## Supplementary file for

## Catalytic depolymerization of Kraft lignin to high yield alkylated-phenols over

## CoMo/SBA-15 catalyst in supercritical ethanol

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**Fig. S1** (a) FT-IR spectra and (b) thermogravimetric (TG) analyses of commercially available KL (Sigma-Aldrich, USA) with the product number 370959 and 471003.



Fig. S2 Core level spectra of (a, b) Si 2p and O 1s in SBA-15.



Fig. S3 NH<sub>3</sub>-TPD spectra of SBA-15, Mo/SBA-15 and CoMo/SBA-15 catalysts.



Fig. S4 FTIR spectra of solid residues obtained at different times (a) and temperatures (b).



Fig. S5 (a) Mass spectra from the peak13.20 and the standard guaiacol. (b) Mass spectra from the peak 16.25 and the standard 4-methylguaiacol.



**Fig. S6** (a) Mass spectra from the peak 18.67 and the standard 4-ethylguaiacol. (b) Mass spectra from the peak 21.05 and the standard 4-propylguaiacol.



Fig. S7 (a) Mass spectra from the peak 23.14 and the standard isoeugenol. (b) Mass spectra from the peak 27.81 and the standard homovanillic acid.



**Fig. S8** Assignments of peaks/peak picking in the aliphatic hydrogenated regions of the Kraft lignin and light oil fraction obtained at 290 °C, 4h and CoMo/SBA-15 catalyst.



**Fig. S9** Assignments of peaks/peak picking in the oxygenated and side-chain regions of the Kraft lignin and light oil obtained at 290 °C, 4h and CoMo/SBA-15 catalyst.



**Fig. S10** Assignments of peaks/peak picking in the aromatic regions of the Kraft lignin and light oil obtained at 290 °C, 4h and CoMo/SBA-15 catalyst.

Catalyst	Metal content (wt%)			
Cataryst	Мо	Со		
SBA-15	-	-		
Mo/SBA-15	9.8	-		
Co/SBA-15	-	9.7		
CoMo/SBA-15	4.9	4.8		

 Table S1. The ICP–OES results of the prepared catalyst.

Parameters	Conditions	Heavy-oil (wt%)	Light-oil (wt%)	Solid residue (wt%)	Gas (wt%)
	0.5	$19.82 \pm 0.25$	$10.26\pm0.10$	$63.12 \pm 1.03$	$6.33 \pm 0.22$
	2	$22.75\pm0.30$	$21.00\pm0.27$	$48.50\pm0.88$	$7.65\pm0.16$
Time (h)	3	$18.32\pm0.19$	$27.00\pm0.35$	$44.81\pm0.73$	$10.02\pm0.60$
	4	$17.88\pm0.22$	$29.17\pm0.50$	$41.50 \pm 0.59$	$11.80\pm0.17$
	6	$19.25\pm0.18$	$25.39\pm0.44$	$45.06\pm0.37$	$10.61\pm0.29$
	240	$17.60\pm0.75$	$15.00\pm0.26$	$61.02 \pm 1.53$	$6.70\pm0.32$
	260	$14.75\pm0.24$	$21.00\pm0.33$	$56.50 \pm 1.15$	$8.11\pm0.18$
Temperature (°C)	270	$13.17\pm0.70$	$25.00\pm0.28$	$52.81\pm0.88$	$9.40\pm0.56$
	290	$17.88\pm0.13$	$29.17\pm0.70$	$41.50\pm0.57$	$11.40\pm0.66$
	320	$19.70\pm0.23$	$26.00\pm0.64$	$44.06\pm0.68$	$10.01\pm0.49$
	Control	$17.88 \pm 0.51$	$29.17\pm0.68$	$41.50\pm0.67$	$11.00 \pm 0.28$
Catalyst	SBA-15	$15.60\pm0.33$	$32.60\pm0.56$	$38.50\pm0.79$	$13.00\pm0.14$
	Co/SA-15	$13.70\pm0.43$	$38.90\pm0.75$	$33.50\pm0.38$	$14.20\pm0.55$
	Mo/SBA-15	$12.96\pm0.11$	$43.50\pm0.61$	$31.70\pm0.77$	$14.01\pm0.39$
	CoMo/SBA-15	$8.42\pm0.17$	$48.20\pm0.82$	$28.50\pm0.26$	$14.60\pm0.24$

 Table S2. Results of KL depolymerization under different reaction conditions.

Wavenumber (cm <sup>-1</sup> )	Characteristics
3093–3619	O–H group.
2928	C–H (aliphatic and aromatic).
2831	C–H (methoxy group).
1712/1733	C=O in unconjugated ketones, carbonyl and ester groups).
1602	aromatic ring (C=C) vibration.
1512	aromatic ring (C=C) vibration.
1460	-OCH <sub>3</sub>
1270, 1150, 1118	G ring and C–O vibrations.
1035	C–O vibration in ether, acid or ester groups.
860-741	para-substituted aromatic ring.

 Table S3. FT-IR spectra of KL and bio-oils obtained under different reaction conditions.

label	$\delta_C / \delta_H$	assignment		
CH <sub>3</sub>	0-1/10-16	$CH_3 \gamma$ to $Ar$		
CH <sub>3</sub> , CH <sub>2</sub>	1-1.4/16-23	CH <sub>3</sub> , CH <sub>2</sub> $\beta$ to Ar, CH $\gamma$ to Ar		
CH <sub>2</sub> , CH	1.4-1.9/23-30	$CH_2$ , $CH \beta$ to $Ar$		
CH <sub>3</sub> , CH <sub>2</sub> , CH	1.9-3.5/30-52.5	$CH_3$ , $CH_2$ , $CH \alpha$ to $Ar$		
$C_{\beta}$	53.78/3.43	$C_{\beta}$ -H <sub><math>\beta</math></sub> in phenylcoumaran substructures (C)		
$\mathbf{B}_{\boldsymbol{eta}}$	54.07/3.04	$C_{\beta}$ -H <sub><math>\beta</math></sub> in resinol substructures (B)		
–OCH <sub>3</sub>	55.98/3.75	C– H in methoxyl		
$A_{\gamma}$	60.55/3.40 and 60.40/3.59	$C_{\gamma}$ -H <sub><math>\gamma</math></sub> in $\beta$ -O-4' substructures (A)		
$\mathbf{D}_{\gamma}$	62.02/4.09	$C_{\gamma}$ – $H_{\gamma}$ in coniferyl alcohol substructures (D)		
$C_{\gamma}$	63.34/3.72 and 64.23/3.30	$C_{\gamma}$ – $H_{\gamma}$ in phenylcoumaran substructures (C)		
$A_{\alpha}$	71.58/4.73	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in $\beta$ -O-4' substructures (A)		
$\mathbf{B}_{\gamma}$	71.44/3.75, 73.64/3.83 and 75.85/3.53	$C_{\gamma}$ -H <sub><math>\gamma</math></sub> in resinol substructures (B)		
$A_{\beta}$	84.68/4.26	$C_{\beta}$ -H <sub><math>\beta</math></sub> in $\beta$ -O-4' substructures (A)		
$B_{\alpha}$	85.56/4.60	$C_{\alpha}$ -H <sub><math>\alpha</math></sub> in resinol substructures (B)		
$G_2'$	109.99/7.14	$C_2$ - $H_2$ in oxidized guaiacyl units (G')		
$G_2$	113.38-109.99/6.63-6.90	C <sub>2</sub> -H <sub>2</sub> in guaiacyl units (G)		
G <sub>5</sub>	115.88/6.75	C <sub>2</sub> -H <sub>2</sub> in guaiacyl units (G)		
G <sub>6</sub>	119.41-121.47/6.52-6.76	C <sub>6</sub> -H <sub>6</sub> in guaiacyl units (G)		
$G_6'$	120.00/6.95	C <sub>6</sub> -H <sub>6</sub> in oxidized guaiacyl units (G')		
$G_4$	126.13/6.97	C <sub>2</sub> -H <sub>2</sub> in guaiacyl units (G)		
H <sub>2,6</sub>	129.13/7.14	$C_{2,6}$ -H <sub>2,6</sub> in <i>p</i> -hydroxyphenyl units (H)		

**Table S4.** Assignment of  ${}^{13}C-{}^{1}H$  cross-signal in the HSQC spectra of Kraft lignin.

Entr	Somnlo	Elemental compositions (%)					HHV
У	Sample	С	Η	0	Ν	S	(MJ/kg) <sup>d</sup>
1	KL	61.69	6.09	30.09	0.65	1.48	25.92
2	Control (heavy oil) <sup>a</sup>	65.47	6.86	26.23	0.39	1.05	28.44
3	Mo/SBA-15 (heavy oil) <sup>b</sup>	67.61	6.92	24.81	0.32	0.34	29.24
4	CoMo/SBA-15 (heavy oil) <sup>c</sup>	67.74	7.29	24.44	0.41	0.12	29.75
5	CoMo/SBA-15 (light oil) <sup>c</sup>	71.23	8.31	20.16	0.23	0.07	32.60

 Table S5. Elemental analyses of Kraft lignin and bio-oil fractions.

<sup>*a*</sup> Reaction conditions = 3.0g lignin, 90 mL ethanol, 290 °C and 4h.

<sup>b</sup> Reaction conditions = 3.0g lignin, 90 mL ethanol, 290 °C, 4h and 0.3 g Mo/SBA-15 catalyst (10 wt%).

<sup>c</sup> Reaction conditions = 3.0g lignin, 90 mL ethanol, 290 °C, 4h and 0.3g CoMo/SBA-15 catalyst (10 wt%).

<sup>*d</sup></sup><i>HHV*(*MJ/kg*) =  $[(34 \times C) + (124.3 \times H) + (6.3 \times N) + (19.3 \times S) - (9.8 \times O)]/100$ </sup>