

Supplementary Material

Selective catalytic synthesis of new terpenic chlorides using NaDCC as an eco-friendly and highly stable FAC agent

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List of supplementary materials:

- ❖ Spectral data of terpenic chlorinated compounds.
- ❖ NMR and MS spectra of terpenic chlorinated compounds.

❖ Spectral data of terpenic chlorinated compounds

Carvone monochloride b

^1H NMR (300 MHz) d 6.70 (m, 1H, =CH), 5.20 (s, 1H, =CH₂), 4.99 (s, 1H, =CH₂), 4.03 (s, 2H, -CH₂Cl), 2.91 (m, 1H, CH), 2.53 (m, 2H, CH₂), 2.27 (m, 2H, CH₂), 1.75 (s, 3H, -CH₃).

^{13}C NMR (75 MHz) d 198.94 (C=O), 146.57 (=C-), 144.11 (=CH-), 135.6 (=C-), 115.16 (=CH₂), 46.95 (CH₂Cl), 43.02 (CH₂), 37.81 (CH), 31.38 (CH₂), 15.67 (CH₃).

MS (EI): m/z = 184.0176 [M]⁺.

Carvone dichloride c

^1H NMR (300 MHz) d 6.75 (m, 1H, =CH), 3.53 (m, 2H, -CH₂Cl), 2.61 (m, 2H, CH₂), 2.48 (m, 1H, CH), 2.32 (m, 2H, CH₂), 1.75 (s, 3H, CH₃), 1.24 (s, 3H, CH₃).

^{13}C NMR (75 MHz) d 199.60 (C=O), 144.56 (=CH-), 135.20 (=C-), 72.79 (-C-Cl), 52.29 (CH₂Cl), 41.57 (CH), 39.23 (CH₂), 26.26 (CH₂), 21.82 (CH₃), 15.55 (CH₃).

MS (EI): m/z = 220.0758 [M]⁺.

Carvone vinyl allyl dichloride d

^1H NMR (300 MHz) d 6.73 (m, 1H, =CH), 6.12 (s, 1H, =CHCl), 4.24 (m, 2H, -CH₂Cl), 3.01 (m, 1H, CH), 2.56 (m, 1H, CH₂), 2.37 (m, 1H, CH₂), 1.76 (s, 3H, CH₃), 1.17 (m, 2H, CH₂).

^{13}C NMR (75 MHz) d 197.97 (C=O), 143.70 (=CH-), 139.70 (=C-), 135.77 (=C-), 119.94 (=CHCl), 42.65 (-CH₂Cl), 39.57 (CH₂), 39.26 (CH), 31.15 (CH₂), 15.63 (CH₃).

MS (EI): m/z = 219.061 [M]⁺.

Carvone trichloride e

^1H NMR (300 MHz) d 6.55 (m, 1H, =CH₂), 3.44–3.97 (m, 4H, -CH₂Cl), 2.10–2.62 (m, 4H), 1.44–1.70 (m, 4H).

^{13}C NMR (75 MHz) d 197.44 (C=O), 143.04 (=CH-), 135.54 (=C-), 73.22 (-C-Cl), 50.80 (CH₂Cl), 49.89 (CH₂Cl), 41.04 (CH₂), 39.32 (-CH), 29.69 (CH₂), 15.54 (CH₃).

MS (EI): m/z = 254.048 [M]⁺.

Limonene oxide monochloride g

^1H NMR (300 MHz) d 4.96 (s, 1H, =CH₂), 4.95 (s, 1H, =CH₂), 4.02 (s, 2H, -CH₂Cl), 2.99 (m, 1H, -O-CH-), 2.36 (m, 1H, CH), 1.35–1.93 (m, 6H), 1.30 (s, 3H, -CH₃).

^{13}C NMR (75 MHz) d 148.76 (=C-), 113.65 (=CH₂), 58.97 (O-CH), 57.12 (O-C), 47.37 (CH₂Cl), 32.42 (CH), 30.70 (CH₂), 28.46 (CH₂), 24.66 (CH₂), 22.96 (CH₃).

MS (EI): m/z = 185.0668 [M]⁺.

Limonene oxide dichloride h

^1H NMR (300 MHz) d 3.55 (s, 2H, -CH₂Cl), 3.02 (m, 1H, -O-CH-), 2.05 (m, 1H, CH), 2.01–1.50 (m, 6H), 1.33 (s, 3H, -CH₃), 1.13 (s, 3H, -CH₃).

^{13}C NMR (75 MHz) d 73.37 (-C-Cl), 58.78 (O-CH), 57.66 (O-C), 53.23 (CH₂Cl), 39.78 (CH), 30.58 (CH₂), 24.89 (CH₂), 22.85 (CH₃), 22.96 (CH₃), 20.45 (CH₂).

MS (EI): m/z = 220.9909 [M]⁺.

Nootkatone monochloride i

^1H NMR (300 MHz) d 5.66 (m, 1H, =CH), 5.09 (s, 1H, =CH₂), 4.90 (s, 1H, =CH₂), 4.01 (s, 2H, -CH₂Cl), 2.47 (m, 2H, CH₂), 2.29 (m, 1H, CH), 2.15 (m, 2H, CH₂), 1.91 (m, 2H, CH₂), 1.27 (m, 1H, CH), 1.04–1.97 (m, 2H, CH₂), 1.04 (s, 3H, -CH₃), 0.87 (d, 3H, -CH₃).

^{13}C NMR (75 MHz) d 199.08 (C=O), 169.58 (=C-), 148.67 (=C-), 124.74 (=CH-), 113.80 (=CH₂), 47.70 (CH₂Cl), 44.17 (CH₂), 41.95 (CH₂), 40.30 (CH), 39.35 (-C-), 35.72 (CH), 32.85 (CH₂), 31.91 (CH₂), 16.67 (CH₃), 14.87 (CH₃).
MS (EI): m/z = 252.107 [M]⁺.

Nootkatone vinyl allyl dichloride k

^1H NMR (300 MHz) d 6.07 (s, 1H, =CHCl), 5.77 (s, 1H, =CH), 4.26 (m, 2H, -CH₂Cl), 2.60 (m, 2H, CH₂), 2.30 (m, 2H, CH₂), 2.12 (m, 1H, CH), 2.00 (m, 2H, CH₂) 1.41 (m, 1H, CH), 1.24 (m, 2H, CH₂), 1.13 (s, 3H, CH₃), 0.96 (m, 3H, CH₃).
 ^{13}C NMR (75 MHz) d 199.25 (C=O), 168.76 (=C-), 141.67 (=C-), 125.10 (=CH-), 118.79 (=CHCl), 43.87 (CH₂Cl), 41.97 (CH₂), 40.38 (CH₂), 40.32 (-C-), 39.41 (-CH), 37.42 (-CH), 32.68 (CH₂), 31.59 (CH₂), 16.74 (CH₃), 14.89 (CH₃).
MS (EI): m/z = 286.088 [M]⁺.

Nootkatone dichloride l

^1H NMR (300 MHz) d 5.76 (m, 1H, =CH), 3.57 (m, 1H, CH₂Cl), 3.64 (m, 1H, CH₂Cl), 2.40 (m, 1H, CH), 2.23–2.50 (m, 2H, CH₂), 2.06 (m, 2H, CH₂), 1.99–2.29 (m, 2H, CH₂), 1.87–2.12 (m, 2H, CH₂), 1.25 (m, 1H, CH), 1.19 (s, 3H, -CH₃), 1.10 (m, 3H, -CH₃), 0.98 (m, 3H, -CH₃).
 ^{13}C NMR (75 MHz) d 199.50 (C=O), 170.08 (=C-), 124.62 (=CH-), 73.33 (-CCl), 53.47 (CH₂Cl), 42.04 (CH₂), 40.53 (CH), 39.96 (CH), 39.68 (CH), 32.72 (CH₂), 27.70 (CH₂), 26.67 (CH₂), 21.22 (CH₃), 16.75 (CH₃), 14.96 (CH₃).
MS (EI): m/z = 288.097 [M]⁺.

Pulegone monochloride n

^1H NMR (300 MHz) d 5.15 (s, 1H, =CH₂), 5.08 (s, 1H, =CH₂), 2.51–2.82 (m, 2H, CH₂), 1.99–2.31 (m, 1H, CH), 1.88–2.19 (m, 2H, CH₂), 1.79 (m, 3H, CH₃), 1.10–1.52 (m, 2H, CH₂), 0.97 (m, 3H, CH₃).
 ^{13}C NMR (75 MHz) d 203.90 (C=O), 143.44 (=C-), 115.29 (=CH₂), 76.09 (-C-Cl), 45.56 (CH₂), 37.97 (CH₂), 34.42 (CH), 29.64 (CH₂), 21.50 (CH₃), 20.61 (CH₃).
MS (EI): m/z = 187.099 [M]⁺.

Pulegone monochloride n'

^1H NMR (300 MHz) d 5.20 (s, 1H, =CH₂), 5.14 (s, 1H, =CH₂), 2.35–2.80 (m, 2H, CH₂), 2.30 (m, 2H, CH₂), 1.96 (m, 1H, CH), 1.90 (m, 3H, -CH₃), 1.84 (m, 2H, CH₂), 1.06 (d, 3H, -CH₃).
 ^{13}C NMR (75 MHz) d 203.37 (C=O), 141.35 (=C-), 115.22 (=CH₂), 77.96 (-C-Cl), 46.76 (CH₂), 38.46 (CH₂), 33.87 (CH), 30.53 (CH₂), 20.58 (CH₃), 18.02 (CH₃).
MS (EI): m/z = 187.093 [M]⁺.

Pulegone vinyl allyl dichloride o

^1H NMR (300 MHz) d 5.71 (s, 1H, =CH₂), 5.57 (s, 1H, =CH₂), 4.22 (m, 2H, -CH₂Cl), 2.84 (m, 2H, CH₂), 2.35 (m, 2H, CH₂), 1.83–2.00 (m, 3H), 1.06 (m, 3H, CH₃).
 ^{13}C NMR (75 MHz) d 203.53 (C=O), 144.26 (=C-), 119.82 (=CH₂), 74.28(-C-Cl), 45.23(CH₂), 44.42(-CH₂Cl), 38.29 (CH₂), 34.57 (-CH), 29.43 (CH₂), 21.81 (CH₃).
MS (EI): m/z = 221.055 [M]⁺.

Pulegone vinyl allyl dichloride o'

^1H NMR (300 MHz) d 5.72 (s, 1H, =CH₂), 5.53 (s, 1H, =CH₂), 4.05–4.25 (m, 2H, -CH₂Cl), 2.68–2.89 (m, 2H, CH₂), 2.12–2.44 (m, 2H, CH₂), 1.94 (m, 1H, CH), 1.19–1.67 (m, 2H, CH₂), 0.91–1.04 (m, 3H, CH₃).
 ^{13}C NMR (75 MHz) d 202.49 (C=O), 142.86 (=C-), 119.97 (=CH₂), 77.37 (-C-Cl), 47.54 (CH₂), 43.48 (-CH₂Cl), 39.03 (CH₂), 34.48 (-CH), 31.08 (CH₂), 21.10 (CH₃).

MS (EI): $m/z = 221.054 [M]^+$.

Perillyl aldehyde monochloride g

^1H NMR (300 MHz) d 9.43 (s, 1H, HC=O), 6.83 (m, 1H, =CH), 5.23 (s, 1H, =CH₂), 5.02 (s, 1H, =CH₂), 4.09 (s, 2H, -CH₂Cl), 2.02–2.53 (m, 2H, CH₂), 2.25 (m, 1H, CH), 1.42–1.72 (m, 2H, CH₂), 0.87–1.24 (m, 2H, -CH₂).

^{13}C NMR (75 MHz) d 193.74 (HC=O), 149.82 (=CH-), 148.25 (=C-), 141.15 (=C-), 114.20 (=CH₂), 47.56 (CH₂Cl), 36.21 (CH), 32.13 (CH₂), 26.51 (CH₂), 21.46 (CH₂).

MS (EI): $m/z = 184.0441 [M]^+$.

Perillyl aldehyde vinyl allyl dichloride r

^1H NMR (300 MHz) d 9.45 (s, 1H, HC=O), 6.82 (m, 1H, =CH), 6.12 (m, 1H, =CHCl), 4.29 (s, 2H, -CH₂Cl), 1.58–2.59 (m, 2H, CH₂), 1.99 (m, 1H, CH), 1.26 (m, 2H, -CH₂), 0.87–2.30 (m, 2H, CH₂).

^{13}C NMR (75 MHz) d 193.51 (HC=O), 148.79 (=CH-), 141.26 (=C-), 141.10 (=C-), 119.00 (=CHCl), 40.11 (CH₂Cl), 37.86 (CH), 31.81 (CH₂), 26.28 (CH₂), 21.40 (CH₂).

MS (EI): $m/z = 218.0015 [M]^+$.

Limona ketone Dichloride u

^1H NMR (300 MHz) d 4.00 (m, 1H, -CHCl), 2.78 (m, 1H, CH), 1.90–2.30 (m, 2H, CH₂), 2.12 (m, 3H), 1.70–1.84 (m, 2H, CH₂), 1.53–1.70 (m, 2H, CH₂), 1.30 (m, 3H, CH₃).

^{13}C NMR (75 MHz) d 210.96 (C=O), 71.54 (-C-Cl), 64.75(-CHCl), 44.65 (CH), 32.64 (CH₂), 31.81 (CH₂), 28.14 (CH₃), 27.29 (CH₃), 23.28 (CH₂).

MS (EI): $m/z = 207.066 [M]^+$.

Limona ketone epoxide y

^1H NMR (300 MHz) 2.97 (m, 1H), 2.22 (m, 1H), 2.08 (s, 3H, CH₃), 2.06–1.94 (m, 3H), 1.37–1.72 (m, 3H), 1.28 (s, 3H, CH₃).

^{13}C NMR (75 MHz) d 210.46 (C=O), 58.22 (O-CH), 57.51(O-C), 46.38 (CH), 29.71 (CH₂), 27.87 (CH₂), 26.03 (CH₃), 23.04 (CH₂), 21.78 (CH₃).

MS (EI): $m/z = 154.0931 [M]^+$.

❖ NMR and MS spectra of terpenic chlorinated compounds

➤ Derivatives of carvone **a**

• Carvone monochloride **b**

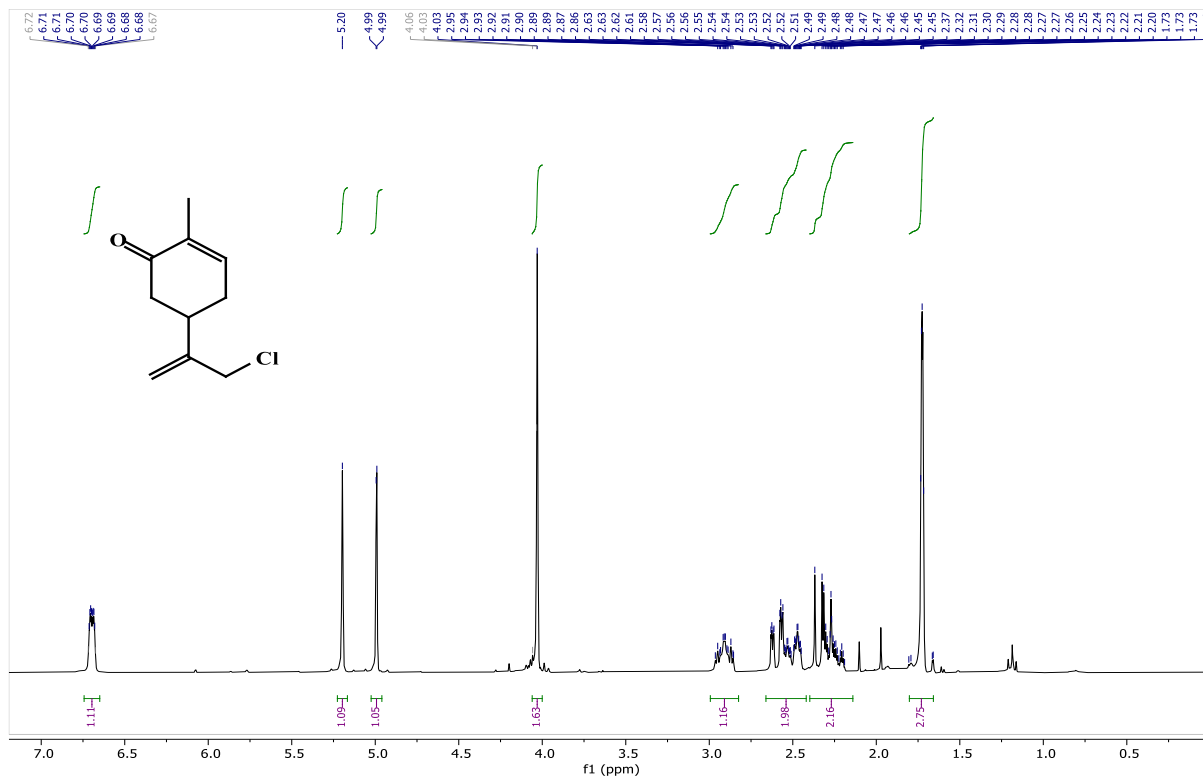


Figure S1: ¹H NMR spectrum of the carvone monochloride **b**.

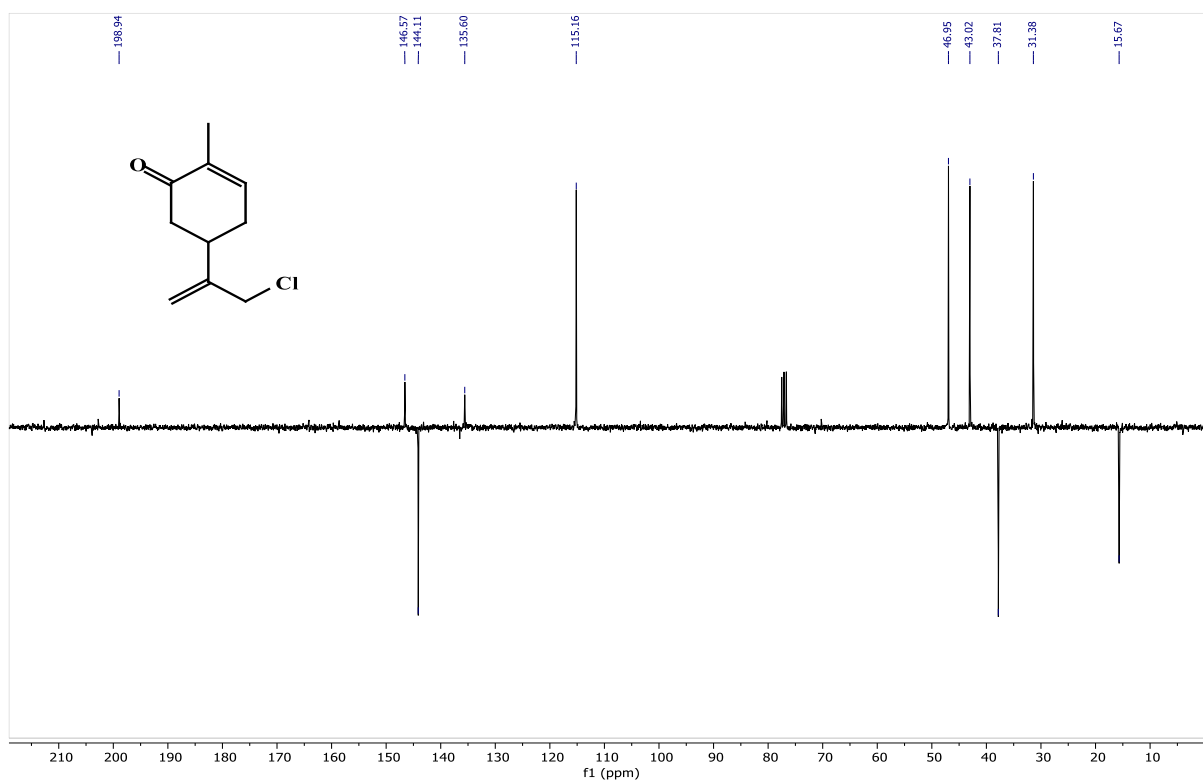


Figure S2: APT spectrum of the carvone monochloride **b**.

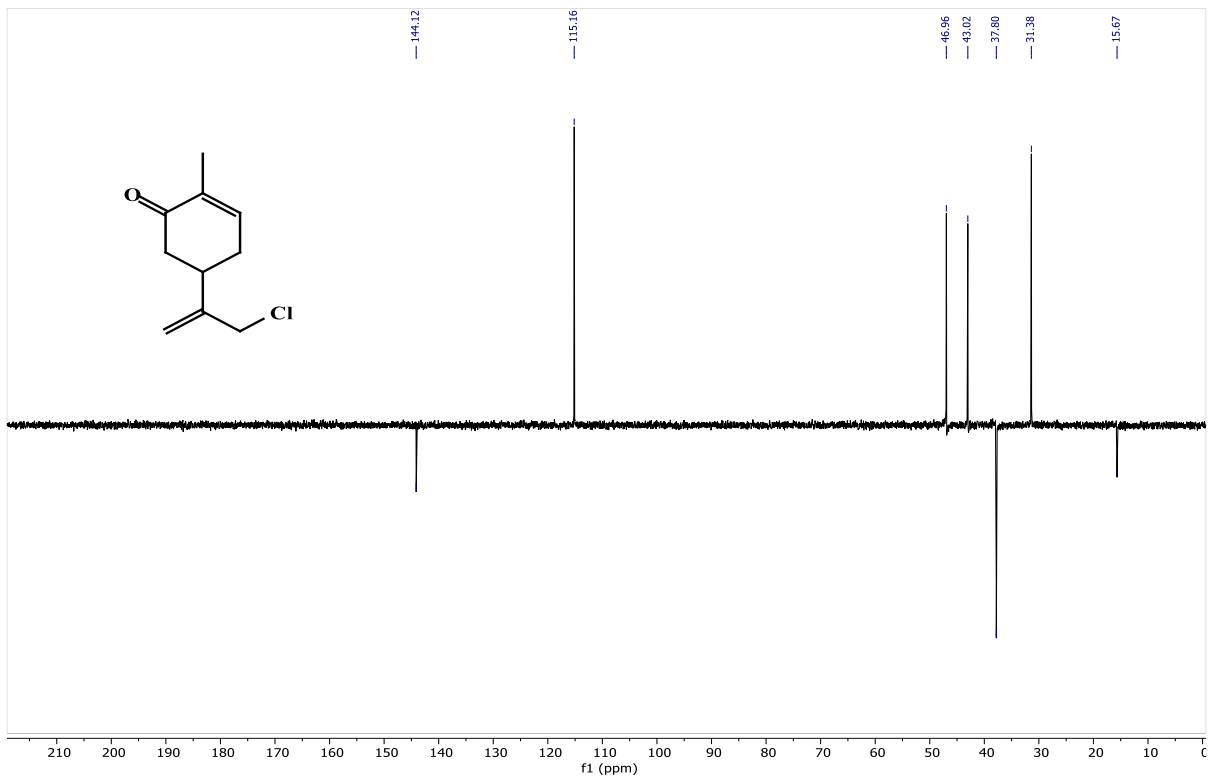


Figure S3: DEPT 135 spectrum of the carvone monochloride **b**.

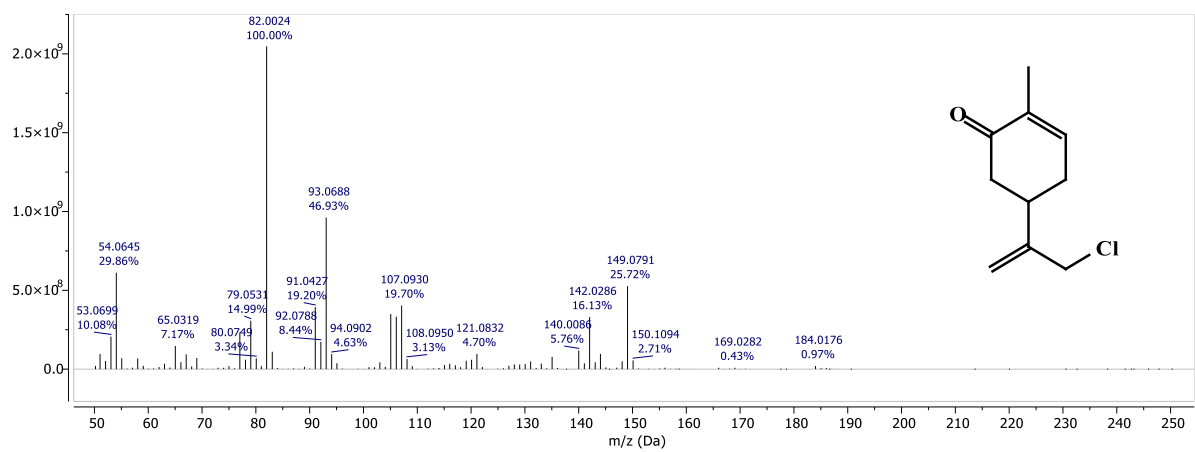


Figure S4: MS spectrum of the carvone monochloride **b**.

• Carvone dichloride **c**

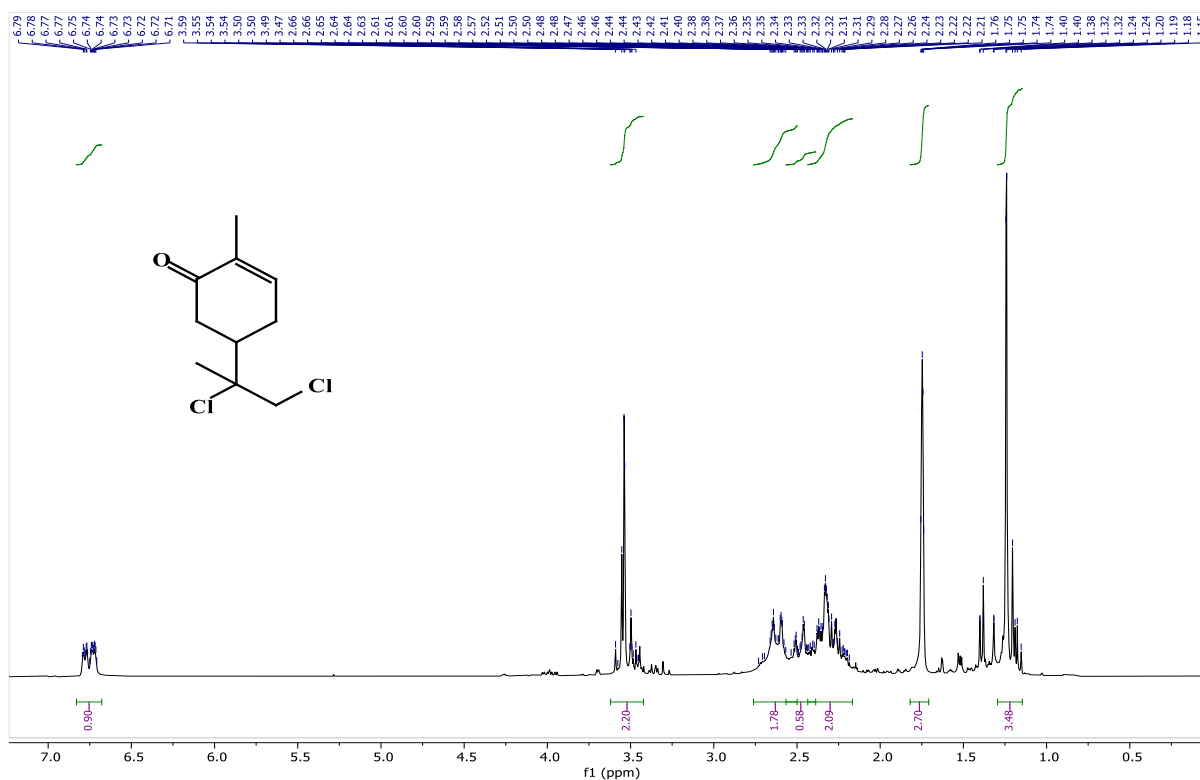


Figure S5: ¹H NMR spectrum of the carvone dichloride **c**.

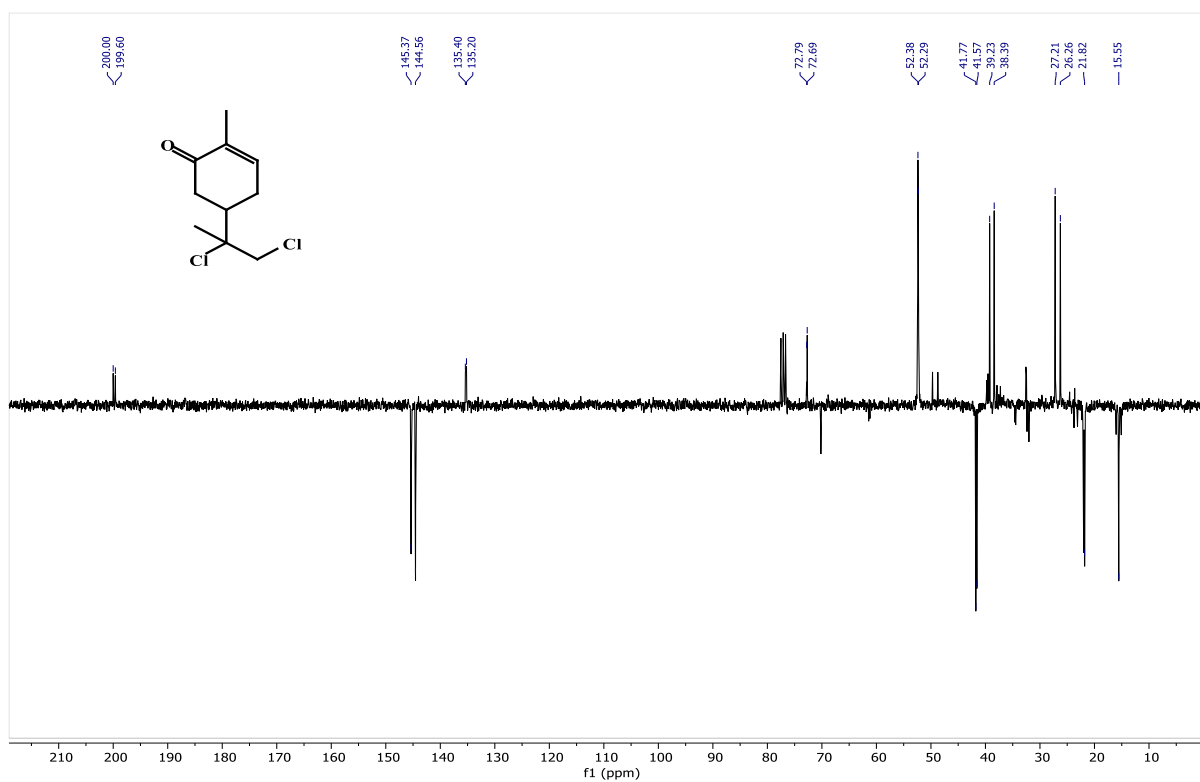


Figure S6: APT spectrum of the carvone dichloride **c**.

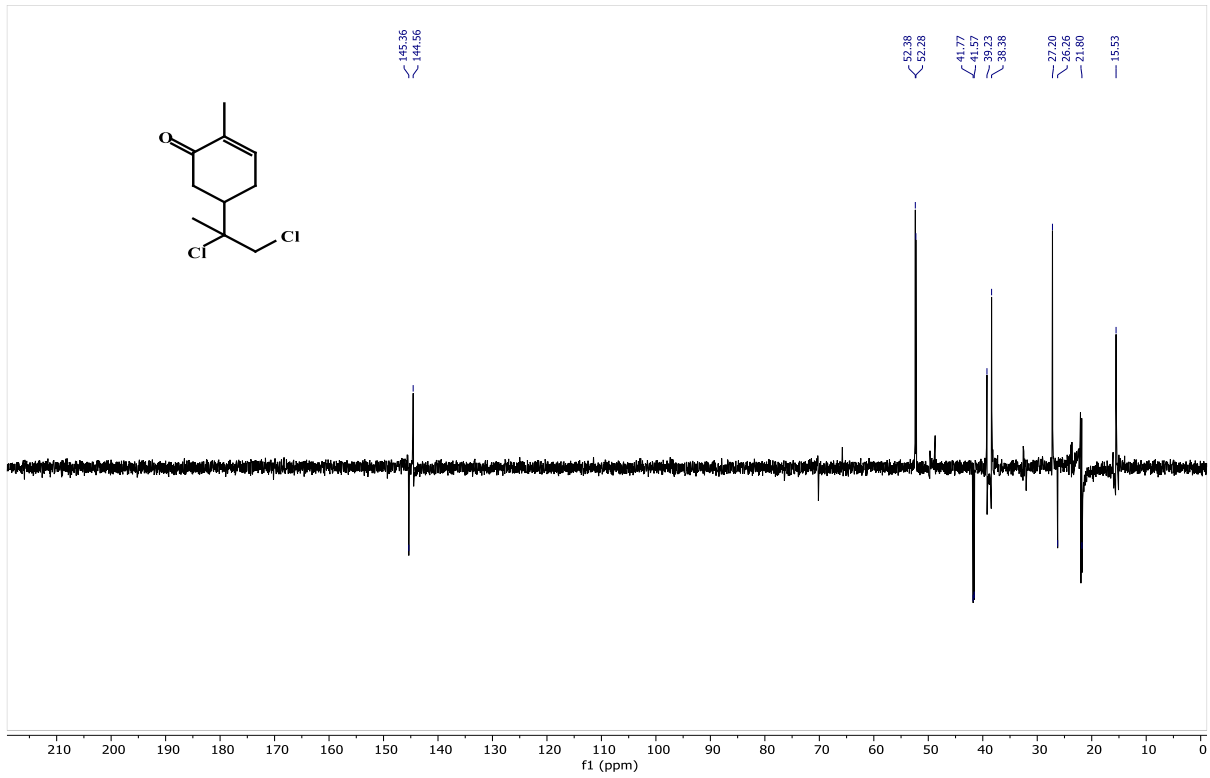


Figure S7: DEPT 135 spectrum of the carvone dichloride **c**.

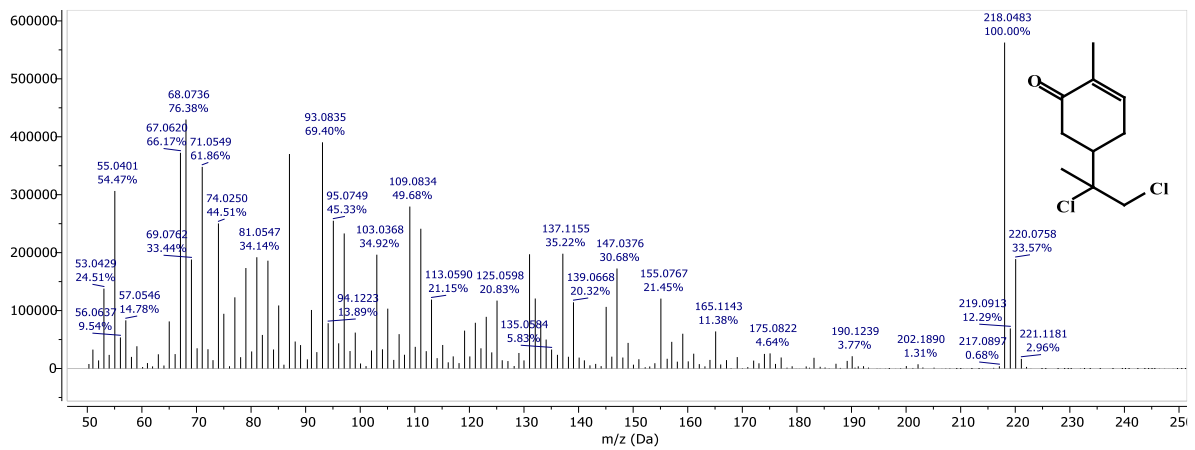


Figure S8: MS spectrum of the carvone dichloride **c**.

• Carvone vinyl allyl dichloride **d**

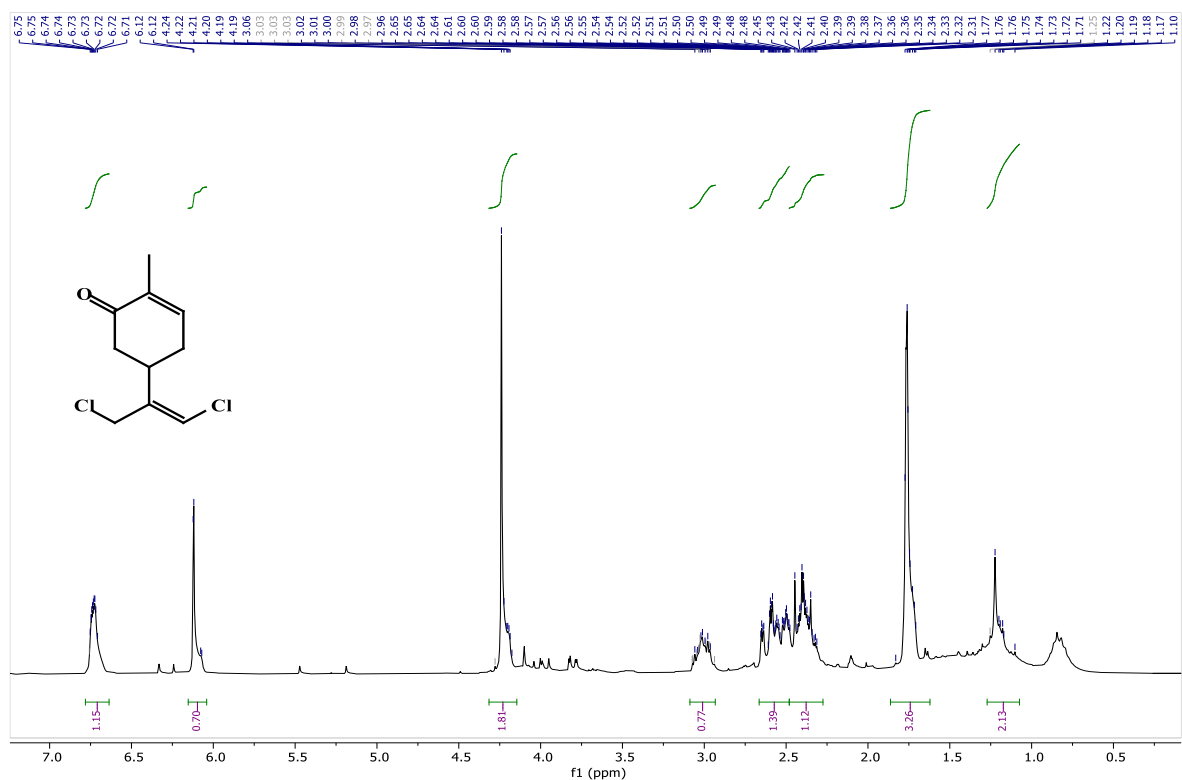


Figure S9: ^1H NMR spectrum of the carvone vinyl allyl dichloride **d**.

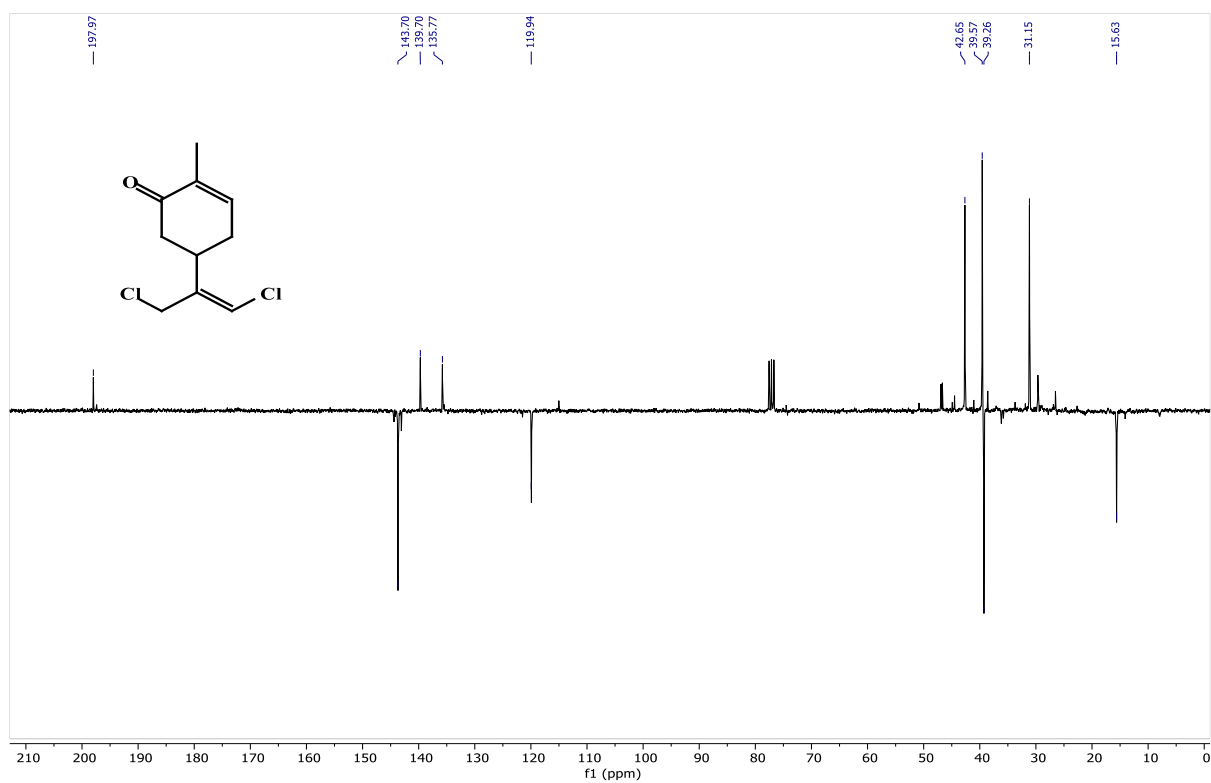


Figure S10: APT spectrum of the carvone vinyl allyl dichloride **d**.

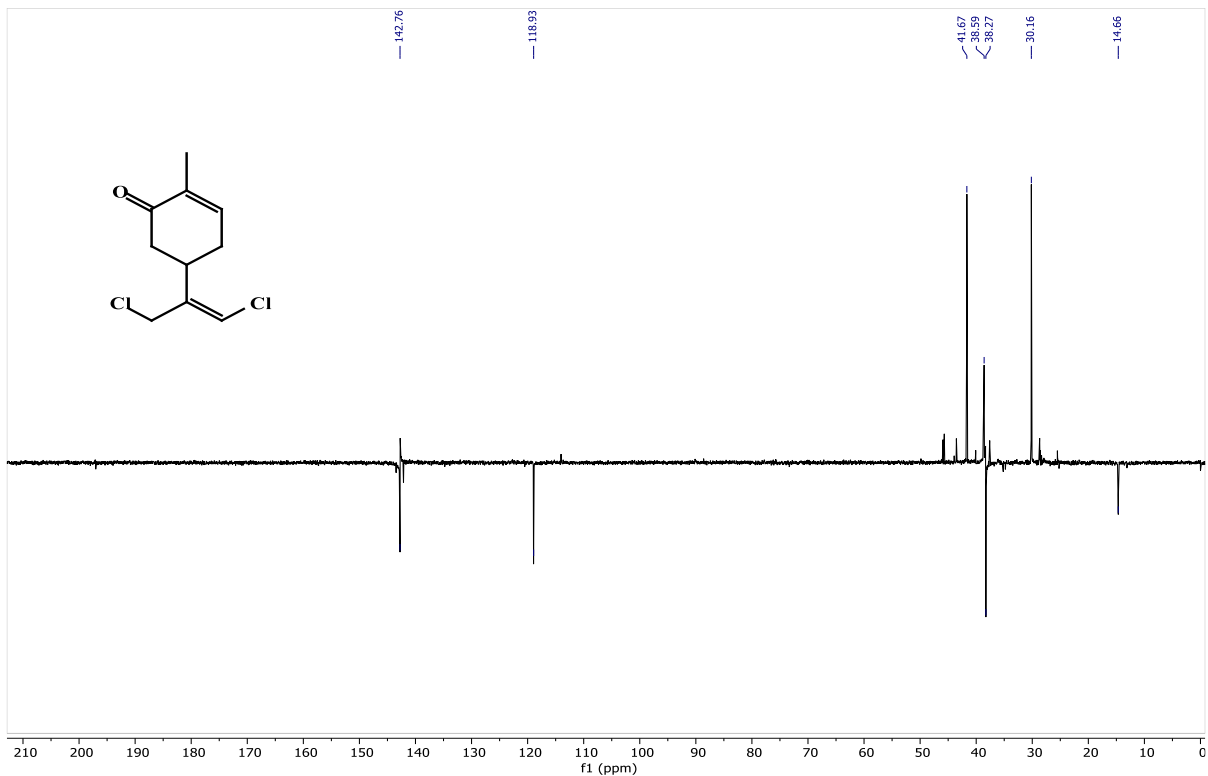


Figure S11: DEPT 135 spectrum of the carvone vinyl allyl dichloride **d**.

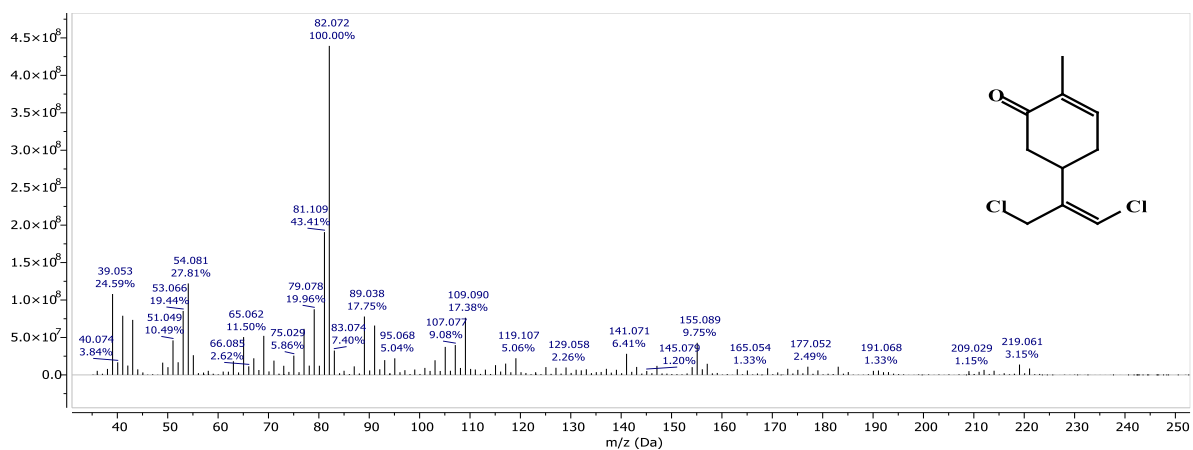


Figure S12: MS spectrum of the carvone vinyl allyl dichloride **d**.

• Carvone trichloride e

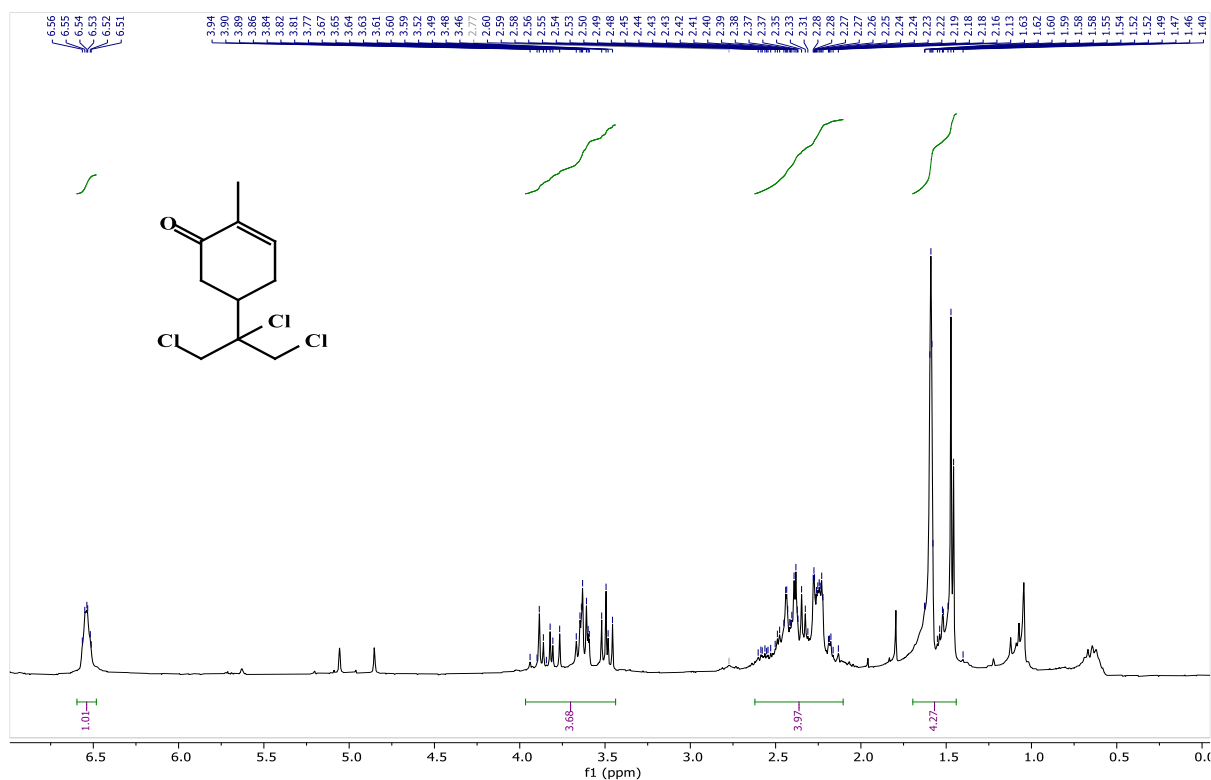


Figure S13: ¹H NMR spectrum of the carvone trichloride e in a complex mixture.

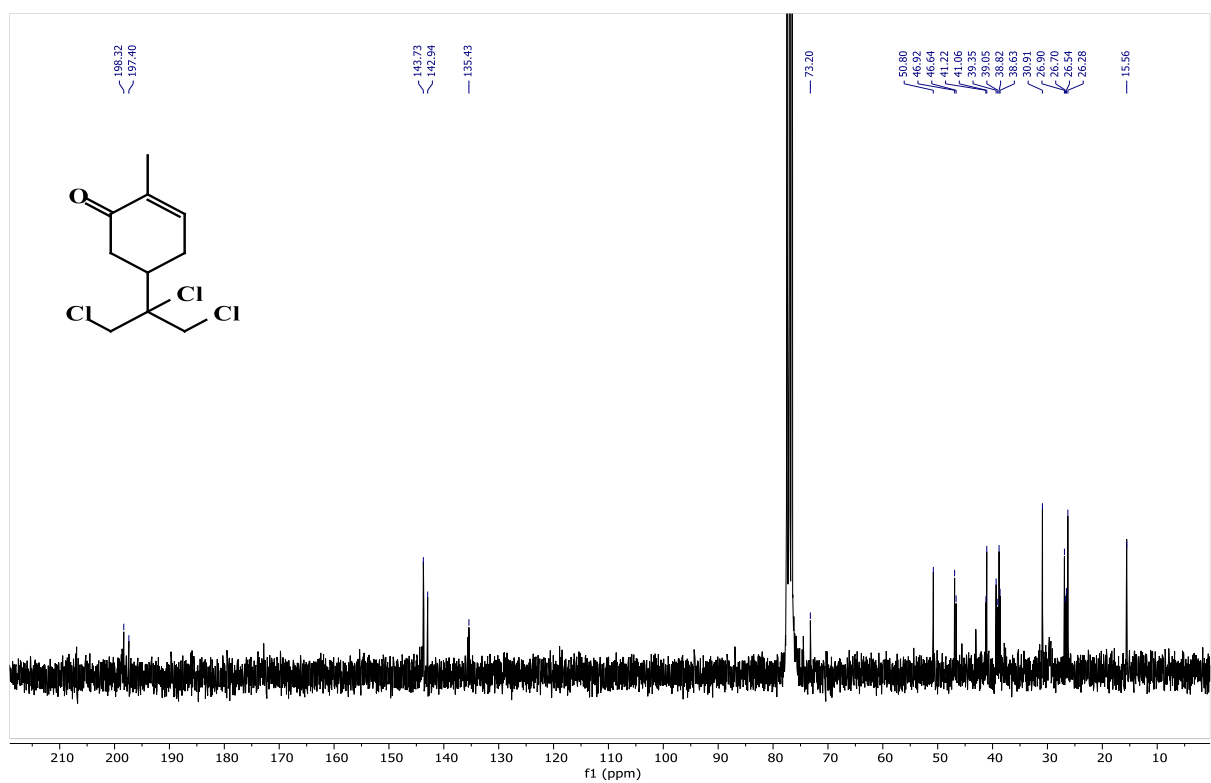


Figure S14: ¹³C spectrum of the carvone trichloride e in a complex mixture.

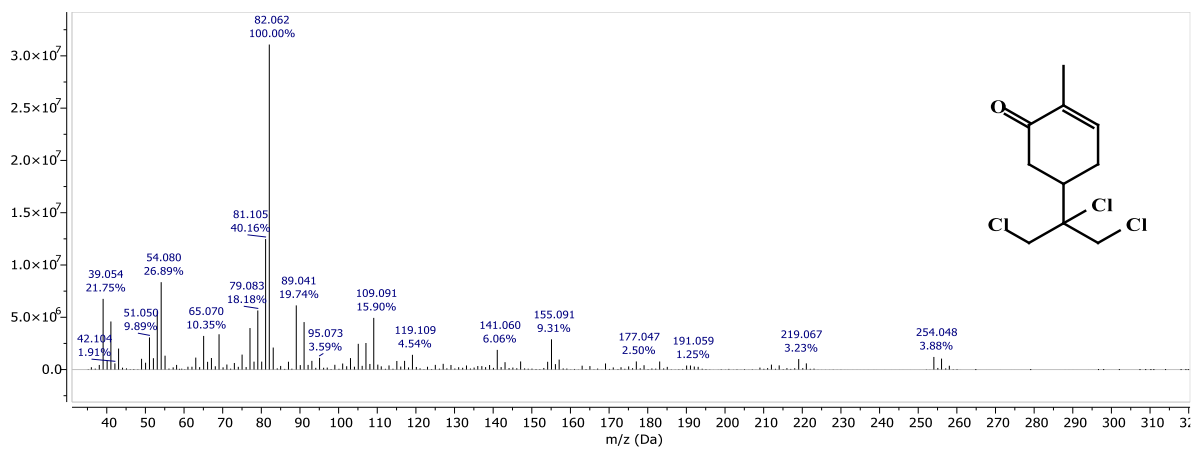


Figure S15: MS spectrum of the carvone trichloride **e**.

➤ Derivatives of limonene oxide **f**

• Limonene oxide monochloride **g**

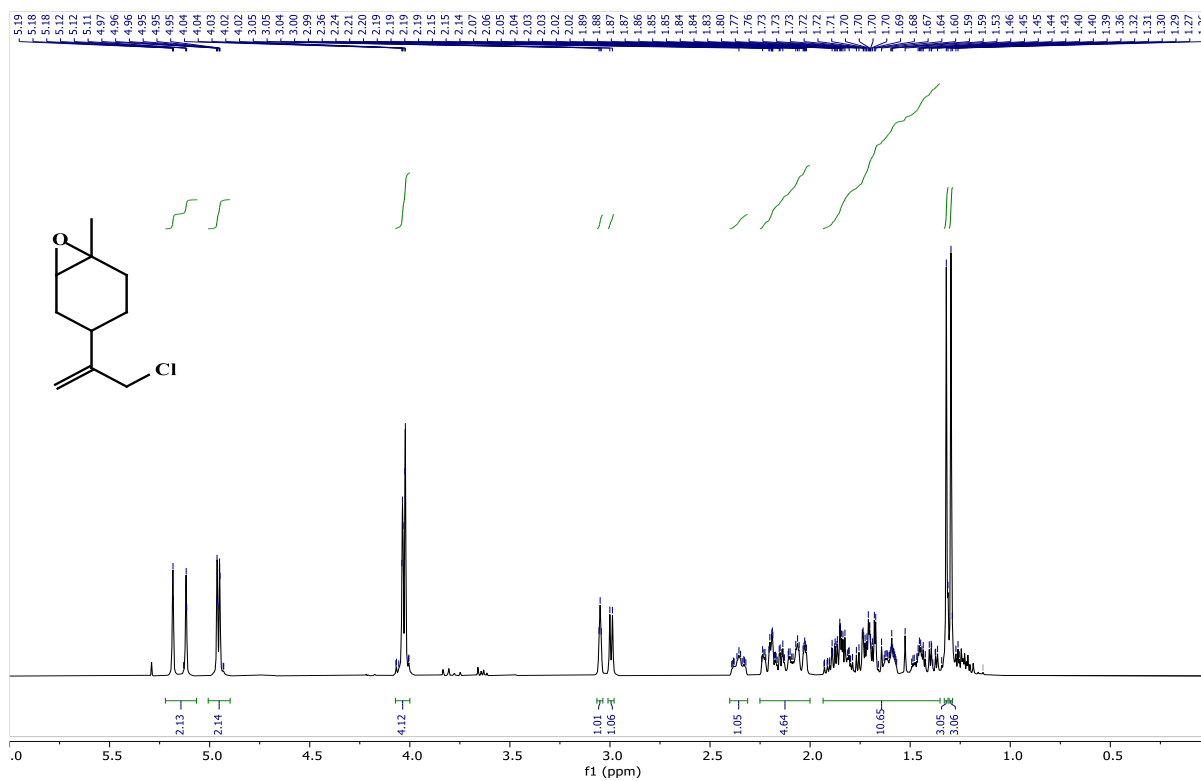


Figure S16: ^1H NMR spectrum of the limonene oxide monochloride **g**.

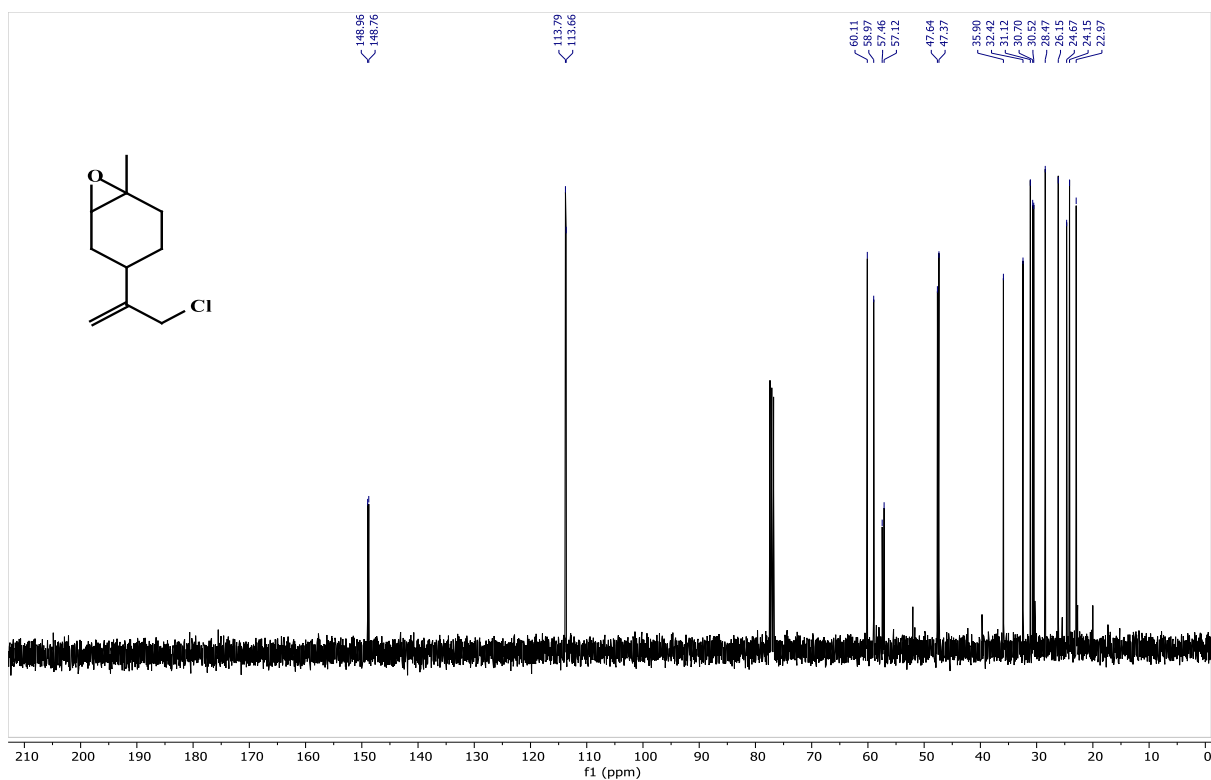


Figure S17: ^{13}C spectrum of the limonene oxide monochloride **g**.

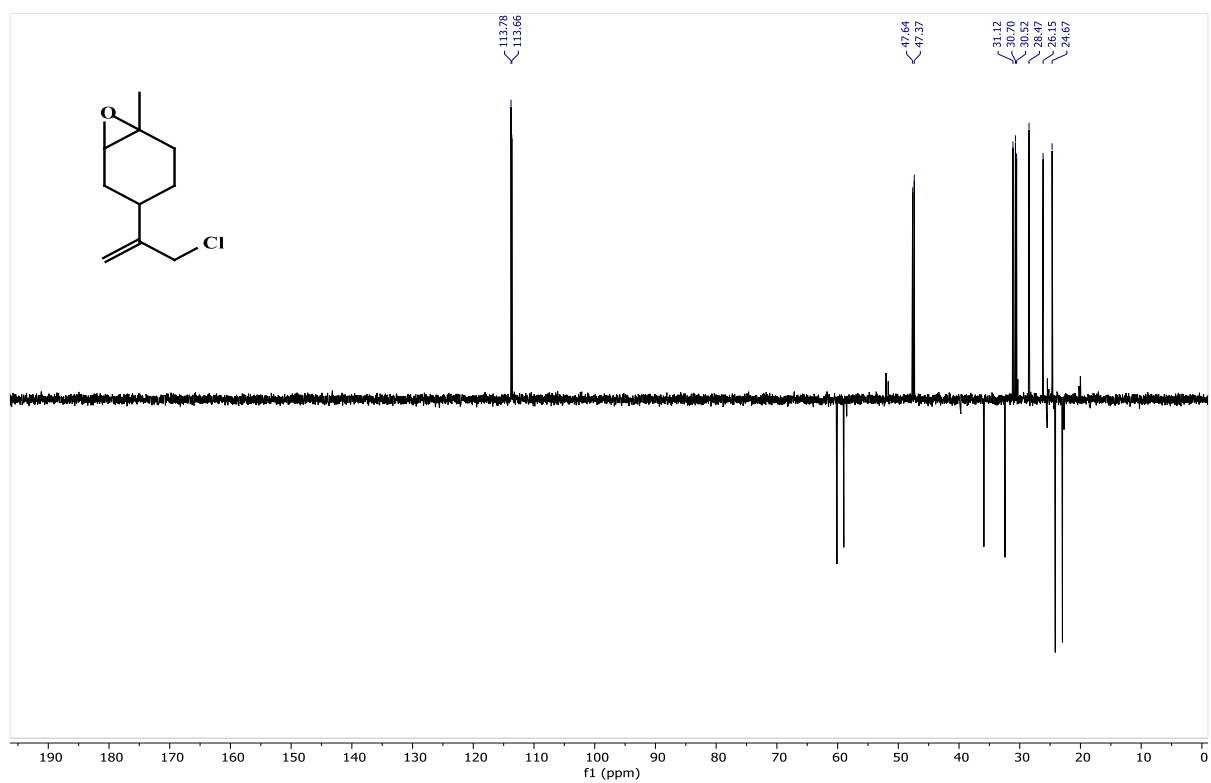


Figure S18: DEPT 135 spectrum of the limonene oxide monochloride **g**.

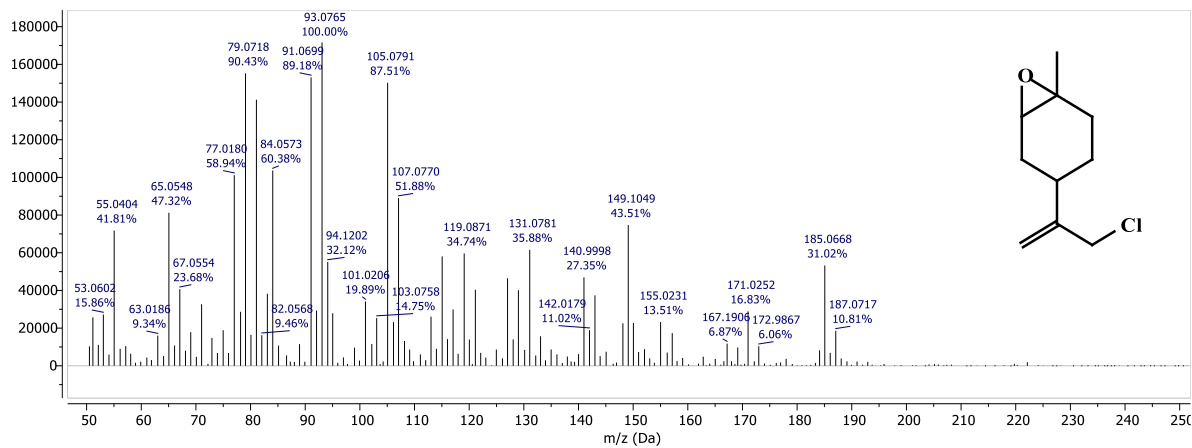


Figure S19: MS spectrum of the limonene oxide monochloride **g**.

- Limonene oxide dichloride **h**

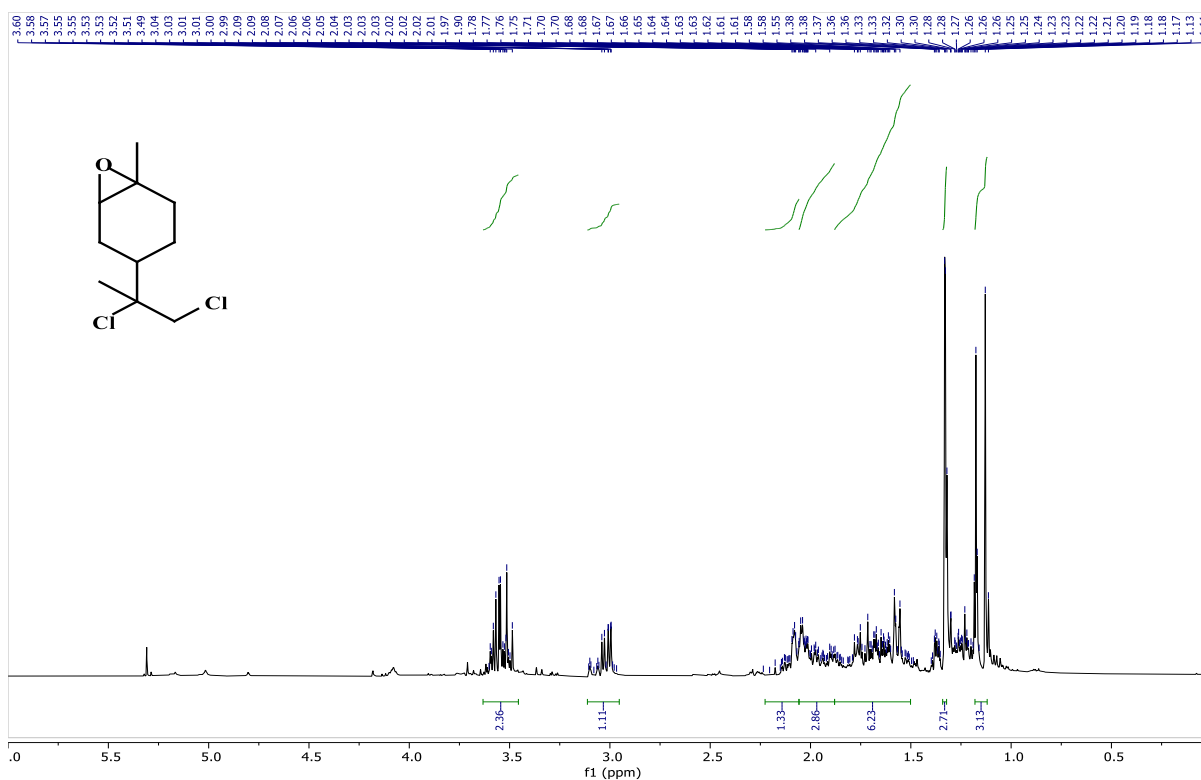


Figure S20: ^1H NMR spectrum of the limonene oxide dichloride **h**.

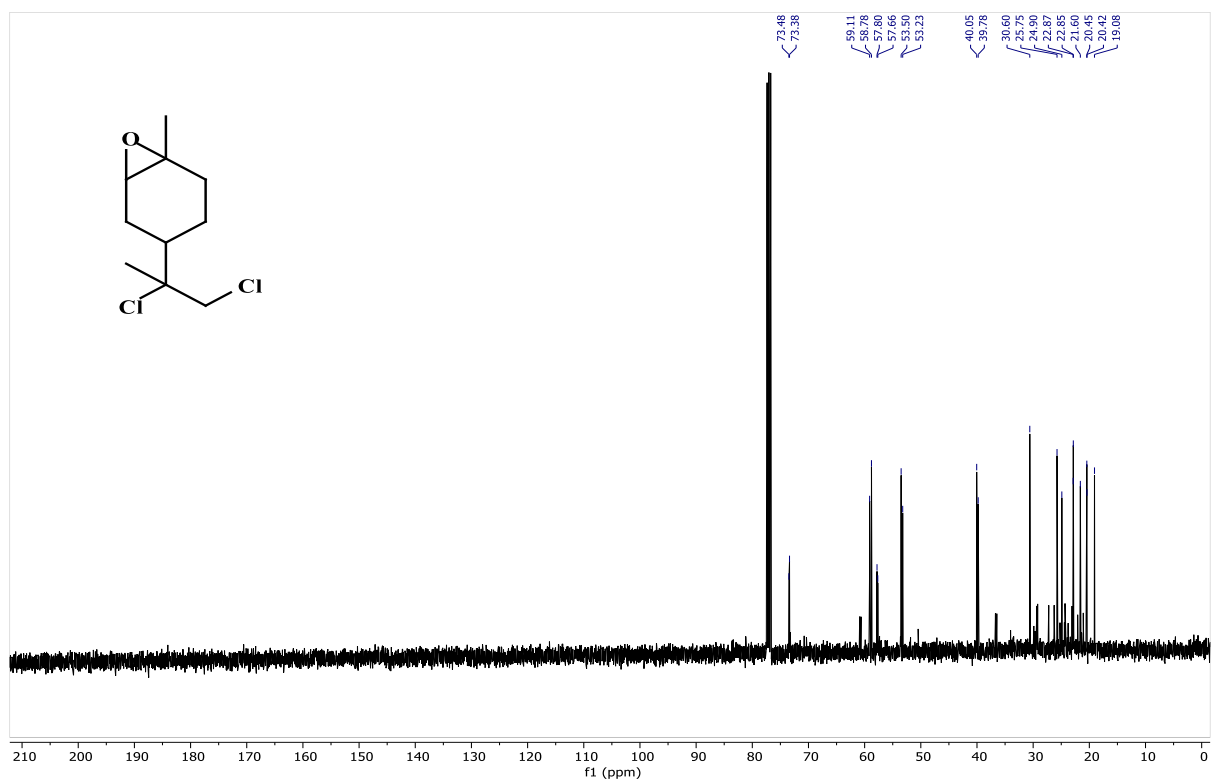


Figure S21: ^{13}C spectrum of the limonene oxide dichloride **h**.

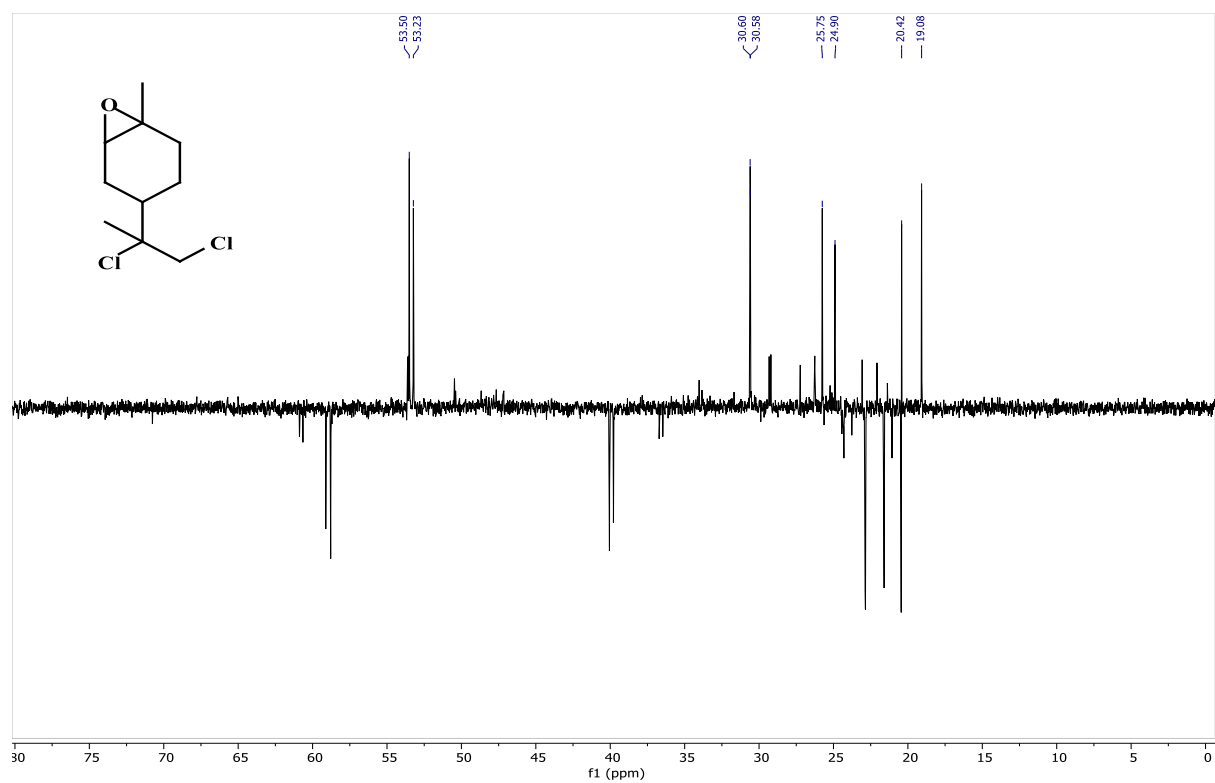


Figure S22: DEPT 135 spectrum of the limonene oxide dichloride **h**.

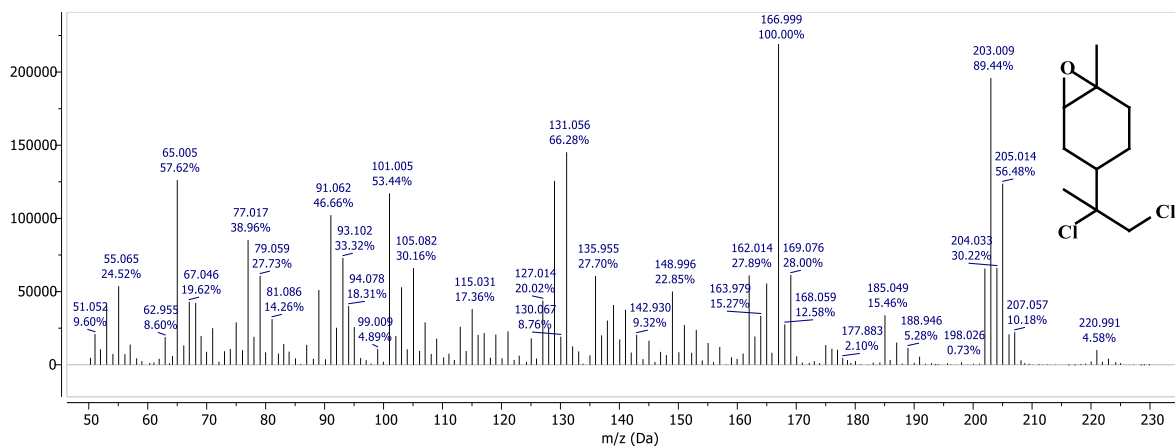


Figure S23: MS spectrum of the limonene oxide dichloride **h**.

➤ Derivatives of nootkatone **i**

● Nootkatone monochloride **j**

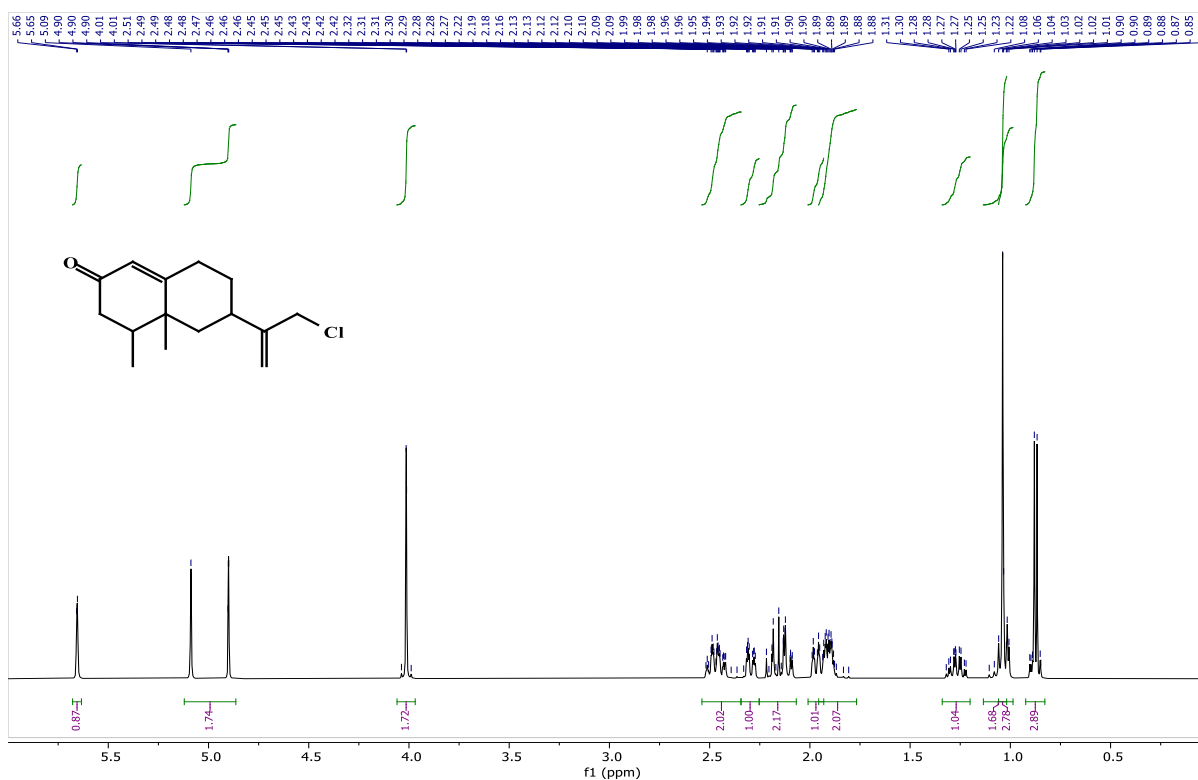
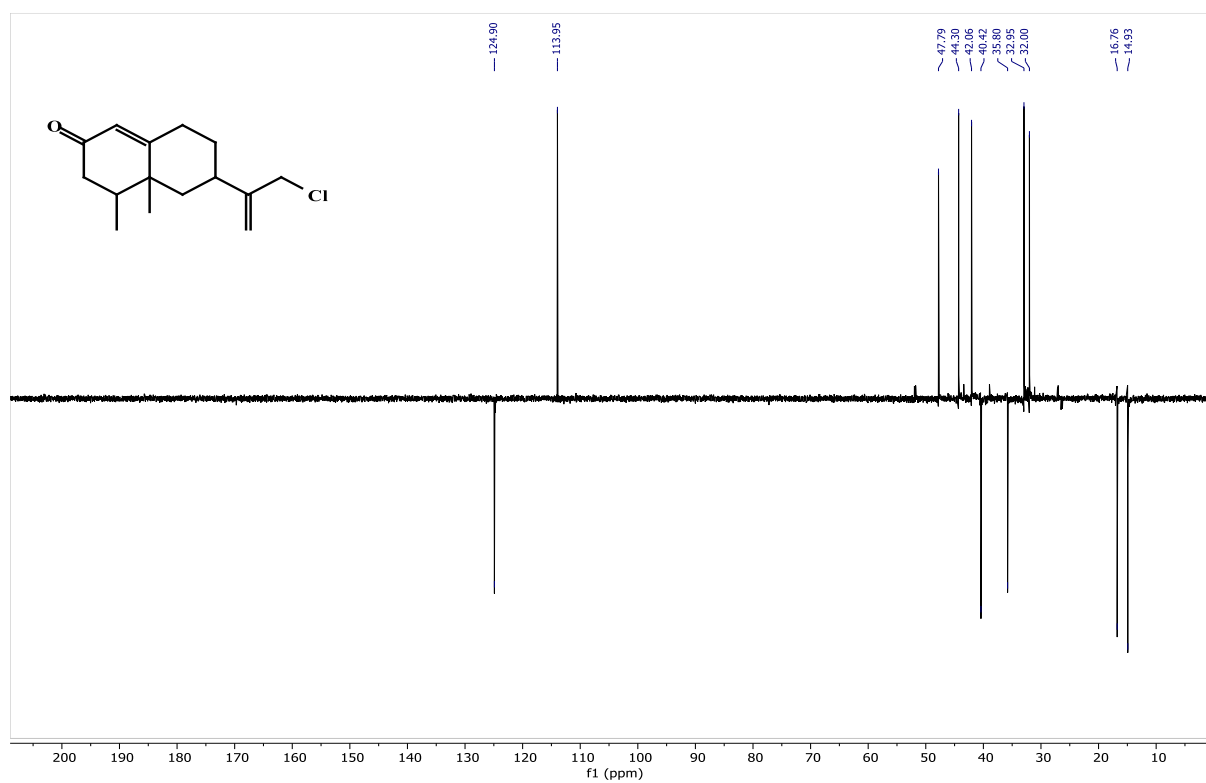
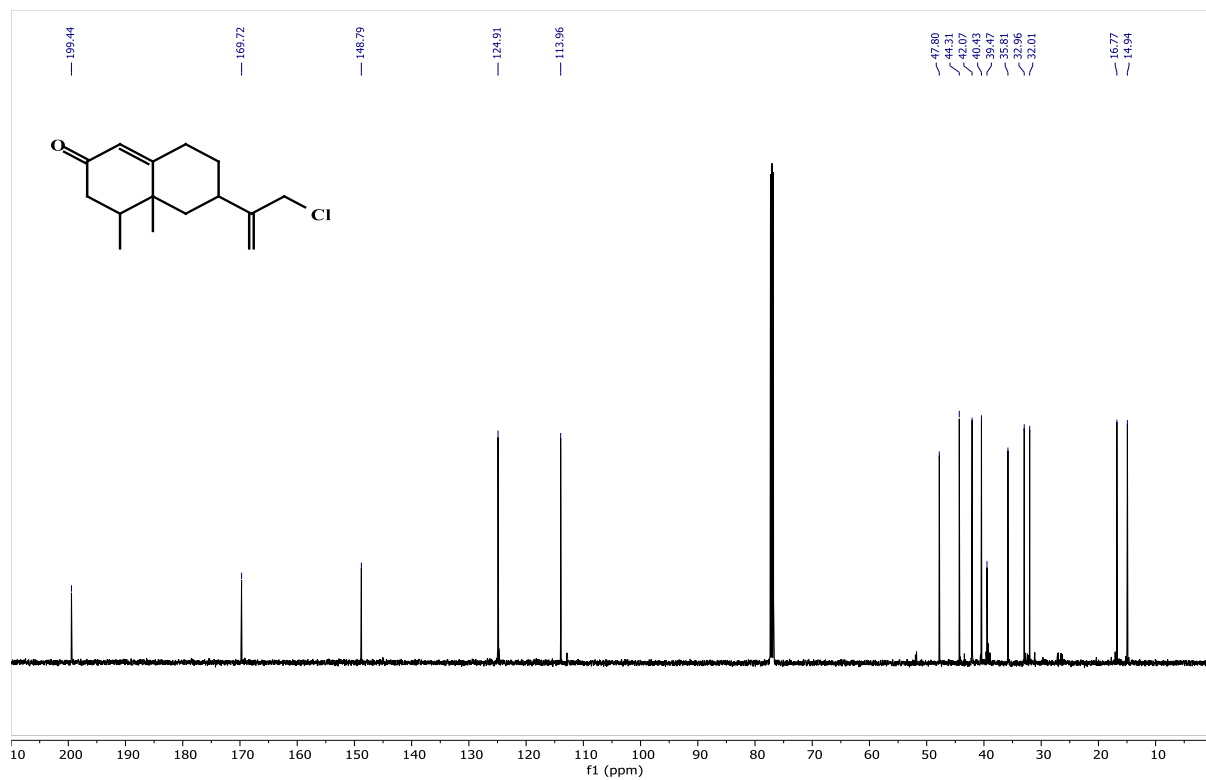


Figure S24: ¹H NMR spectrum of the nootkatone monochloride **j**.



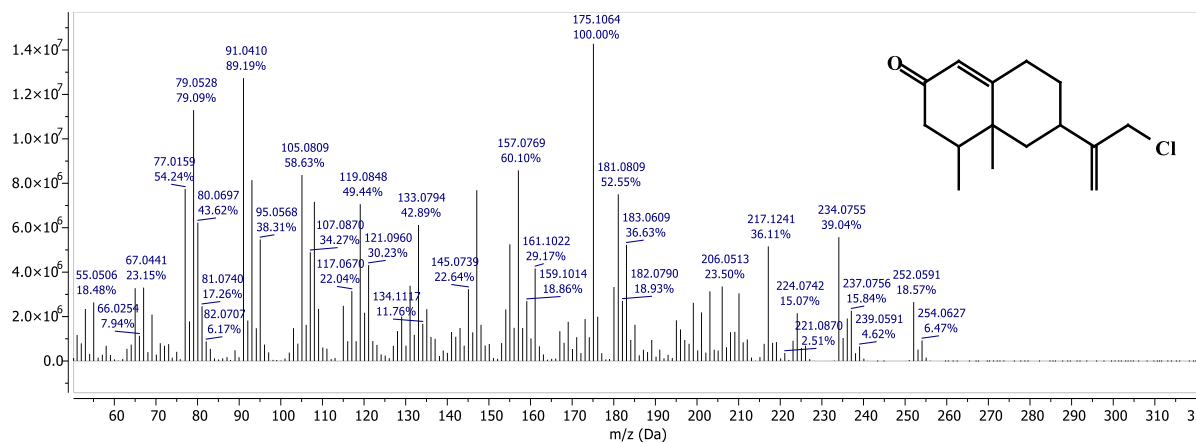


Figure S27: MS spectrum of the nootkatone monochloride **j**.

- Nootkatone vinyl allyl dichloride **k**

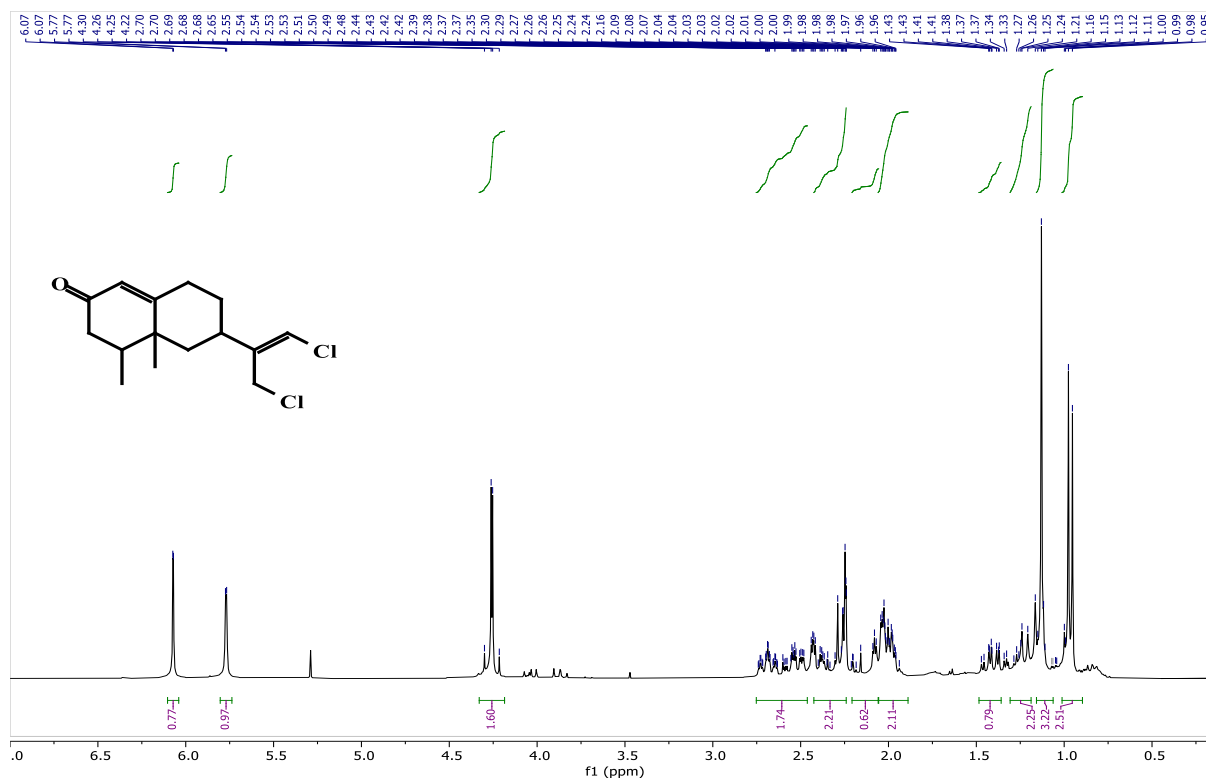


Figure S28: ¹H NMR spectrum of the nootkatone vinyl allyl dichloride **k**.

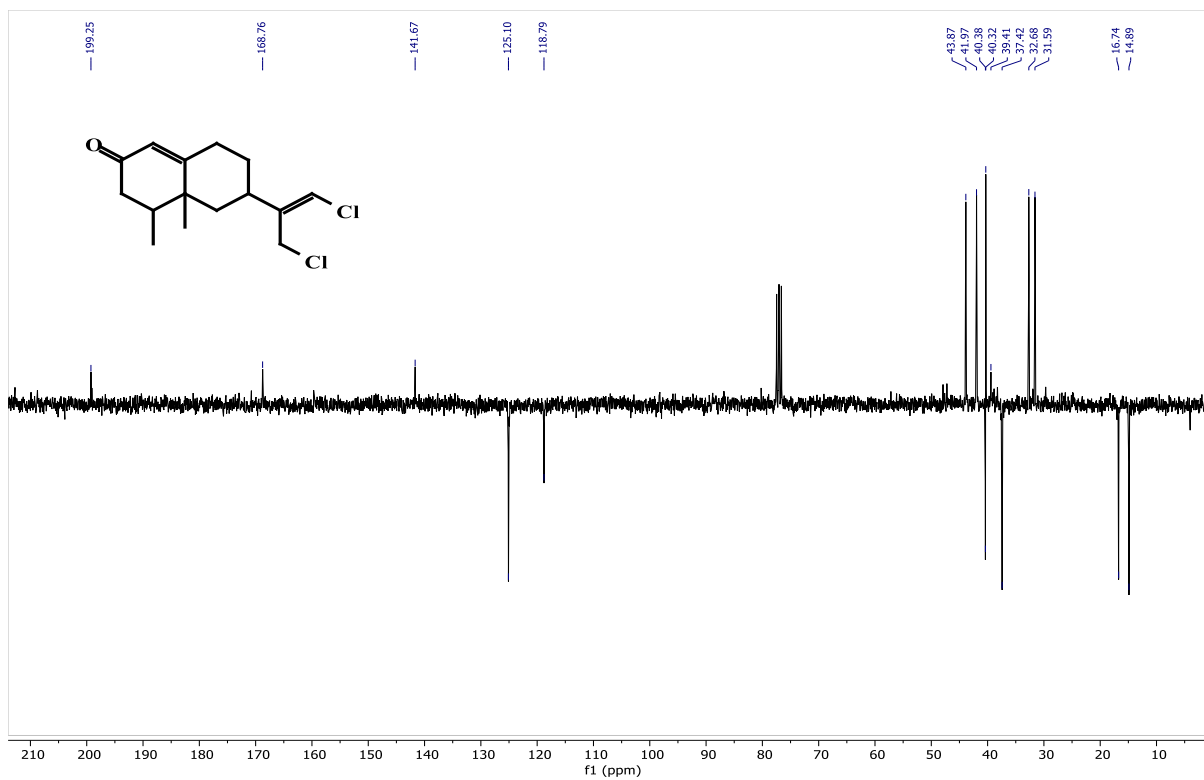


Figure S29: APT spectrum of the nootkatone vinyl allyl dichloride k.

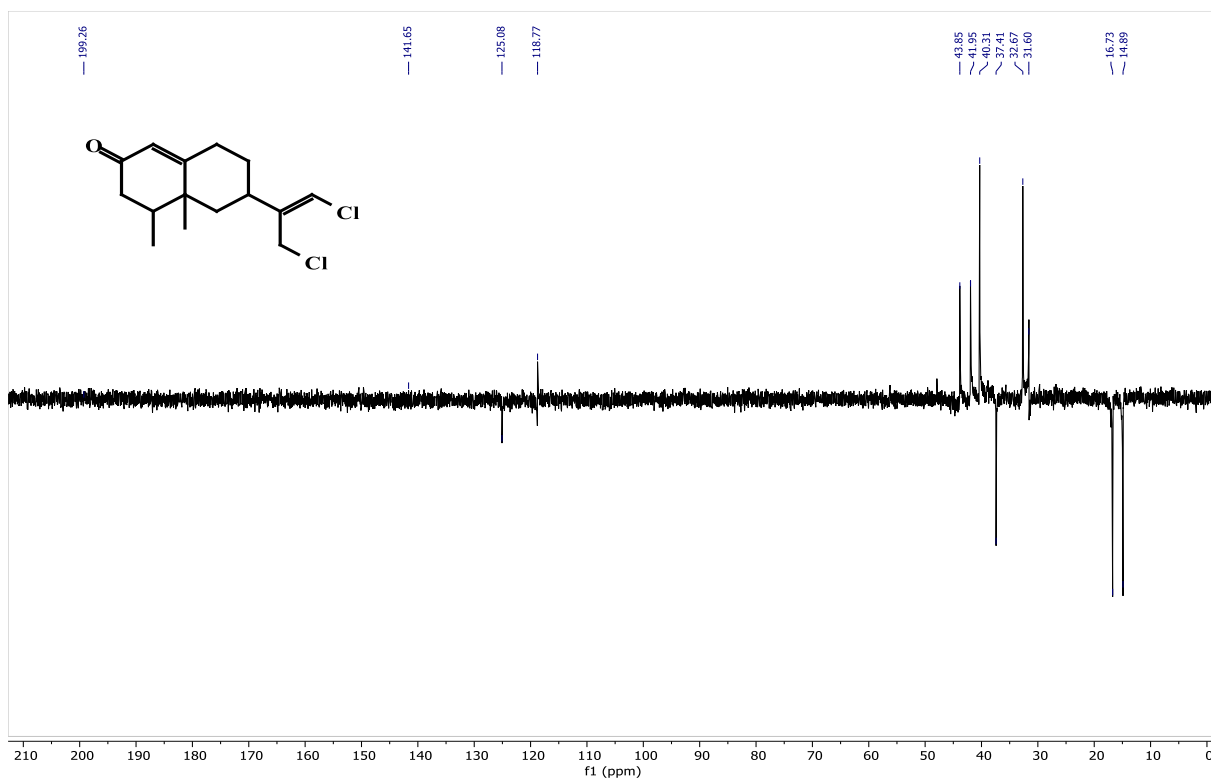


Figure S30: DEPT 135 spectrum of the nootkatone vinyl allyl dichloride k.

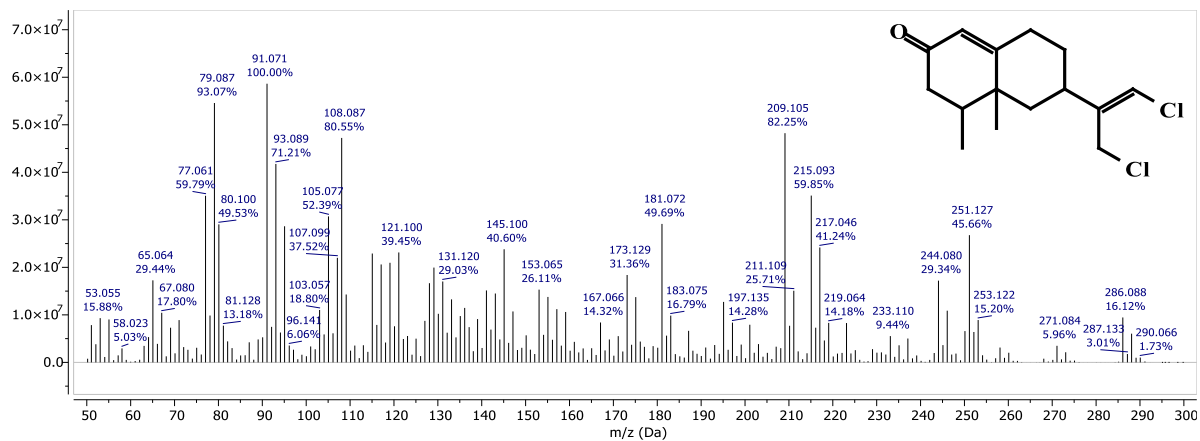


Figure S31: MS spectrum of the nootkatone vinyl allyl dichloride **k**.

- Nootkatone dichloride **l**

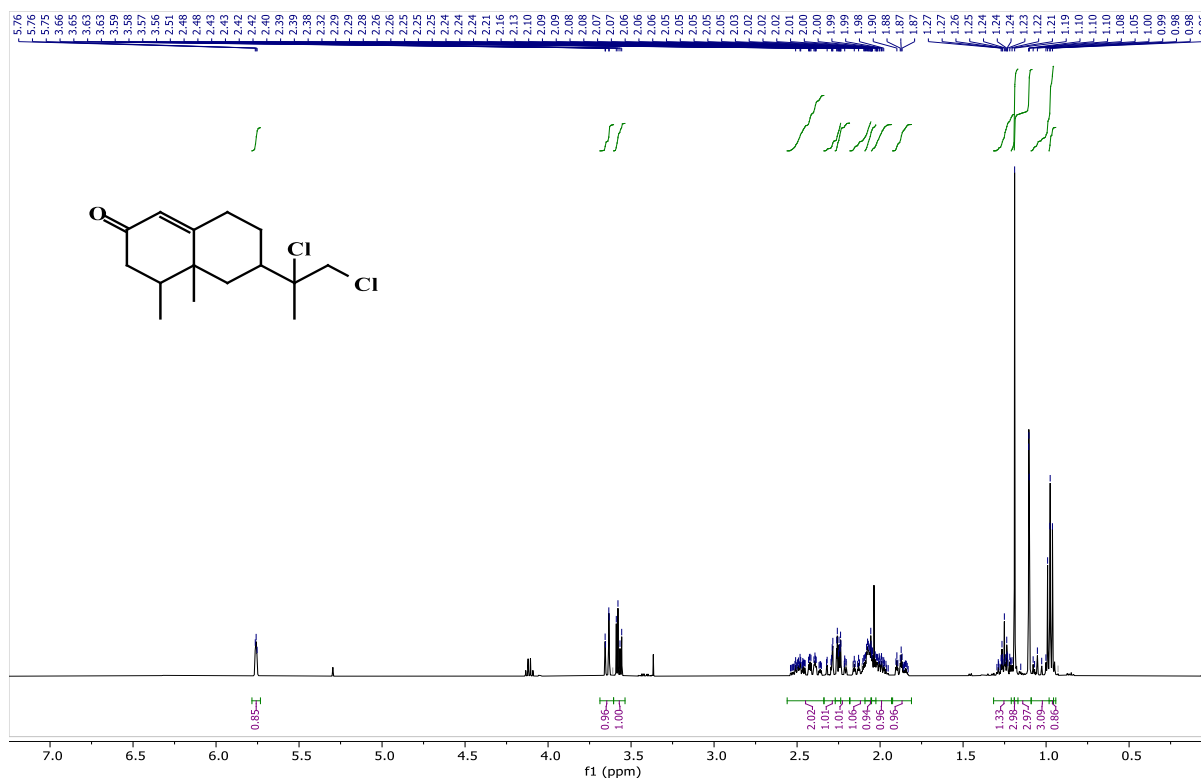


Figure S32: ^1H NMR spectrum of the nootkatone dichloride **l**.

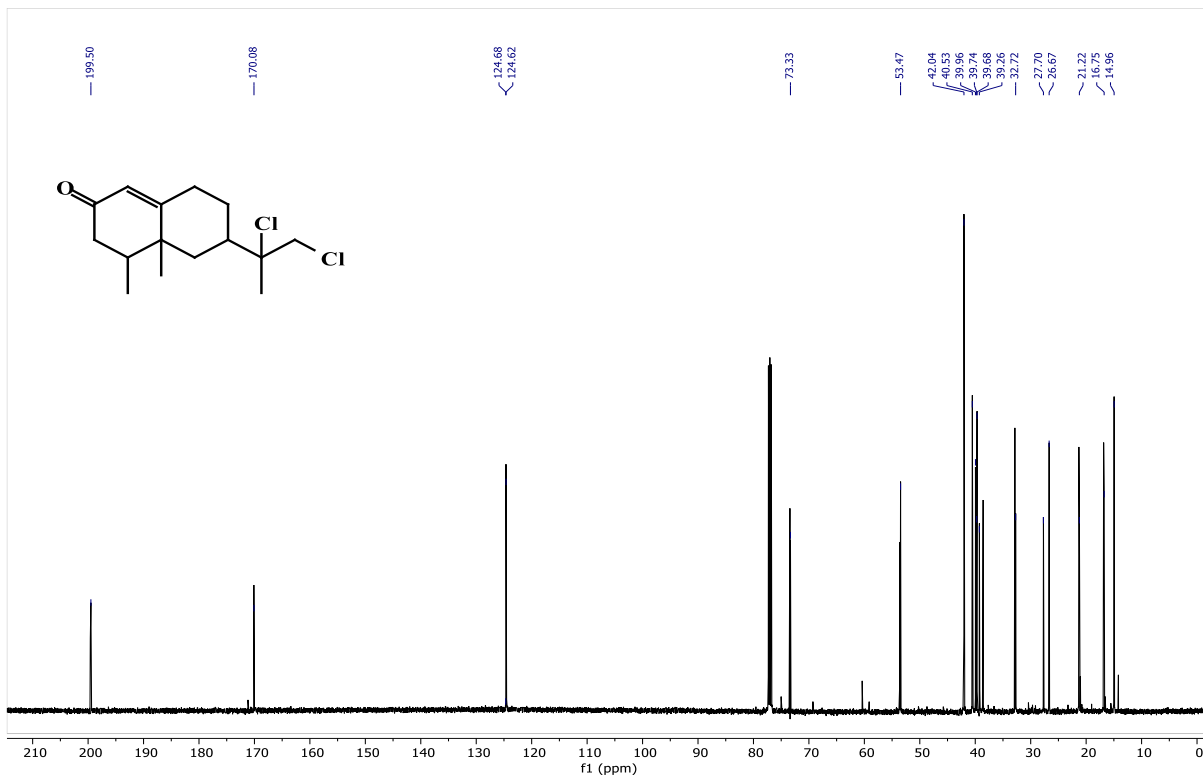


Figure S33: ^{13}C spectrum of the nootkatone dichloride **1**.

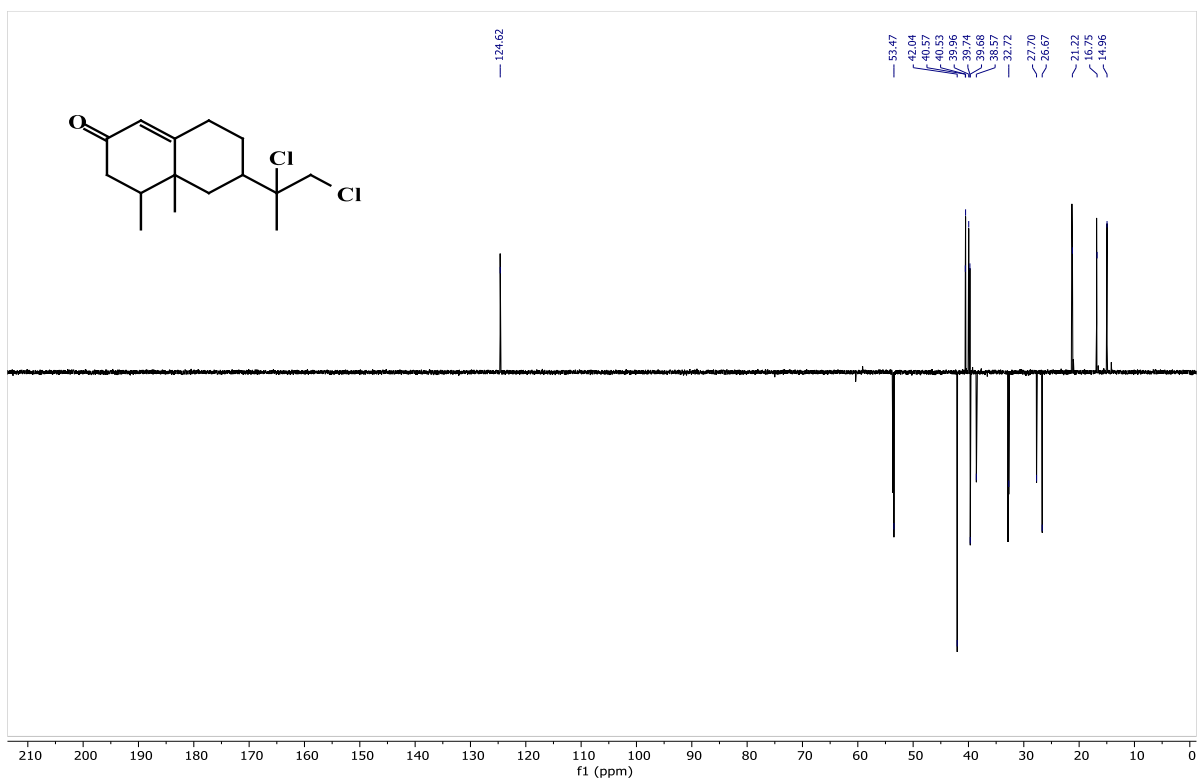


Figure S34: DEPT 135 spectrum of the nootkatone dichloride **1**.

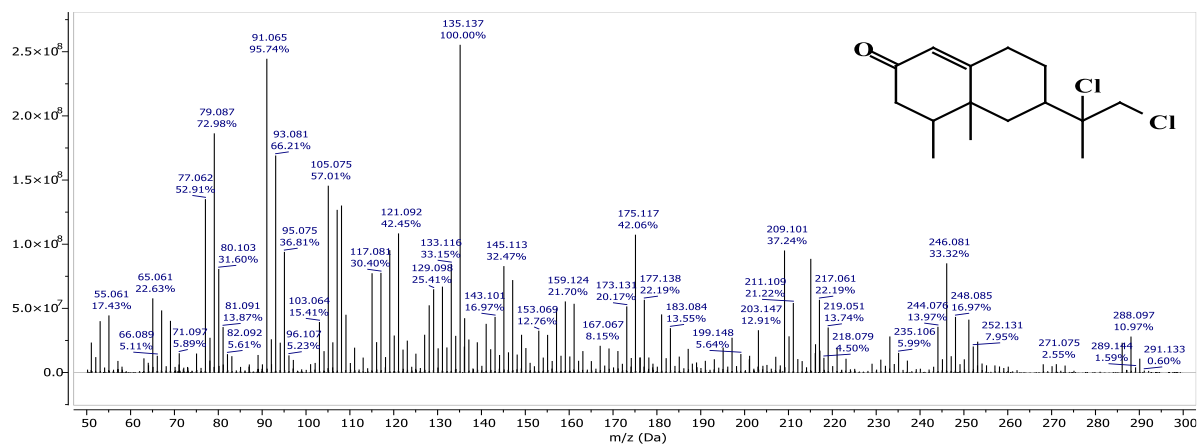


Figure S35: MS spectrum of the nootkatone dichloride **I**.

➤ Derivatives of pulegone **m**

• Pulegone monochloride **n**

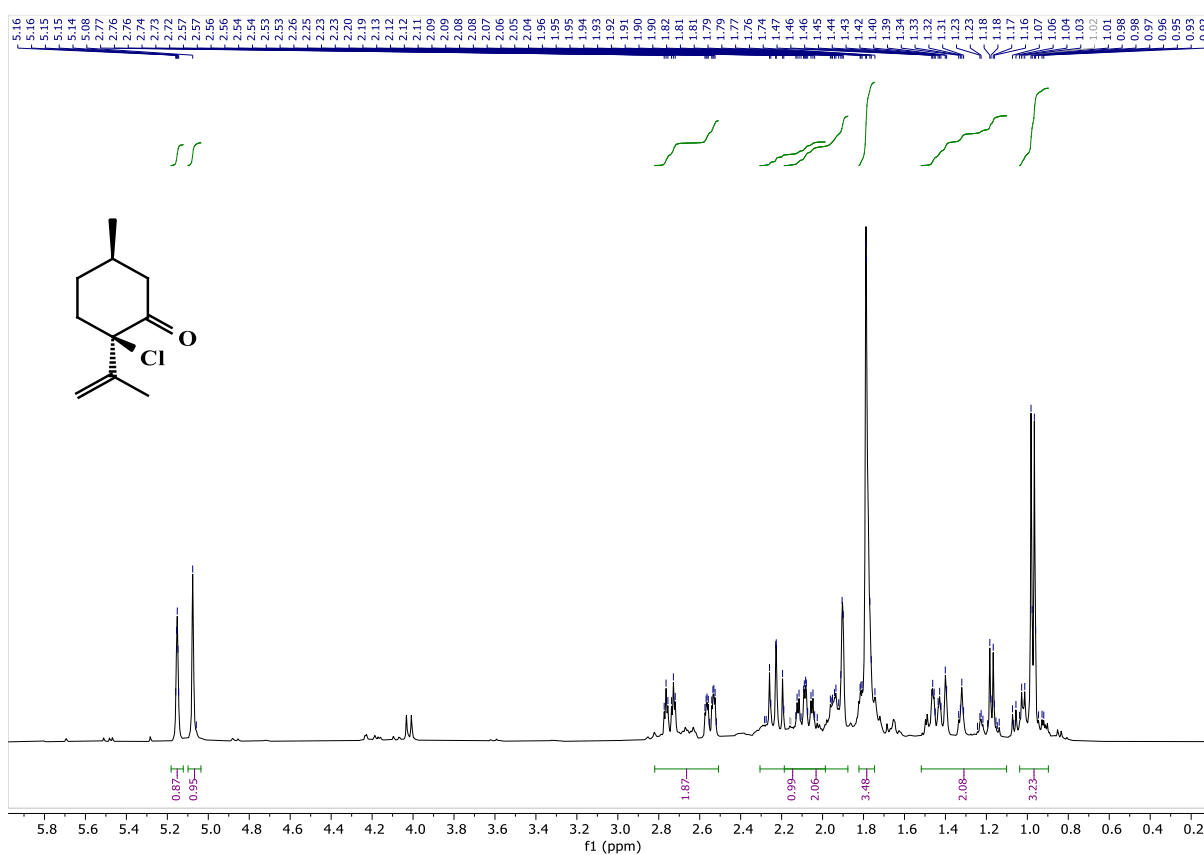


Figure S36: ¹H NMR spectrum of the pulegone monochloride **n**.

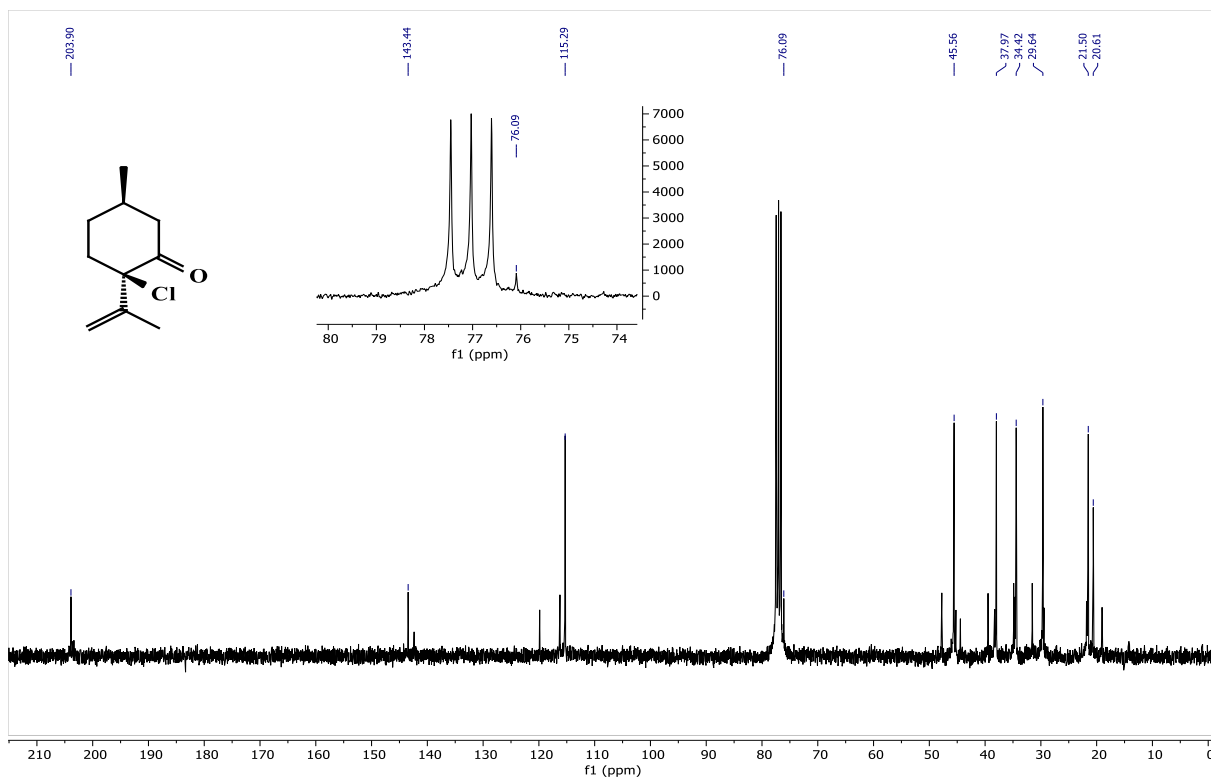


Figure S37: ^{13}C spectrum of the pulegone monochloride **n**.

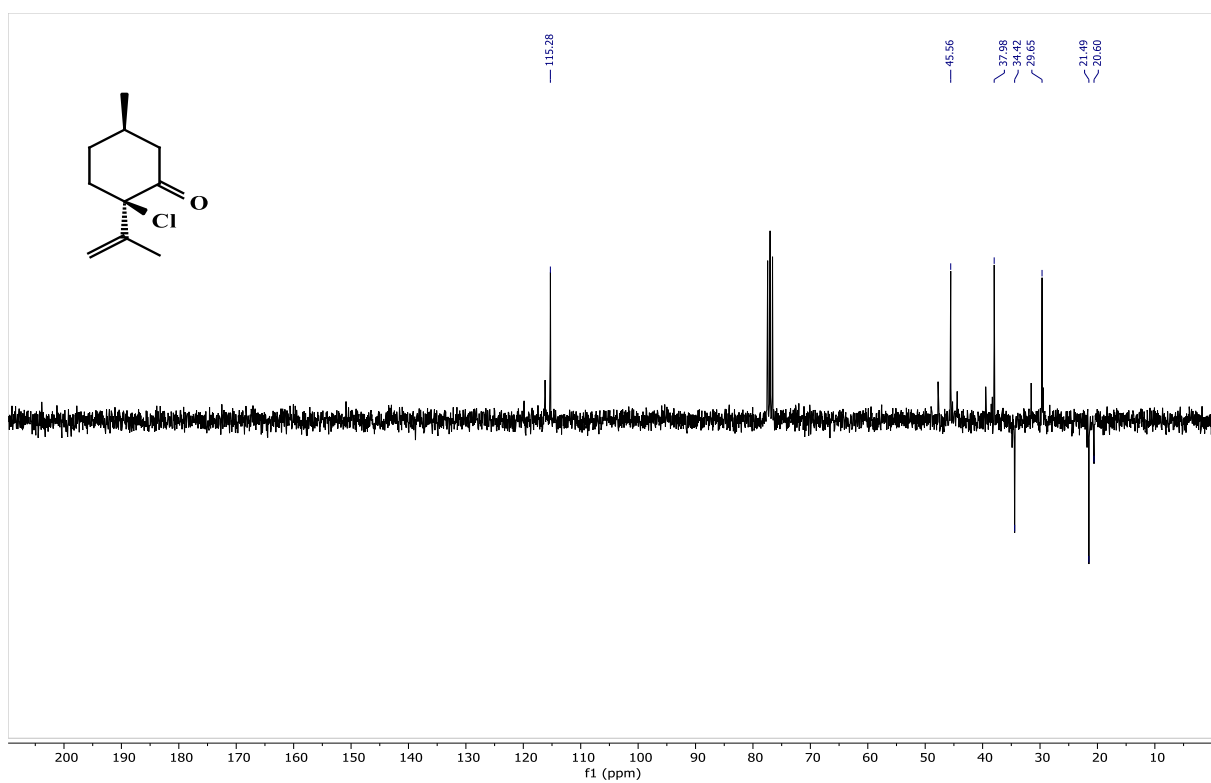


Figure S38: APT spectrum of the pulegone monochloride **n**.

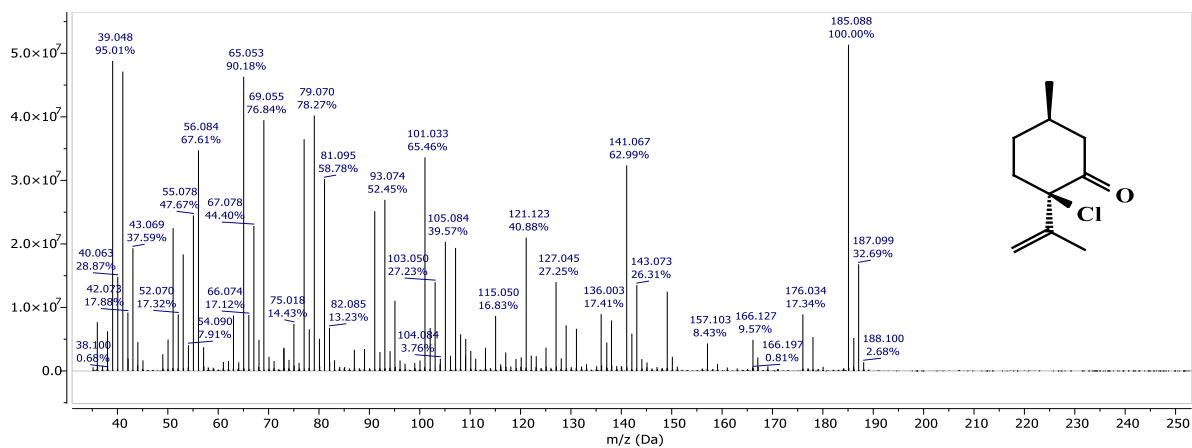


Figure S39: MS spectrum of the pulegone monochloride **n**.

- Pulegone monochloride **n'**

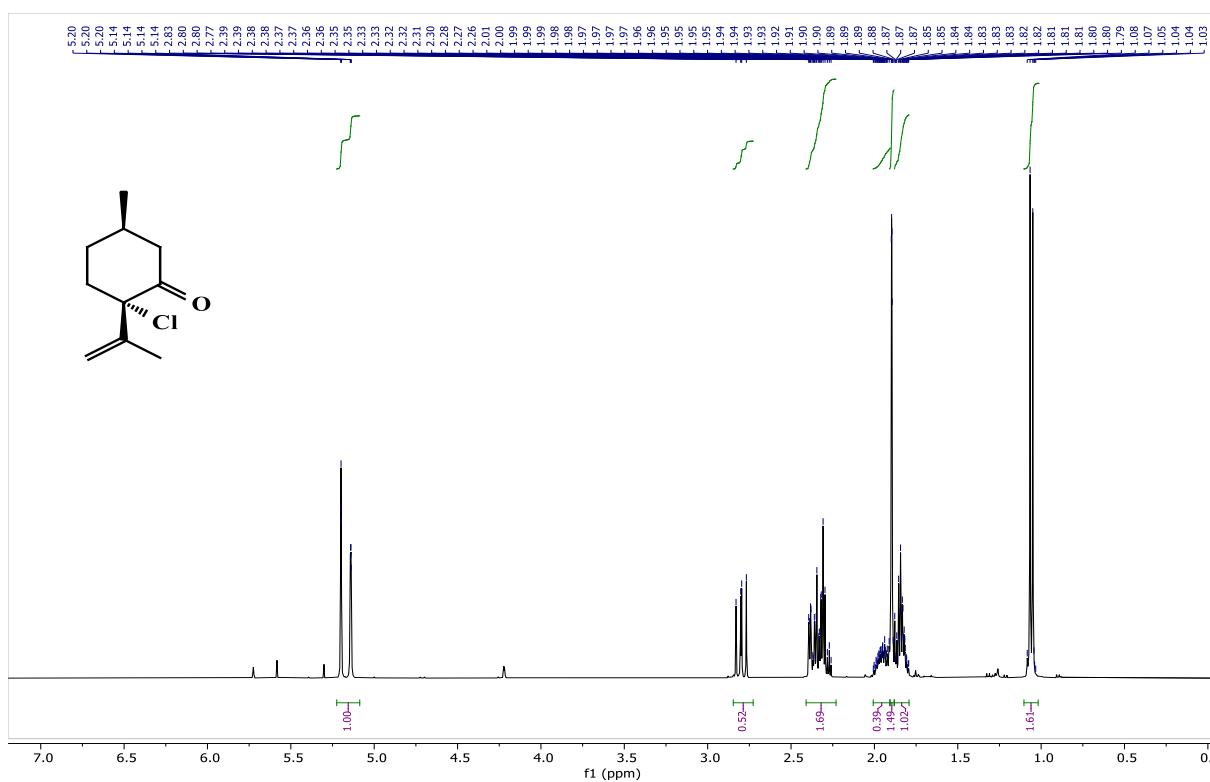


Figure S40: ¹H NMR spectrum of the pulegone monochloride **n'**.

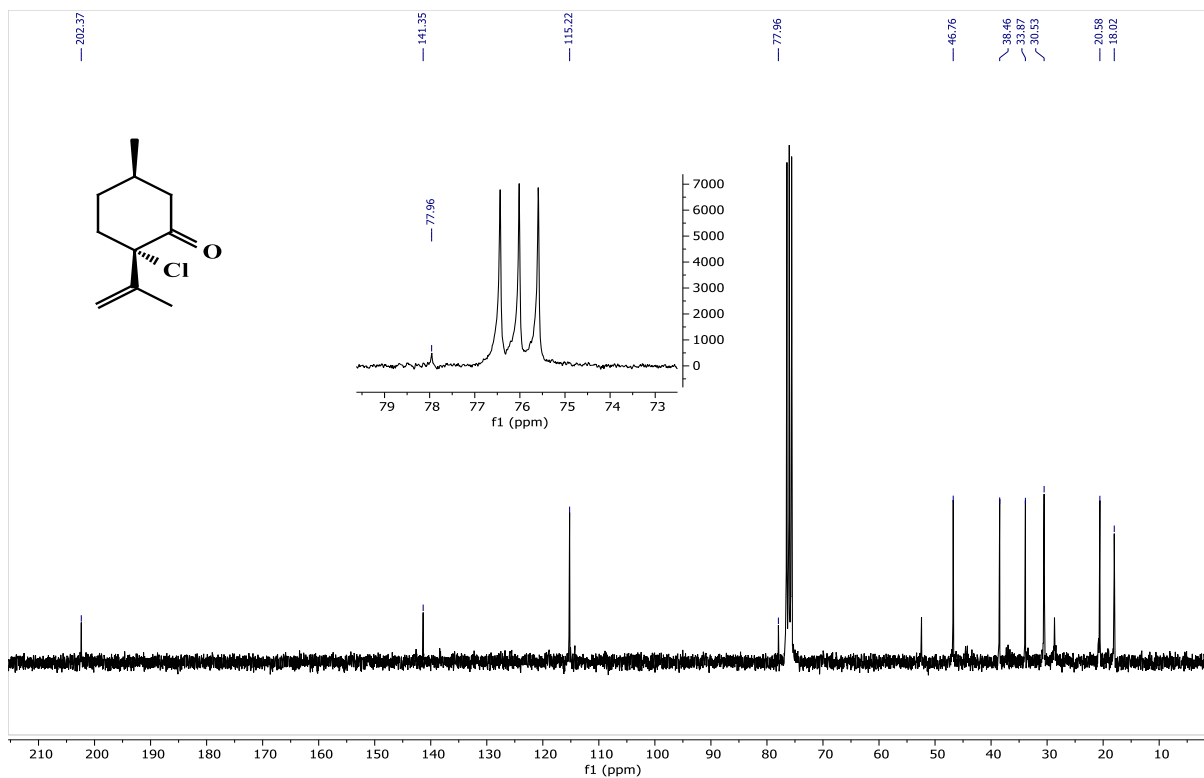


Figure S41: ^{13}C spectrum of the pulegone monochloride n'.

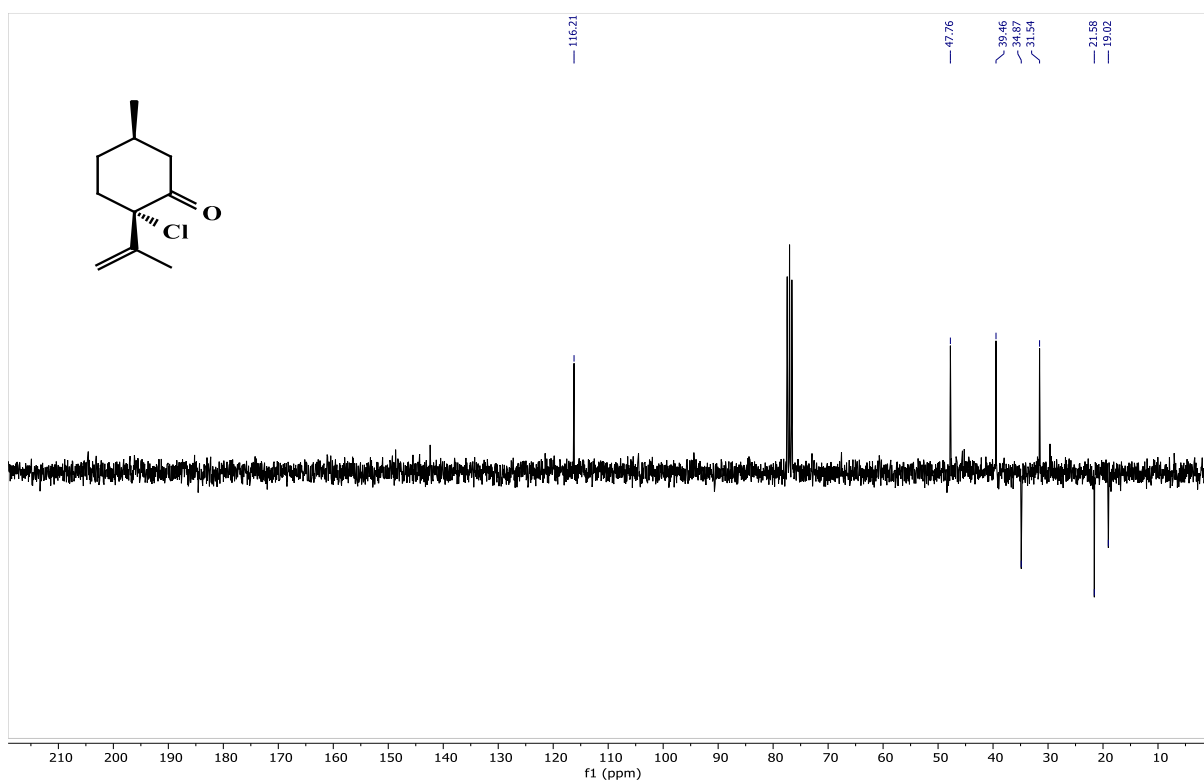


Figure S42: APT spectrum of the pulegone monochloride n'.

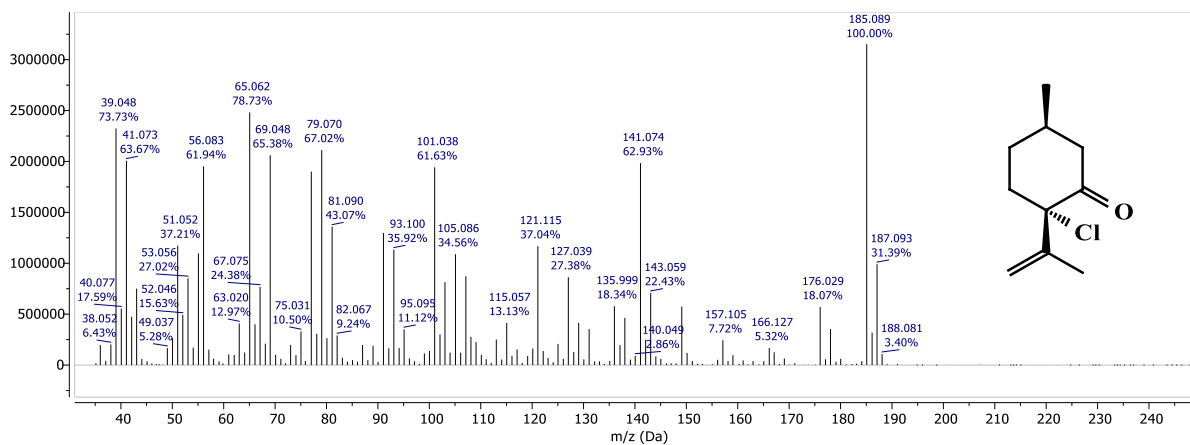


Figure S43: MS spectrum of the pulegone monochloride **n'**.

- Pulegone allyl dichloride **o**

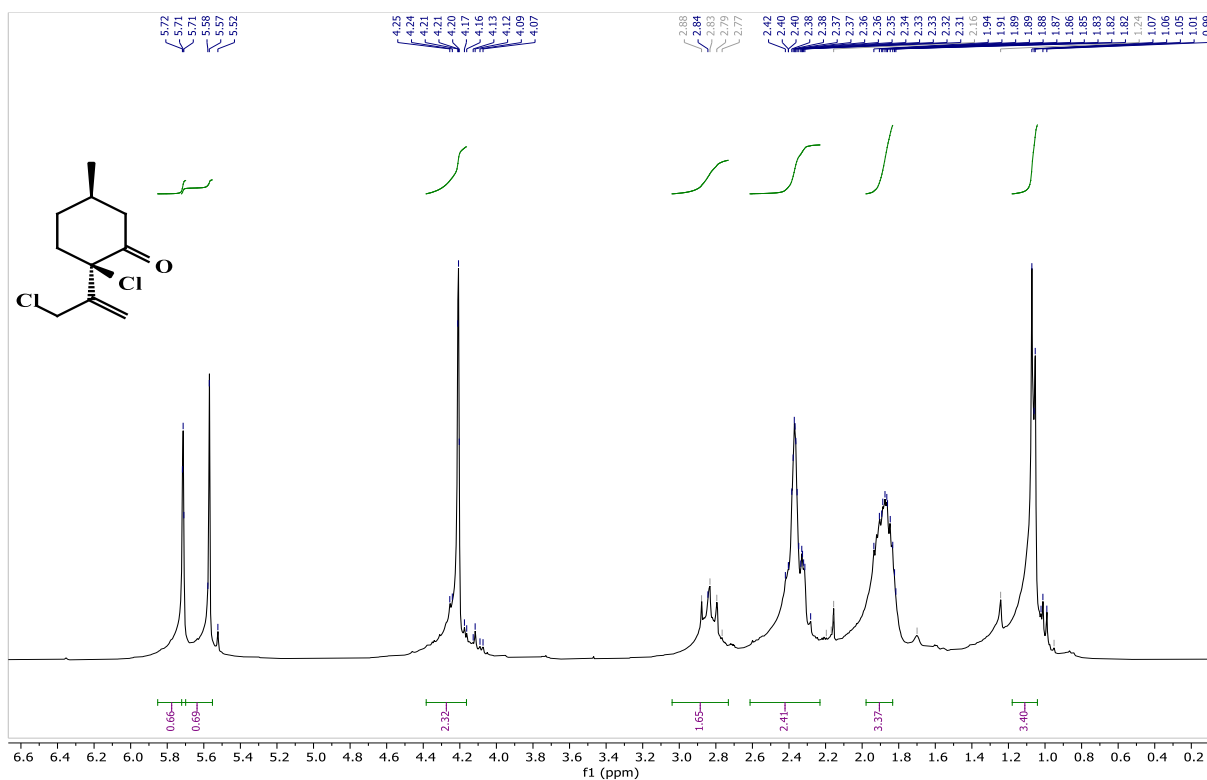


Figure S44: ¹H NMR spectrum of the pulegone vinyl allyl dichloride **o**.

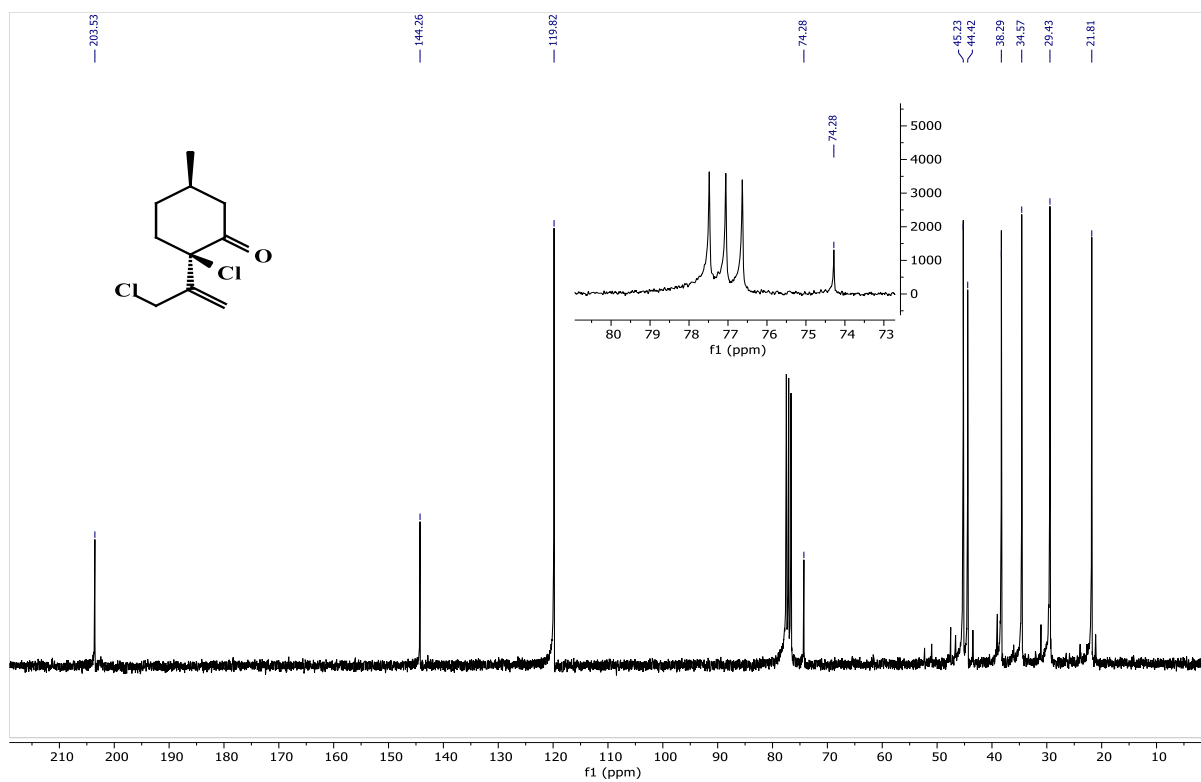


Figure S45: ¹³C spectrum of the pulegone vinyl allyl dichloride **o**.

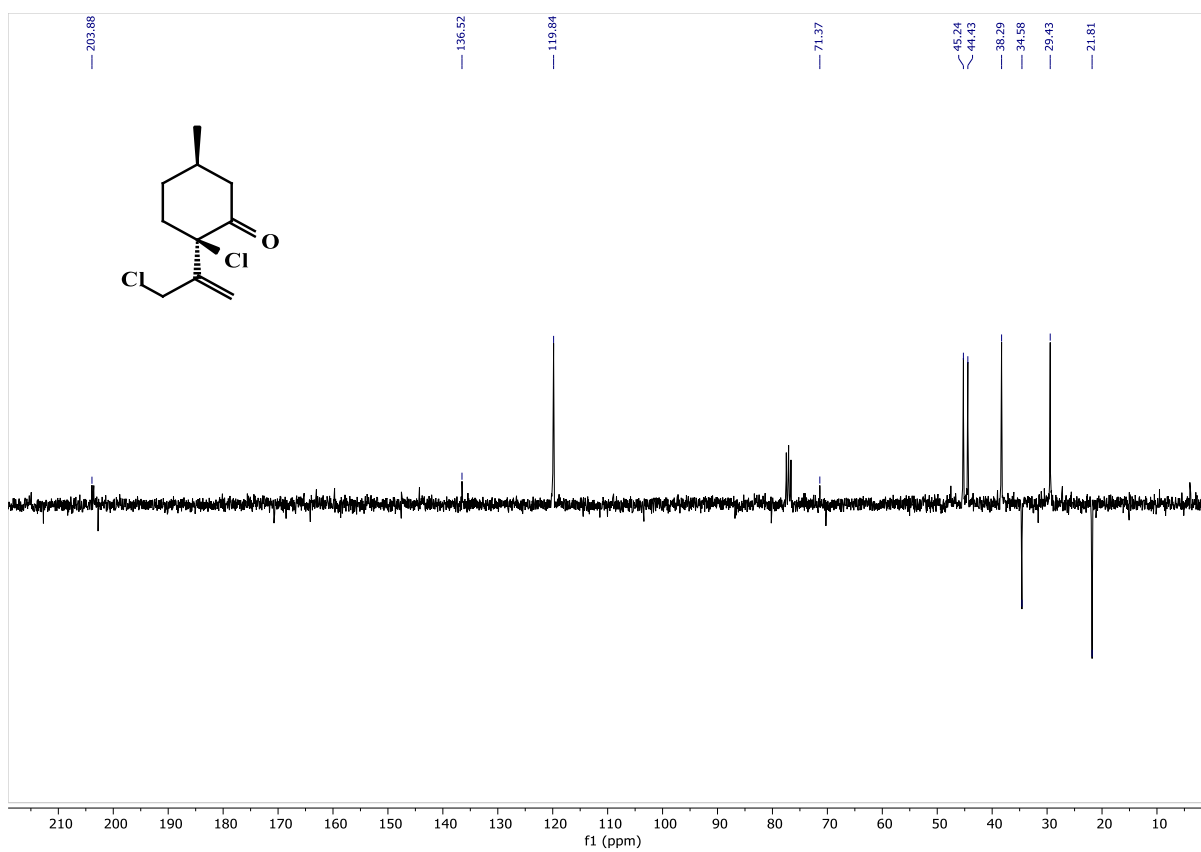


Figure S46: APT spectrum of the pulegone vinyl allyl dichloride **o**.

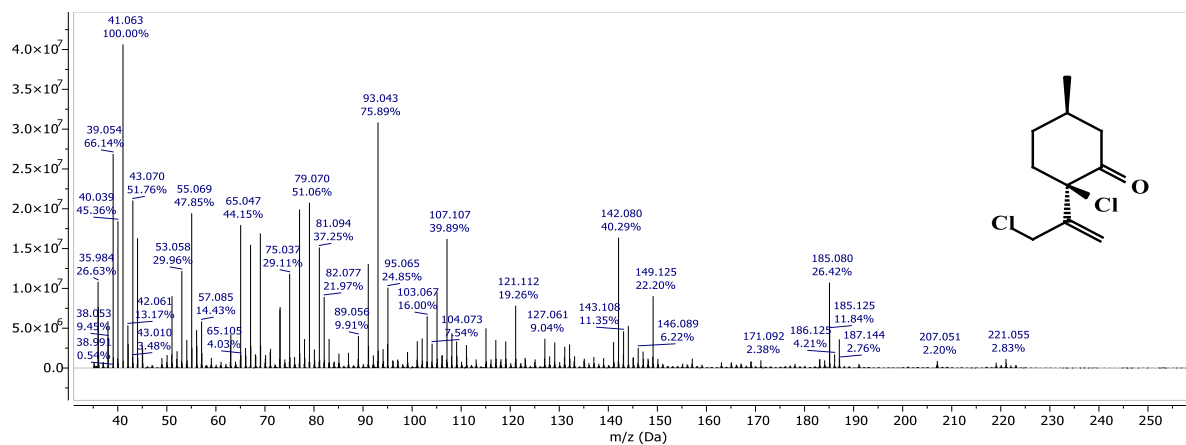


Figure S47: MS spectrum of the pulegone vinyl allyl dichloride **9**.

- Pulegone allyl dichloride **9'**

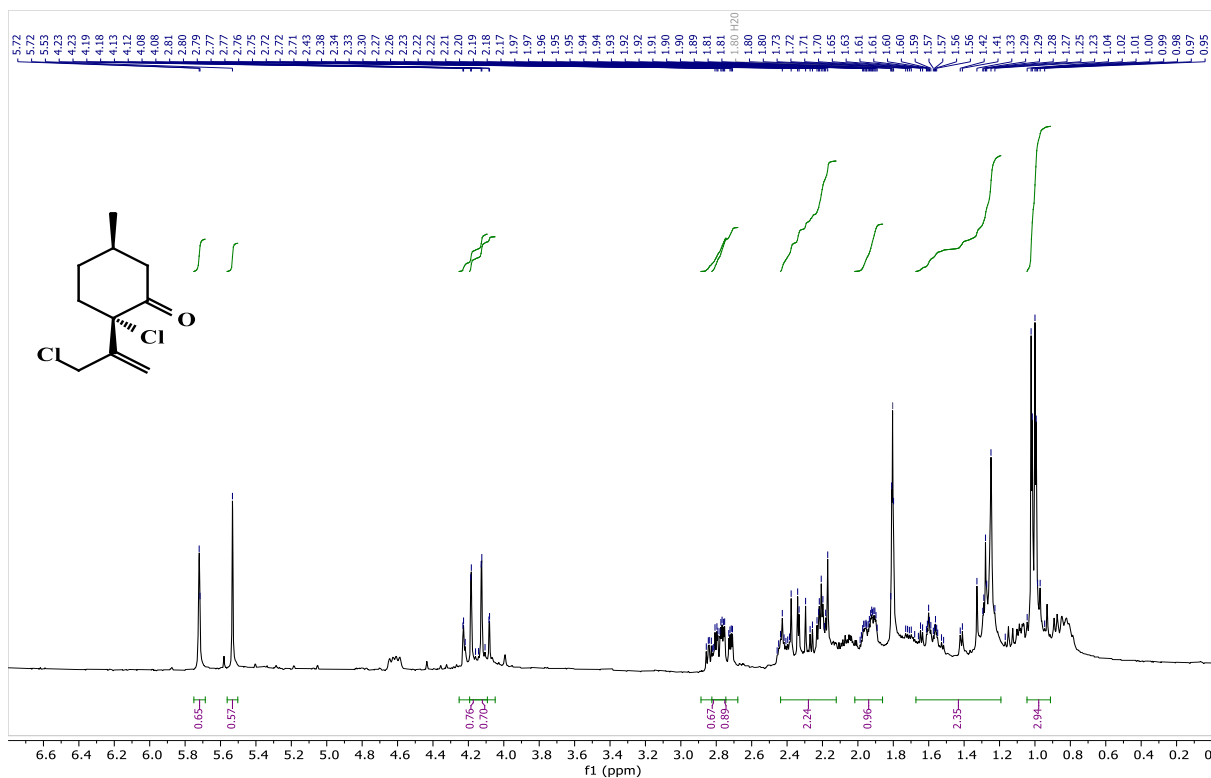


Figure S48: ^1H NMR spectrum of the pulegone vinyl allyl dichloride **9'**.

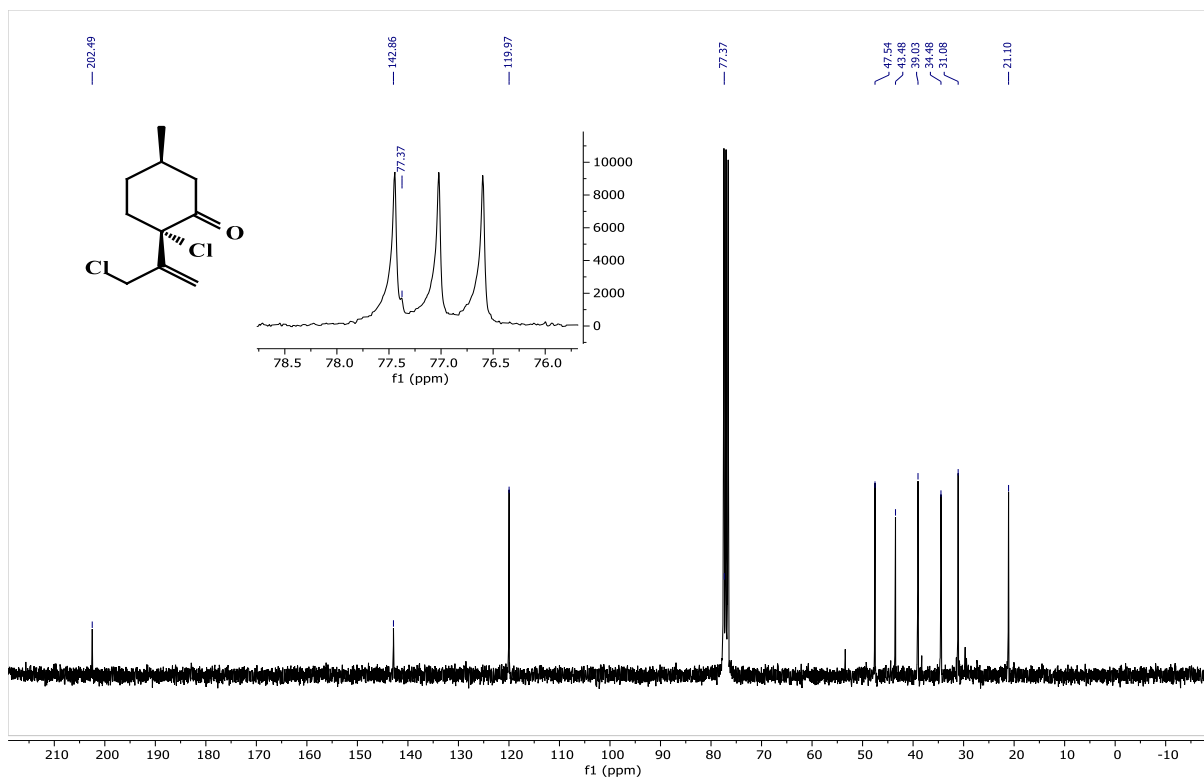


Figure S49: ^{13}C spectrum of the pulegone vinyl allyl dichloride o'.

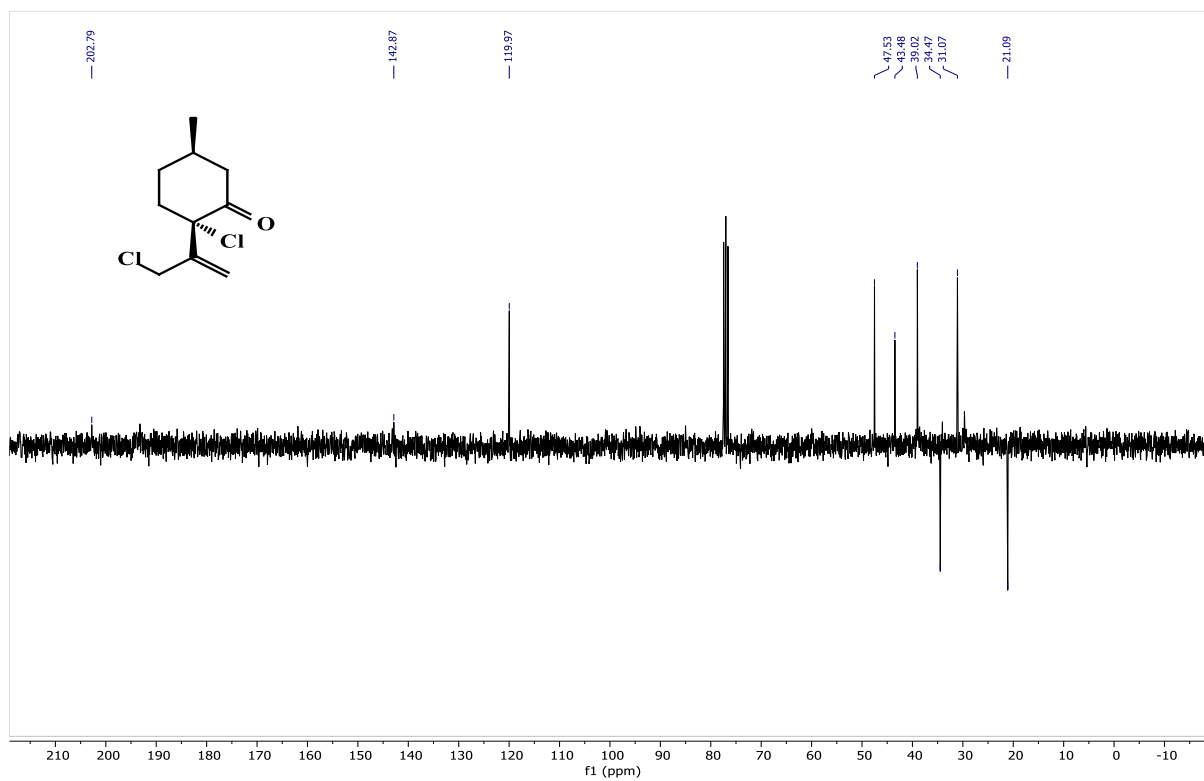


Figure S50: APT spectrum of the pulegone vinyl allyl dichloride o'.

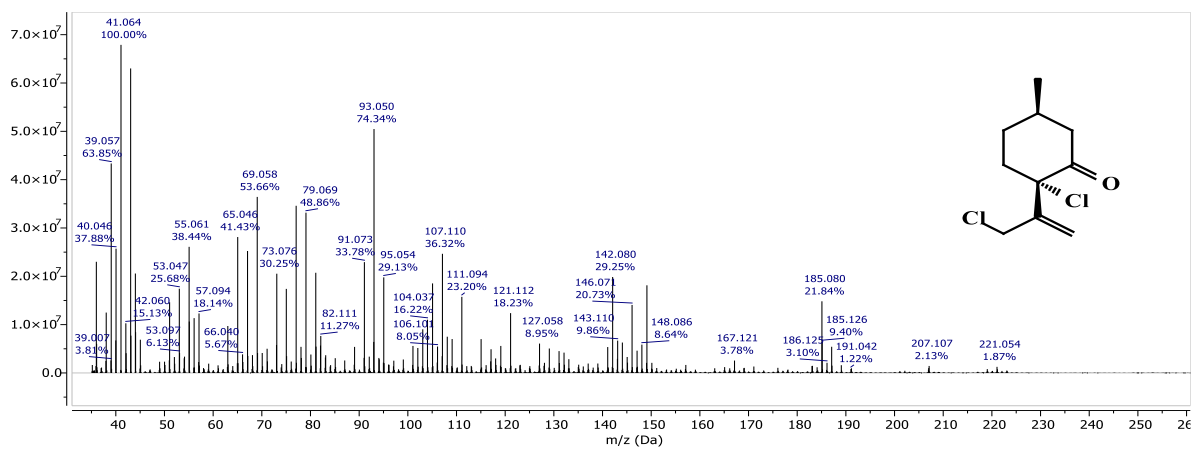


Figure S51: MS spectrum of the pulegone vinyl allyl dichloride **o'**.

➤ Derivatives of both perillyl aldehyde **p** and perillyl alcohol **s**

- Perillyl aldehyde monochloride **q**

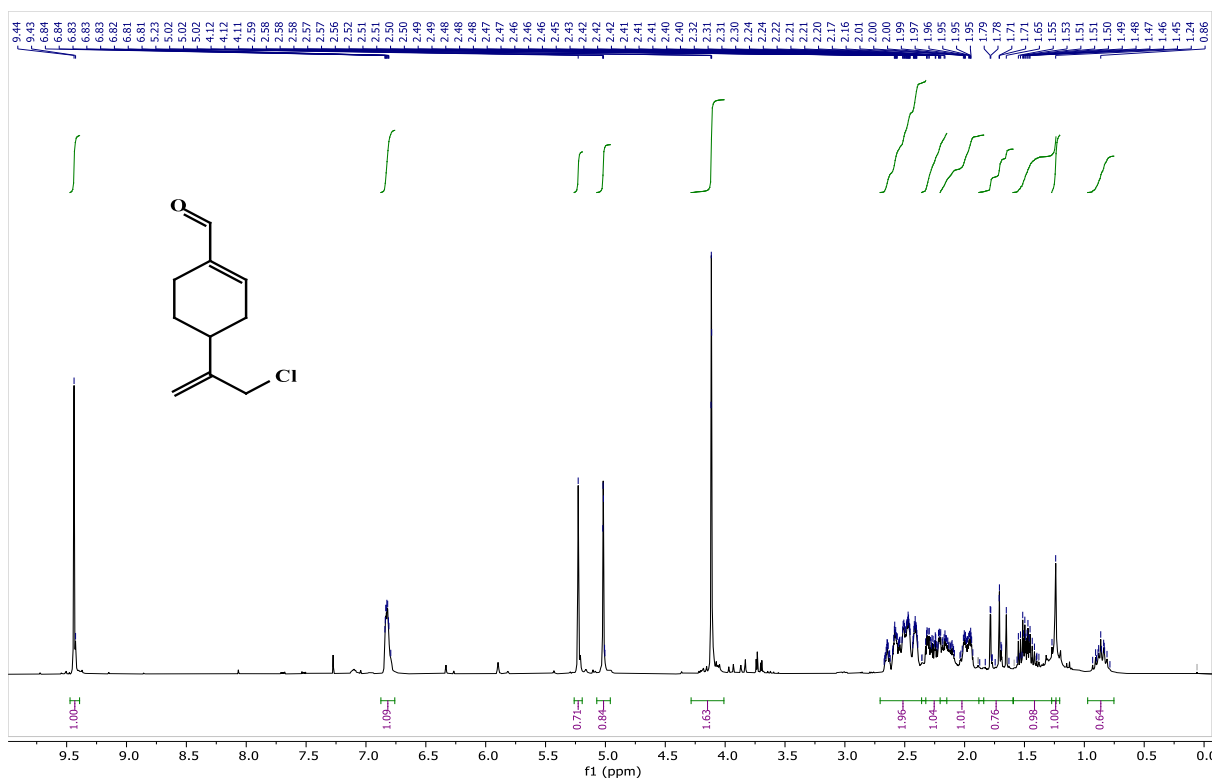


Figure S52: ^1H NMR spectrum of the perillyl aldehyde monochloride **q**.

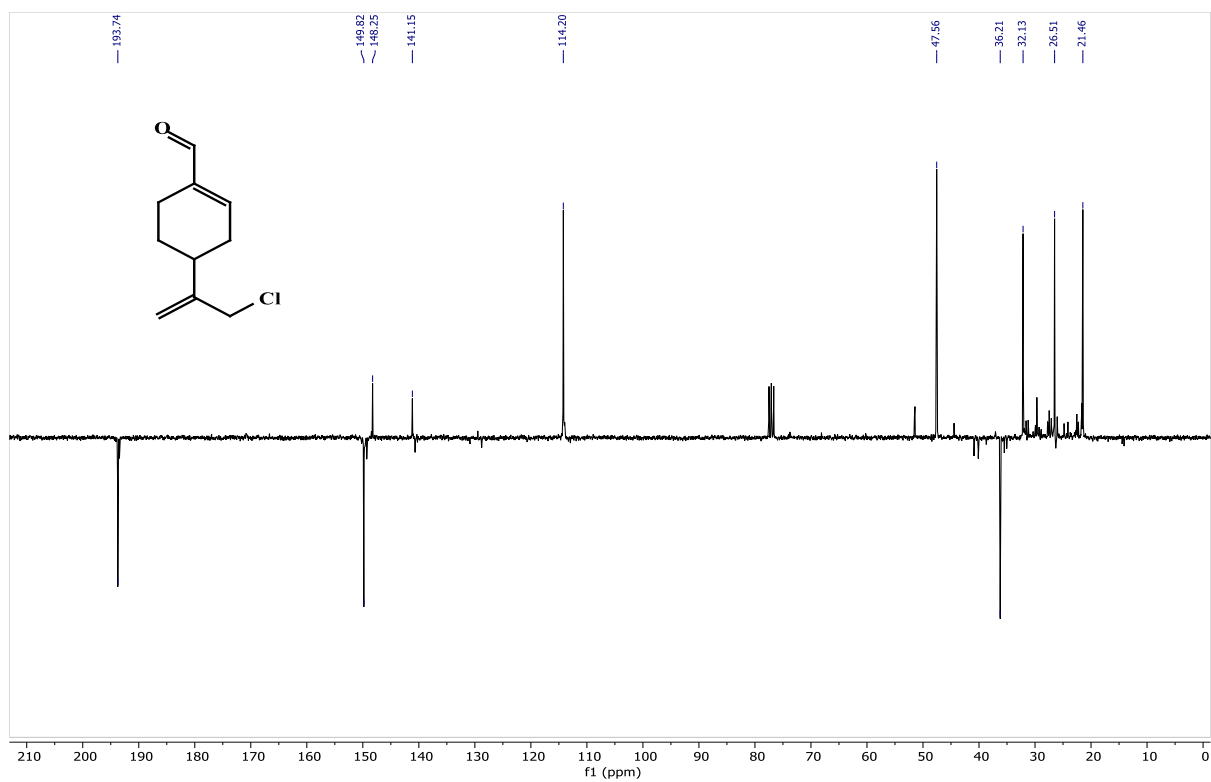


Figure S53: APT spectrum of the perillyl aldehyde monochloride **q**.

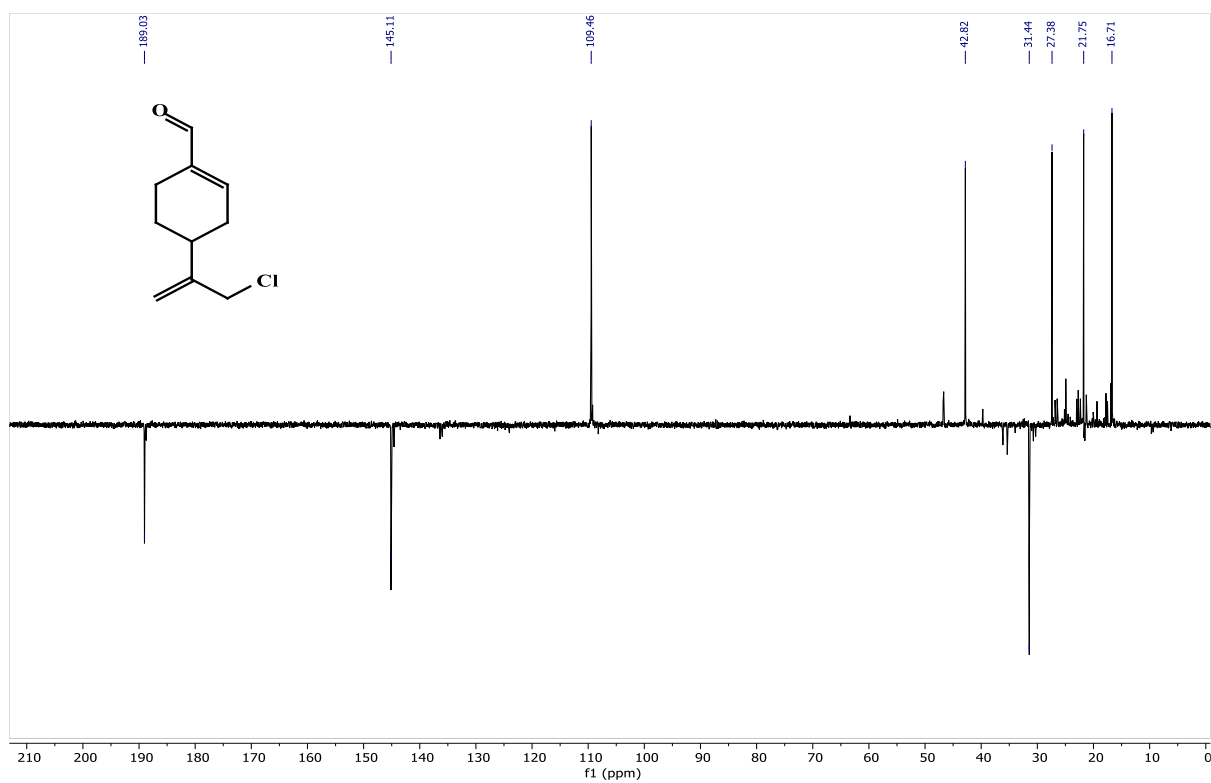


Figure S54: DEPT 135 spectrum of the perillyl aldehyde monochloride **q**.

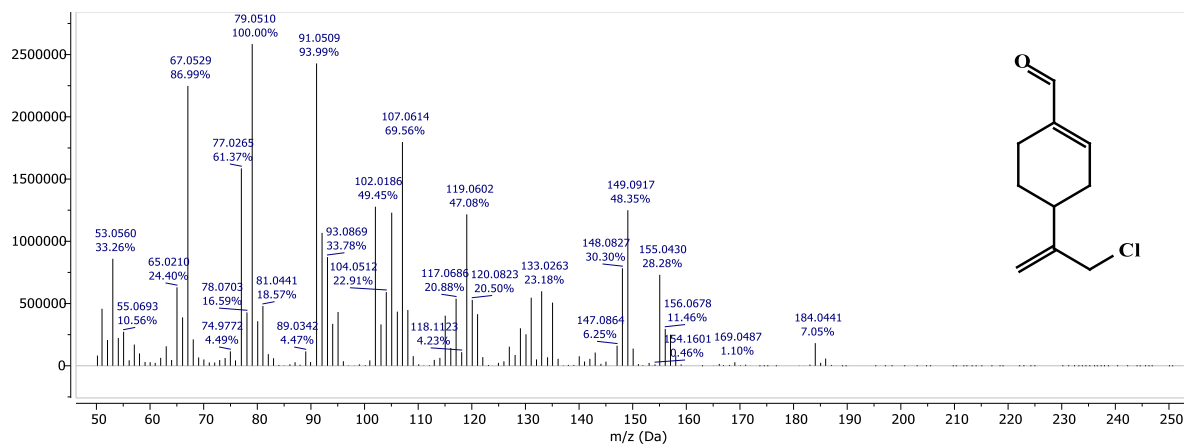


Figure S55: MS spectrum of the perillyl aldehyde monochloride **q**.

- Perillyl aldehyde vinyl allyl dichloride **r**

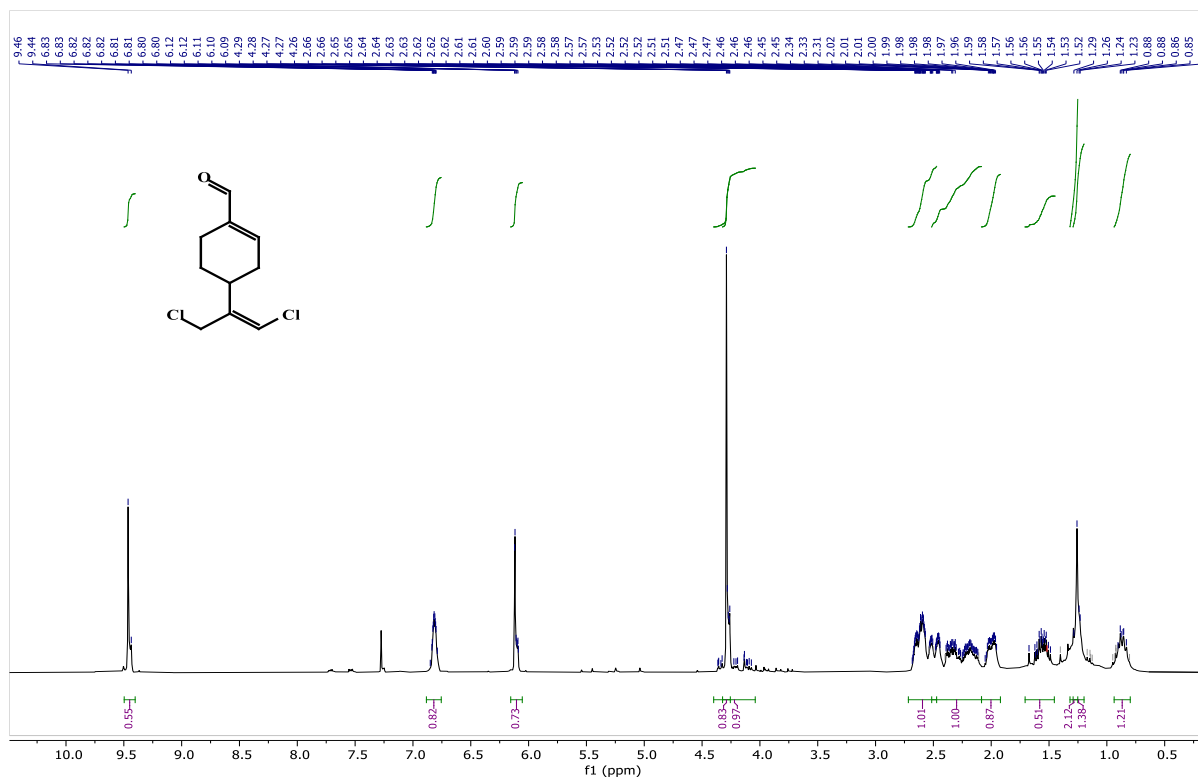


Figure S56: ^1H NMR spectrum of the perillyl aldehyde vinyl allyl dichloride **r**.

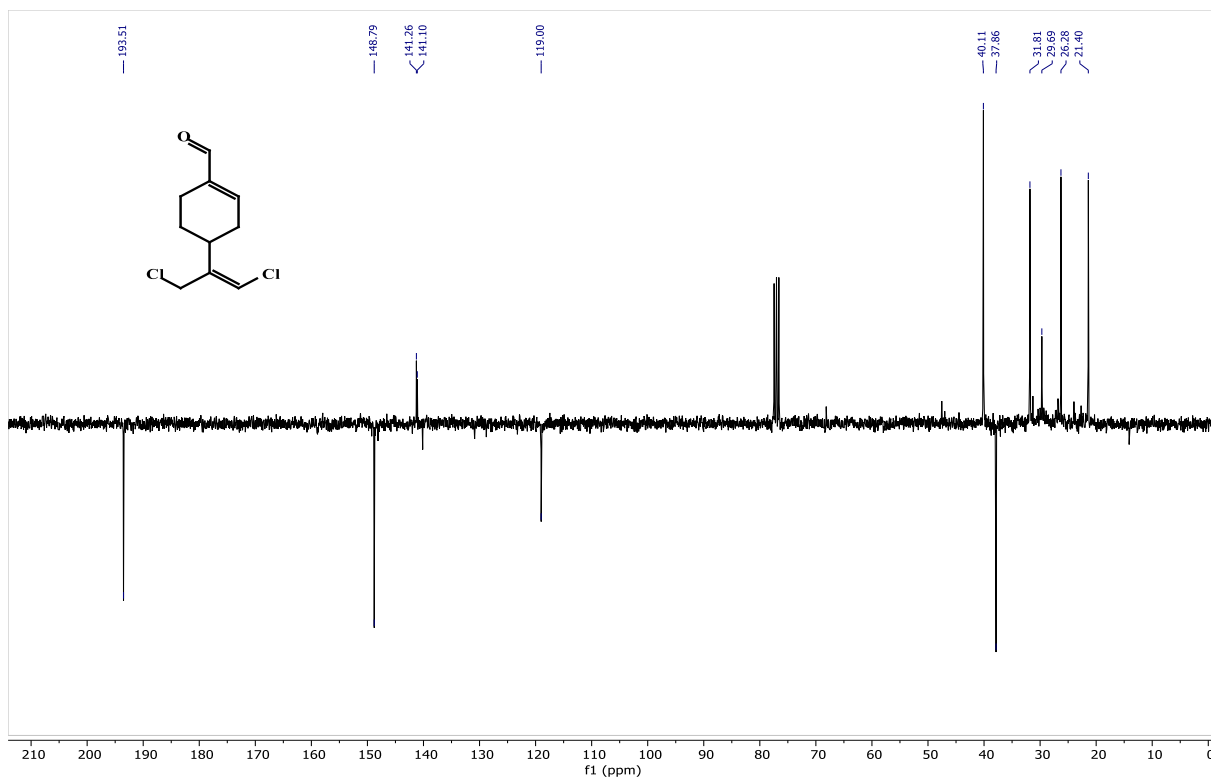


Figure S57: APT spectrum of the perillyl aldehyde vinyl allyl dichloride **r**.

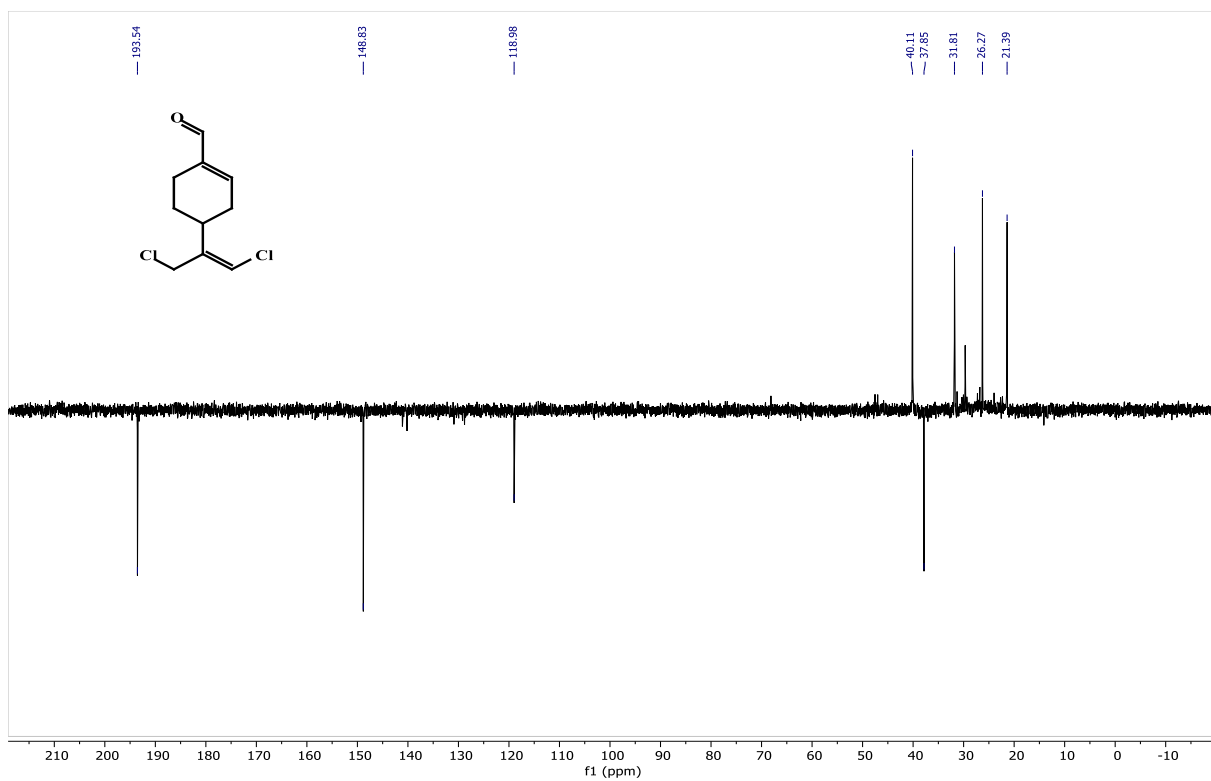


Figure S58: DEPT 135 spectrum of the perillyl aldehyde vinyl allyl dichloride **r**.

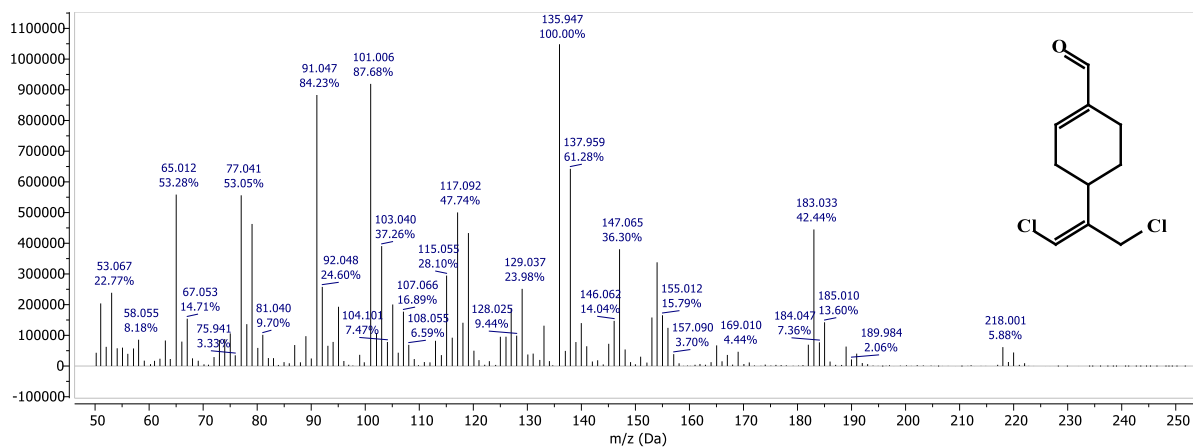


Figure S59: MS spectrum of the perillyl aldehyde vinyl allyl dichloride **r**.

➤ Derivatives of limona ketone **t**

- Limona ketone dichloride **u**

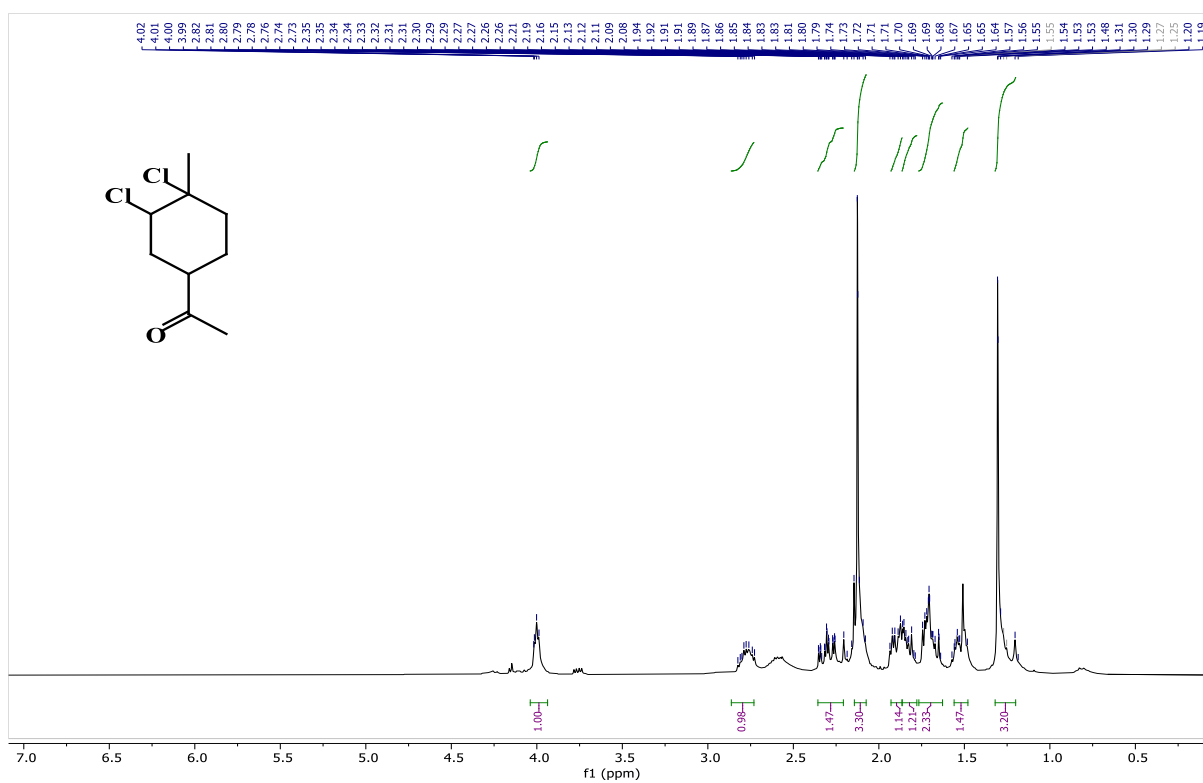


Figure S60: ^1H NMR spectrum of the limona ketone dichloride **u**.

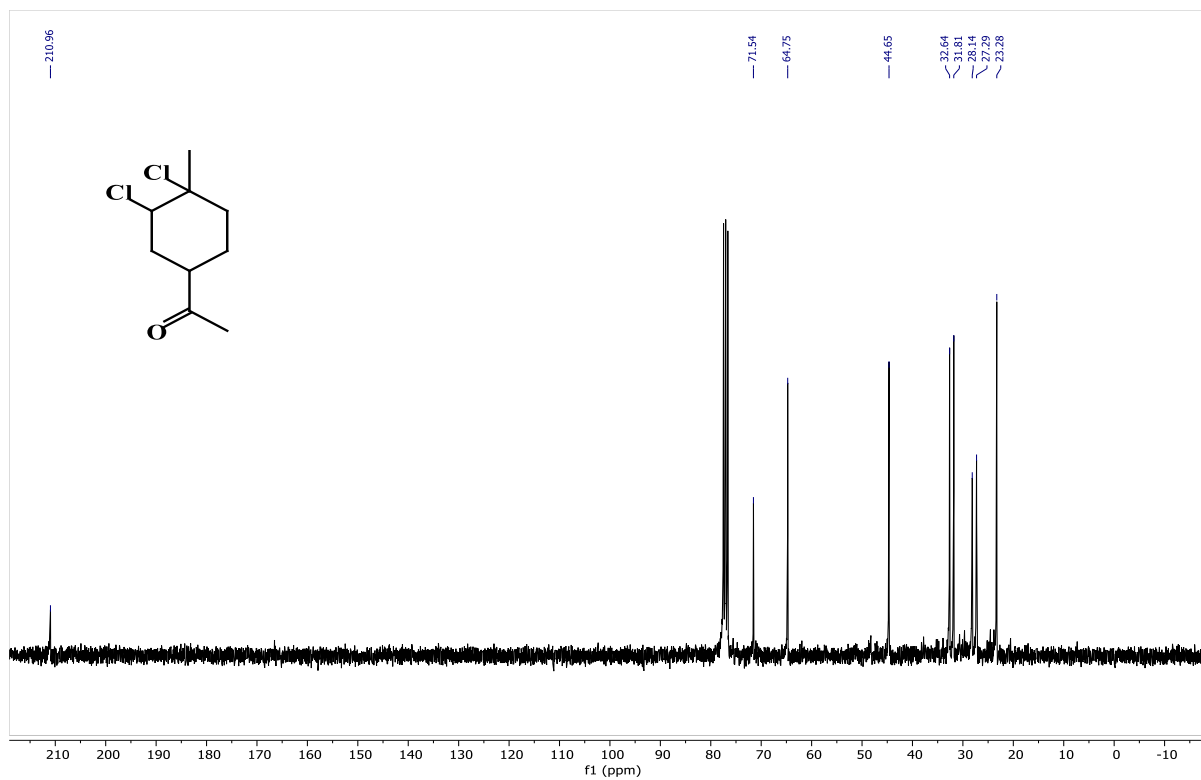


Figure S61: ^{13}C spectrum of the limona ketone dichloride **u**.

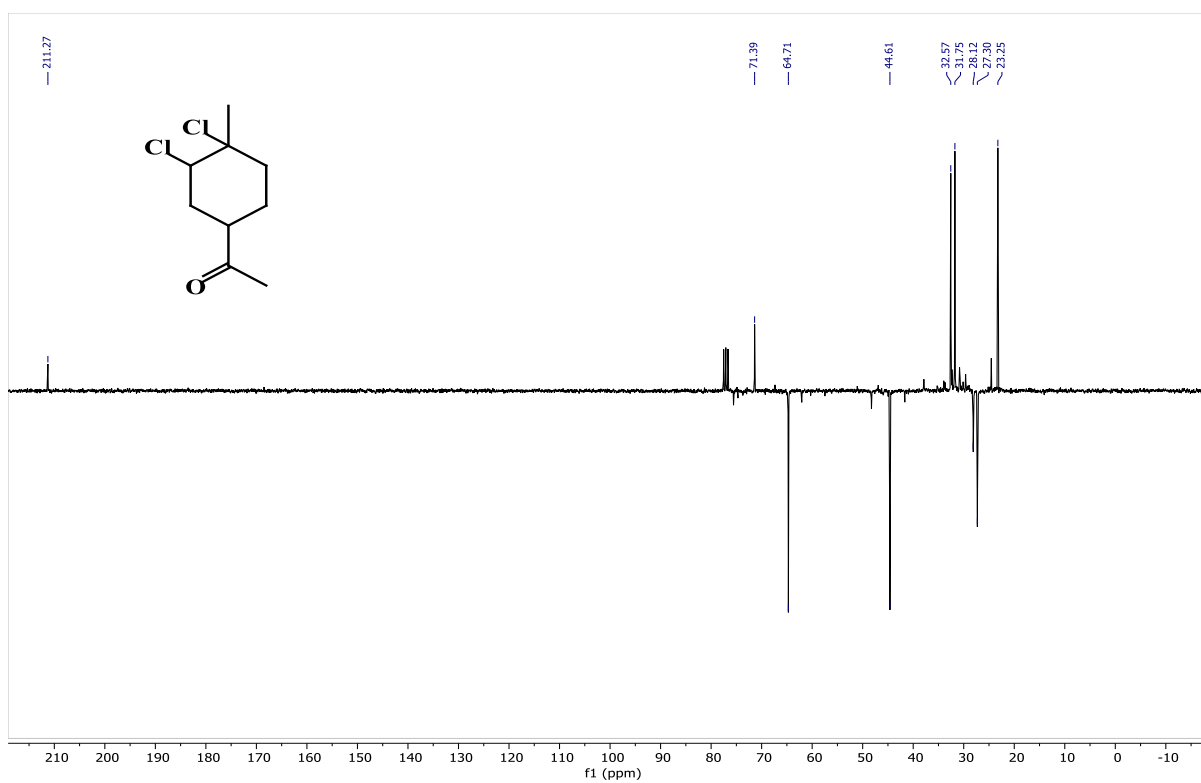


Figure S62: APT spectrum of the limona ketone dichloride **u**.

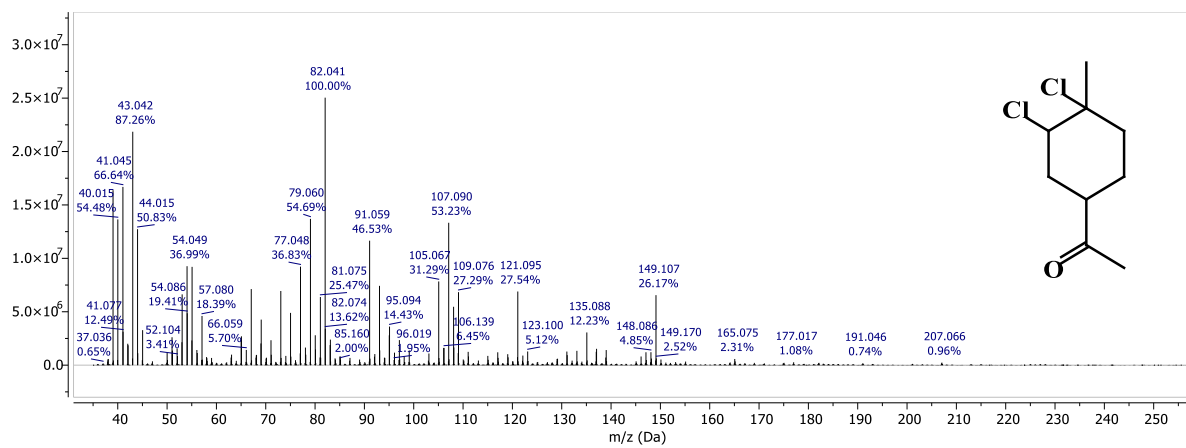


Figure S63: MS spectrum of the limona ketone dichloride **u**.

- Limona ketone epoxide **y**

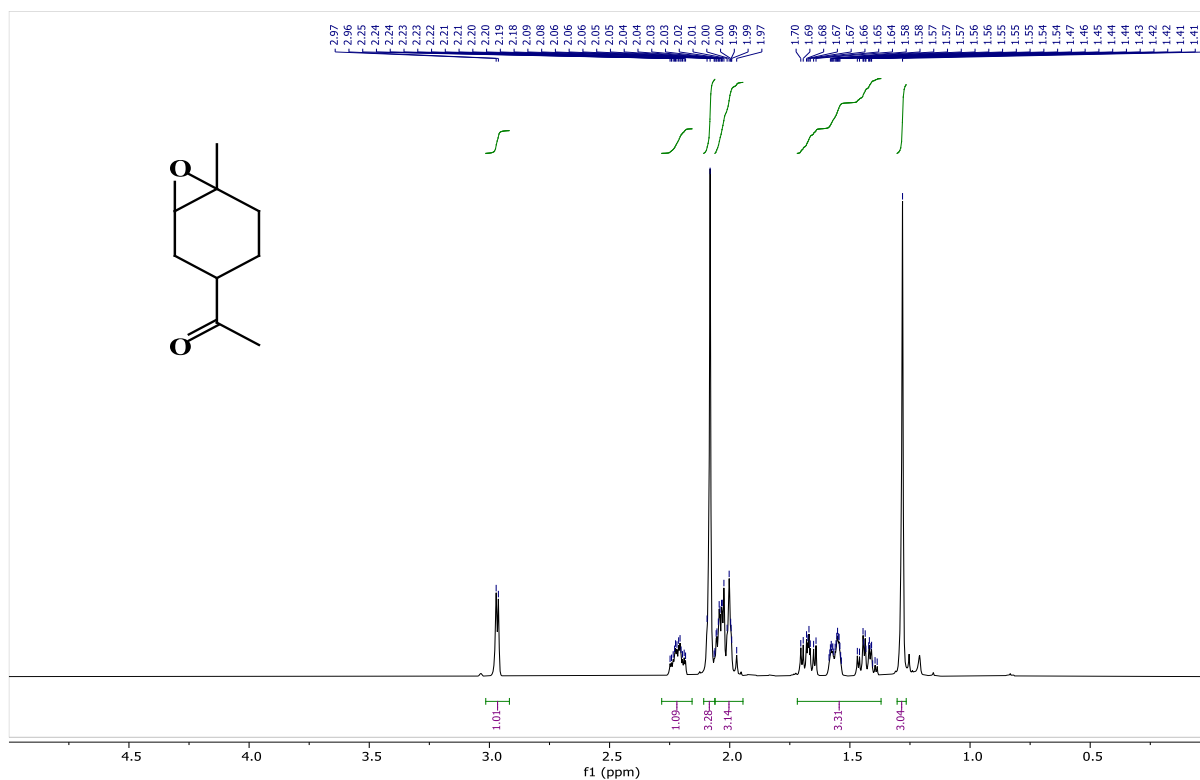


Figure S64: ^1H NMR spectrum of the limona ketone epoxide **y**.

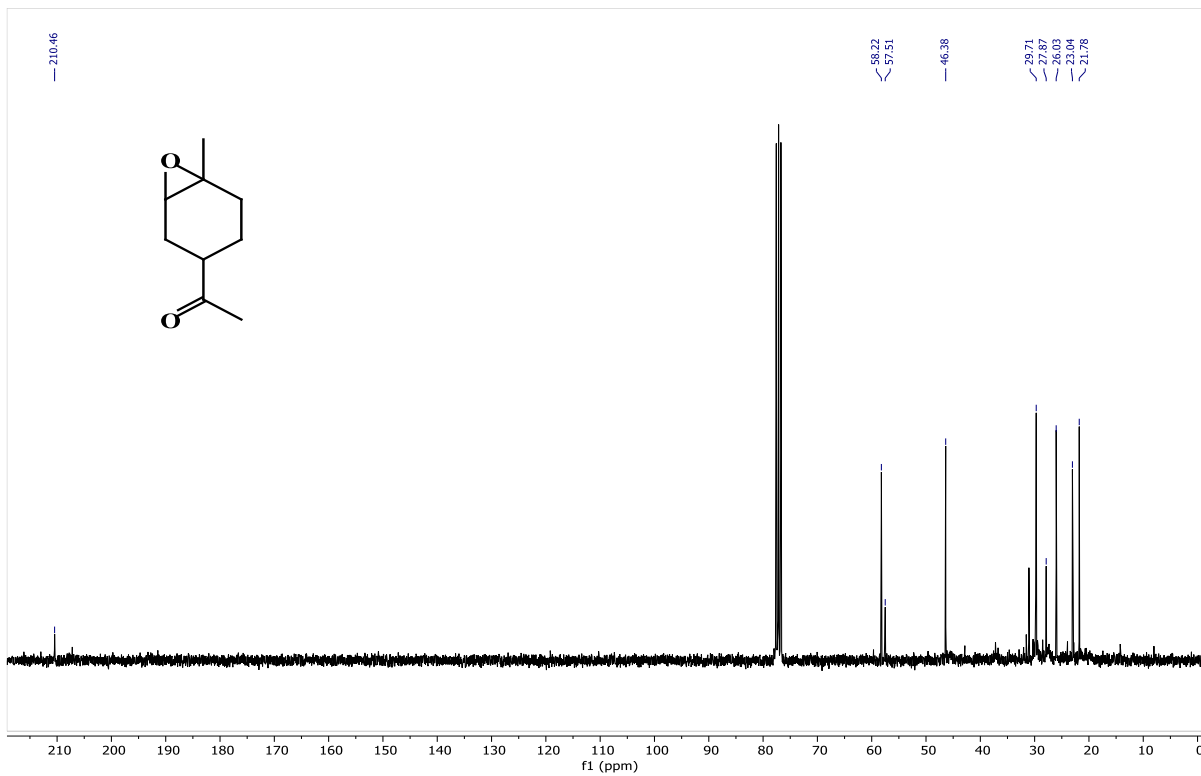


Figure S65: ^{13}C spectrum of the limona ketone epoxide v.

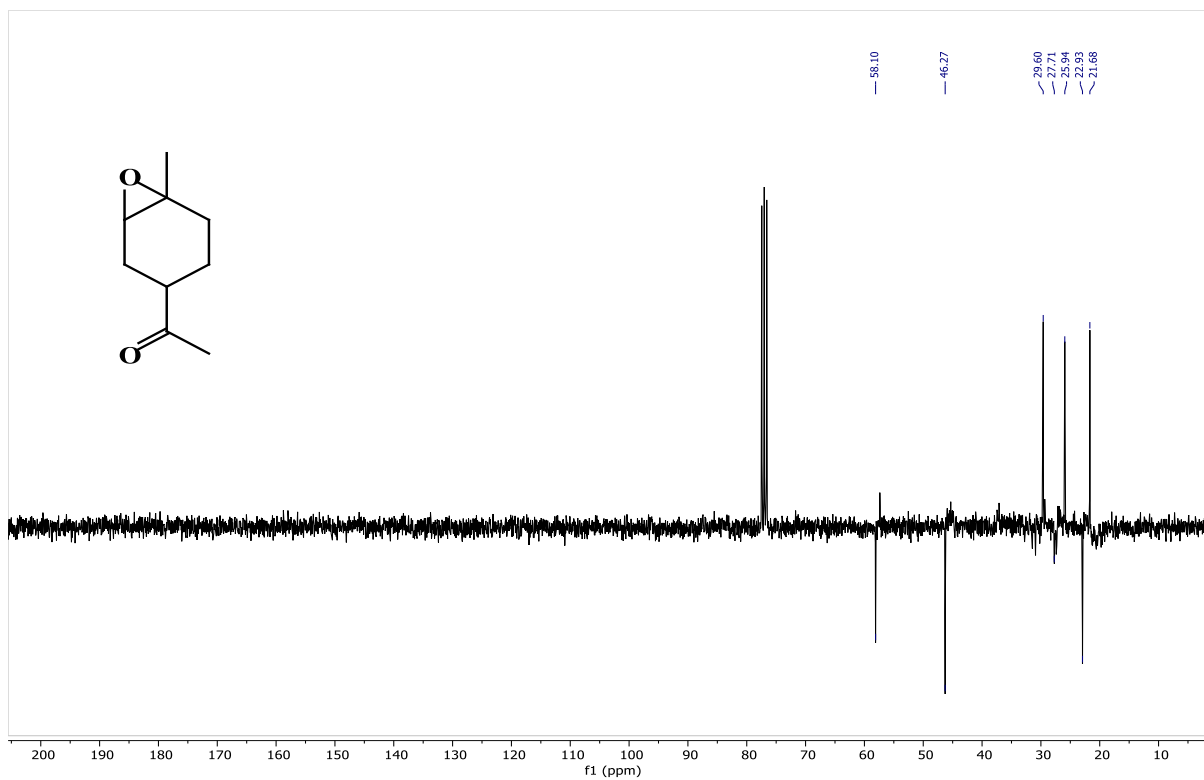


Figure S66: APT spectrum of the limona ketone epoxide v.

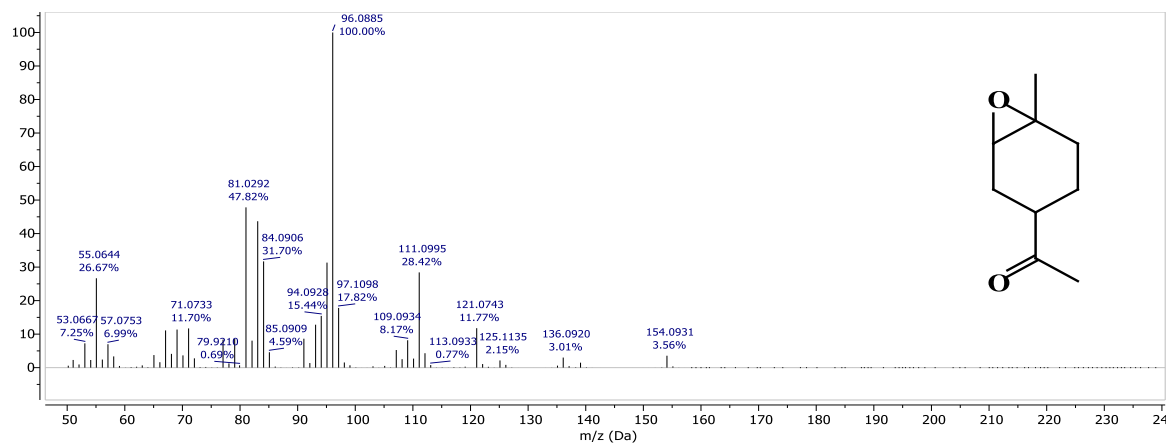


Figure S67: MS spectrum of the limona ketone dichloride **y**.