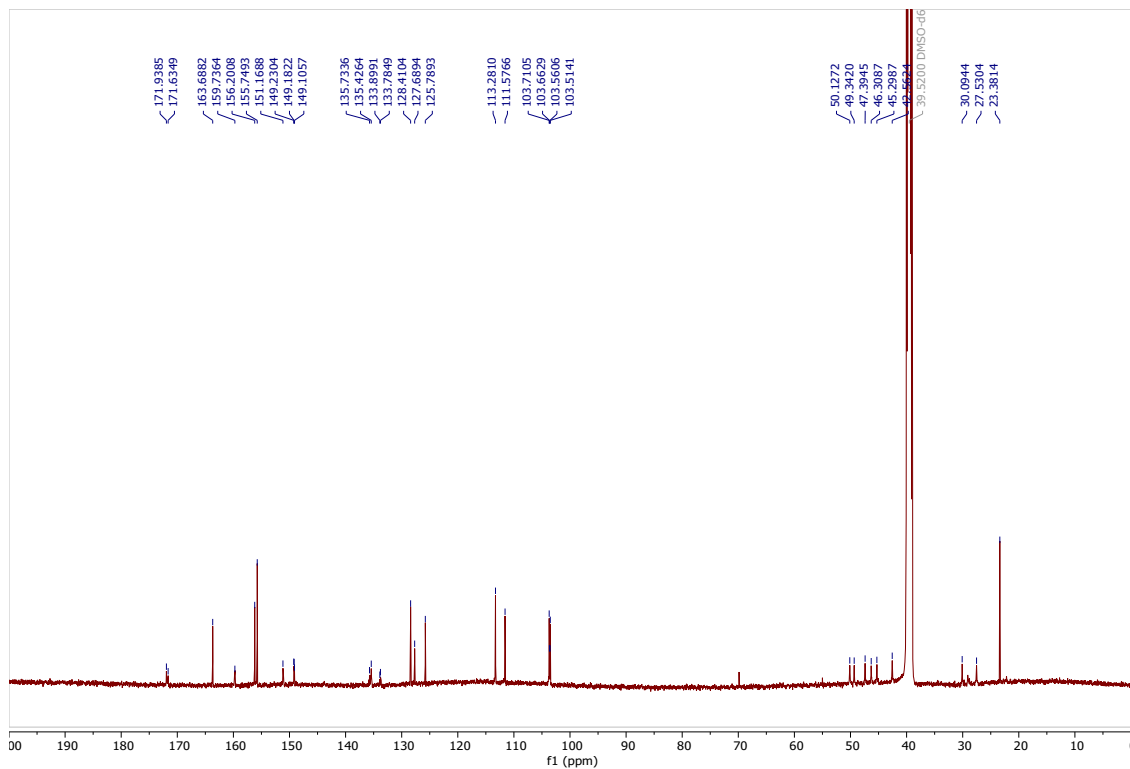
<sup>1</sup>H NMR spectrum of CU09 (DMSO-d<sub>6</sub>)



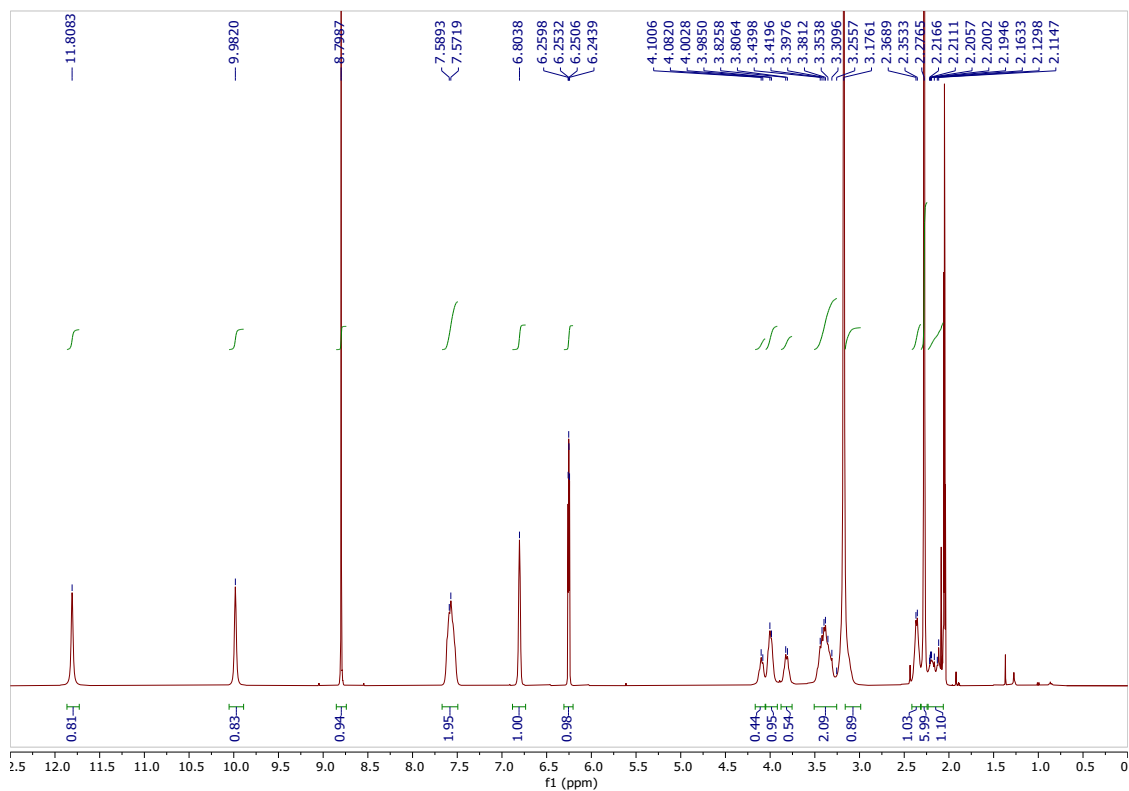
<sup>13</sup>C NMR spectrum of CU09 (DMSO-d<sub>6</sub>)



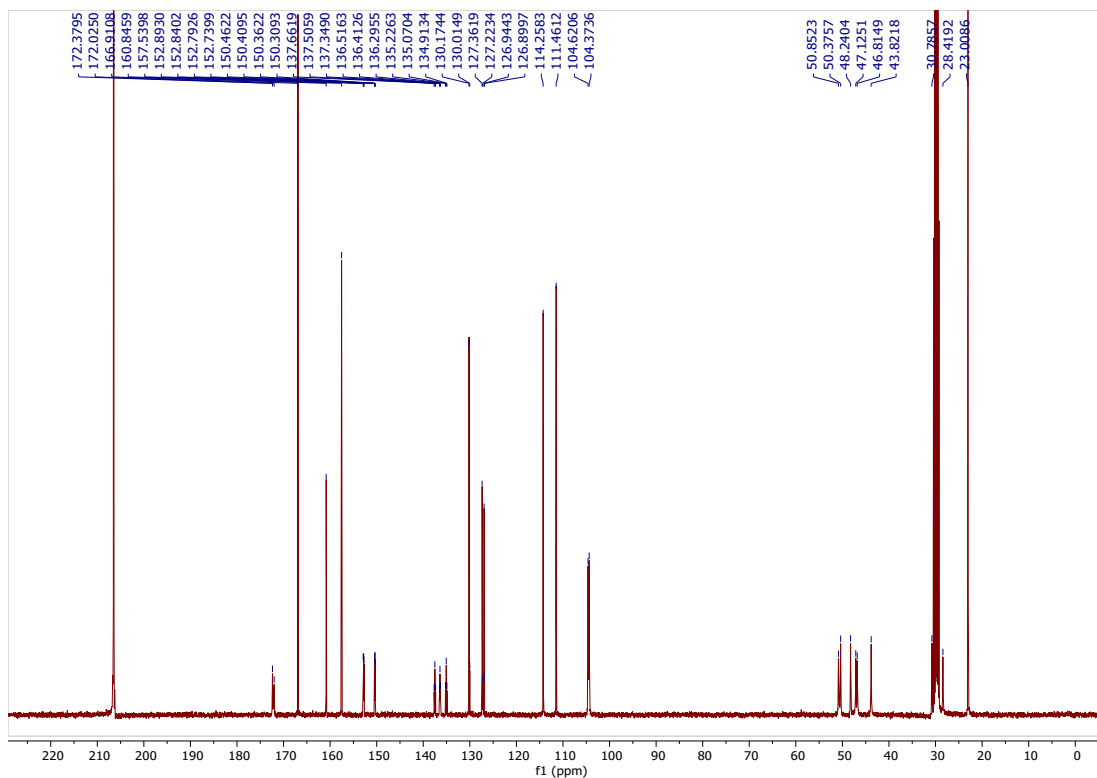
<sup>1</sup>H NMR spectrum of CU10 (DMSO-d<sub>6</sub>)



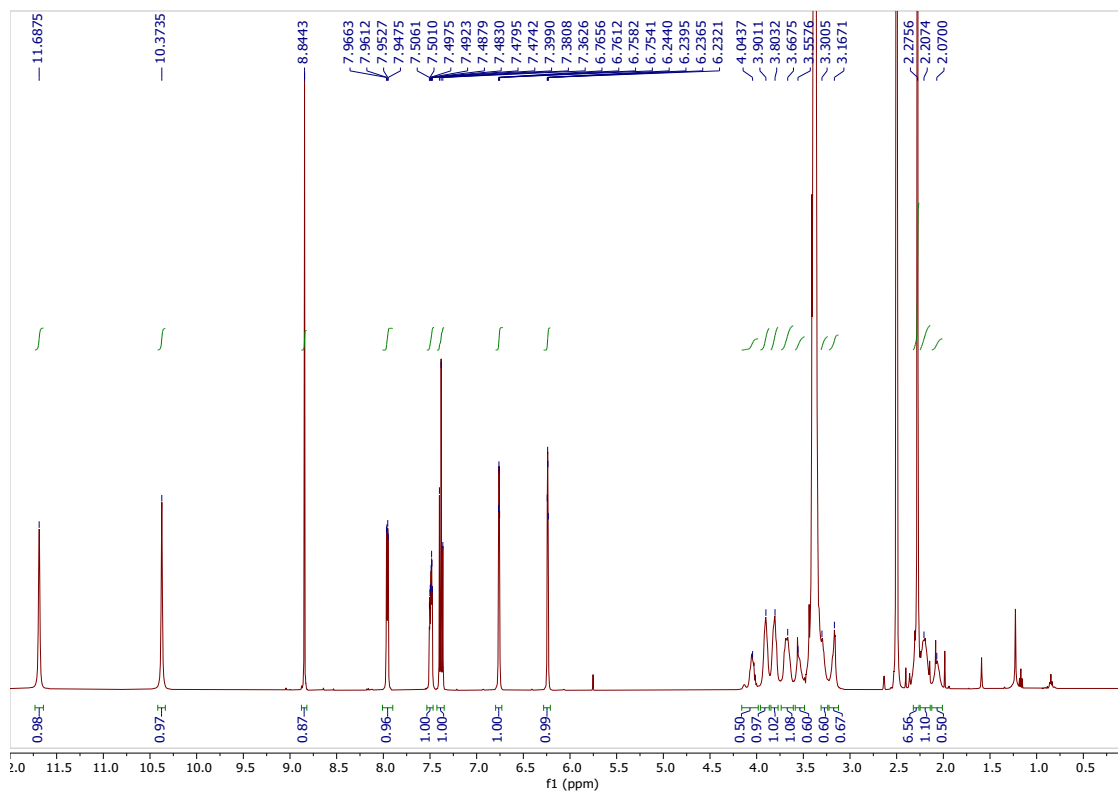
$^{13}\text{C}$  NMR spectrum of CU10 (DMSO- $d_6$ )



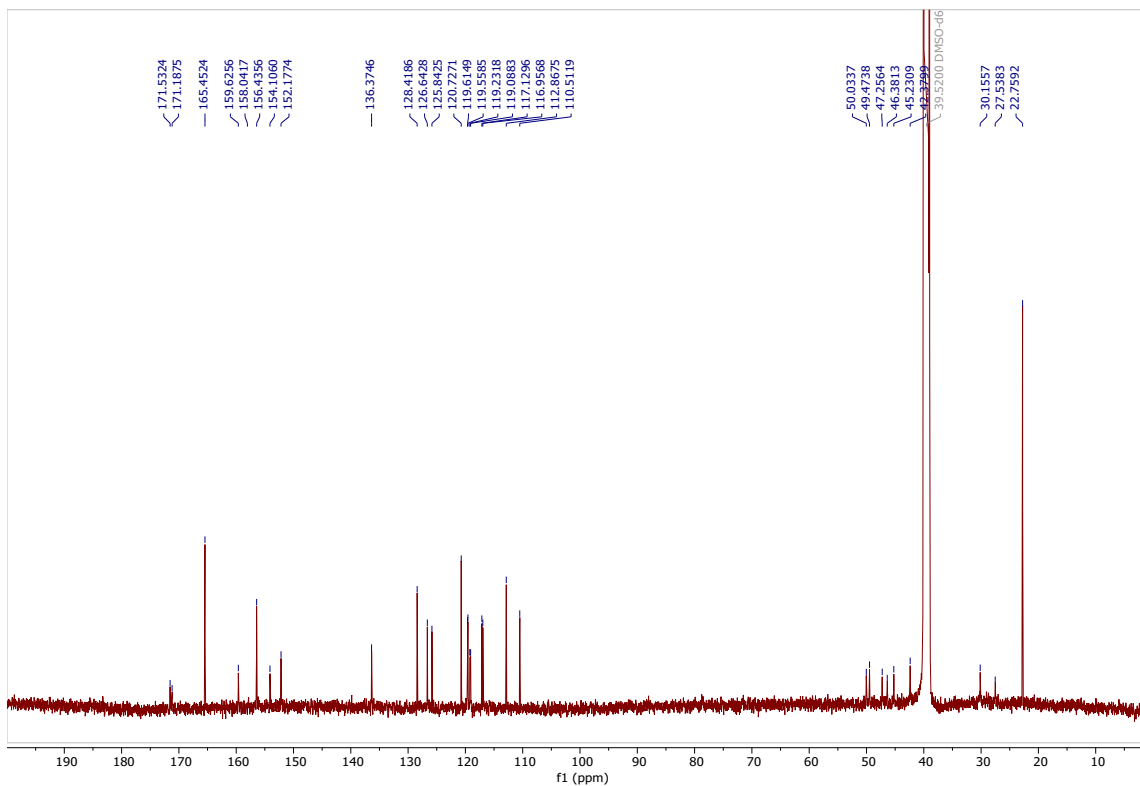
$^1\text{H}$  NMR spectrum of CU11 (Acetone- $d_6$ )



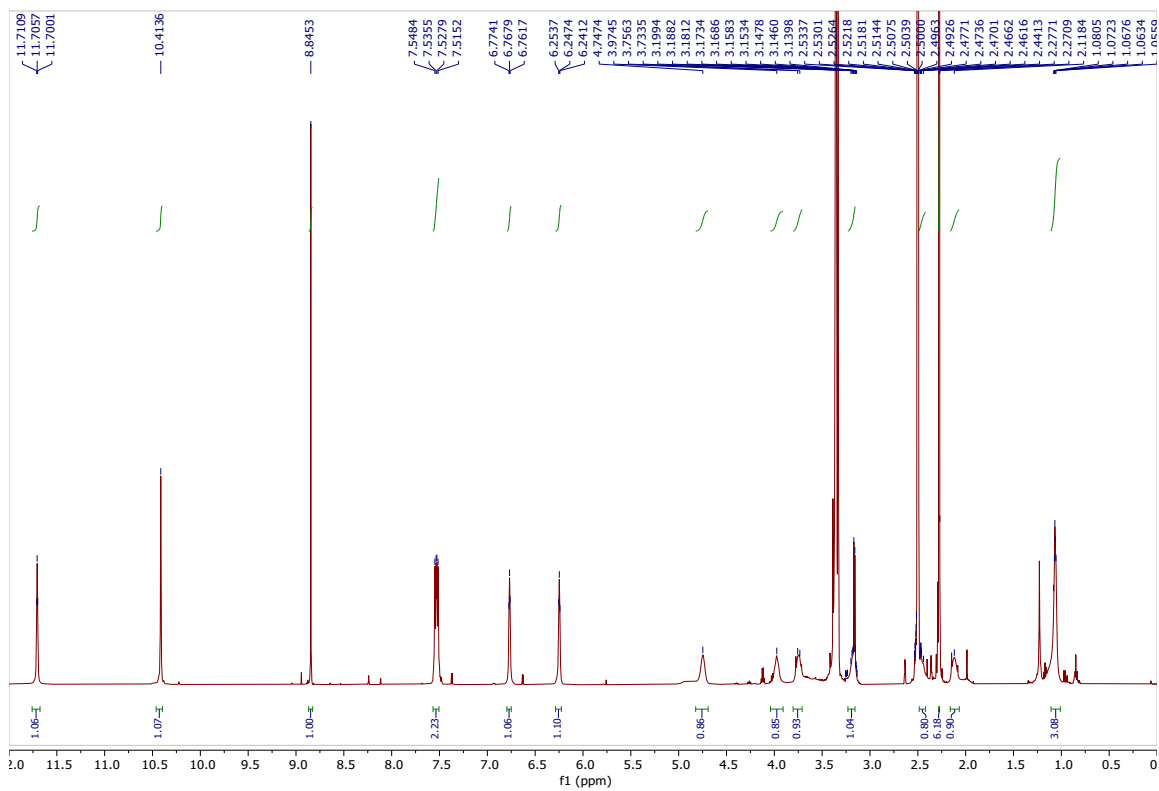
<sup>13</sup>C NMR spectrum of CU11 (Acetone-d<sub>6</sub>)



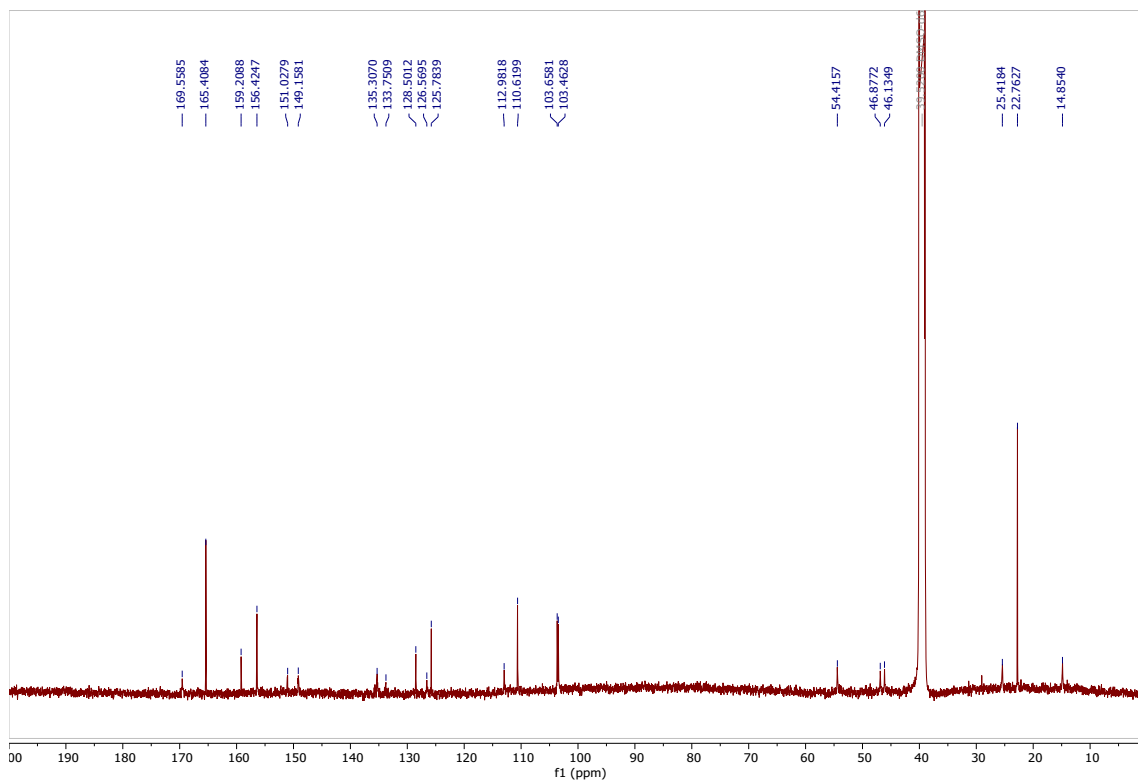
<sup>1</sup>H NMR spectrum of CU12 (Acetone-d<sub>6</sub>)



<sup>13</sup>C NMR spectrum of CU12 (Acetone-d<sub>6</sub>)

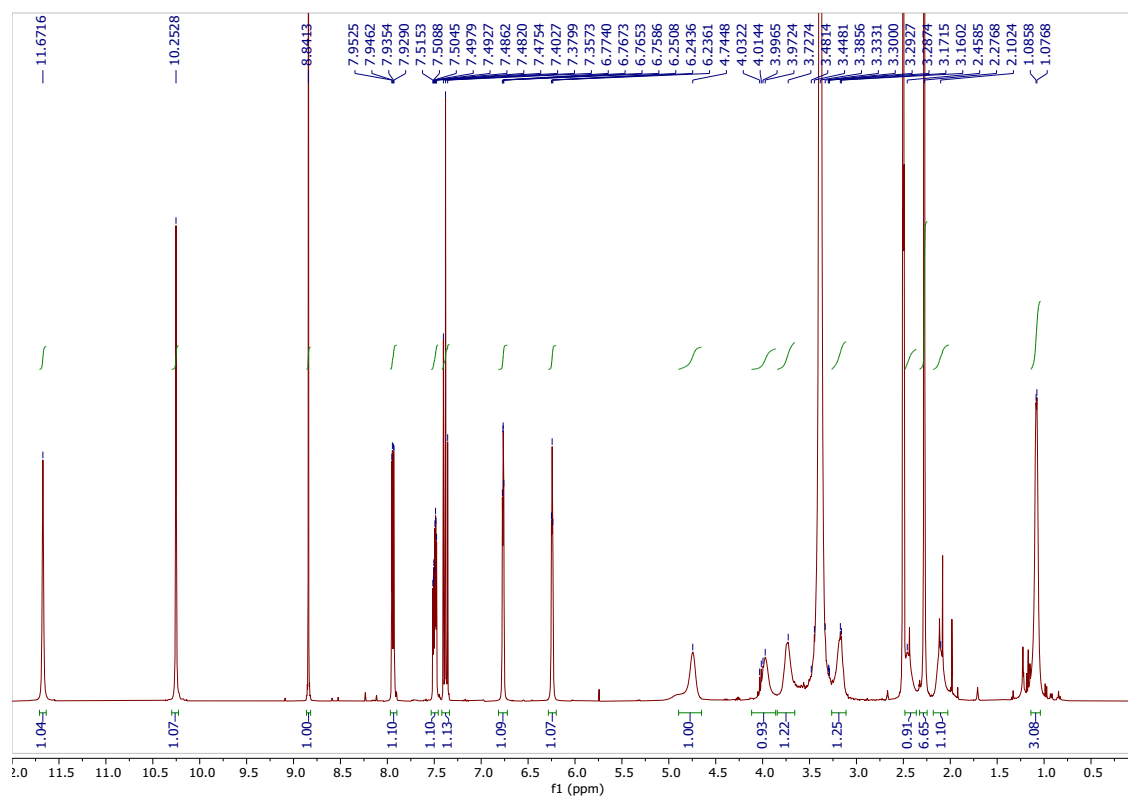


<sup>1</sup>H NMR spectrum of CU13 (Acetone-d<sub>6</sub>)

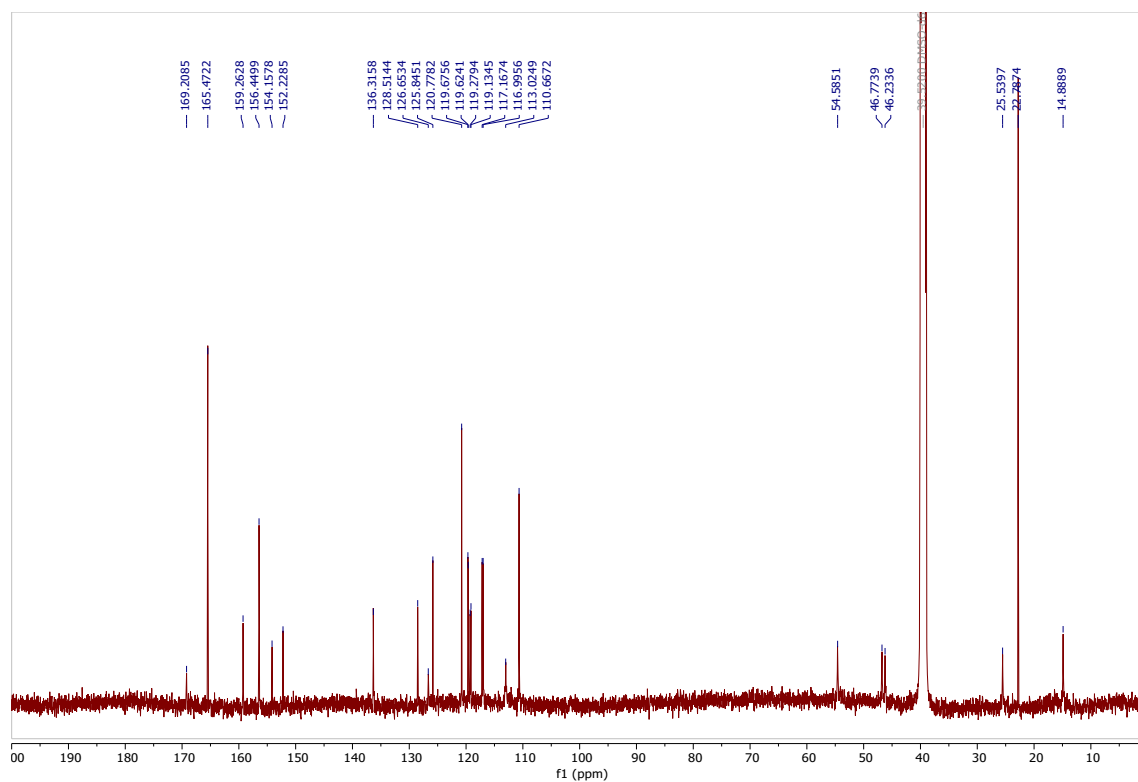




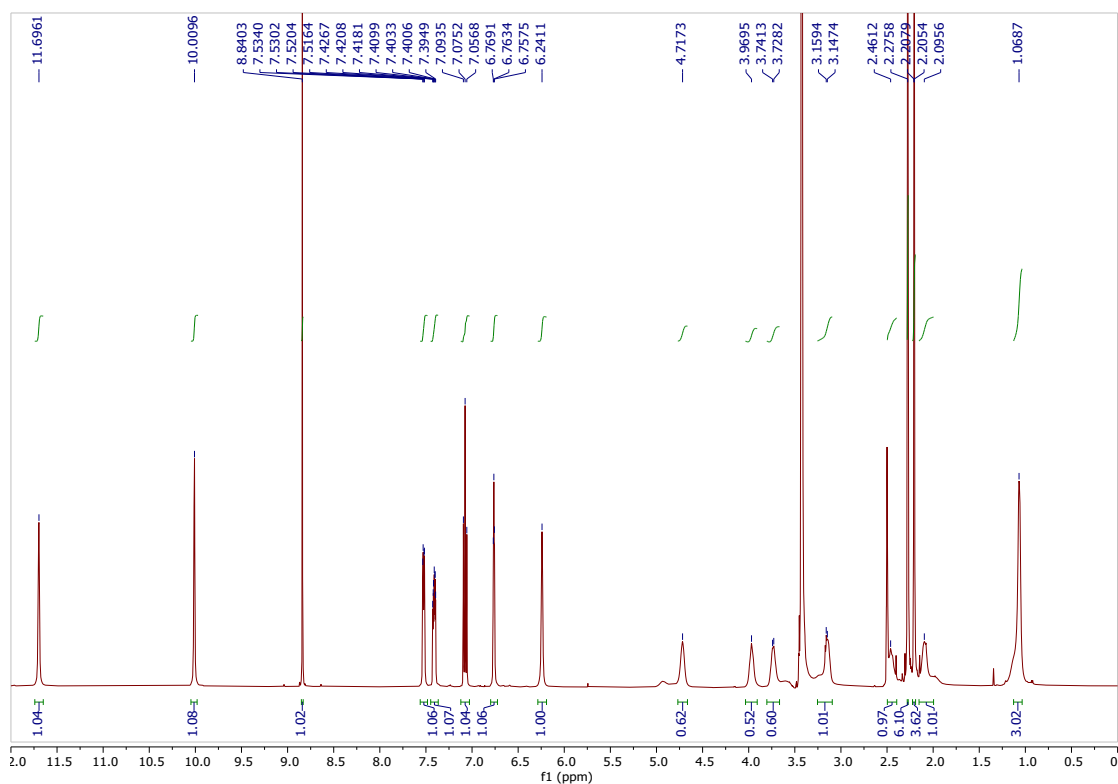
$^{13}\text{C}$  NMR spectrum of CU13 (Acetone- $d_6$ )



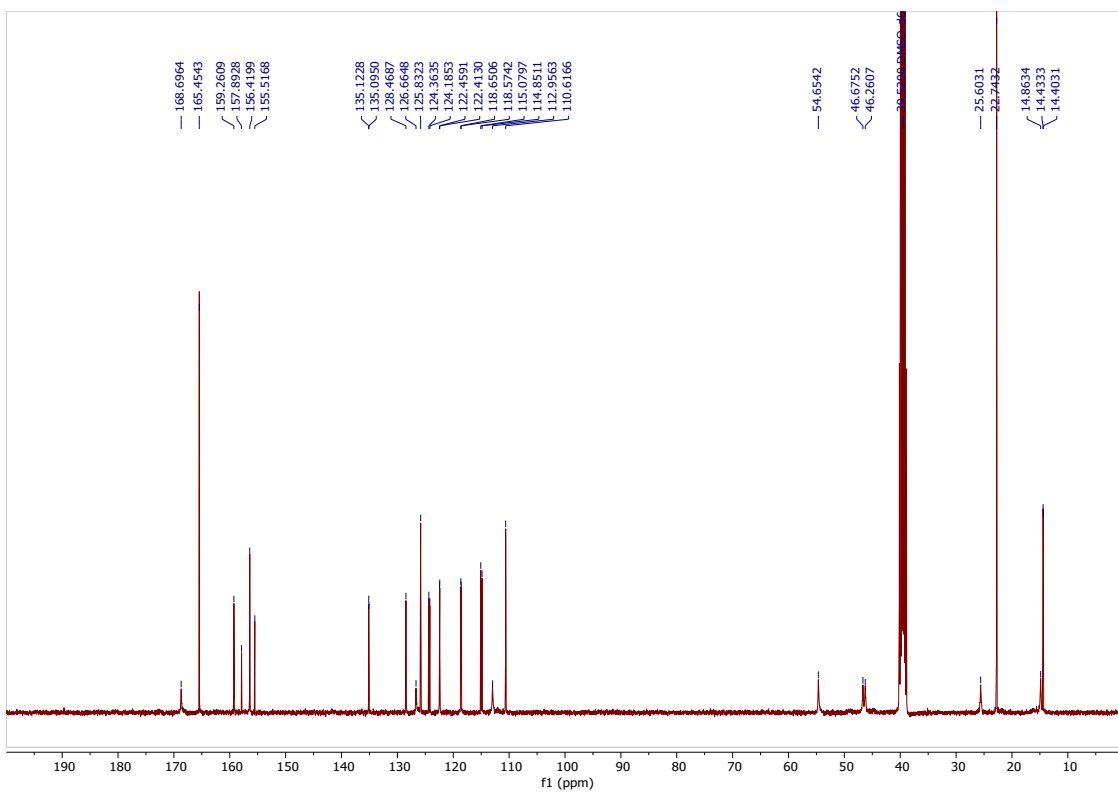
$^1\text{H}$  NMR spectrum of CU14 (DMSO- $d_6$ )



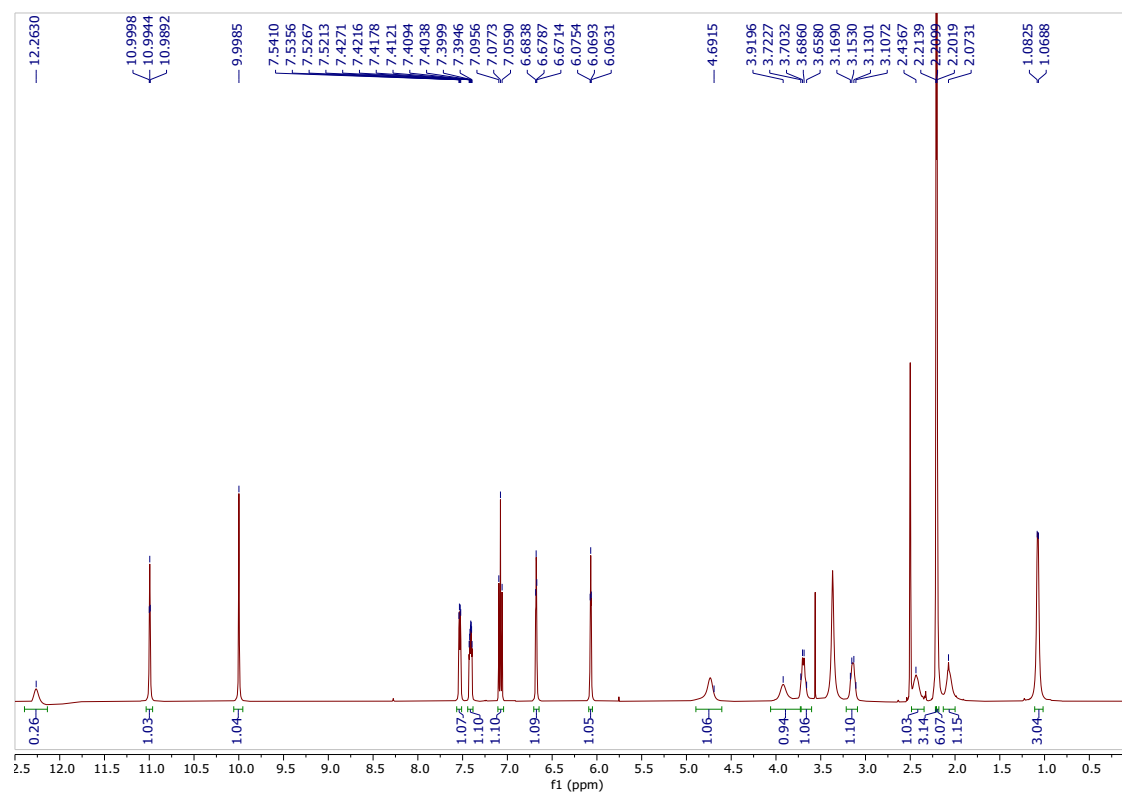
$^{13}\text{C}$  NMR spectrum of CU14 (DMSO- $d_6$ )



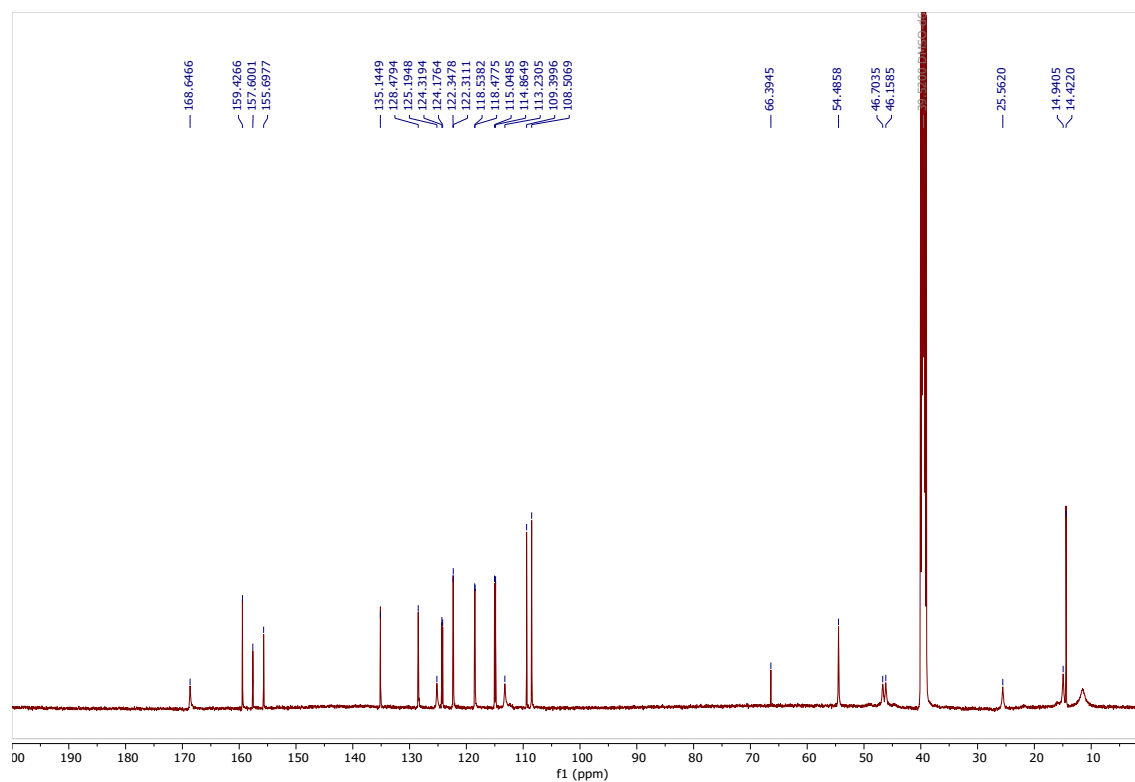
$^1\text{H}$  NMR spectrum of CU15 (DMSO- $d_6$ )



### $^{13}\text{C}$ NMR spectrum of CU15 (DMSO- $d_6$ )



### $^1\text{H}$ NMR spectrum of CU16 (DMSO- $d_6$ )



$^{13}\text{C}$  NMR spectrum of **CU16** ( $\text{DMSO-}d_6$ )