

ELECTRONIC SUPPORTING INFORMATION

Tetranuclear lanthanide-based silsesquioxanes: towards a combination of a slow relaxation of the magnetization and a luminescent thermometry

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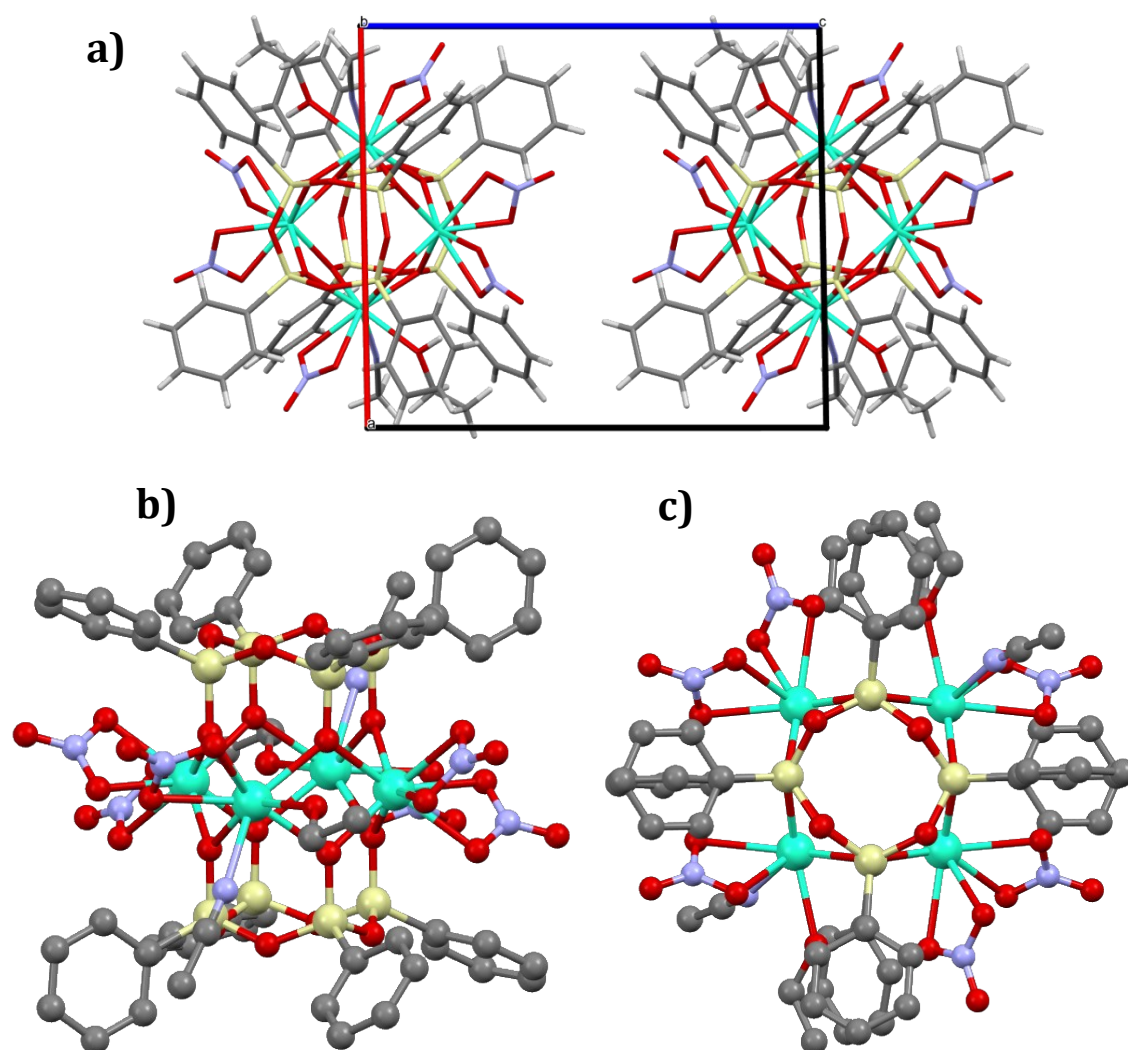


Figure S1. a) Perspective view of the crystal packing for **2** along the crystallographic axis *b*; b) and c) Molecular structure of **2** showing the prism-like polyhedron in the form of a new year paper lantern. Colour code: light green Dy/Tb; light yellow Si; red O; blue N; grey C. Hydrogen atoms and crystallized acetonitrile molecules have been omitted for clarity.

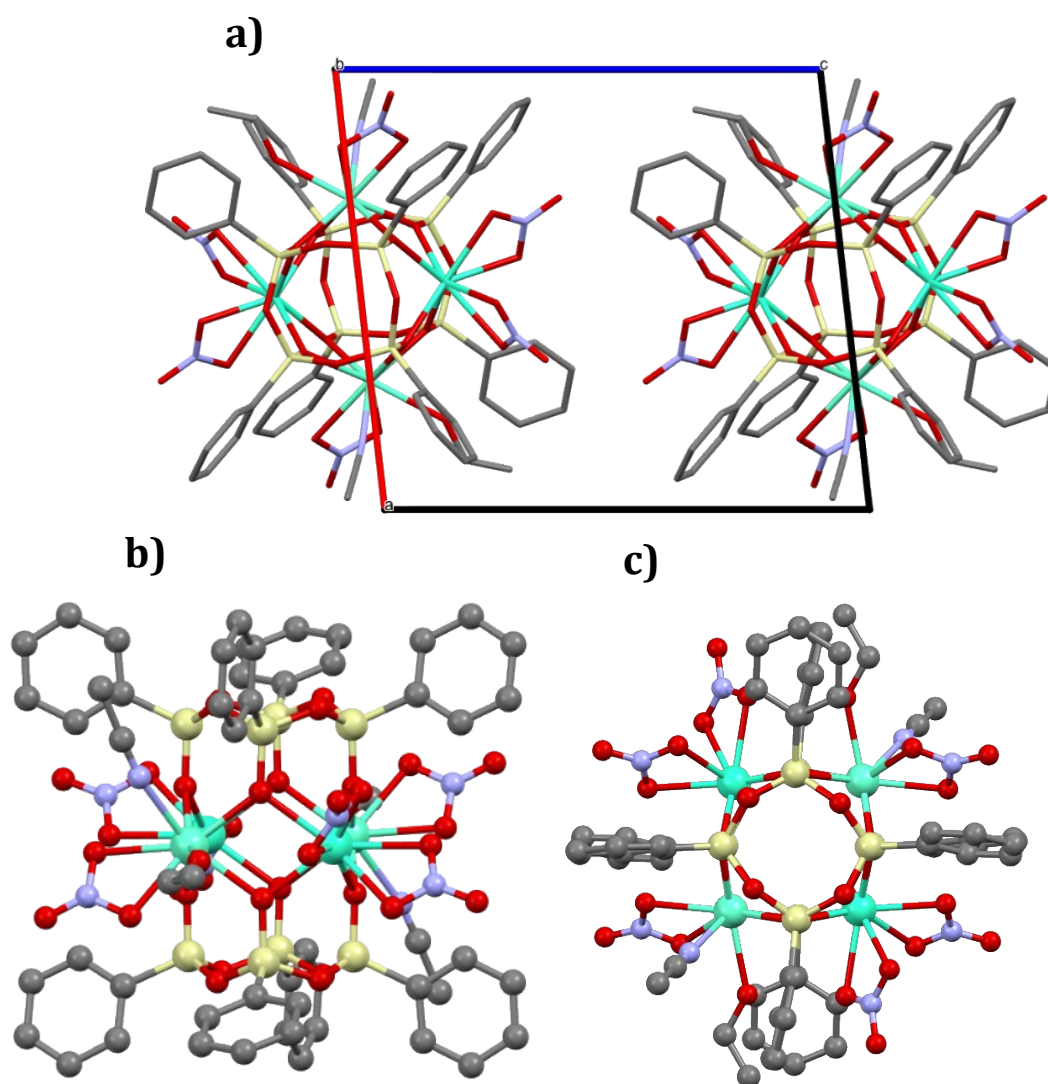


Figure S2. a) Perspective view of the crystal packing for **3** along the crystallographic axis *b*; b) and c) Molecular structure of **3** showing the prism-like polyhedron in the form of a new year paper lantern. colour code: light green Eu/Tb/Y; light yellow Si; red O; blue N; grey C. Hydrogen atoms and crystallized acetonitrile molecules have been omitted for clarity.

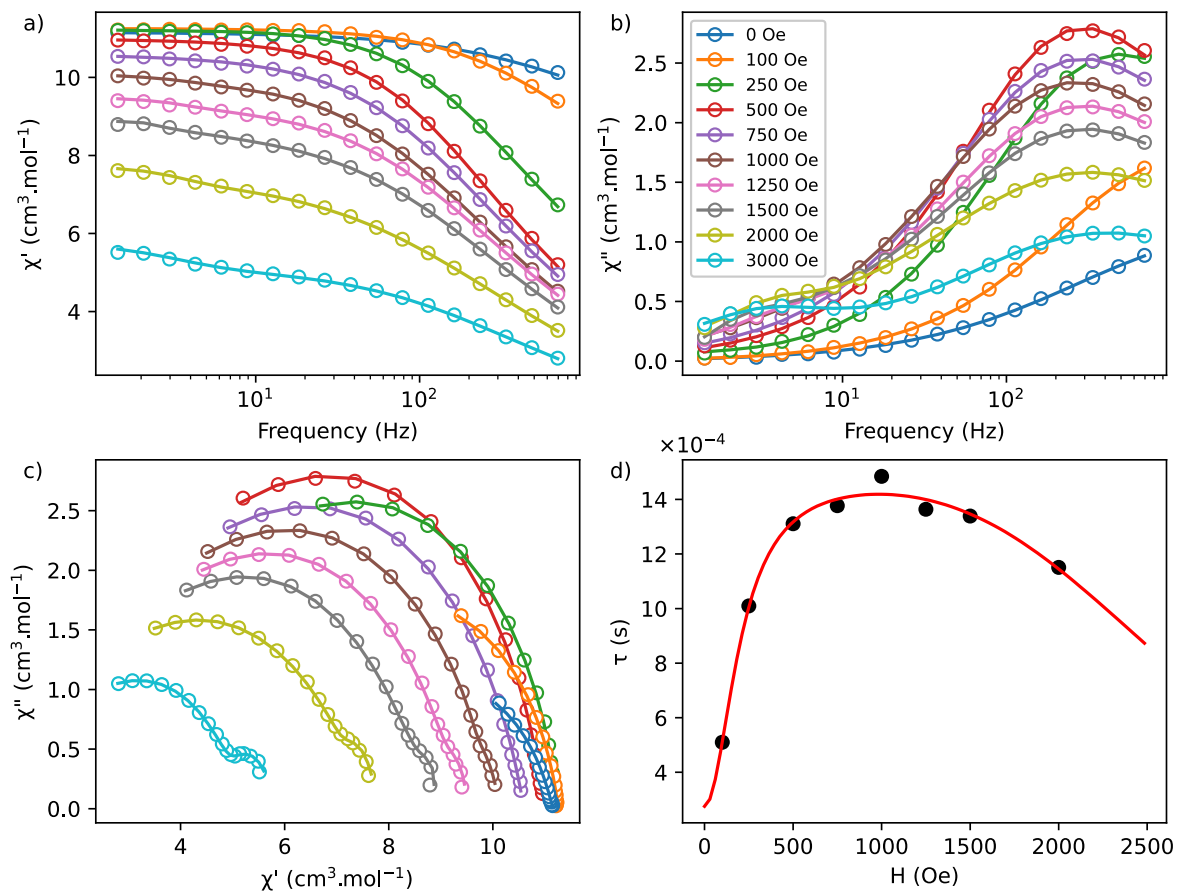


Figure S3. In-phase, χ' , (a) and out-of-phase, χ'' , (b) components of the ac magnetic susceptibility as a function of frequency at different magnetic fields measured for **1**; c) Cole-Cole (Argand) plots obtained using the field dependence of the ac susceptibility data for **1**. The solid lines correspond to the best fit obtained with a generalized Debye model; d) Field dependence of the relaxation time for **1** at 1.8 K. The red solid line represents the fits done by using the equation (2).

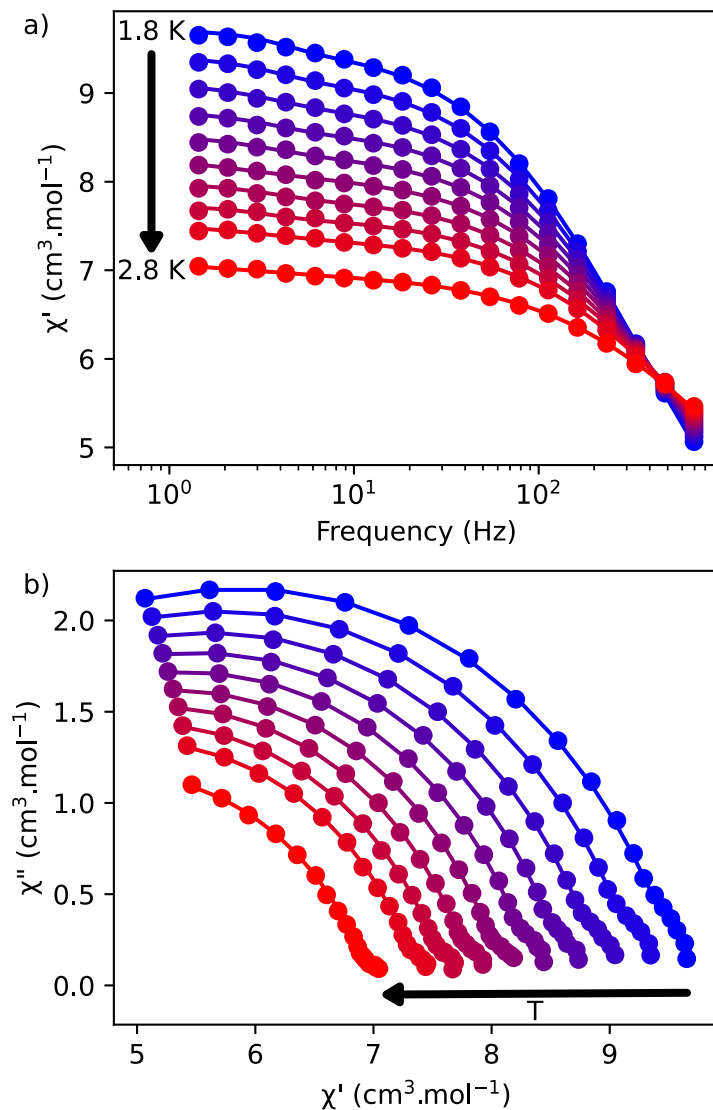


Figure S4. a) In phase, χ' , component of the ac magnetic susceptibility as a function of frequency at different temperatures measured under an optimal dc field of 1000 Oe for **1**; b) Cole-Cole (Argand) plots obtained using the ac susceptibility data for **1** under an optimal magnetic field of 1000 Oe. The solid lines correspond to the best fit obtained with a generalized Debye model.

Table S1: Crystal data, data collection and structure refinement details for **1-3**.

Identification code	1 • 4MeCN	2 • 4MeCN	3 • 4MeCN _{sq}
Empirical formula	C ₈₀ H ₁₁₀ DyEu ₃ N ₁₄ O ₃₆ Si ₈	C ₈₀ H ₁₁₀ Dy ₃ N ₁₄ O ₃₆ Si ₈ Tb	C ₈₀ H ₁₁₀ EuN ₁₄ O ₃₆ Si ₈ TbY ₂
Formula weight	2686.95	2714.97	2557.23
Crystal size, mm	0.15×0.13×0.12	0.18×0.15×0.15	0.09×0.09×0.06
Crystal system	Triclinic	Triclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a</i> , Å	13.27830(13)	13.2584(2)	12.8840(2)
<i>b</i> , Å	13.34642(15)	13.3275(2)	13.9471(2)
<i>c</i> , Å	16.09392(18)	16.0668(2)	15.3005(2)
α , deg.	68.0627(10)	68.1098(15)	67.2399(15)
β , deg.	85.3464(9)	85.3999(14)	82.6403(13)
γ , deg.	80.1123(9)	80.1109(15)	86.2713(13)
<i>V</i> , Å ³	2605.87(5)	2594.78(7)	2514.08(7)
<i>Z</i>	1	1	1
Density (calc.), g/cm ³	1.712	1.737	1.689
Absorption coefficient, mm ⁻¹	2.668	2.987	2.638
<i>F</i> (000)	1343	1351	1294
Theta range, deg.	2.051-35.300	2.055-35.427	2.190-35.386
Index ranges	-21 ≤ <i>h</i> ≤ 20, -21 ≤ <i>k</i> ≤ 21, -26 ≤ <i>l</i> ≤ 25	-19 ≤ <i>h</i> ≤ 21, -21 ≤ <i>k</i> ≤ 21, -25 ≤ <i>l</i> ≤ 26	-19 ≤ <i>h</i> ≤ 20, -22 ≤ <i>k</i> ≤ 21, -24 ≤ <i>l</i> ≤ 23
Reflections collected	99913	97988	66219
Independent reflections, <i>R</i> _{int}	21545, 0.0477	21453, 0.0611	20143, 0.1160
Reflections with <i>I</i> > 2σ(<i>I</i>)	17201	17586	12723
Parameters refined	713	758	544
Goodness-of-fit on <i>F</i> ²	1.040	1.031	1.030
<i>R</i> ₁ / w <i>R</i> ₂ for <i>I</i> > 2σ(<i>I</i>)	0.0282 / 0.0615	0.0354 / 0.0821	0.0883 / 0.1886
<i>R</i> ₁ / w <i>R</i> ₂ for all data	0.0439 / 0.0656	0.0491 / 0.0891	0.1274 / 0.2038
<i>T</i> _{max} , <i>T</i> _{min}	0.696, 0.633	0.632, 0.523	0.844, 0.752
Δρ _{max} / Δρ _{min} , e·Å ⁻³	1.580 / -1.583	2.749 / -2.495	2.758 / -2.895

Table S2: Main structural parameters for **1-3**.

Sample	Ln-O distance, Å	Ln-N distance, Å	Ln-Ln distance, Å	∠ O-Ln-O, °
1	Ln-O(Ln): 2.310(9)-2.356(9); Ln-NO ₃ : 2.436(4)-2.563(3); Ln-EtOH: 2.460(3), 2.468(8)	2.551(4), 2.554(10)	3.726(3)-3.777(3)	(Ln)O-Ln-O(Ln): 70.99(9)-81.46(12); NO ₃ : 50.66(15)-51.52(8)
2	Ln-O(Ln): 2.243(8)-2.376(8); Ln-NO ₃ : 2.380(10)-2.567(8); Ln-EtOH: 2.394(8), 2.456(3)	2.524(4), 2.527(12)	3.648(2)-3.742(2)	(Ln)O-Ln-O(Ln): 70.98(10)-83.4(2) NO ₃ : 50.26(15)-52.13(8)
3	Ln-O(Ln): 2.289(4)-2.356(4); Ln-NO ₃ : 2.453(5)-2.534(6); Ln-EtOH: 2.442(5), 2.477(6)	2.486(6), 2.520(7)	3.646(5)-3.718(4)	(Ln)O-Ln-O(Ln): 72.04(14)-83.11(16) NO ₃ : 51.0(2)-52.70(18)

Table S3: Main parameters for **1-3** regarding their magnetic properties.

Sample	Experimental χT values, $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	Calculated χT values, $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	Calculated χT values for Ln^{3+}	Magnetization values at 7 kOe, Nb
1	14.2 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	$\text{Eu}_{3.08}\text{Dy}_{0.92}$ 13.03 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	Dy^{3+} ($J=15/2$, $g = 4/3$): 14.17 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$ Eu^{3+} is diamagnetic	8.8
2	53.9 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	$\text{Dy}_{3.18}\text{Tb}_{0.82}$ 54.74 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	Dy^{3+} ($J=15/2$, $g = 4/3$): 14.17 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$ Tb^{3+} (7F_6 , $S = 3$, $L = 3$, $g = 3/2$): $\chi T = 11.82$ $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	29.1
3	11.9 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	$\text{Y}_{2.11}\text{Eu}_{0.87}\text{Tb}_{1.01}$ 11.93 $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$	Tb^{3+} (7F_6 , $S = 3$, $L = 3$, $g = 3/2$): $\chi T = 11.82$ $\text{cm}^3 \cdot \text{K} \cdot \text{mol}^{-1}$ Eu^{3+} is diamagnetic	6.8