Supporting Information

Molecular Dynamics Study of Fluorosulfonyl Ionic Liquids as

Electrolyte for Electrical Double Layer Capacitors

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Figure S1. Mean Square Displacement (MSD) curves of the four fluorinated sulfonyl ILs using OPLS-VSIL force field.



Figure S2. (a) Pristine graphene electrode; (b) defect graphene electrode; (c) Localchargedistributionofdefectivegrapheneelectrode

	$D^{+}(10^{-10}m^{2}/s)$	$D^{-}(10^{-10}m^{2}/s)$
EMIMTFSI	0.332	0.211
EMIMFSI	0.552	0.394
EMIMTSAC	0.34	0.219
EMIMTFSAM	0.511	0.299

Table S1. Self diffusion parameters of four fluorinated sulfonyl ionic liquids

		OPLS	OPLS-VSIL
EMIMTFSI	EMIM	0.6	0.8
	TFSI	0.7	0.7
EMIMFSI	EMIM	0.6	0.9
	FSI	0.7	0.9
EMIMFSAC	EMIM	0.8	1
	TSAC	0.6	1
EMIMTFSAM	EMIM	0.8	0.9
	TFSAM	0.6	0.8

Table S2. Values of Beta Factors

	OPLS		OPLS-VSIL	
	$D^+(10^{-10}m^2/s)$	$D^{-}(10^{-10}m^2/s)$	$D^{+}(10^{-10}m^{2}/s)$	$D^{-}(10^{-10}m^2/s)$
EMIMTFSI	0.332	0.211	1.041	0.785
EMIMFSI	0.552	0.394	1.642	1.481
EMIMTSAC	0.34	0.219	1.6	1.3
EMIMTFSAM	0.511	0.299	1.728	1.542

 Table S3. Comparison of self diffusion coefficients simulated by OPLS and OPLS

 VSIL force field

nena				
	OPLS	OPLS-VSIL		
EMIMTFSI	6.69 S • cm ⁻¹	21.45 S • cm ⁻¹		
EMIMFSI	14 S • cm ⁻¹	46.14 S • cm ⁻¹		
EMIMTSAC	7.12 S • cm ⁻¹	36.46 S • cm ⁻¹		
EMIMTFSAM	$12.23 \text{ S} \cdot \text{cm}^{-1}$	$45.92 \text{ S} \cdot \text{cm}^{-1}$		

 Table S4. Comparison of Conductivity simulated by OPLS and OPLS-VSIL force field