

Supporting Information for

Design and synthesis of Unnatural Coordination Glycopolymer Particles (CGPs): unleashing the potential of catechol-saccharide derivatives.

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S1. Secondary products obtained after methanolysis

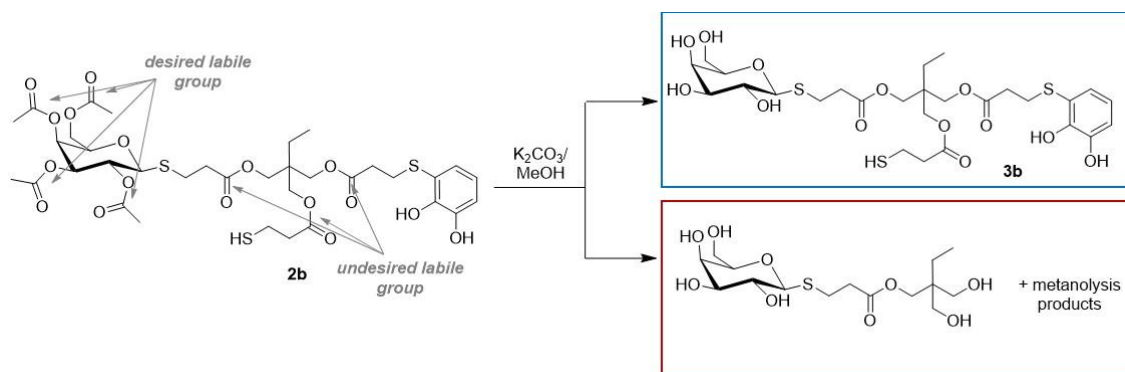


Figure S1. Secondary products obtained after methanolysis under $K_2CO_3/MeOH$ conditions.

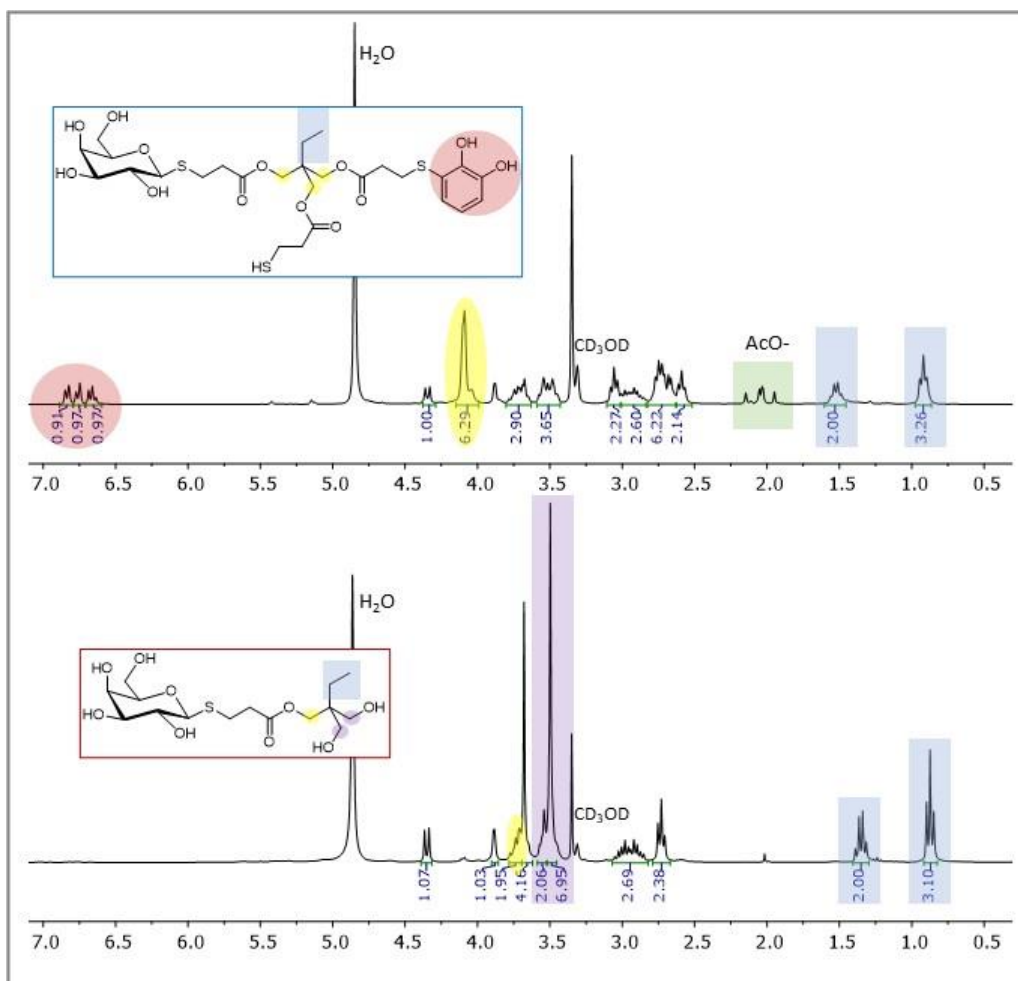


Figure S2. Comparison of 1H NMR of **3b** and secondary products obtained after methanolysis.

S2. UV-Vis of compound 4a

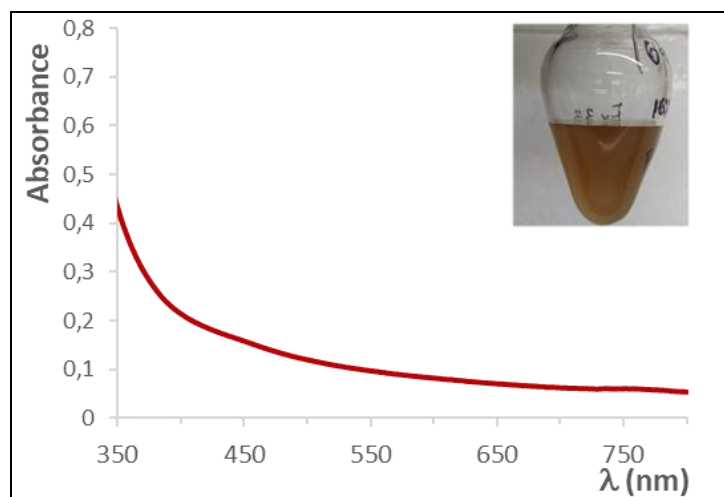


Figure S3. UV-Vis spectra of compound 4a were obtained in the same medium used for the preparation of CGPs, but in the absence of Fe(III), after 24 hours.

S3. IR spectra

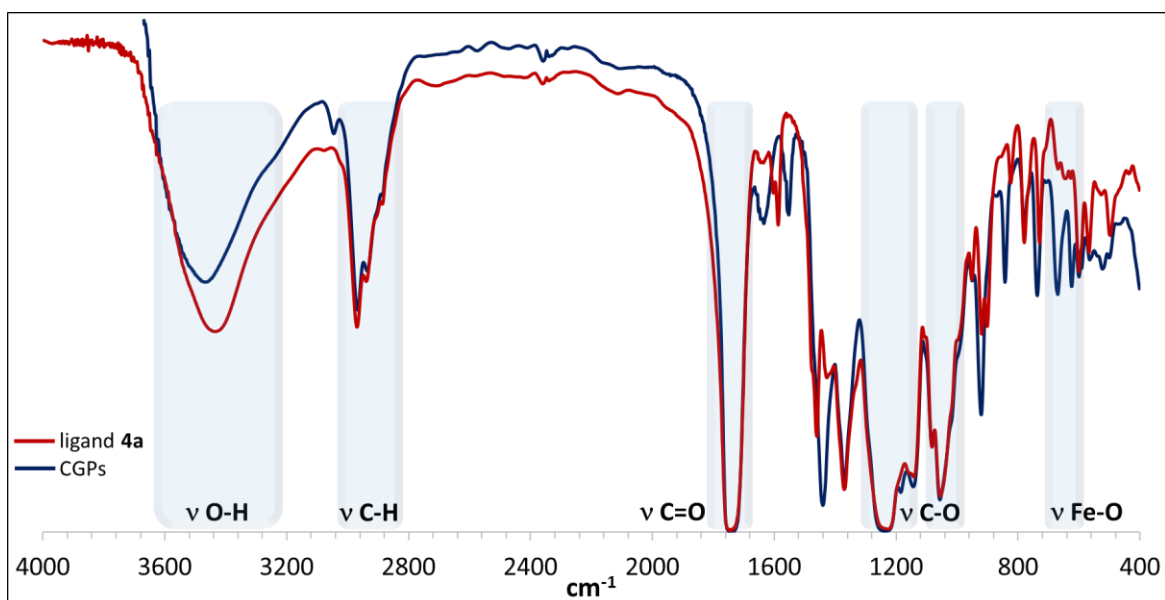


Figure S4. Full IR spectra (KBr) of ligand 4a and CGPs.

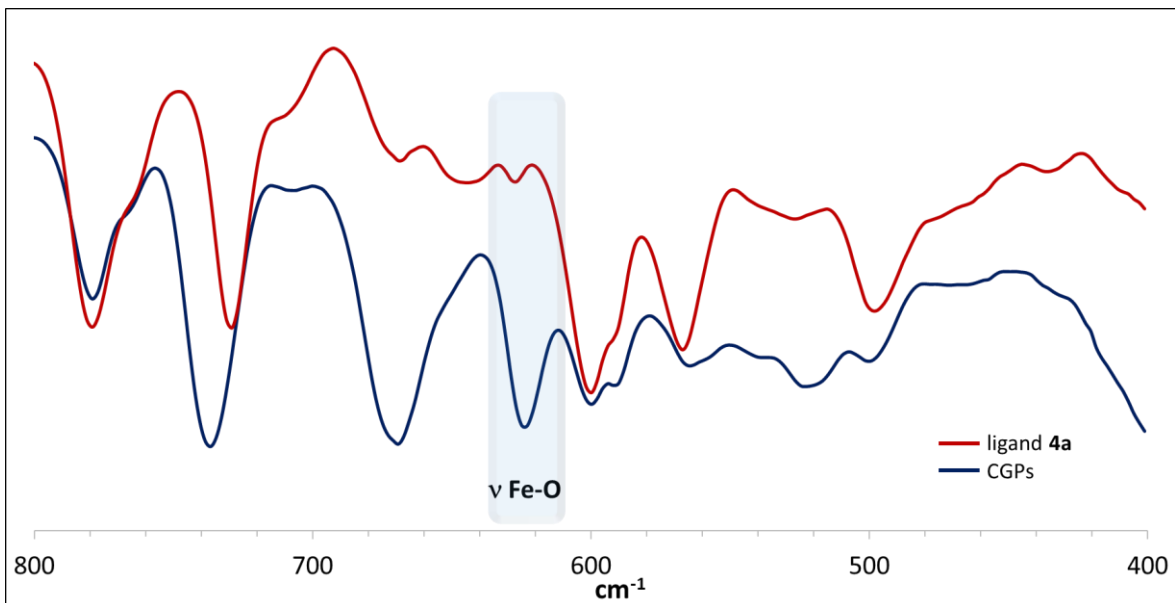


Figure S5. IR spectra (KBr) of ligand **4a** and CGPs in the 400-800 cm^{-1} region.

S4. SEM and EDX of CGPs

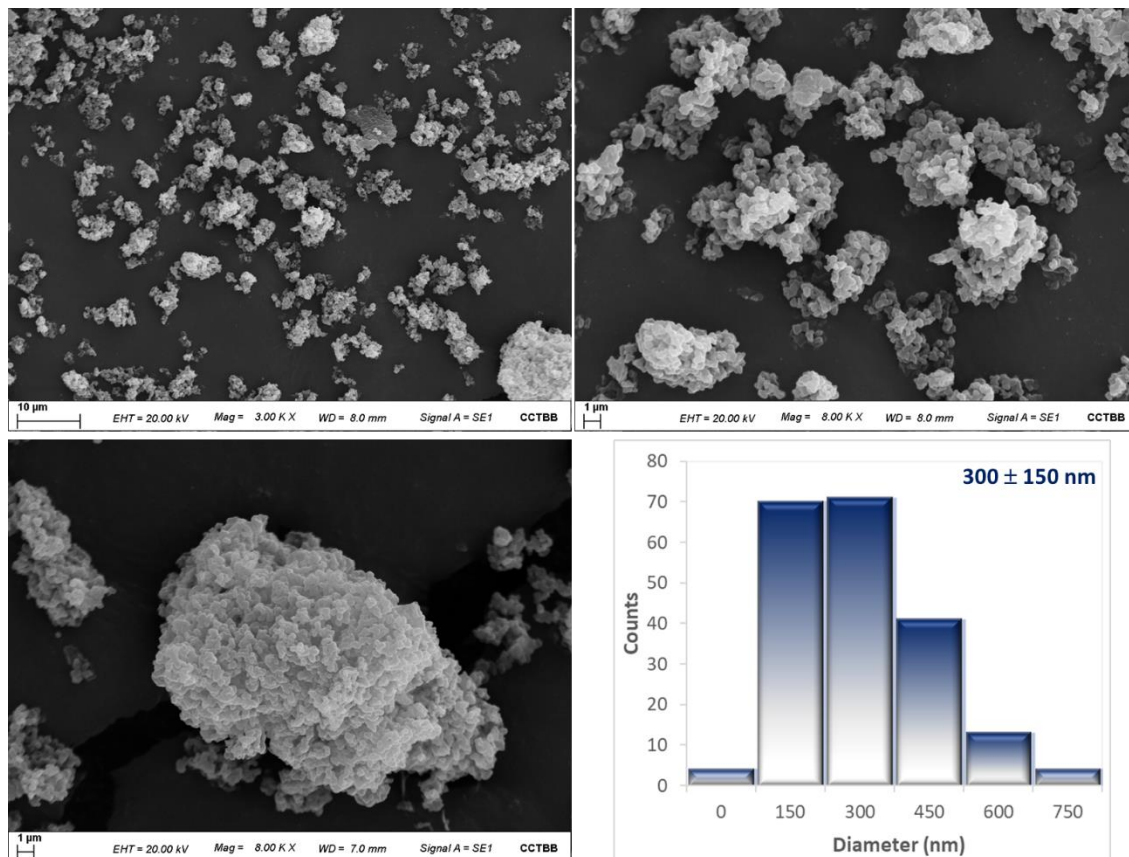


Figure S6. SEM images of CGPs and size distribution histogram.

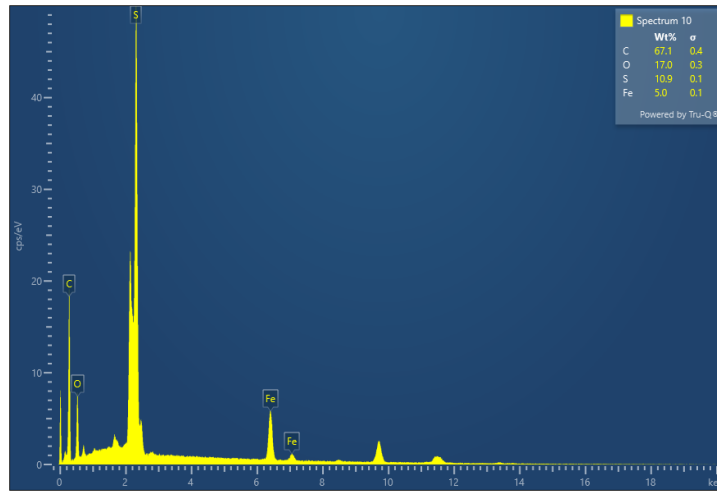


Figure S7. EDX spectrum of CGPs.

S5. DLS and Zeta Potential of CGPs samples in different conditions



Figure S8. CGPs solutions in EtOH, AcOEt, DMSO, H₂O and PBS buffer (pH ~7,4).

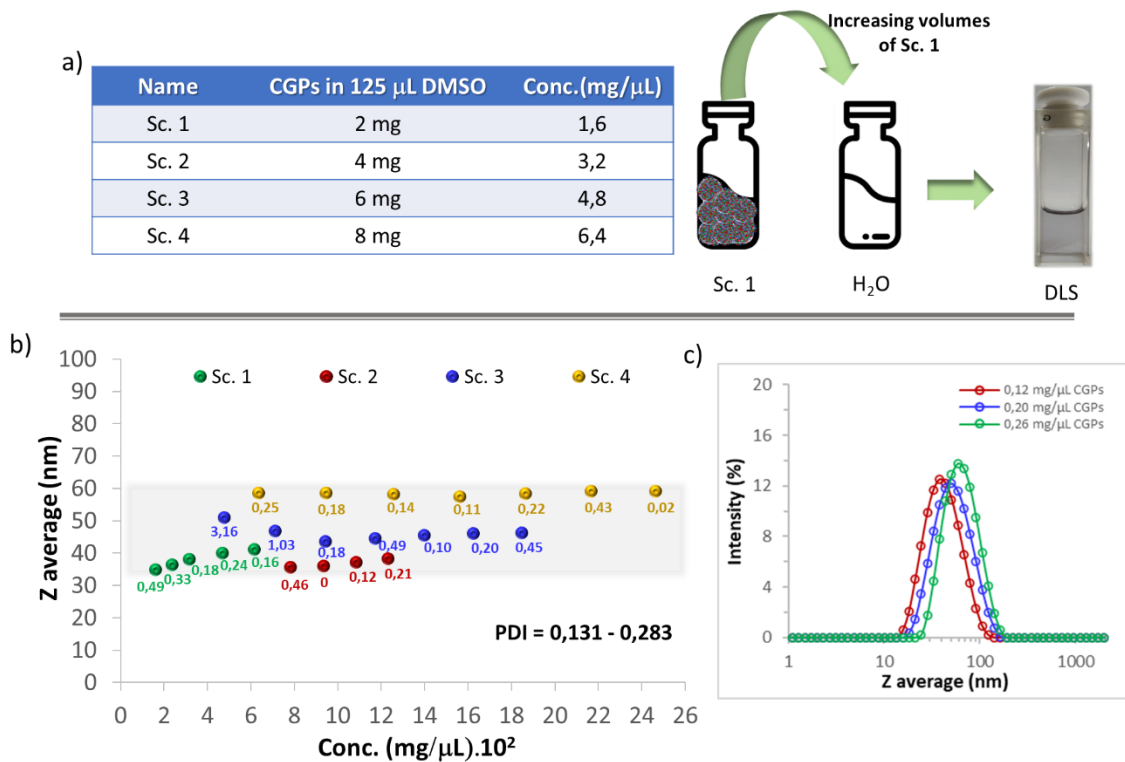


Figure S9. Figure S9. DLS measurements (represented as Z-average mean size) of different concentrations of CGPs/DMSO solutions in H₂O. Standard deviation values are indicated below each data point.

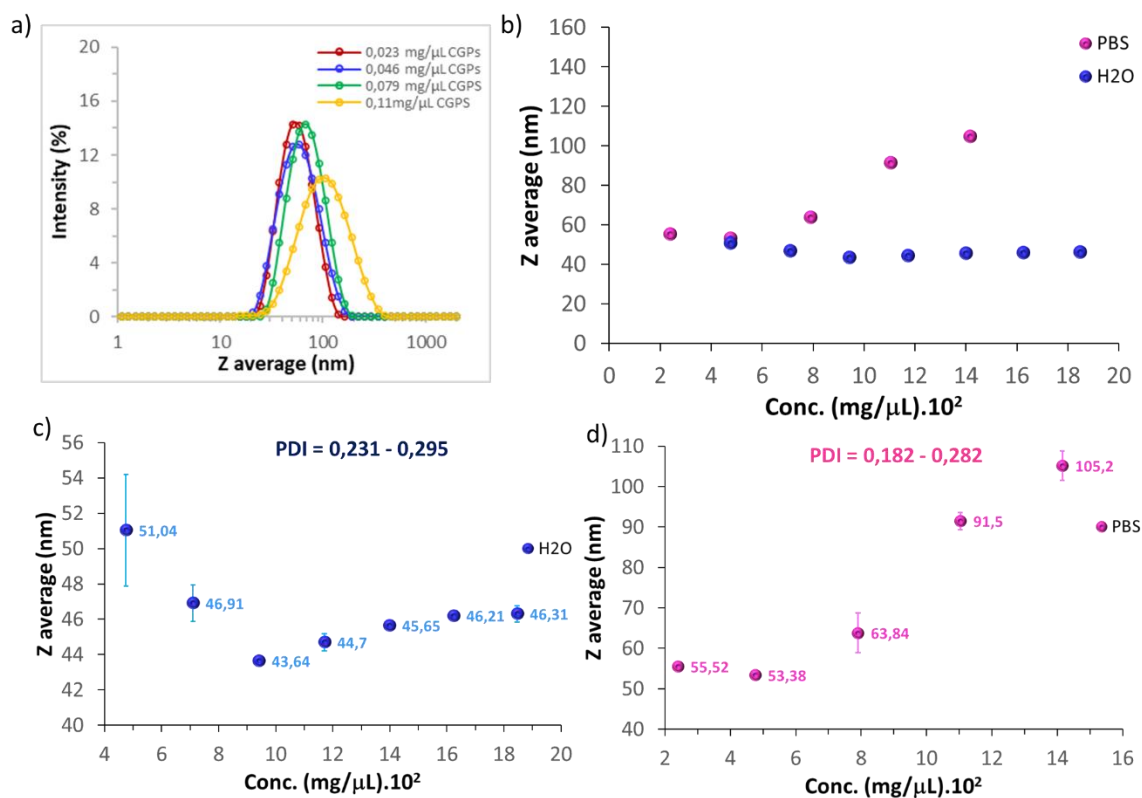


Figure S10. a) DLS measurements (represented as Z-average mean size) of different concentrations of CGPs/DMSO solutions in PBS buffer (pH ~ 7.4); b) A comparison depicting the Z-average mean size of distinct concentrations of CGPs/DMSO in both H₂O and PBS buffer solutions; c) The same graph as in b), illustrating an upward extension on the y-axis from 40 to 56 nm, highlighting the standard deviation values for each measurement and the PDI range; d) The identical graph as in b), displaying an upward extension on the y-axis from 40 to 110 nm, accentuating the standard deviation values for each measurement and the PDI range.

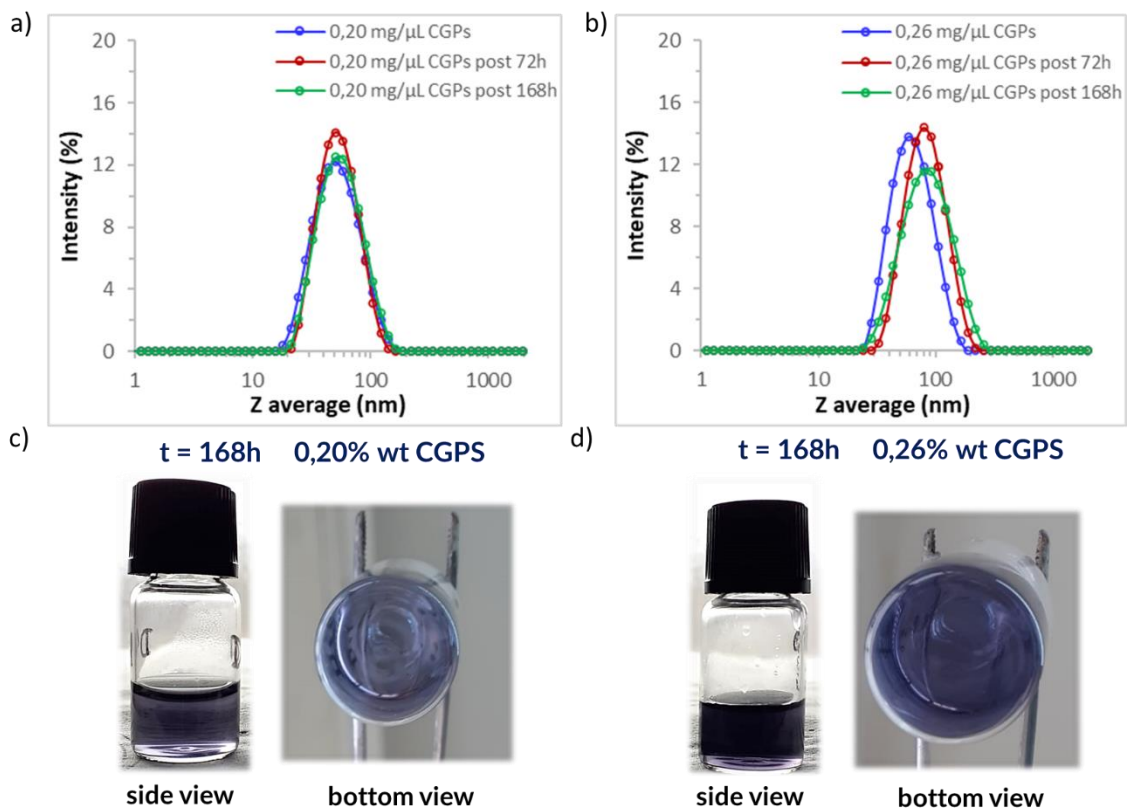


Figure S11. DLS (represented in Z average) of different concentrations of CGPs/DMSO solutions in H₂O at t = 0 h, t = 72 h and t = 168 h.

S6. Characterization of compounds 1

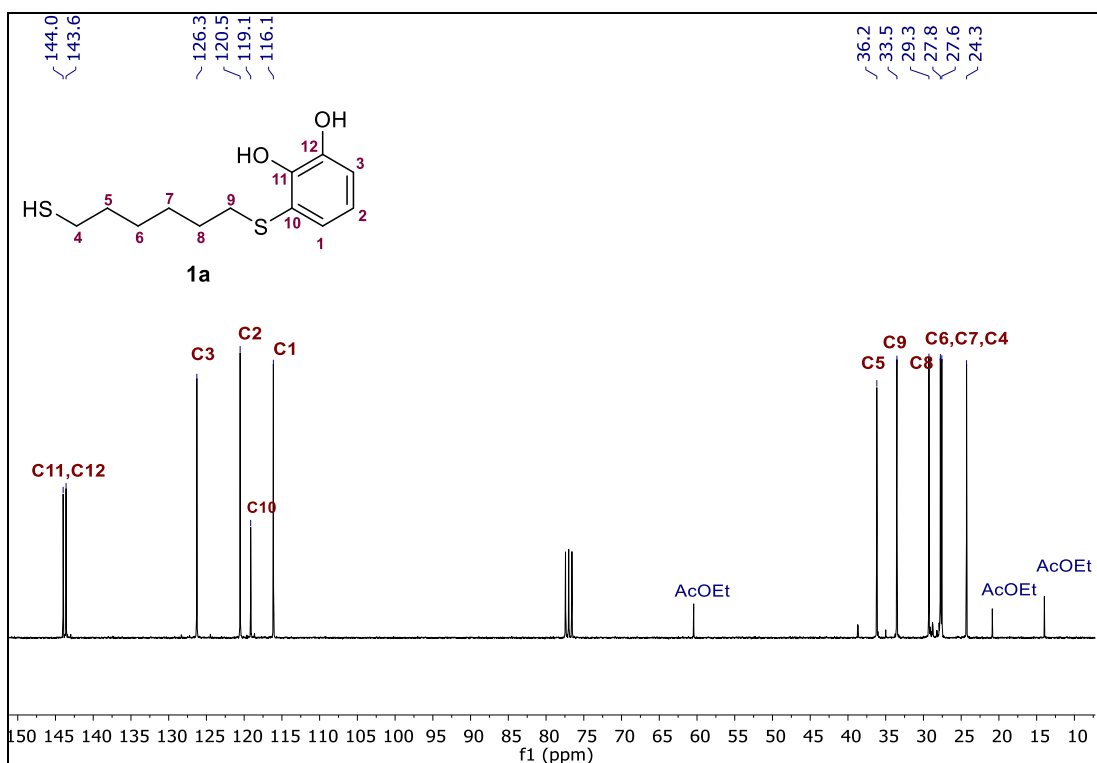
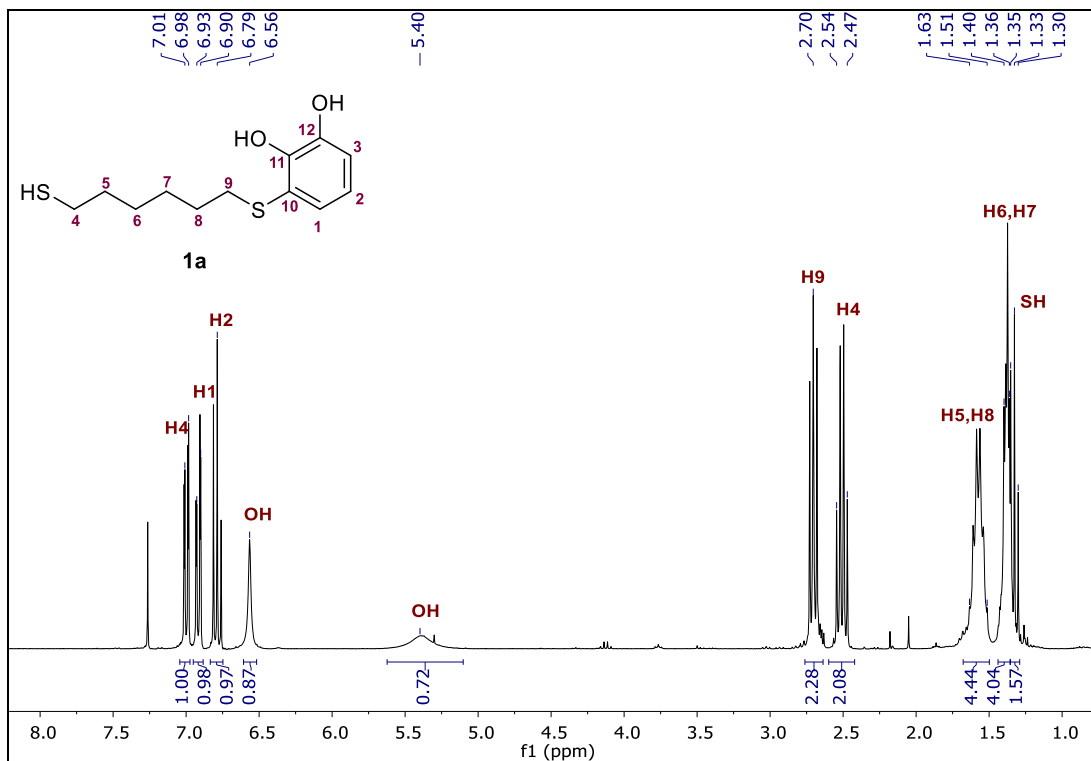


Figure S13. ^{13}C NMR of **1a in CDCl_3**

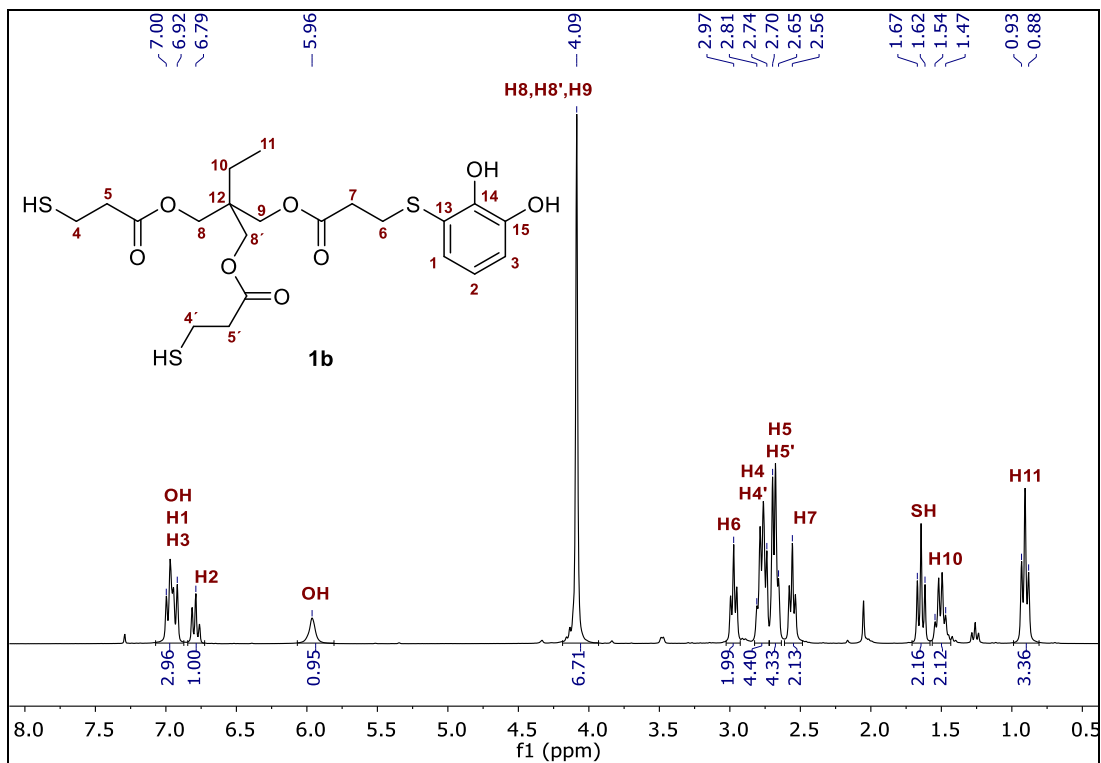


Figure S14. ¹H NMR of 1b in CDCl₃

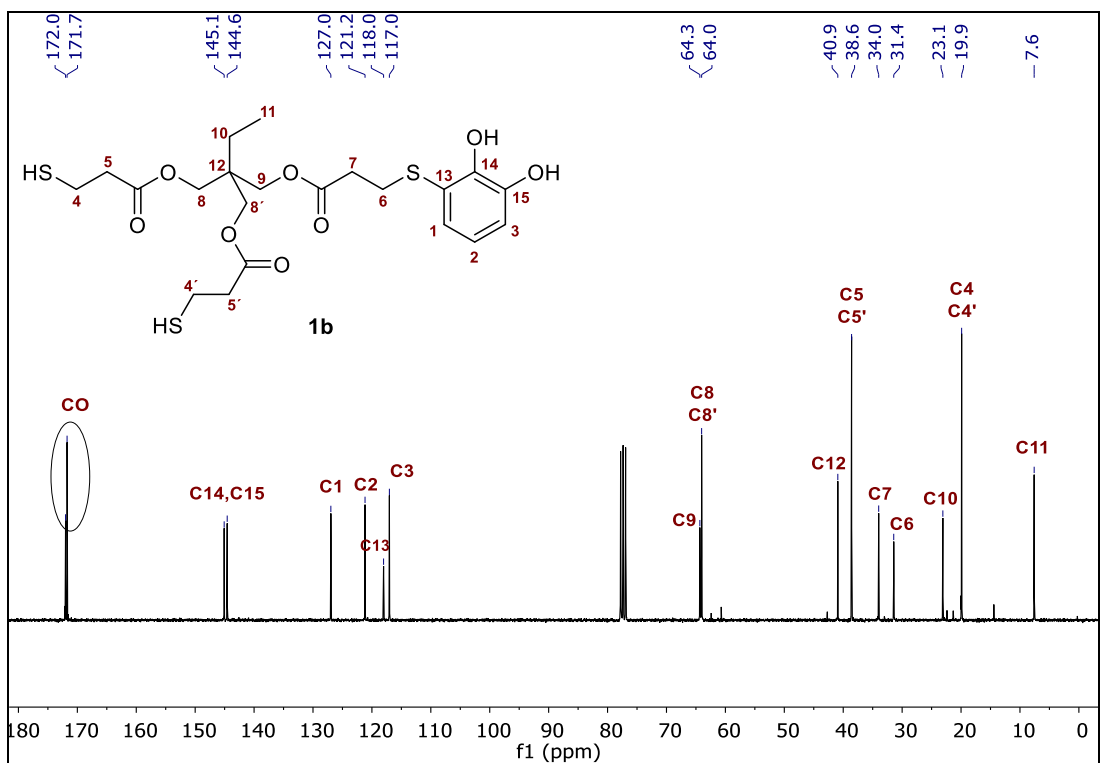


Figure S15. ¹³C NMR of 1b in CDCl₃

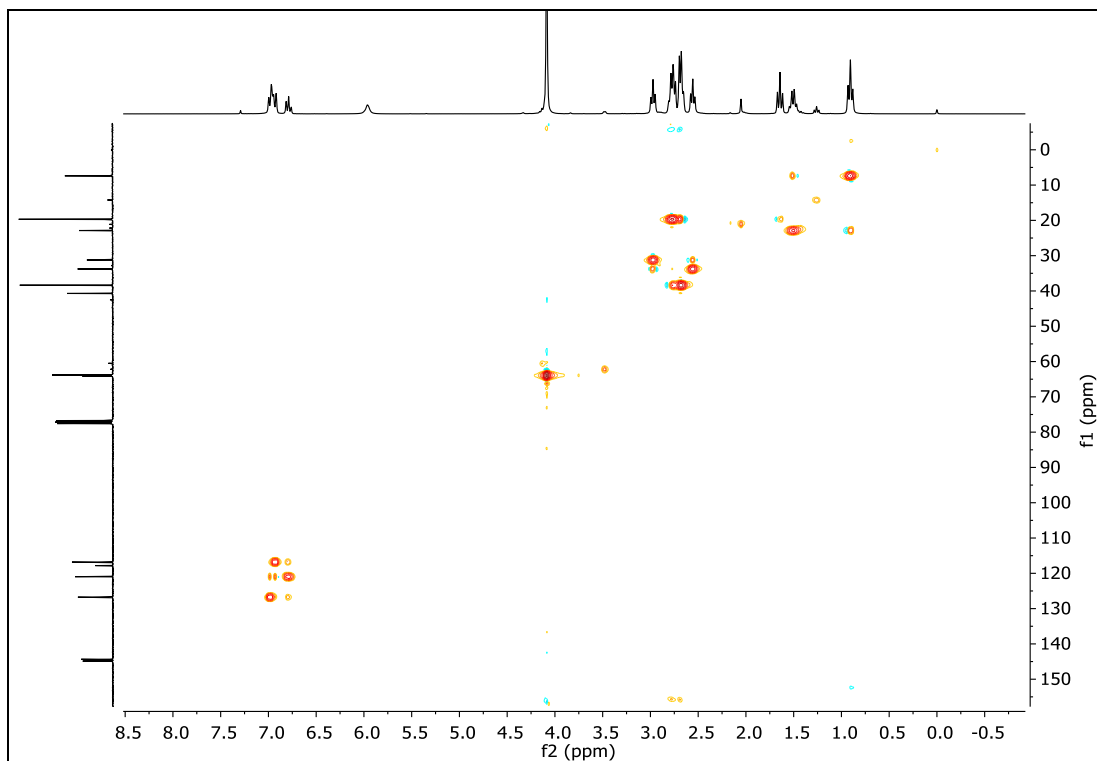


Figure S16. HSQC of **1b** in CDCl_3

S7. Characterization of compounds **2**

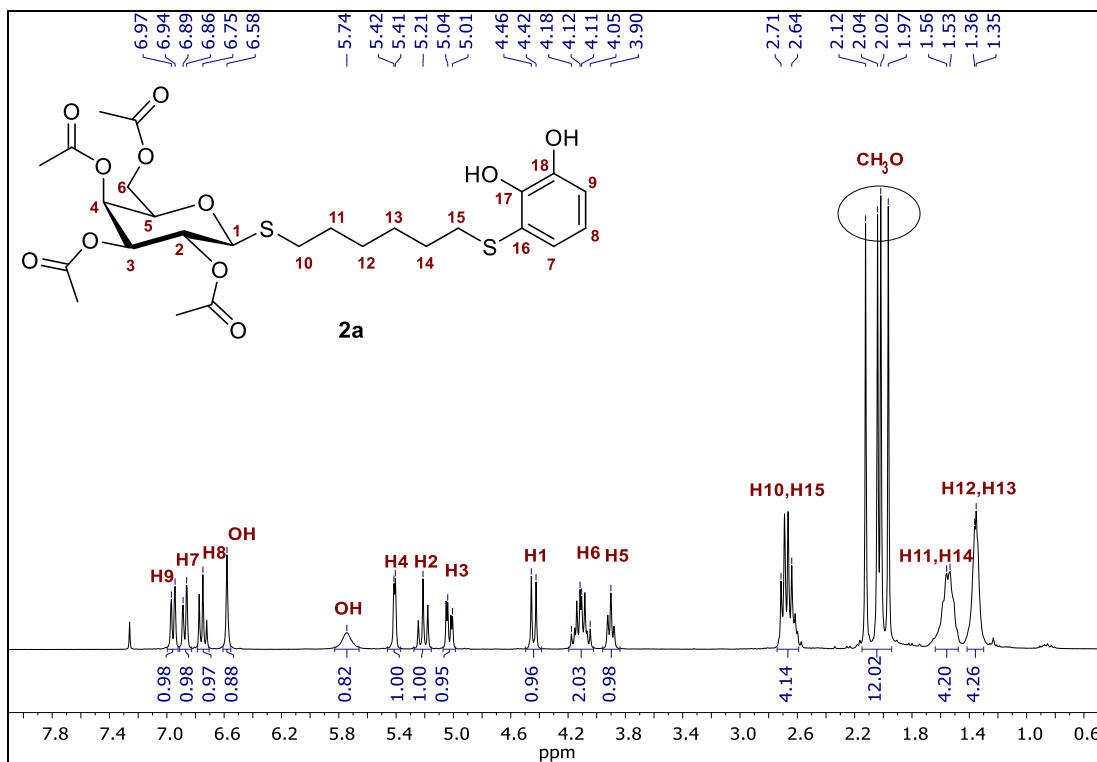


Figure S17. ^1H NMR of **2a** in CDCl_3

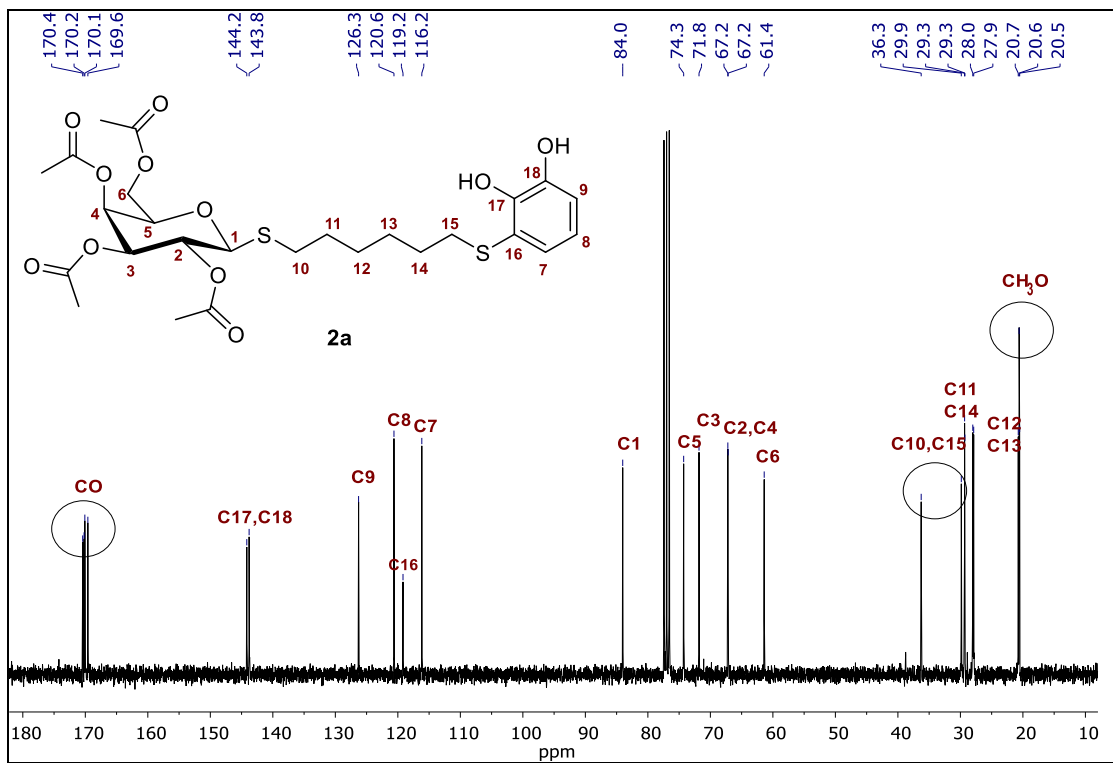


Figure S18. ^{13}C NMR of **2a** in CDCl_3

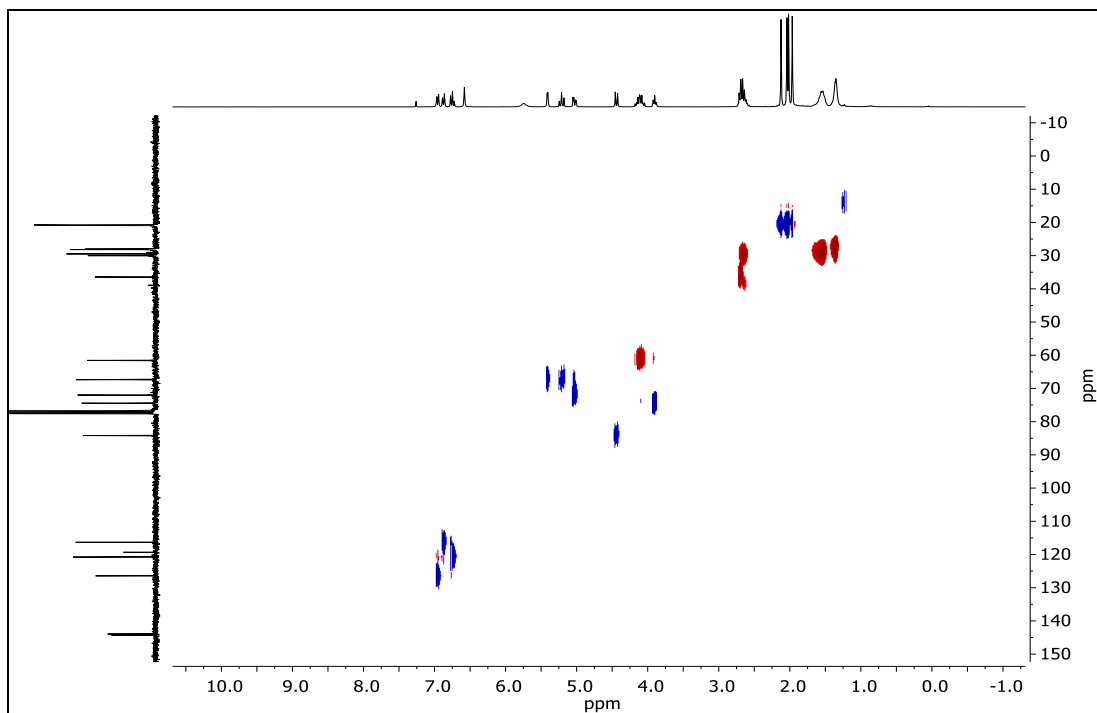


Figure S19. HSQC of **2a** in CDCl_3

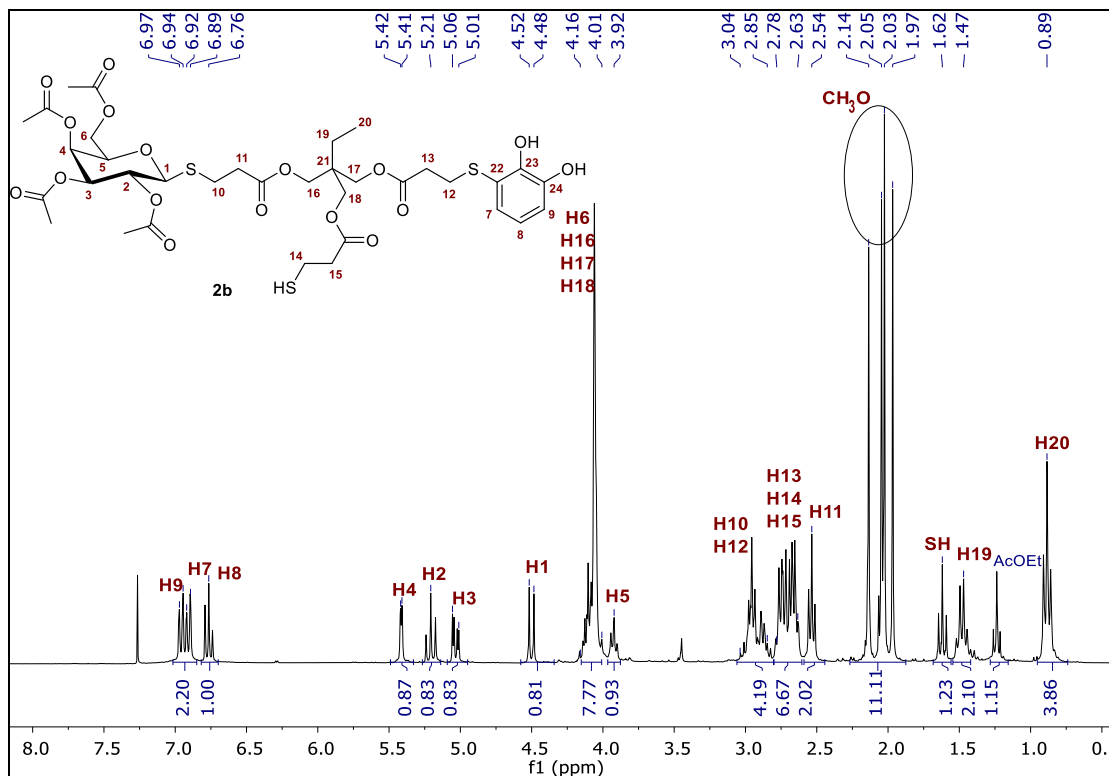


Figure S20. ^1H NMR of 2b in CDCl_3

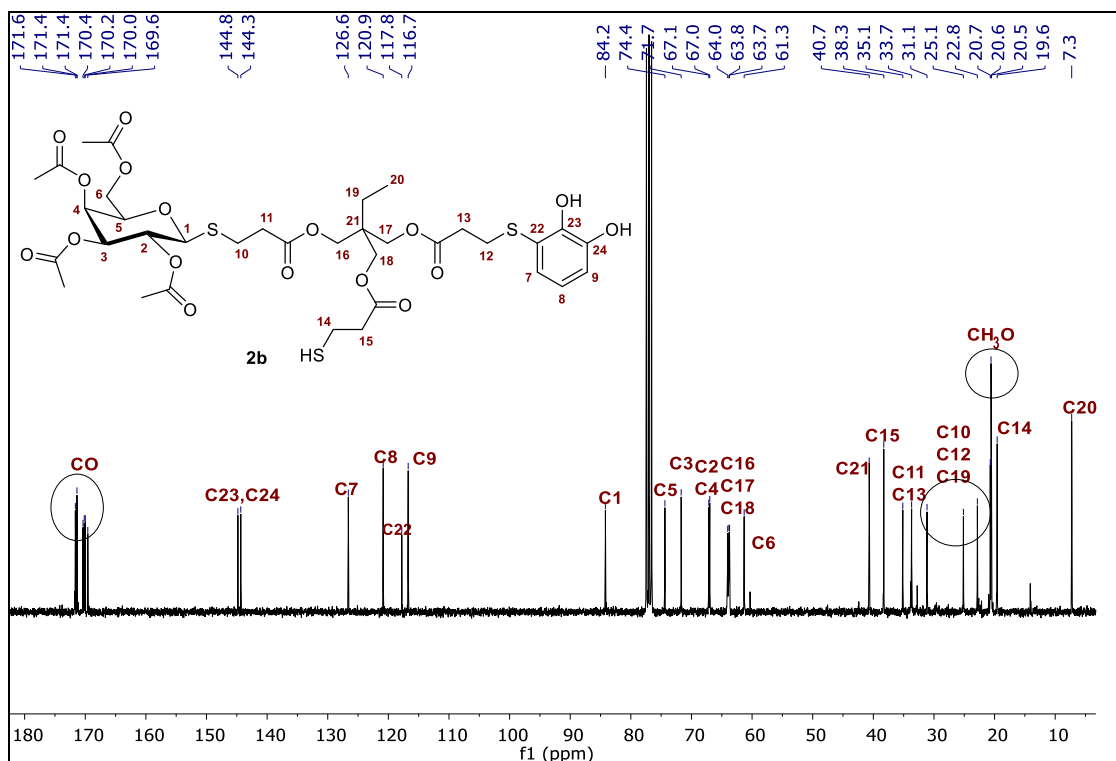


Figure S21. ^{13}C NMR of 2b in CDCl_3

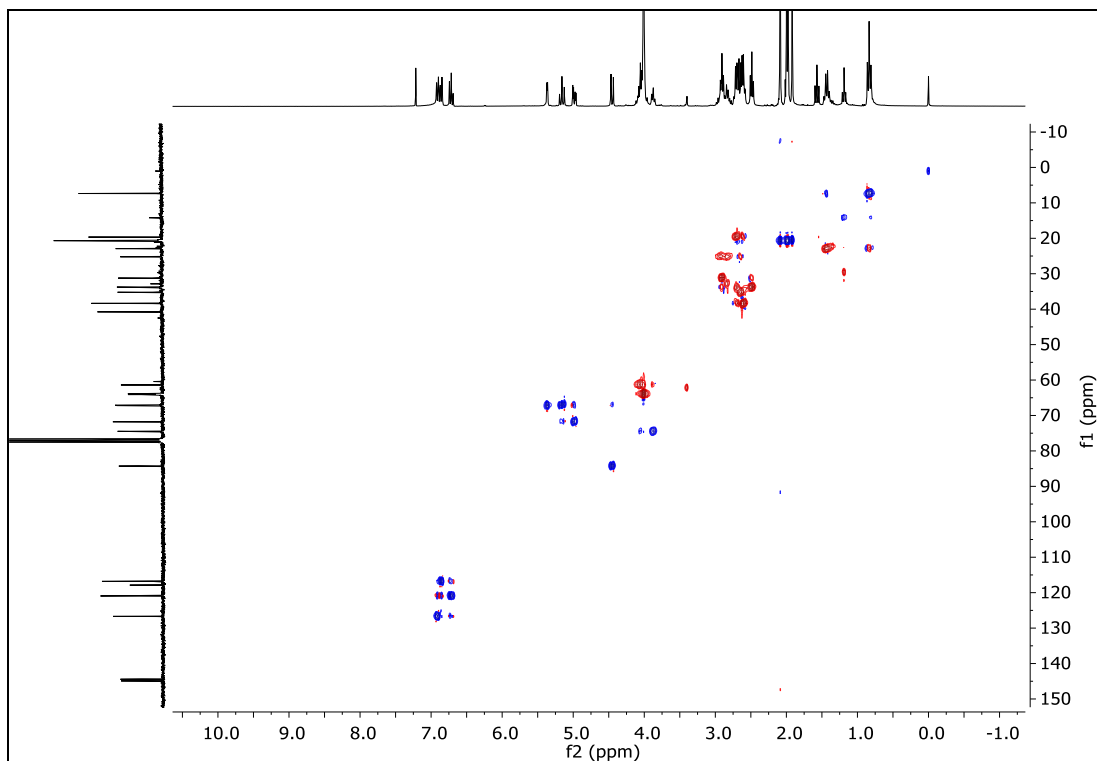


Figure S22. HSQC of **2b** in CDCl_3

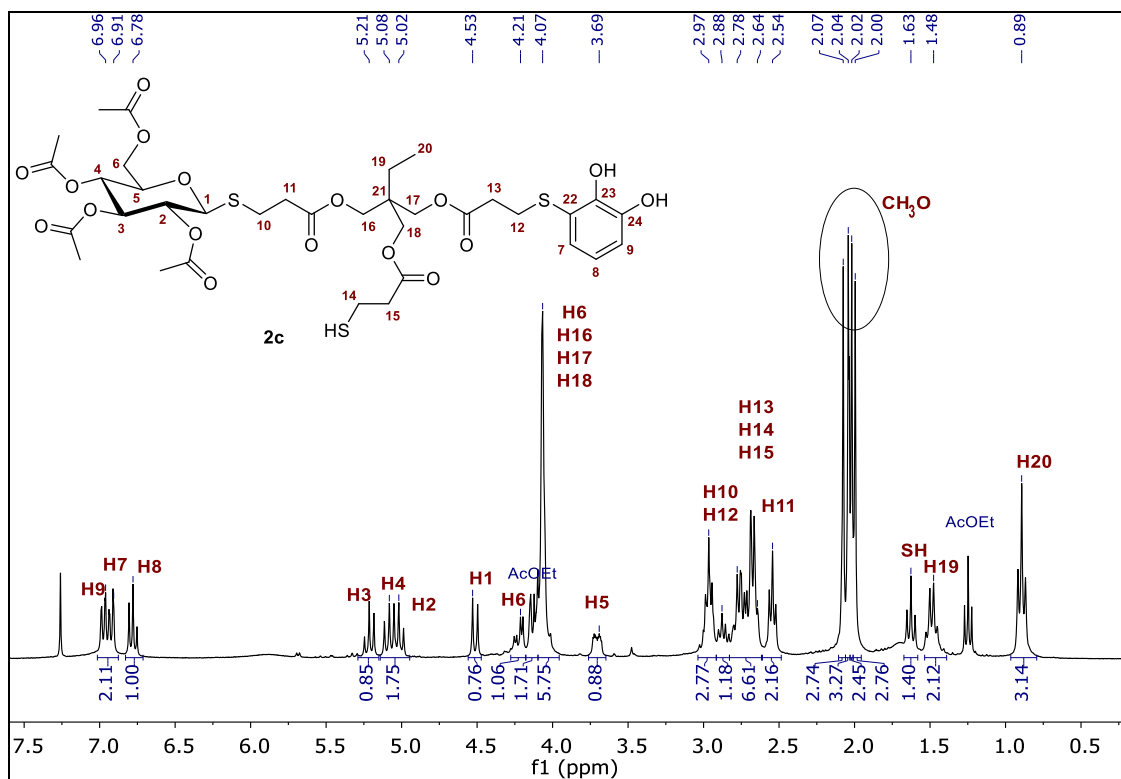
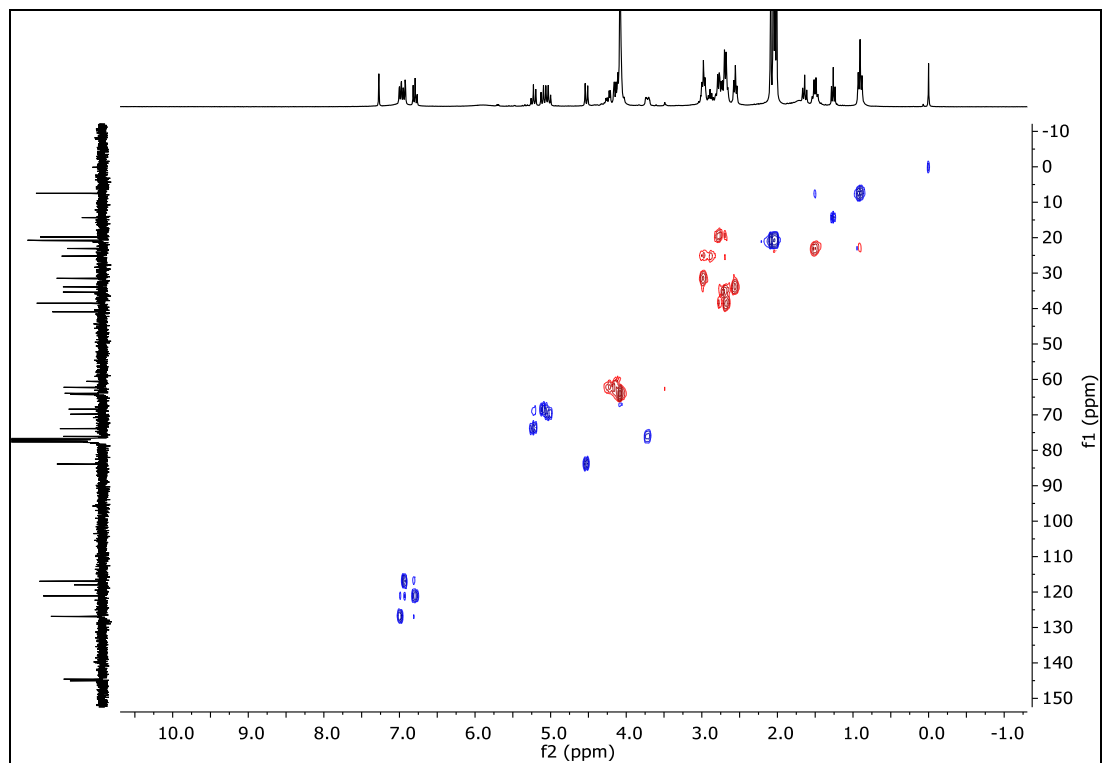
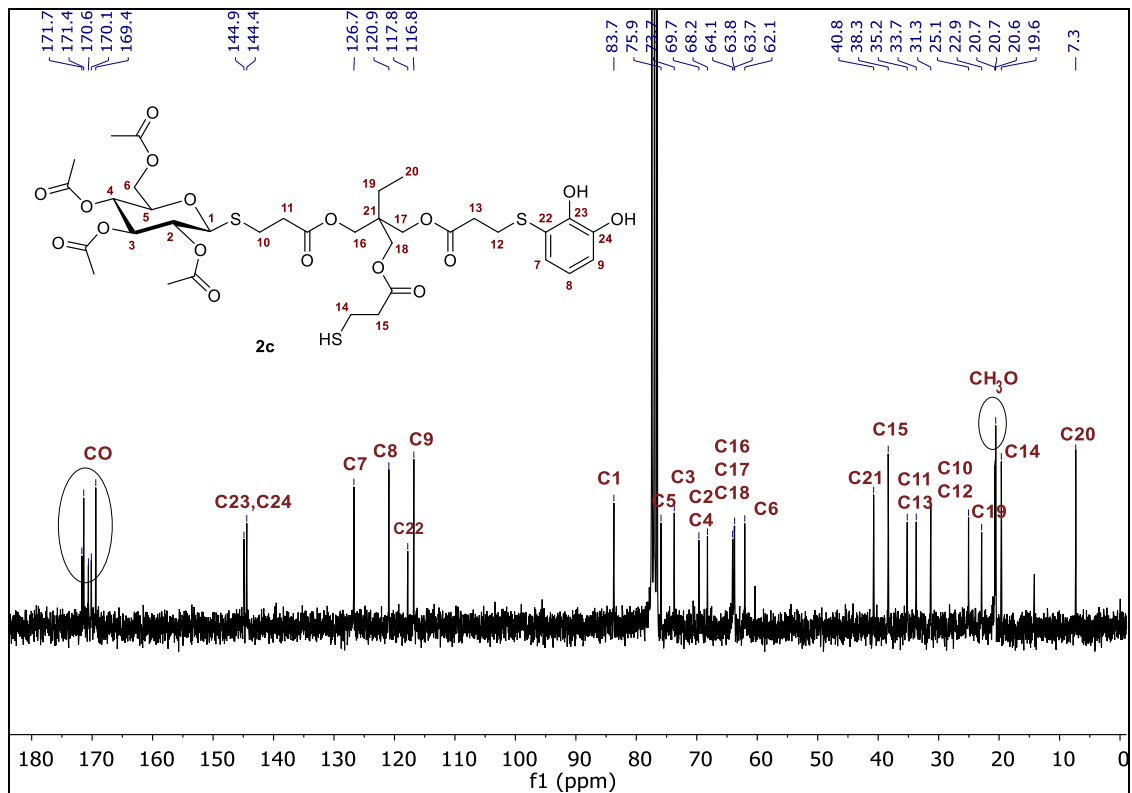


Figure S23. ^1H NMR of **2c** in CDCl_3



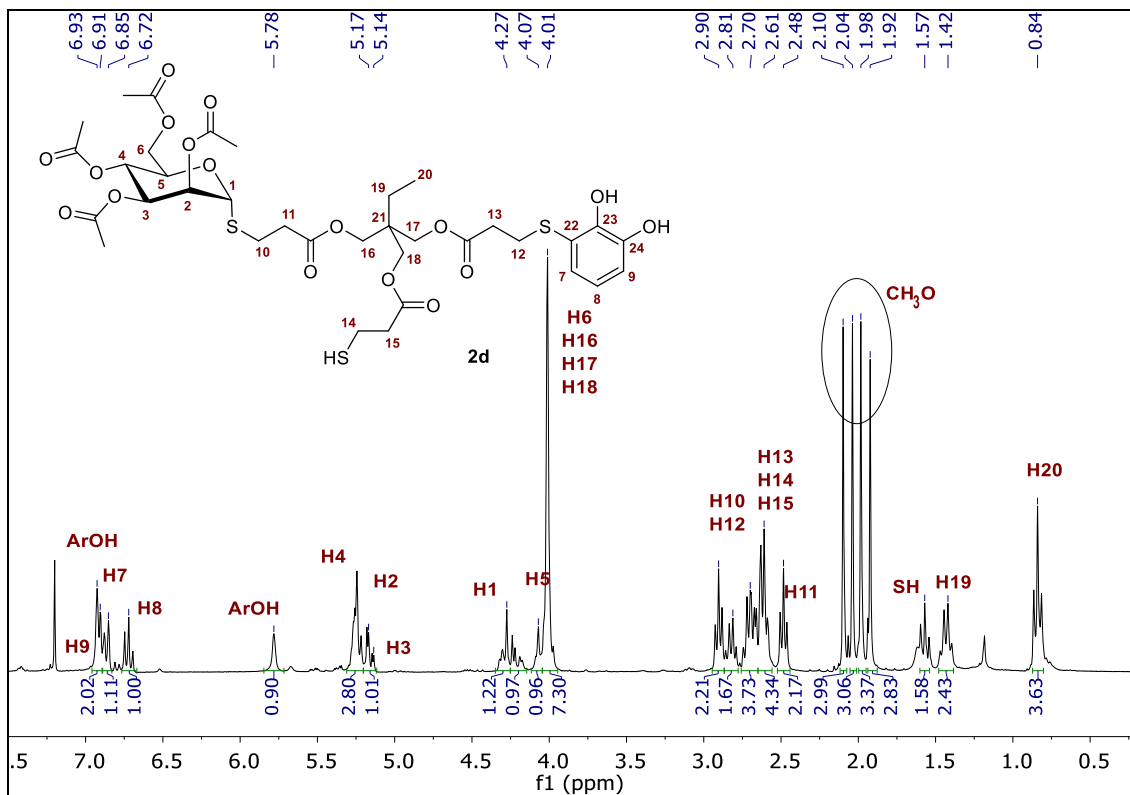


Figure S26. ¹H NMR of 2d in CDCl₃

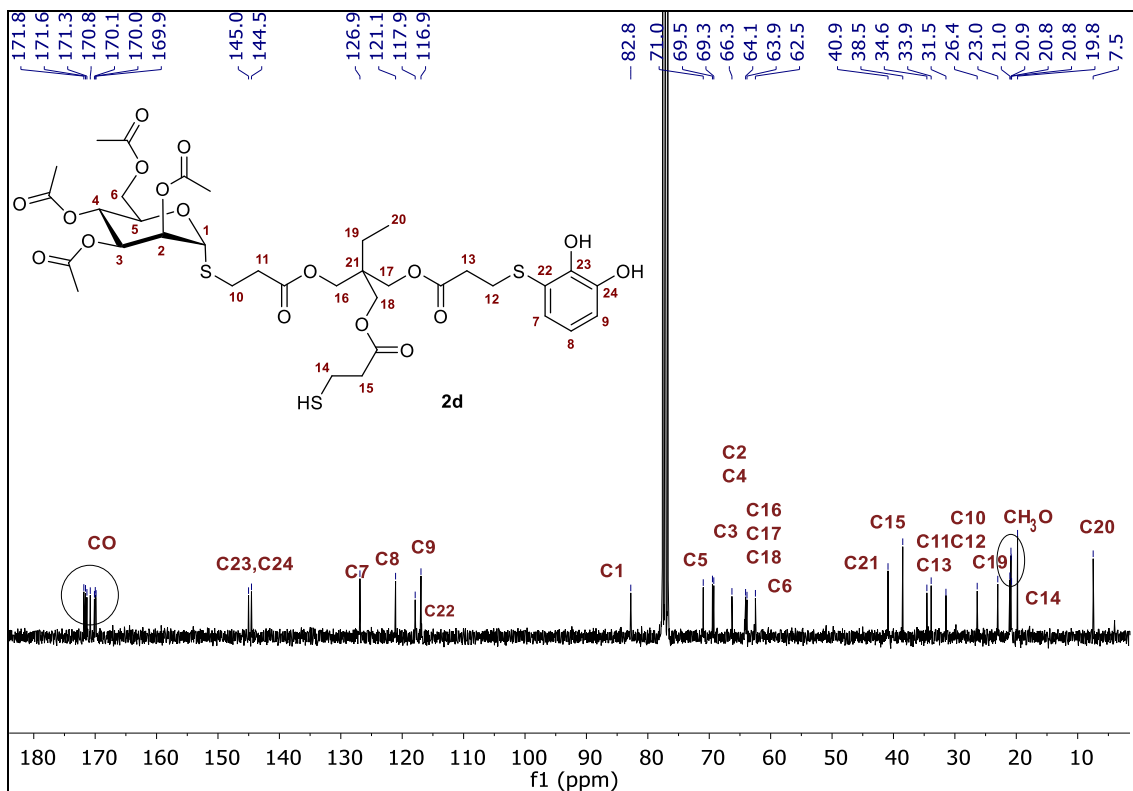


Figure S27. ¹³C NMR of 2d in CDCl₃

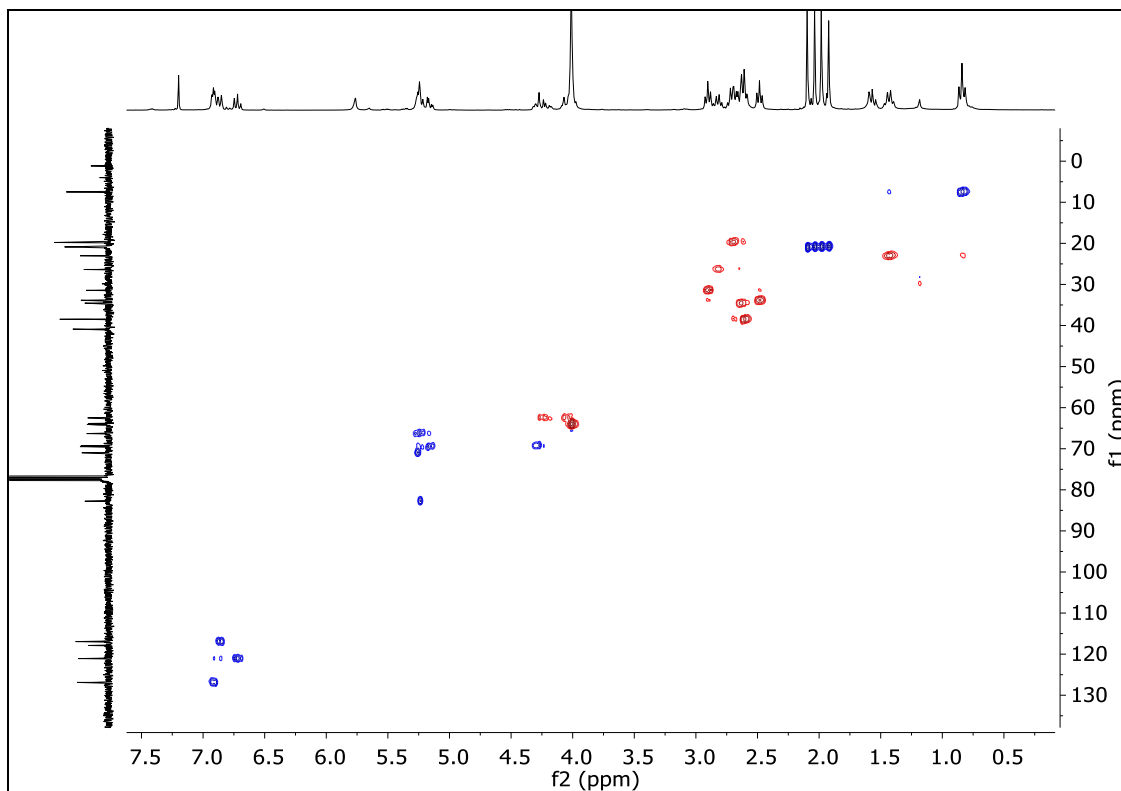


Figure S28. HSQC of **2d** in CDCl_3

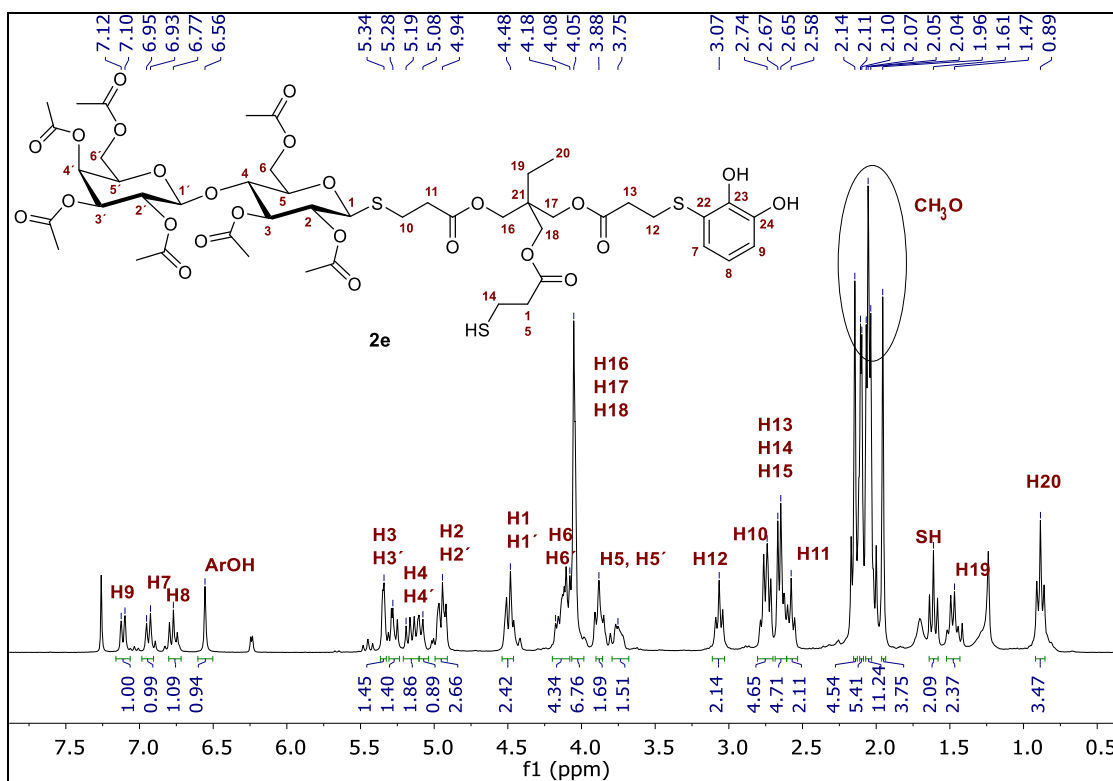
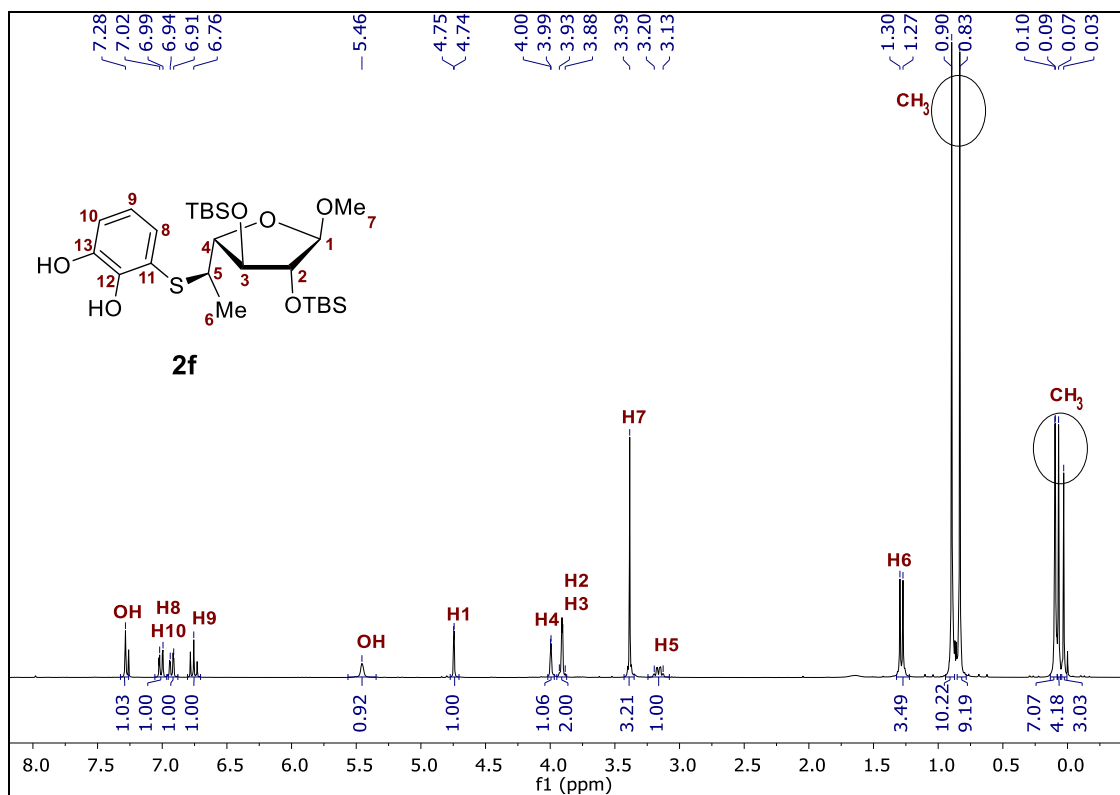
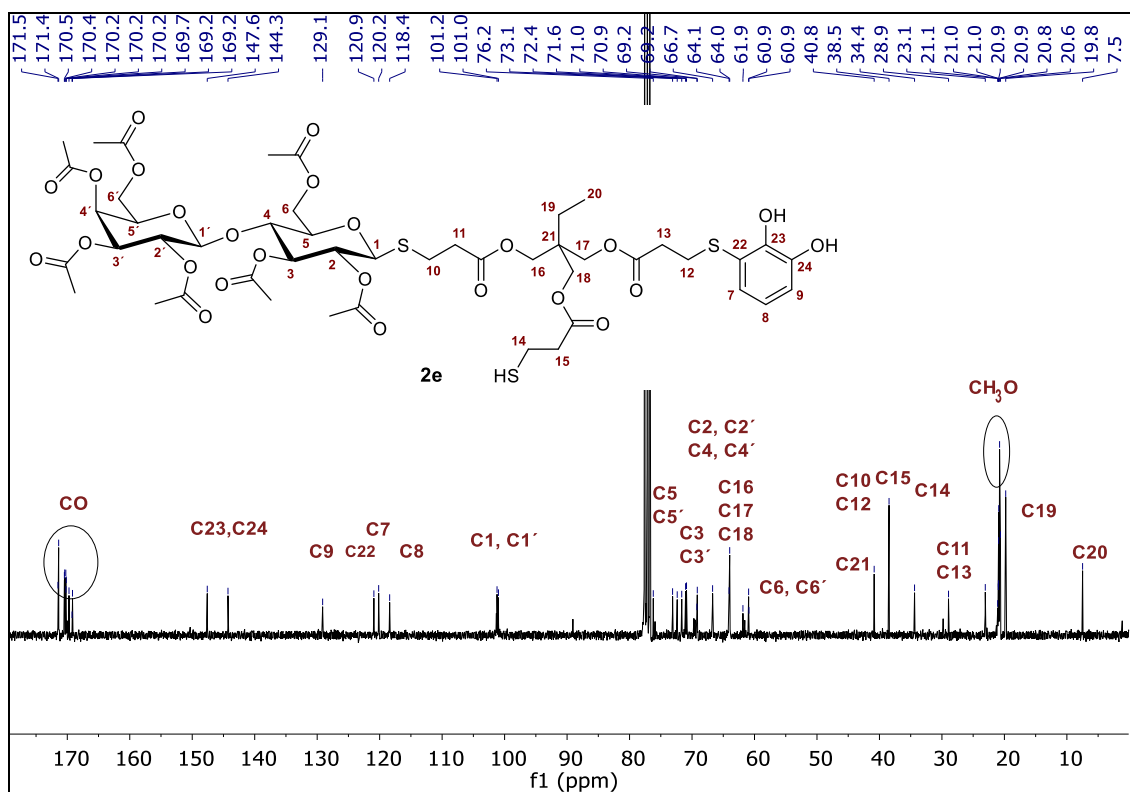
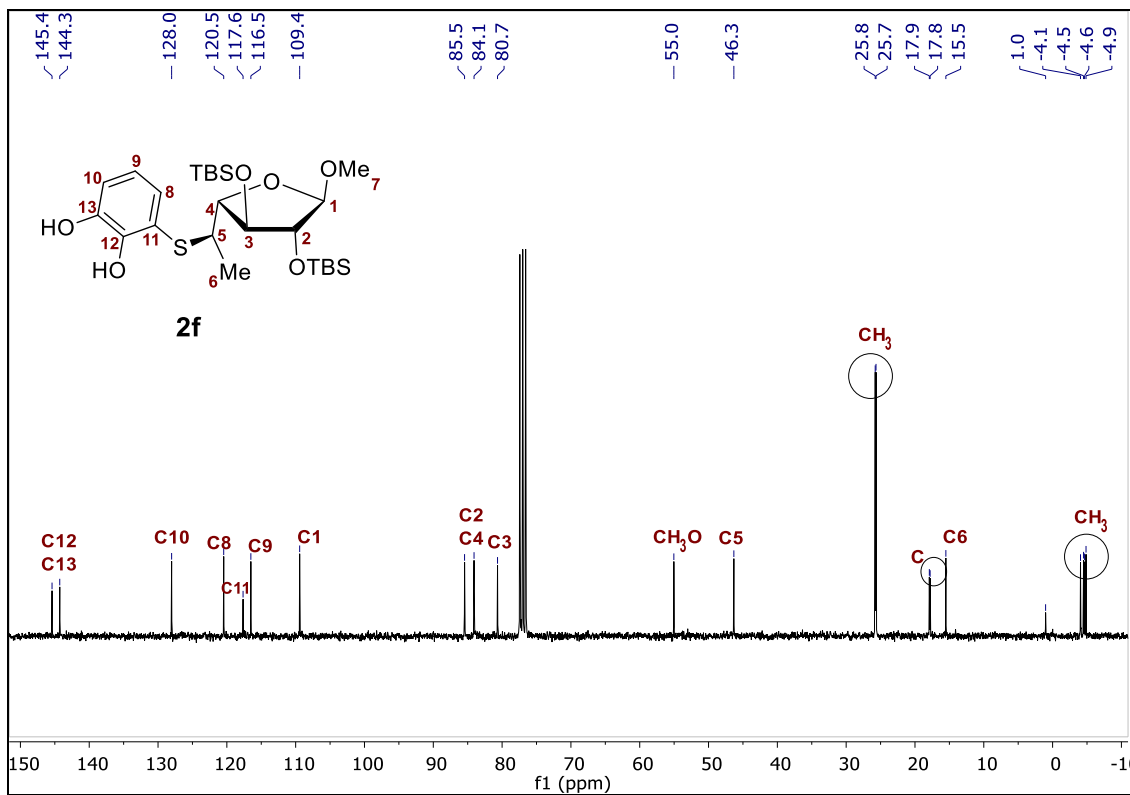
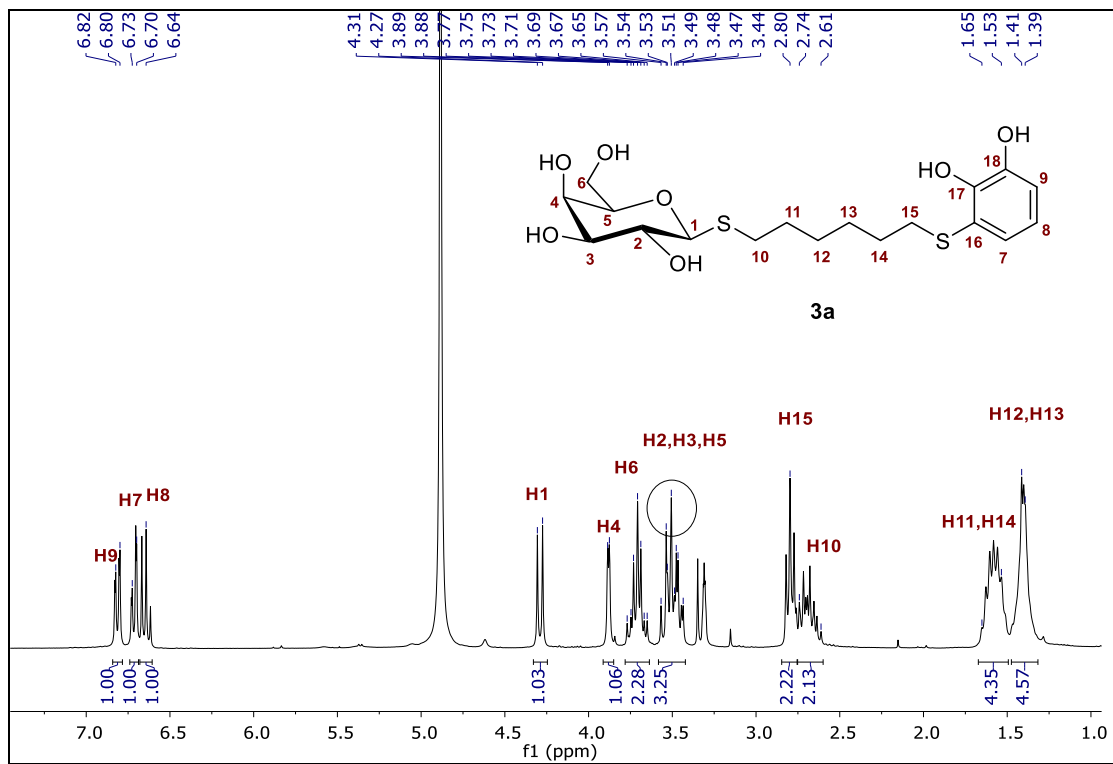


Figure S29. ^1H NMR of **2e** in CDCl_3





S8. Characterization of compounds 3



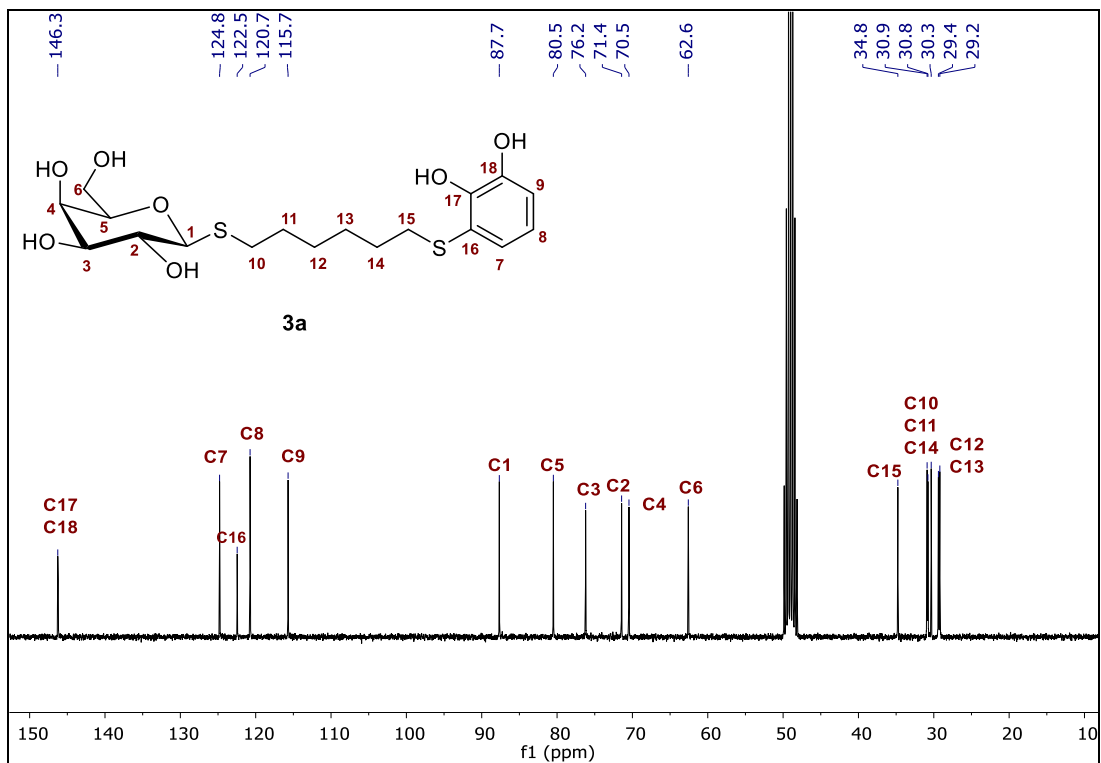


Figure S34. ^{13}C NMR of **3a** in CD_3OD

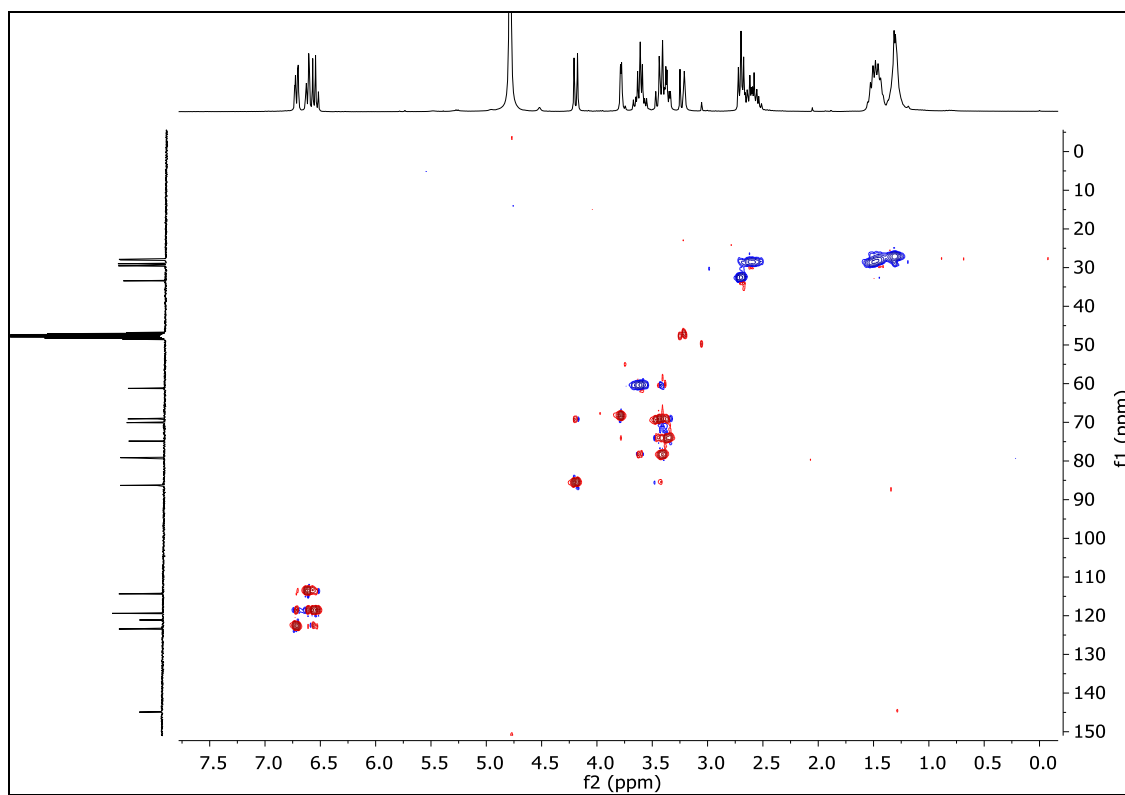


Figure S35. HSQC of **3a** in CD_3OD

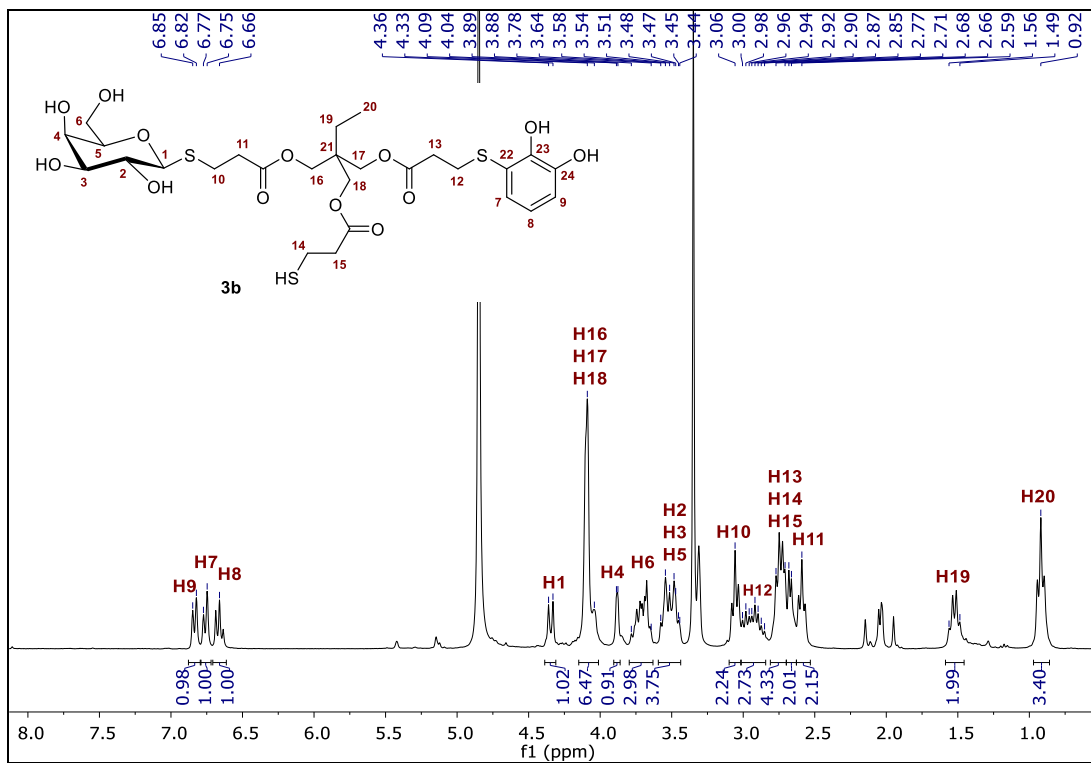


Figure S36. ¹H NMR of 3b in CD₃OD

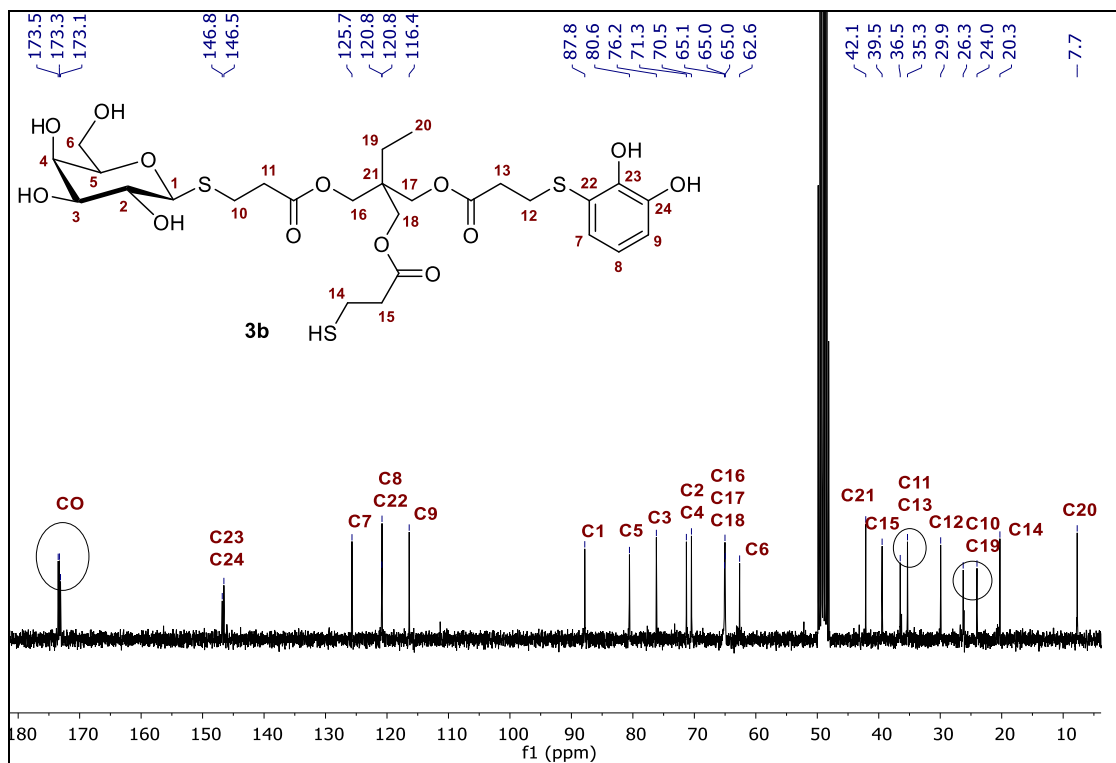


Figure S37. ¹³C NMR of 3b in CD₃OD

S9. Characterization of compounds 4

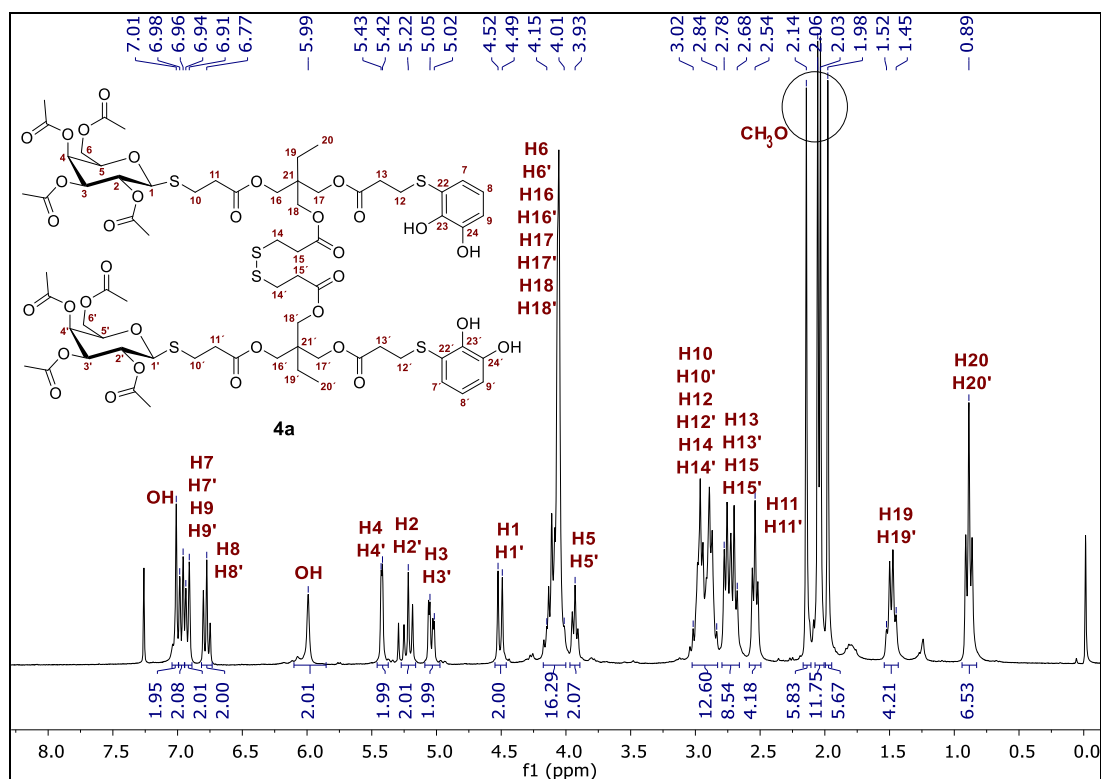


Figure S38. ¹H NMR of 4a in CDCl₃

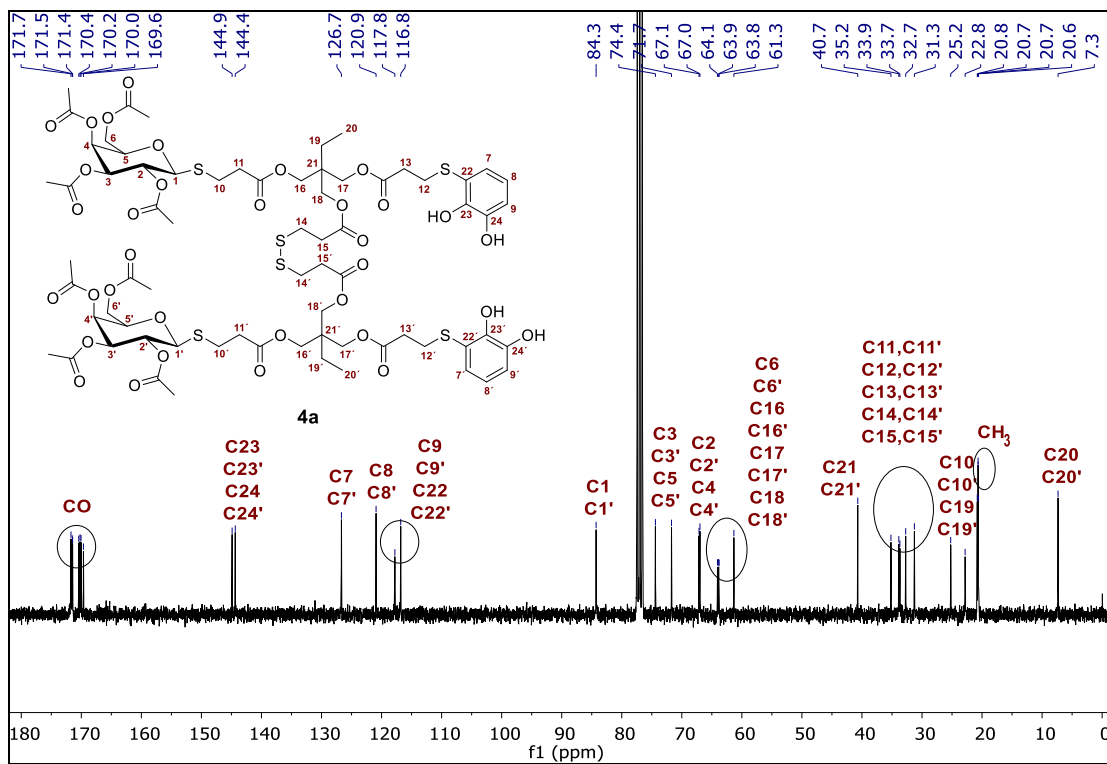


Figure S39. ¹³C NMR of 4a in CDCl₃