

Supporting information for: Microwave-Assisted Commercial Copper-Catalyzed Aerobic Oxidative Synthesis of AChE Quinazolinone Inhibitors under Solvent Free Conditions

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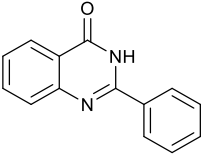
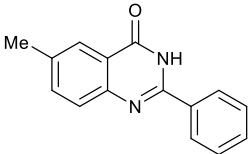
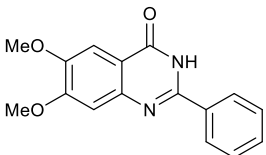
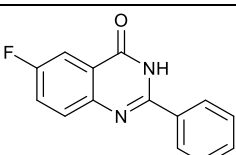
1. General methods

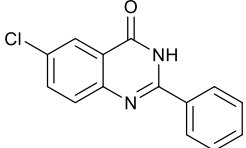
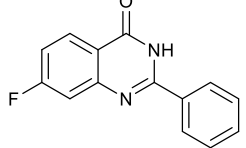
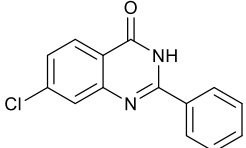
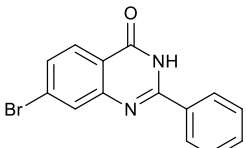
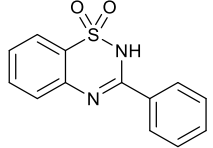
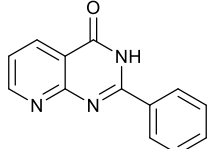
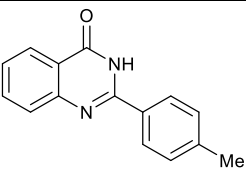
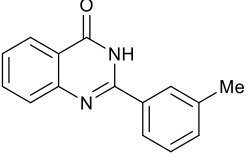
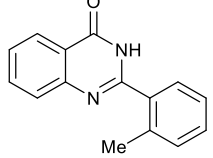
The Duran[®] Culture Tube with PBT screw caps were dried in the oven (120 °C) over night and cooled to room temperature. All aminobenzamide substrates and aldehydes were purchased from commercially available sources. The thin-layer chromatography (TLC) was performed on aluminum plates coated with Kieselgel 60 (0.20 mm, UV 254) and visualized under ultraviolet light. Silica gel (70 – 230 mesh) was purchased from Silicycle. ¹HNMR spectra were recorded at 300 MHz in DMSO-d₆ and referenced internally to the residual DMSO-d₆ signal 2.50 ppm). ¹³CNMR spectra were recorded at 75 MHz in DMSO-d₆ and referenced to the central peak of DMSO-d₆ (39.92ppm). Chemical shifts were reported in ppm (δ scale) and coupling constants (*J*) were reported in Hertz

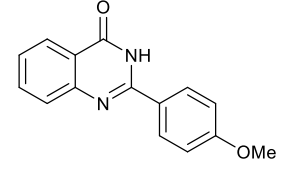
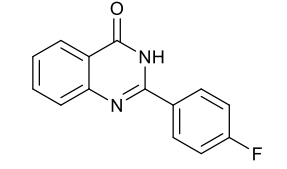
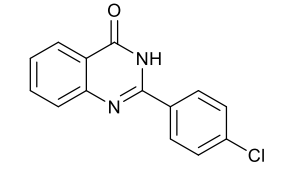
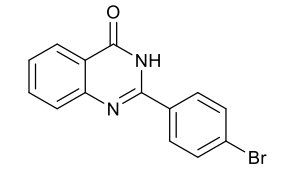
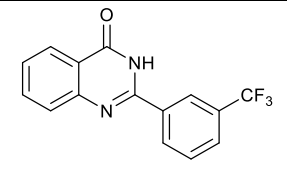
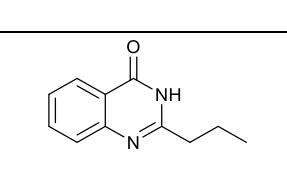
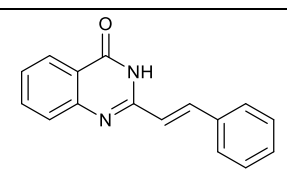
(Hz). High Performance Microwave Digestion System, Milestone MA182 ETHOS UP with PTFE vessel.

2. Procedure for synthesis of quinazolinone

A mixture of *o*-aminobenzamide (0.5 mmol), aldehydes (2.5 mmol), CuI (20 mol%) and Cs₂CO₃ (0.75 mmol) under oxygen atmosphere were mixed in Duran[®] Culture Tube with PBT screw caps. The culture tube was inserted to PTFE vessel including 15 mL of purified water, then was kept in the microwave reactor (High Performance Microwave Digestion System, Milestone ETHOS UP) at 130 °C for 2 hours. After that the reaction mixture was cooled down to room temperature, the crude mixture was purified by dry loading silica gel column chromatography (gradient elution: 20%-70% v/v EtOAc/hexane) to yield a corresponding quinazolinone products. The spectroscopic data of the known quinazolinones were agreement with the previously reported data, the references as shown in below.

Compound	Quinzanolione	Yield	Spectroscopic data
3a		90	Sahoo, S.; Pal, S., <i>The Journal of Organic Chemistry</i> 2021 , 86(24), 18067-18080.
3b		56	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.
3c		40	Chen, J.; Liang, E.; Shi, J.; We, Y.; Wen, K.; Yao, X.; Tang, X., <i>RSC Advance</i> 2021 , 11, 4966-4970.
3d		79	Sardar, B.; Jamatia, R.; Samanta, A.; Srimani, D., <i>The Journal of Organic Chemistry</i> 2022 , 87(9), 5556-5567.

3e		46	Sardar, B.; Jamatia, R.; Samanta, A.; Srimani, D., <i>The Journal of Organic Chemistry</i> 2022 , 87(9), 5556-5567.
3f		36	Chen, J.; Liang, E.; Shi, J.; We, Y.; Wen, K.; Yao, X.; Tang, X., <i>RSC Advance</i> 2021 , 11, 4966-4970.
3g		62	Chen, J.; Liang, E.; Shi, J.; We, Y.; Wen, K.; Yao, X.; Tang, X., <i>RSC Advance</i> 2021 , 11, 4966-4970.
3h		14	Chen, J.; Liang, E.; Shi, J.; We, Y.; Wen, K.; Yao, X.; Tang, X., <i>RSC Advance</i> 2021 , 11, 4966-4970.
3i		41	Jang, Y.; Lee, S. B.; Hong, J.; Lee, J.; Hong, S., <i>Organic & Biomolecular Chemistry</i> 2020 , 18, 5435-5441.
3j		48	Jang, Y.; Lee, S. B.; Hong, J.; Lee, J.; Hong, S., <i>Organic & Biomolecular Chemistry</i> 2020 , 18, 5435-5441.
4a		83	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.
4b		56	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.
4c		54	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.

4d		60	Sardar, B.; Jamatia, R.; Samanta, A.; Srimani, D., <i>The Journal of Organic Chemistry</i> 2022 , 87(9), 5556-5567.
4e		58	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.
4f		observed	Sahoo, S.; Pal, S., <i>The Journal of Organic Chemistry</i> 2021 , 86(24), 18067-18080.
4g		observed	Parua, S.; Das, S.; Sikari, R.; Sinha, S.; Paul, N. D., <i>The Journal of Organic Chemistry</i> 2017 , 82(14), 7165-7175.
4h		73	Laha, J. K.; Tummalapalli, K. S. S.; Nair, A.; Patel, N., <i>The Journal of Organic Chemistry</i> 2015 , 80(22), 11351-11359.
4i		92	Tain, X.; Song, L.; Li, E.; Wang, Q.; Yu, W.; Chang, J., <i>RSC Advance</i> , 2015 , 5, 62194-62201.
4j		55	Sahoo, S.; Pal, S., <i>The Journal of Organic Chemistry</i> 2021 , 86(24), 18067-18080.

3. NMR spectra of the quinazolone products

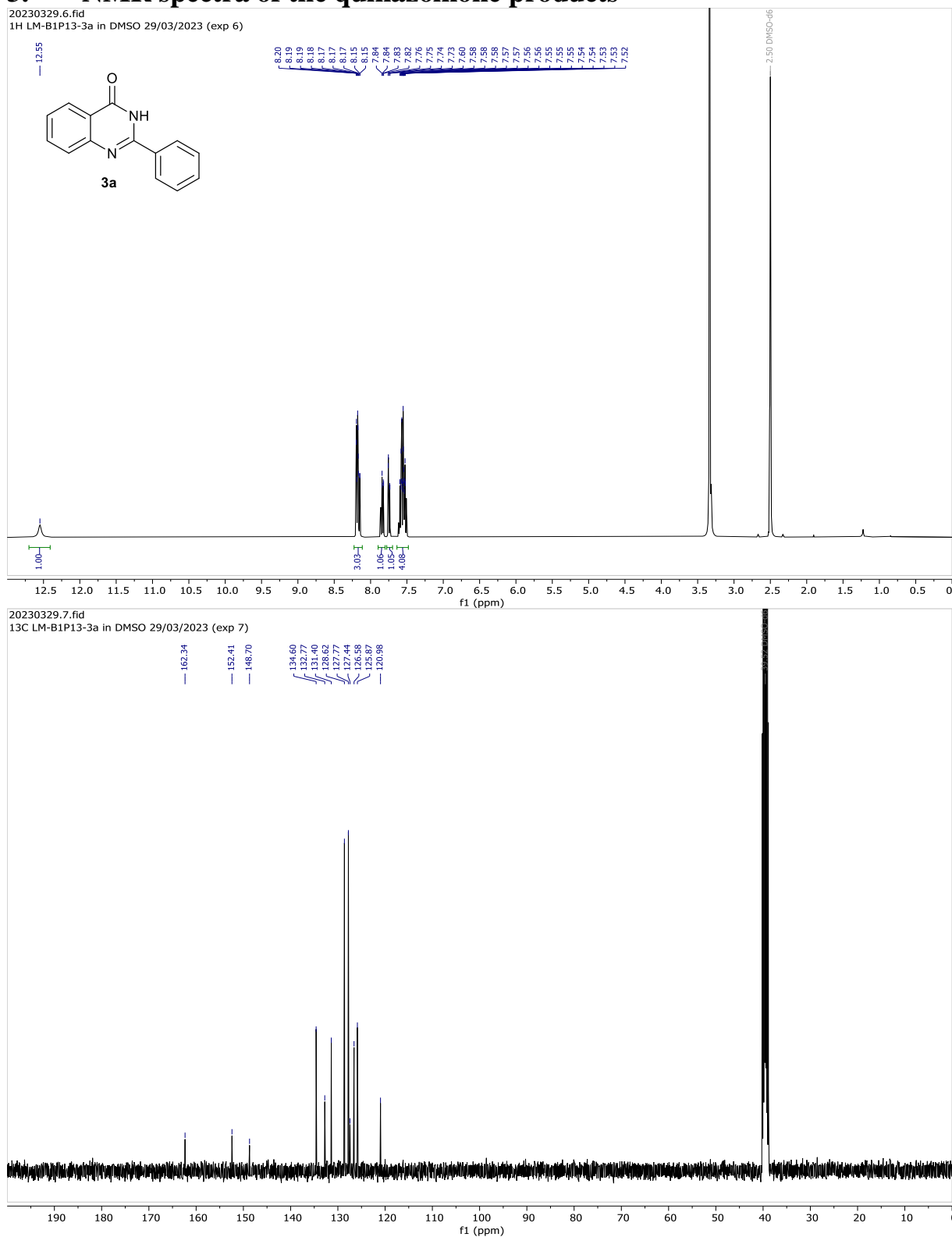


Figure 1 ^1H NMR and ^{13}C NMR spectra of **3a**

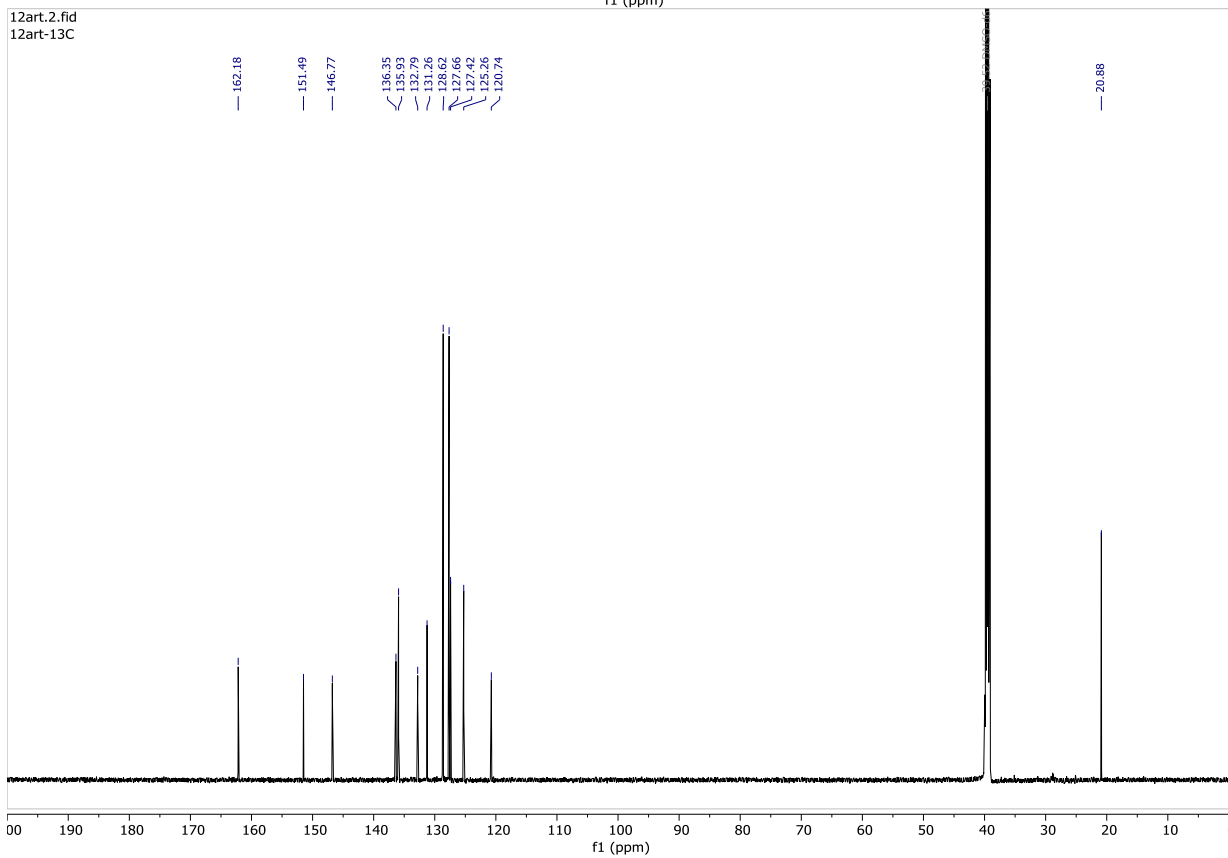
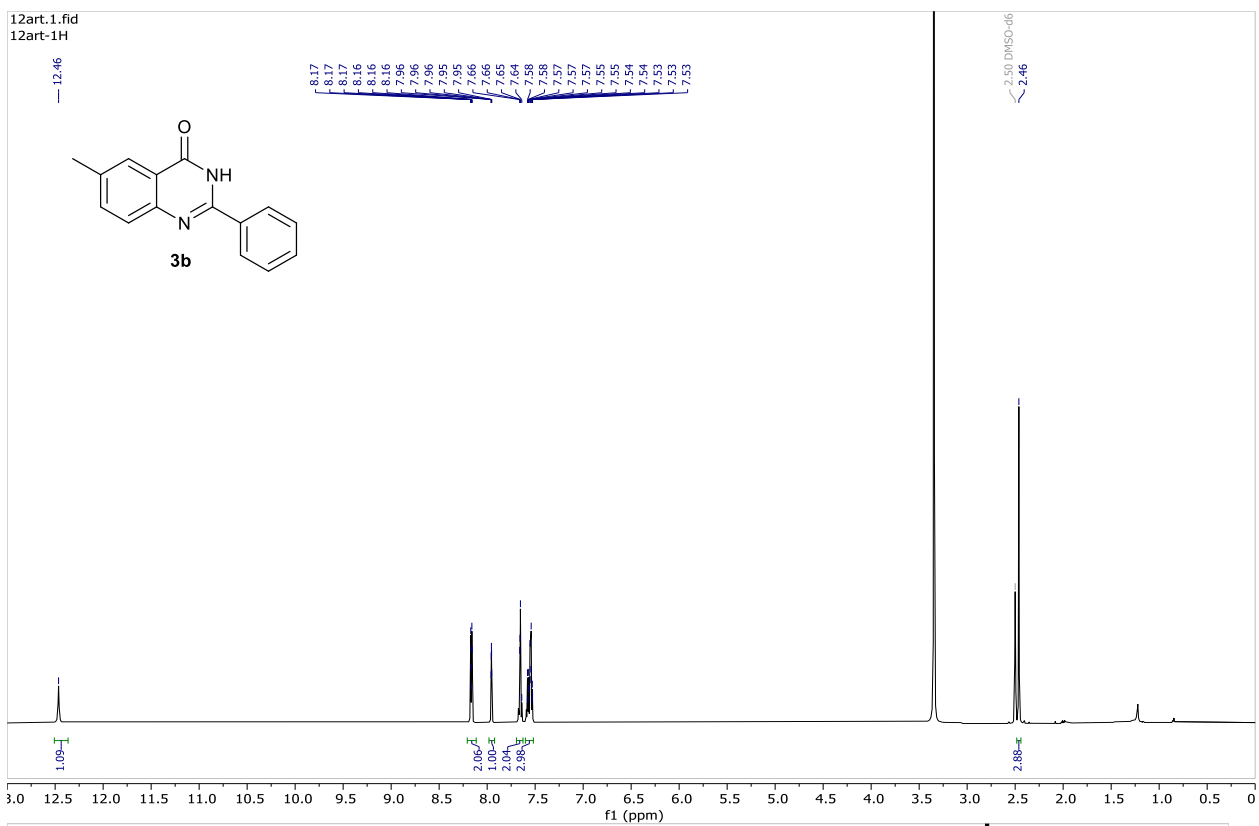


Figure 2 ^1H NMR and ^{13}C NMR spectra of **3b**

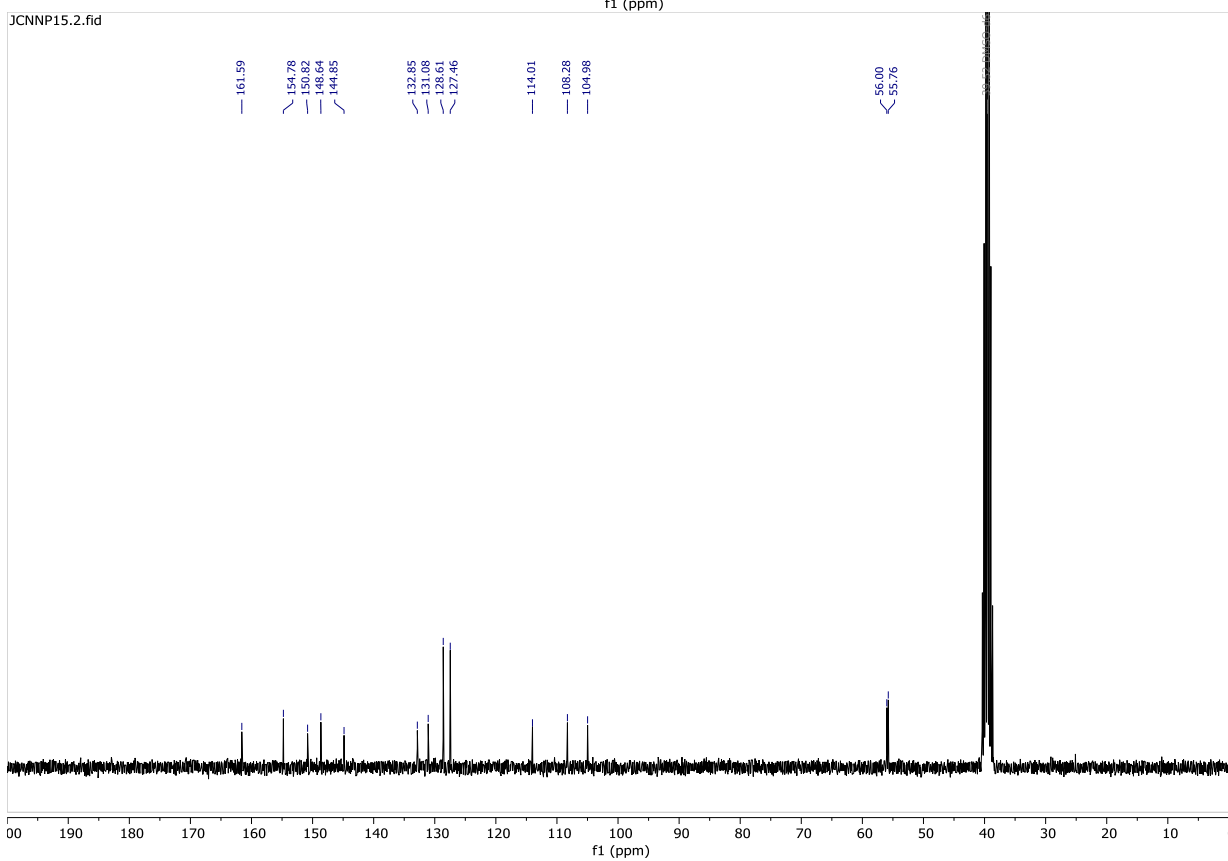
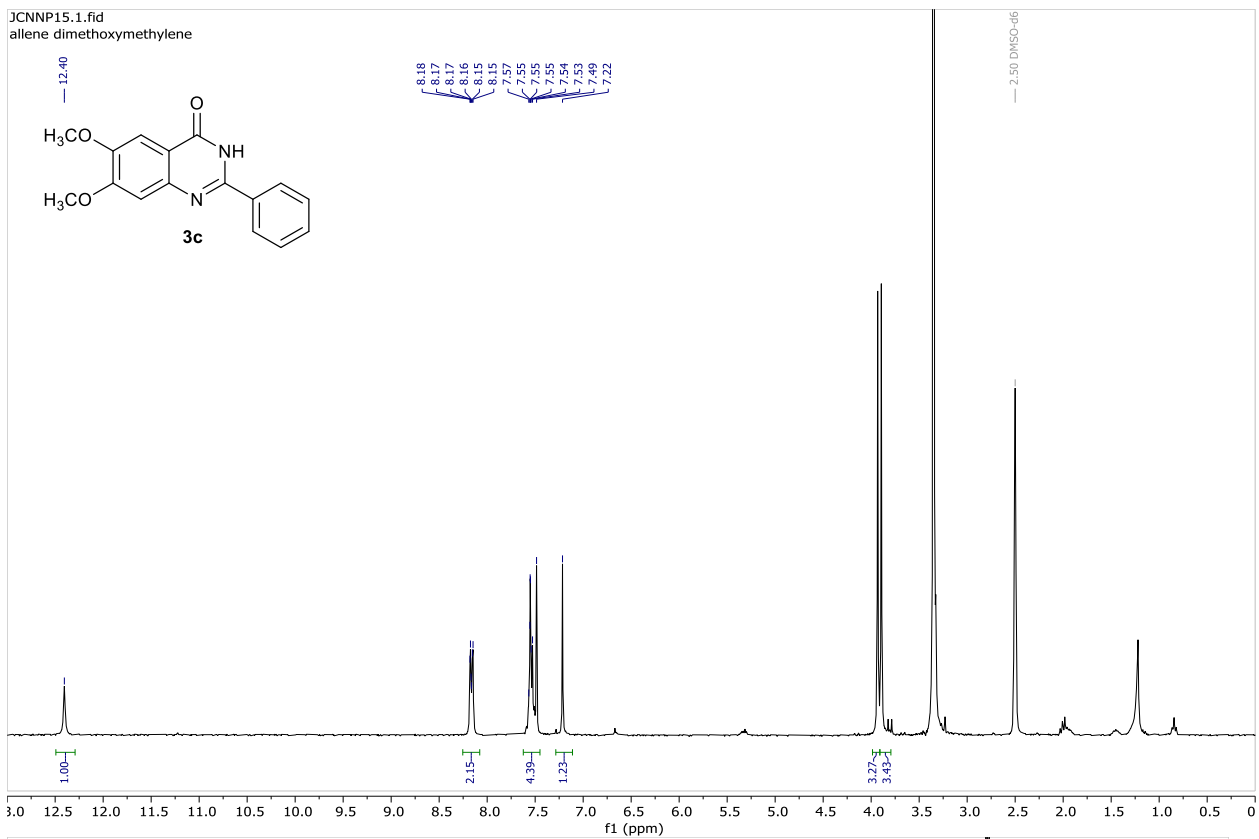


Figure 3 ¹H NMR and ¹³C NMR spectra of **3c**

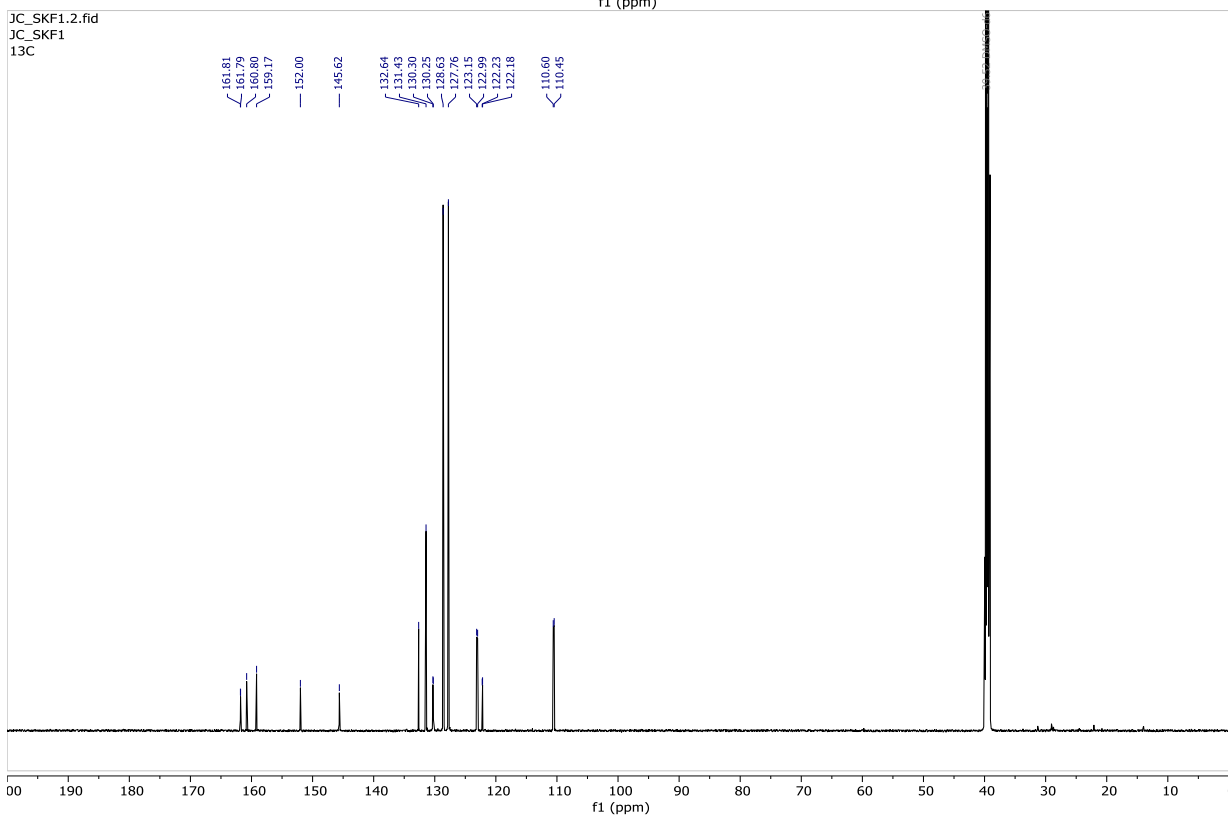
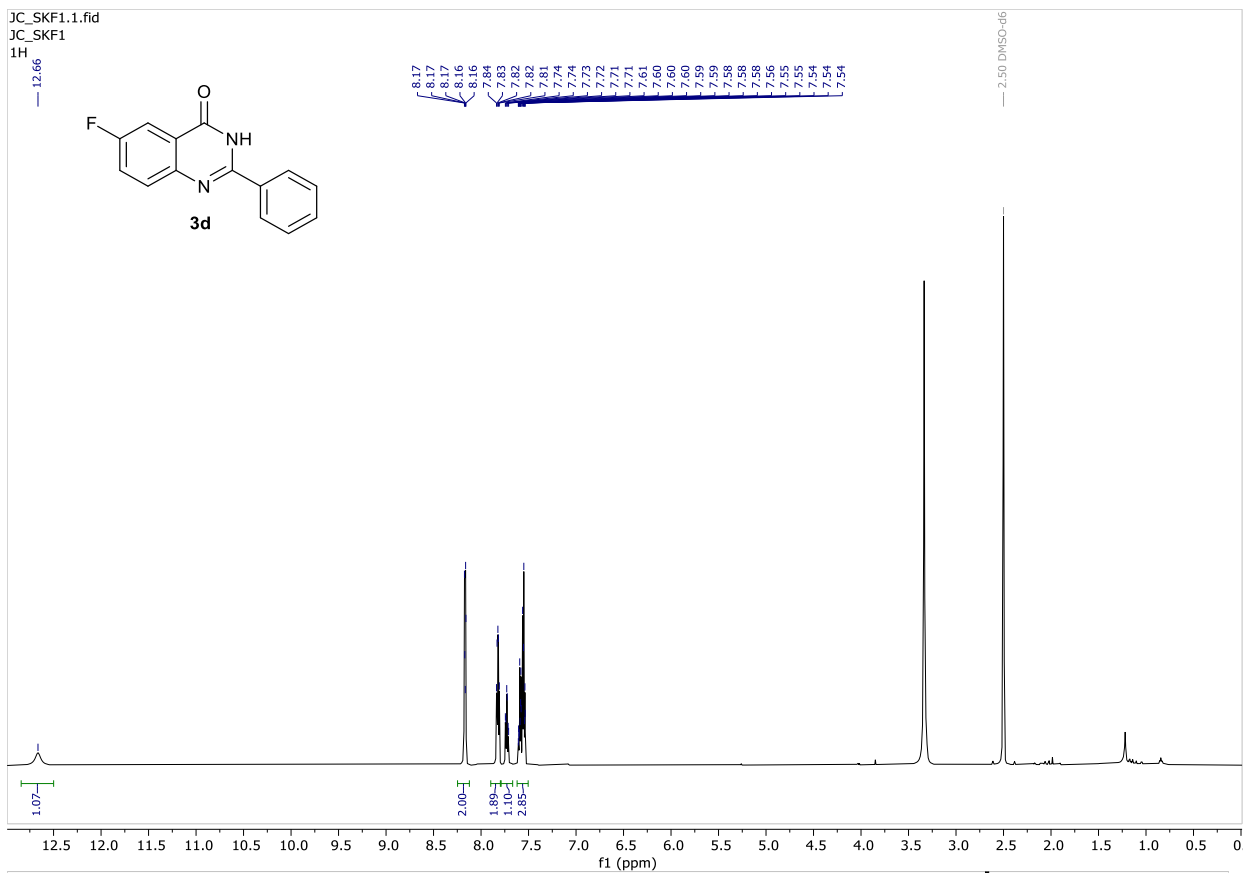


Figure 4 ^1H NMR and ^{13}C NMR spectra of **3d**

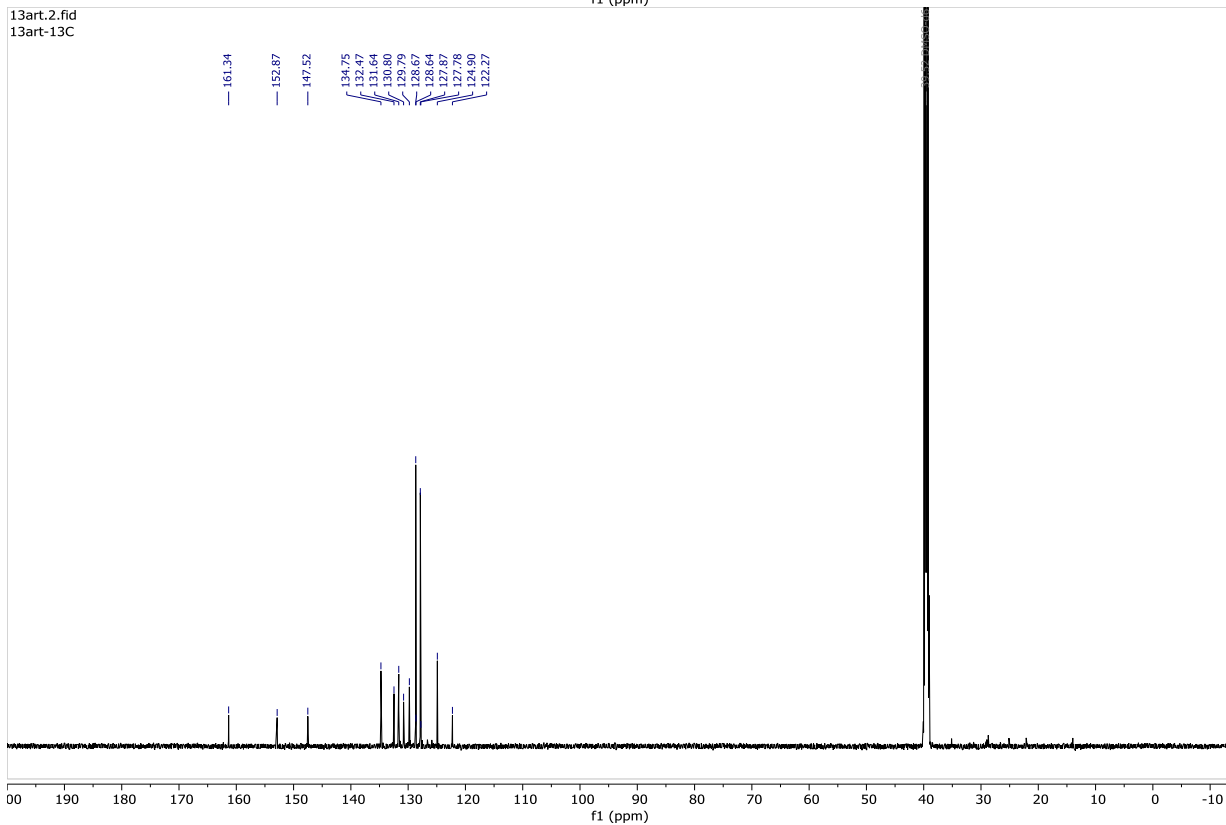
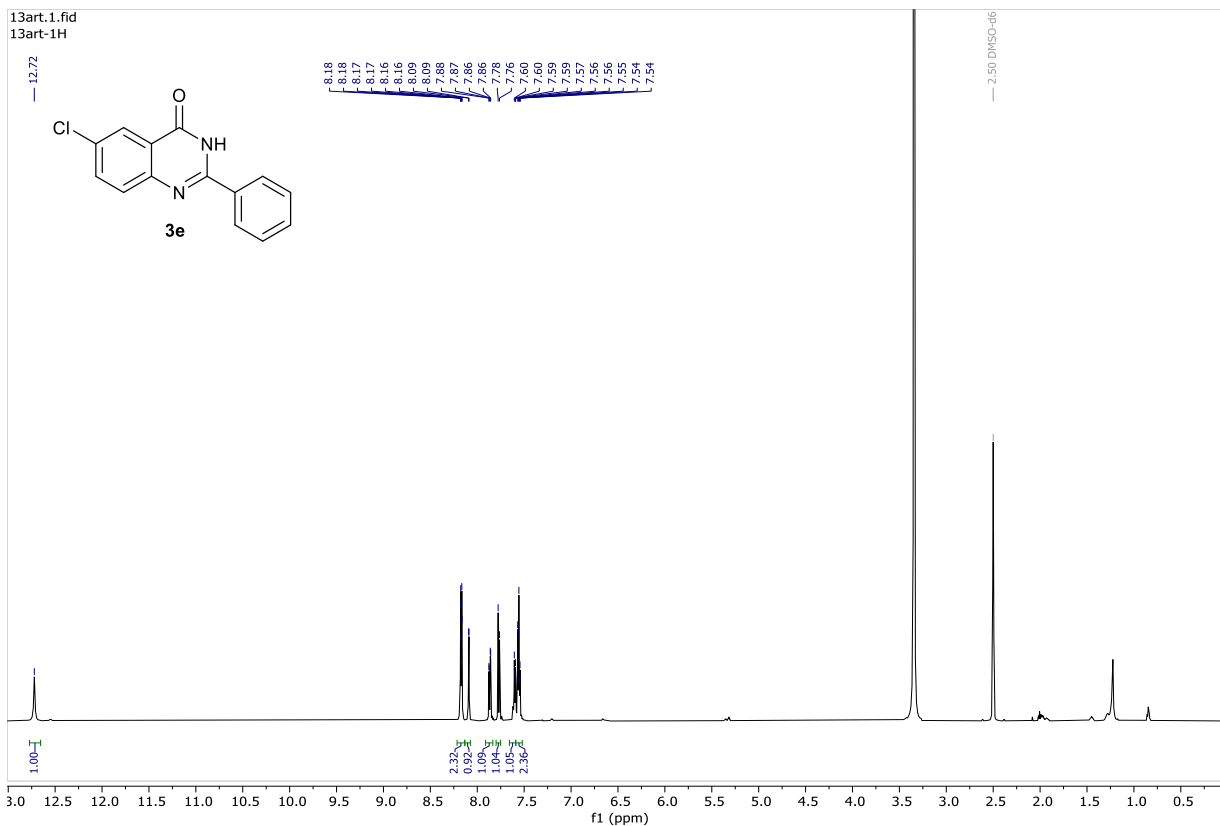


Figure 5 ^1H NMR and ^{13}C NMR spectra of **3e**

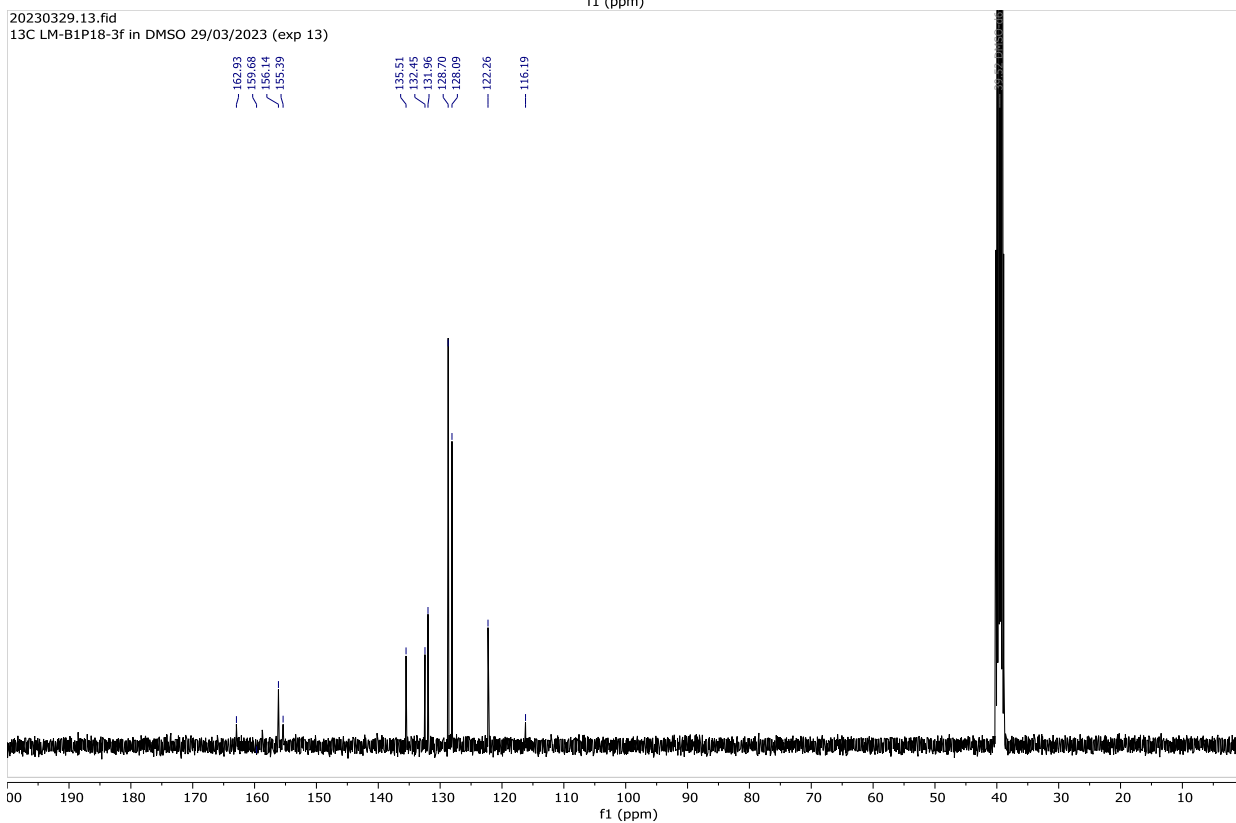
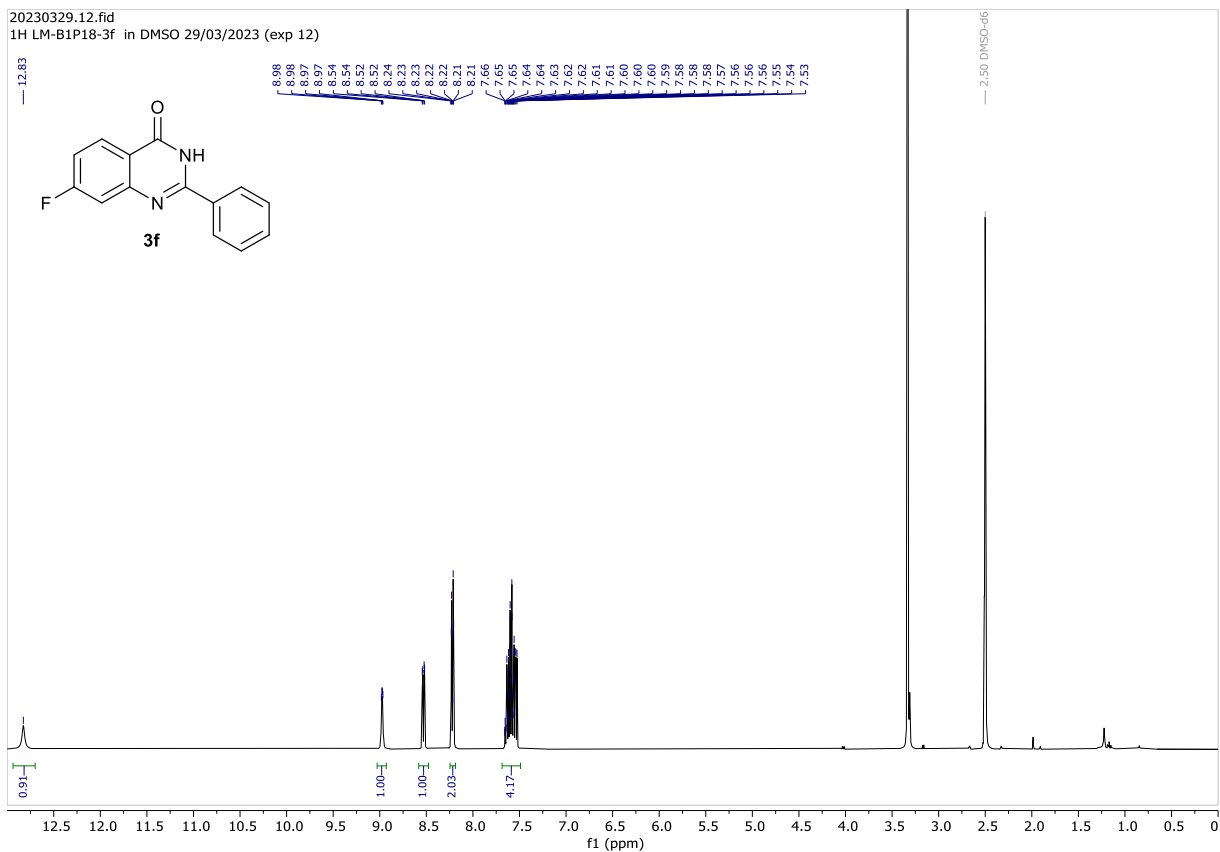


Figure 6 ^1H NMR and ^{13}C NMR spectra of **3f**

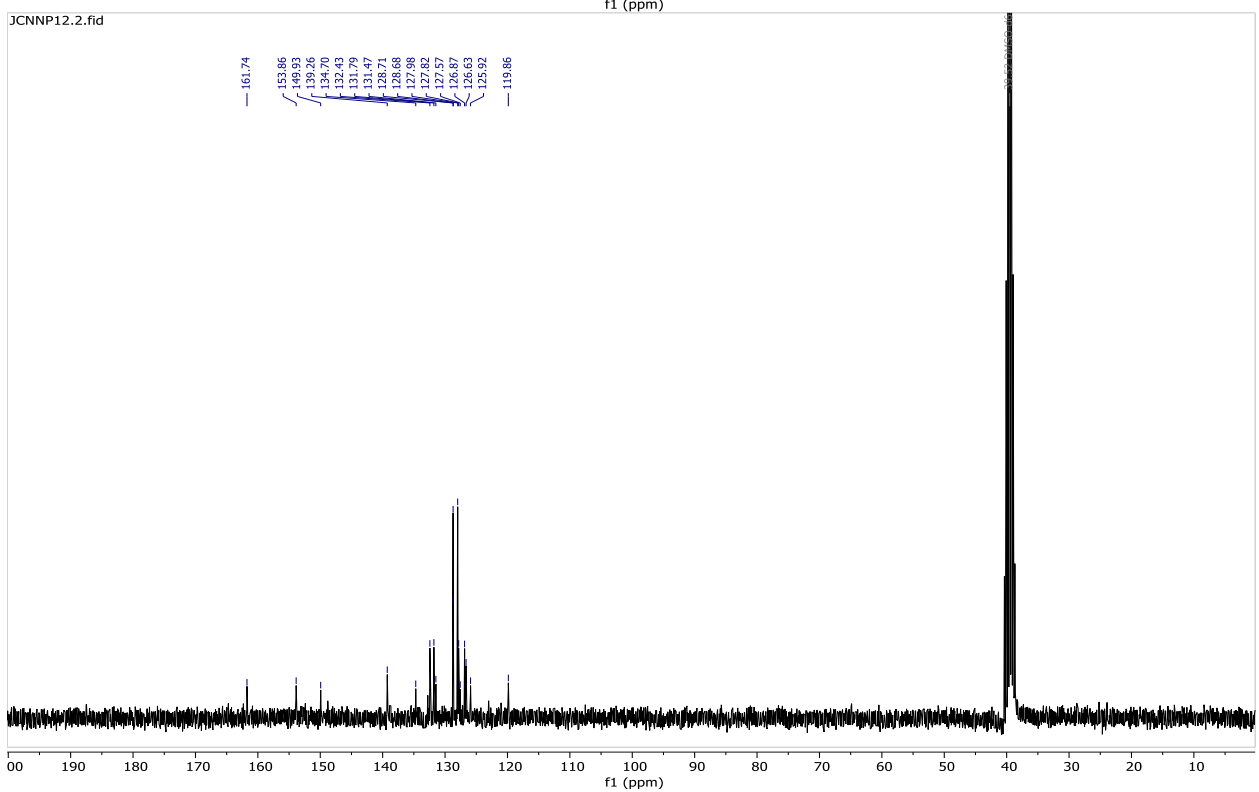
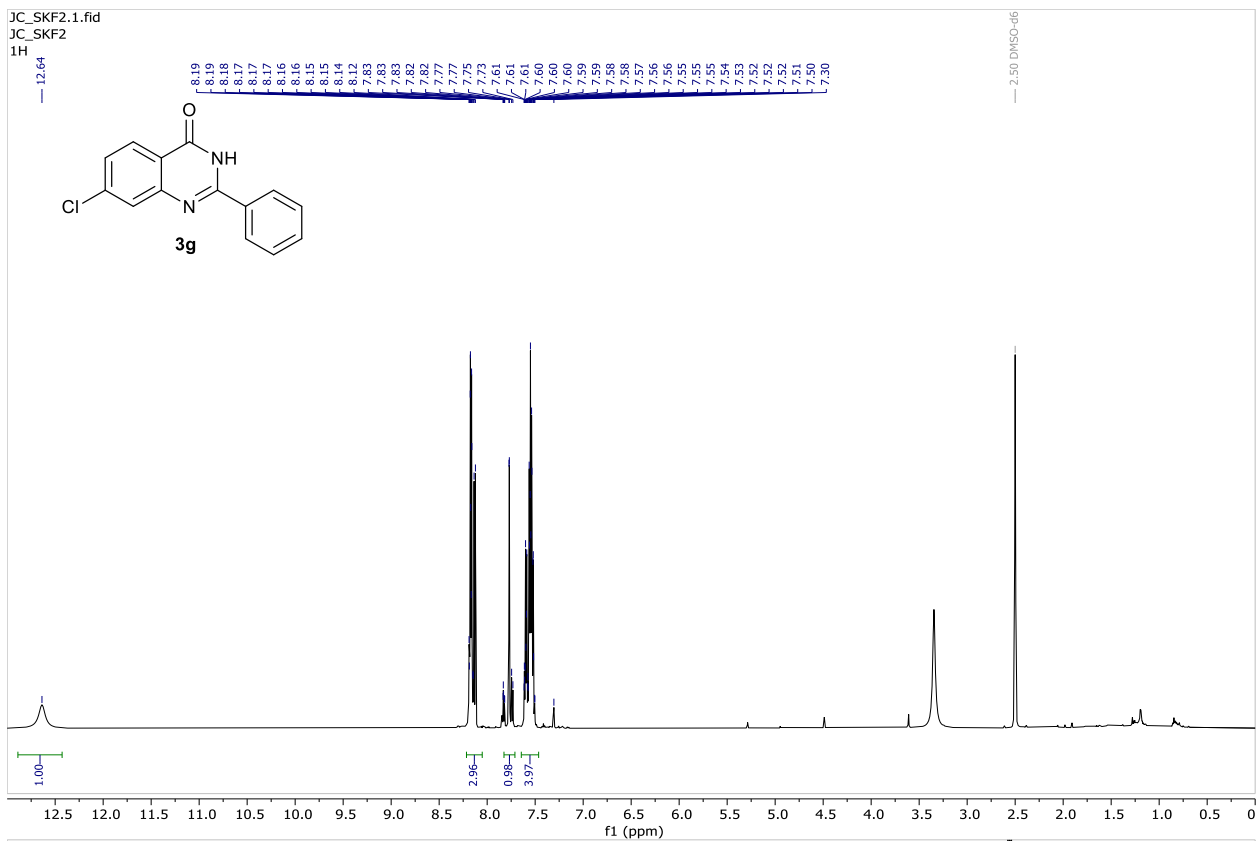


Figure 7 ^1H NMR and ^{13}C NMR spectra of **3g**

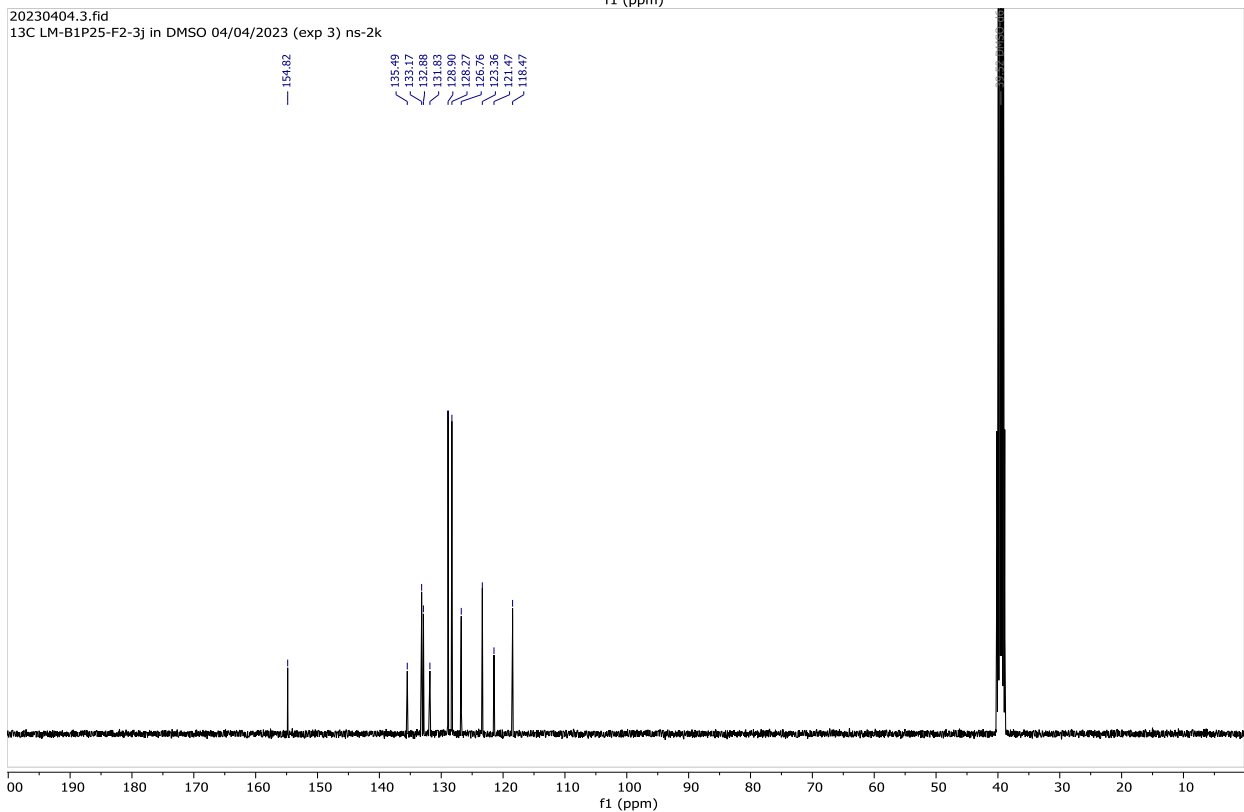
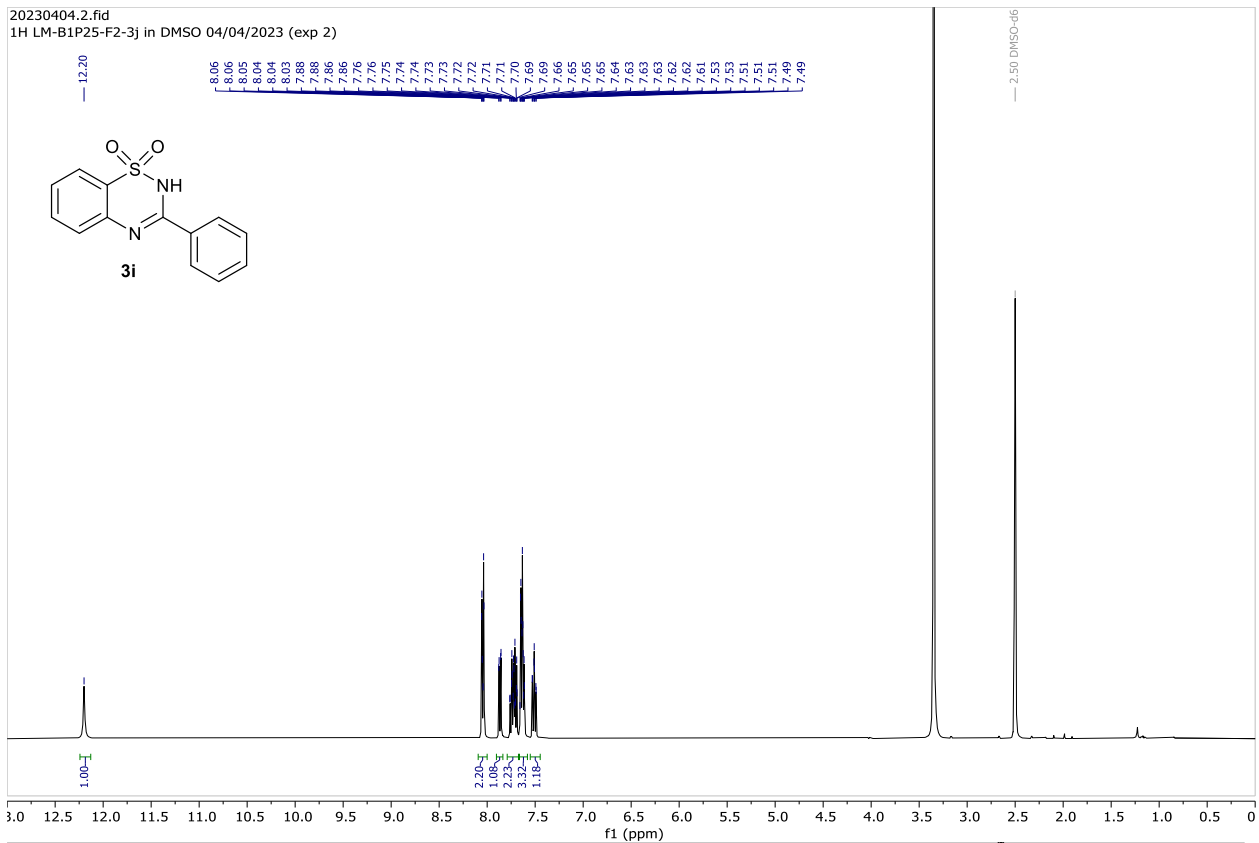


Figure 8 ^1H NMR and ^{13}C NMR spectra of **3i**

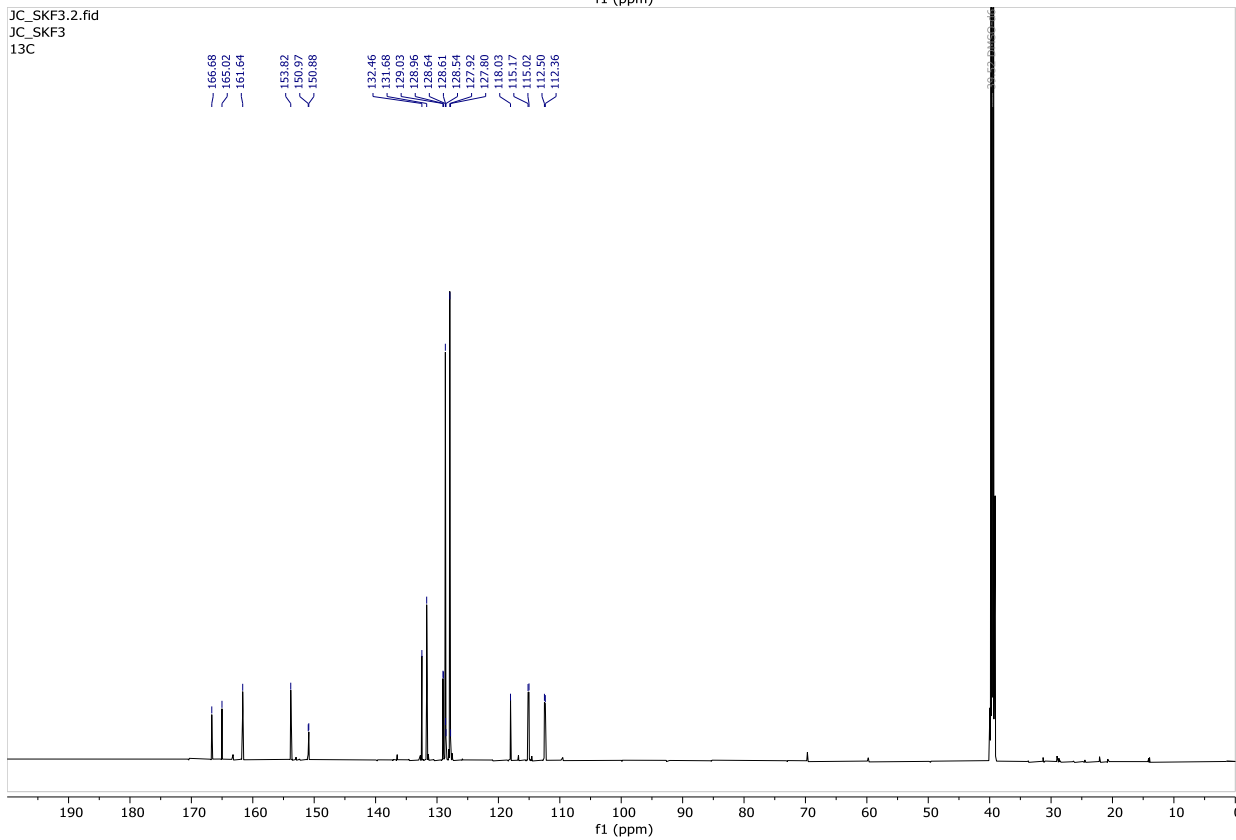
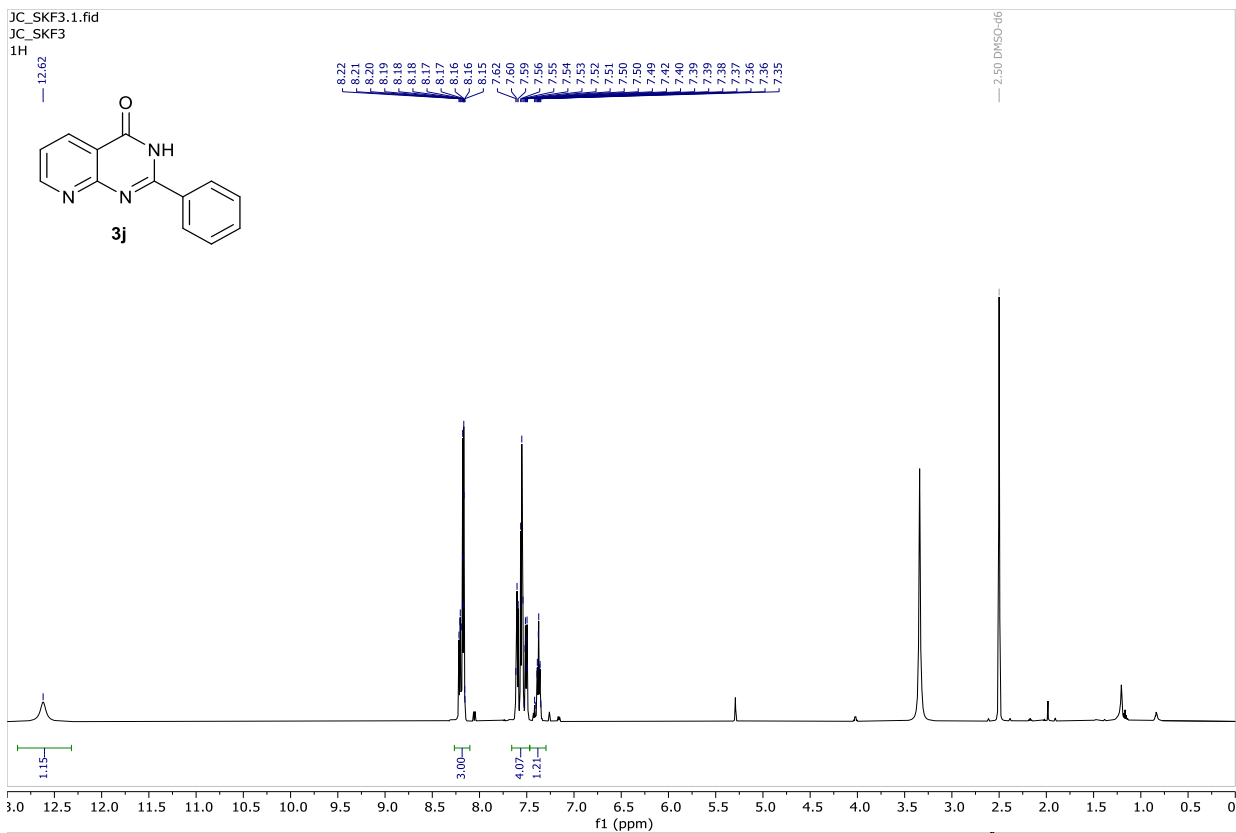


Figure 9 ^1H NMR and ^{13}C NMR spectra of **3j**

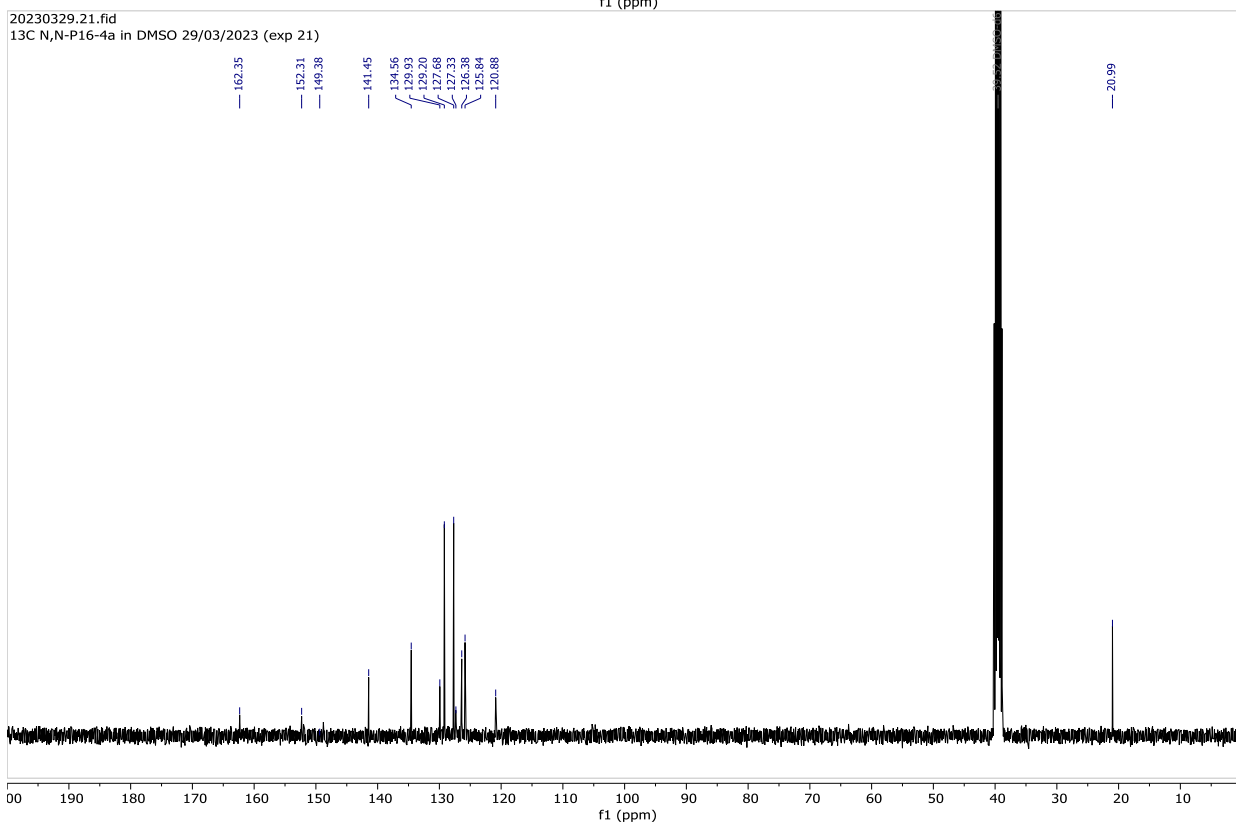
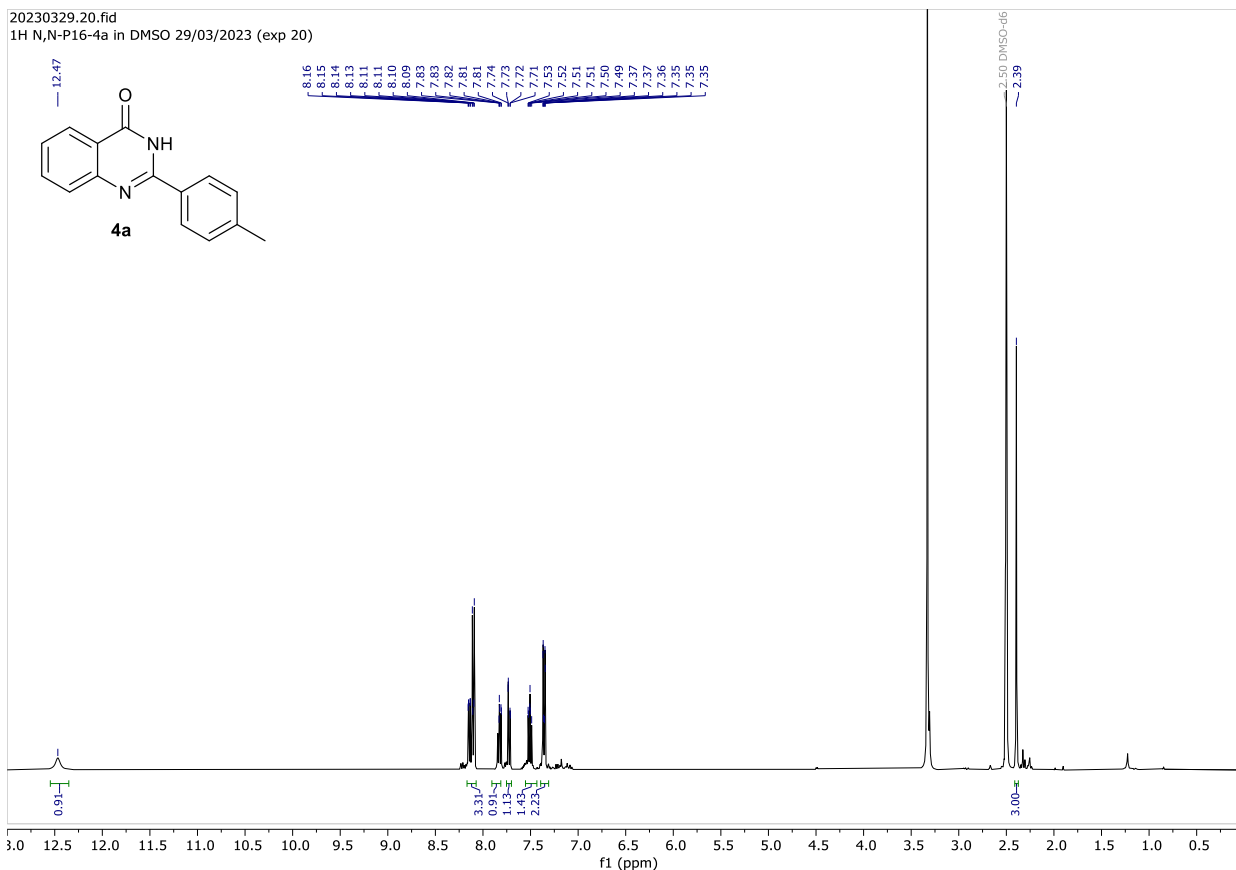


Figure 10 ¹H NMR and ¹³C NMR spectra of **4a**

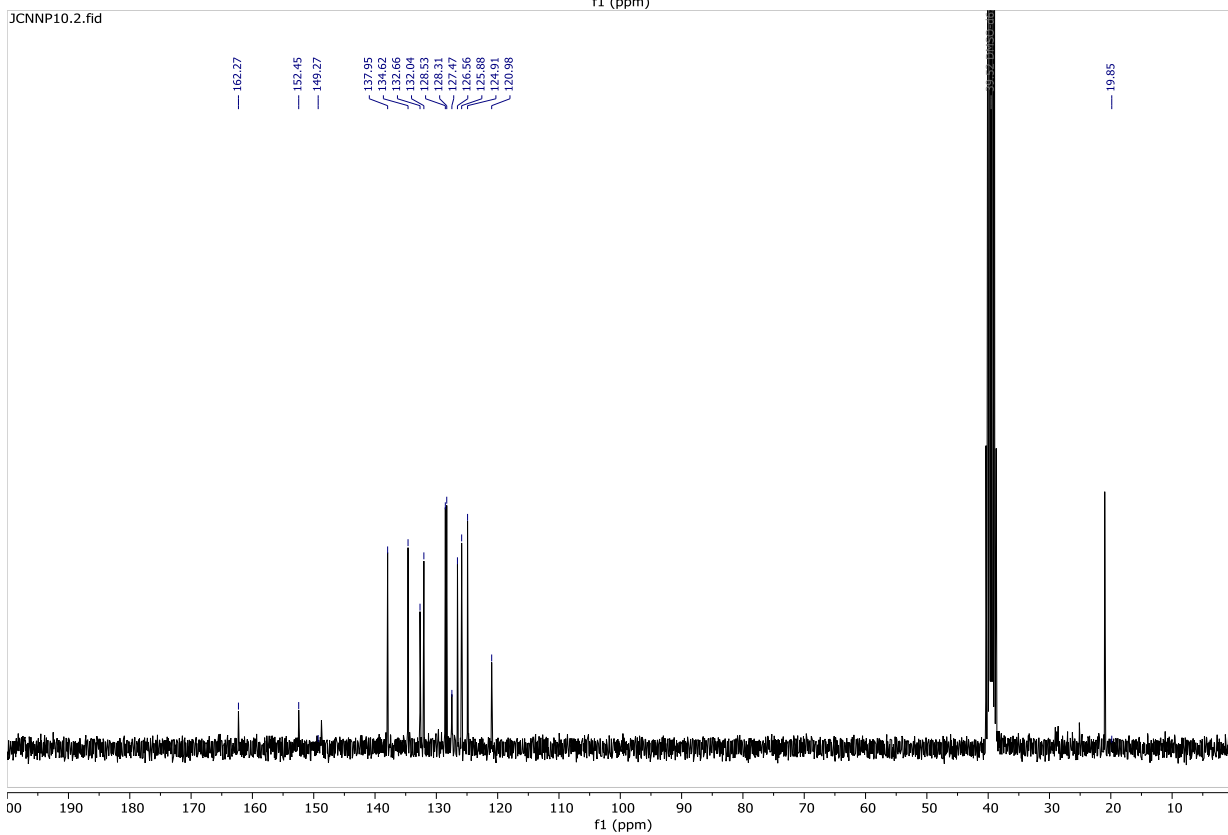
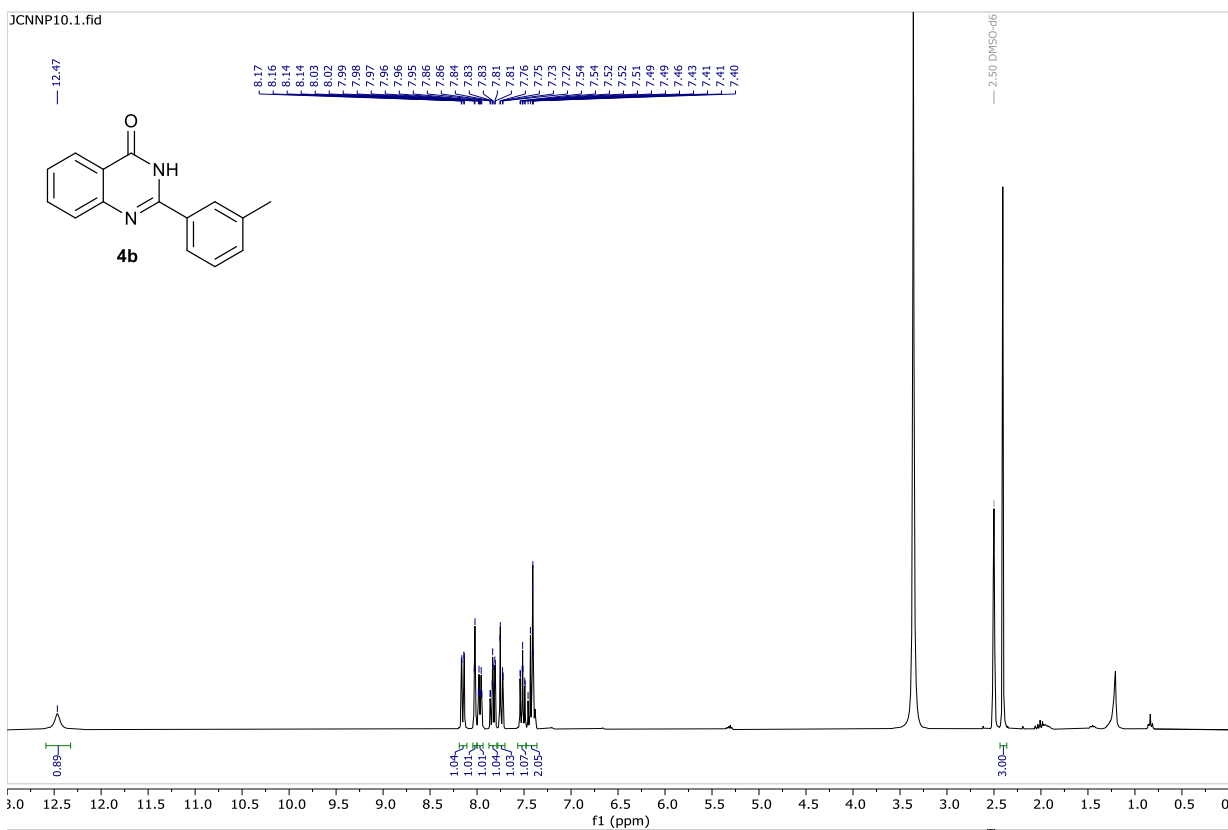


Figure 11 ^1H NMR and ^{13}C NMR spectra of **4b**

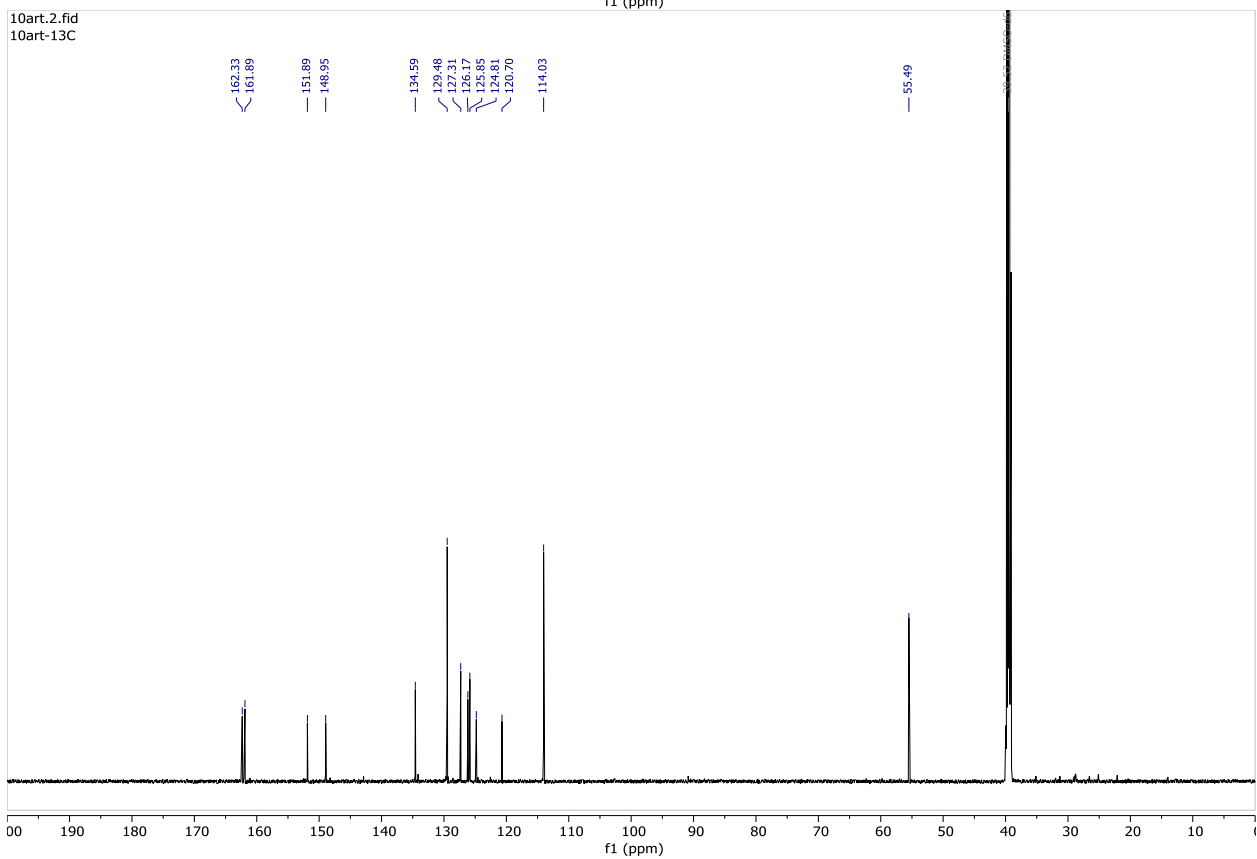
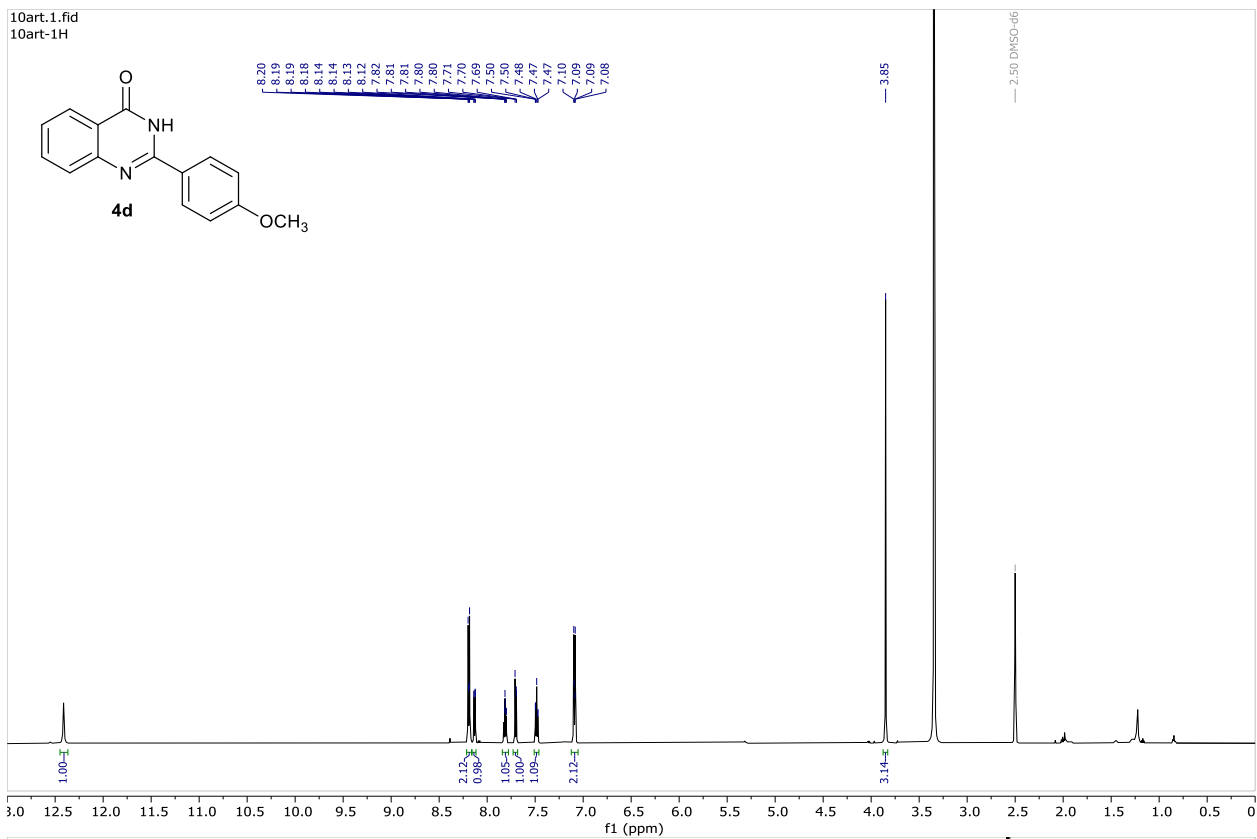


Figure 13 ^1H NMR and ^{13}C NMR spectra of **4d**

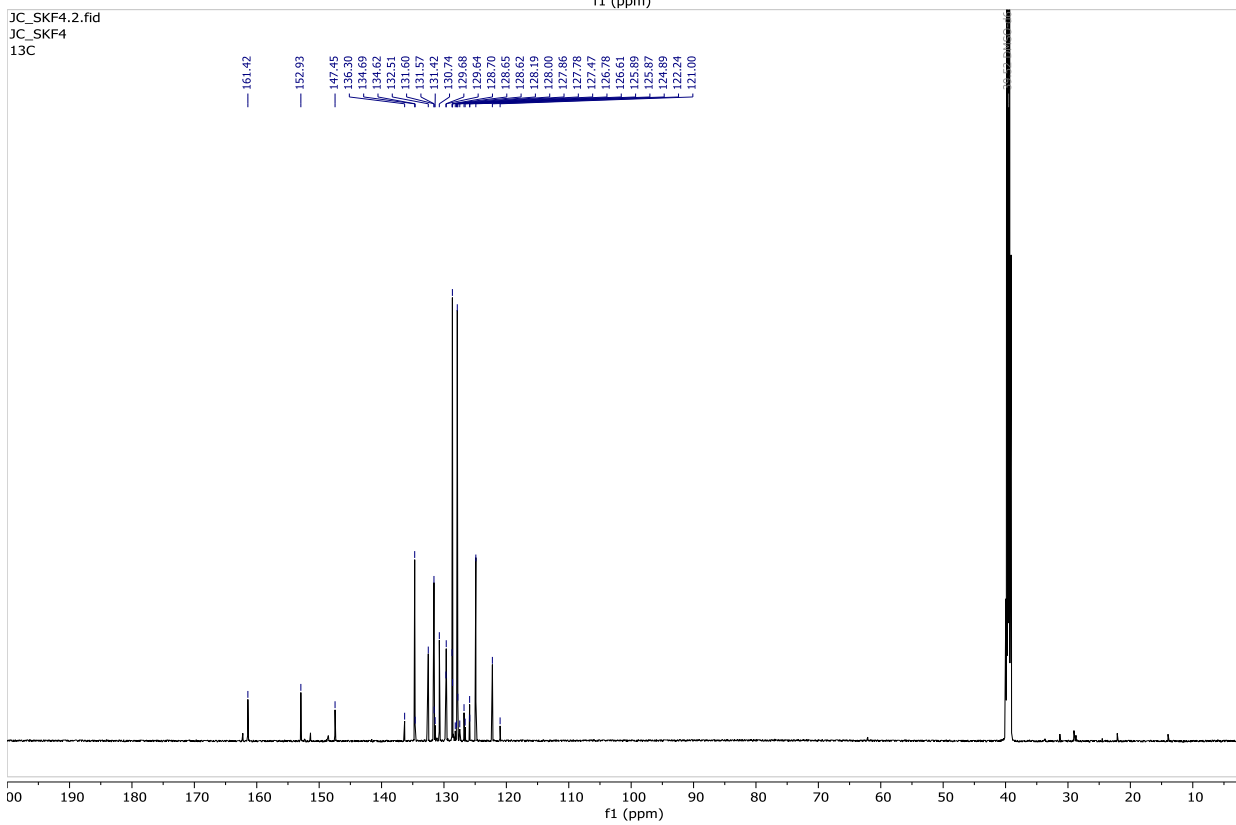
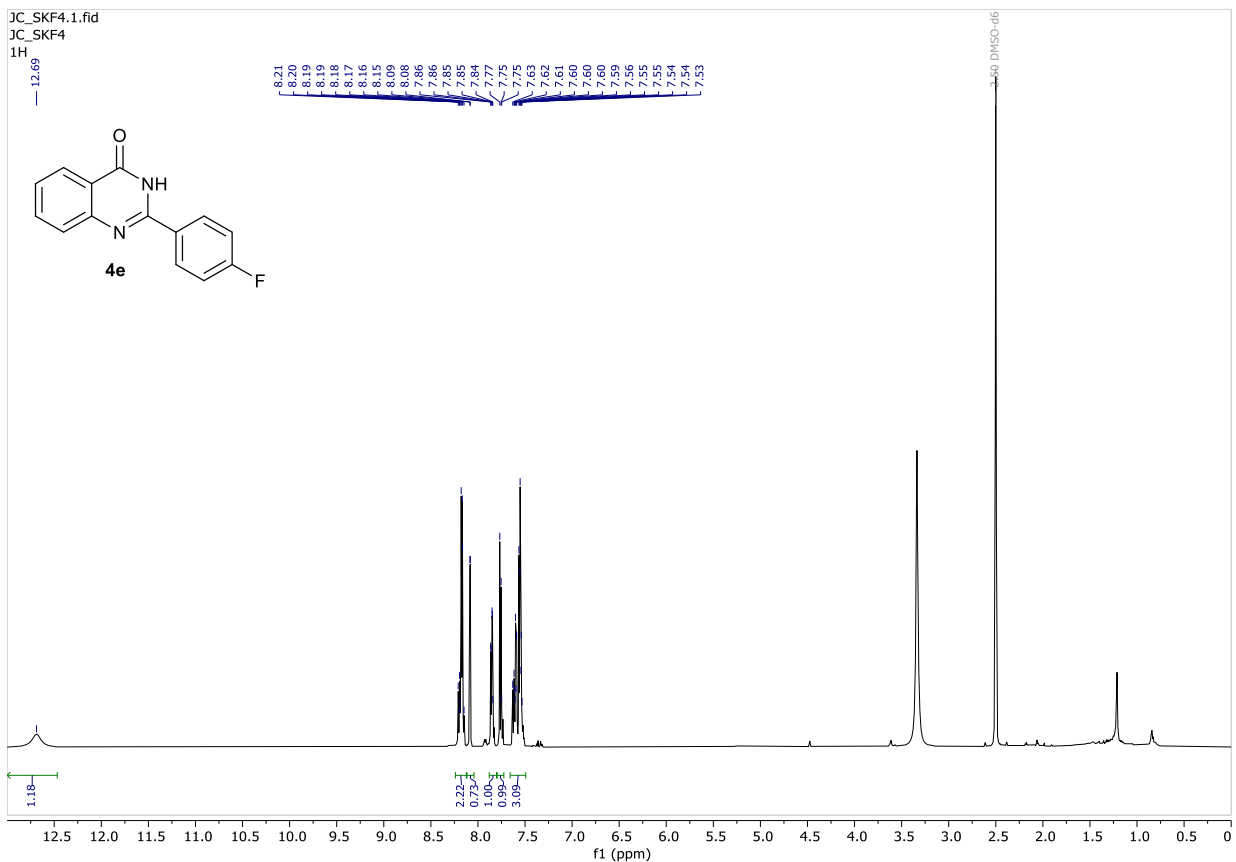


Figure 14 ^1H NMR and ^{13}C NMR spectra of **4e**

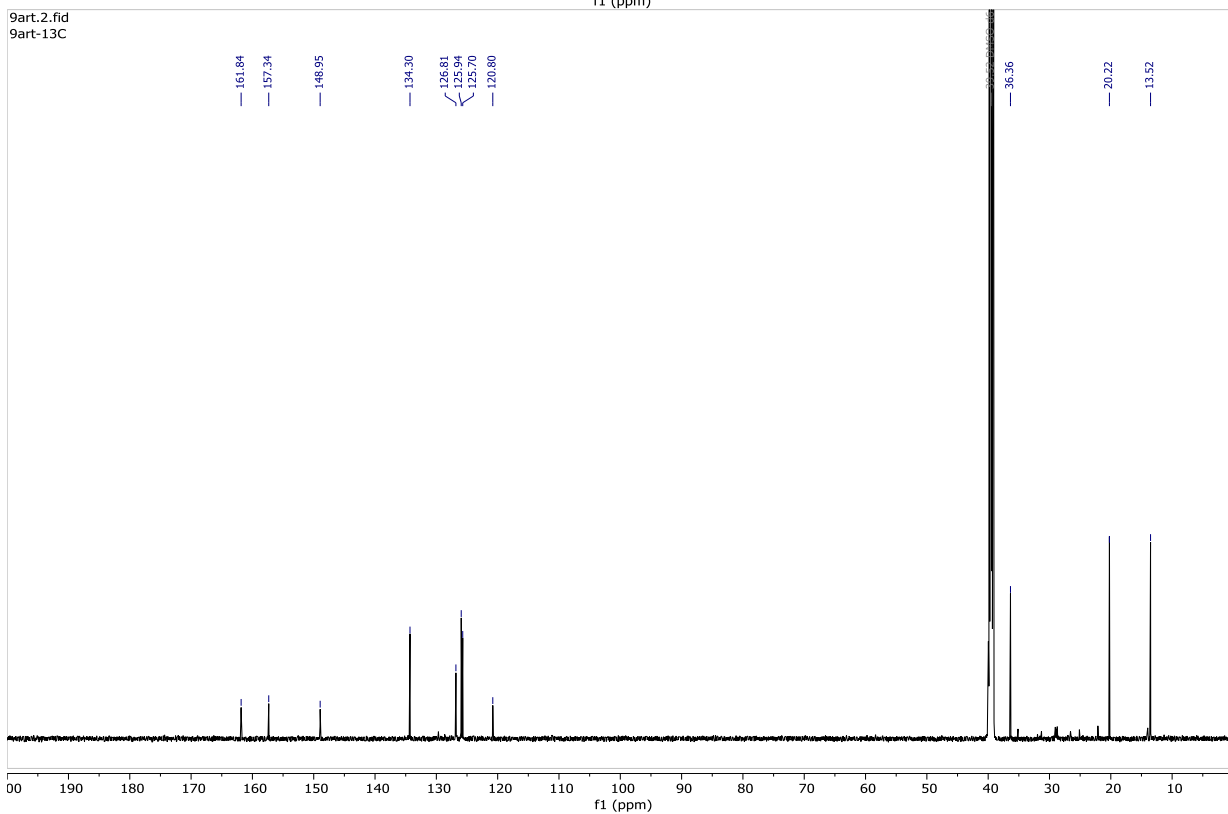
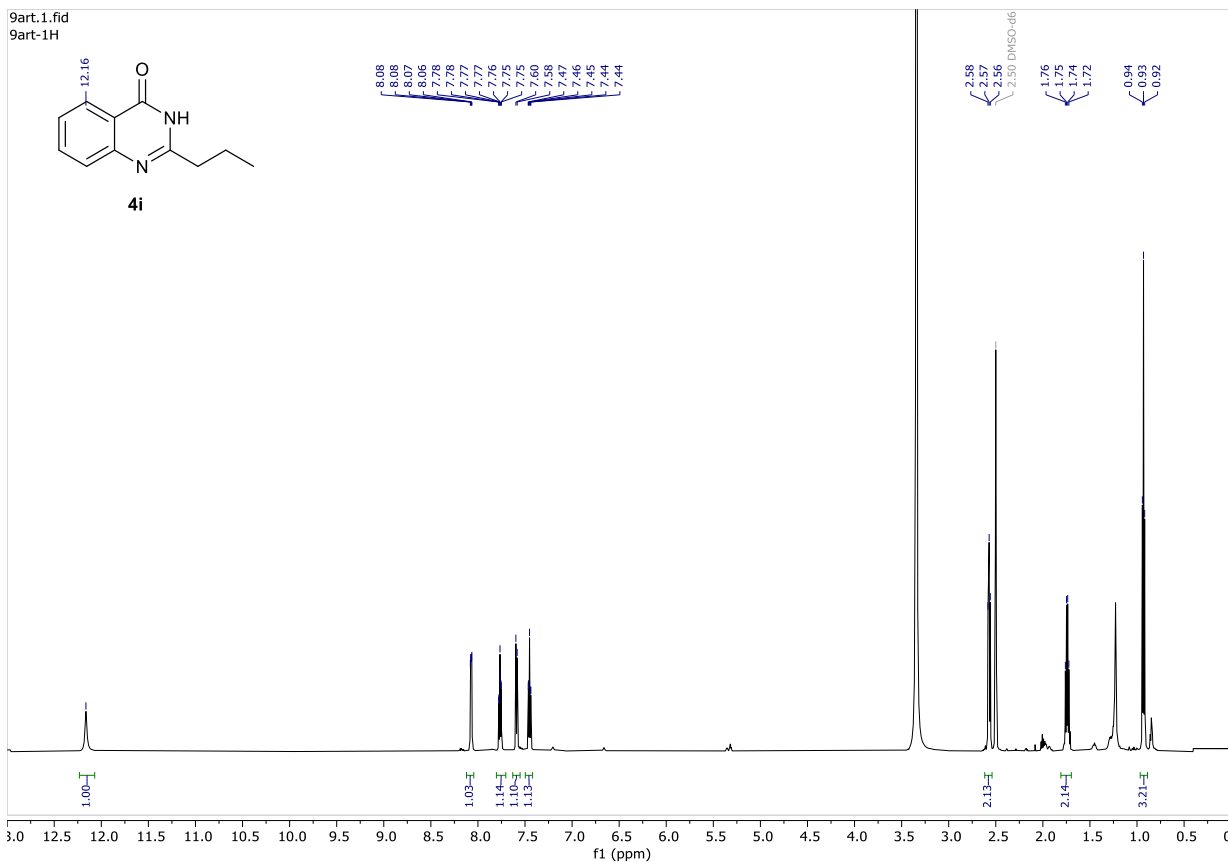
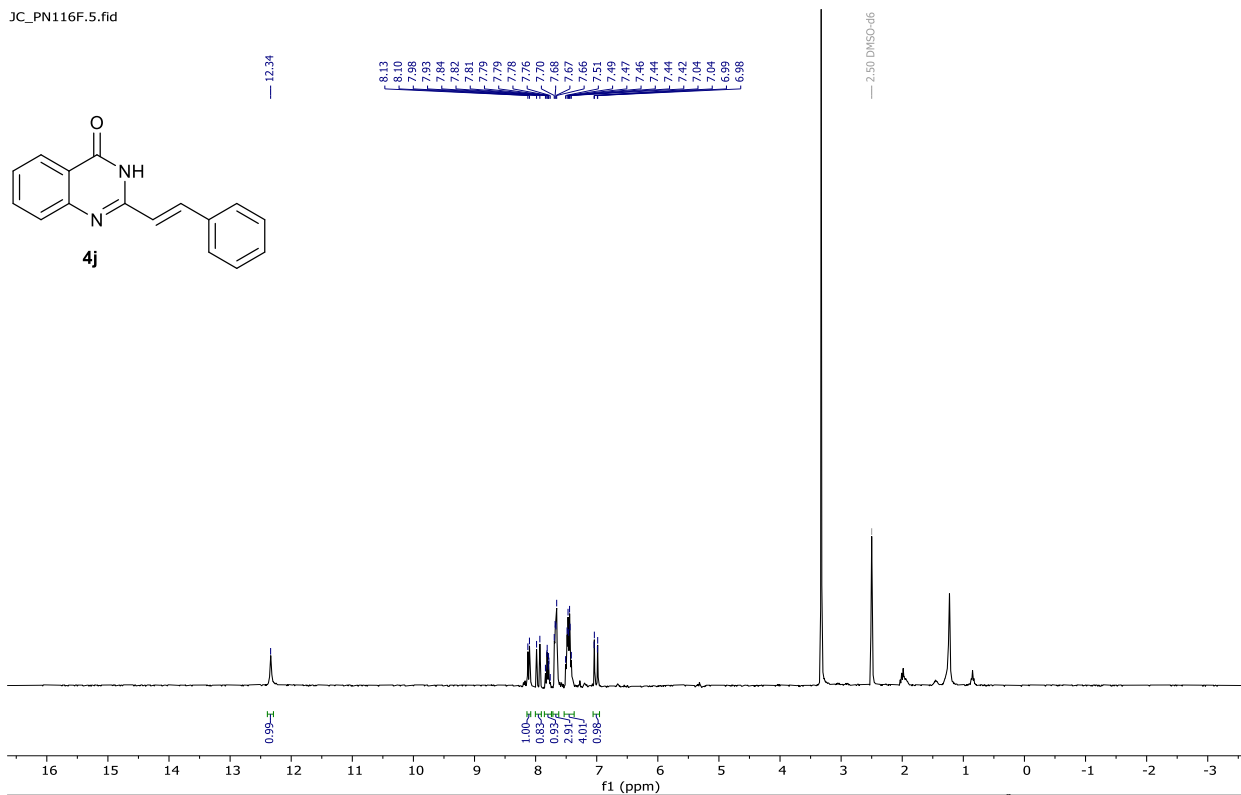
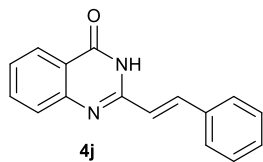


Figure 16 ^1H NMR and ^{13}C NMR spectra of **4i**

JC_PN116F.5.fid



20230616.6.fid
13C-Cinanyl in DMSO 16/06/2023 (exp 6) ns-2k

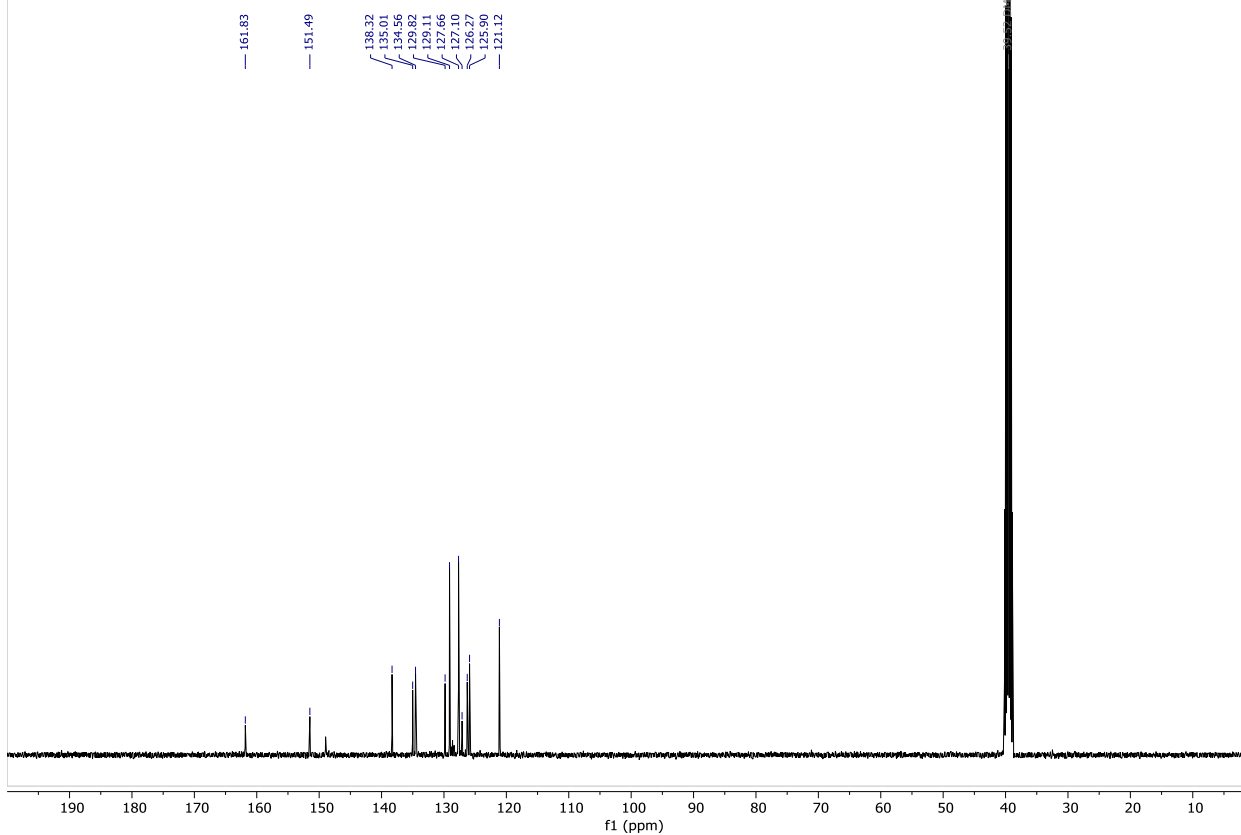


Figure 17 ^1H NMR and ^{13}C NMR spectra of **4j**

4. Acetylcholinesterase (AChE) activity

3 mM Ellman's reagent in Tris/HCl pH 8 125 μ l, 1.5 mM acetylthiocholine iodide in water 25 μ l and Tris/HCl pH 8 buffer 50 μ l and sample 25 μ l were mixed in 96-well plate. The mixture was added with AChE solution in 1 mg/ml BSA 25 μ l then immediately measured at 405 nm every 50 second for 5 min.

$$\% \text{inhibition} = \frac{(\text{Mean velocity}_{\text{control}} - \text{Mean velocity}_{\text{sample}}) \times 100}{\text{Mean velocity}_{\text{control}}}$$
